Supporting information for

Hierarchical SnS@ZnIn₂S₄ marigold flower 2 D nano heterostructure as an efficient photocatalyst for sunlight-driven hydrogen generation

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ESI Fig. I



ESI Fig. I X-ray diffraction patterns of pure SnS

ESI Fig. II



ESI Fig. II. Nitrogen adsorption-desorption isotherms obtained from the samples: (S1) $ZnIn_2S_4$ and (S3) $SnS@ZnIn_2S_4$.

Code	Average Pore Size	Pore Volume	Surface Area
S1(PureZnIn ₂ S ₄)	7.5171e+000nm	2.8089e-001cc/g	74.7352m ² /g
S3 (2%)	5.1514e+000nm	1.8722e-oo1cc/g	78.3929m ² /g

ESI Table 1Textual property values BET Study

ESI Fig. III



ESI Fig. III. UV-DRS spectra of pure Sn

ESI Fig. IV



ESI Fig. IV. Photoluminescence spectra of pure SnS



ESI-V Reusability study of H₂ (µmole) evolution from H₂S

Time versus volume of H_2 (µmole) evolution from H_2S of the S3 prepared with a different percent of SnS loading (S3) 2%.

Table 2: The H₂ generation rates for as-synthesized Sample S3 and Repeatability

Sr. No.	Sample code	H ₂ evolution rate from H ₂ S (µmolh ⁻¹ g ⁻¹)		
1	S3	6429		
2	Cycle 1	6222		
3	Cycle 2	5900		
4	Cycle 3	5500		



ESI-VI Reusability study of H_2 (µmole) evolution from H_2O

Table 3: The H₂ generation rates for as-synthesized Sample S3 and Repeatability Study

Sr. No.	Sample code	H ₂ evolution rate from H ₂ O (μmolh ⁻¹ g ⁻¹)		
1	S3	650		
2	Cycle 1	610		
3	Cycle 2	575		
4	Cycle 3	531		

Table 4: Comparison of rate of Photocatalytic H2 Production of similarheterostructure system reported previously

Photocatalyst	Sacrificial	Source	H ₂	H ₂	References
	reagent		Production	Production	
	system		rate From	rate From	
			H ₂ O	H_2S	
ZnFe ₂ O ₄ /ZnIn ₂ S ₄	0.35 M	300 W	79.0 µmol	-	1
	$Na_2S + 0.25$	Xe-lamp	h^{-1}		
	$M Na_2 SO_3$	-			
CdS	Na_2S (5 mL,	300 W	2.7 mmol h ⁻¹	-	2
QDs/graphene/ZnIn ₂ S ₄	$0.1 \text{ mol } L^{-1}) +$	Xe-lamp			
	Na ₂ SO ₃				
	(5 mL, 0.04				
	mol L ⁻¹)				
Cu-Doped ZnIn ₂ S ₄	0.25 M	300 W	151.5	-	3
	$Na_2S + 0.35$	Xe-lamp	µmol/h		
	$M Na_2 SO_3$	_			
AgIn ₅ S ₈ nanoparticles	0.25 M	300 W	265.9 µmol	-	4
anchored on 2D	$Na_2S + 0.25$	Xe-lamp	g ⁻¹ h ⁻¹		
layered ZnIn ₂ S ₄	$M Na_2 SO_3$	_	_		
NiS/ZnIn ₂ S ₄	$0.5 \text{ M} \text{Na}_2 \text{SO}_3$	300 W	104.7	-	5
	+ 0.43 M	Xe-lamp	µmol/h		
	Na_2S	-			
SnS@ZnIn ₂ S ₄	Na ₂ S/Na ₂ S ₂ O ₃	Sunlight	650 μmol h ⁻ ¹ g ⁻¹	$\begin{array}{c} 6429 \ \mu mol \\ h^{-1}g^{-1} \end{array}$	Present Work

References

- 1. Y. Chen, G.Tian, W. Zhou, Y.Xiao, J. Wang, X. Zhang and H. Fu, *Nanoscale*, 2017, 9, 5912.
- 2. J. Hou, C. Yang, H. Cheng, Z. Wang, S. Jiao and H. Zhu *Phys. Chem. Chem. Phys.*, 2013, **15**, 15660-15668
- 3. S. Shen, L. Zhao, Z. Zhou and L. Guo, J. Phys. Chem. C, 2008, 112, 41.
- 4. Z. Guana, Z. Xuac, Q. Lia, P. Wanga, G. Lib, and J. Yanga, *APPL CATAL B-ENVIRON*. 2018.
- 5. L. Wei, Y. Chen, J. Zhao and Z. Li, J. Nanotechnol. 2013, 4, 949.