Supplementary information for

## Yolk-albumen-shell structure of mixed Ni-Co oxide with ultrathin carbon

## shell for high-sensitivity glucose sensors

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**Figure S1**. XRD pattern of (a) NiNP@C and (b) CoNP@C. (using Cu K<sub> $\alpha$ </sub> radiation with wave length  $\lambda = 0.15405$  nm and Ni filter with a step size of 0.02°)



Figure S2. SEM images (at an acceleration voltage of 15 kV) of NiNP@C



Figure S3. SEM images (at an acceleration voltage of 15 kV) of YASNiCo@C



Figure S4. SEM images (at an acceleration voltage of 15 kV) of CoNP@C



**Figure S5.** EELS chemical composition maps of NiNP@C from the area marked by the red square on the STEM micrograph. Individual Ni  $L_{2,3}$ -edges at 855 eV (red), O K-edge at 532 eV (blue) and C K-edge at 284 eV (yellow) as well as the composite (O-C and Ni-O-C) elemental maps.



**Figure S6.** EELS chemical composition maps of CoNP@C from the area marked by the red square on the STEM micrograph. Individual Co  $L_{2,3}$ -edges at 779 eV (green), O K-edge at 532 eV (blue) and C K-edge at 284 eV (yellow) as well as the composite (O-C and Co-O-C) elemental map.



**Figure S7**. Amperometric response of carbon paper to the addition of 1 mM glucose at +0.55 V vs Hg/HgO to a solution of 0.1 M NaOH.



**Figure S8**. Amperometric response of YASNiCo@C catalyst coated on carbon paper after the addition of 1 mM of glucose and different concentrations of interferences at an applied potential of +0.55 V vs Hg/HgO to a solution of 0.1 M NaOH.

<u>10</u> po	rea in interatares.						
Ref.	Catalyst	Sensitivity	linear range	applied	limit of detection	electrolyte	
		$(\mu A \ cm^{-2} \ mM^{-1})$	(µM)	potential (V)	(µM)		
Cobalt oxide							
S1	Co <sub>3</sub> O <sub>4</sub> nanostructures	246.8	0.5-1000	0.54	0.012	0.1 M NaOH	
S2	Cobalt nitride (Co <sub>4</sub> N)	1137.2	600-10000	0.55	0.1	0.1 M NaOH	
S3	Cobalt nitride nanosheets	921.18	10-8000	0.55	0.1	0.1 M NaOH	
S4	Mesoporous Co <sub>3</sub> O <sub>4</sub> Nanobundle	88	0.6-160	0.58	0.6	0.1 M NaOH	
S5	Co-CoO-Co <sub>3</sub> O <sub>4</sub> nanocomposites	949.3	5-600	0.55	0.92	0.1 M KOH	
<b>S</b> 6	Cobalt oxide nanoflowers	693.02	5-60	0.60	0.04	0.01M NaOH	
<b>S</b> 7	$3D/Co_3O_4$ thorn-like nanostructures	180	1-1000	0.55	0.046	0.1 M NaOH	
<b>S</b> 8	3D hierarchical porous Co <sub>3</sub> O <sub>4</sub> film	366.03	1-500	0.60	1	0.1 M NaOH	
S9	Porous CoOOH nanosheet	526.8	3-1109	0.52	1.37	0.1 M NaOH	
S10	Nanoporous cobalt oxide nanowires	300.8	5-570	0.60	5	0.3 M NaOH	
S11	Co <sub>3</sub> O <sub>4</sub> nanoparticles	471.5	1-300	0.59	0.1	0.1 M NaOH	
S12	CoNPs/ITO	1720	5-180	0.59	0.25	1 M NaOH	
Nickel oxide							
S13	Sandpaper-supported nickel coatings	1163.3	10-4300	0.60	0.25	0.1 M NaOH	
S14	Ni layers modifie dboron-dope ddiamon	839.3	10-5640	0.40	1.23	1 M NaOH	
S15	Nanostructured NiO electrode	206.9	100-10000	0.55	1.16	0.5 M NaOH	
S16	Ni(OH) <sub>2</sub> /PGE	948	4-3500	0.57	2	0.1 M NaOH	
S17	Nf-Ni(OH)2@oPPyNW	1049.2	1-4863	0.54	0.3	0.1 M NaOH	
S18	NiO nanostructures	1915	100-5000	0.48	0.7	0.1 M NaOH	
S19	Hedgehog-like NiO nanostructures	1052.8	500-4500	0.49	1.2	0.1 M NaOH	
S20	NiO nanosheets	36.13	up to 10000	0.50	0.9	0.5 M NaOH	
S21	Macro-mesoporous Ni(OH) <sub>2</sub>	243	10-8300	0.50	1	0.1 M NaOH	
Cobalt-Nickel bimetallic oxide							
S22	Ni-Co Layered Double Hydroxide	292.84	5-50	0.55	0.8	0.1 M NaOH	
S23	NiCo2O4 nanowires	72.4	0.37-2000	0.44	0.37	0.1 M NaOH	
S24	NiCo2O4 hollow nanospheres	1917	10-300	0.55	0.6	0.2 M NaOH	
S25	NiCo2O4 hollow nanorods	1685.1	3-1000	0.60	0.16	0.1 M NaOH	
S26	Ni-Co bimetal nanowires	695	5-10000	0.45.	1.2	0.1 M NaOH	
S27	NiCo-LDH	1235	5-1200	0.50	1.6	0.1 M NaOH	
S28	Ni-Co alloy nanostructures	536.2	1-5000	0.50	0.39	0.1 M NaOH	
MOF-derived metal oxide							
S29	CuO nanorod	1523.5	up to 1250	0.60	1	0.1 M NaOH	
S30	GS@ZIF-67 hybrids	1521.1	1-805.5	0.55	0.36	0.1 M NaOH	
S31	Cu Nanospheres@Porous carbon	28.67	0.15-5620	0.45	0.48	0.1 M NaOH	
S32	Cu-based structures	1255	3.76-1400	0.50	3.76	0.1 M NaOH	
S33	Co nanoparticles with porous carbon	227	100-1100	0.50	5.69	0.1 M NaOH	
S34	CuO/NiO-Carbon Nanocomposite	586.7	0.1-4500	0.65	0.037	0.1 M NaOH	
S35	NiCo LDH nanosheets/graphene	344	5-800	0.60	0.6	0.1 M NaOH	
This	YASNiCo@C	1964	5-1000	0.55	0.75	0.1 M NaOH	

## **Table S8.** Comparison of the sensitivity of YASNiCo@C electrode with other electrodes reported in literatures.

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