

Hierarchical Fe-modified MgAl₂O₄ as Ni-catalyst support for methane dry reforming

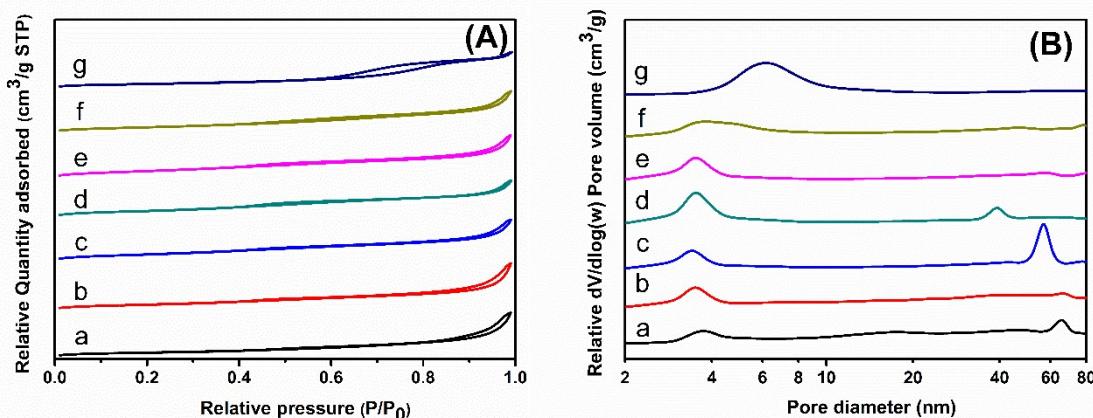
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Supplementary figures

Fig. S1. (A) N₂ adsorption-desorption isotherms and (B) corresponding pore size distributions of support materials calcined at 800 °C for 4 h. (a) Al₂O₃, (b) Mg-0.25, (c) Mg-0.5, (d) Mg-1.0, (e) FeMg-0.75, (f) FeMg-6.0, (g) MgAl₂O₄

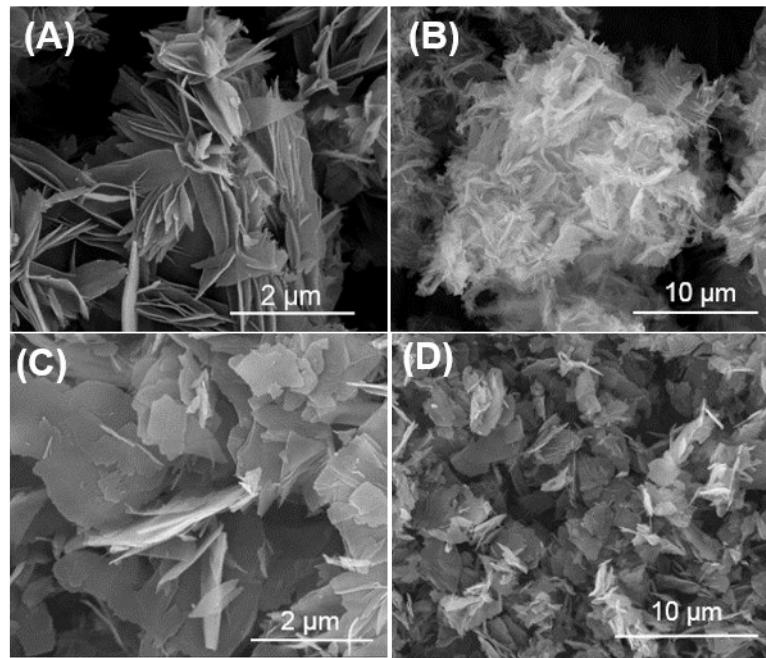


Fig. S2. SEM images of samples calcined at 800 °C for 4 h. (A) Mg-0.25, (B) Mg-0.5, (C) (D) Mg-1.0.

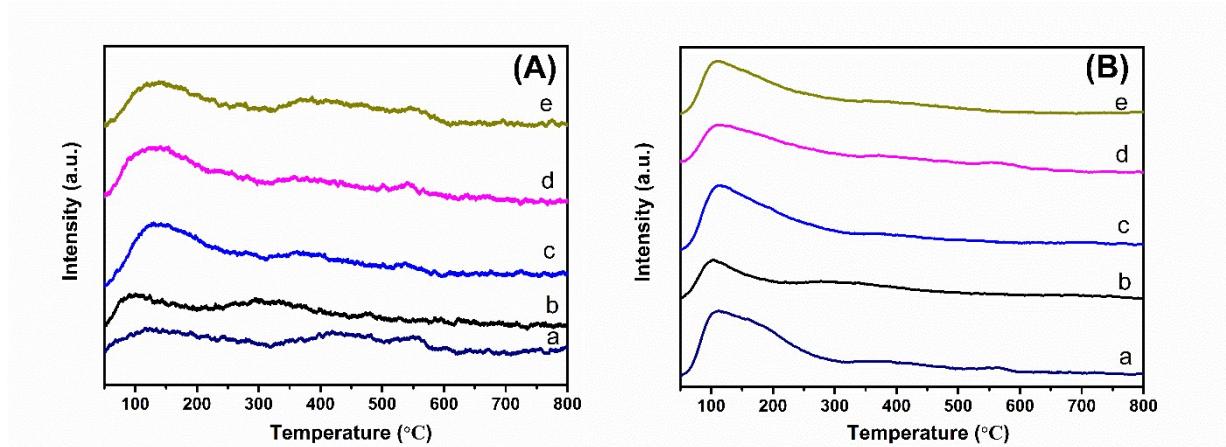


Fig. S3. (A) CO₂-TPD profiles and (B) NH₃-TPD profiles during thermal treatment of calcined support materials at 800 °C for 4 h. (a) MgAl₂O₄, (b) Al₂O₃, (c) Mg-0.5, (d) FeMg-0.75, (e) FeMg-6.0. Respective pre-treatments: 60 ml/min 4% NH₃/He or 100% CO₂ at room temperature for 1 h.

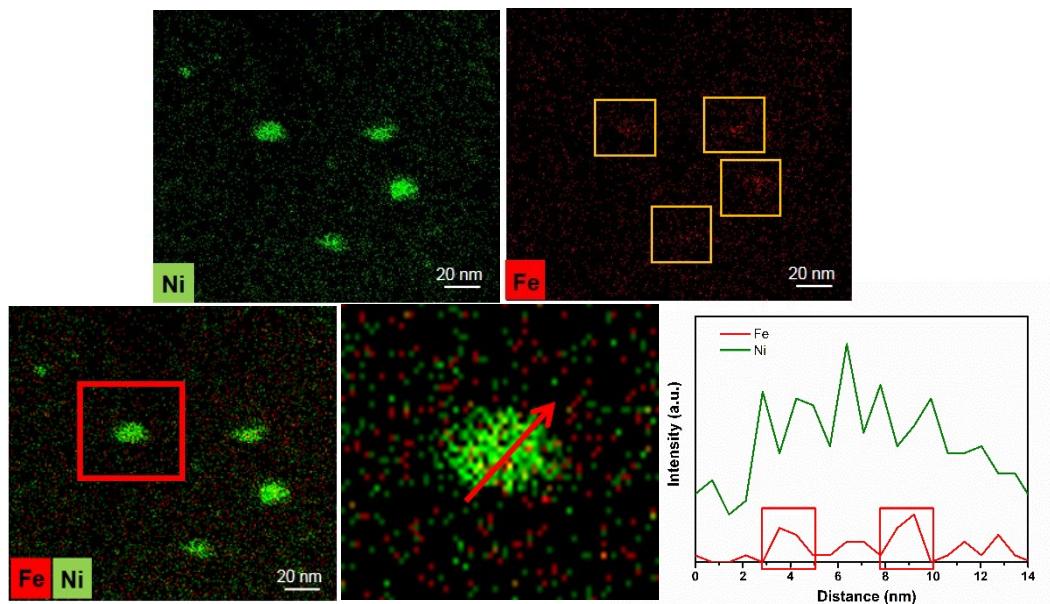


Fig. S4. EDX mapping and line scan of reduced Ni/FeMg-0.75.

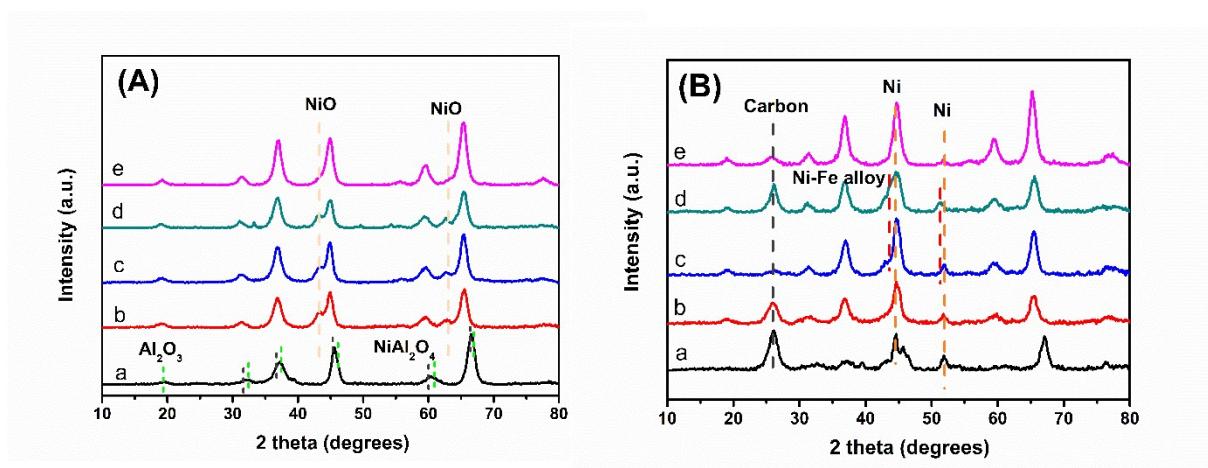


Fig. S5. XRD scans of (A) calcined catalysts and (B) spent catalysts after DRM 750 °C for 12 h (200 ml/min CH₄:CO₂:Ar with volumetric ratio of 1:1:3, 111.3 kPa, W_{Ni}/F_{CH4}⁰ = 0.33 kg_{Ni} . s . mol_{CH4}⁻¹). (a) Ni/Al₂O₃, (b) Ni/Mg-0.5, (c) Ni/FeMg-0.75, (d) Ni/FeMg-6.0, (e) Ni/MgAl₂O₄.

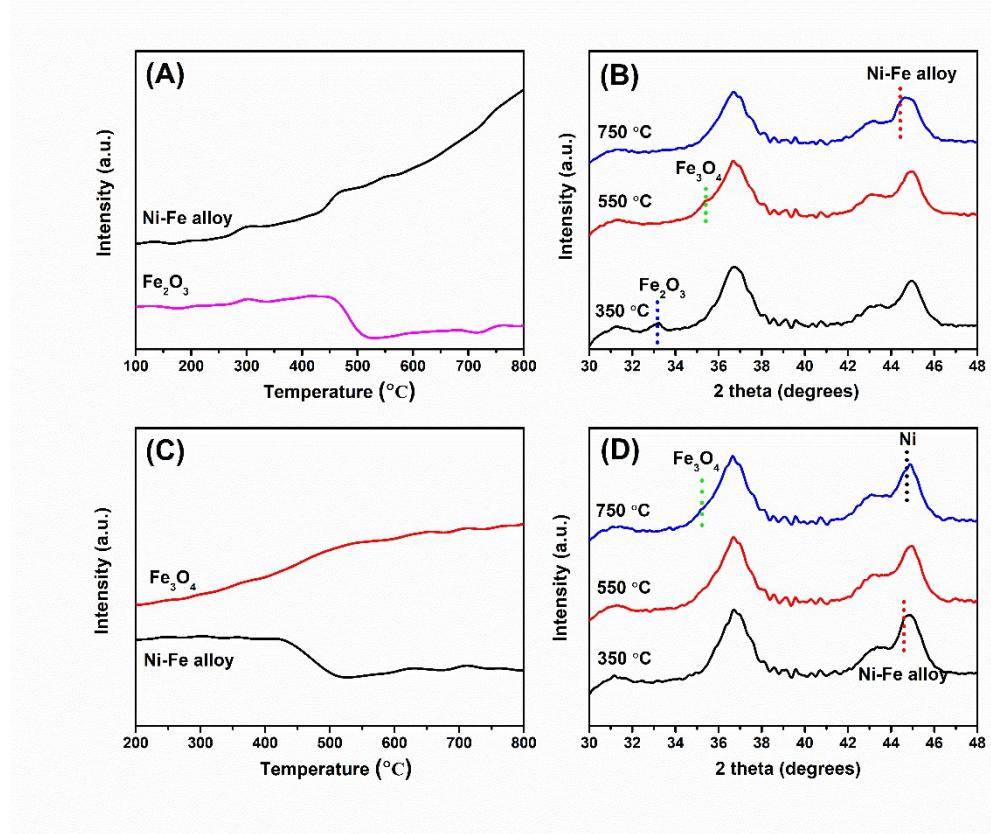


Fig. S6. In situ XRD for Ni/FeMg-6.0. (A) integral intensity variation during H_2 -TPR for diffraction areas $32.5\text{--}33.5^\circ$ (Fe_2O_3) and $43.7\text{--}44.2^\circ$ (Fe-Ni alloy); (B) diffraction peaks at different temperature during H_2 -TPR; (C) integral intensity variation during CO_2 -TPO for diffraction areas $35\text{--}36^\circ$ (Fe_3O_4) and $43.7\text{--}44.2^\circ$ (Fe-Ni alloy); (D) diffraction peaks at different temperature during CO_2 -TPO.

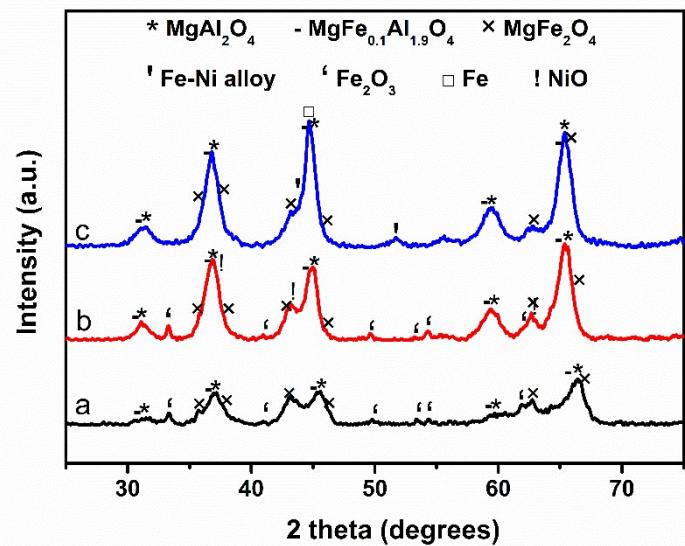


Fig. S7. XRD scans of Ni/FeMg-6.0 sample in different states. (a) FeMg-6.0 support, (b) calcined catalyst, (c) reduced catalyst.

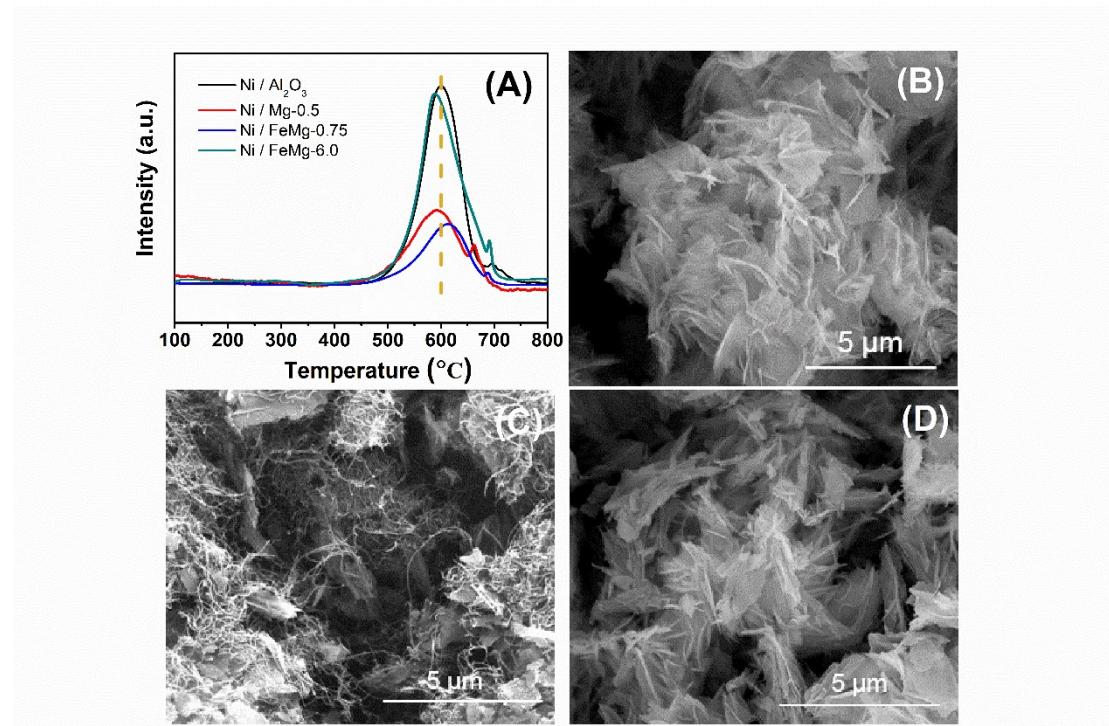


Fig. S8. (A) O₂-TPO profiles for spent hierarchical catalysts and SEM images for Ni/FeMg-0.75 sample in different stages: (B) before reaction, (C) after reaction, (D) after O₂-TPO.

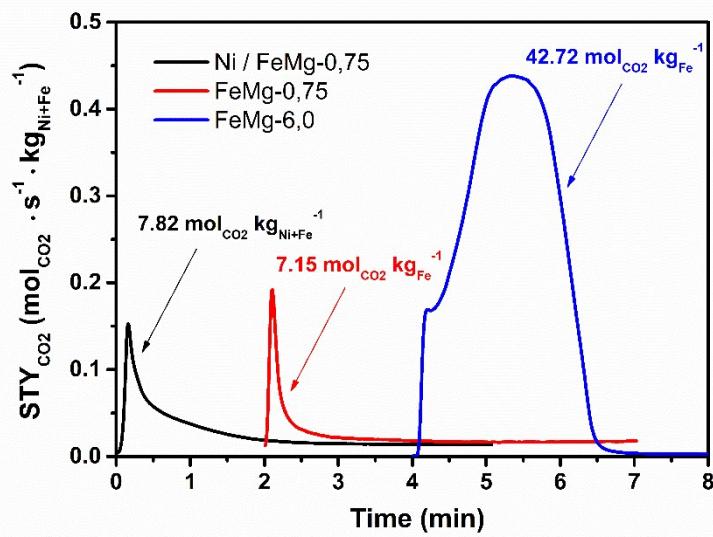


Fig. S9. CO₂ space-time yield (STY, mol_{CO₂} · s⁻¹ · kg_{Ni+Fe}⁻¹) upon O₂ oxidation of deposited carbon for the three different spent materials (using a 2 min offset for the different curves).