

Graphene Oxide (GO) Catalysed MW-assisted One-Pot Synthesis of Densely Substituted Furan

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Experimental Section

General information

All chemicals and solvents were purchased from commercial suppliers and used without further purification. All reactions were carried out in oven-dried glassware with magnetic stirring. Commercially available chemicals were used without further purification. The starting materials aryl glyoxals (**1**)¹ and graphene oxide² were synthesized according to the reported procedure. Monitoring of the reaction was performed by thin-layer chromatography using TLC silica gel 60 F254 plates. Melting points were measured in open capillary tubes and were uncorrected. Flash column chromatography was performed using silica gel 60 (particle size 0.040–0.063 mm) typically using a n-hexane/ethyl acetate eluent system. FT-IR spectra were recorded with PerkinElmer 782 spectrophotometer. NMR spectra were calibrated to the solvent residual signals of CDCl₃. ¹H NMR spectra were recorded at a 300 MHz and 500 MHz Bruker instrument. Data are reported as follows: chemical shift, multiplicity (s: singlet, d: doublet, t: triplet, q: quartet, qui: quintuplet, m: multiplet), coupling constant (*J* in Hz) and integration. ¹³C NMR spectra were recorded at 75 MHz and 125 MHz using broadband proton decoupling and chemicals shifts are reported in ppm using residual solvent peaks as reference. High resolution mass spectra (HRMS) were recorded on a QTOF I (quadrupole-hexapole-TOF) mass spectrometer with an orthogonal Z-spray-electrospray interface. Reported yields are based upon isolation following purification by silica gel column chromatography; isolated material was judged to be homogeneous based upon TLC and NMR.

General procedure of the one-pot furan synthesis: Glyoxal **1** (2.0 mmol), 1,3 diketone **2** (2.0 mmol) and compound **3** (0.8 mmol) were taken in a MW-tube and radiated in MW reactor at 150W for 6 min (3 min X 2). Then, graphene oxide (60 mg) was added to the reaction vessel and whole mixture was heated at 70 °C for 5 h. Then, EtOH (5 mL) was added to the reaction vessel and the product was extracted from the surface of graphene oxide (GO) by ultrasonication. The graphene oxide (GO) was separated by decantation. The separated organic layer was evaporated to get the crude mass which was purified by column chromatography (20% ethyl acetate/n-hexane) to afford the desired product.

Compound **4a**: Yellow solid, mp 170 °C. δ_H (500 MHz, Chloroform-*d*) 8.47 (1 H, s), 7.44–7.39 (2 H, m), 7.31 – 7.21 (3 H, m), 7.18 – 7.05 (2 H, m), 6.95 (2 H, d, *J* 8.0), 2.68 (3 H, s), 2.23 (3 H, s), 1.84 (3 H, s). ¹³C NMR (126 MHz, CDCl₃) δ 196.9, 157.5, 148.9, 137.2, 136.3, 129.1, 128.0, 127.9, 125.4, 125.2, 123.8, 122.8, 120.5, 119.9, 112.5, 111.5, 108.9, 30.1, 21.3, 14.8. ESI-MS: *m/z* =[M+H]⁺calcd. for C₂₂H₂₀NO₂: 330.1489; found 330.1487.

Compound **4b**: δ_H (300 MHz, Chloroform-*d*) 8.54 (1 H, s), 7.36 – 7.25 (3 H, m), 7.24 – 7.08 (2 H, m), 7.06 – 6.96 (5 H, m), 2.60 (3 H, s), 1.76 (3 H, s). δ_C (75 MHz, Chloroform-*d*) 197.1, 157.8, 148.6, 136.3, 133.0, 130.6, 128.5, 127.9, 127.3, 126.4, 125.2,

123.8, 122.7, 120.5, 119.8, 113.3, 111.5, 108.5, 30.1, 14.9 . **ESI-MS:** $m/z = [M+H]^+$ calcd. for $C_{22}H_{18}NO_2$: 316.1332; found 316.1334.

Compound **4c**: Brown viscous liquid. δ_H (300 MHz, Chloroform-*d*) 8.50 (1 H, s), 7.36 (1 H, dt, *J* 8.2, 0.9), 7.30-7.26 (1 H, m), 7.24 – 7.13 (3 H, m), 7.09 – 6.95 (4 H, m), 2.60 (3 H, s), 1.76 (3 H, s). ^{13}C NMR (75 MHz, CDCl₃) δ 197.0, 158.0, 147.9, 136.2, 133.0, 129.2, 128.6, 127.5, 126.5, 125.7, 123.7, 123.0, 120.6, 119.9, 113.8, 111.5, 108.2, 30.0, 14.9. **ESI-MS:** $m/z = [M+H]^+$ calcd. for $C_{21}H_{17}ClNO_2$: 350.0942; found 350.0942.

Compound **4d**: Yellow solid, mp 180 °C.. δ_H (500 MHz, Chloroform-*d*) 8.51 (1 H, s), 7.46 (1 H, d, *J* 8.2), 7.37 (1 H, d, *J* 7.9), 7.32 – 7.20 (5 H, m), 7.18 – 7.08 (2 H, m), 2.68 (3 H, s), 1.84 (3 H, s). ^{13}C NMR (126 MHz, CDCl₃) δ 196.7, 158.0, 147.7, 136.3, 131.5, 129.5, 127.6, 126.7, 125.6, 123.7, 123.0, 121.2, 120.8, 119.8, 114.0, 111.6, 108.4, 30.1, 14.9. **ESI-MS:** $m/z = [M+H]^+$ calcd. for $C_{21}H_{17}BrNO_2$: 394.0437; found 394.0436.

Compound **4e**: Yellow solid, mp 180 °C. δ_H (300 MHz, Chloroform-*d*) 8.3 (1 H, s), 7.4 – 7.3 (4 H, m), 7.2 – 7.1 (2 H, m), 7.06 - 7.01 (1 H, m), 6.8 – 6.6 (2 H, m), 4.0 (2 H, q, *J* 7.1), 3.7 (3 H, s), 2.7 (3 H, s), 0.9 (3 H, t, *J* 7.1). δ_C (75 MHz, Chloroform-*d*) 164.5, 158.8, 157.9, 148.7, 136.1, 128.0, 126.9, 123.9, 123.7, 122.0, 120.3, 119.8, 116.4, 113.8, 112.7, 111.1, 108.6, 55.3, 14.7, 59.9, 13.7. **ESI-MS:** $m/z = [M+H]^+$ calcd. for $C_{23}H_{22}NO_4$: 376.1543; found 376.1545.

Compound **4f**: Light yellow solid, mp 140 °C. δ_H (500 MHz, Chloroform-*d*) 8.35 (1 H, s), 7.38 (1 H, d, *J* 8.2), 7.3-7.25 (4 H, m), 7.21 – 6.99 (4 H, m), 4.01 (2 H, q, *J* 7.1), 2.70 (3 H, s), 0.86 (3 H, t, *J* 7.1). δ_C (126 MHz, Chloroform-*d*) 164.2, 158.7, 147.6, 136.2, 131.4, 129.7, 127.6, 126.9, 123.9, 122.2, 121.1, 120.2, 120.0, 116.7, 115.1, 111.2, 107.9, 60.0, 29.9, 14.5, 13.7. **ESI-MS:** $m/z = [M+H]^+$ calcd. for $C_{22}H_{19}BrNO_3$: 424.0543; found 424.0542.

Compound **4g**: Yellow liquid. δ_H (500 MHz, Chloroform-*d*) 8.41 (1 H, s), 7.47 (1 H, d, *J* 8.2), 7.40 (1 H, d, *J* 7.9), 7.30 – 7.24 (2 H, m), 7.20 (1 H, d, *J* 2.4), 7.16 – 7.06 (2 H, m), 6.93 (1 H, dd, *J* 5.1, 1.3), 2.68 (3 H, s), 1.87 (3 H, s). δ_C (126 MHz, Chloroform-*d*) 196.6, 157.4, 146.7, 136.3, 131.7, 127.9, 125.5, 125.2, 125.0, 123.9, 122.9, 120.7, 120.4, 120.0, 112.1, 111.5, 108.7, 30.2, 14.9. **ESI-MS:** $m/z = [M+H]^+$ calcd. for $C_{19}H_{16}NO_2S$: 322.0896; found 322.0895.

Compound **4h**: Yellow solid, mp 210 °C. δ_H (300 MHz, Chloroform-*d*) 8.52 (1 H, s), 7.46 - 7.43 (1 H, m), 7.29 – 7.18 (4 H, m), 7.17 – 7.05 (3 H, m), 2.61 (3 H, s), 1.79 (3 H, s). δ_C (75 MHz, Chloroform-*d*) 196.2, 158.3, 148.0, 134.9, 131.6, 129.4, 129.3, 126.7, 126.0, 125.4, 124.9, 122.2, 121.5, 114.1, 113.2, 108.2, 100.1, 30.2, 15.0. **ESI-MS:** $m/z = [M+H]^+$ calcd. for $C_{21}H_{16}Br_2NO_2$: 471.9542; found 471.9542.

Compound 4i: Brown viscous liquid. δ_{H} (300 MHz, Chloroform-*d*) 8.29 (1 H, s), 7.37–7.30 (2 H, m), 7.20 – 7.13 (1 H, m), 7.11 – 7.00 (2 H, m), 2.51 (3 H, s), 2.08 (3 H, s), 1.83 (3 H, s). δ_{C} (75 MHz, Chloroform-*d*) 197.0, 156.7, 148.3, 136.1, 128.1, 123.6, 122.6, 120.3, 119.8, 112.6, 111.4, 108.9, 30.0, 14.6, 12.0. **ESI-MS:** $m/z = [\text{M}+\text{H}]^+$ calcd. for $\text{C}_{16}\text{H}_{16}\text{NO}_2$: 254.1176; found 254.1175.

Compound 4j: Yellow viscous liquid. δ_{H} (300 MHz, Chloroform-*d*) 7.98 – 7.81 (2 H, m), 7.09 – 6.98 (4 H, m), 6.94 – 6.87 (2 H, m), 3.81 (3 H, s), 2.64 (3 H, s), 2.50 (3 H, s), 2.28 (3 H, s). δ_{C} (75 MHz, Chloroform-*d*) 195.8, 160.1, 158.5, 155.6, 135.4, 133.9, 130.2, 128.3, 125.8, 124.8, 122.1, 114.0, 105.1, 55.4, 30.6, 21.0, 15.0. **ESI-MS:** $m/z = [\text{M}+\text{H}]^+$ calcd. for $\text{C}_{21}\text{H}_{21}\text{O}_3\text{S}$: 353.1206; found 353.1204.

Compound 4k: White solid, mp 110 °C. δ_{H} (300 MHz, Chloroform-*d*) 7.98 – 7.87 (2 H, m), 7.44 – 7.33 (3 H, m), 7.25 – 7.18 (2 H, m), 7.10 – 7.01 (2 H, m), 2.66 (3 H, s), 2.48 (3 H, s). δ_{C} (75 MHz, Chloroform-*d*) 195.3, 159.3, 155.6, 135.9, 131.6, 129.6, 129.2, 129.1, 128.7, 127.0, 126.8, 124.8, 106.1, 30.7, 15.1. **ESI-MS:** $m/z = [\text{M}+\text{H}]^+$ calcd. for $\text{C}_{19}\text{H}_{16}\text{ClO}_2\text{S}$: 343.0554; found 343.0555.

Compound 4l: Yellow viscous liquid. δ_{H} (300 MHz, Chloroform-*d*) 7.60 – 7.46 (2 H, m), 7.43 – 7.18 (3 H, m), 6.32 (1 H, d, *J* 3.1), 6.14 – 6.12 (1 H, m), 2.66 (3 H, s), 2.40 (3 H, s), 2.07 (3 H, s). δ_{C} (75 MHz, Chloroform-*d*) 195.7, 157.8, 152.8, 150.1, 143.5, 129.8, 128.5, 128.2, 125.7, 124.1, 112.1, 111.2, 107.5, 29.7, 14.6, 13.8. **ESI-MS:** $m/z = [\text{M}+\text{H}]^+$ calcd. for $\text{C}_{18}\text{H}_{17}\text{O}_3$: 281.1172; found 281.1172.

Compound 4m: Yellow viscous liquid. δ_{H} (300 MHz, Chloroform-*d*) 7.36 – 7.26 (2 H, m), 7.24 – 7.12 (2 H, m), 6.20 (1 H, d, *J* 3.1), 6.12 – 5.92 (1 H, m), 2.54 (3 H, s), 2.29 (3 H, s), 1.96 (3 H, s). δ_{C} (75 MHz, Chloroform-*d*) 195.4, 158.0, 153.0, 149.0, 143.1, 134.0, 128.8, 128.2, 126.9, 124.2, 112.2, 111.9, 107.5, 29.7, 14.6, 13.8. **ESI-MS:** $m/z = [\text{M}+\text{H}]^+$ calcd. for $\text{C}_{18}\text{H}_{16}\text{ClO}_3$: 315.0782; found 315.0783.

Compound 4n: Yellow viscous liquid. δ_{H} (300 MHz, Chloroform-*d*) 7.46 – 7.33 (2 H, m), 6.89 – 6.76 (2 H, m), 6.27 (1 H, d, *J* 3.1), 6.09 (1 H, dq, *J* 3.1, 1.0), 3.80 (3 H, s), 2.62 (3 H, s), 2.37 (3 H, s), 2.03 (3 H, s). δ_{C} (75 MHz, Chloroform-*d*) 195.8, 159.6, 157.3, 152.6, 150.3, 143.8, 127.2, 125.3, 124.0, 122.6, 114.3, 112.0, 109.9, 107.4, 55.4, 29.7, 14.6, 13.8. **ESI-MS:** $m/z = [\text{M}+\text{H}]^+$ calcd. for $\text{C}_{19}\text{H}_{19}\text{O}_4$: 311.1278; found 311.1274.

Compound 4o: Brown viscous liquid. δ_{H} (300 MHz, Chloroform-*d*) 7.49 – 7.37 (2 H, m), 7.36 – 7.28 (2 H, m), 6.28 (1 H, d, *J* 3.1), 6.10 (1 H, dq, *J* 3.1, 1.0), 2.62 (3 H, s), 2.36 (3 H, s), 2.03 (3 H, s). δ_{C} (75 MHz, Chloroform-*d*) 195.4, 158.1, 153.0, 149.0, 143.1, 131.7, 128.7, 127.1, 124.2, 122.2, 112.2, 112.0, 107.5, 29.7, 14.6, 13.8. **ESI-MS:** $m/z = [\text{M}+\text{H}]^+$ calcd. for $\text{C}_{18}\text{H}_{16}\text{BrO}_3$: 359.0277; found 359.0278.

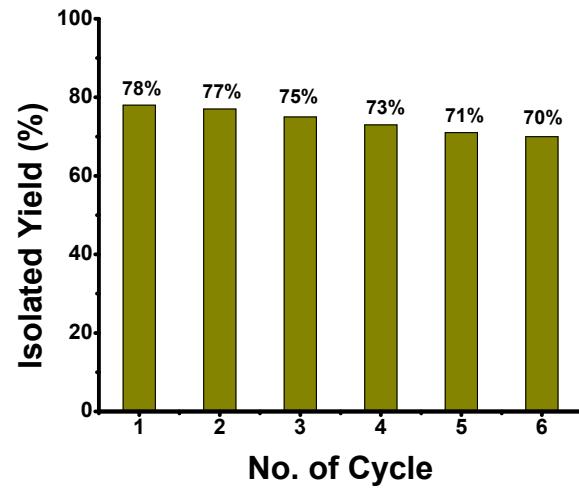
General procedure of the one-pot pyrrole synthesis: Glyoxal **1** (2.0 mmol), 1,3 diketone **2a** (2.0 mmol) and indole (2.0 mmol) were taken in a MW-tube and radiated in MW reactor at 150W for 6 min (3 min X 2). Then, aniline **5** (2.0 mmol) was added to the reaction vessel and whole mixture was heated at 70 °C for 5 h. Then, the crude mass was purified by column chromatography (25% ethyl acetate/n-hexane) to afford the desired product.

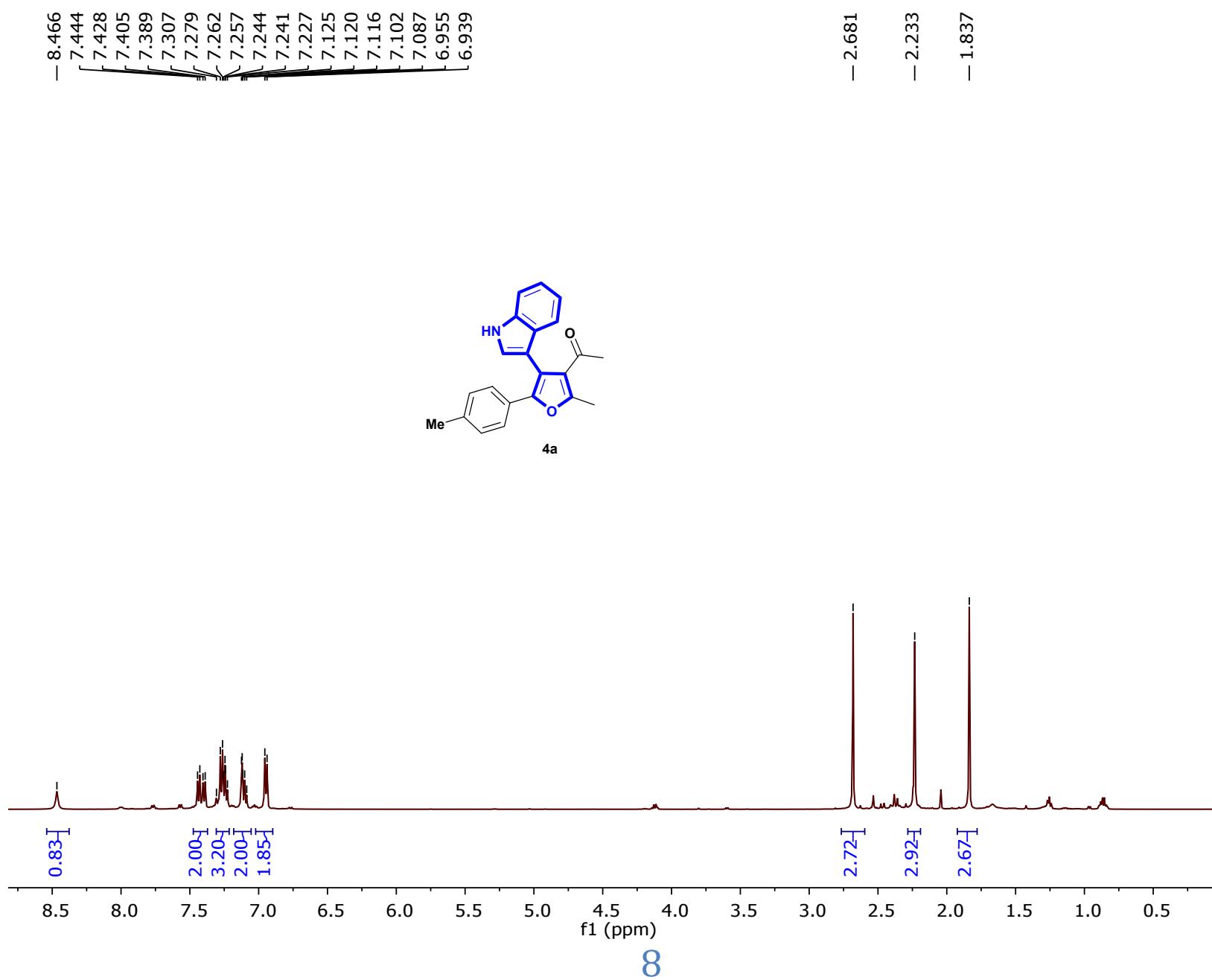
Compound **5a**: Light yellow solid, mp 205 °C. δ_{H} (300 MHz, DMSO-*d*₆) 11.00 (1 H, s), 7.33 (1 H, d, *J* 8.0), 7.26 (1 H, d, *J* 7.8), 7.22 – 7.09 (5 H, m), 7.05 (1 H, t, *J* 7.5), 6.93 (3 H, dd, *J* 12.5, 8.0), 6.54 (2 H, d, *J* 8.4), 3.56 (3 H, s), 2.29 (6 H, s), 1.74 (3 H, s). δ_{C} (75 MHz, DMSO-*d*₆) 195.9, 157.8, 137.5, 135.7, 134.9, 134.8, 132.4, 131.9, 129.6, 128.7, 128.4, 125.0, 122.4, 120.0, 119.0, 118.9, 114.8, 112.8, 111.5, 109.4, 54.7, 29.6, 20.6, 13.0. **ESI-MS:** *m/z* =[M+H]⁺calcd. for C₂₉H₂₇N₂O₂: 435.2067; found 435.2067.

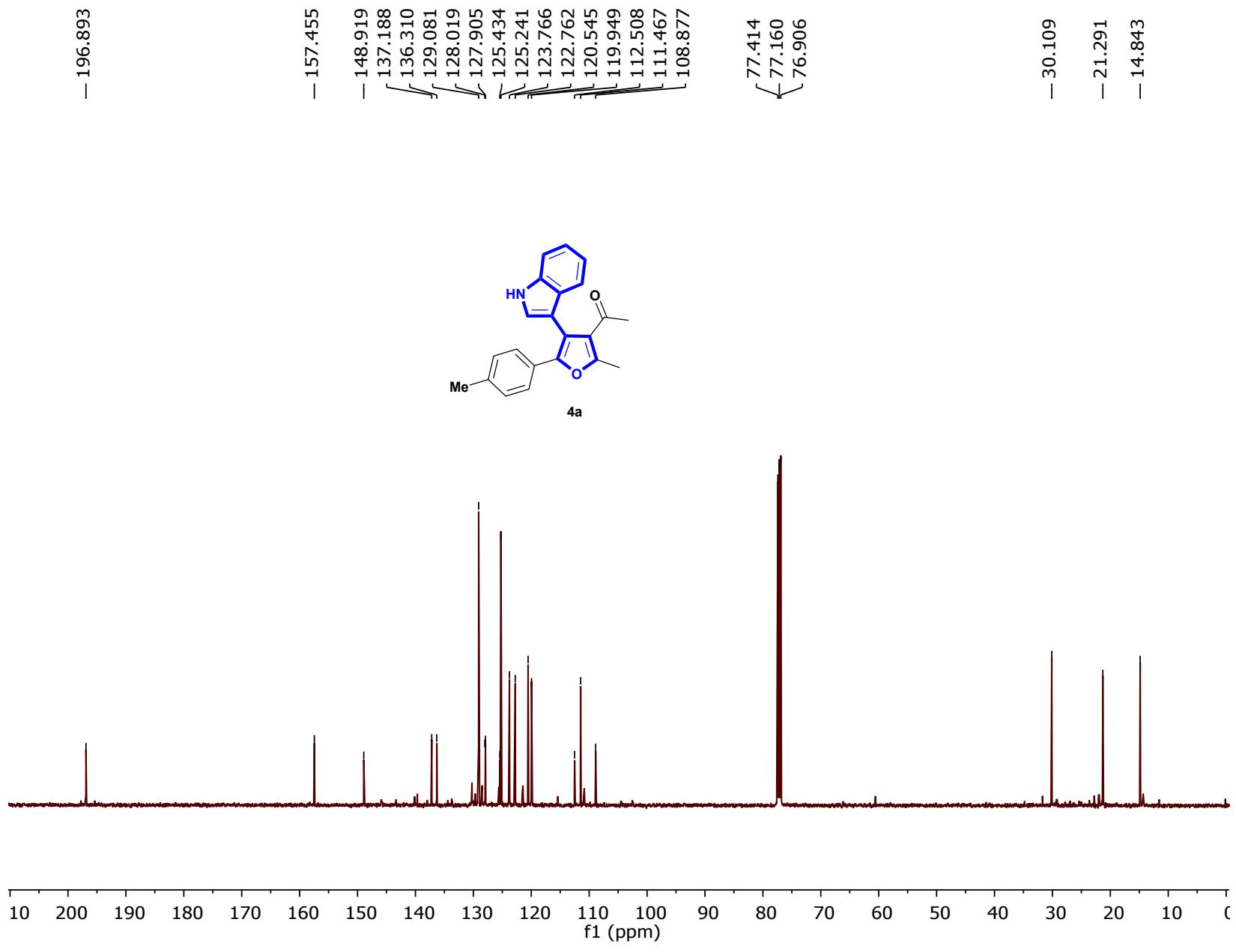
Compound **5b**: Off white solid, mp 275 °C. δ_{H} (300 MHz, DMSO-*d*₆) 11.34 – 10.78 (1 H, m), 7.35 (1 H, d, *J* 8.1), 7.29 – 7.11 (6 H, m), 7.10 – 6.90 (6 H, m), 2.30 (6 H, s), 1.75 (3 H, s). δ_{C} (75 MHz, DMSO-*d*₆) 195.8, 137.8, 135.8, 134.6, 132.2, 131.4, 131.2, 130.9, 129.7, 128.5, 128.3, 127.4, 125.2, 123.9, 122.5, 121.1, 119.1, 118.7, 115.6, 111.6, 108.8, 29.7, 20.6, 12.9. **ESI-MS:** *m/z* =[M+H]⁺calcd. for C₂₈H₂₄ClN₂O: 439.1572; found 439.1570.

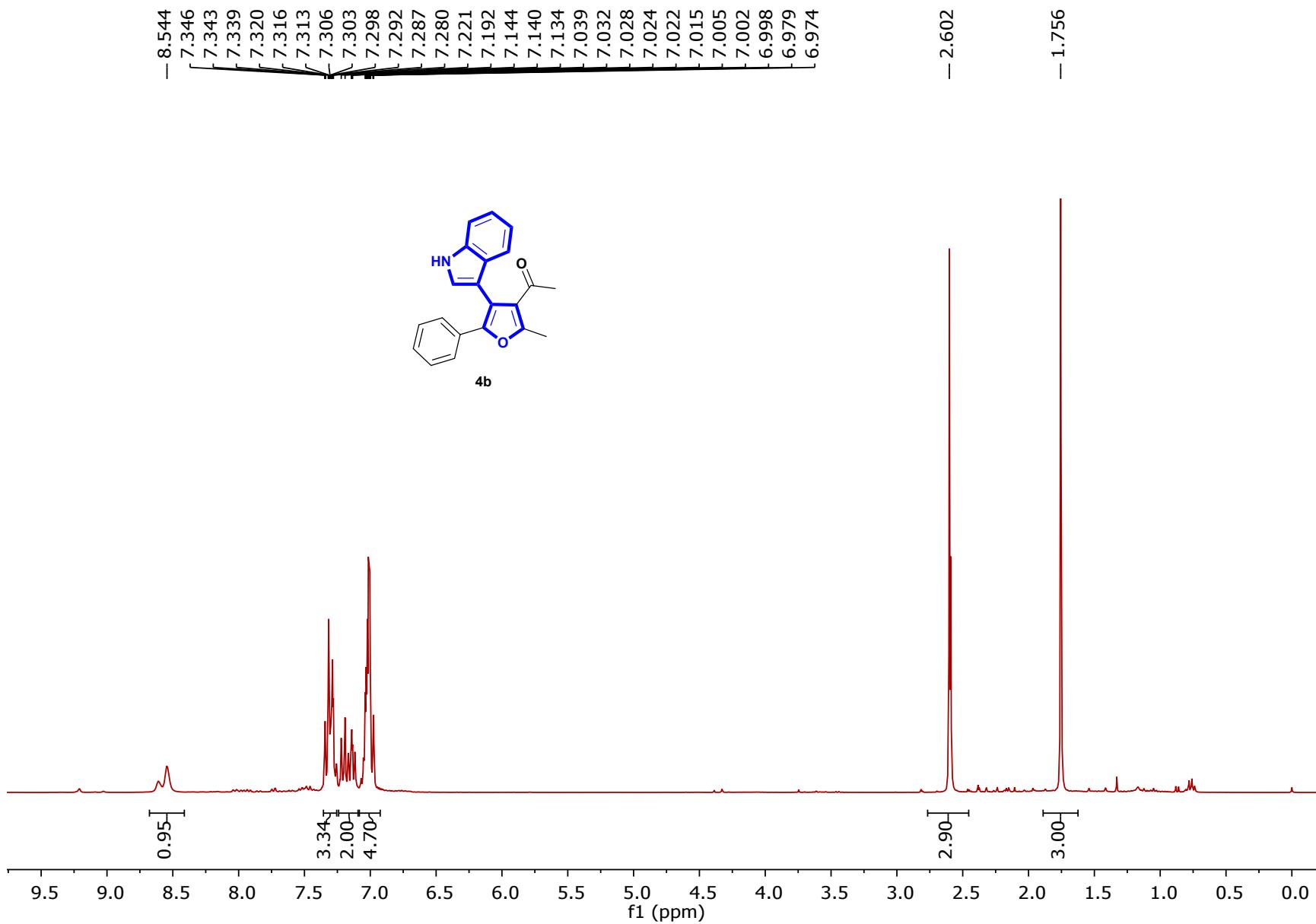
Compound **5c**: Light red solid, mp 288 °C. δ_{H} (300 MHz, DMSO-*d*₆) 11.03 (1 H, s), 7.55 – 7.40 (2 H, m), 7.40 – 7.29 (3 H, m), 7.30 – 7.15 (2 H, m), 7.14 – 6.87 (7 H, m), 2.32 (3 H, s), 1.76 (3 H, d, *J* 1.4). δ_{C} (75 MHz, DMSO-*d*₆) 196.0, 136.3, 136.1, 135.7, 135.4, 135.0, 132.7, 132.4, 132.2, 131.6, 130.7, 130.5, 129.1, 129.3, 127.6, 127.4, 125.1, 122.8, 121.0, 119.0, 118.7, 115.4, 111.5, 108.9, 108.7, 29.7, 12.9. **ESI-MS:** *m/z* =[M+H]⁺calcd. for C₂₇H₂₂ClN₂O: 425.1415; found 425.1416.

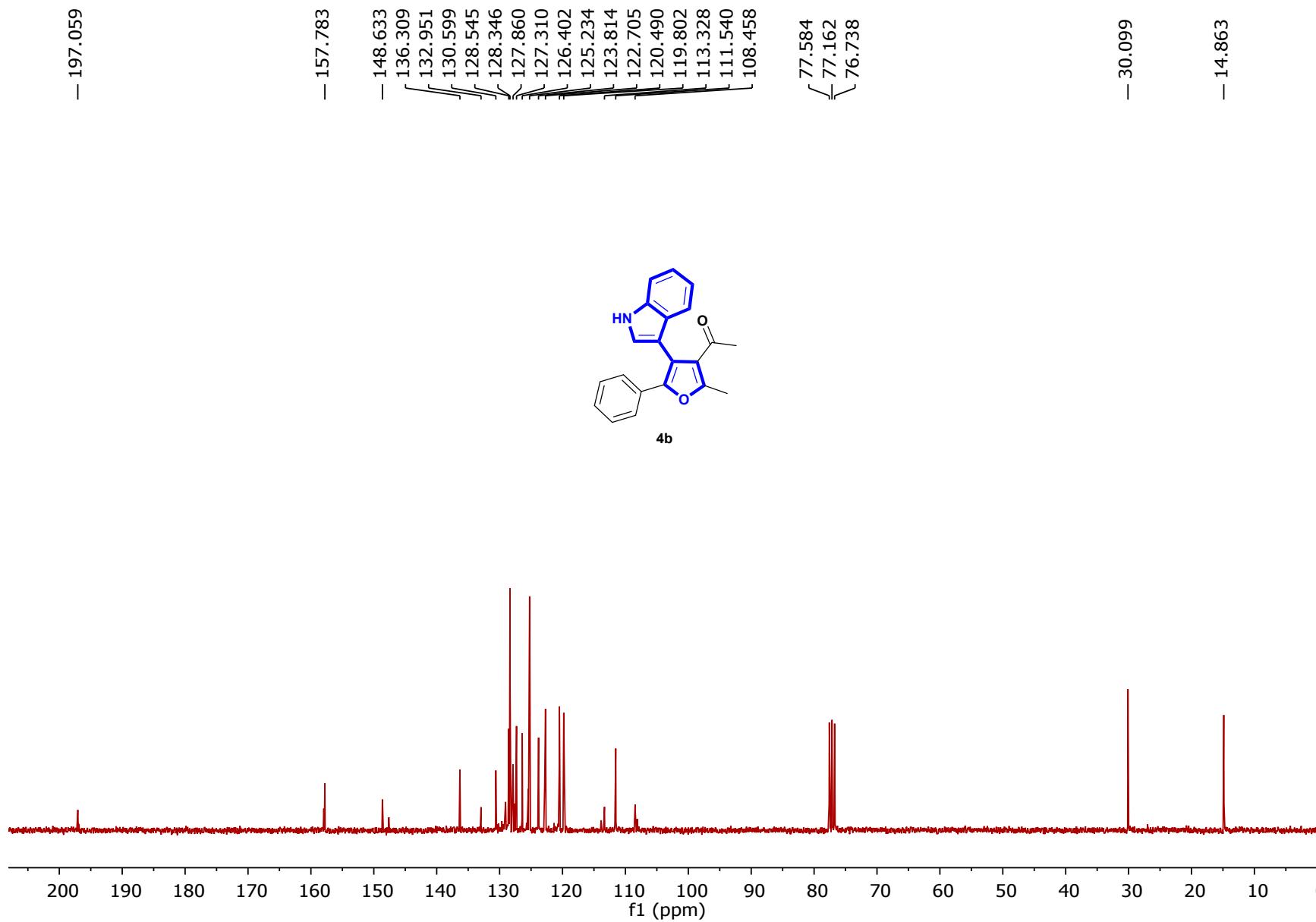
Figure 1. Reusability test of GO for the synthesis 4a

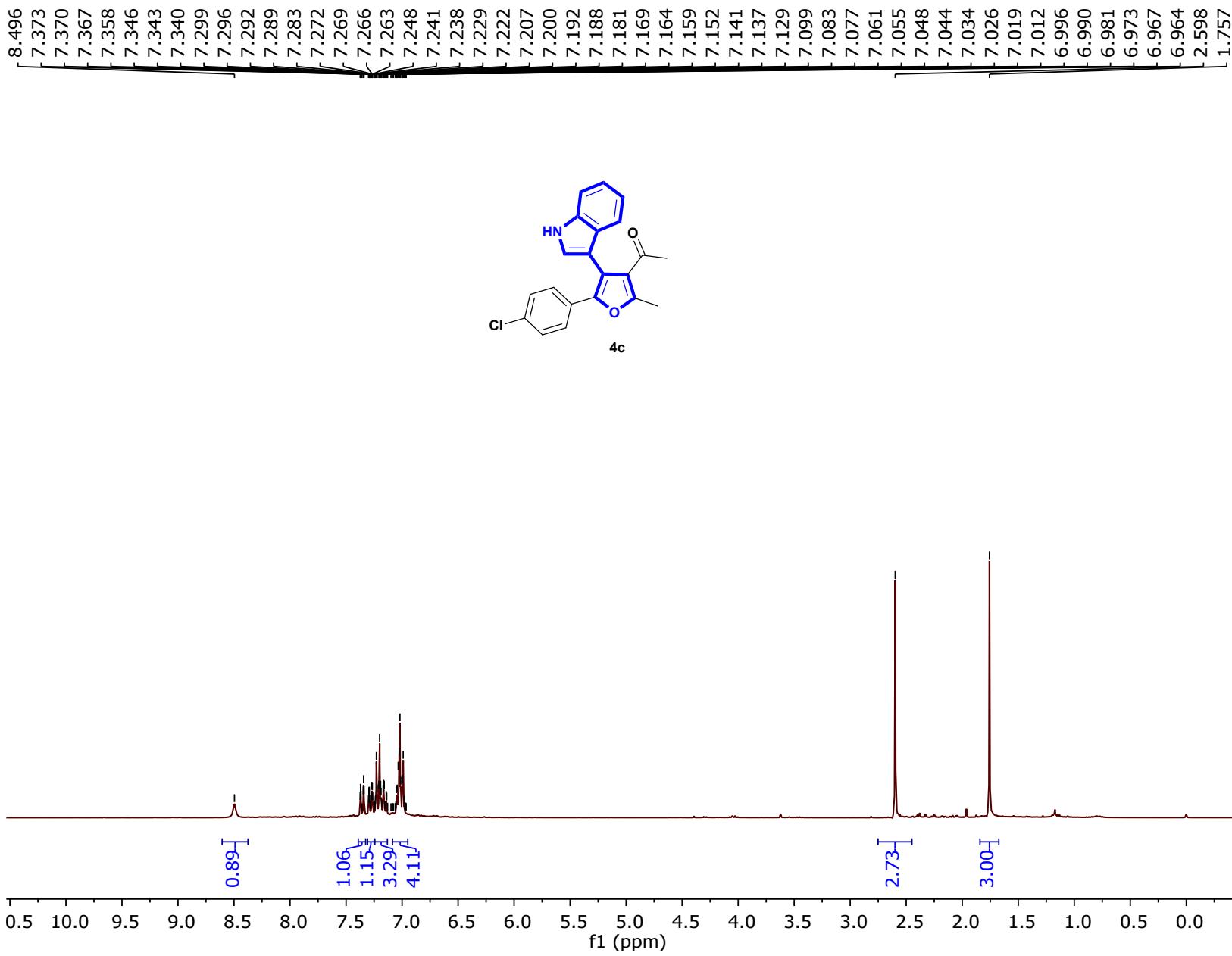




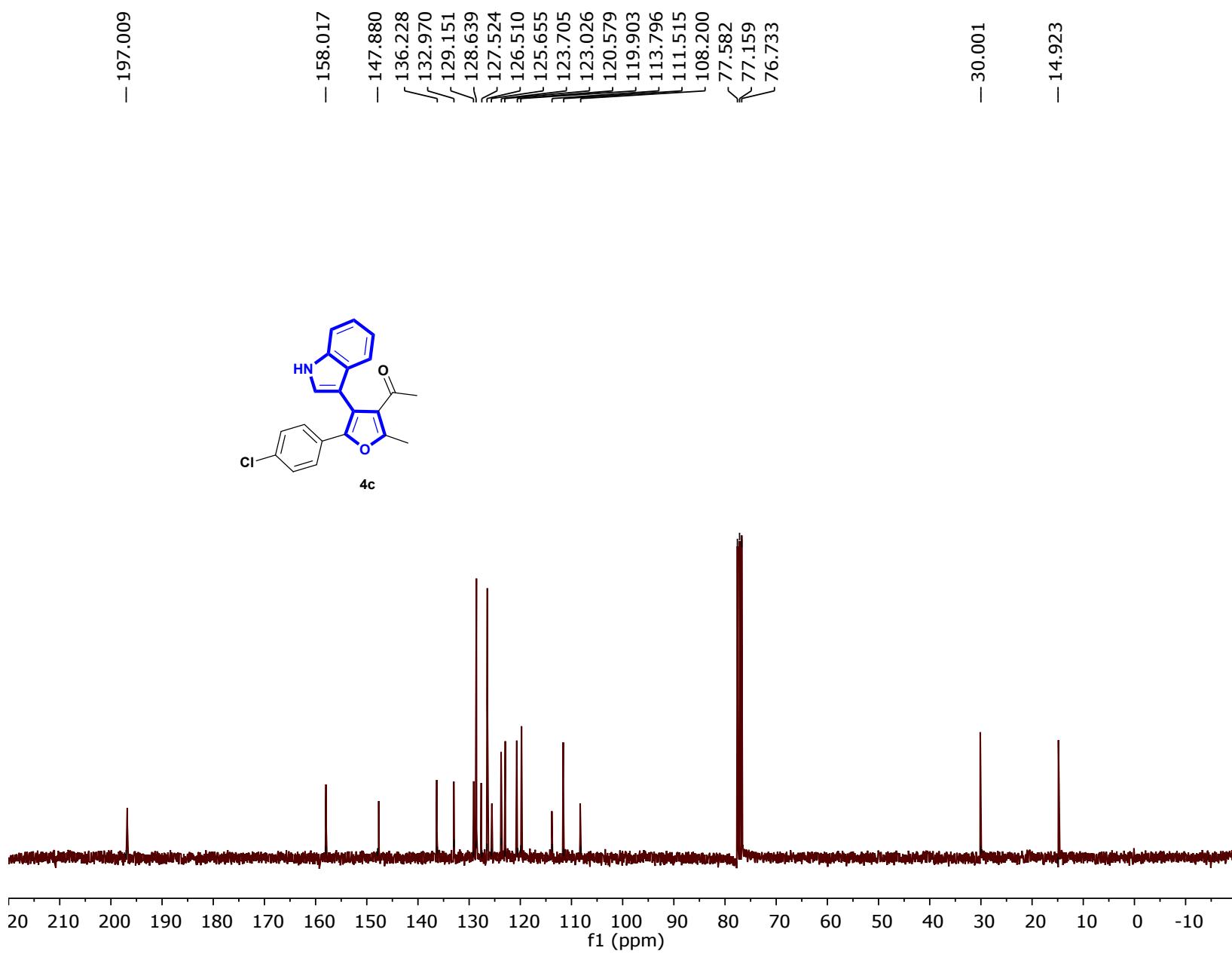


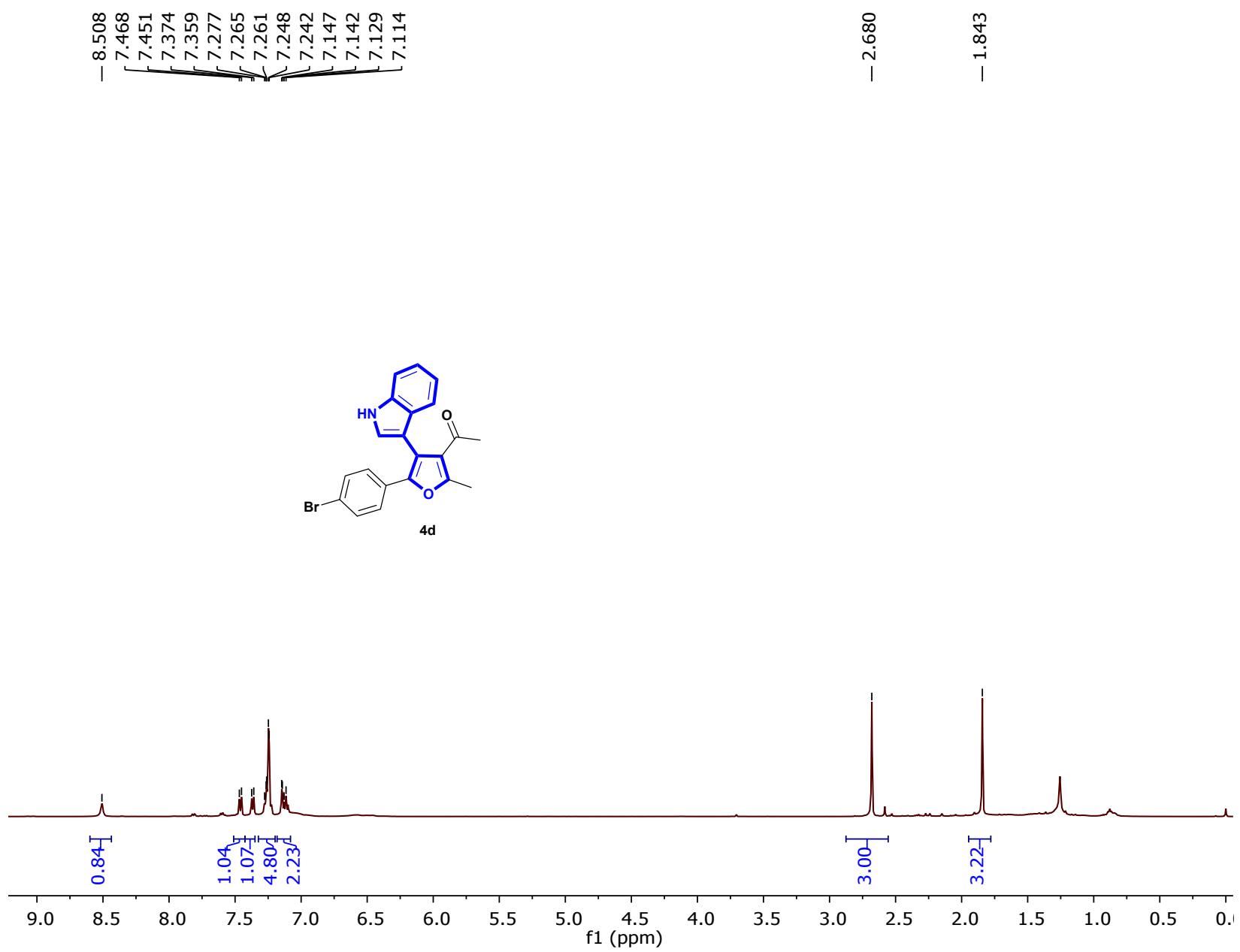




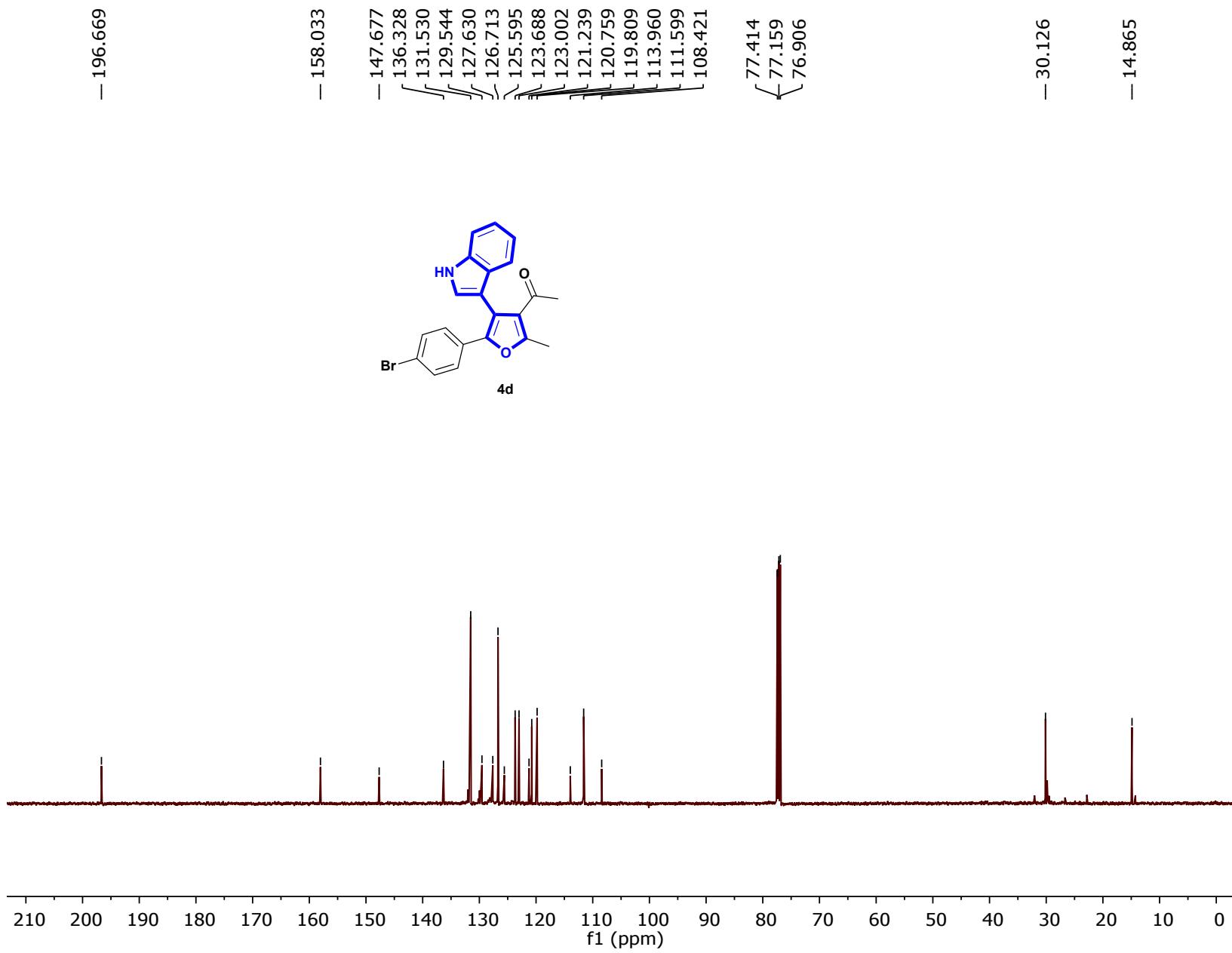


12

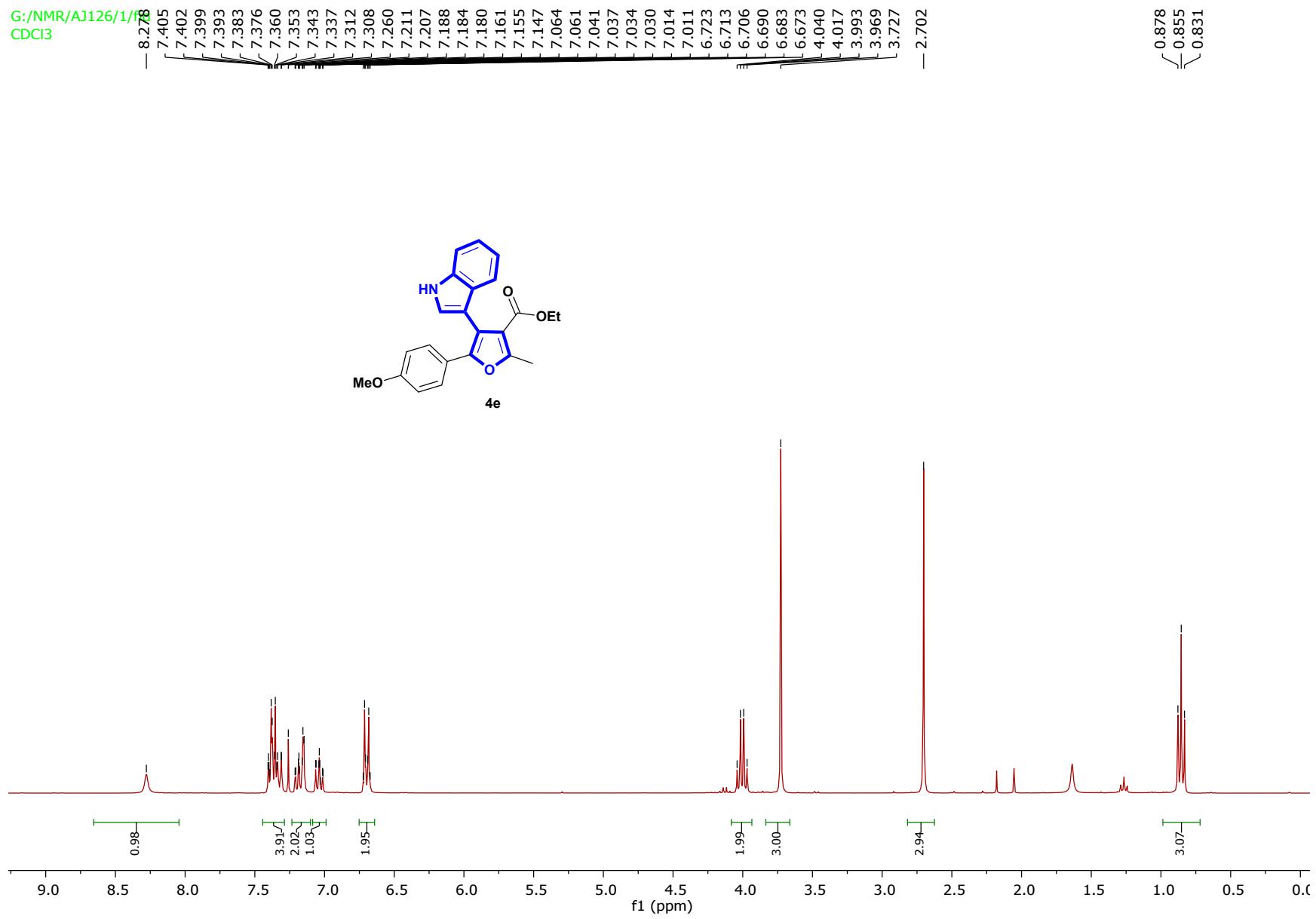




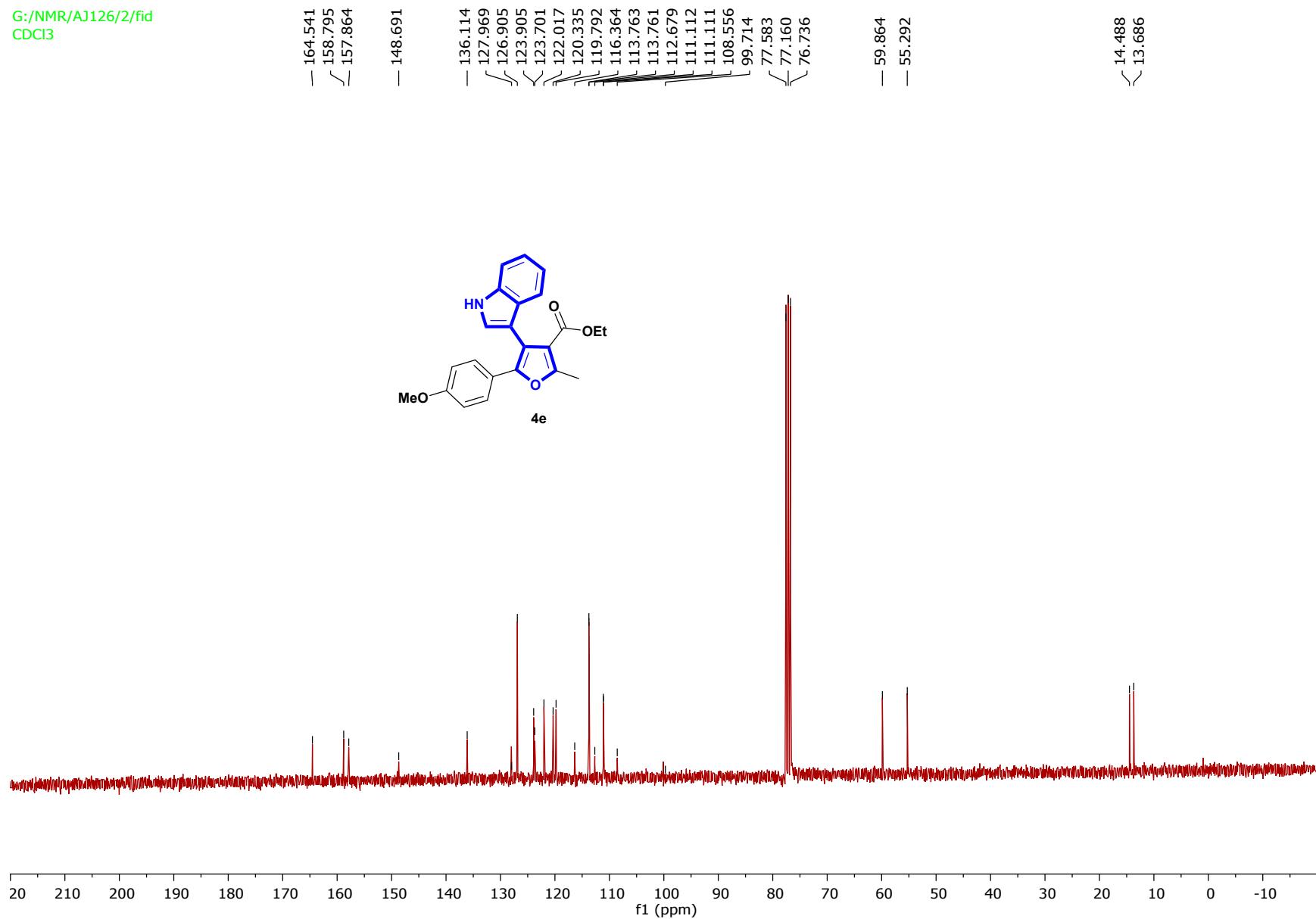
14



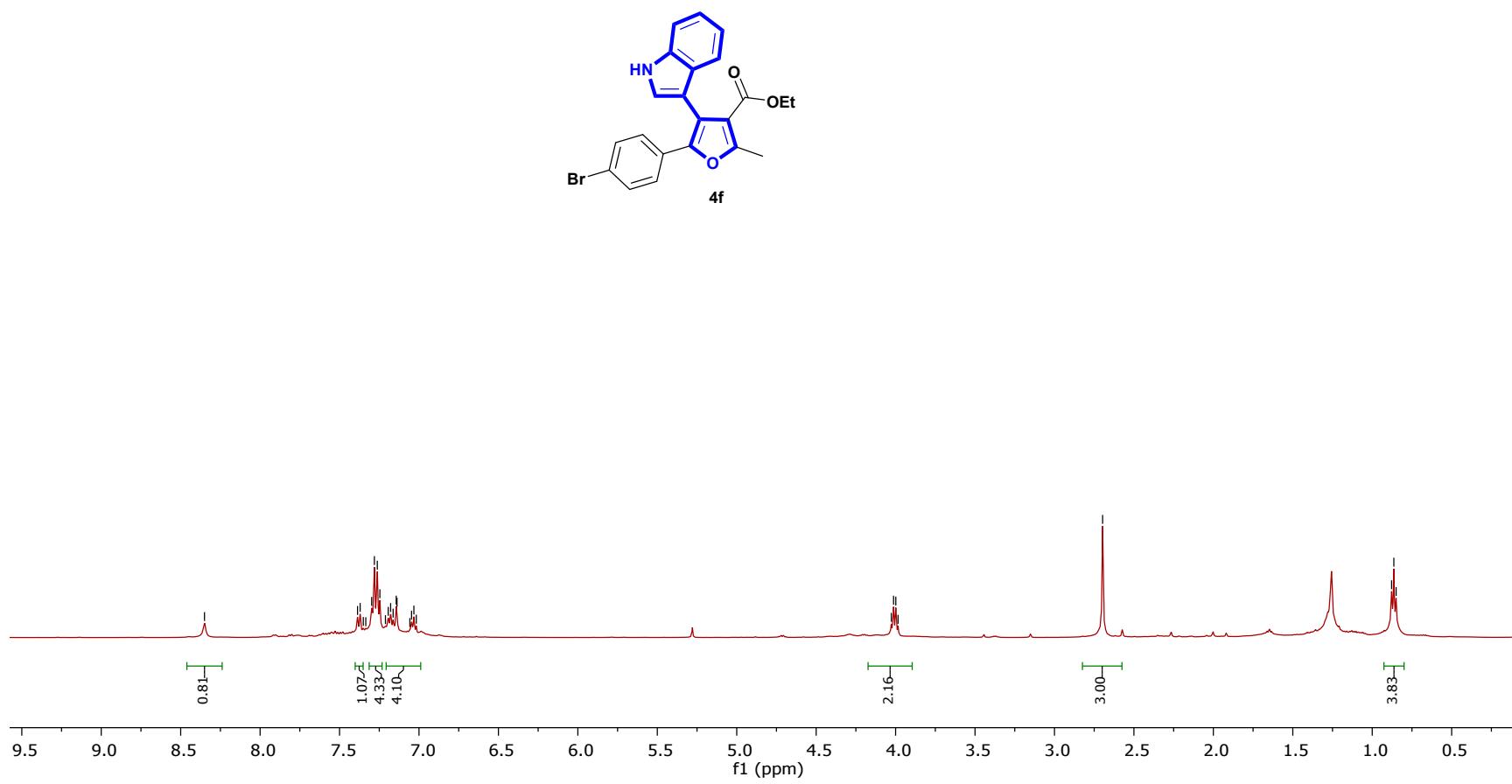
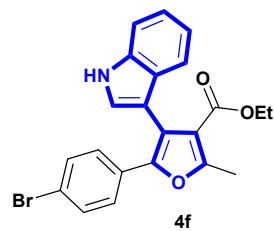
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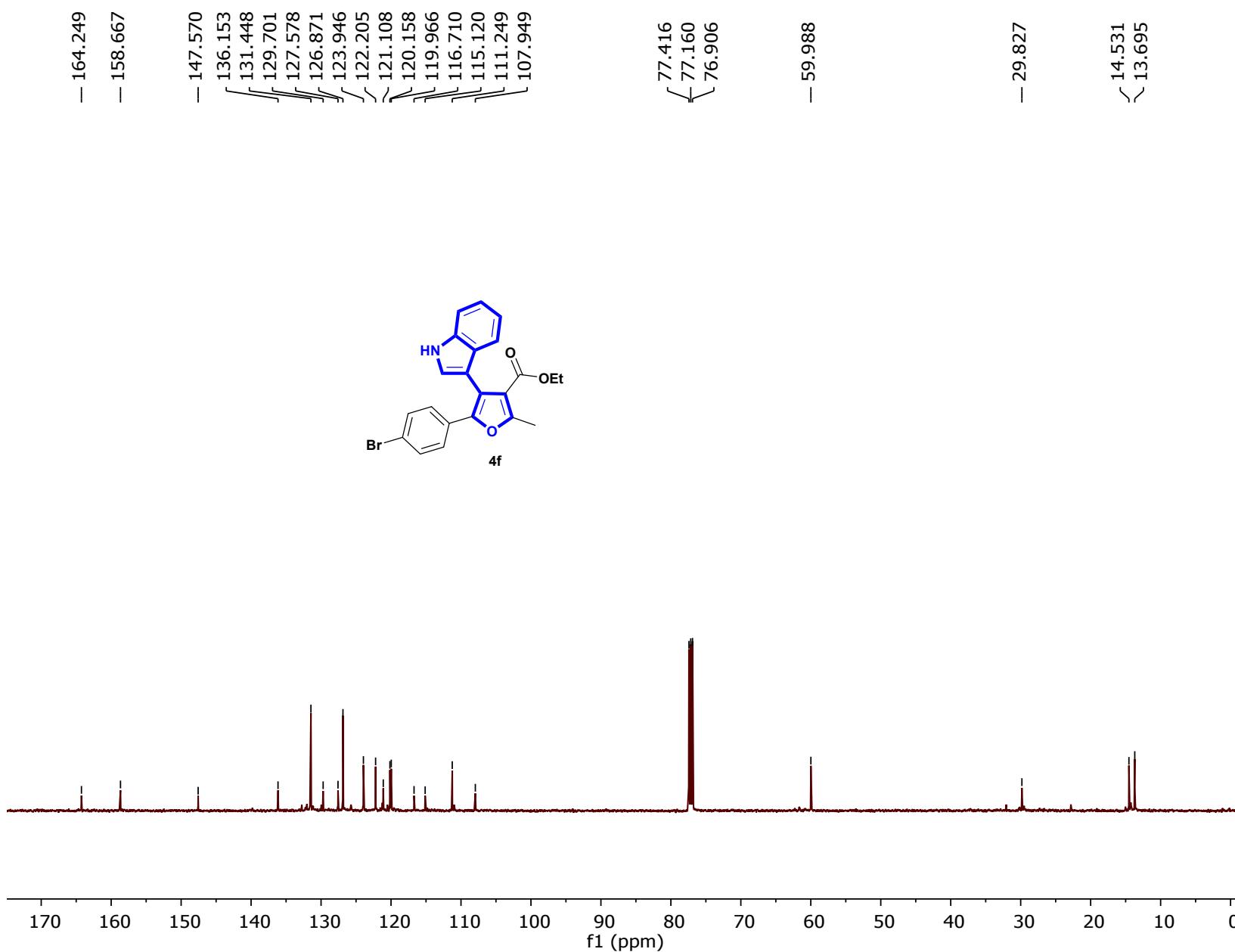


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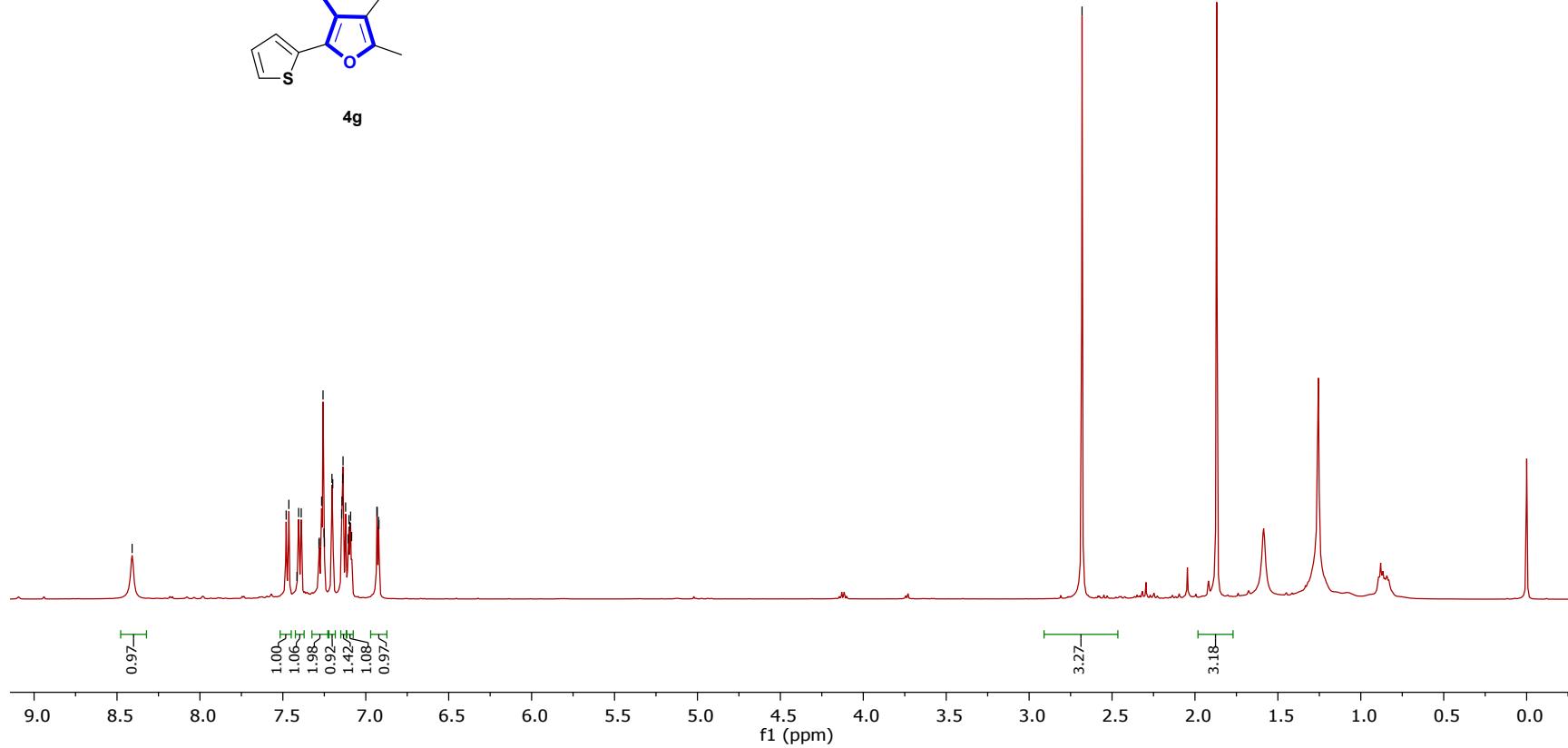




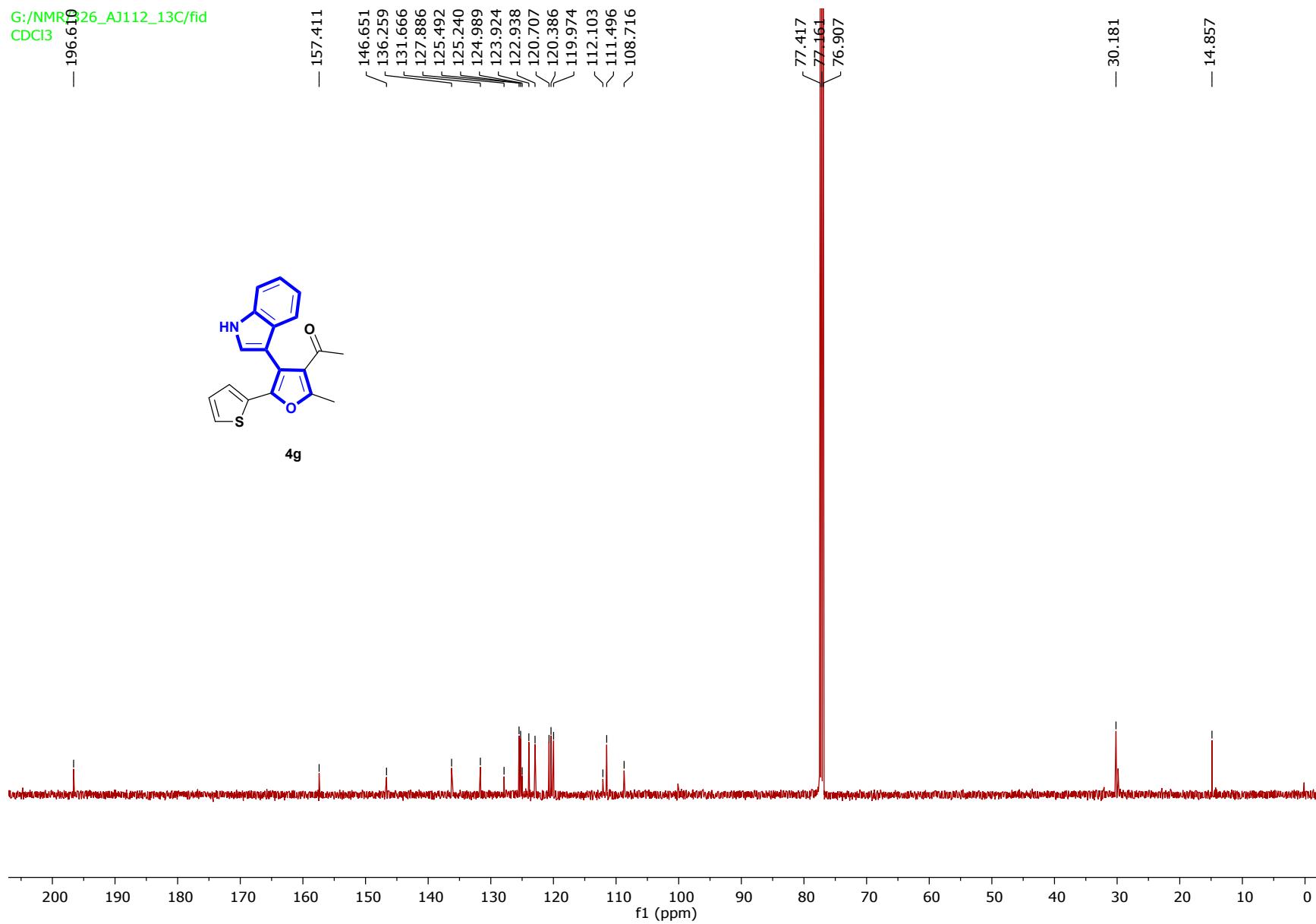
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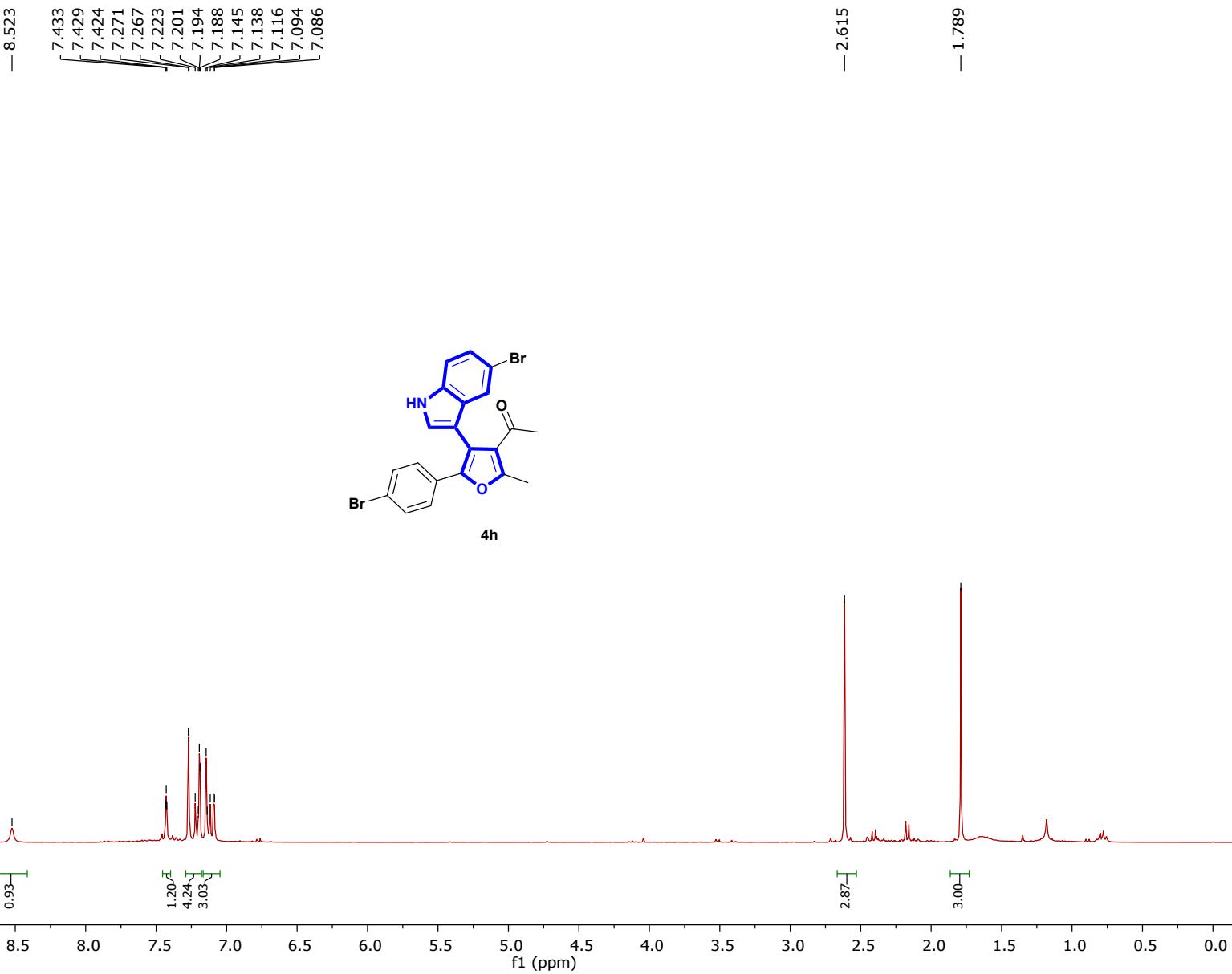
4g



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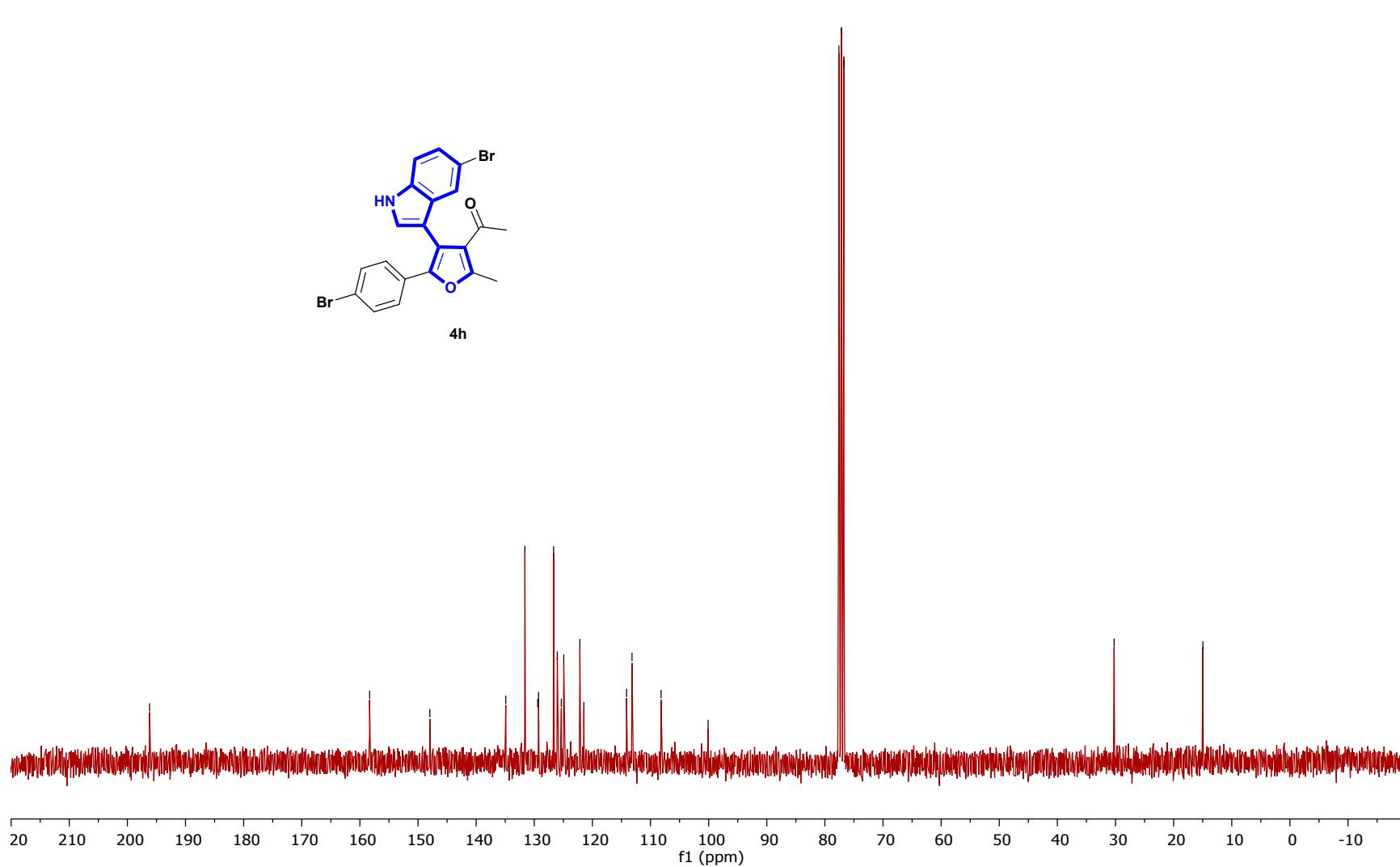
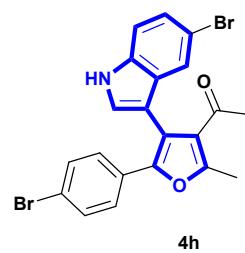
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129.263
126.671
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~ 108.179

— 100.092

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77.160
76.736

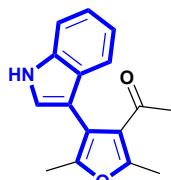
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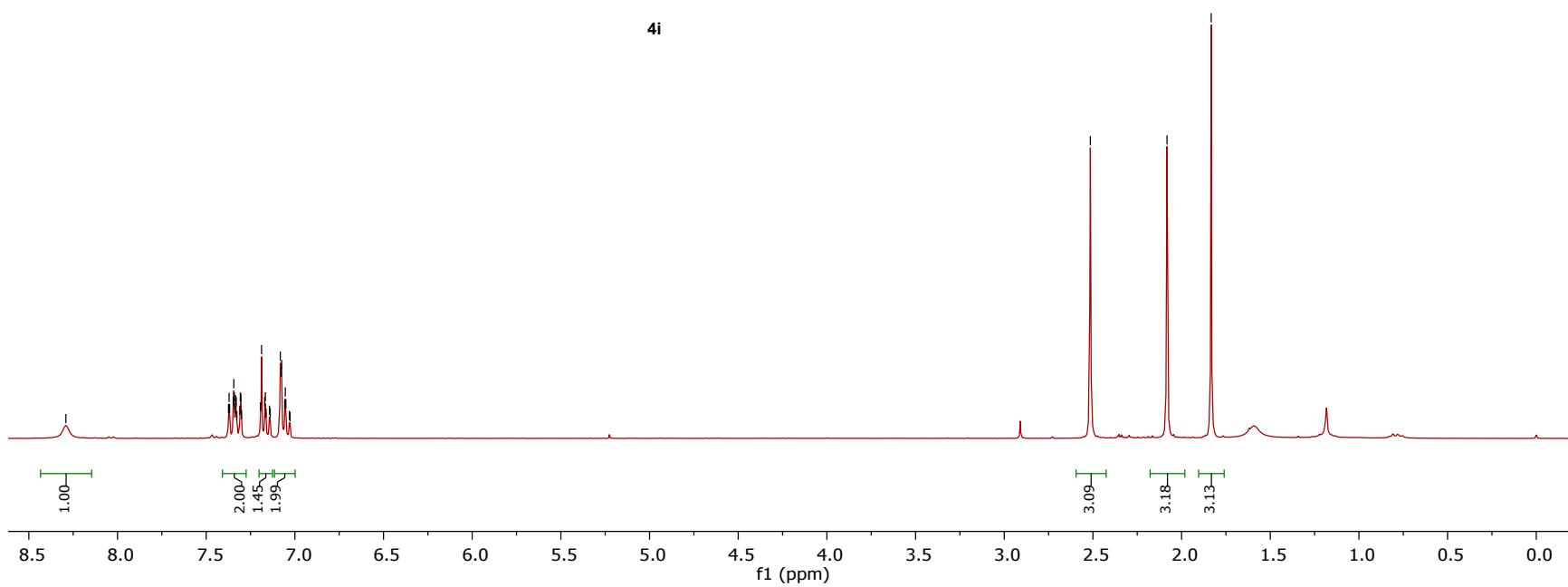


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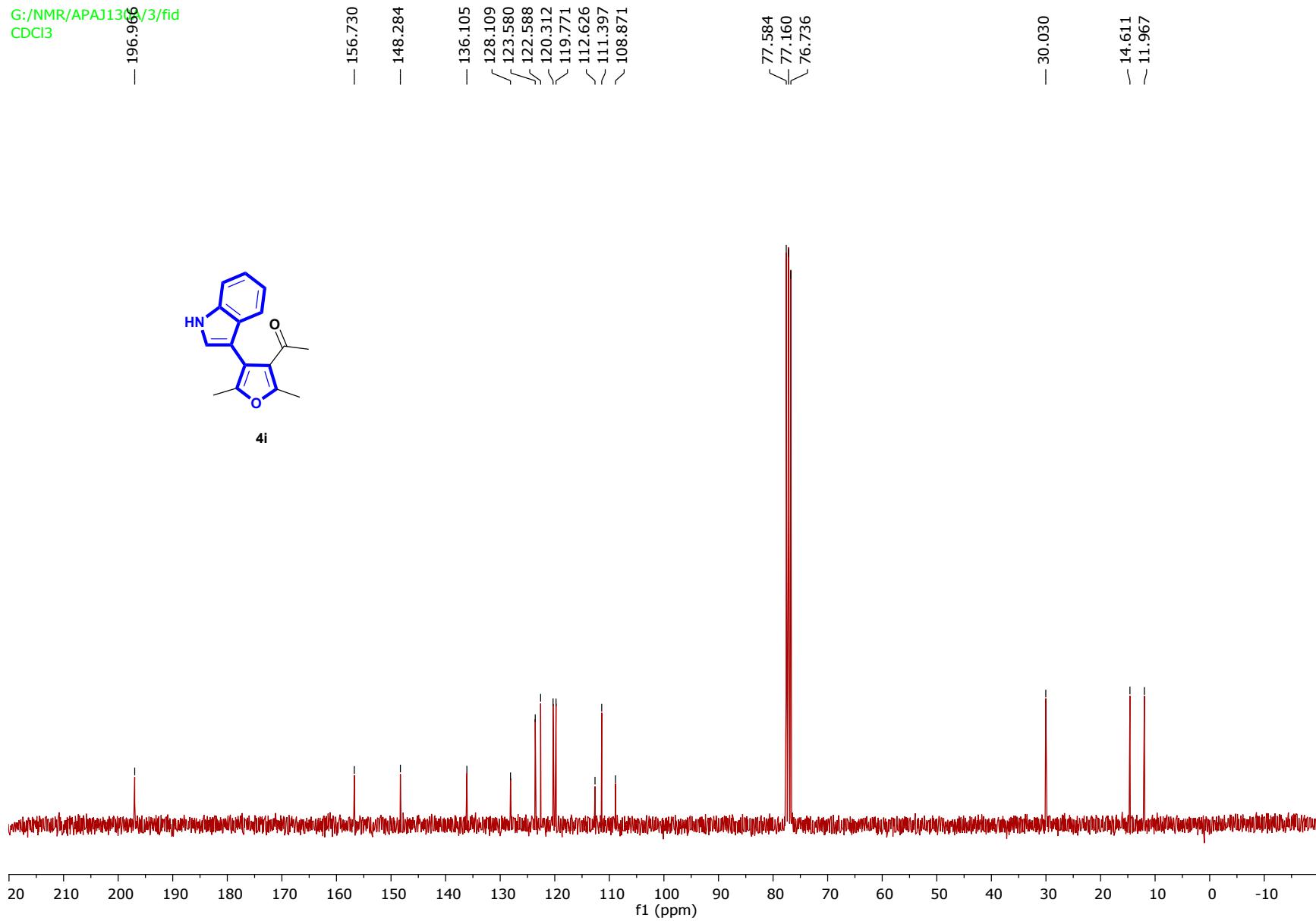
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7.188
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4i



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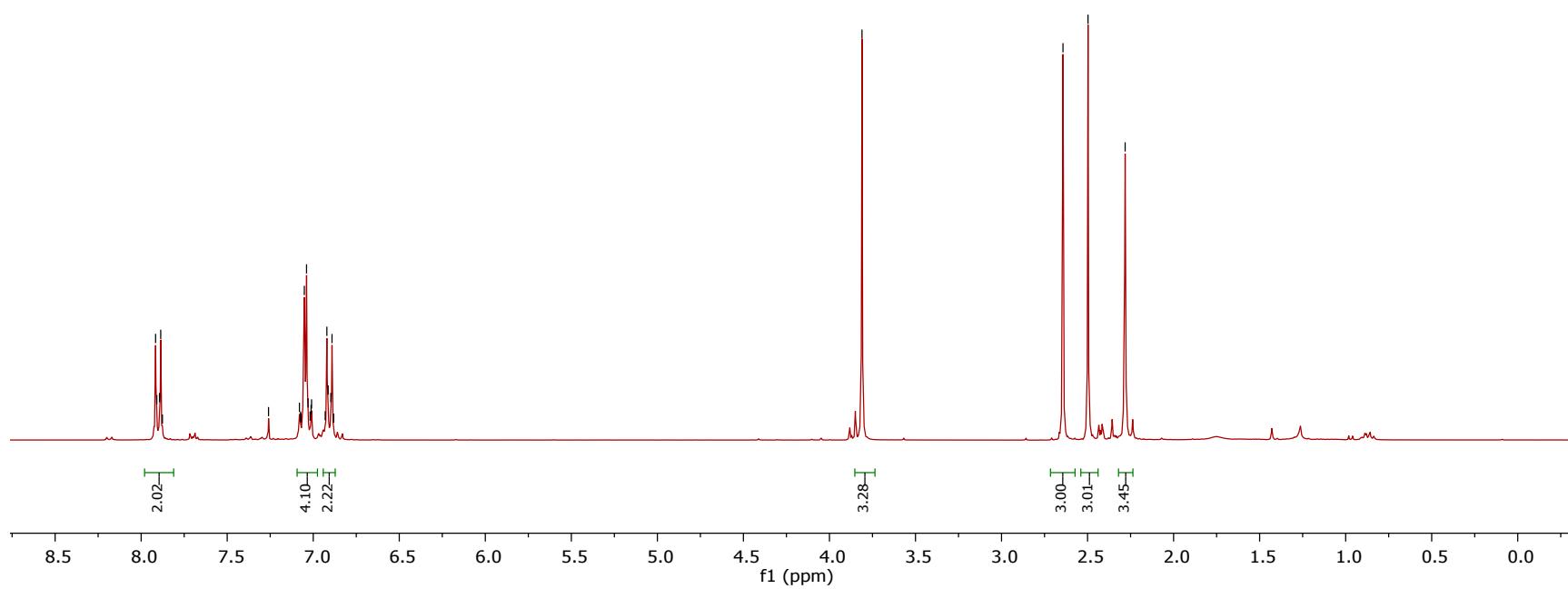


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~ 2.643
~ 2.497
~ 2.282



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CDCl₃

— 195.81

— 160.064
— 158.512
— 155.553

— 135.386
— 133.916
— 130.172
— 128.347
— 125.761
— 124.827
— 122.095

— 113.965

— 105.102

— 55.365

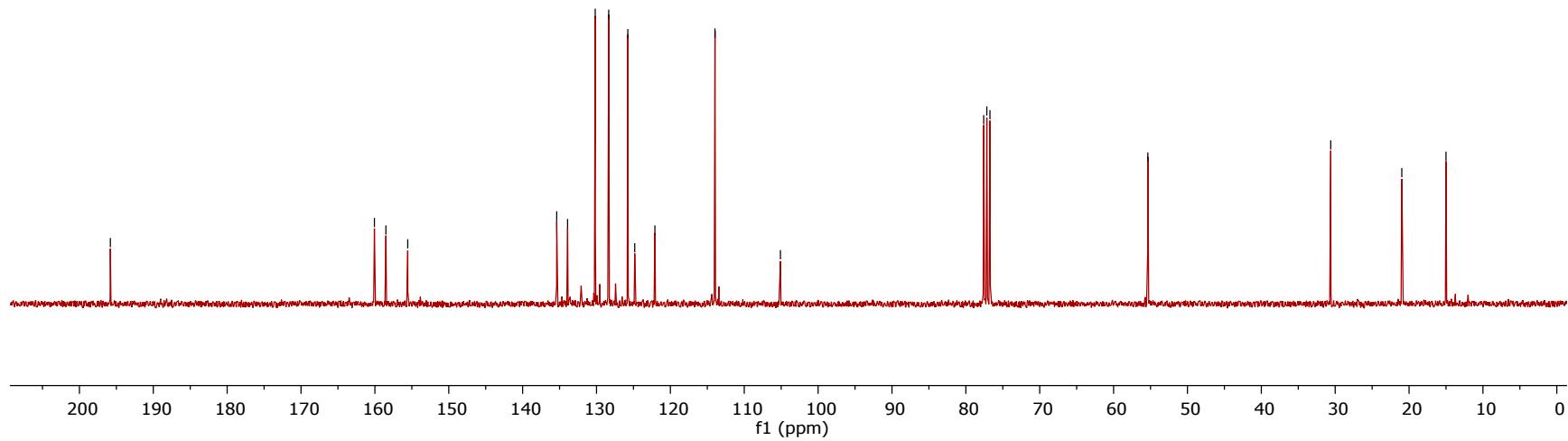
— 30.604

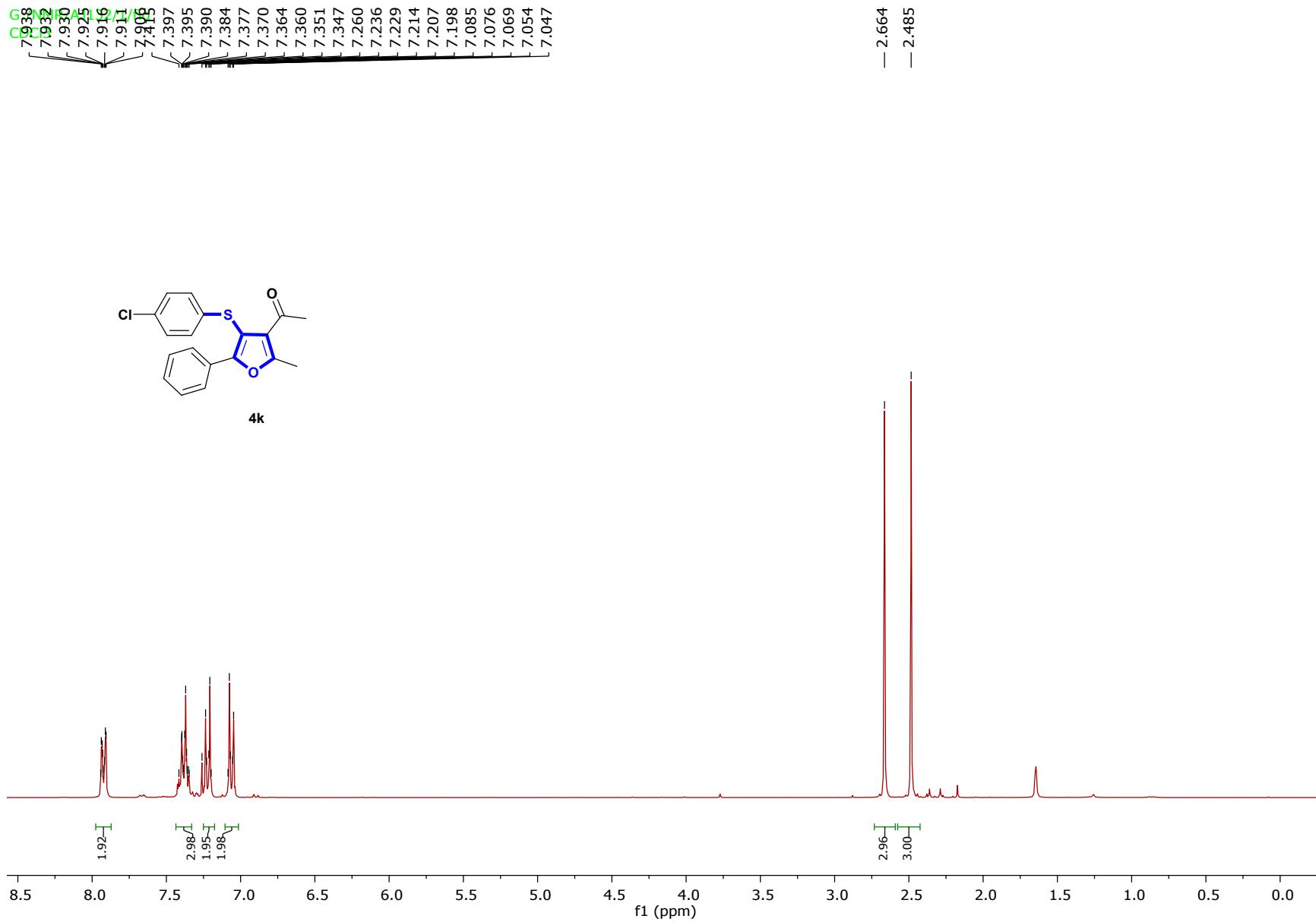
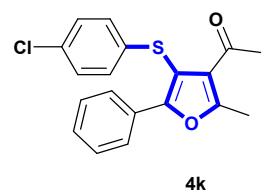
— 20.984

— 15.001

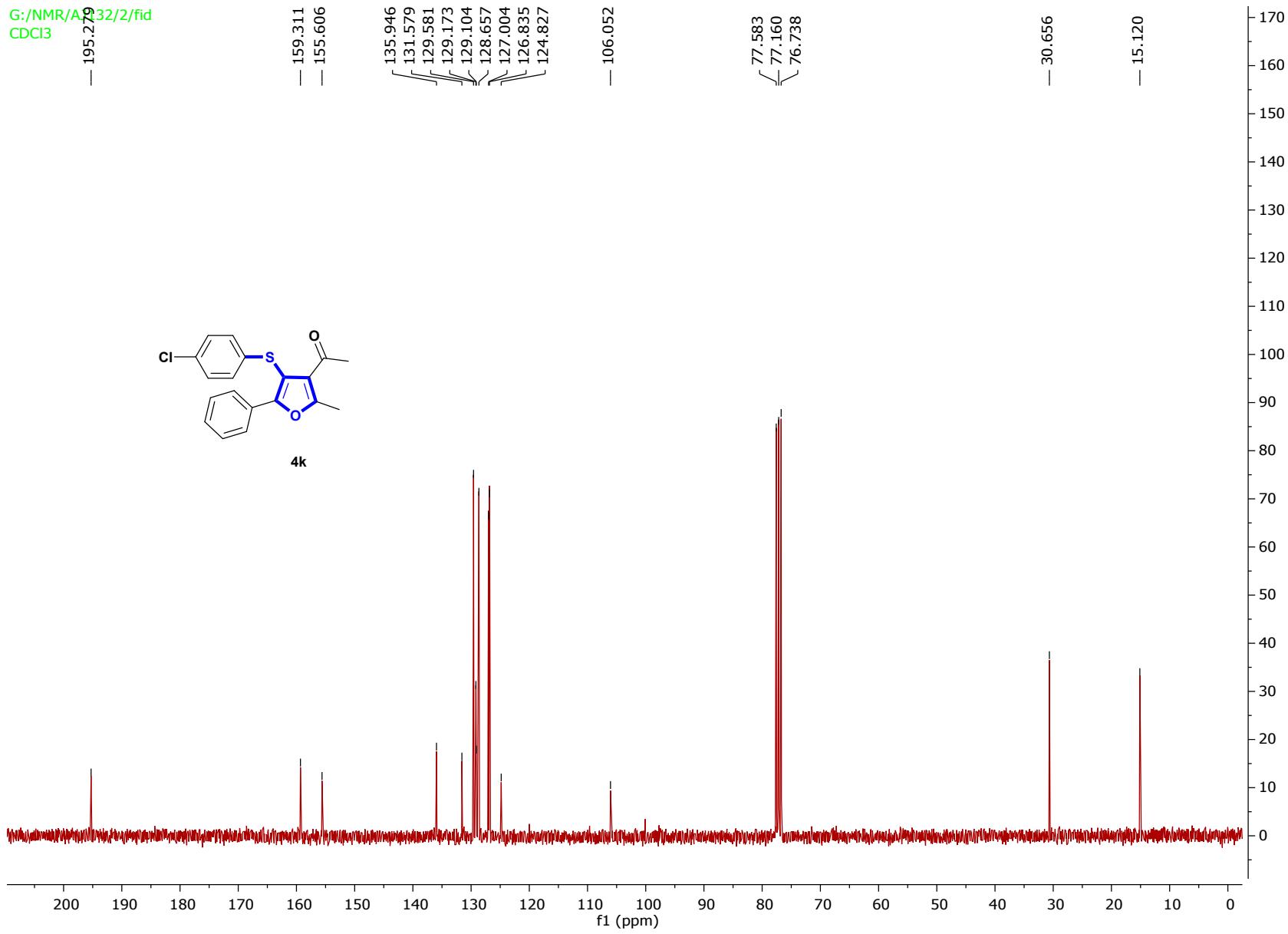


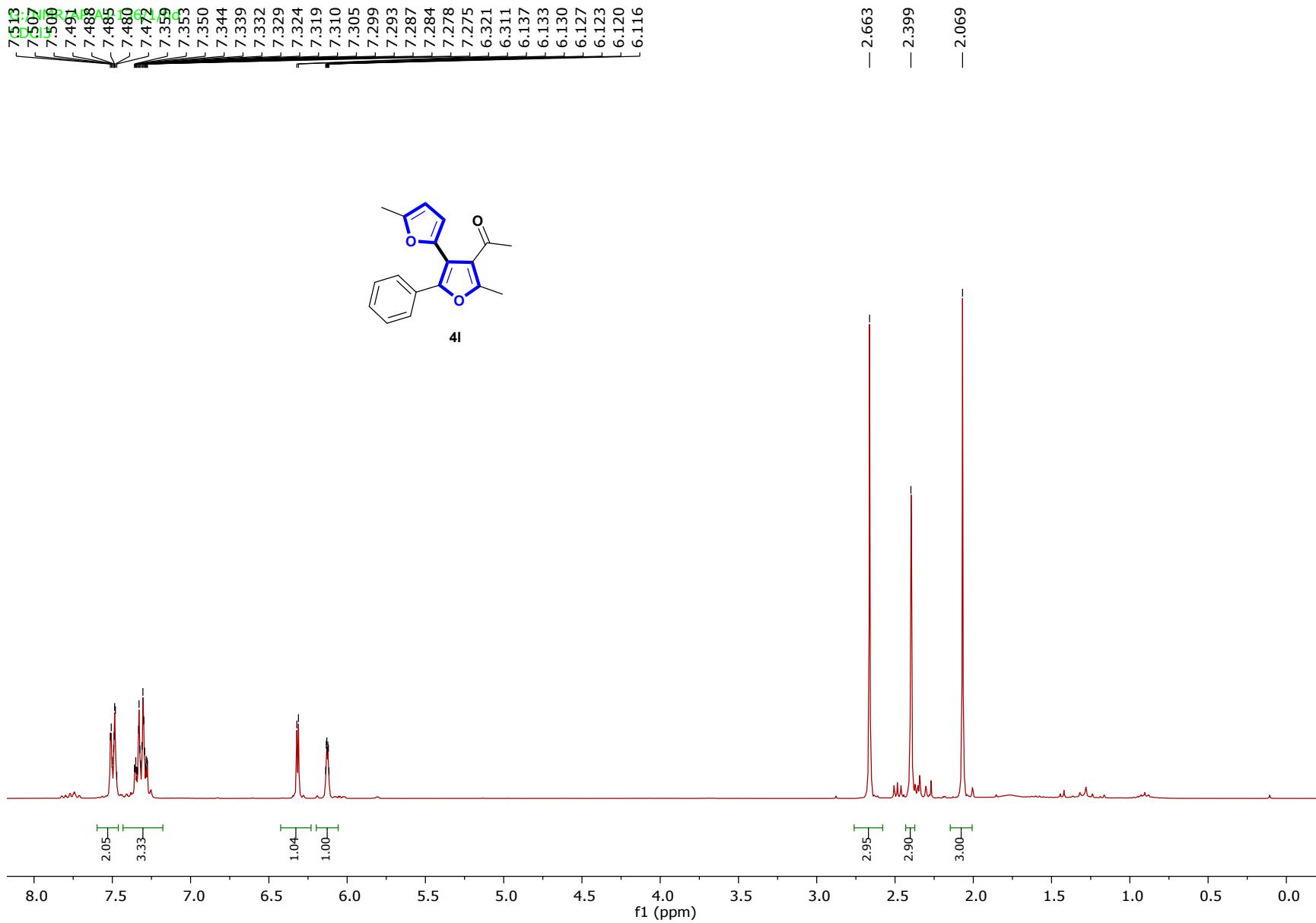
4j





G:/NMR/Ar32/2/fid
CDCl₃





G:/NMR/APAJ13618/fid
CDCl₃

— 195.618

~ 157.801
~ 152.761
~ 150.110
~ 143.473

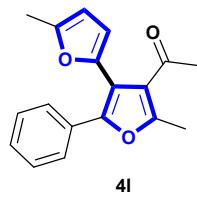
129.802
128.522
128.192
125.697
124.149

112.078
111.412
107.479

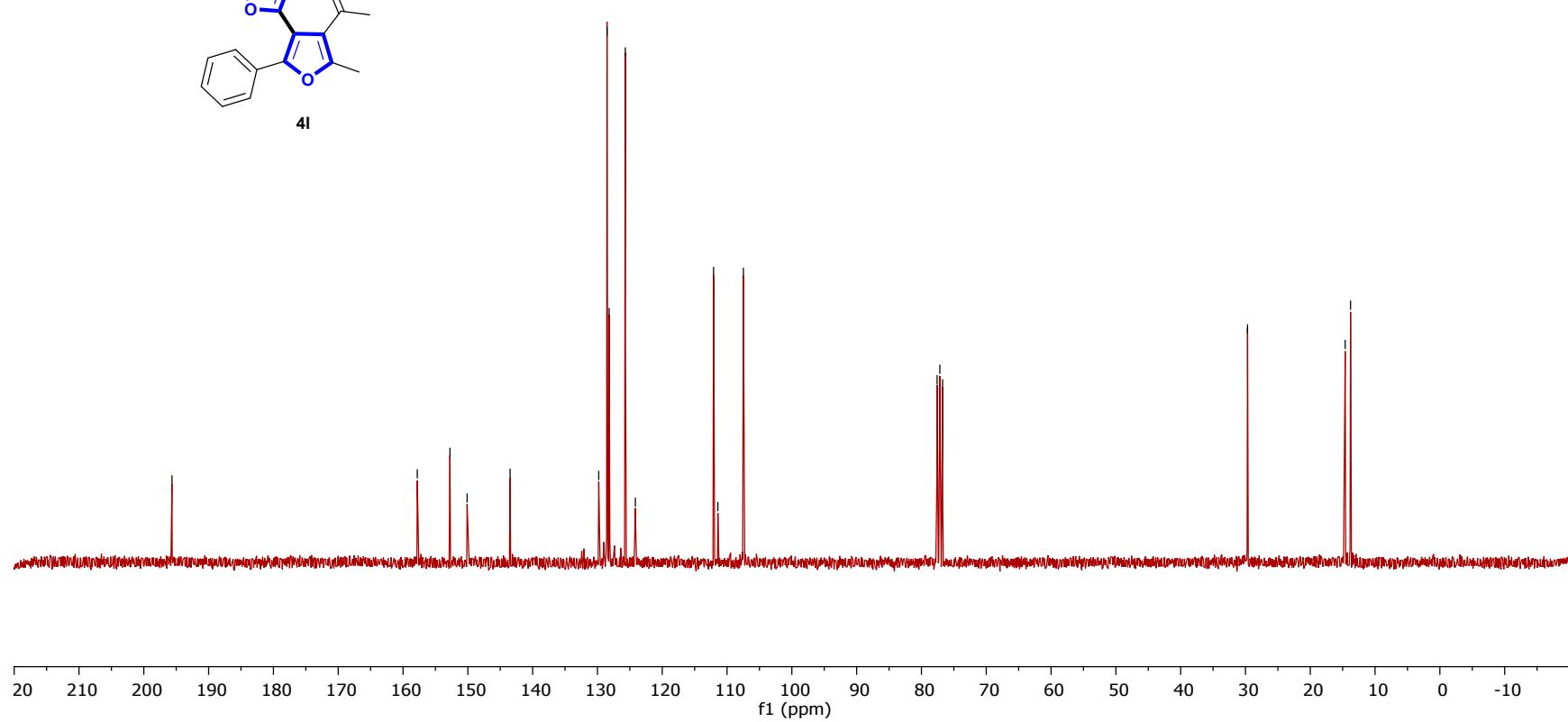
77.585
77.161
76.737

— 29.682

14.612
13.785



41

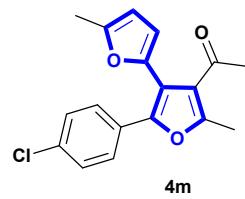


G:/NMR
CDCl₃

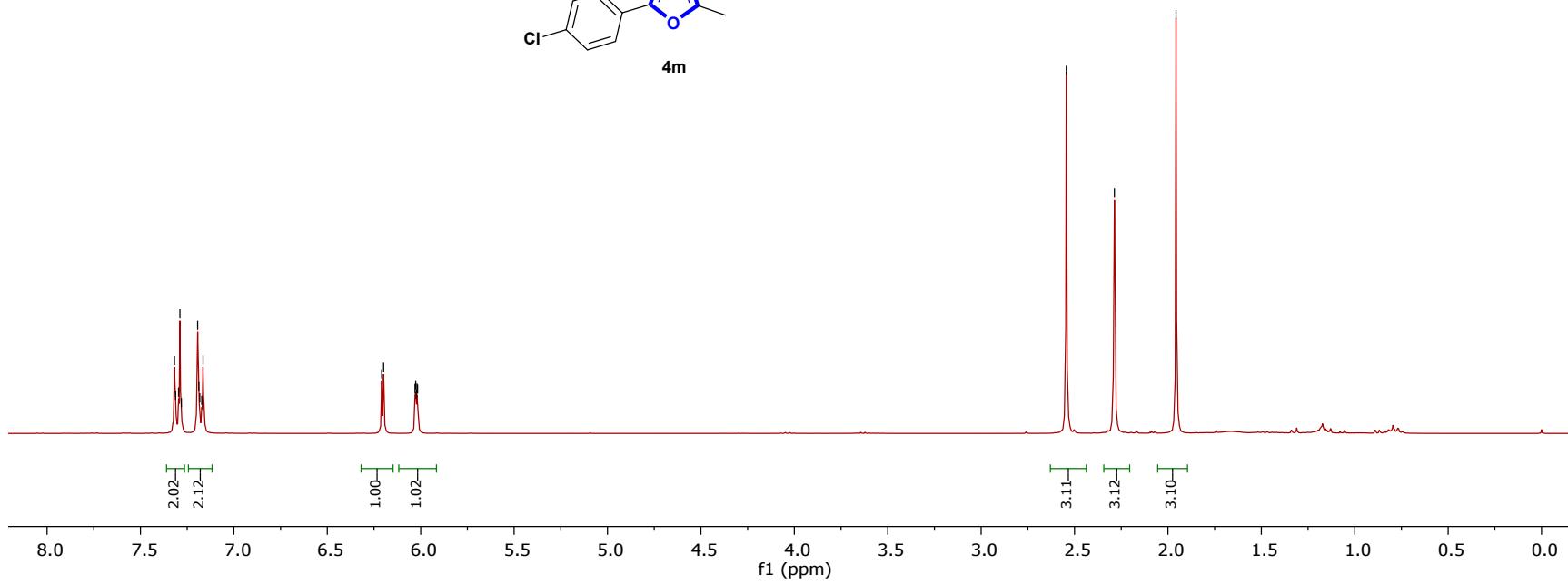
7.318
7.312
7.296
7.289
7.281
7.194
7.188
7.183
7.172
7.165

6.209
6.198
6.030
6.027
6.023
6.020
6.017

— 2.544
— 2.286
— 1.957



4m



G:/NMR/APAJ38/2/fid
CDCl₃

— 195.446

\ 157.998
~ 152.960
— 149.033
/ 143.107

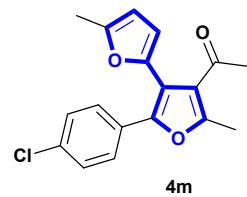
\ 133.972
/ 128.764
/ 128.244
~ 126.852
~ 124.186

\ 112.213
~ 111.851
~ 107.543

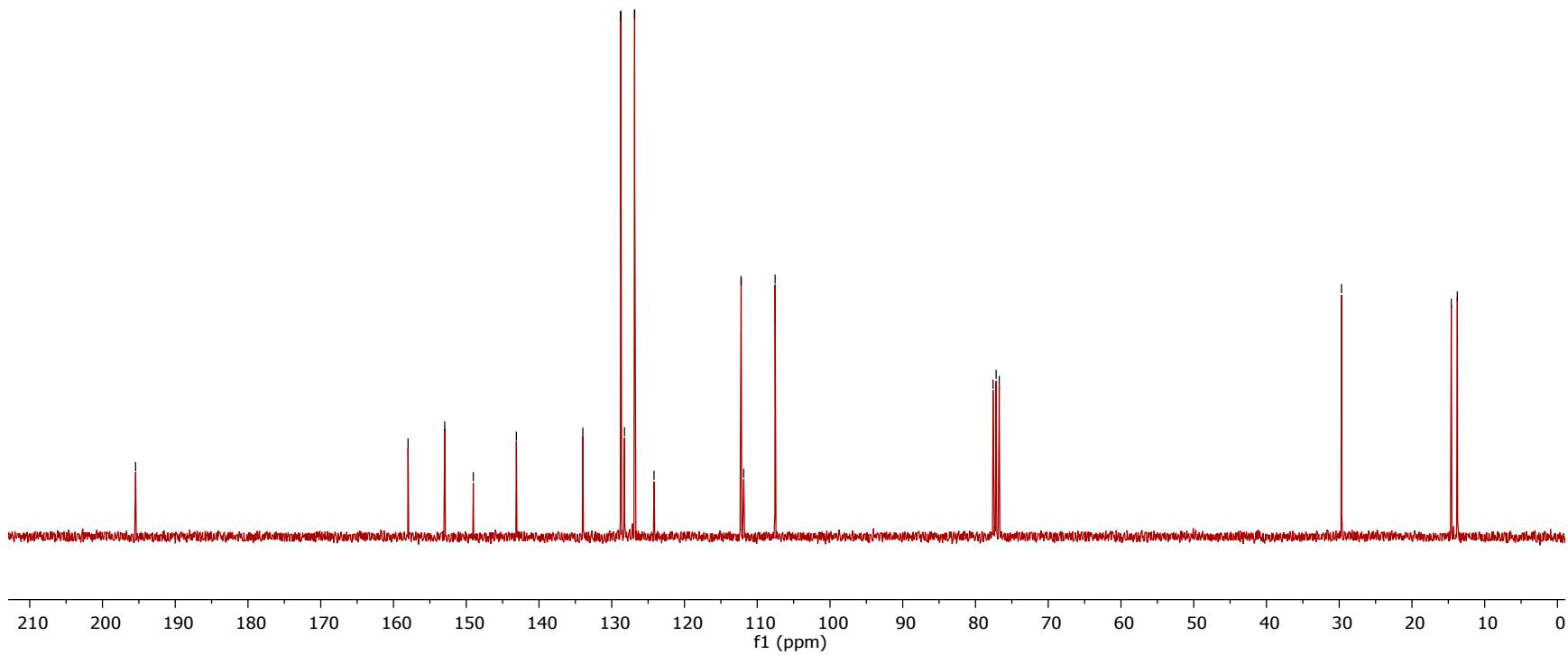
\ 77.585
/ 77.160
/ 76.736

— 29.687

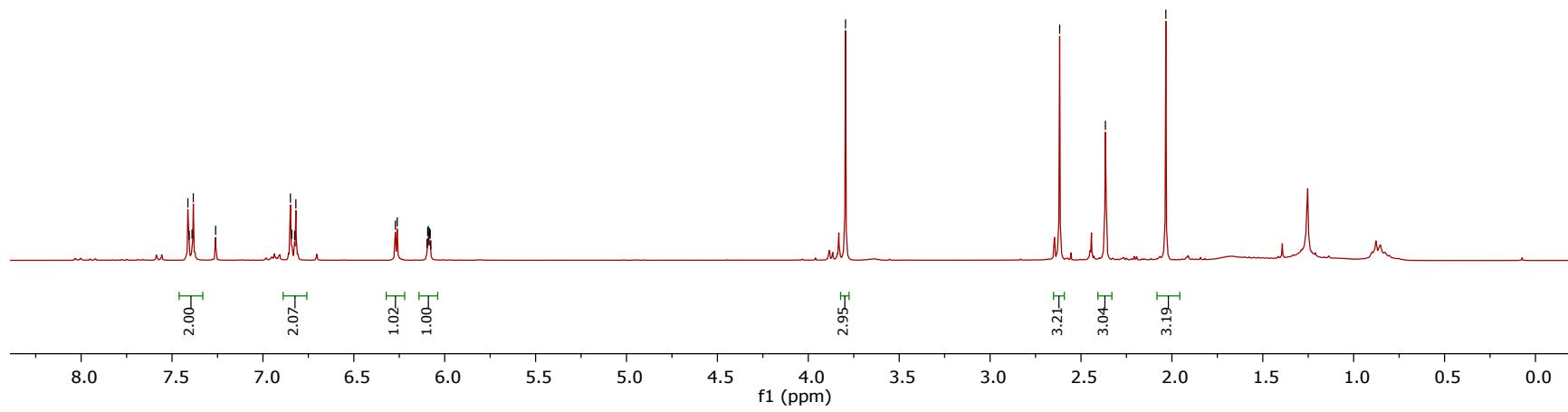
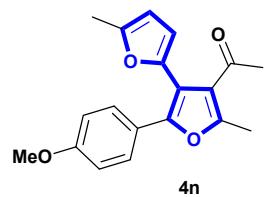
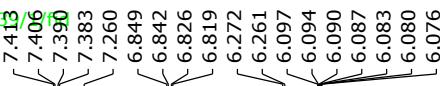
\ 14.597
~ 13.763



4m



G:/NMR/APAJ133
CDCl₃



G:/NMR/APAJ139/2/fid
CDCl₃⁷⁵⁶

— 195.736

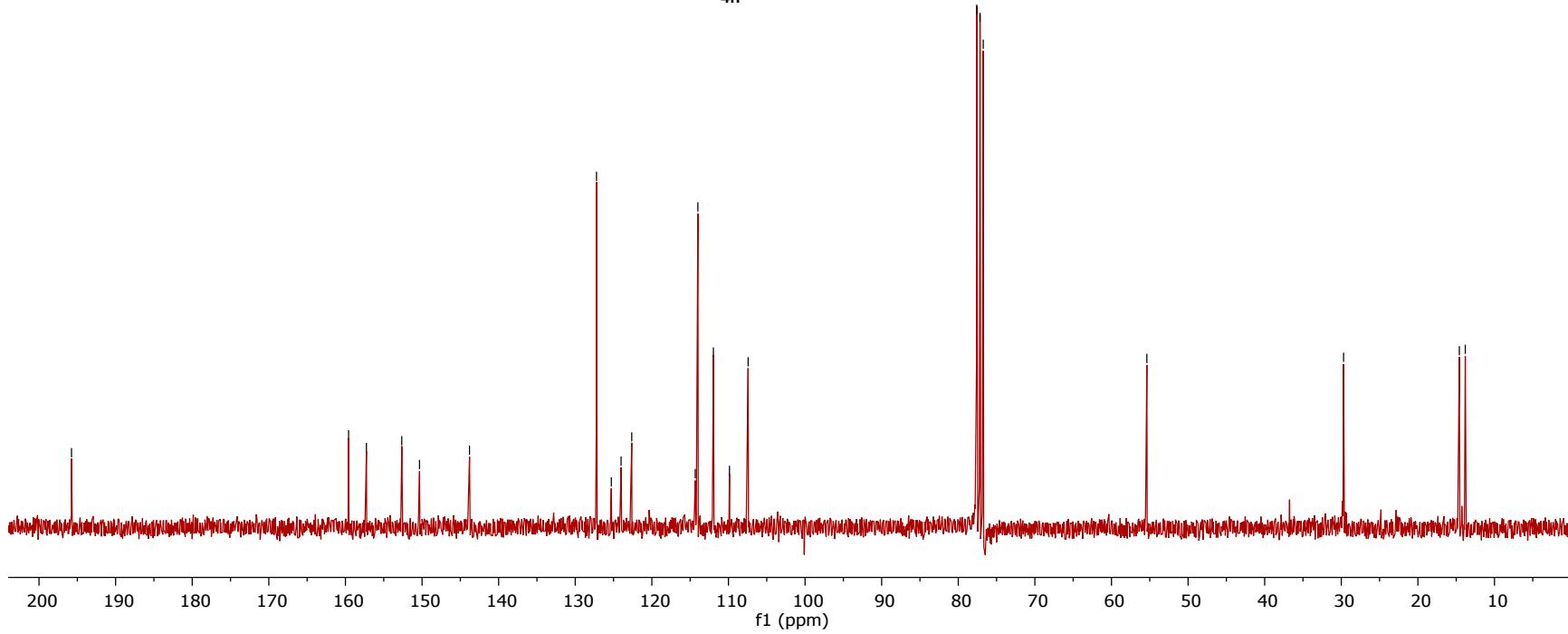
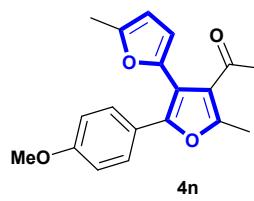
— 159.587

~ 157.258

~ 152.646

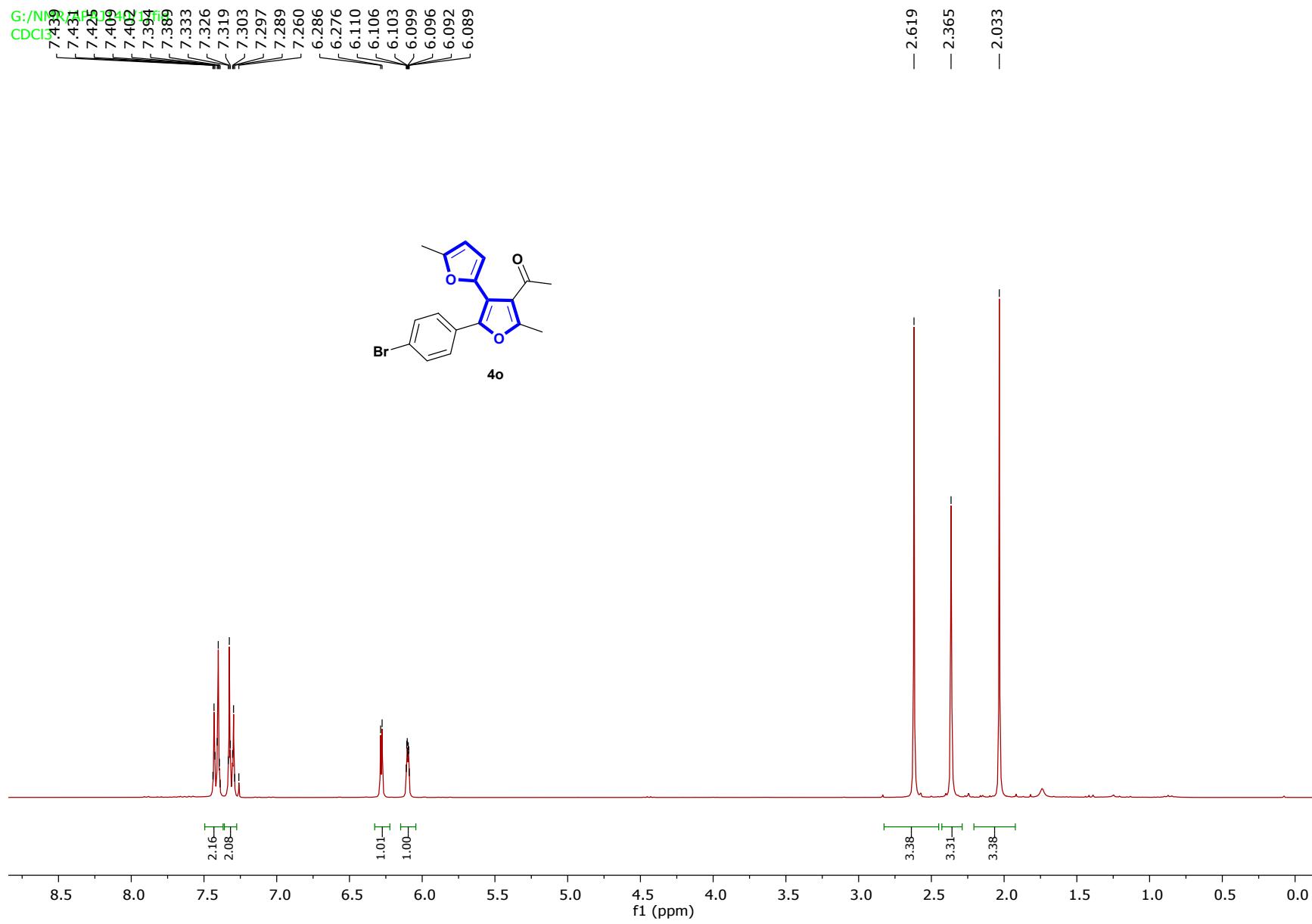
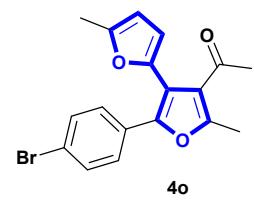
— 150.327

— 143.804

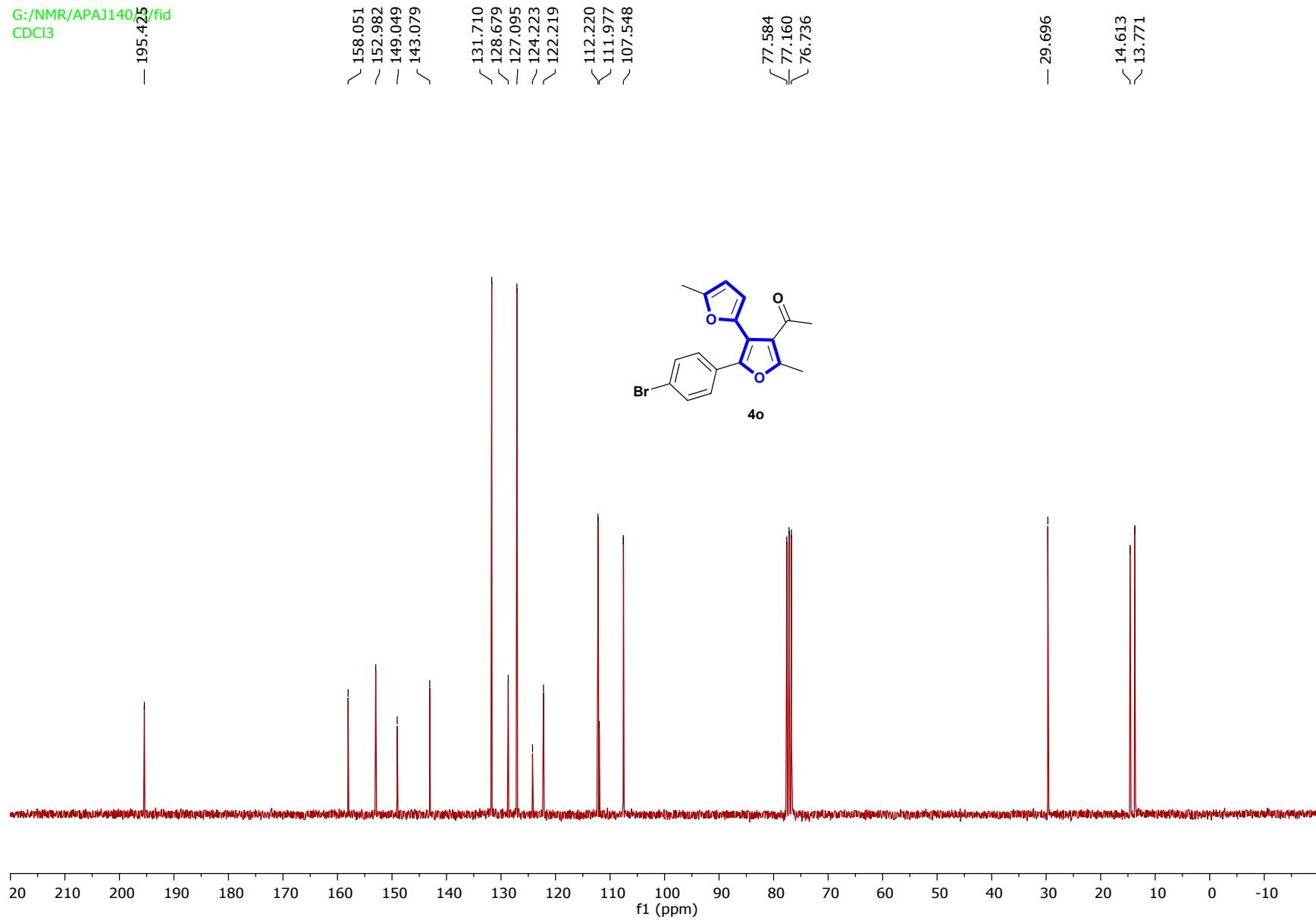


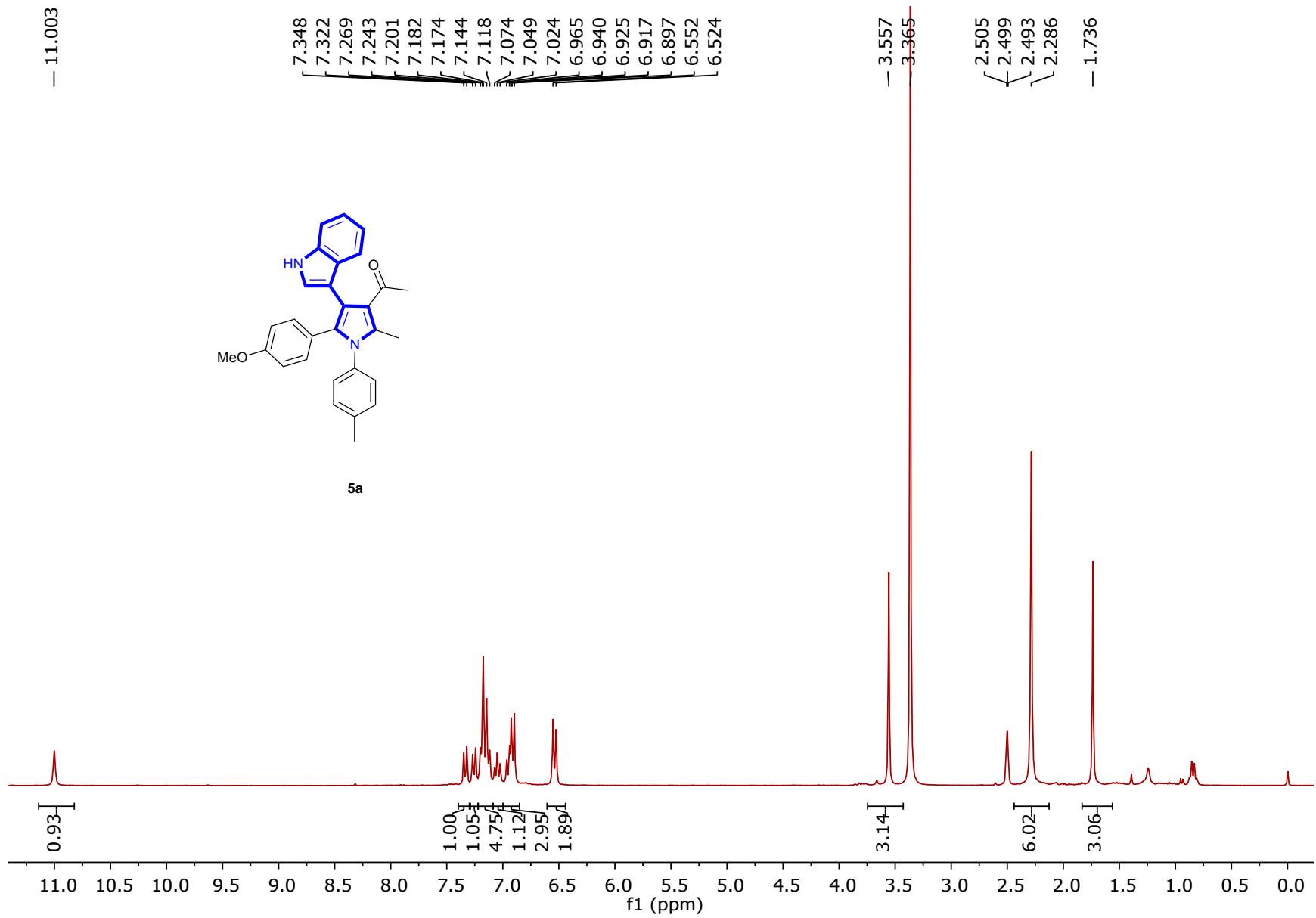
G:NMR 1H
CDCl₃

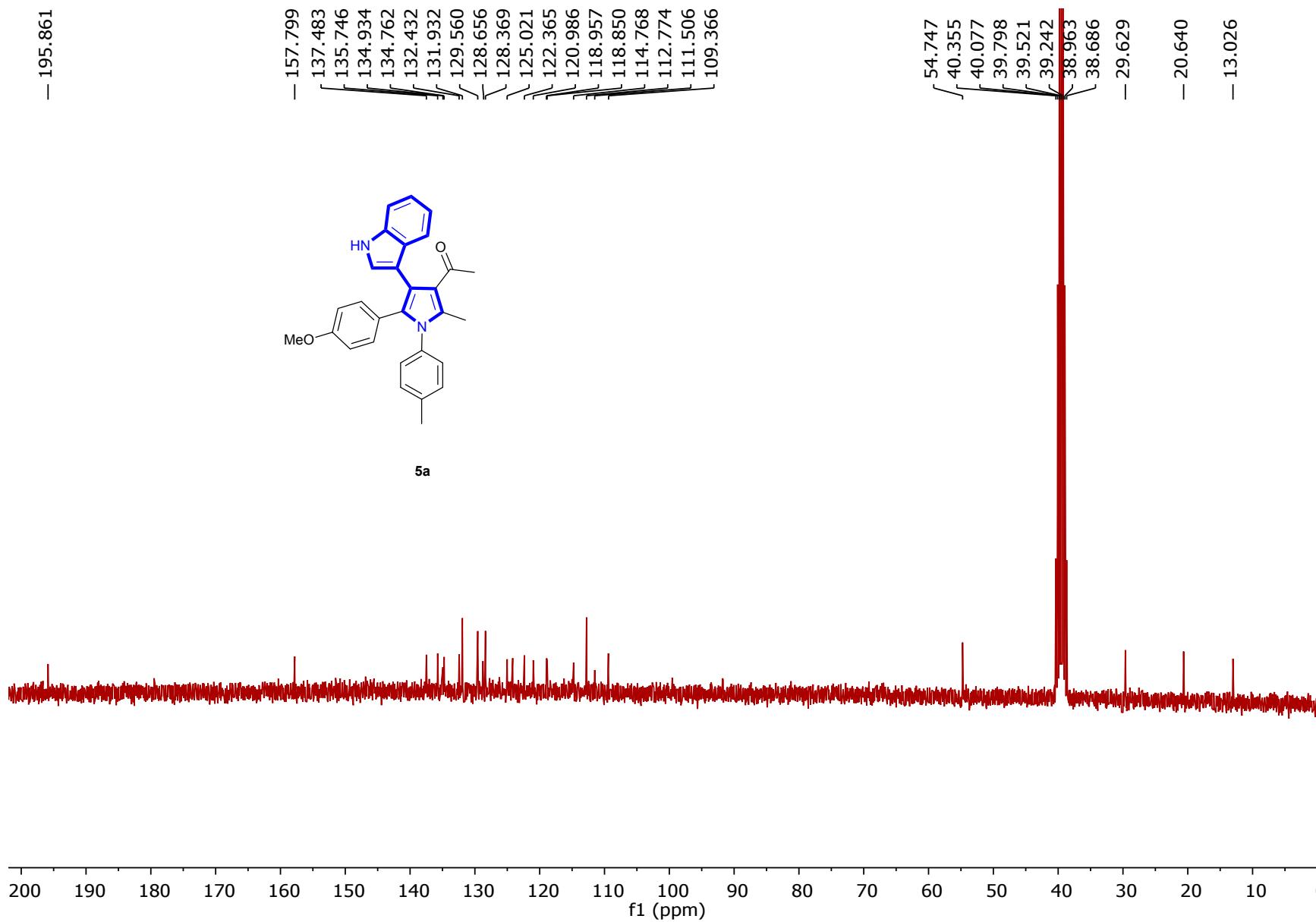
7.419
7.421
7.425
7.409
7.402
7.394
7.389
7.333
7.326
7.319
7.303
7.297
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6.110
6.106
6.103
6.099
6.096
6.092
6.089



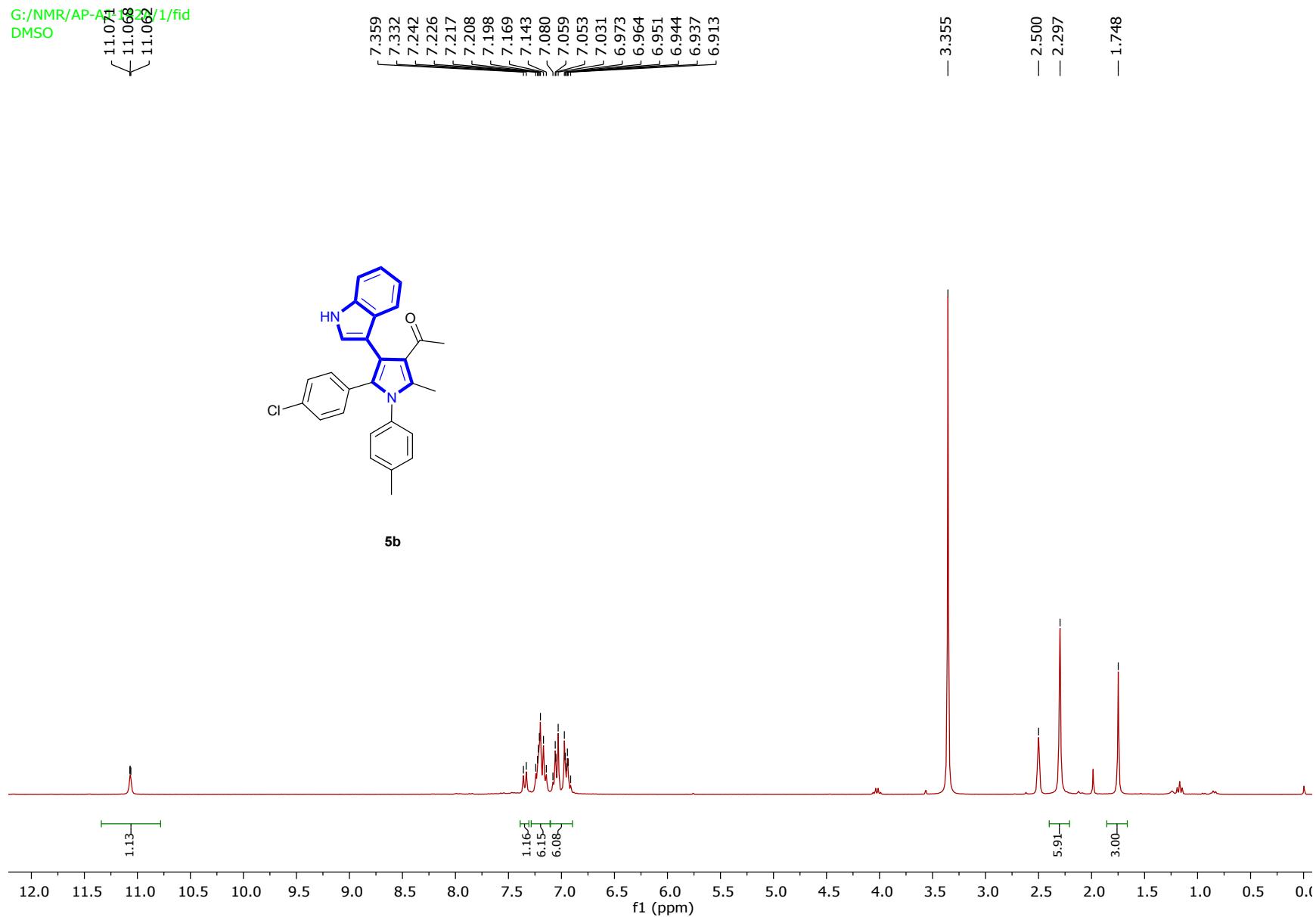
G:/NMR/APAJ140²⁵/fid
CDCl₃

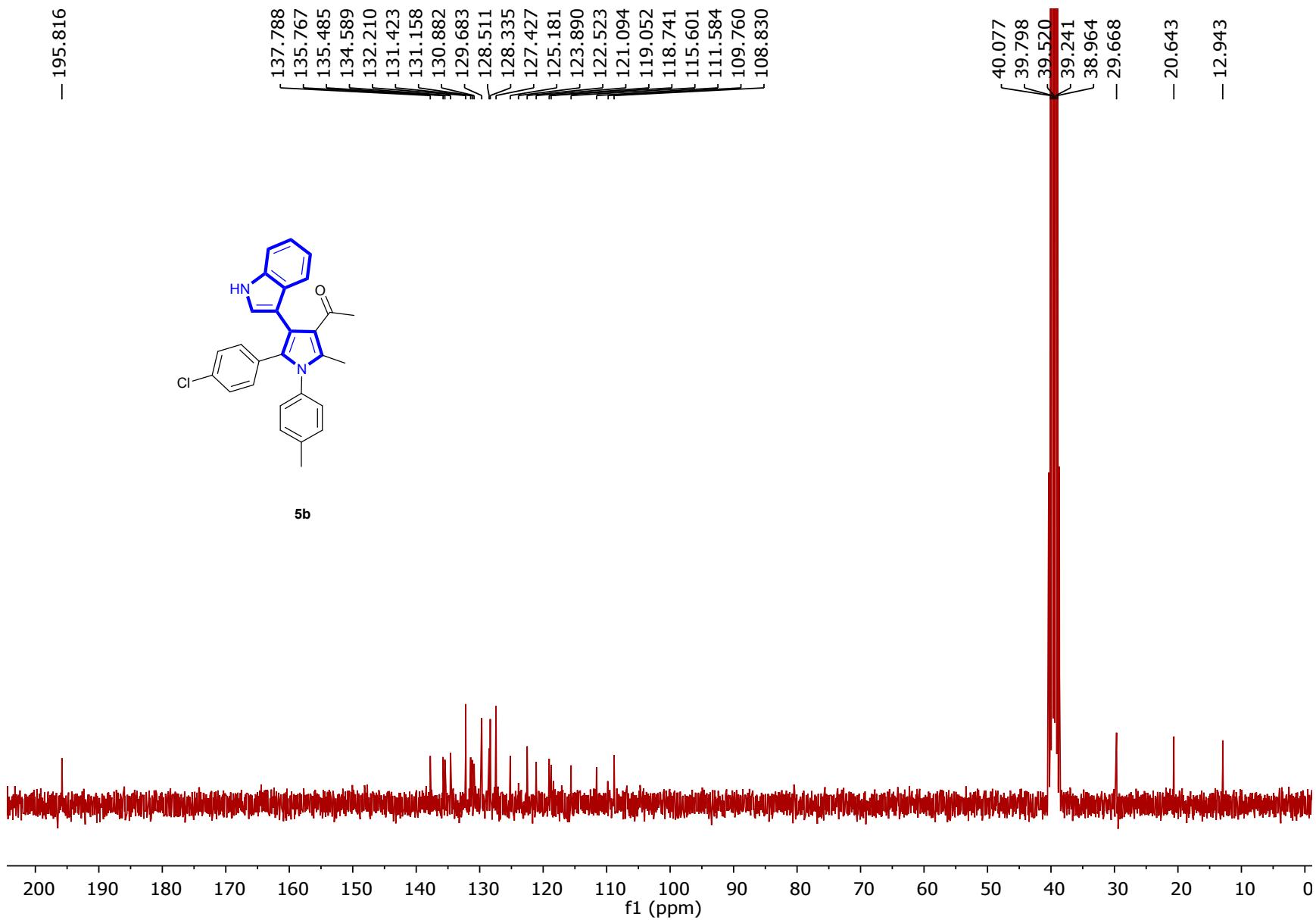


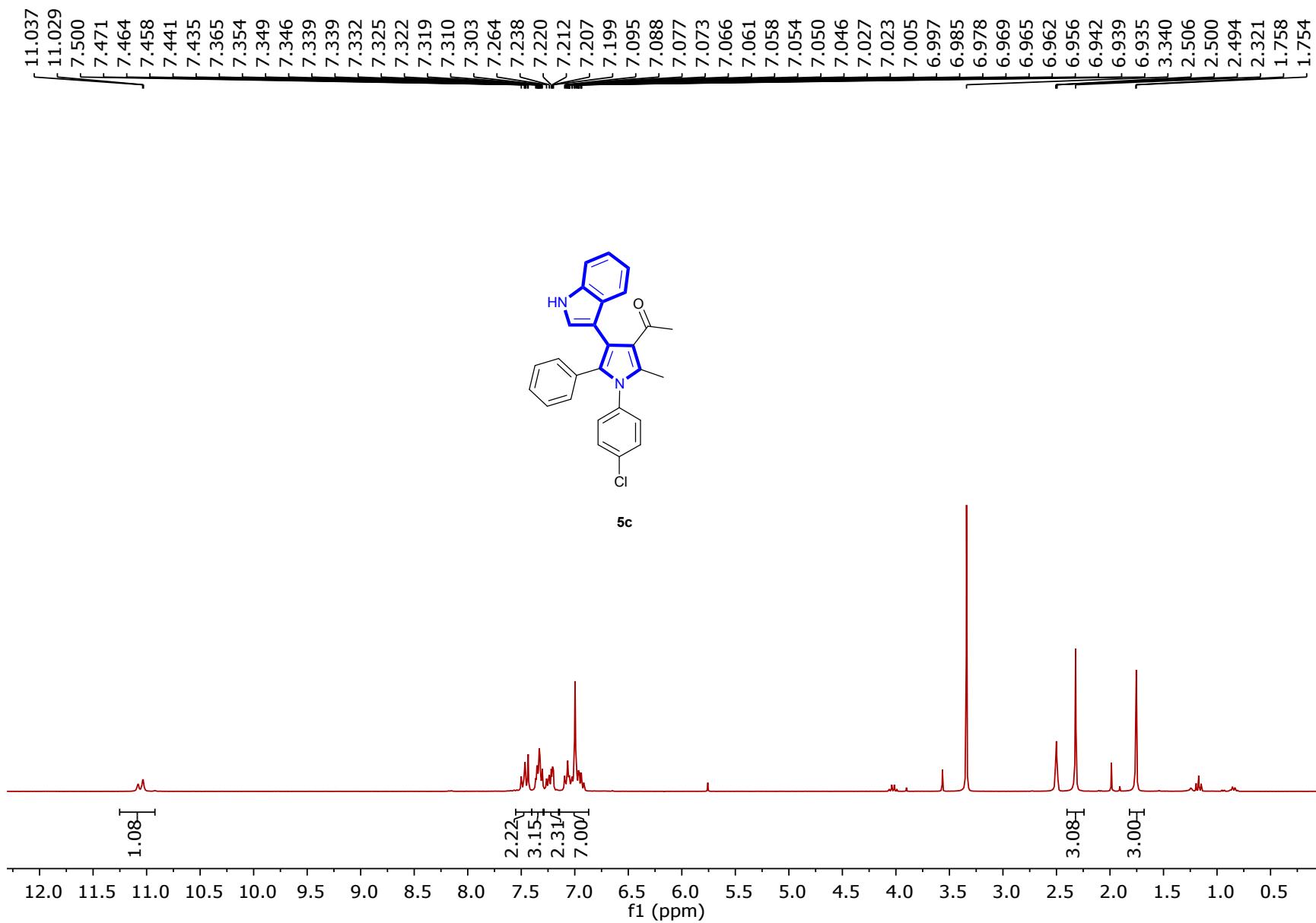


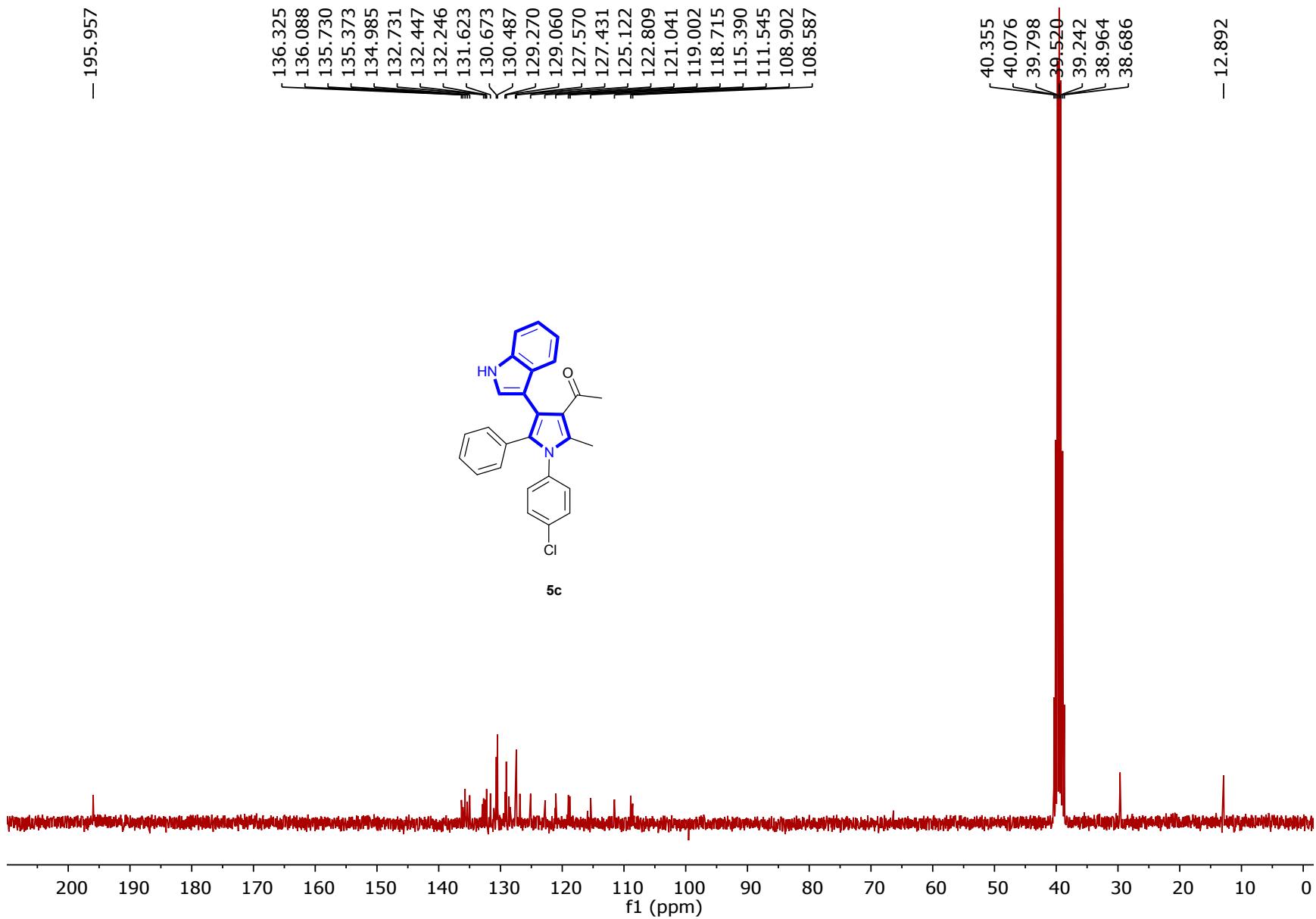


G:/NMR/AP-AP-1828/1/fid
DMSO









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1. (a) B. Eftekhari-Sis, M. Zirak, A. Akbari, *Chem. Rev.* **2013**, *113*, 2958-3043. (b) F. Kipnis, J. Ornfelt, *J. Am. Chem. Soc.* **1946**, *68*, 2734.
2. C. Bodhak, S. Hazra, and A. Pramanik, *ChemistrySelect* 2018, **3**, 7707 – 7712.