

# **Sulphonic Acid Functionalized Porphyrin Anchored with meso-Substituted Triazolium Ionic Liquid Moiety: Heterogeneous Photo-Catalyst for Metal/Base Free C-C Cross-Coupling and C-N/C-H activation using Aryl Chloride under visible light irradiations**

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*#Equal contribution, Shital contributed for C-N coupling*

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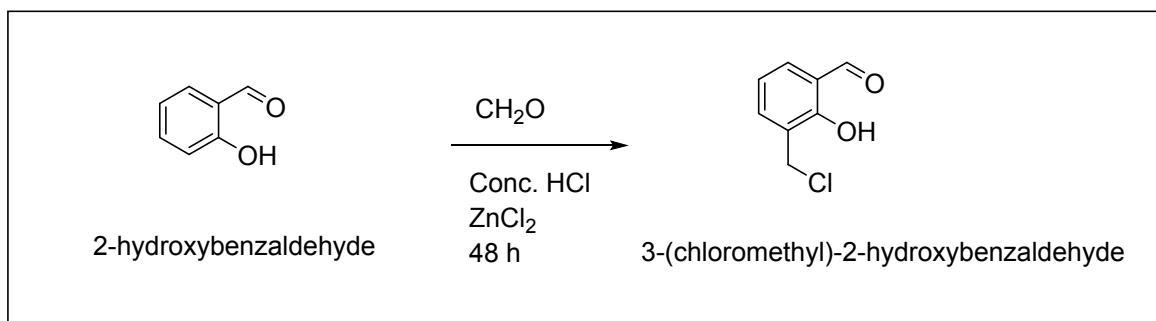
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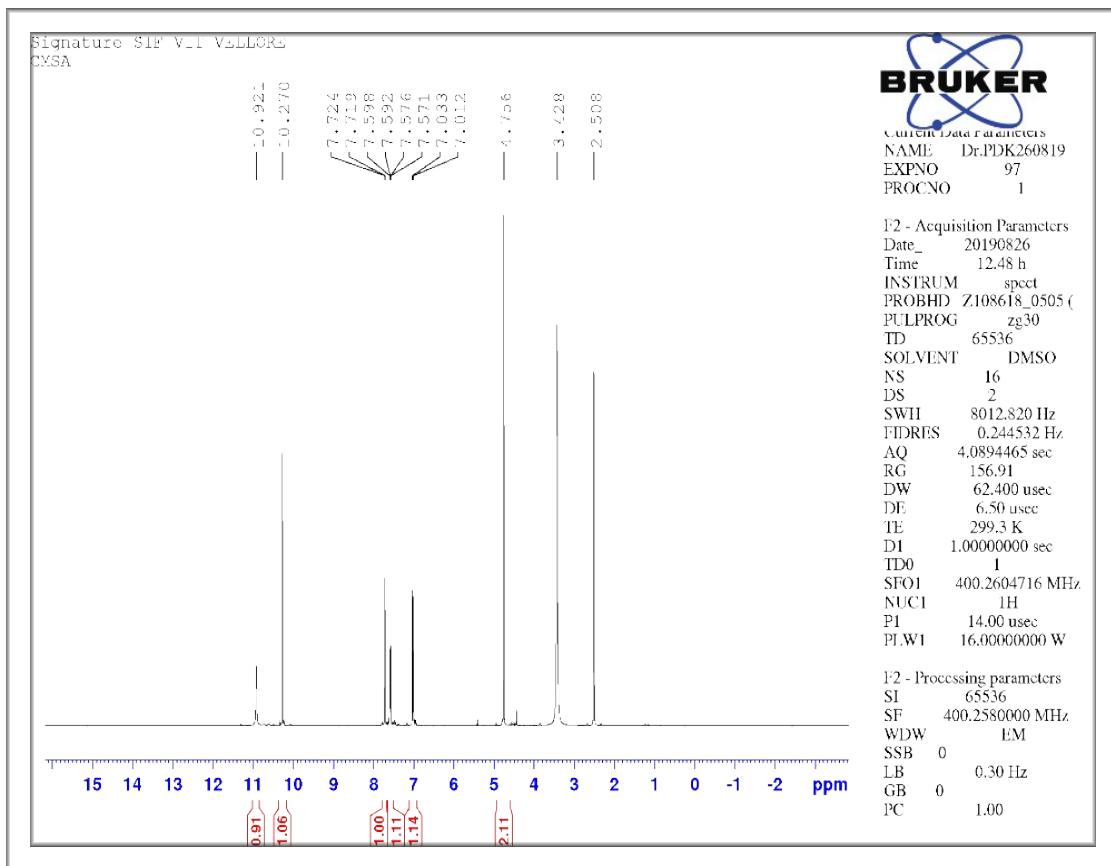
## Synthesis of SAFPTILM

### Synthesis of 3-(chloromethyl)-2-hydroxybenzaldehyde (1a)



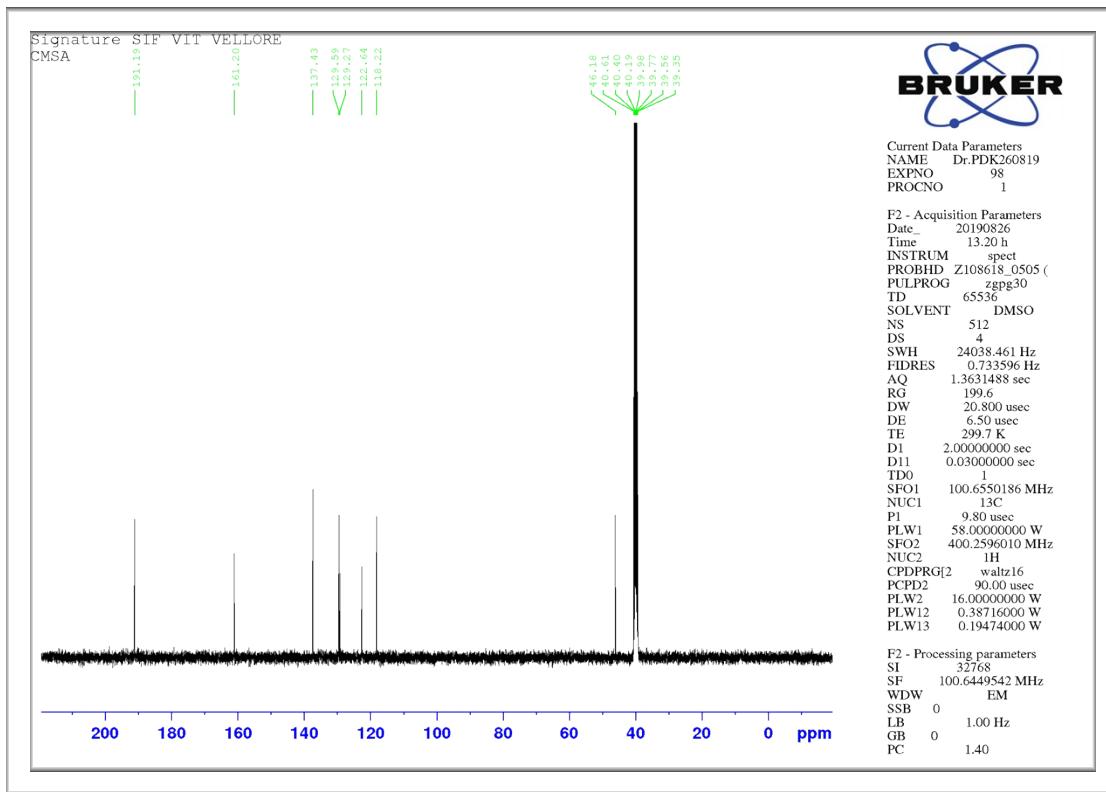
**Scheme. S1.** Synthesis of 1a

The synthesis of 1a was achieved by adapting procedure from the known literature. In a 50 mL round bottom flask (RBF), the mixture of Conc. HCl (30 mL) and formaldehyde (3 mL, 100 mmol) was cooled at 0 °C and allowed constant stirring for 30 min. To this mixture, salicylaldehyde (2.92 mL, 24 mmol) was charged gradually and reaction mixture was kept on magnetic stirring for 48 h at room temperature. The precipitated desired 1a solid (white) was separated and washed with 3 x 10 mL distilled water, followed by acetone and recrystallized from n-hexane. The resulting compound (1a) was dried in hot air oven to give 3.5 g (Yield: 87 %)



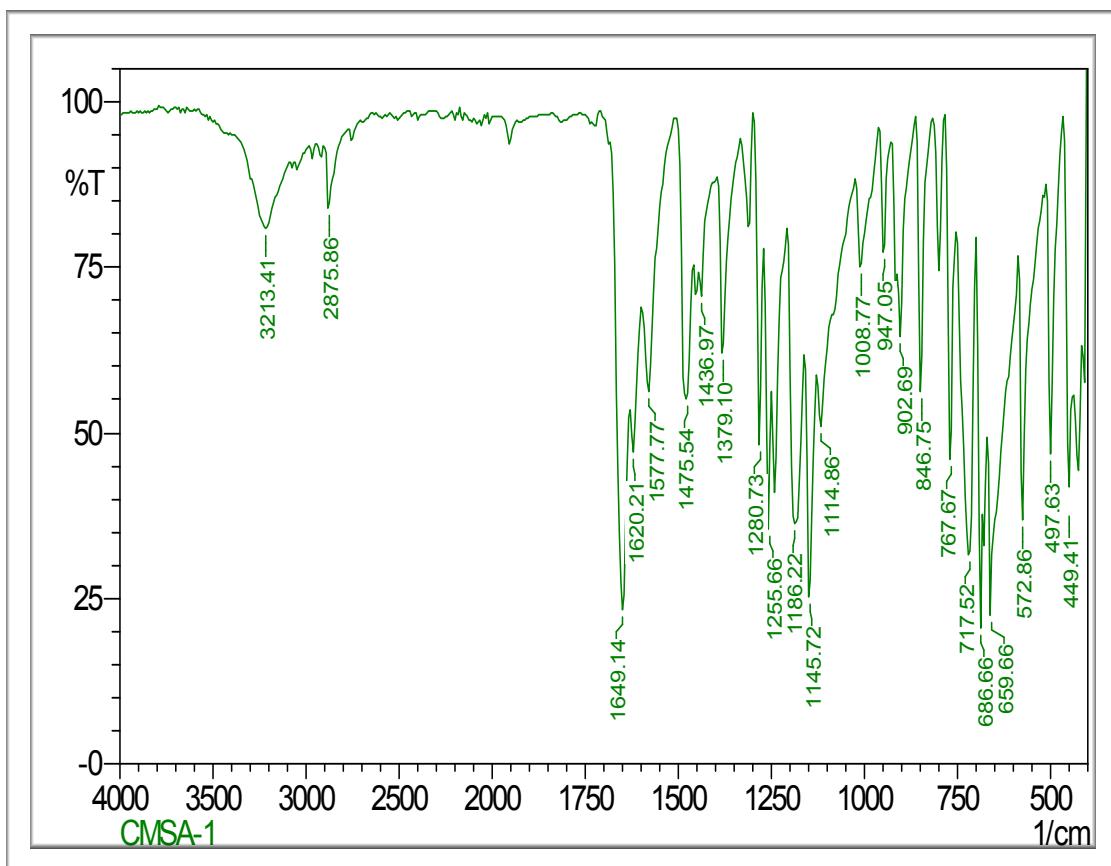
**Fig. S1**  $^1\text{H}$  NMR Spectrum of 3-(chloromethyl)-2-hydroxybenzaldehyde

$^1\text{H}$  NMR (400 MHz, DMSO-d<sub>6</sub>)  $\delta$ :  $\delta$  10.92 (s, 1H), 10.27 (s, 1H), 7.72 (d, 1H), 7.58 (q, 1H), 7.02 (d, 1H), 4.75 (s, 2H).



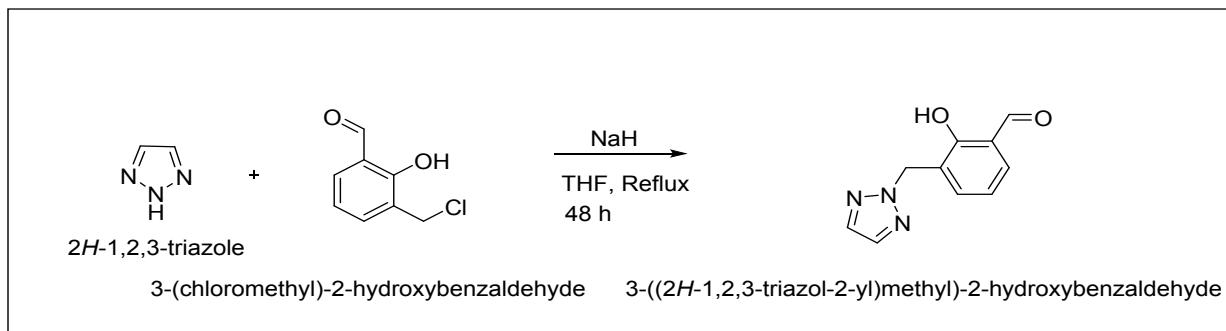
**Fig. S2**  $^{13}\text{C}$  NMR Spectrum of 3-(chloromethyl)-2-hydroxybenzaldehyde

$^{13}\text{C}$  NMR (100 MHz, DMSO-d6)  $\delta$ : 191.19, 161.20, 137.43, 129.59, 129.27, 122.44, 118.22, 44.16



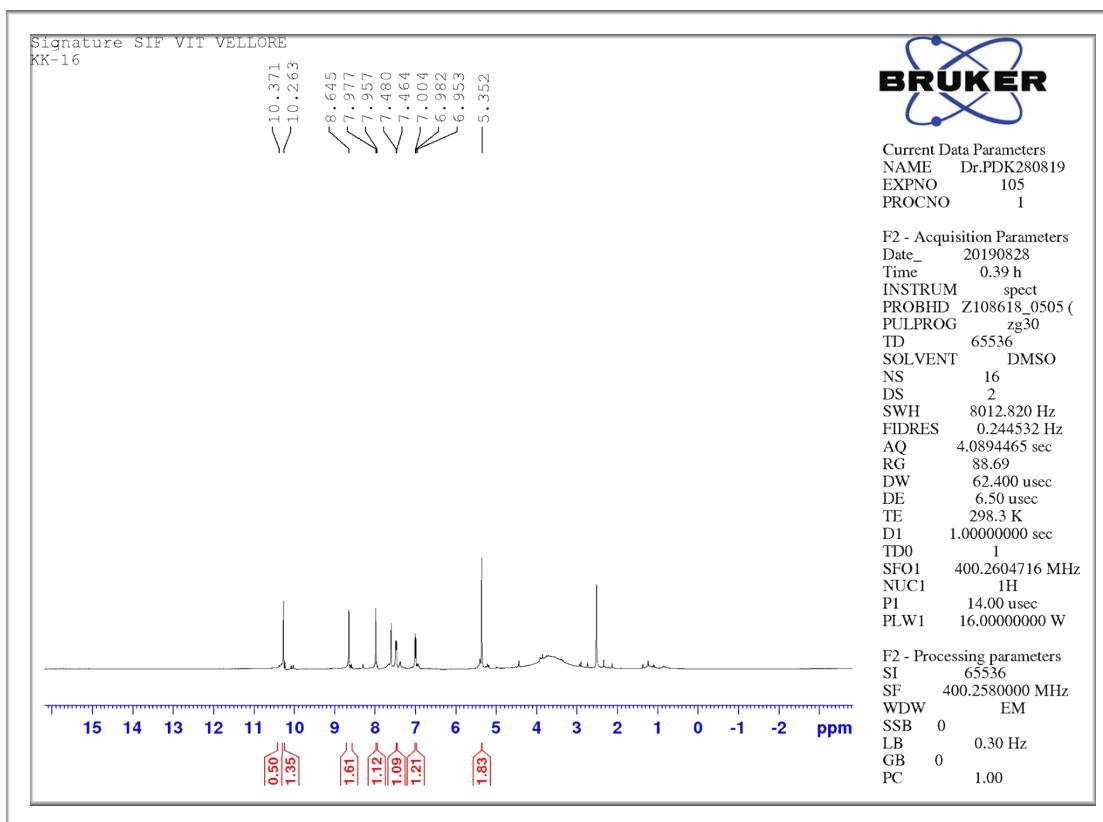
**Fig. S3** FT-IR: (KBr,  $\nu/\text{cm}^{-1}$ ):– 3213.41, 2875.86, 1649.14, 1620.21, 1577.77, 1475.54, 1436.97, 1379.10, 1280.73, 1255.66, 1186.22, 1145.72, 1114.86, 1008.77, 947.05, 902.69, 846.75, 767.67, 717.52, 686.66, 659.66, 572.86, 497.63, 449.41.

### Synthesis of 3-((2H-1,2,3-triazole-2-yl)methyl)-2-hydroxybenzaldehyde (1b)

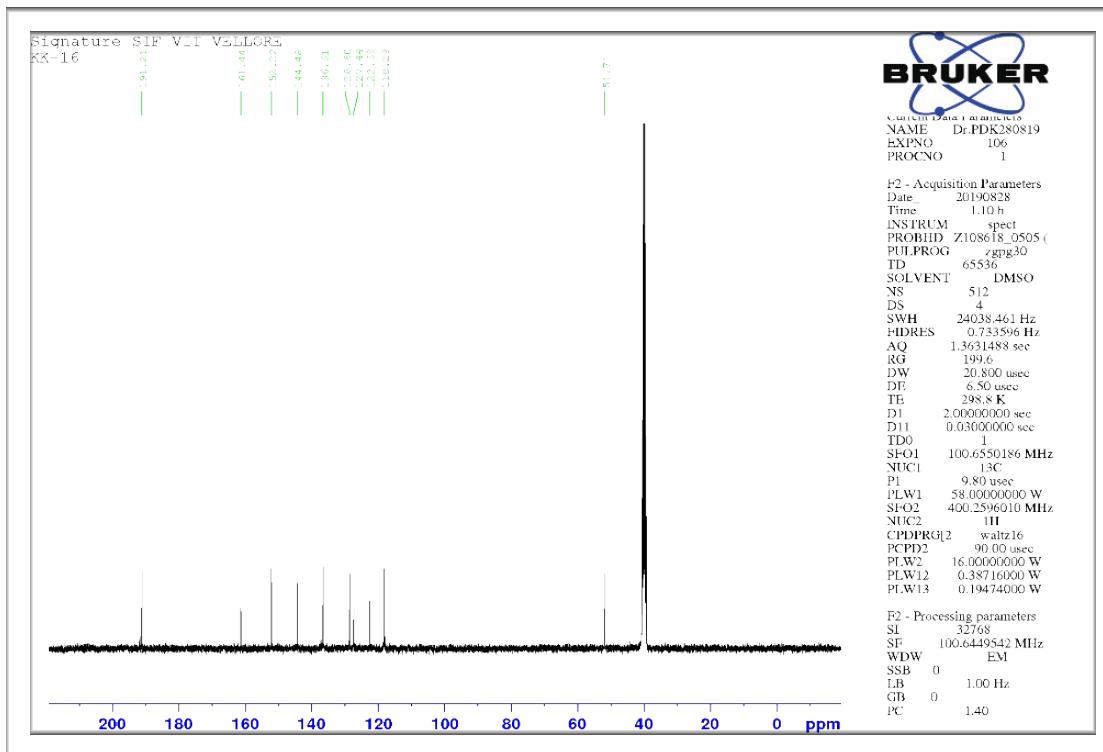


**Scheme. S2** Synthesis of 1b

Into a 100 mL RBF sodium hydride was placed and mixed with 30 mL n-hexane. The reaction mixture was stirred at room temperature (RT) for 40 min., after which time the excess solvent was decanted under vacuum to remove paraffin oil.<sup>28</sup> The flask containing NaH (1.32 g, 55 mmol) was charged with 15 mL tetrahydrofuran (THF), cooled in ice bath to maintain 2 to 3 °C temperature, and solution of triazole (3.45 g, 50 mmol) in THF was added dropwise and stirred for 2 h at cold condition. The reaction mixture was loaded with chloromethylated salicylaldehyde (1a) (8.50 g, 50 mmol) in THF very slowly at the same temperature. The reaction mixture was then reflux for 48 h (60 °C). After cooling, the salt was removed by filtration and the solvent was evaporated under vacuum by rotavapour to get sticky yellowish liquid. The product was charged with the mixture of ethyl acetate and saturated brine solution to remove remaining salt. The organic layer was separated and dried using anhydrous sodium sulphate for overnight. The ethyl acetate layer was concentrated using rotavapour to get sticky yellowish solid. The resulting solid was rinsed by diethyl ether (3 X 10 mL) and dried in hot air oven to afford yellowish fine powder 1b. Yield-84%, 8.5 gram.



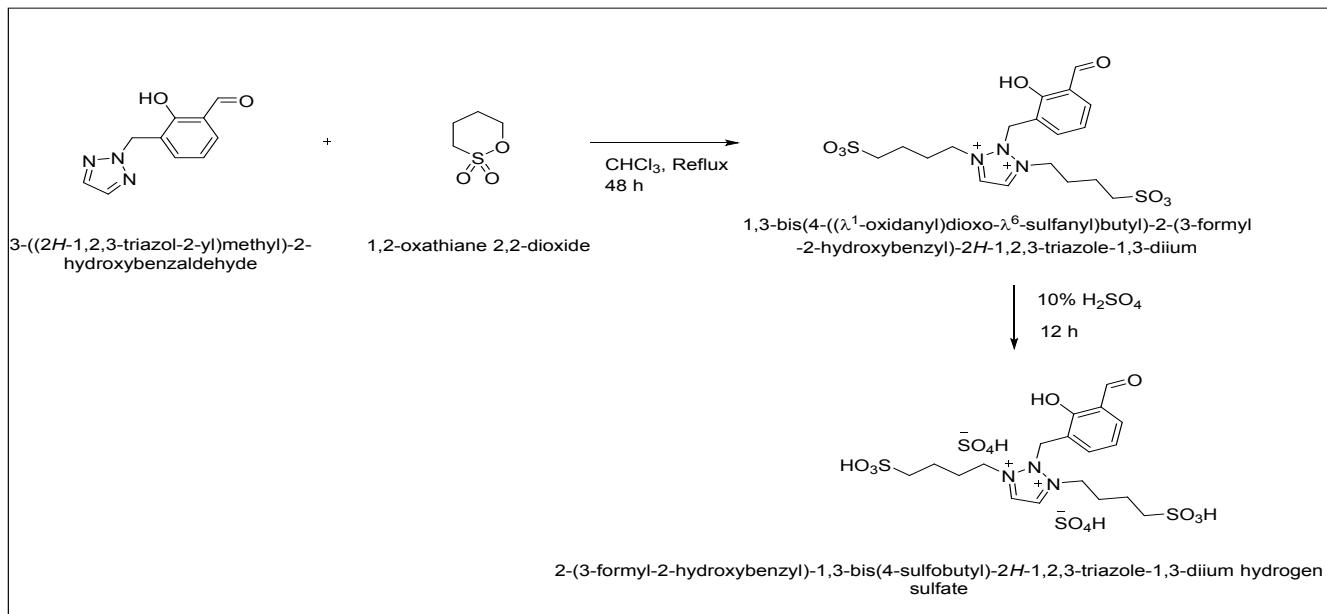
**Fig. S4**  $^1\text{H}$  NMR Spectrum of 3-((2H-1,2,3-triazole-2-yl)methyl)-2-hydroxybenzaldehyde (1b)  
 $^1\text{H}$  NMR (400 MHz, DMSO-d6)  $\delta$ :  $\delta$  10.37 (s, 1H), 10.26 (s, 1H), 8.64 (s, 2H), 7.96 (d, 1H), 7.47 (d, 1H), 6.98 (t, 1H), 5.35 (s, 2H).



**Fig. S5**  $^{13}\text{C}$  NMR Spectrum of of 3-((2H-1,2,3-triazole-2-yl)methyl)-2-hydroxybenzaldehyde (1b)

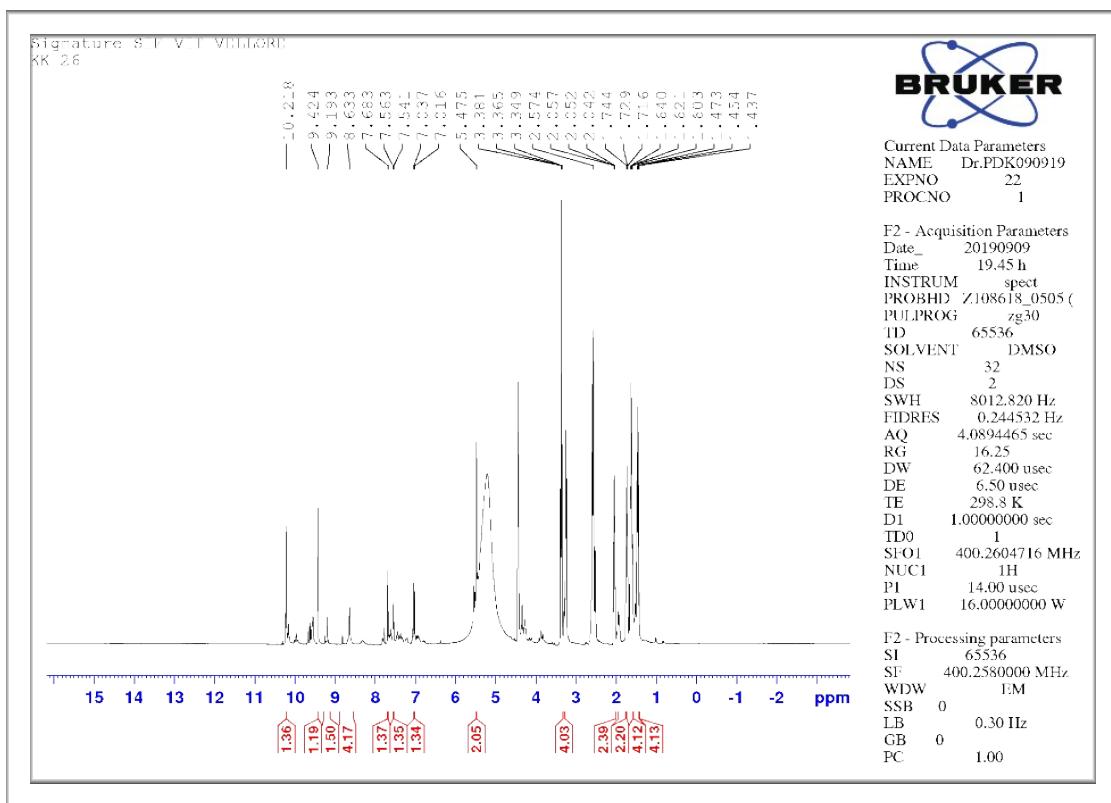
$^{13}\text{C}$  NMR (100 MHz, DMSO-d6)  $\delta$ : 191.20, 161.44, 152.21, 144.48, 136.61, 128.60, 127.43, 122.68, 118.29, 51.71

**Synthesis of 2-(3-formyl-2-hydroxybenzyl)1,3-bis(4-sulfobutyl)-2H-1,2,3-triazole1,3-diium hydrogen sulfate (1c)**



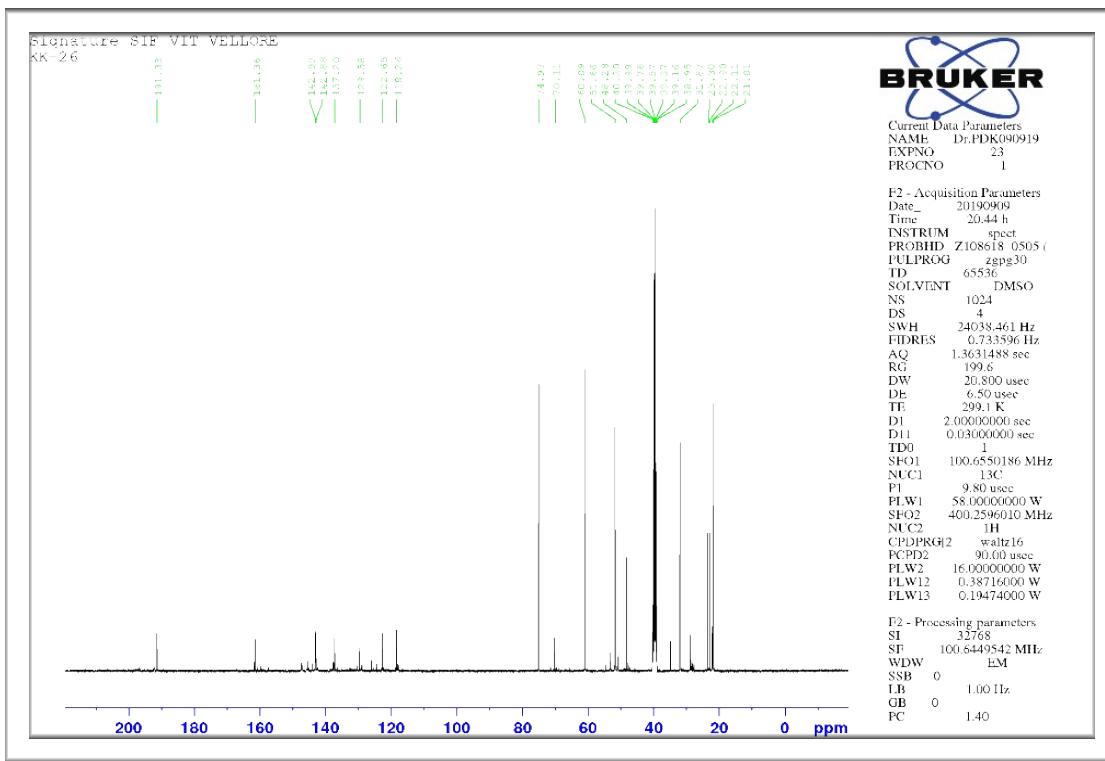
**Scheme. S3** Synthesis of 1c

1, 4-Butylsultone (6.80 g, 50 mmol) was added slowly to a 30 mL solution of triazole salicylaldehyde (5.08 g, 25 mmol) (1b) in acetonitrile into a flask and the mixture was refluxed with constant stirring for 48 h. The solvent was evaporated to dryness by rotavapor and the resulting zwitter ion intermediate was quenched with 10% ( 1ml of H<sub>2</sub>SO<sub>4</sub> in 10 ml of H<sub>2</sub>O) aq. H<sub>2</sub>SO<sub>4</sub> solution.<sup>29</sup> The reaction mixture was refluxed for 12 h, after which time the excess sulfuric acid was evaporated under vacuum. The solid sticky product was washed with diethyl ether (3 x 10 mL), to achieve pure sticky product (1c) confirm by by <sup>1</sup>H NMR and <sup>13</sup>C NMR. Yield=89%, 15 g.



**Fig. S6**  $^1\text{H}$  NMR Spectrum of 2-(3-formyl-2-hydroxybenzyl)1,3-bis(4-sulfobutyl)-2H-1,2,3-triazole1,3-diium hydrogen sulfate (1c)

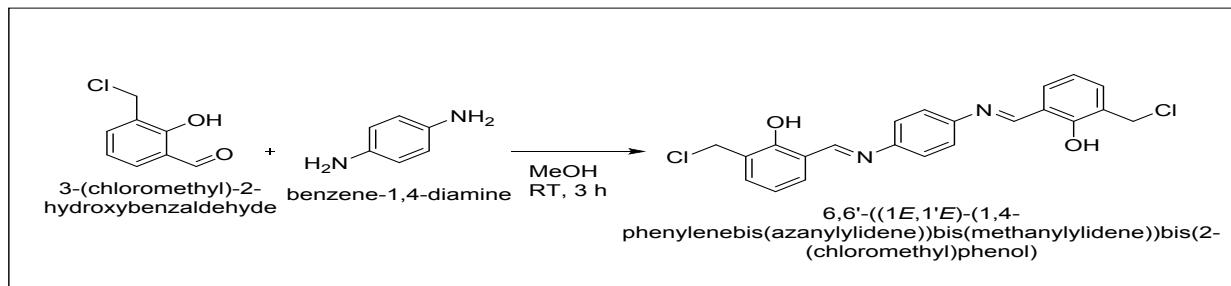
$^1\text{H}$  NMR (400 MHz, DMSO-d6)  $\delta$ :  $\delta$  10.21 (s, 1H), 9.42 (s, 1H), 9.19 (s, 1H), 8.63 (s, 4H), 7.68 (s, 1H), 7.55 (d, 1H), 7.02 (d, 1H), 5.47 (s, 2H), 3.36 (t, 4H), 2.05 (t, 2H), 1.72 (t, 2H), 1.62 (t, 4H), 1.45 (t, 4H).



**Fig. S7**  $^{13}\text{C}$  NMR Spectrum of 2-(3-formyl-2-hydroxybenzyl)1,3-bis(4-sulfobutyl)-2H-1,2,3-triazole1,3-diium hydrogen sulfate (1c)

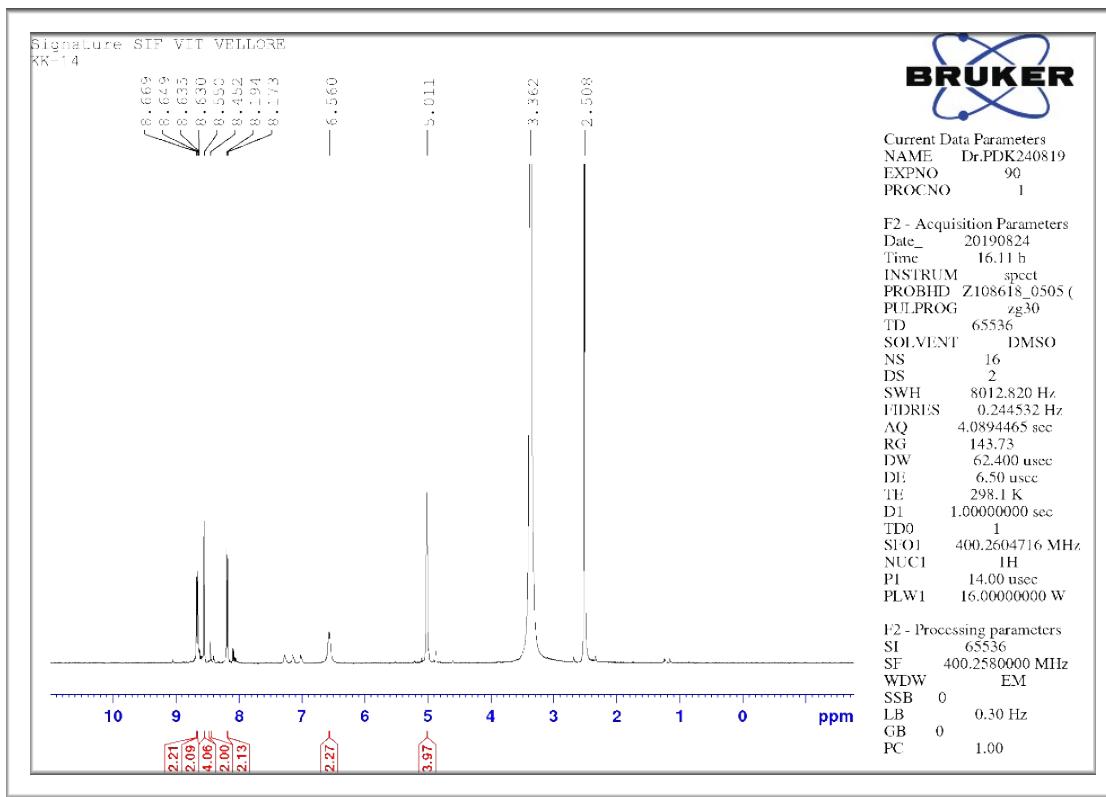
$^{13}\text{C}$  NMR (100 MHz, DMSO-d6)  $\delta$ : 191.32, 161.35, 142.99, 142.87, 137.19, 129.58, 122.64, 118.25, 74.97, 70.10, 60.89, 51.66, 48.27, 31.87, 23.30, 22.10, 21.81.

**Synthesis of phenylene bis (azanylylidene))bis(methanylylidene))bis(2-(chloromethyl)phenol) (2a)**



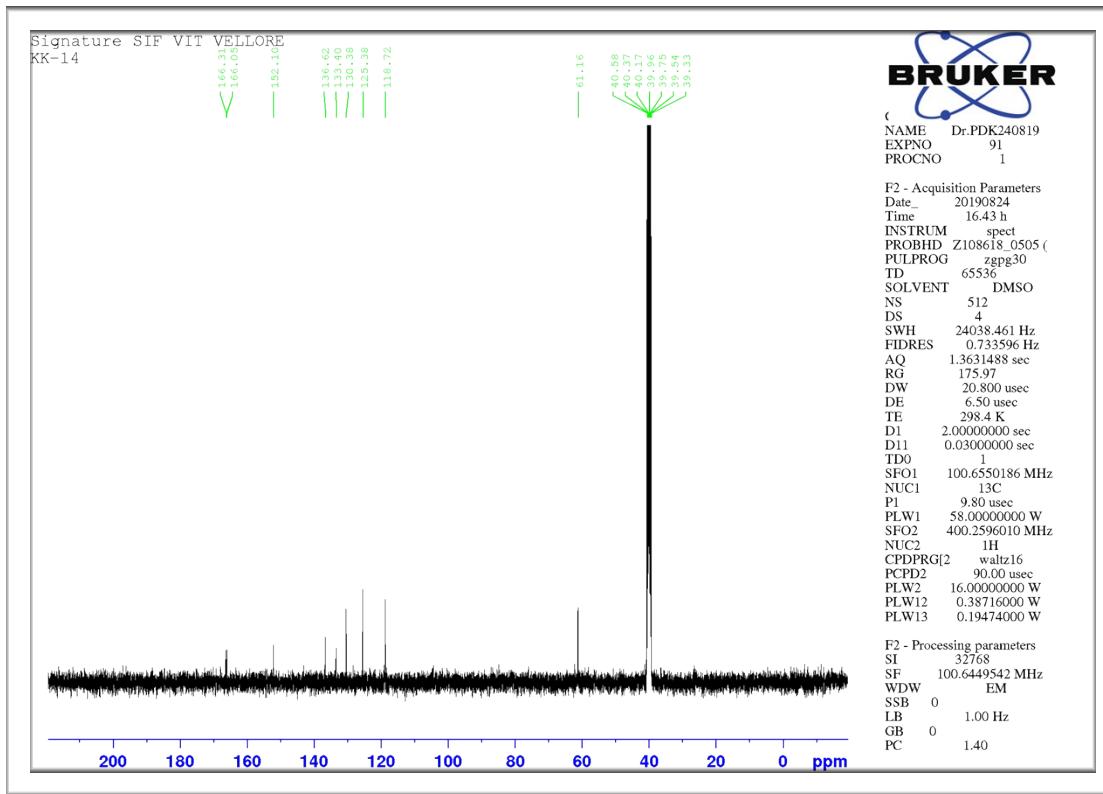
**Scheme. S4** Synthesis of 2a

p-Phenylenediamine (2.16 g, 20 mmol) and chloromethylated salicylaldehyde (1a, 6.80 g, 40 mmol) were weighed and mixed into a 100 mL RBF containing 30 mL methanol and the mixture was refluxed with a magnetic stirring for 3 h.<sup>30</sup> The reaction mixture was cooled to RT affording the product as a red colour solid product in quantitative yield. The resulting product was washed with methanol to acquire red colour solid, obtained by filtering and dried in hot air oven at 80 °C. Yield = 92%, 14.5 g.



**Fig. S8**  $^1\text{H}$  NMR Spectrum of phenylene bis (azanylylidene))bis(methanyllylidene))bis(2-(chloromethyl)phenol) (2a)

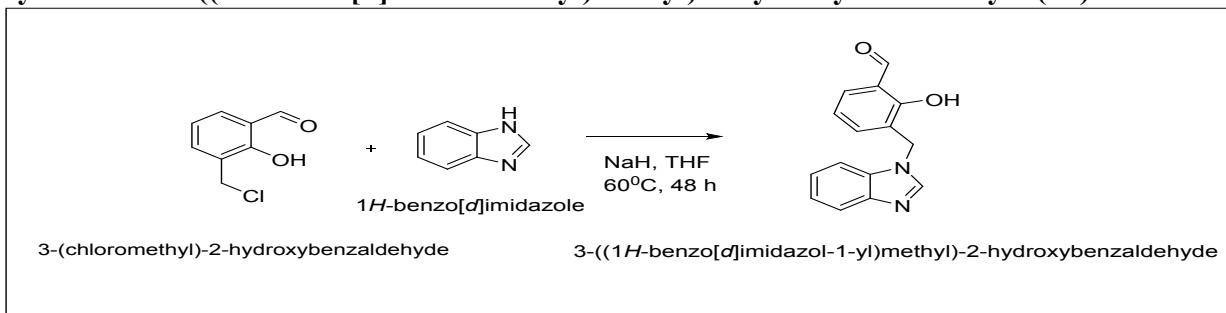
$^1\text{H}$  NMR (400 MHz, DMSO-d6):  $\delta$  8.65 (d, 2H), 8.63 (d, 2H), 8.54 (s, 4H), 8.45 (s, 2H), 8.18 (d, 2H), 6.55 (s, 2H), 5.01 (s, 4H)



**Fig. S9**  $^{13}\text{C}$  NMR Spectrum of phenylene bis (azanylylidene))bis(methanyllylidene))bis(2-(chloromethyl)phenol) (2a)

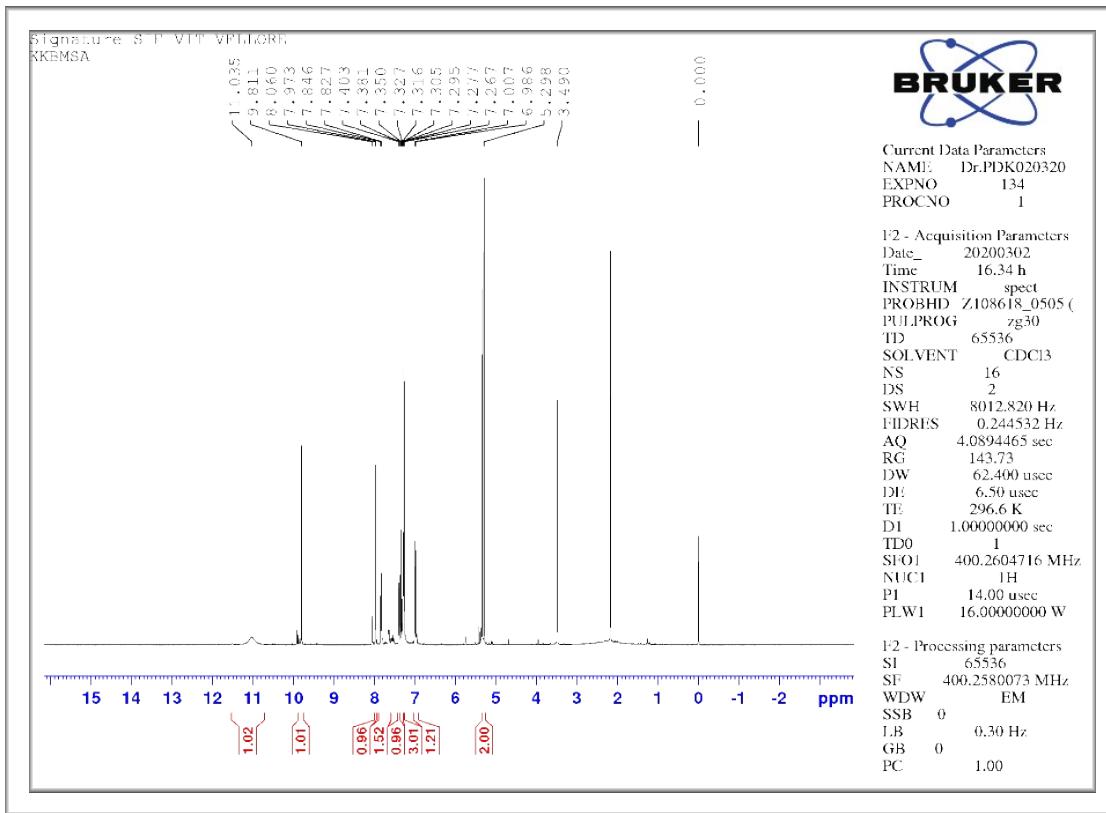
$^{13}\text{C}$  NMR (100 MHz, DMSO-d6)  $\delta$ : 166.048, 152.10, 136.61, 133.40, 130.38, 125.37, 118.71, 61.16

### Synthesis of 3-((1H-benzo[d]imidazole-1-yl)methyl)-2-hydroxybenzaldehyde(2b)



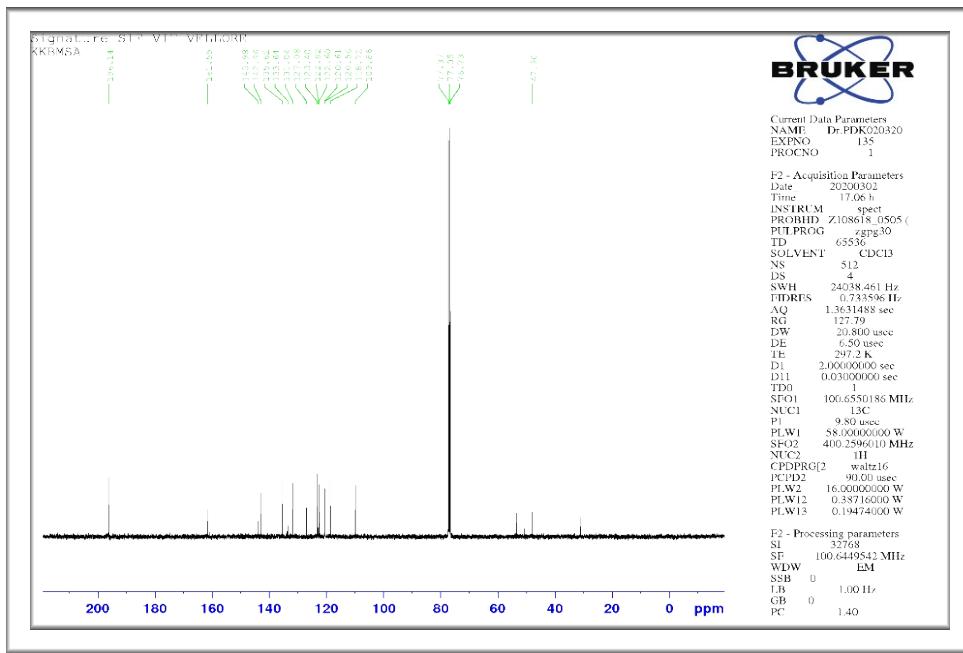
**Scheme. S5** Synthesis of 2b

To a 100 mL RBF sodium hydride was placed and 30 mL n-hexane was added and the reaction mixture was stirred at RT for 40 min., after which time the excess solvent was decanted under vacuum to remove paraffin oil (same ref no as for 1a). The flask containing NaH (0.96g, 40 mmol) was charged with 15 mL tetrahydrofuran (THF), cooled in ice bath to maintain 2 to 3 °C temperature, and solution of benzimidazole (2.36 g, 20 mmol) in THF was added gradually and stirred for 2 h at cold condition. The reaction mixture was charged with chloromethylated salicylaldehyde (1a) (3.41 g, 20 mmol) in THF steadily at the same temperature. The reaction mixture was then refluxed for 48 h (60 °C). After cooling, the salt was discarded by filtration and the solvent was concentrated under vacuum by rotavapour to get sticky yellowish liquid. To the product the mixture of ethyl acetate and saturated brine solution was added to remove remaining salt. The organic layer was separated and dried using anhydrous sodium sulphate for 12 h. The ethyl acetate layer was evaporated using rotavapour to get sticky yellowish solid.<sup>31</sup> The resulting solid was washed by diethyl ether (3 X 10 mL) and dried in hot air oven to obtain yellowish fine powder 2b. Yield- 82 %, 8.2 gram.



**Fig. S10**  $^1\text{H}$  NMR Spectrum of Synthesis of 3-((1H-benzo[d]imidazole-1-yl)methyl)-2-hydroxybenzaldehyde(2b)

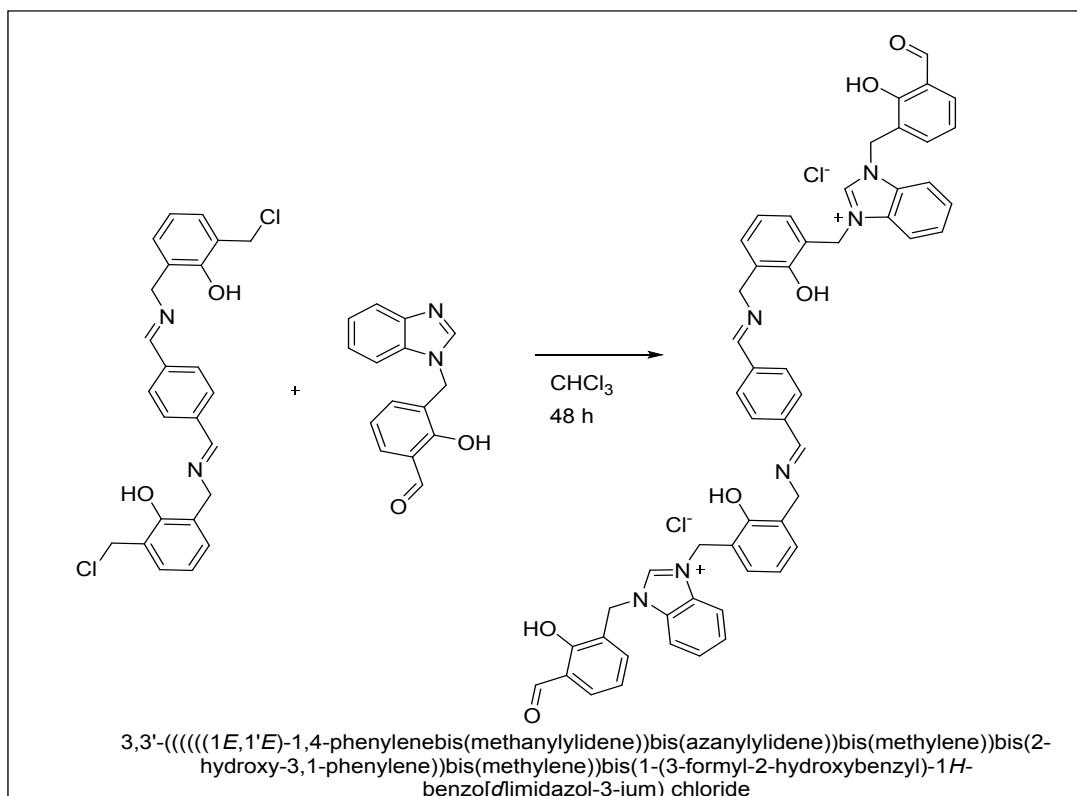
$^1\text{H}$  NMR,  $^1\text{H}$  NMR (400MHz,  $\text{CdCl}_3\text{-d}6$ ) :  $\delta$  11.03 (s, 1H), 9.81 (s, 1H), 7.97 (s, 1H), 7.83 (d 2H), 7.39 (d, 1H), 7.26-7.27 (m, 3H), 6.99 (d, 1H), 5.20 (s, 2H).



**Fig. S11**  $^{13}\text{C}$  NMR Spectrum of Synthesis of 3-((1H-benzo[d]imidazole-1-yl)methyl)-2-hydroxybenzaldehyde(2b)

$^{13}\text{C}$  NMR (100MHz,  $\text{CDCl}_3$ ):  $\delta$  196.13, 161.54, 143.97, 142.95, 135.62, 133.63, 131.86, 127.07, 123.40, 122.60, 120.60, 120.55, 118.72, 109.86, 47.89

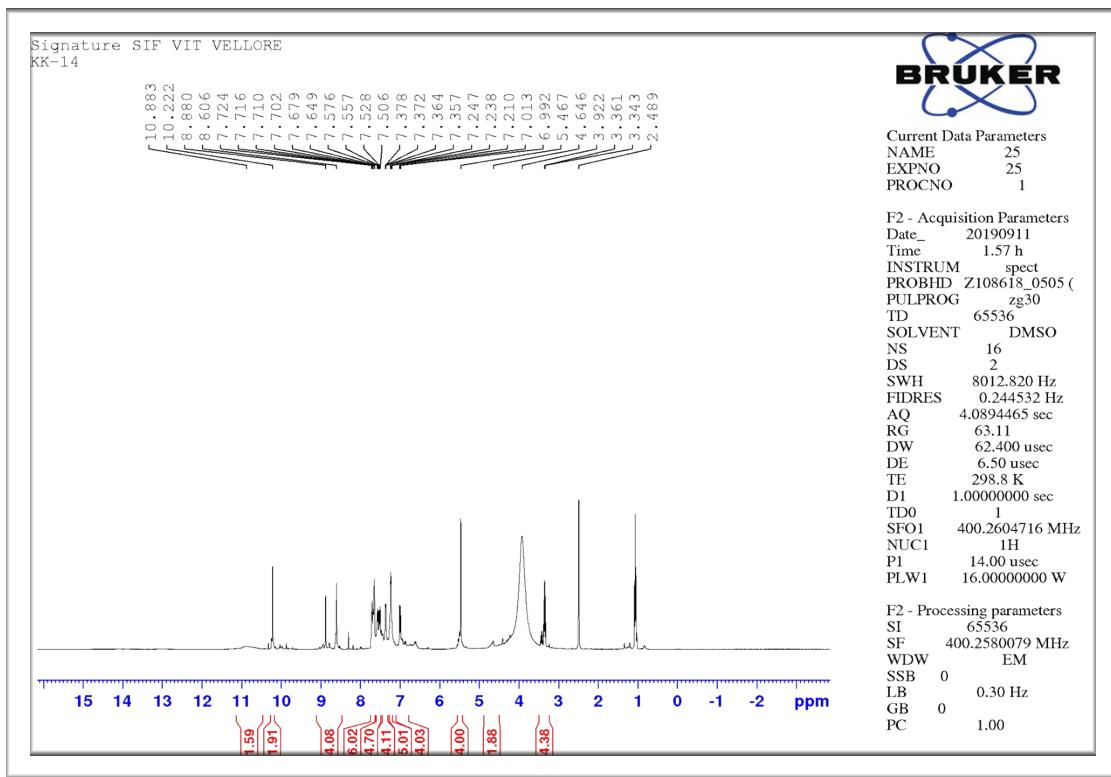
**Synthesis of 1,1'-((((1E, 1'E)-1,4-phenylenebis(methanylylidene))bis(methylene))bis(2-hydroxy-3,1-phenylene))bis(methylene))bis(3-(3-formyl-2-hydroxybenzyl)-1H-benzo[d]imidazole-3-iun) chloride (2c)**



**Scheme. S6** Synthesis of 2c

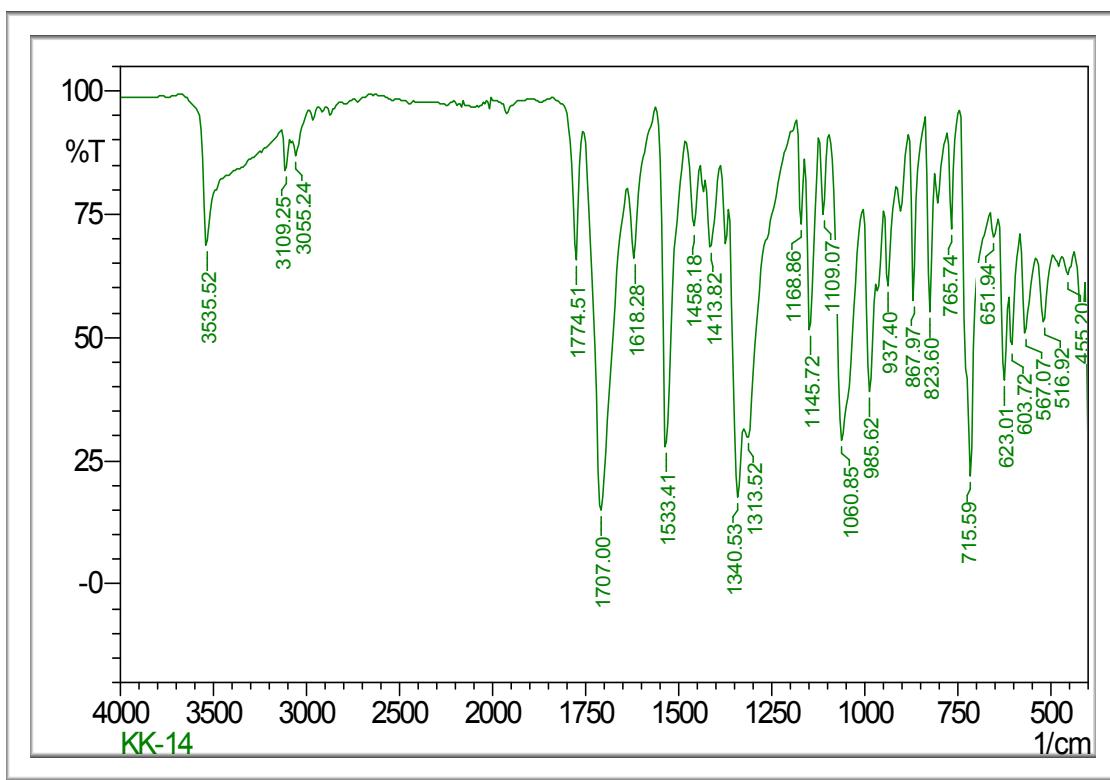
In a 50 mL RBF, the mixture of phenylene bis (azanylylidene))bis(methanylylidene))bis(2-(chloromethyl)phenol) 2a (0.412 g 1 mmol) and benzimidazole salicylaldehyde 2b (0.504 g, 2 mmol) was suspended in 30 mL chloroform. The mixture was set aside on constant stirring for 48 h.<sup>32</sup> After the completion of reaction, the precipitated NaCl salt was separated by filtration and rinsed with chloroform twice. The resulting filtrate was concentrated under vacuum to acquire precipitate, dried in hot air oven to give 2c.Yield: 74 %, 14.2 g.

In a 50 mL RBF, the mixture of phenylene bis (azanylylidene))bis(methanylylidene))bis(2-(chloromethyl)phenol) 2a (0.412 g 1 mmol) and benzimidazole salicylaldehyde 2b (0.504 g, 2 mmol) was suspended in 30 mL chloroform. The mixture was set aside on constant stirring for 48 h.<sup>32</sup> After the completion of reaction, the precipitated NaCl salt was separated by filtration and rinsed with chloroform twice. The resulting filtrate was concentrated under vacuum to acquire precipitate, dried in hot air oven to give 2c.Yield: 74 %, 14.2 g.



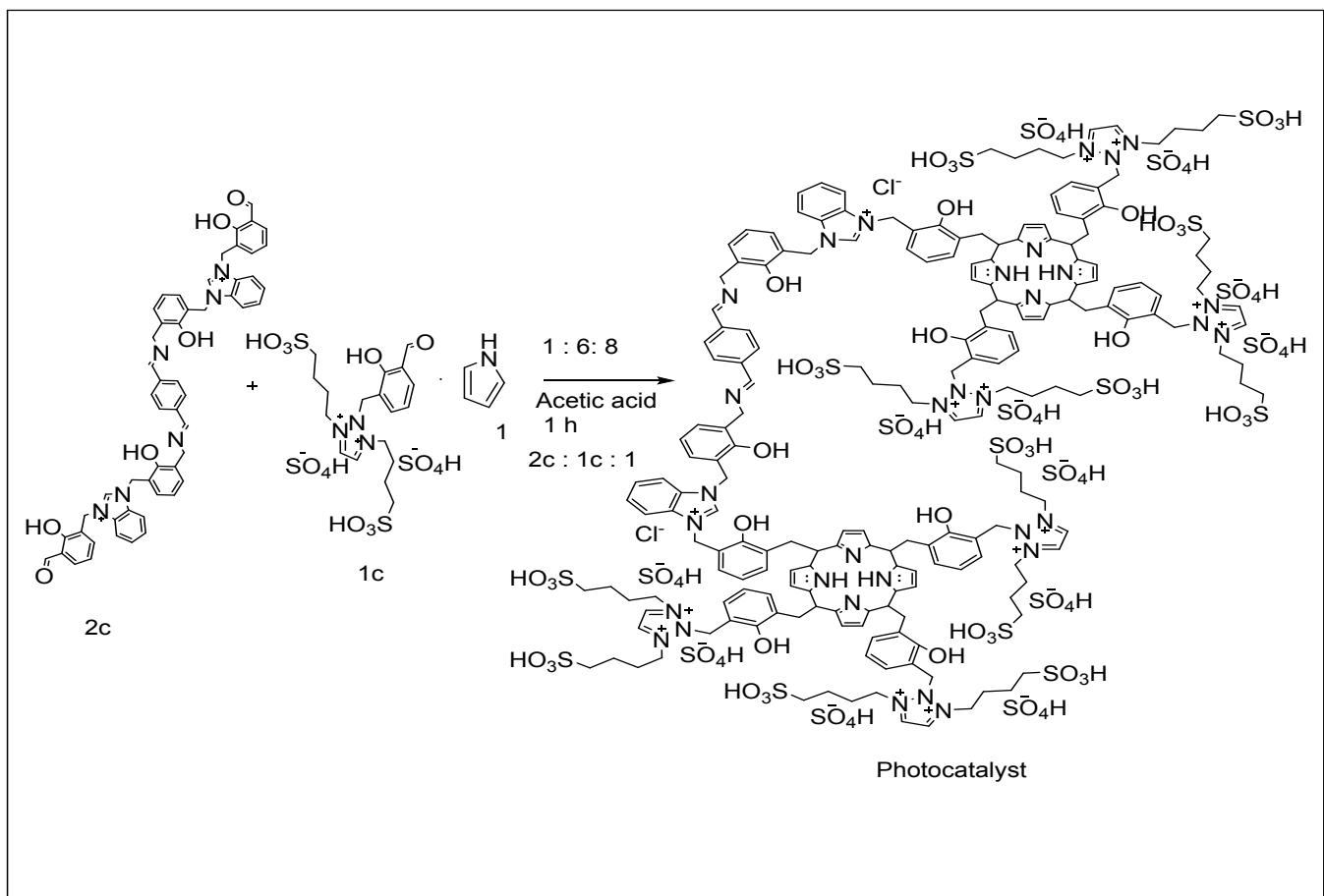
**Fig. S12**  $^1\text{H}$  NMR Spectrum of 1,1'-(((((1E, 1'E)-1,4-phenylenebis(methanlylidene))bis(methylene))bis(2-hydroxy-3,1-phenylene))bis(methylene))bis(3-(3-formyl-2-hydroxybenzyl)-1H-benzo[d]imidazole-3-ium) chloride (2c)

$^1\text{H}$  NMR (400 MHz, DMSO-d6):  $\delta$  10.88 (s, 2H), 10.22 (s, 2H), 8.60 (s, 2H), 8.88 (s, 2H), 7.55-7.57 (m, 6H), 7.51 (d, 5H), 0.00-7.01 (m, 4H), 6.99 (s, 5H), 5.46 (s, 4H), 4.64 (s, 4H), 3.92 (s, 2H), 3.36 (s, 4H)



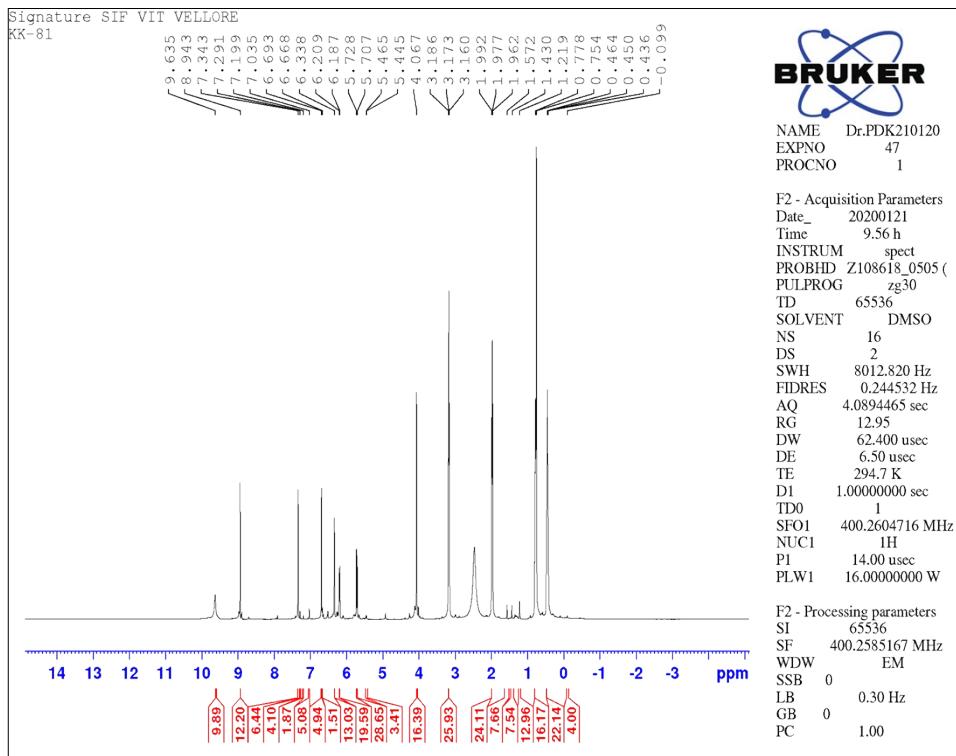
**Fig. S13** FT-IR: (KBr,  $\nu/\text{cm}^{-1}$ ): 3535.52, 3109.25, 3055.24, 1774.51, 1707, 1618.28, 1533.41, 1458.18, 1413.82, 1340.53, 1313.52, 1168.86, 1145.72, 1109.07, 1060.85, 985.62, 937.40, 867.97, 823.60, 765.74, 715.59, 661.94, 623.01, 603.72, 567.07, 516.92, 465.20

**Synthesis of 2-((9E,16E)-15,20-bis(3-((1,3-bis(4-sulfobutyl)-2H-1,2,3-triazole-1,3-diium-2-yl)methyl)-2-hydroxybenzyl)-10-(2-hydroxy-3-((1-(2-hydroxy-3-((2-hydroxy-3-((9E,19E)-10, 15, 20-tris(3-((1,3-bis(4-sulfobutyl)-2H-1,2,3-triazole-1,3-diium-2-yl)methyl)-2-hydroxybenzyl)-1H,4H,5H,6H,10H,11H,14H,15H,16H,20H-porphyrin-5-yl)methyl)benzyl)-1H-benzo[d]imidazole-3-iium-1-yl)methyl)benzyl)imino)benzylidene)amino)methylbenzyl)-1H-benzo[d]imidazole-3-iium-3-yl)methyl)benzyl)-1H,4H,5H,6H,10H,11H,14H,15H,19H,20H-porphyrin-5-yl)methyl)-2-hydroxybenzyl)-1,3-bis(4-sulfobutyl)-2H-1,2,3-triazole-1,3-diium dichloride dodecakis(hydrogen sulphate)**



**Scheme. S7.** Synthesis of Photocatalyst (SAFPTILM)

In a 100 mL RBF, the precursors (1c) (4.03 g, 6 mmol) and (2c) (0.875 g, 1 mmol) were weighed, mixed with 25 mL acetic acid and stirred for 10 min. To this, pyrrole (0.536 g, 8 mmol) was added and the reaction mixture continued for stirring at room temperature for 1 h. The desired dark brown solid SAFPTILM was separated by vacuum filtration, rinsed with water (5 x 15 mL) to remove traces of acetic acid. Finally the product was frequently washed with acetone and dried in oven at 80°C to furnish the SAFPTILM photocatalyst.<sup>33</sup> Yield: 90 %, 4.52 g.



**Fig. S14**  $^1\text{H}$  NMR Spectrum of Synthesis of Synthesis of 2-(3-((15,20-bis(3-((1,3-bis(4-sulfobutyl)-2H-1,2,3-triazole-1,3-dium-2-yl)methyl)-2-hydroxybenzyl)-10-(2-hydroxy-3-((1-(2-hydroxy-3-(((E)-4-(E)-((2-hydroxy-3-((3-(2-hydroxy-3-((10,15,20-tris(3-((1,3-bis(4-sulfobutyl)-2H-1,2,3-triazole-1,3-dium-2-yl)methyl)-2-hydroxybenzyl)-yl)methyl)benzyl)-1H-benzo[d]imidazole-3-ium-1-yl)methyl)benzyl)imino)benzylidene)amino)methyl)benzyl)-1H-benzo[d]imidazole-3-ium-1-yl)methyl)benzyl)imino)benzylidene)amino)methyl)benzyl)-1H-benzo[d]imidazol-3-ium-3-yl)methyl)benzyl)Porphyrin-5-yl)methyl)-2-hydroxybenzyl)-1,3-bis(4-sulfobutyl)-2H-1,2,3-triazole-1,3-dium dichloride dodecakis (hydrogen sulphate)

$^1\text{H}$  NMR 400 MHz, DMSO-d6:  $\delta$  9.63 (s, 10H), 8.94 (s, 12H), 7.34 (s, 6H), 7.29 (s, 4H), 7.19 (s, 2H), 7.03 (s, 5H), 6.69 (s, 5H), 6.66 (s, 2H), 6.33 (s, 13H), 6.19 (d, 20H), 5.71 (d, 29H), 5.45 (d, 3H), 4.06 (s, 16H), 3.17 (t, 26H), 1.97 (t, 24H), 1.57 (s, 8H), 1.43 (s, 8H), 1.21 (s, 13H), 0.76 (d, 16H), 0.45 (t, 22H), -0.0989 (s, 4H).  $^{13}\text{C}$  NMR (100MHz, DMSO)  $\delta$  189.80, 158.99, 150.10, 142.38, 134.64, 127.17, 125.65, 120.50, 116.44, 116.08, 72.89, 49.76, 46.26, 21.33, 20.90, -0.50.

**Table S1. Optimization of Heck Coupling between 4-chloroaniline and acrylonitrile under Photocatalytic Conditions Catalysed by Photo-catalyst**

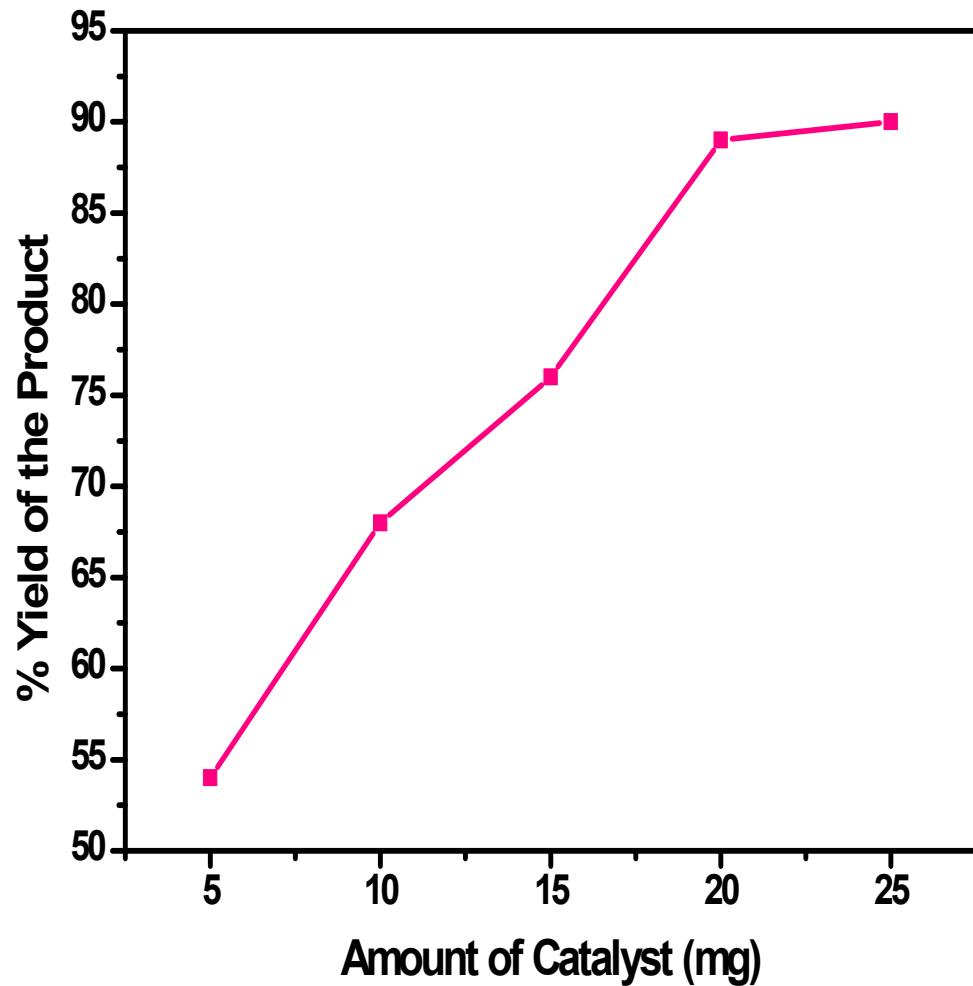
| Entry | Reagent | Catalyst (mg) | Time (h) | Yield (%)       |
|-------|---------|---------------|----------|-----------------|
| 1     | R1 +R2  | 20            | 24       | 89              |
| 2     | R1 +R2  | 5             | 24       | 54              |
| 3     | R1 +R2  | 10            | 24       | 68              |
| 4     | R1 +R2  | 15            | 24       | 76              |
| 5     | R1 +R2  | 20            | 24       | 89              |
| 6     | R1 +R2  | 25            | 24       | 90              |
| 7     | R1 +R2  | 20            | 6        | 61              |
| 8     | R1 +R2  | 20            | 12       | 70              |
| 9     | R1 +R2  | 20            | 18       | 88              |
| 10    | R1 +R2  | 20            | 24       | 89              |
| 11    | R1 +R2  | 20            | 15       | 87 <sup>a</sup> |
| 12    | R1 +R2  | 20            | 13       | 90 <sup>b</sup> |
| 13    | R1 +R2  | 20            | 18       | 90 <sup>c</sup> |
| 14    | R1 +R2  | 20            | 18       | 92 <sup>d</sup> |
| 15    | R1 +R2  | 20            | 18       | 5 <sup>e</sup>  |

**Reaction conditions:** 4-chloroaniline (30 mmol, 3.82 gm), acrylonitrile (32 mmol, 1.59 gm), photocatalysts SAFTILM (20 mg), 5W yellow light, 18 h

a=9W light ; b= 12 W light ; c= 5W Green light;

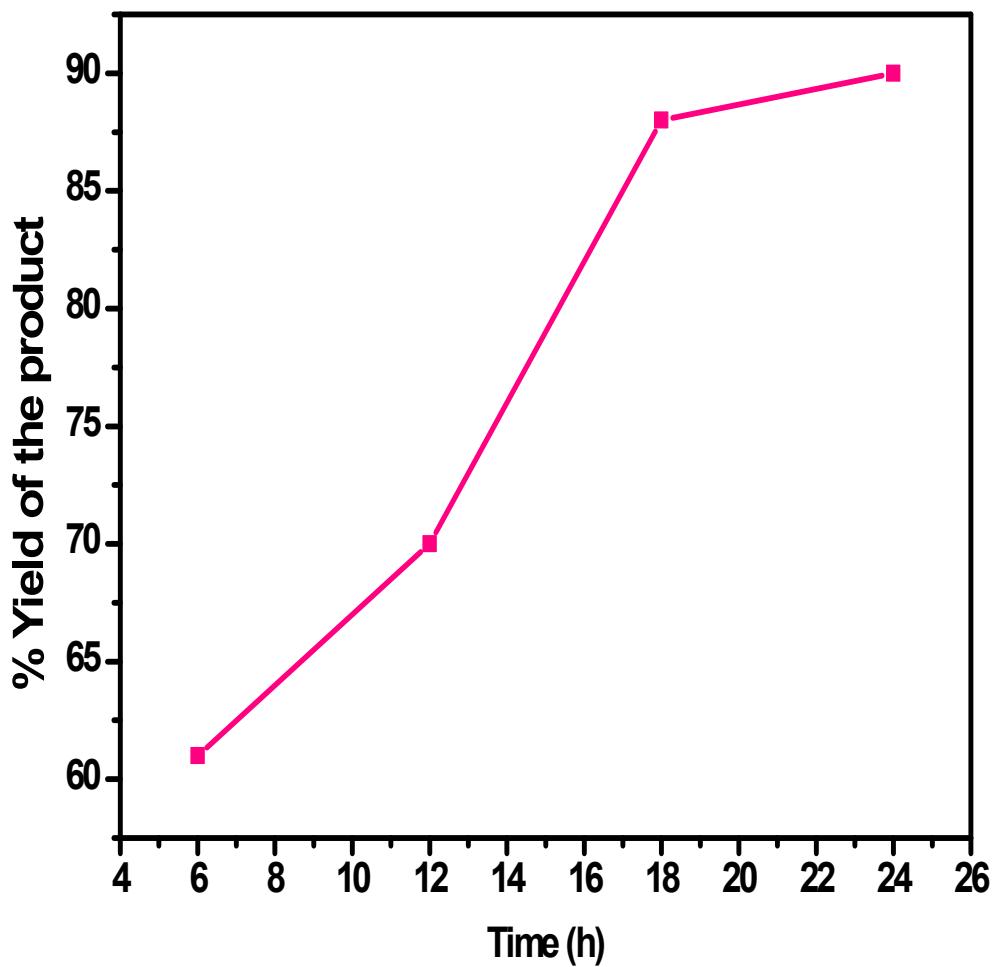
d= 5W yellow light; e= reaction performed in dark

### Effect of photo-catalyst dose



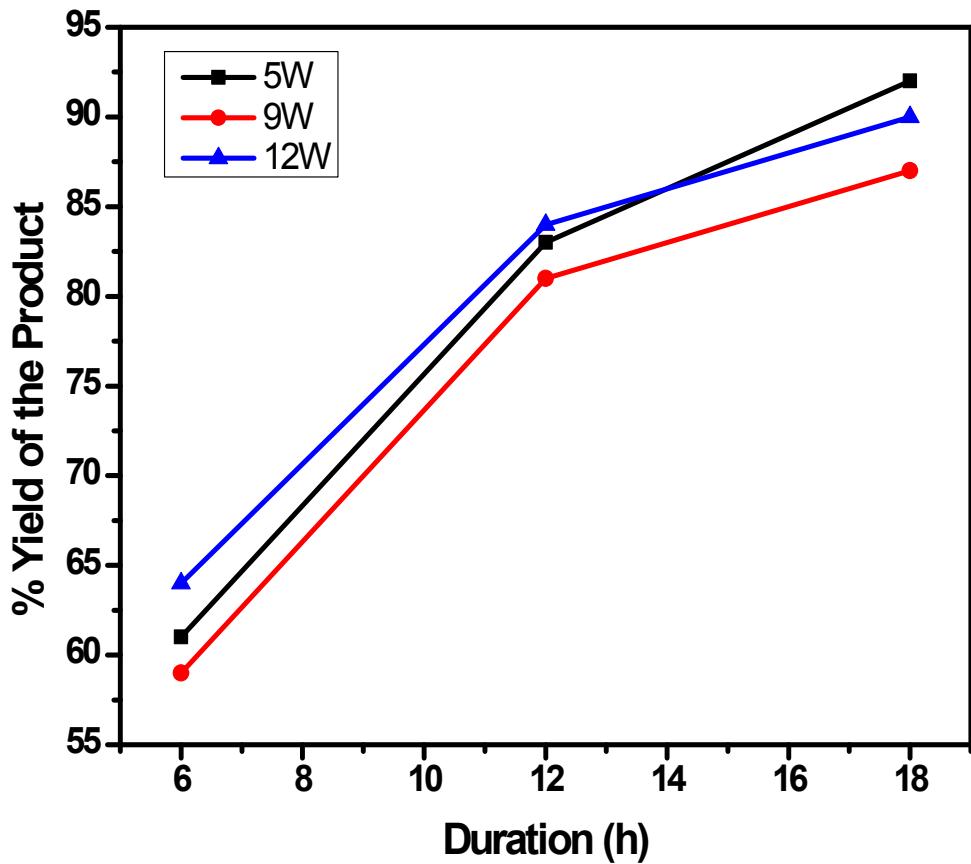
**Fig. S15** Photo-catalyst dose

### Influence of illumination time



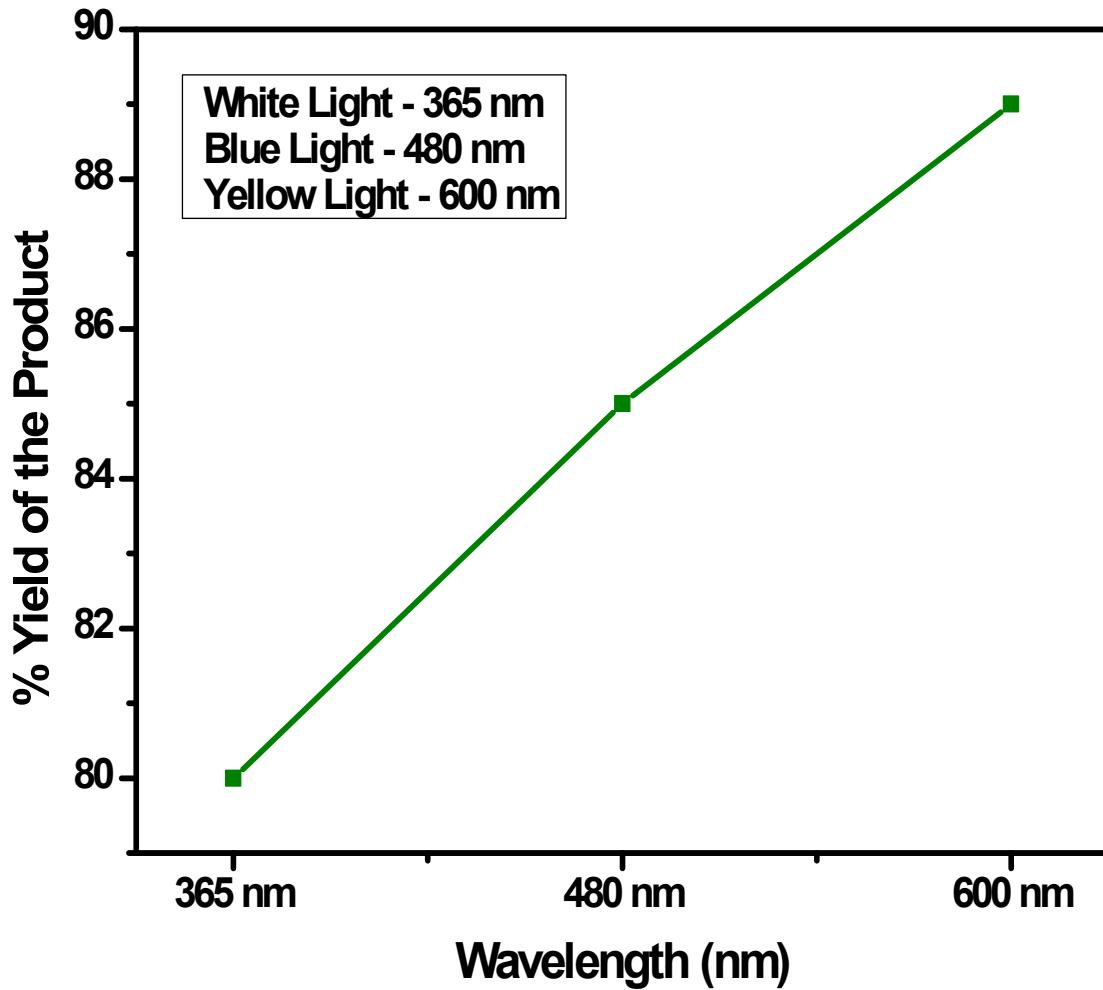
**Fig. S16** Influence of illumination time

### Effect of light intensity



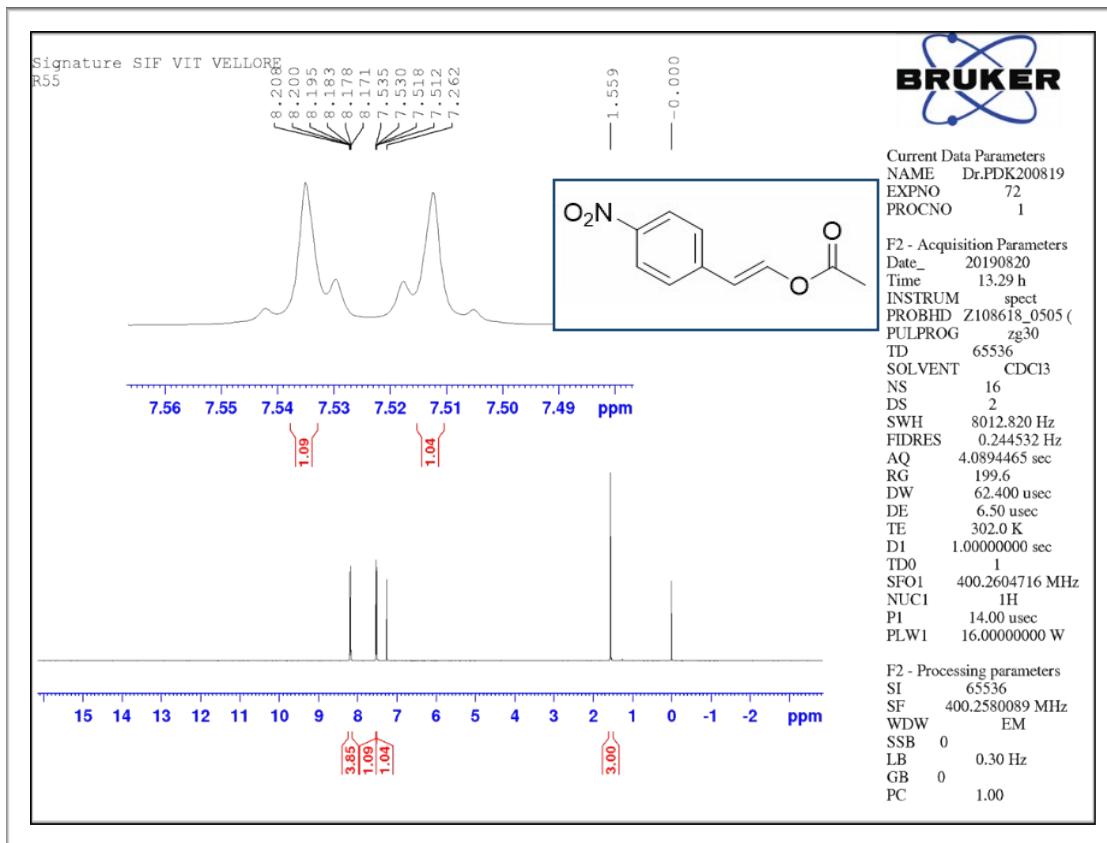
**Fig. S17** Effect of light intensity

The effect of wavelength of the irradiation on the C-C coupling



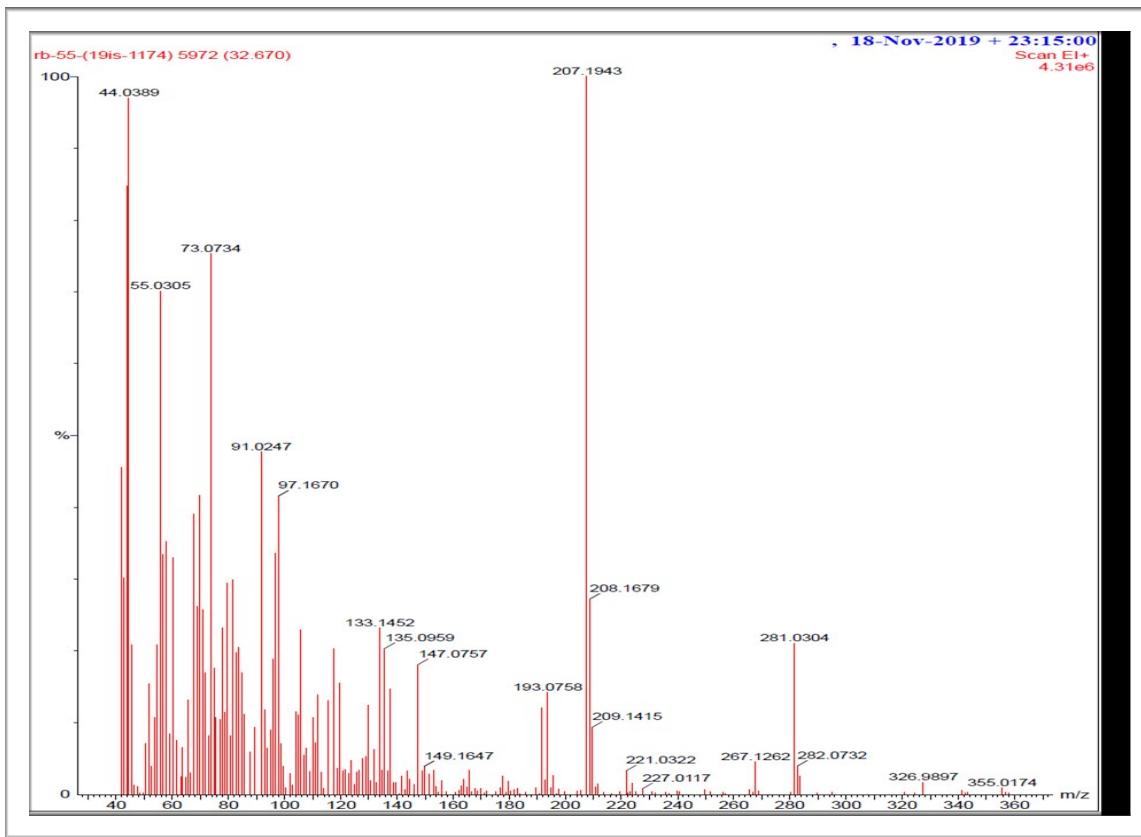
**Fig. S18** The effect of wavelength of the irradiation on the C-C Coupling formation

## Heck coupling derivatives

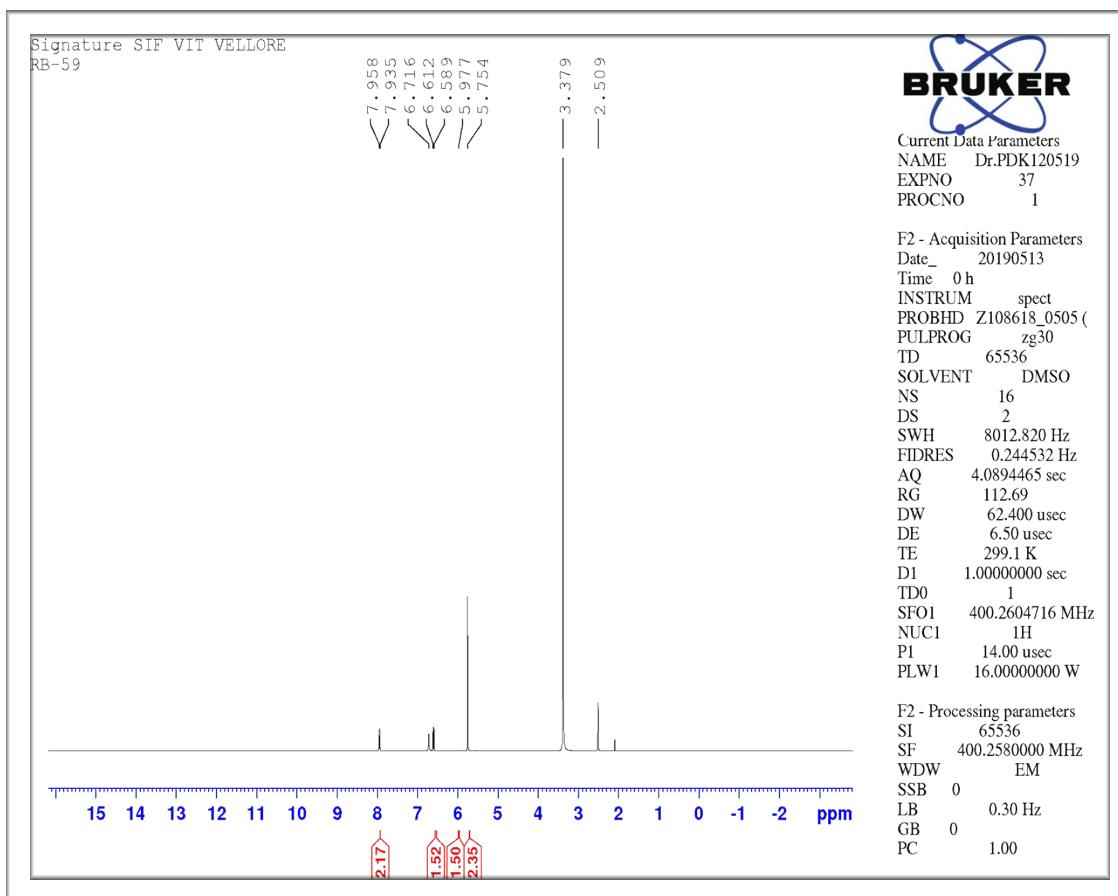


**Fig. S19** <sup>1</sup>H NMR Spectrum of (Z)-4-nitrostyryl acetate

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ : δ 8.20 (d, 2H), 8.18 (d, 2H), 7.53 (d, 1H), 7.51 (d, 1H), 1.55 (s, 3H)



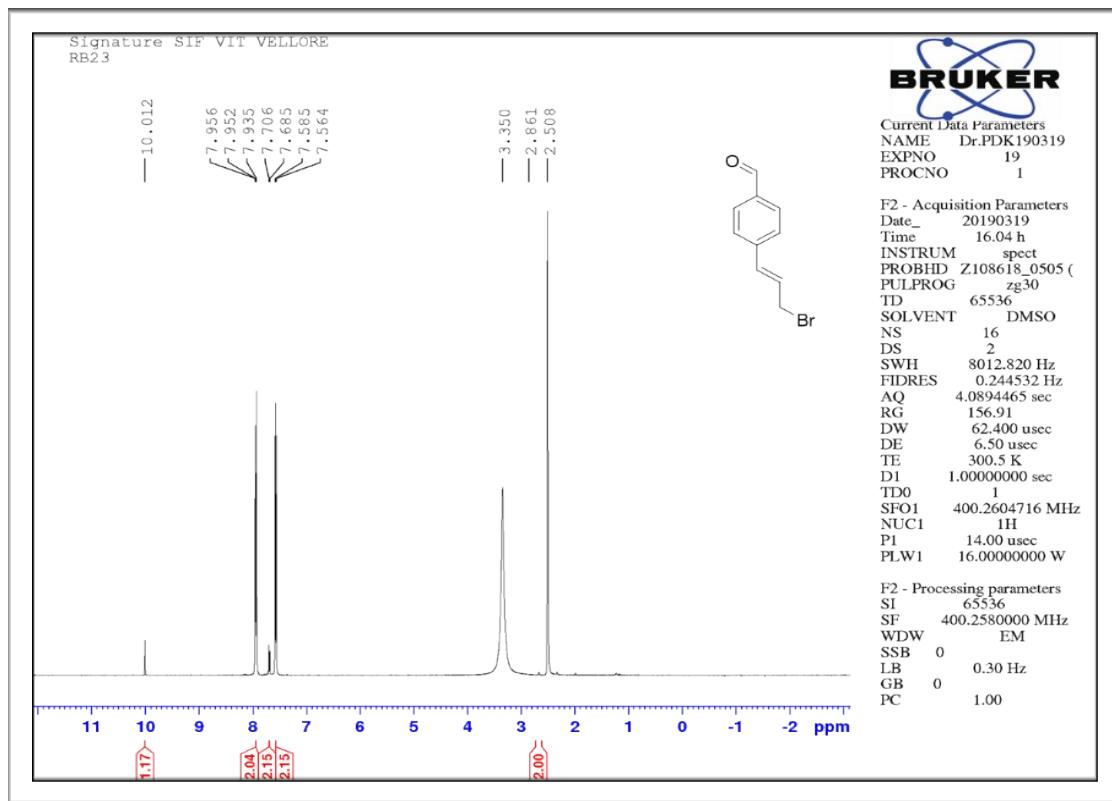
**Fig. S20** GC-MS Spectrum of (Z)-4-nitrostyryl acetate GC-MS: 207.19



**Fig. S21**  $^1\text{H}$  NMR Spectrum of (E)-3-(4-aminophenyl) acrylonitrile

$^1\text{H}$  NMR (400 MHz, DMSO-d6):  $\delta$  7.94 (d, 2H), 6.60 (d, 2H), 5.97 (s, 2H), 5.75 (s, 2H), .

**GC-MS: 144.18**

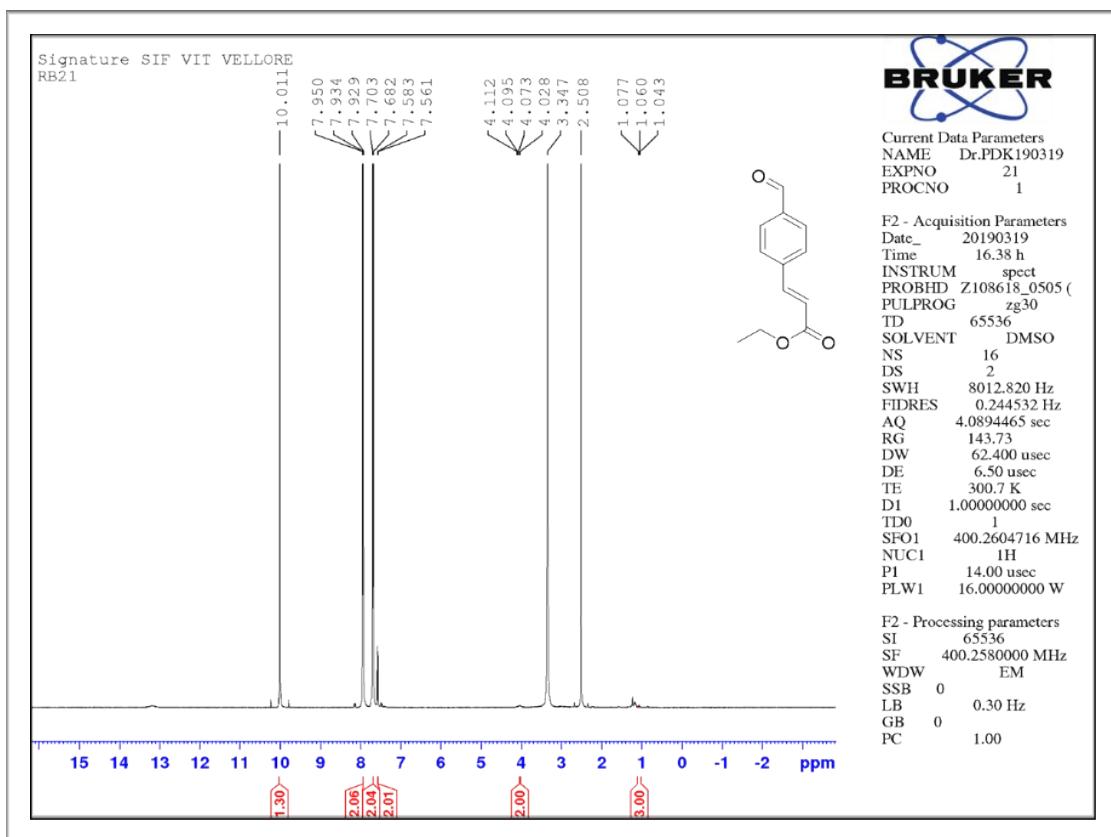


**Fig. S22**  $^1\text{H}$  NMR Spectrum of (E)-4-(3-bromoprop-1-en-1yl)benzaldehyde

$^1\text{H}$  NMR (400 MHz, DMSO-d6):  $\delta$  10.01 (s, 1H), 7.95 (t, 2H), 7.70 (d, 2H), 7.57 (d, 2H), 2.86 (s, 2H).

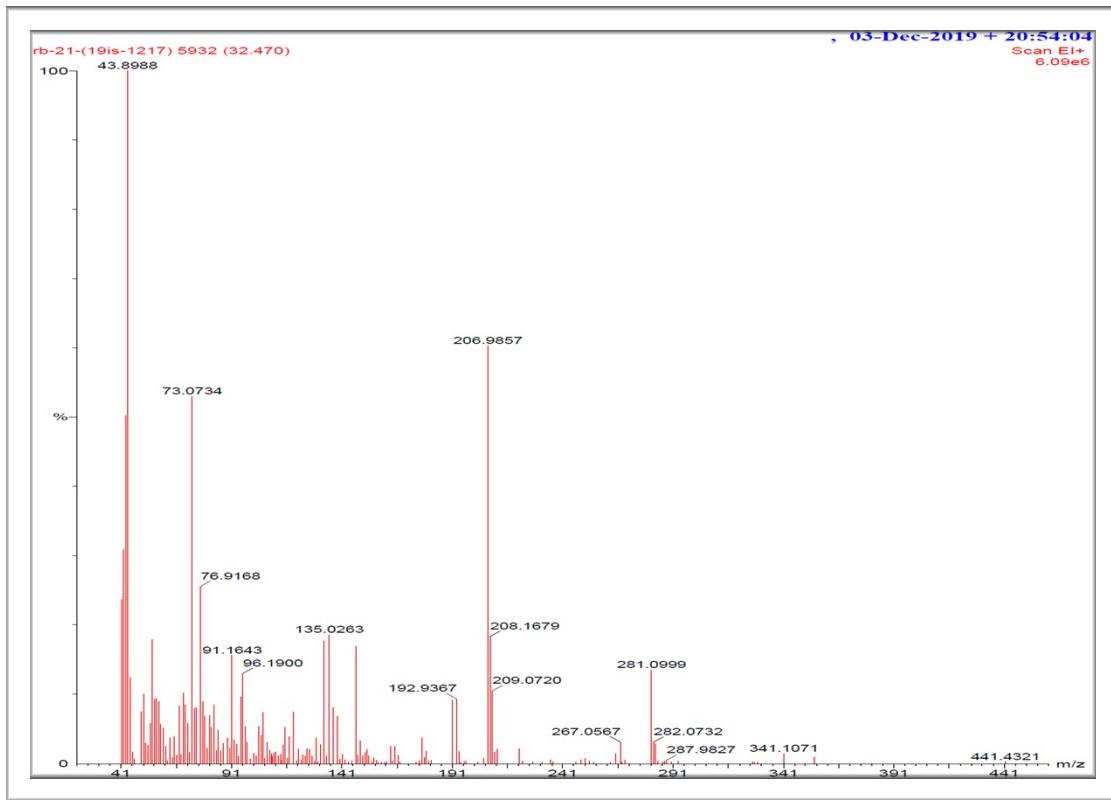
$^{13}\text{C}$  NMR (100 MHz, DMSO-d6):  $\delta$  166.92, 138.25, 131.65, 131.61, 130.12, 129.85, 129.21, 40.2

**GC-MS: 225.09**

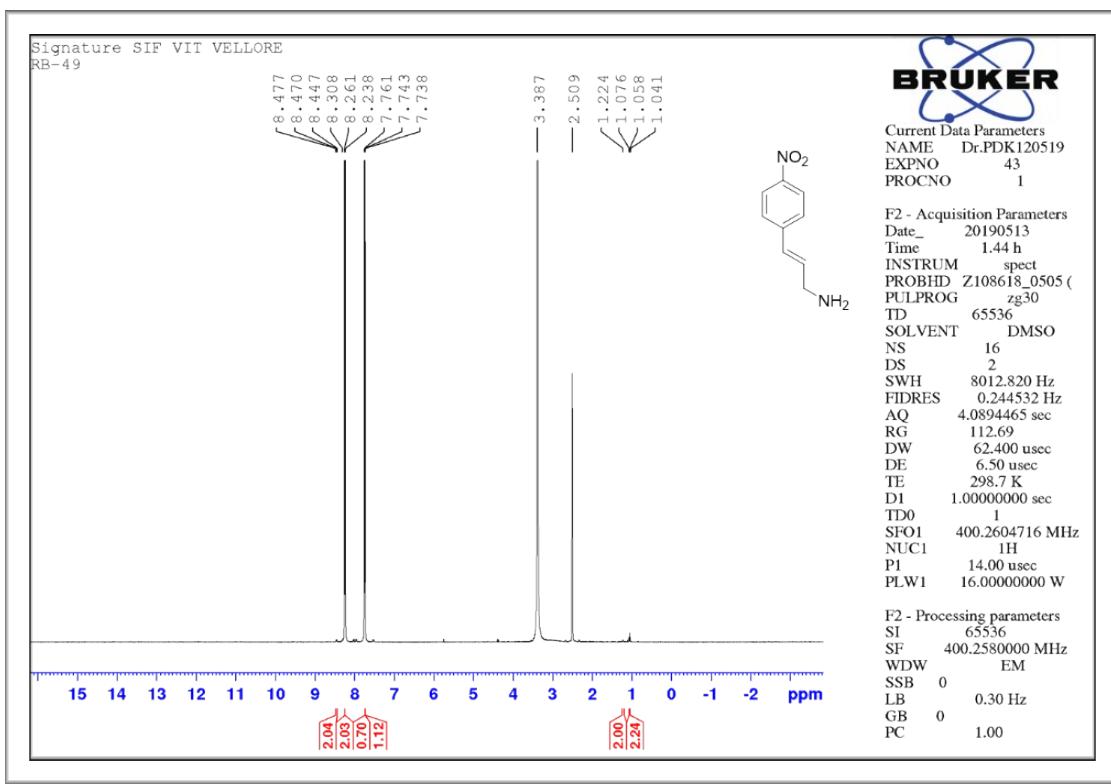


**Fig. S23**  $^1\text{H}$  NMR Spectrum of ethyl (E)-3-(4-formylphenyl)acrylate

$^1\text{H}$  NMR (400 MHz, DMSO-d<sub>6</sub>):  $\delta$  10.01 (s, 1H), 7.93 (t, 2H), 7.69 (d, 2H), 7.57 (d, 2H), 4.07 (q, 2H), 36.36 (t, 3H).

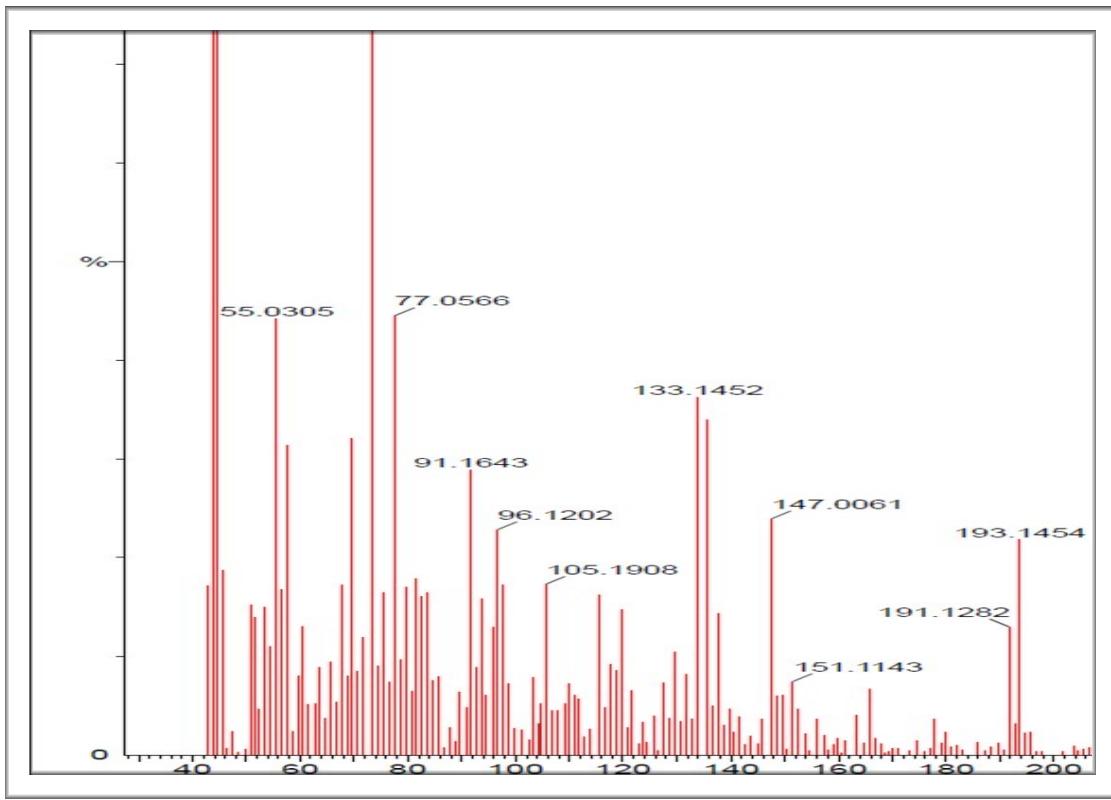


**Fig. S24** GC-MS Spectrum of ethyl (E)-3-(4-formylphenyl) acrylate GC-MS: 204.23

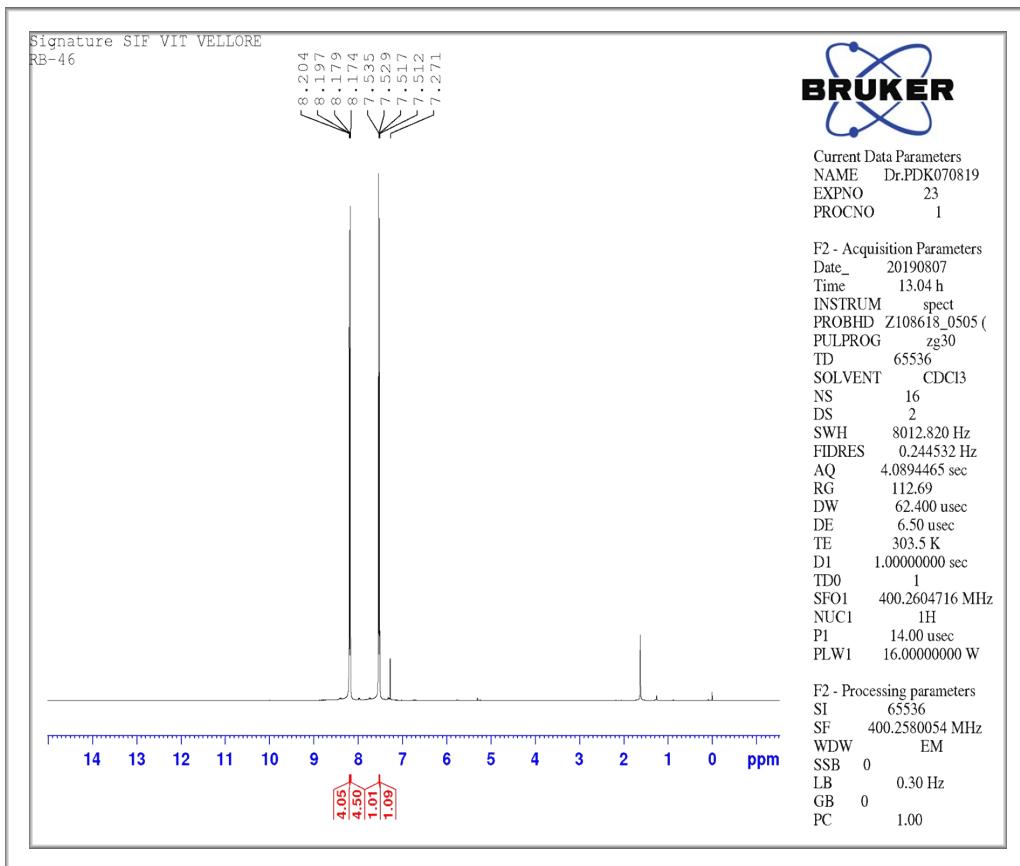


**Fig. S25**  $^1\text{H}$  NMR Spectrum of (E)-3-(4-nitrophenyl)prop-2-en-1-amine

$^1\text{H}$  NMR (400 MHz, DMSO-d<sub>6</sub>):  $\delta$  8.47 (d, 2H), 8.25 (d, 2H), 7.76 (s, 1H), 7.74 (s, 1H), 4.38 (t, 2H), 1.22 (s, 2H)



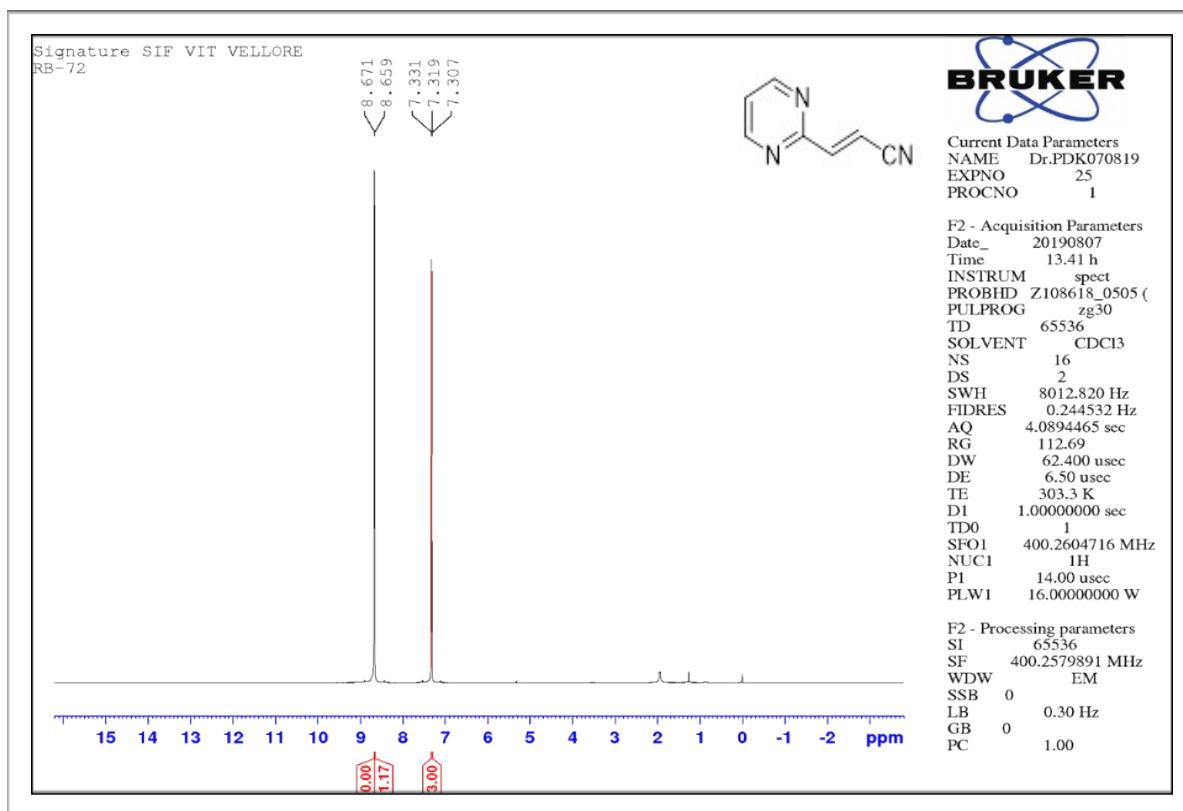
**Fig. S26** GC-MS Spectrum of (E)-3-(4-nitrophenyl)prop-2-en-1-amine GC-MS: 178.19



**Fig. S27**  $^1\text{H}$  NMR Spectrum of (E)-1-nitro-4-styrylbenzene

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ : δ 8.19 (d, 4H), 8.18 (d, 4H), 7.53 (d, 1H), 7.51 (d, 1H).

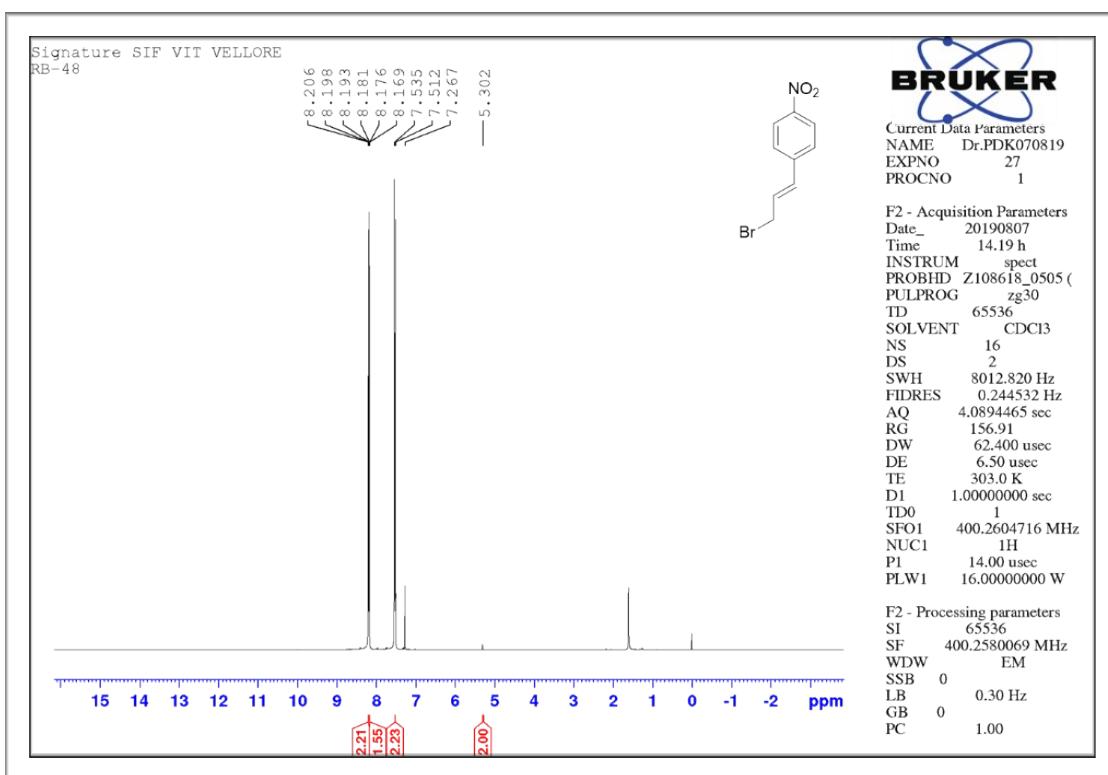
GC-MS: 225.25



**Fig. S28** <sup>1</sup>H NMR Spectrum of (E)-3-(pyrimidin-2-yl)acrylonitrile

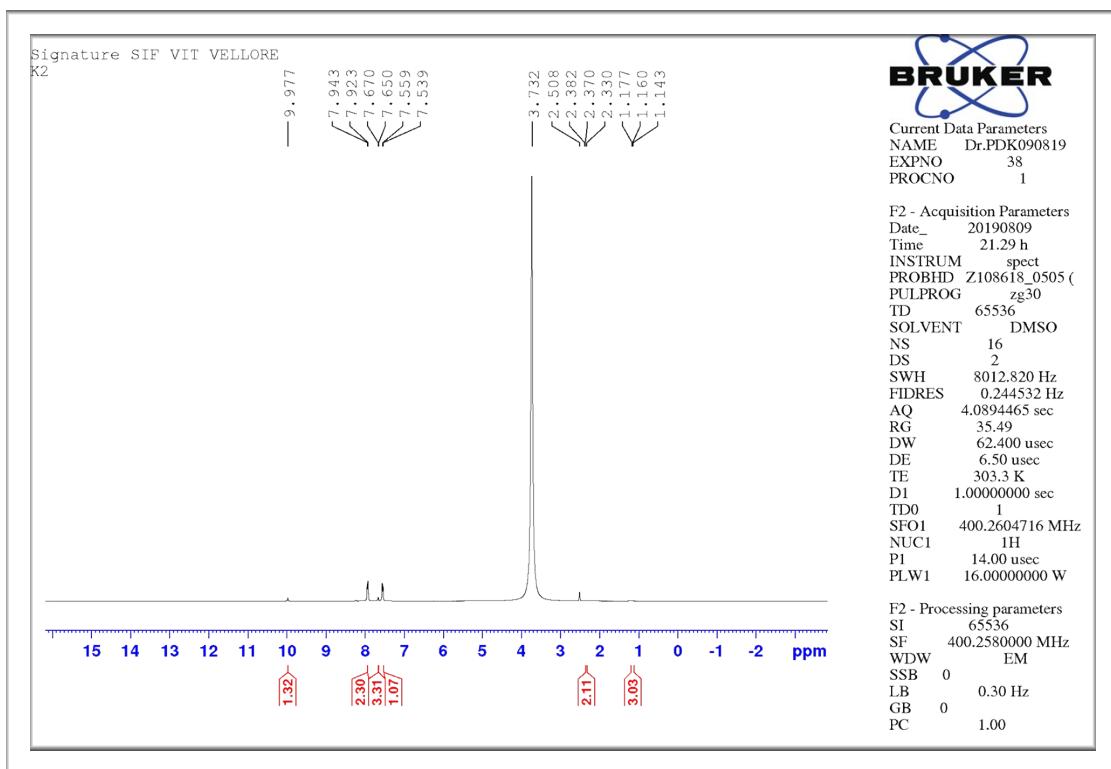
<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ : δ 8.66 (d, 2H), 7.53 (d, 1H), 7.31 (t, 1H), 7.11 (d, 1H)

GC-MS: 130.15



**Fig. S29**  $^1\text{H}$  NMR Spectrum of (E)-1-(3-bromoprop-1-en-1-yl)-4-nitrobenzene

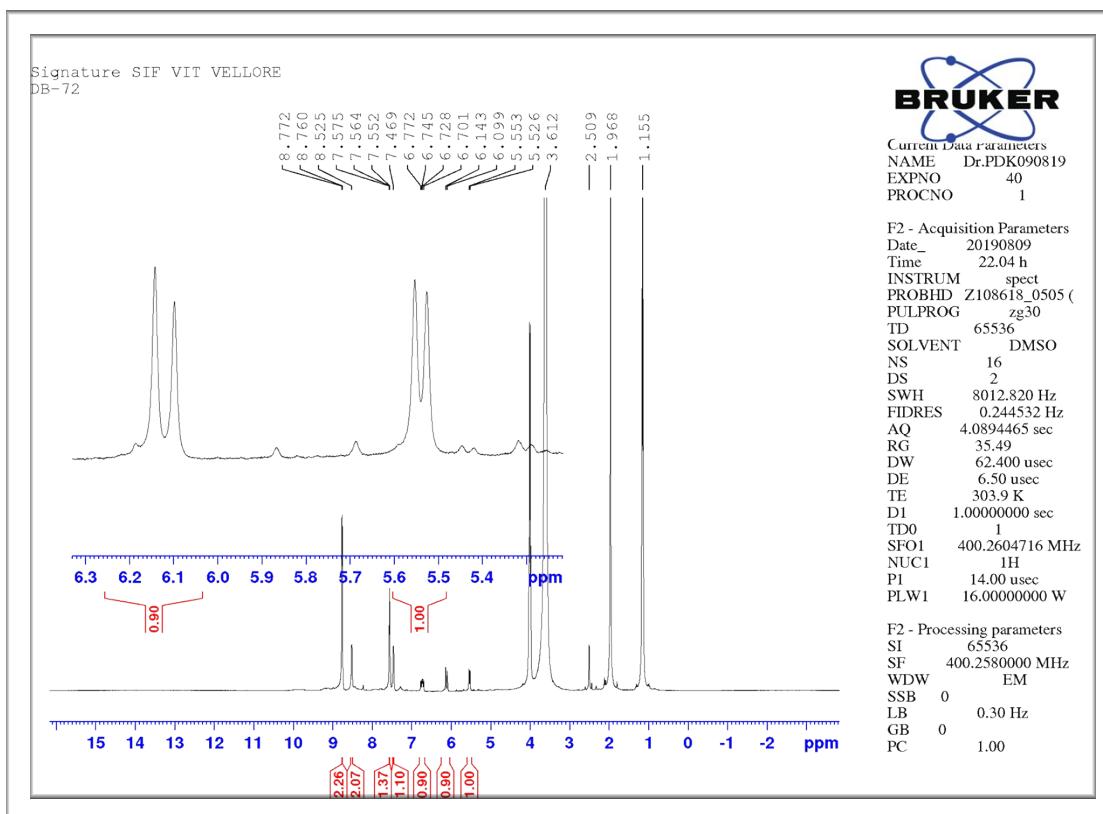
$^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  : $\delta$  8.19 (d, 2H), 8.18 (d, 2H), 7.53 (d, 1H), 7.51 (t, 1H), 1.59 (s, 1H)



**Fig. S30**  $^1\text{H}$  NMR Spectrum of ethyl(E)-3-(4-formylphenyl)acrylate

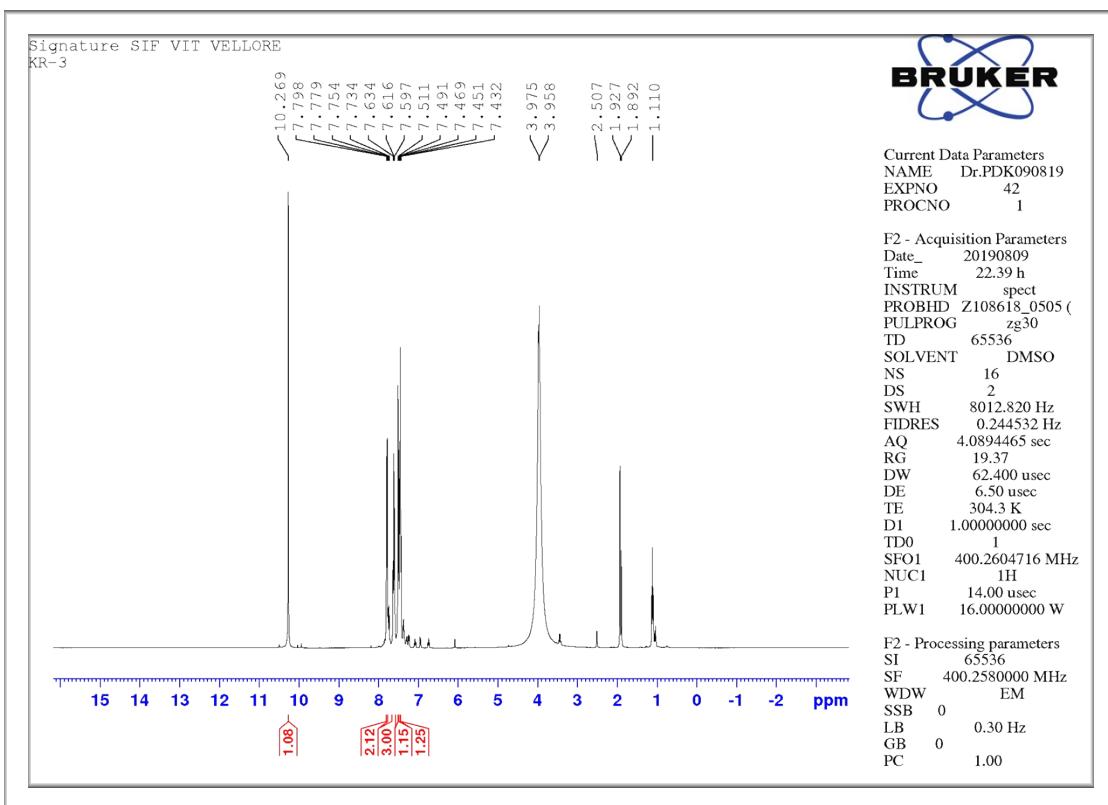
$^1\text{H}$  NMR (400 MHz, DMSO-d6):  $\delta$  9.97 (s, 1H), 7.93 (d, 2H), 7.66 (d, 3H), 7.54 (d, 1H), 2.40 (q, 2H), 1.16 (t, 3H).

$^{13}\text{C}$  NMR (100 MHz, DMSO-d6):  $\delta$  192.69, 167.02, 139.91, 138.32, 138.23, 131.61, 130, 129.62, 129.20, 65.2, 20.4



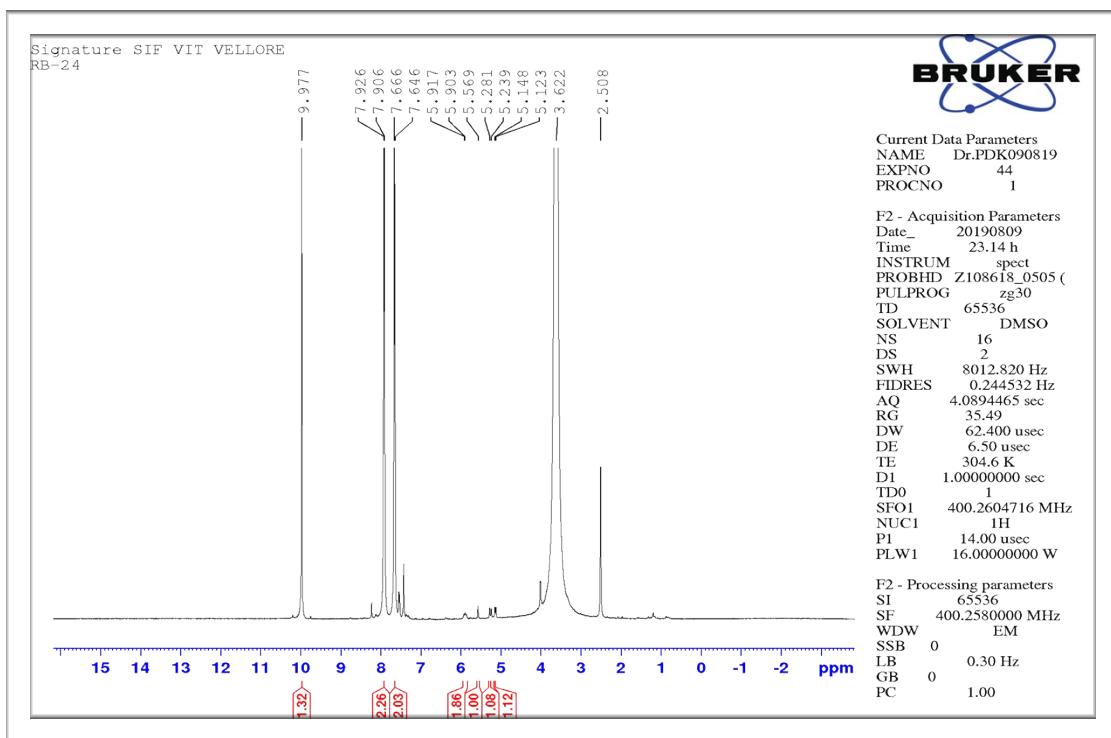
**Fig. S31**  $^1\text{H}$  NMR Spectrum of (E)-(2-(pyridine-4-yl)vinyl)pyrimidine

<sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>): δ 8.77 (d, 2H), 8.52 (s, 2H), 7.56 (t, 1H), 7.46 (s, 1H), 6.74 (t, 1H), 6.12 (d, 1H), 5.54 (d, 1H).

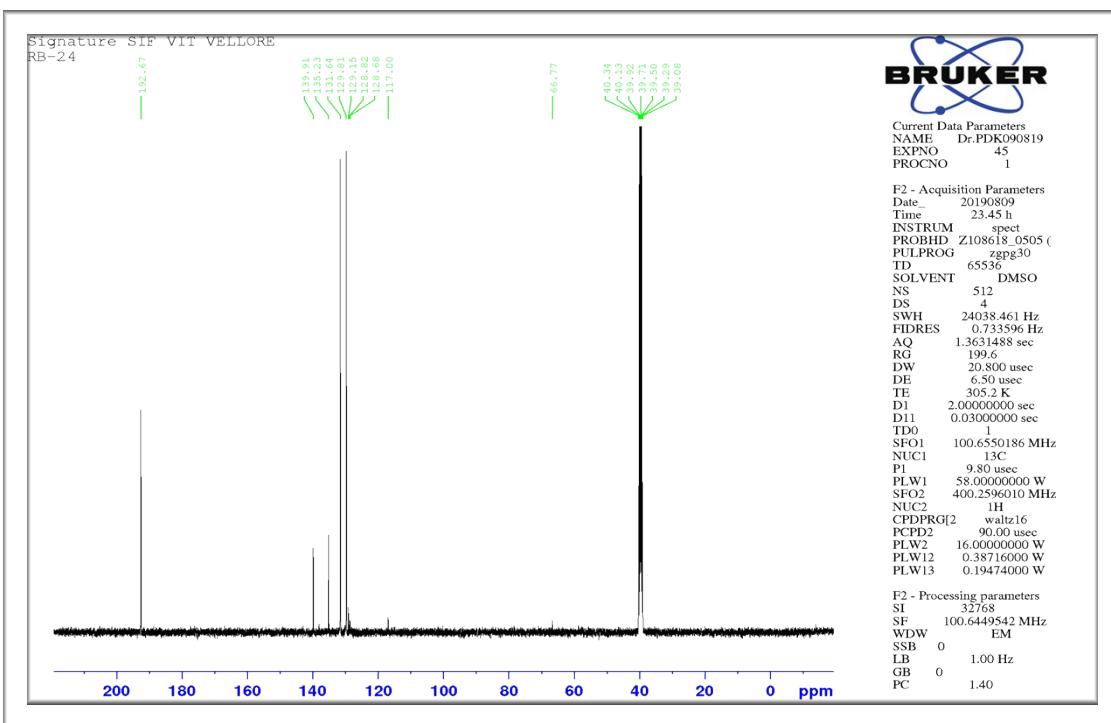


**Fig. S32**  $^1\text{H}$  NMR Spectrum of 2-(thiophen-2-yl)benzaldehyde

$^1\text{H}$  NMR (400 MHz, DMSO-d<sub>6</sub>):  $\delta$  10.26 (s, 1H), 7.76 (q, 2H), 7.61 (t, 3H), 7.50 (d, 1H), 7.45 (t, 1H).



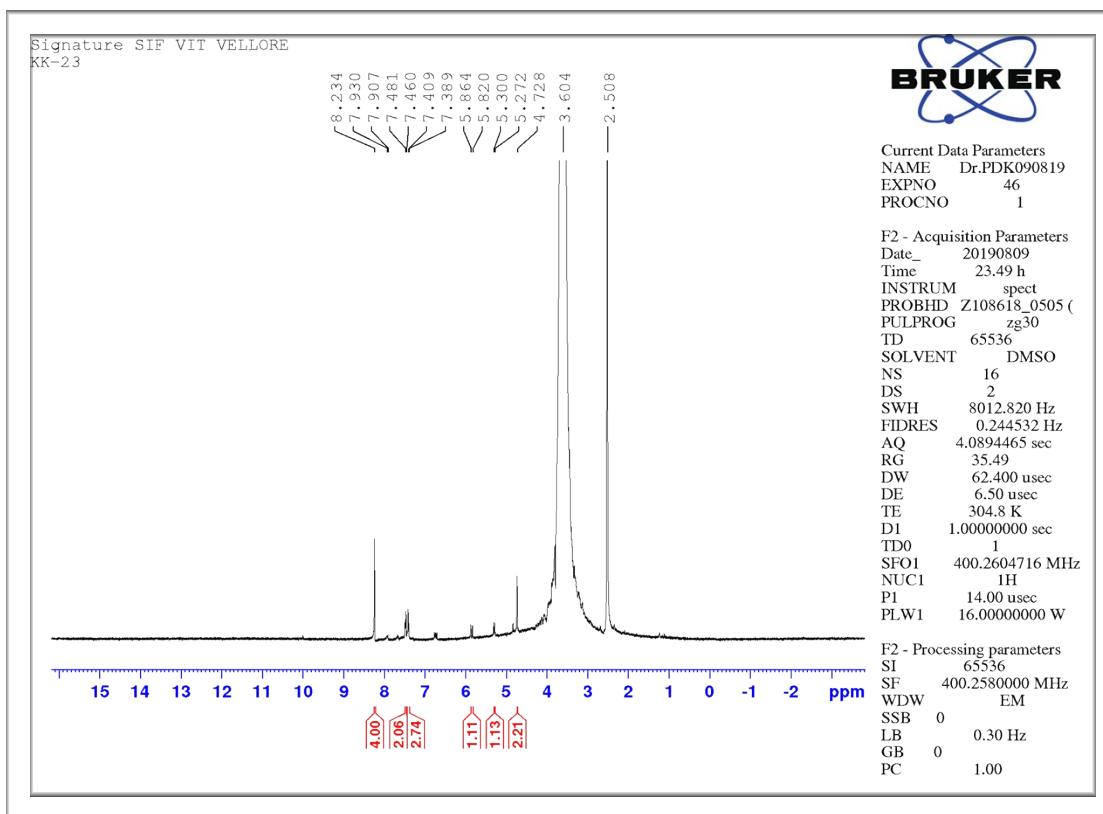
**Fig. S33**  $^1\text{H}$  NMR Spectrum of (Z)-4-(3-hydroxyprop-1-en-yl)benzaldehyde



**Fig. S34**  $^{13}\text{C}$  NMR Spectrum of (Z)-4-(3-hydroxyprop-1-en-yl)benzaldehyde

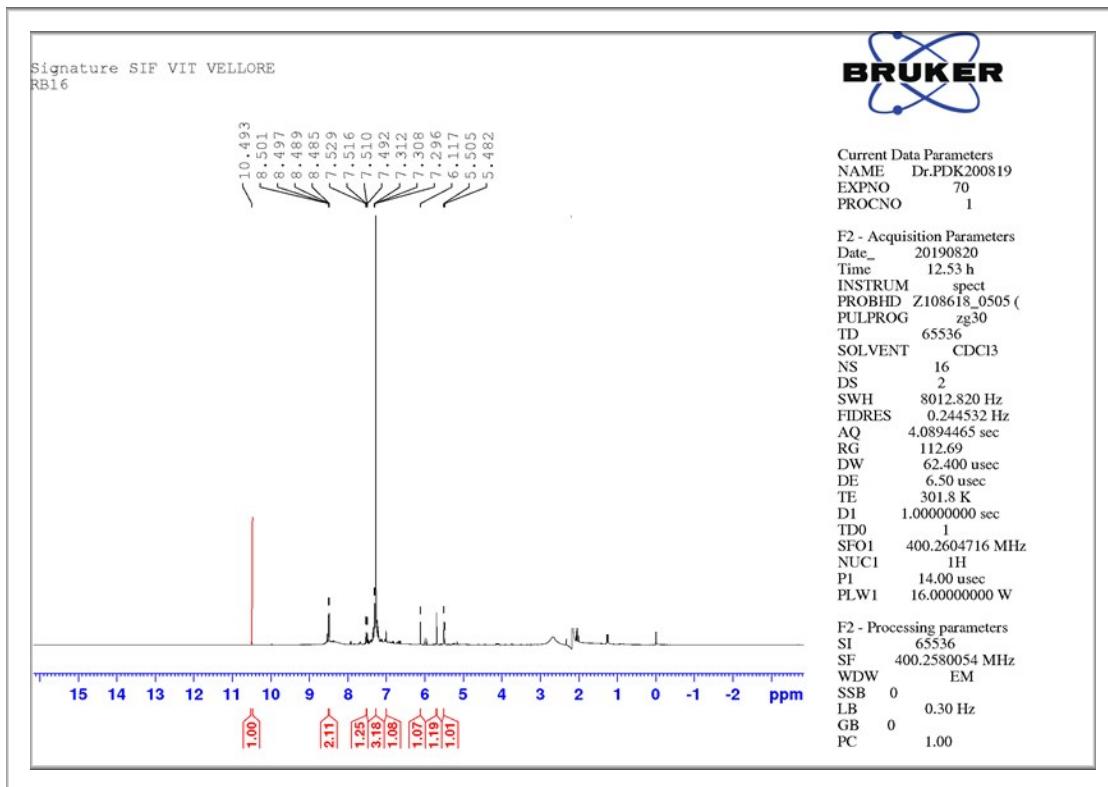
$^1\text{H}$  NMR (400 MHz, DMSO-d6):  $\delta$  9.97 (s, 1H), 7.91 (d, 2H), 7.65 (d, 2H), 5.91 (d, 2H), 5.56 (s, 1H), 5.26 (d, 1H), 5.13 (d, 1H).

$^{13}\text{C}$  NMR (100 MHz, DMSO-d6):  $\delta$  192.67, 139.90, 135.23, 131.63, 129.80, 129.14, 128.82, 128.68, 117, 66.74



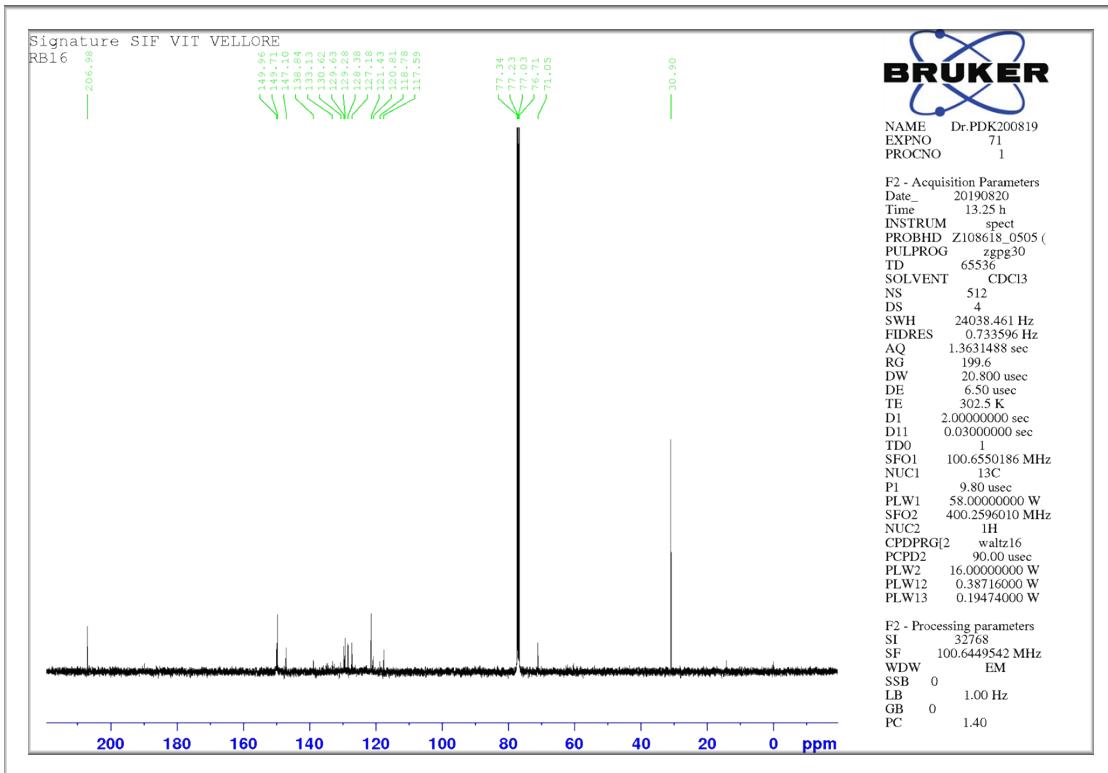
**Fig. S35**  $^1\text{H}$  NMR Spectrum of (Z)-4-(3-bromoprop-1-en-1-yl)benzaldehyde

$^1\text{H}$  NMR (400 MHz, DMSO-d6):  $\delta$  8.23 (s, 4H), 7.47 (d, 2H), 7.39 (d, 3H), 5.84 (d, 1H), 5.29 (d, 1H), 4.72 (s, 2H).



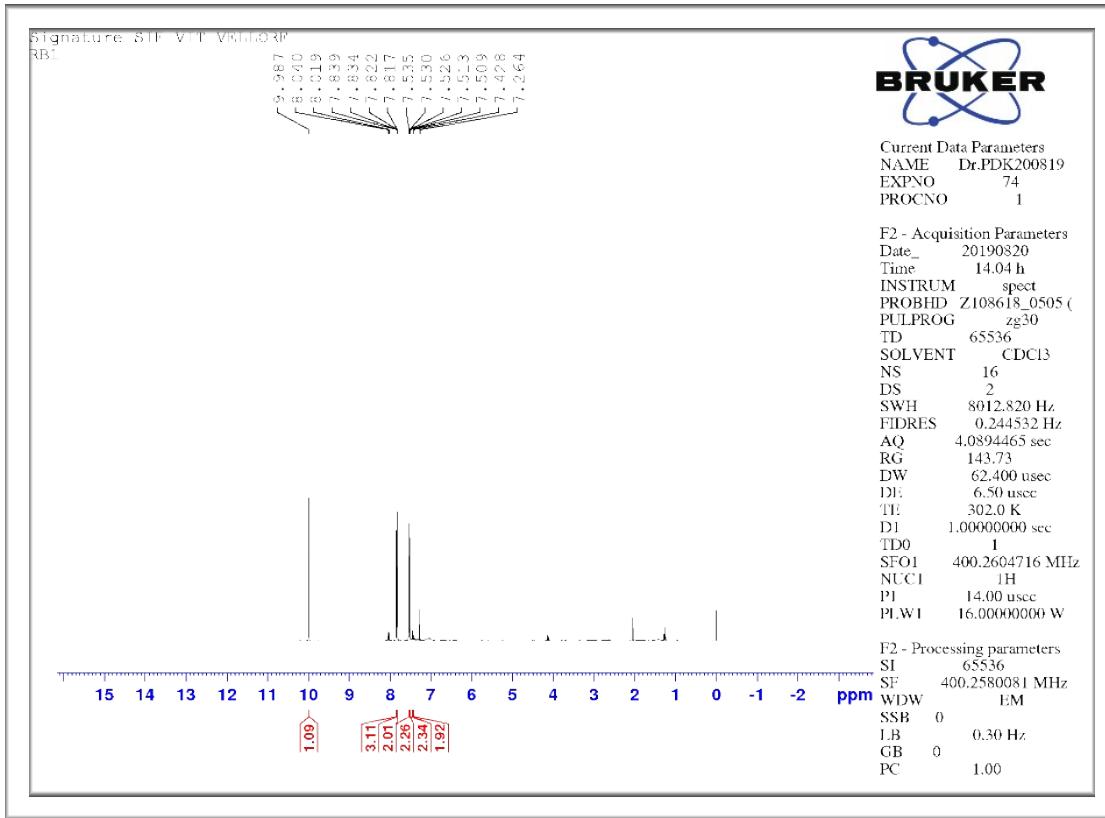
**Fig. S36** <sup>1</sup>H NMR Spectrum of (E)-2-(pyridine-4-yl)vinylbenzaldehyde

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) : δ 10.49 (s, 1H), 8.49 (q, 2H), 7.51 (q, 1H), 7.30 (t, 3H), 7.00 (s, 1H), 6.11 (s, 1H), 5.69 (s, 1H), 5.49 (d, 1H).



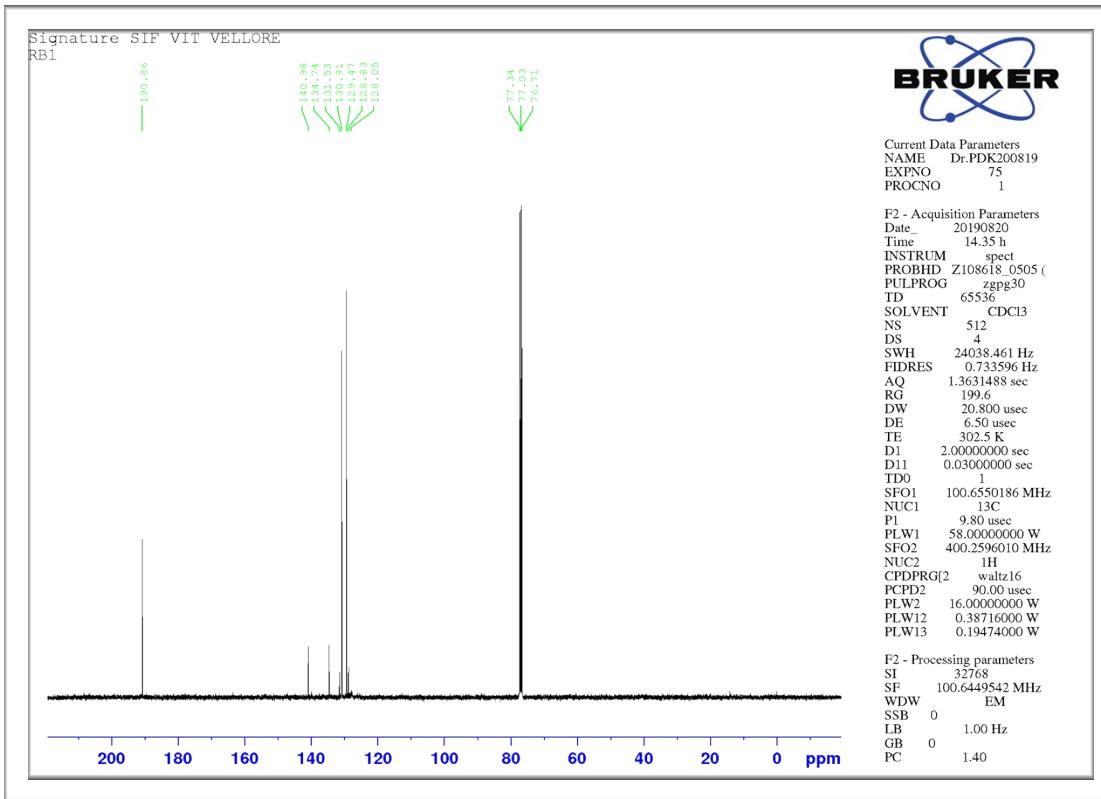
**Fig. S37**  $^{13}\text{C}$  NMR Spectrum of (E)-2-(pyridine-4-yl)vinylbenzaldehyde

$^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ) :  $\delta$  206.97, 149.95, 149.71, 147.10, 138.83, 133.12, 130.62, 129.63, 129.27, 128.37, 127.18, 121.41, 120.80, 118.78, 117.59



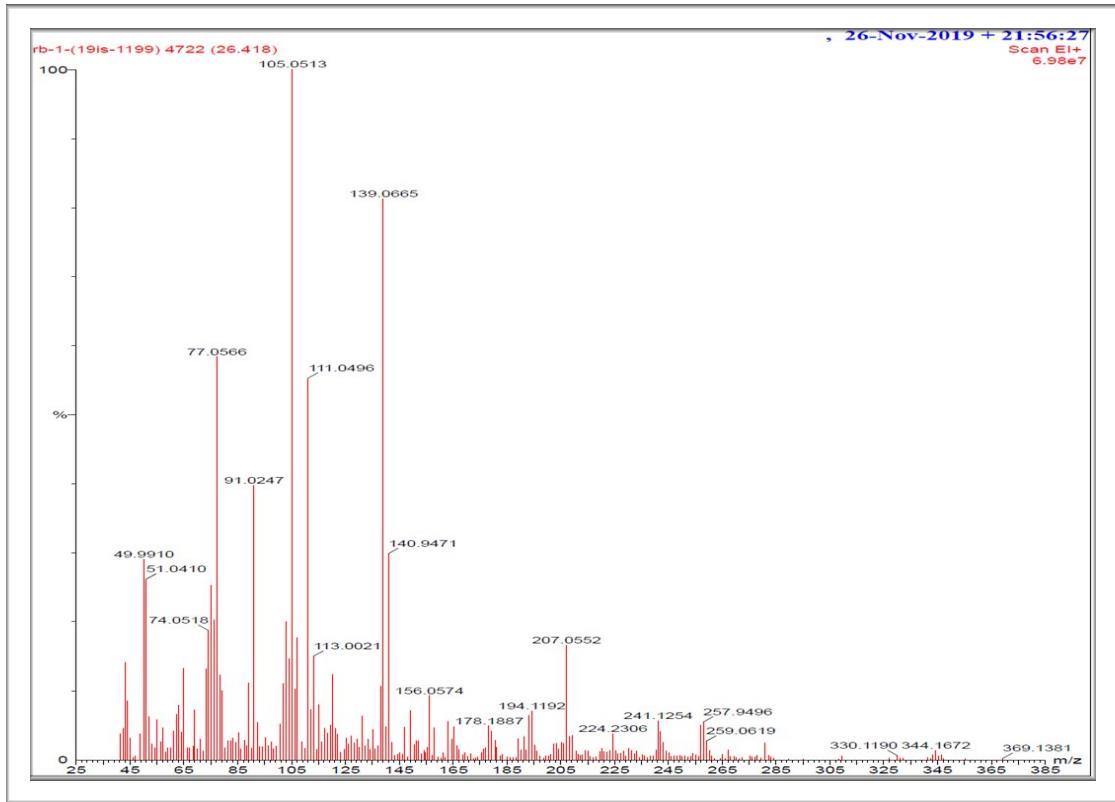
**Fig. S38** <sup>1</sup>H NMR Spectrum of (Z)-4-styrylbenzaldehyde

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) : δ 9.98 (s, 1H), 7.83 (d, 3H), 7.82 (d, 2H), 7.52 (t, 2H), 7.51 (d, 2H), 7.42 (s, 2H).

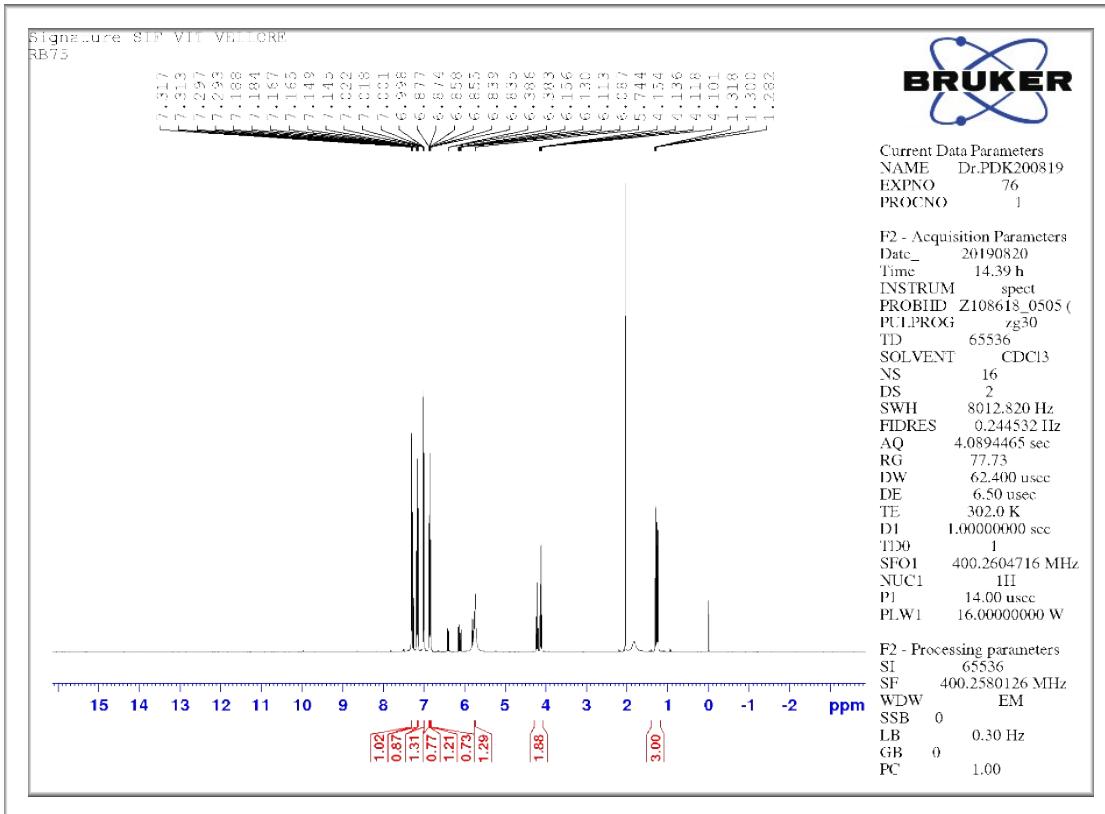


**Fig. S39**  $^{13}\text{C}$  NMR Spectrum of (Z)-4-styrylbenzaldehyde

$^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ) :  $\delta$  190.86, 140.98, 134.73, 131.52, 130.91, 129.47, 128.83, 128.05.

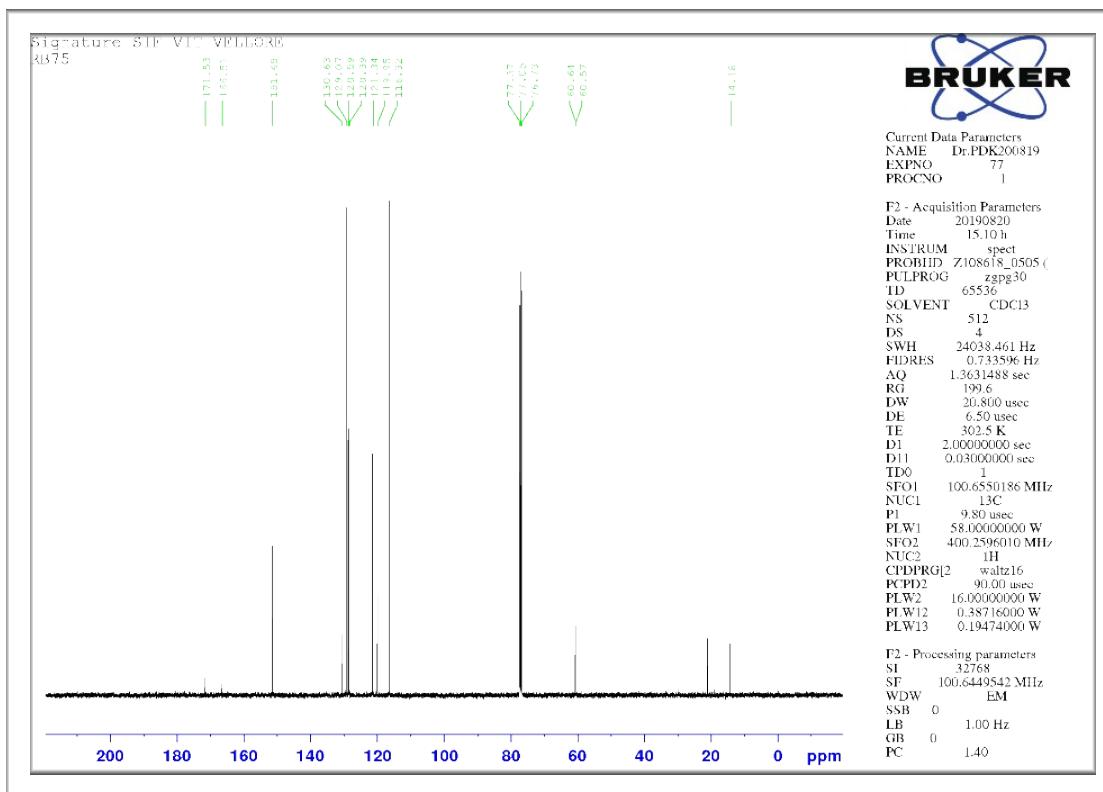


**Fig. S40** GC-MS Spectrum of (Z)-4-styrylbenzaldehyde GC-MS: 208.26



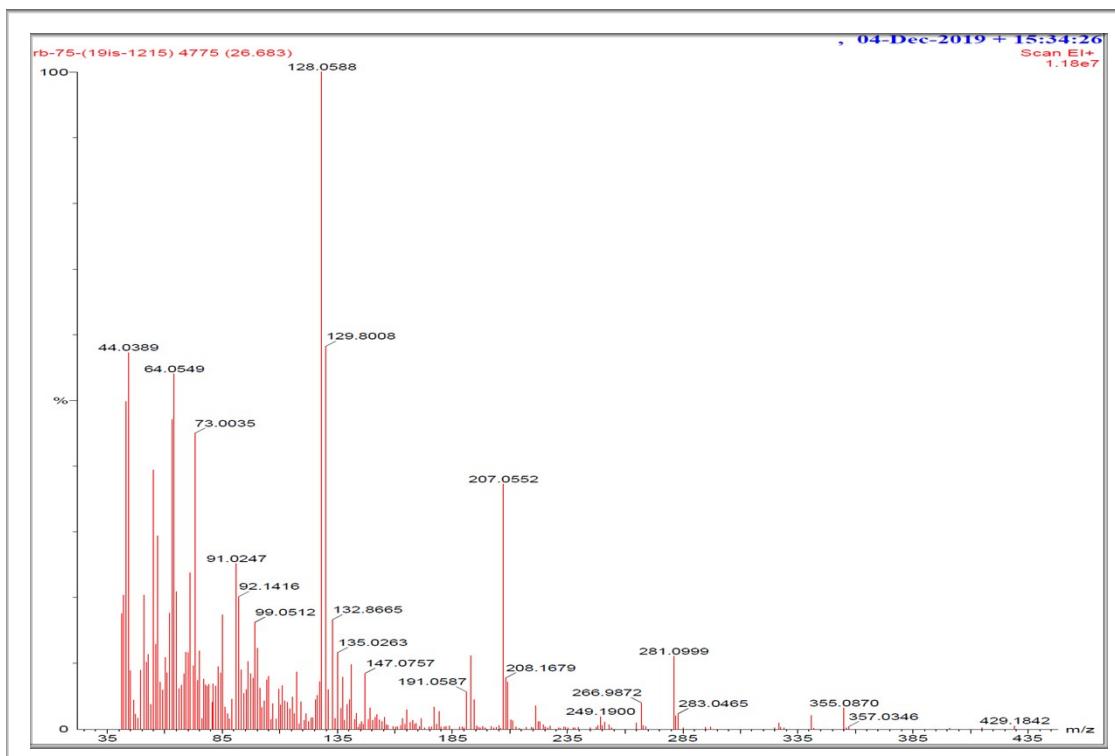
**Fig. S41**  $^1\text{H}$  NMR Spectrum of ethyl(E)-3-(2-hydroxyphenyl)acrylate

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) : δ 7.30 (d, 1H), 7.14 (d, 1H), 7.35 (d, 1H), 6.87 (d, 1H), 6.85 (d, 1H), 6.83 (d, 1H), 5.74 (s, 1H), 4.12 (q, 2H), 1.29 (t, 3H), .

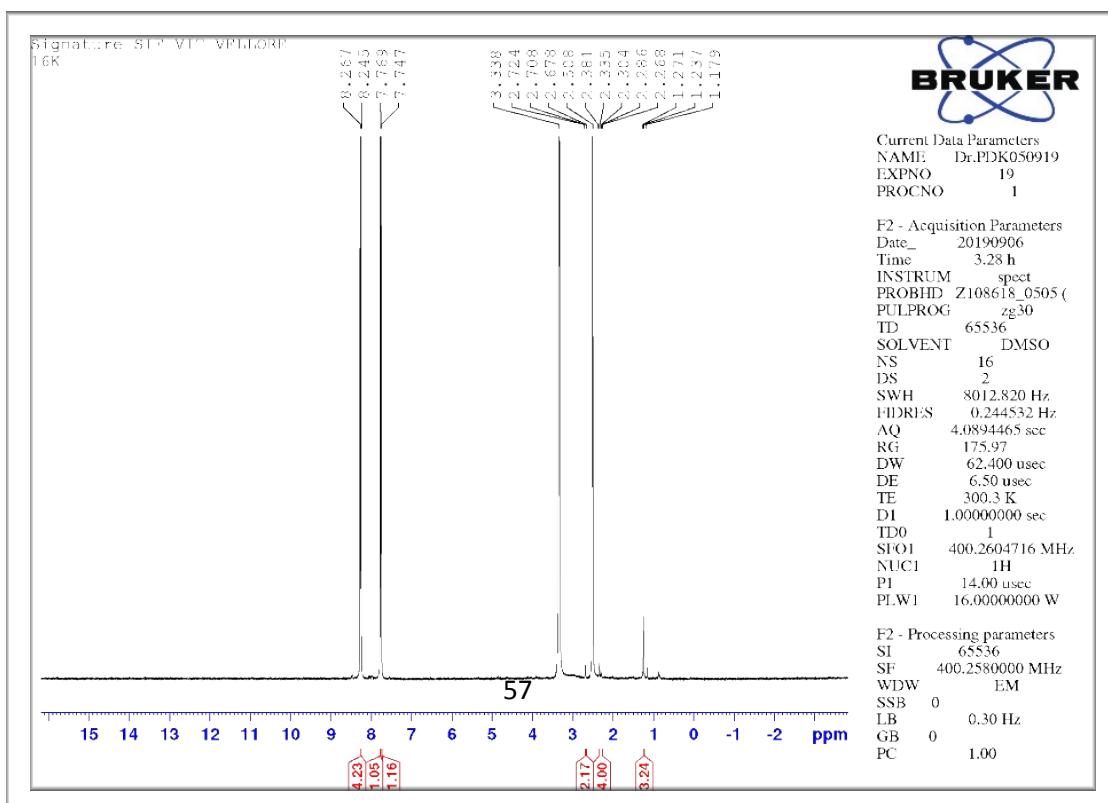


**Fig. S42**  $^{13}\text{C}$  NMR Spectrum of ethyl(E)-3-(2-hydroxyphenyl)acrylate

$^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ) :  $\delta$  166.51, 151.45, 130.63, 129.07, 128.59, 128.38, 121.33, 119.94, 116.32, 60.64, 14.17

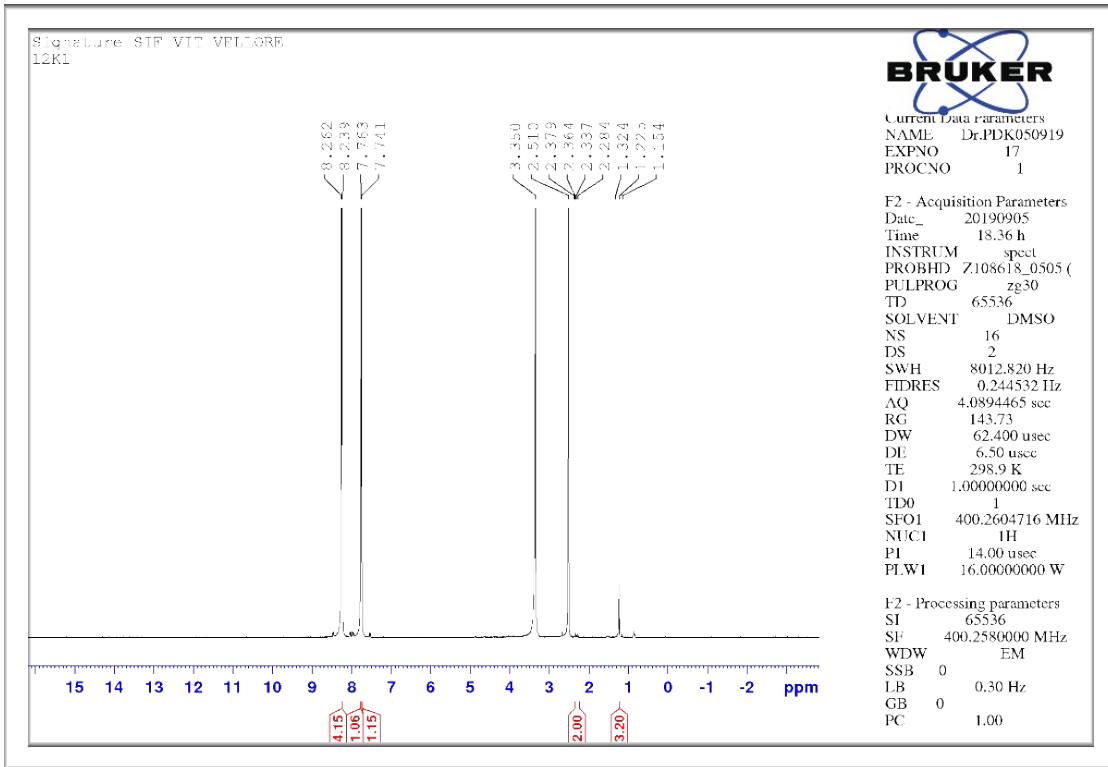


**Fig. S43** GC-MS ethyl(E)-3-(2-hydroxyphenyl)acrylate GC-MS : 192.018



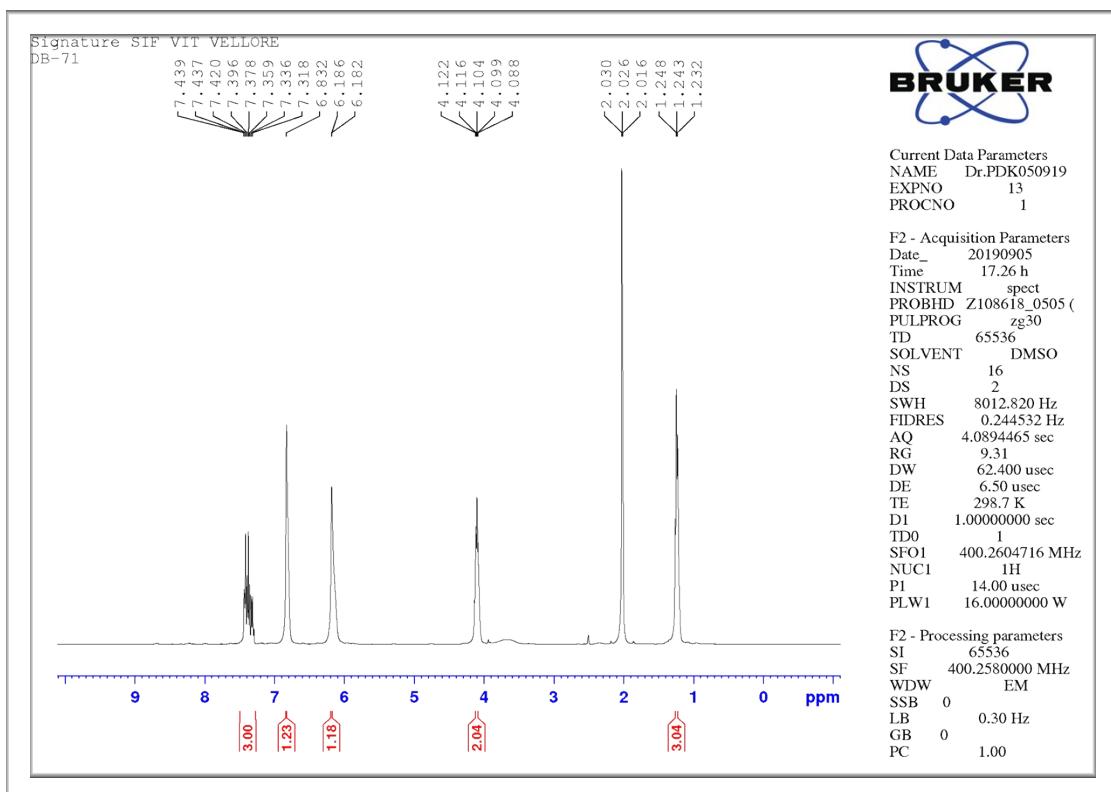
**Fig. S44**  $^1\text{H}$  NMR Spectrum of (Z)-1-(4-nitrophenyl)hept-1-en-one

$^1\text{H}$  NMR (400 MHz, DMSO-d6):  $\delta$  8.25 (d, 4H), 7.75 (d, 2H), 2.70 (t, 2H), 2.26-2.28 (m, 4H), 1.22 (t, 3H)



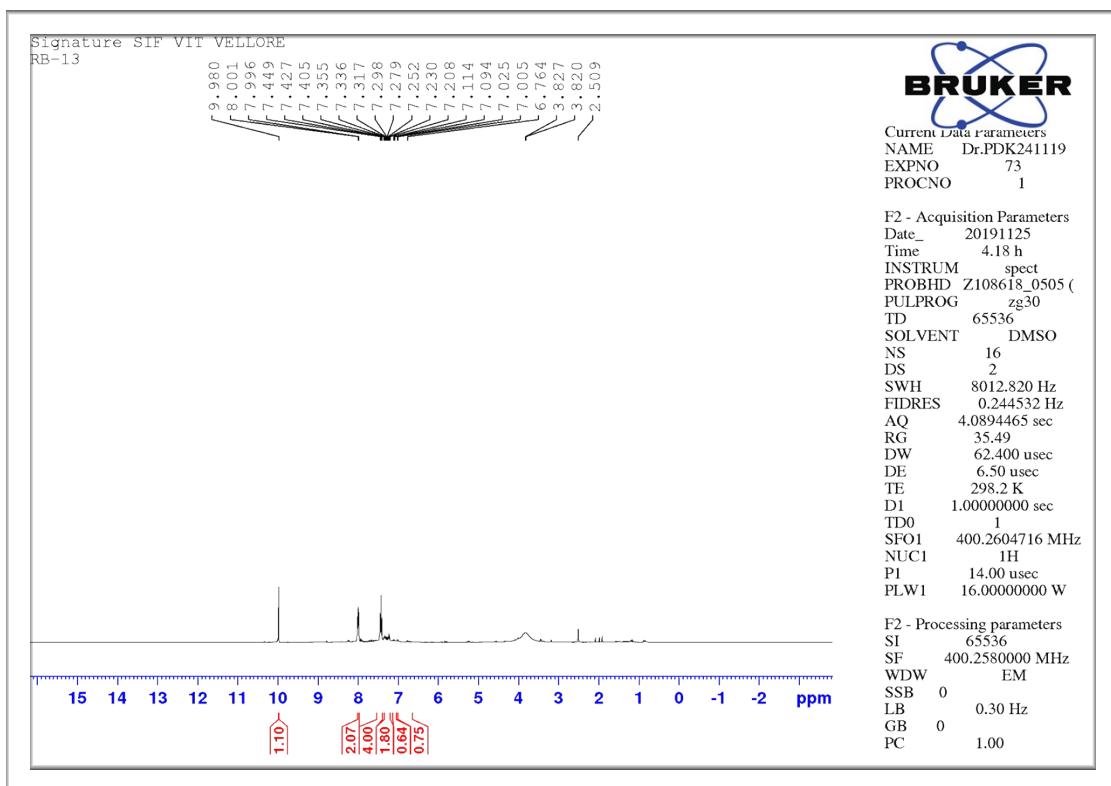
**Fig. S45**  $^1\text{H}$  NMR Spectrum of ethyl(Z)-3(4-nitrophenyl)acrylate

$^1\text{H}$  NMR (400 MHz, DMSO-d6):  $\delta$  8.25 (d, 4H), 7.75 (d, 2H), 2.28-2.33 (m, 2H), 1.23 (t, 3H)



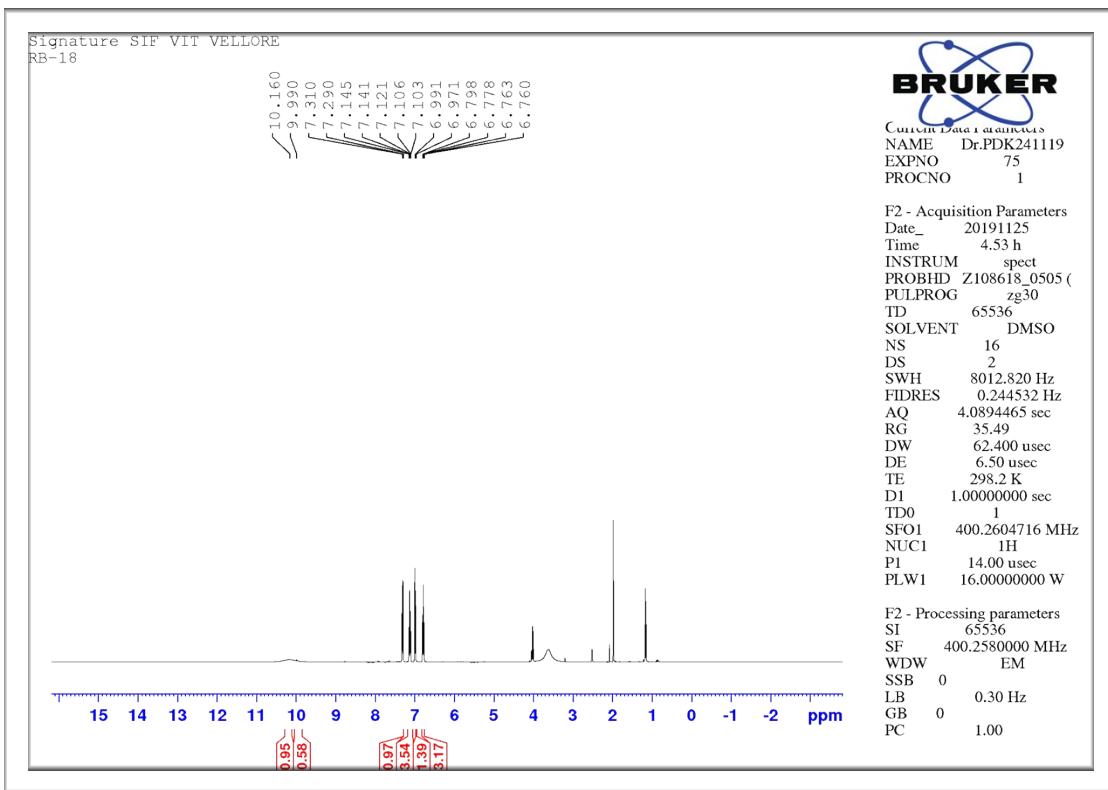
**Fig. S46**  $^1\text{H}$  NMR Spectrum of ethyl(Z)-3-(pyrimidin-2-yl)acrylate

$^1\text{H}$  NMR (400 MHz, DMSO-d<sub>6</sub>):  $\delta$  7.31-7.33 (m, 3H), 6.83 (s, 1H), 6.18 (d, 1H), 4.09-4.10 (m, 2H), 1.24 (t, 3H)



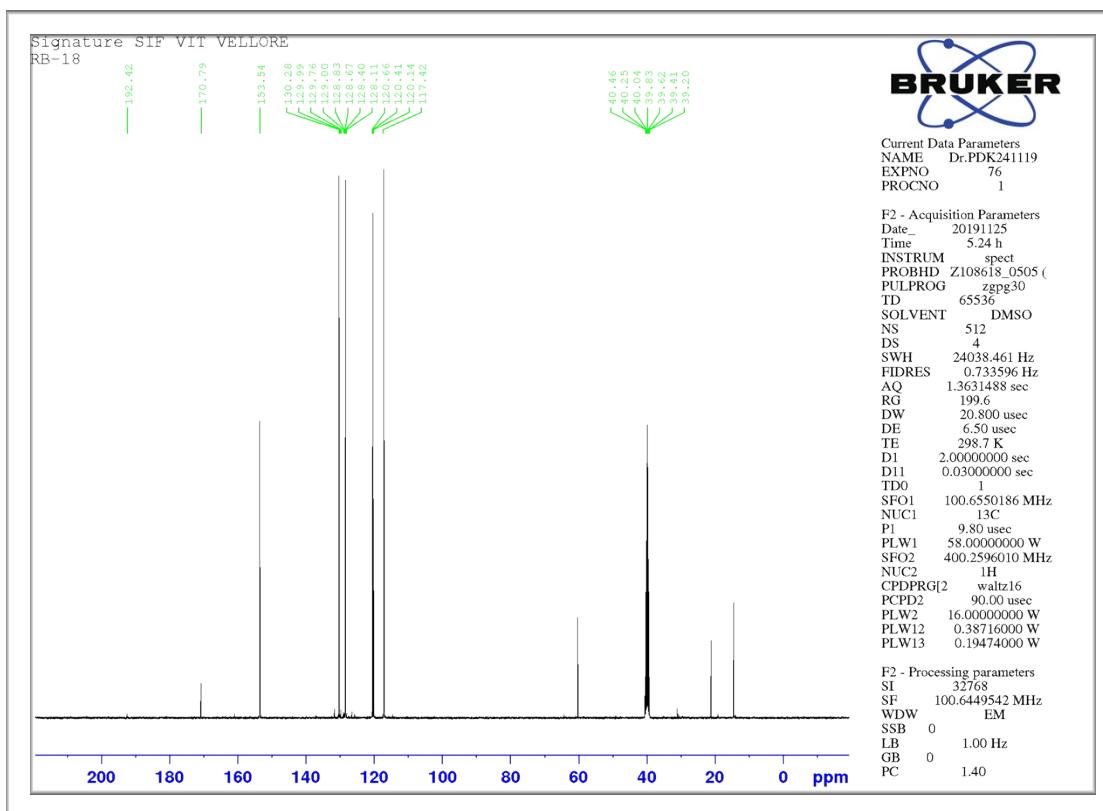
**Fig. S47**  $^1\text{H}$  NMR Spectrum of (E)-4-styrylbenzaldehyde

$^1\text{H}$  NMR (400 MHz, DMSO-d<sub>6</sub>):  $\delta$  9.98 (s, 1H), 8.00 (d, 2H), 7.42 (t, 4H), 7.31 (q, 2H), 7.10 (d, 1H), 6.76 (s, 1H).



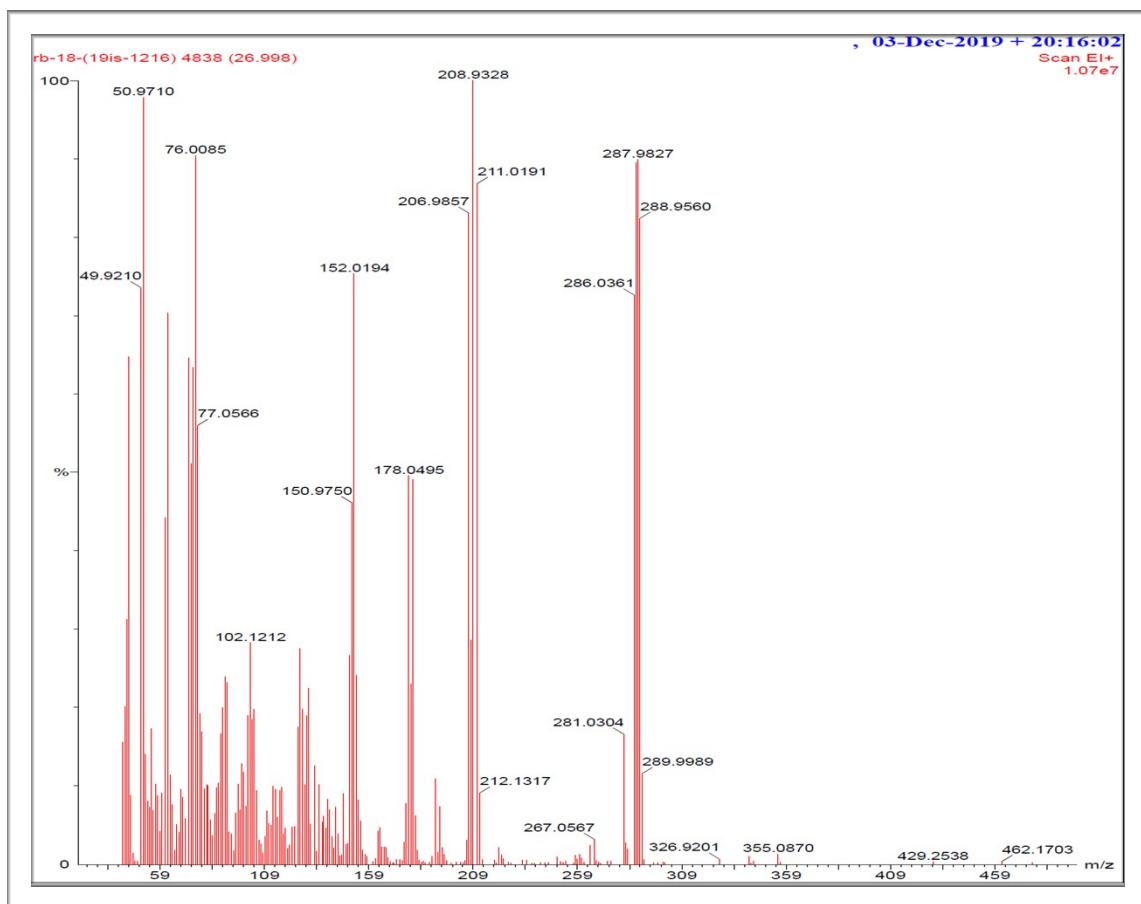
**Fig. S48**  $^1\text{H}$  NMR Spectrum of (Z)-2-hydroxy-5-(2-(pyridine-4-yl)vinyl)benzaldehyde

$^1\text{H}$  NMR (400 MHz, DMSO-d6):  $\delta$  10.16 (s, 1H), 9.98 (s, 1H), 7.30 (d, 1H), 7.12 (t, 4H), 6.98 (d, 1H), 6.77 (q, 3H).

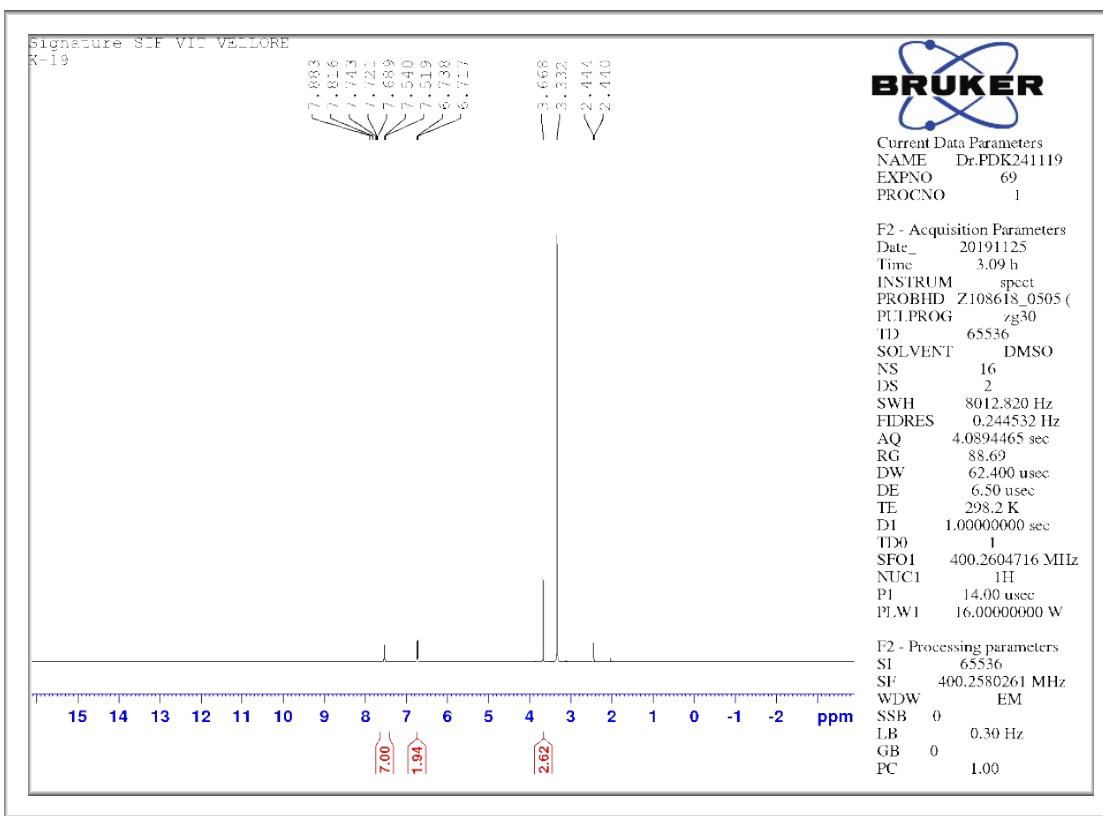


**Fig. S49**  $^{13}\text{C}$  NMR Spectrum of (Z)-2-hydroxy-5-(2-(pyridine-4-yl)vinyl)benzaldehyde

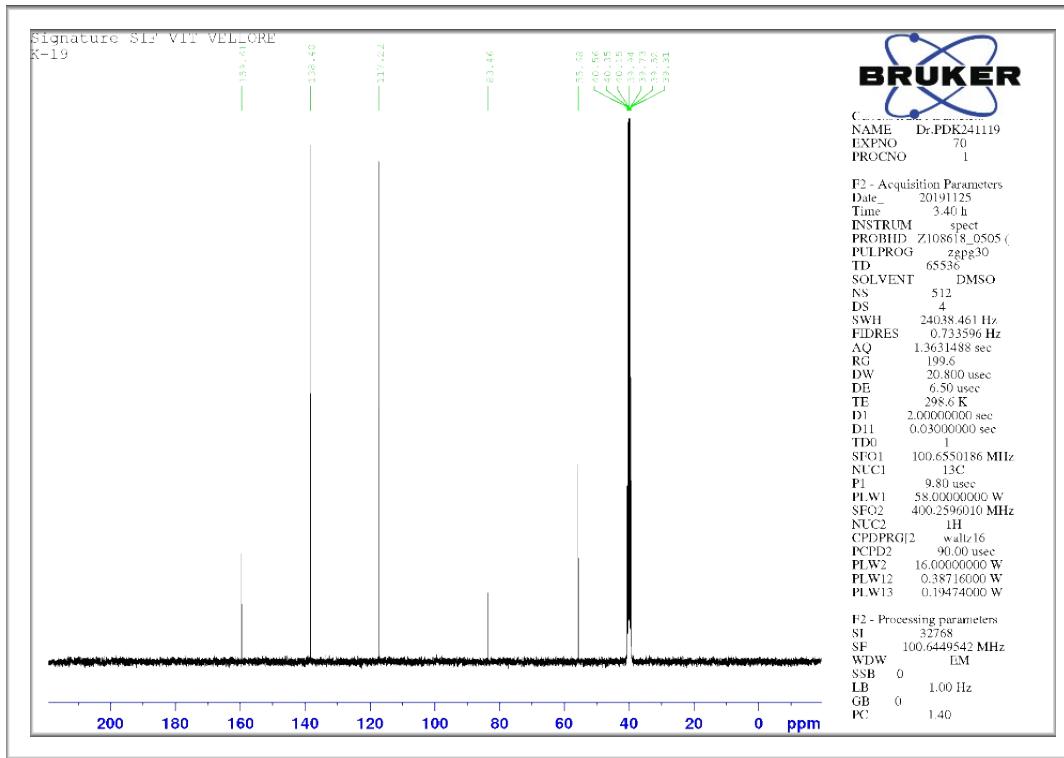
$^{13}\text{C}$  NMR (100 MHz, DMSO-d6):  $\delta$  192.42, 170.79, 153.54, 130.28, 129.99, 129.76, 128.99, 128.83, 128.67, 128.40, 128.11, 120.66, 120.41, 120.14, 117.42



**Fig. S50** GC-MS Spectrum of (Z)-2-hydroxy-5-(2-(pyridine-4-yl)vinyl)benzaldehyde  
GC-MS: 225.25

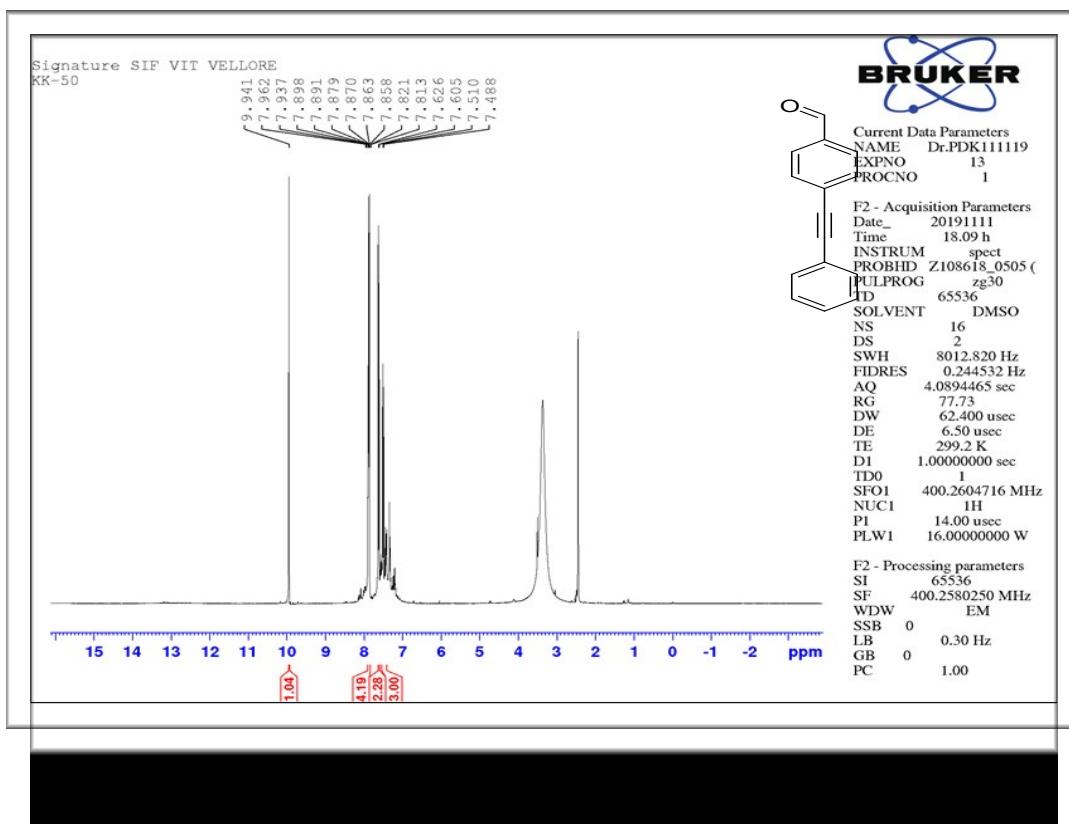


**Fig. S51**  $^1\text{H}$  NMR Spectrum of 4-methoxy-1,1'-biphenyl  
 $^1\text{H}$  NMR (400 MHz, DMSO-d6):  $\delta$  7.54-7.68 (m, 7H), 6.72 (d, 2H), 3.66 (s, 2H).



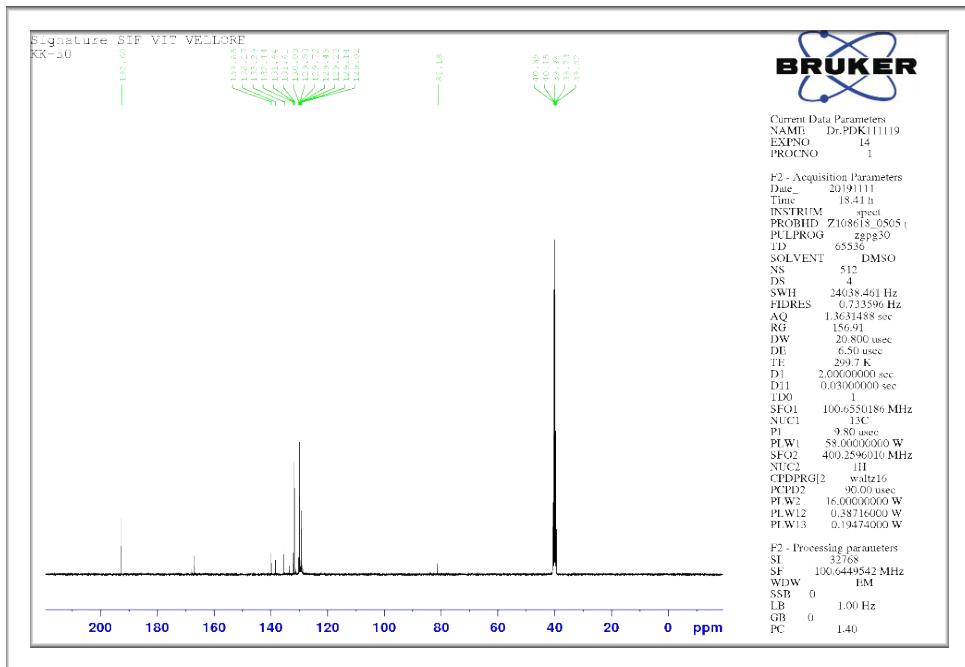
**Fig. S52**  $^{13}\text{C}$  NMR Spectrum of 4-methoxy-1,1'-biphenyl  
 $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$ : 159.61, 138.40, 117.21, 83.46, 55.67.

## Sonogashira Coupling



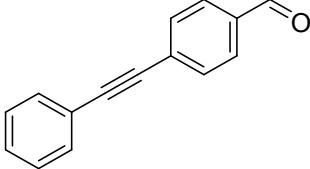
**Fig. S53**  $^1\text{H}$  NMR Spectrum of 4-(phenylethyl)benzaldehyde

400 MHz, DMSO-d6:  $\delta$  9.94 (s, 1H), 7.82-7.85 (m, 4H), 7.61 (d, 2H), 7.50 (d, 3H).



**Fig. S54**  $^{13}\text{C}$  NMR Spectrum of 4-(phenylethyl)benzaldehyde  
 $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$ : 192.60, 137.05, 138.27, 135.74, 132.14, 131.44, 130.09, 129.80, 129.72, 129.43, 129.20, 129.14, 129.02, 81.10

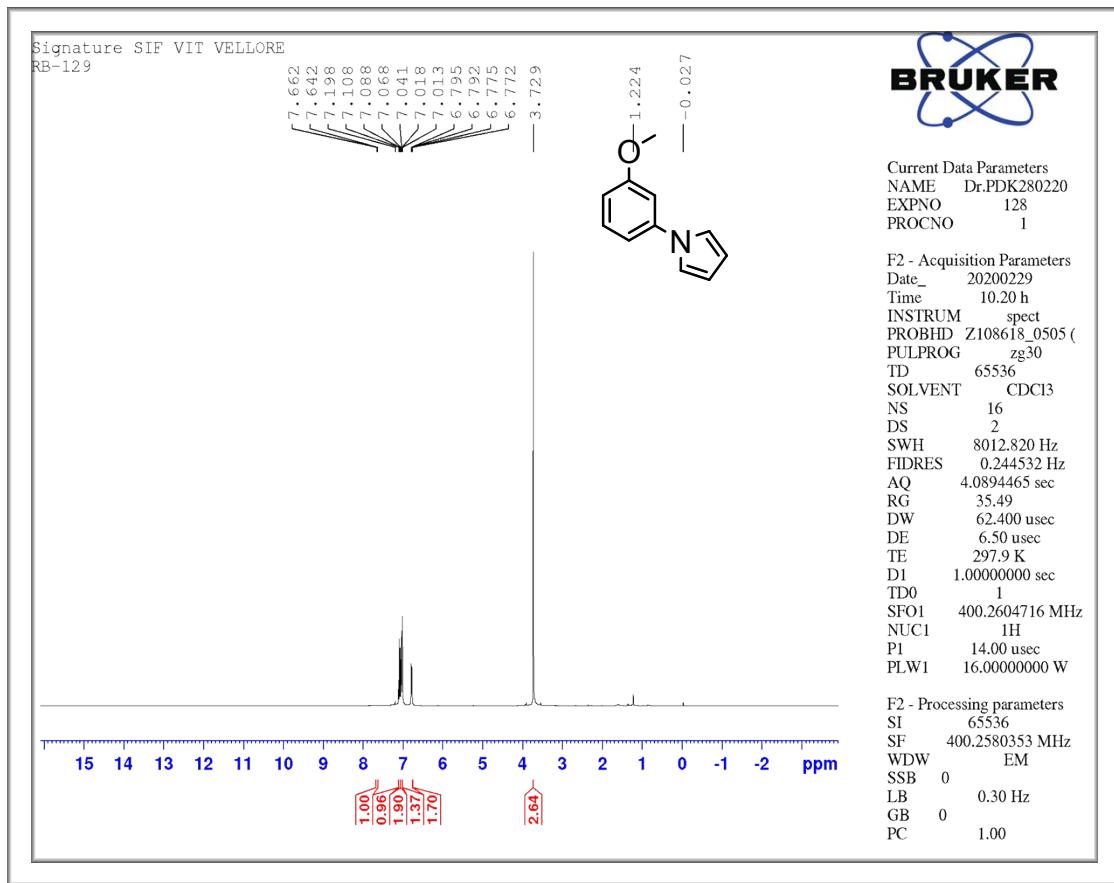
**Table S2.  $^1\text{H}$  and  $^{13}\text{C}$  value for Sonogashira coupling between various Ethynylbenzene and Aryl halide under Photocatalytic Conditions.**

| Sonogashira product   | Coupling | Conformation Data (NMR data)  |
|---|----------|---|
|  |          | 400 MHz, DMSO-d6: $\delta$ 9.94 (s, 1H), 7.82-7.85 (m, 4H), 7.61 (d, 2H), 7.50 (d, 3H).<br>$^{13}\text{C}$ NMR (100 MHz, DMSO) $\delta$ : 192.60, 137.05, 138.27, 135.74, 132.14, 131.44, 130.09, 129.80, 129.72, 129.43, 129.20, 129.14, 129.02, 81.10 |

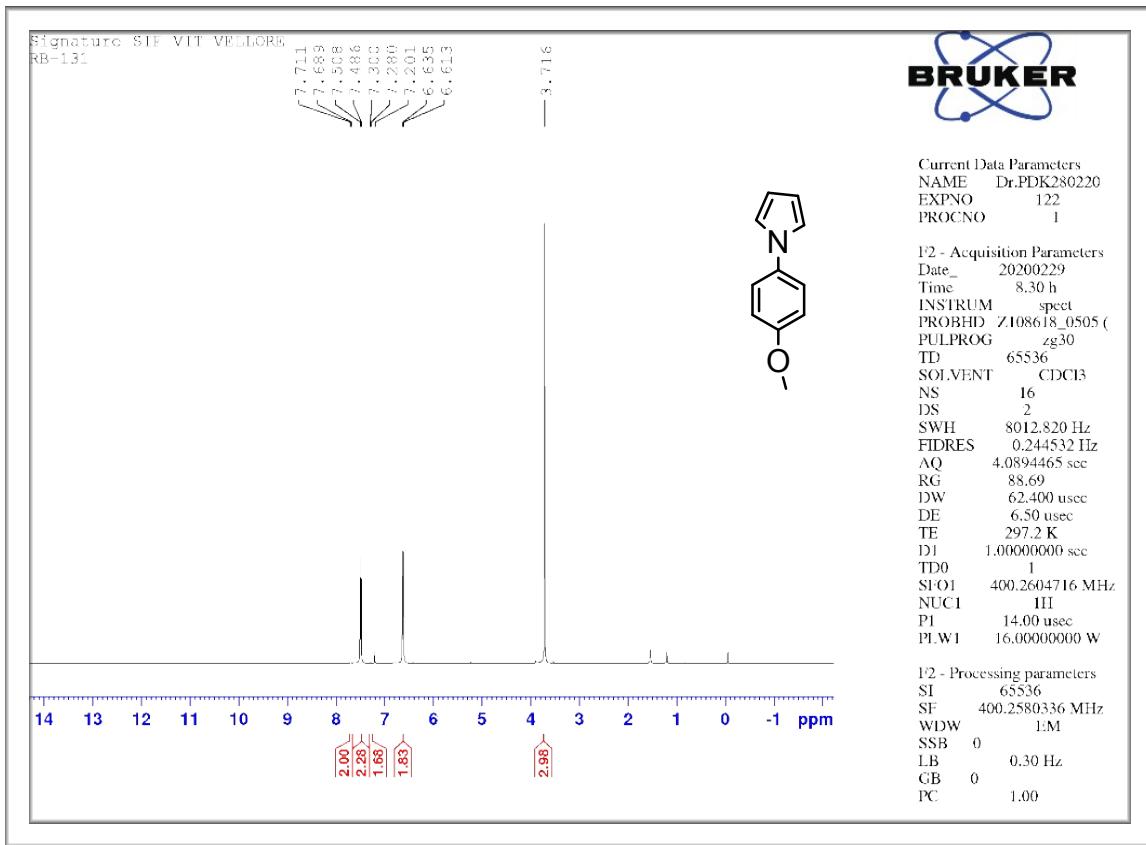
**Table S3. Optimization of Sonogashira between 4-chloroaniline and ethynylbenzene under Photocatalytic Conditions Catalyzed by Photo-catalyst**

| Entry | Reagent | Catalyst (mg) | Time (h) | Yield (%) |
|-------|---------|---------------|----------|-----------|
| 1     | R1 + R2 | 5             | 24       | 56        |
| 2     | R1 + R2 | 10            | 24       | 70        |
| 3     | R1 +R2  | 15            | 24       | 87        |
| 4     | R1 + R2 | 20            | 24       | 96        |
| 5     | R1 + R2 | 25            | 24       | 97        |
| 6     | R1 + R2 | 20            | 6        | 69        |
| 7     | R1 + R2 | 20            | 12       | 82        |
| 8     | R1 + R2 | 20            | 18       | 89        |
| 9     | R1 + R2 | 20            | 24       | 96        |

## Buchwald-Hartwig Cross Coupling Reaction

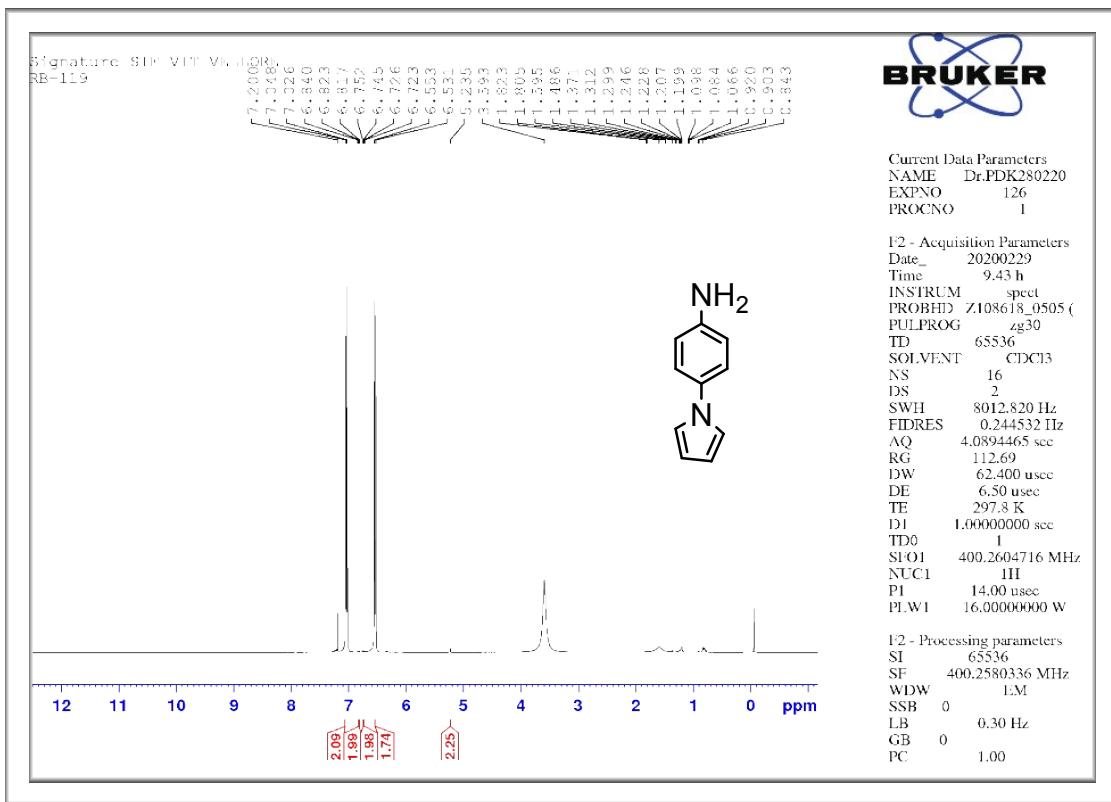


**Fig. S55** <sup>1</sup>H NMR Spectrum of 1-(3-methoxyphenyl)-1-H-pyrrole



**Fig. S56** <sup>1</sup>H NMR Spectrum of 1-(4-methoxyphenyl)-1H-pyrrole

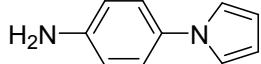
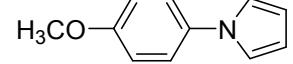
100 MHz, CdCl<sub>3</sub>-d6: δ 7.70 (d, 2H), 7.49 (d, 2H), 7.29 (d, , 2H), 6.62 (d, 2H), 0.00 (s, 3H).



**Fig. S57** <sup>1</sup>H NMR Spectrum of 4-(1H-pyrrole-1-yl)aniline

400 MHz, DMSO-d6:  $\delta$  7.03 (d, 2H), 6.83 (d, 2H), 6.75 (d, 2H), 6.54 (d, 2H), 5.23 (s, 2H).

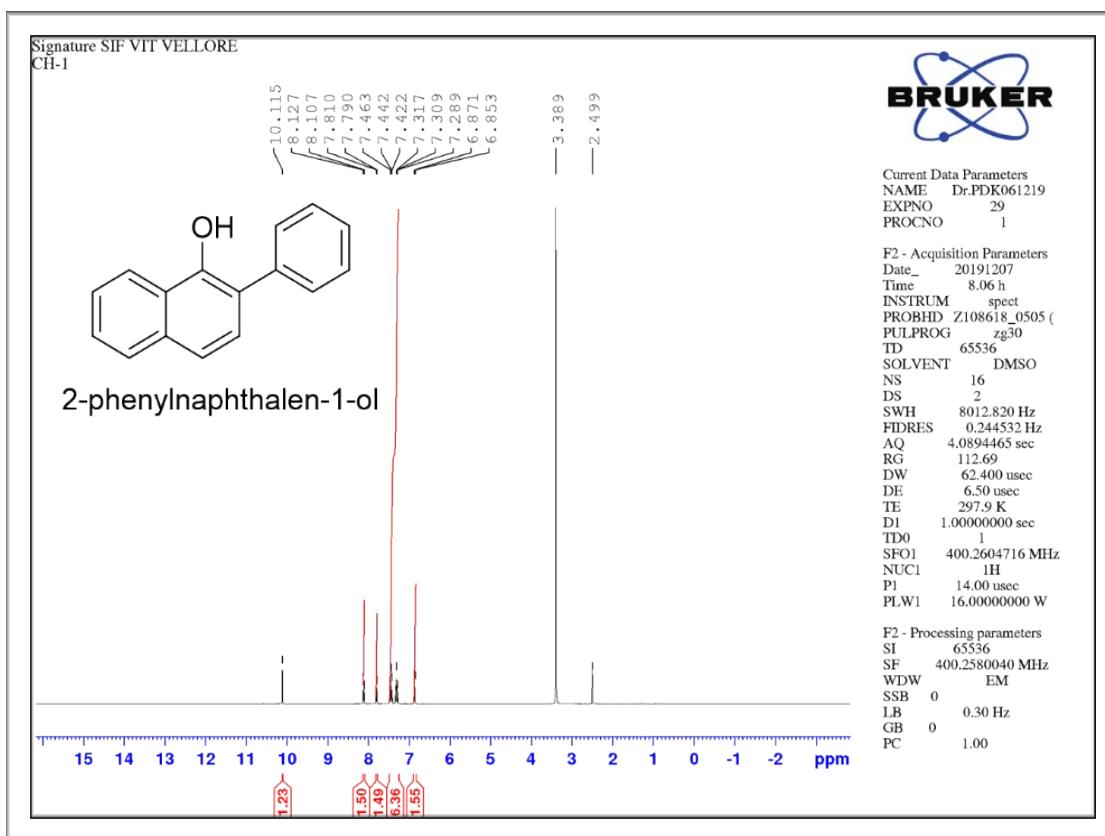
**Table S4.  $^1\text{H}$  and  $^{13}\text{C}$  value for Formation of CN bond between various Aryl halide, Pyrrole and Morpholine under Photocatalytic Conditions.**

| Formation of C-N bond Product   | Conformation Data (NMR data)  |
|---|---|
|  | 400 MHz, DMSO-d6: $\delta$ 7.03 (d, 2H), 6.83 (d, 2H), 6.75 (d, 2H), 6.54 (d, 2H), 5.23 (s, 2H).  |
|  | 400 MHz, CdCl3-d6: $\delta$ 7.70 (d, 2H), 7.49 (d, 2H), 7.29 (d, 2H), 6.62 (d, 2H), 0.00 (s, 3H). |

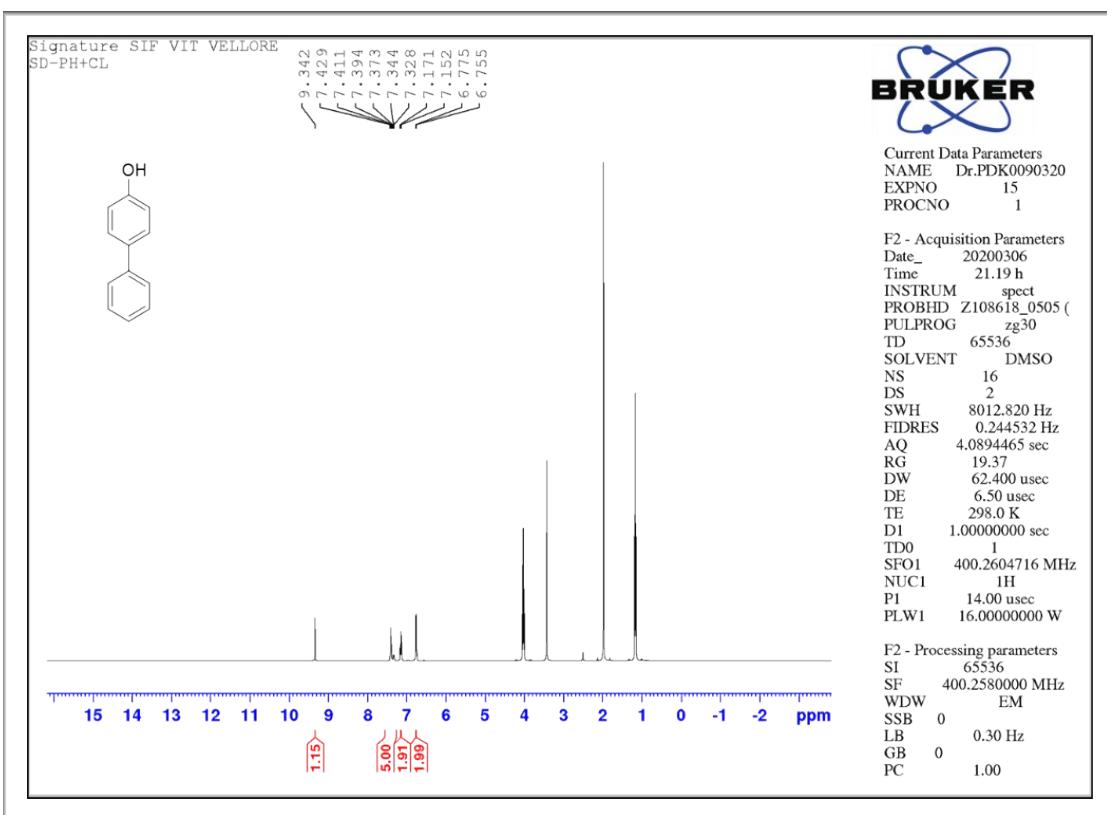
**Table S5. Optimization of Formation of C-N bond between 4-chloroaniline and 1H-pyrrole under Photocatalytic Conditions Catalyzed by Photo-catalyst**

| Entry | Reagent | Catalyst (mg) | Time (h) | Yield (%) |
|-------|---------|---------------|----------|-----------|
| 1     | R1 +R2  | 20            | 24       | 87        |
| 2     | R1 +R2  | 5             | 24       | 52        |
| 3     | R1 +R2  | 10            | 24       | 64        |
| 4     | R1 +R2  | 15            | 24       | 77        |
| 5     | R1 +R2  | 20            | 24       | 87        |
| 6     | R1 +R2  | 25            | 24       | 88        |
| 7     | R1 +R2  | 20            | 6        | 60        |
| 8     | R1 +R2  | 20            | 12       | 72        |
| 9     | R1 +R2  | 20            | 18       | 87        |
| 10    | R1 +R2  | 20            | 24       | 88        |

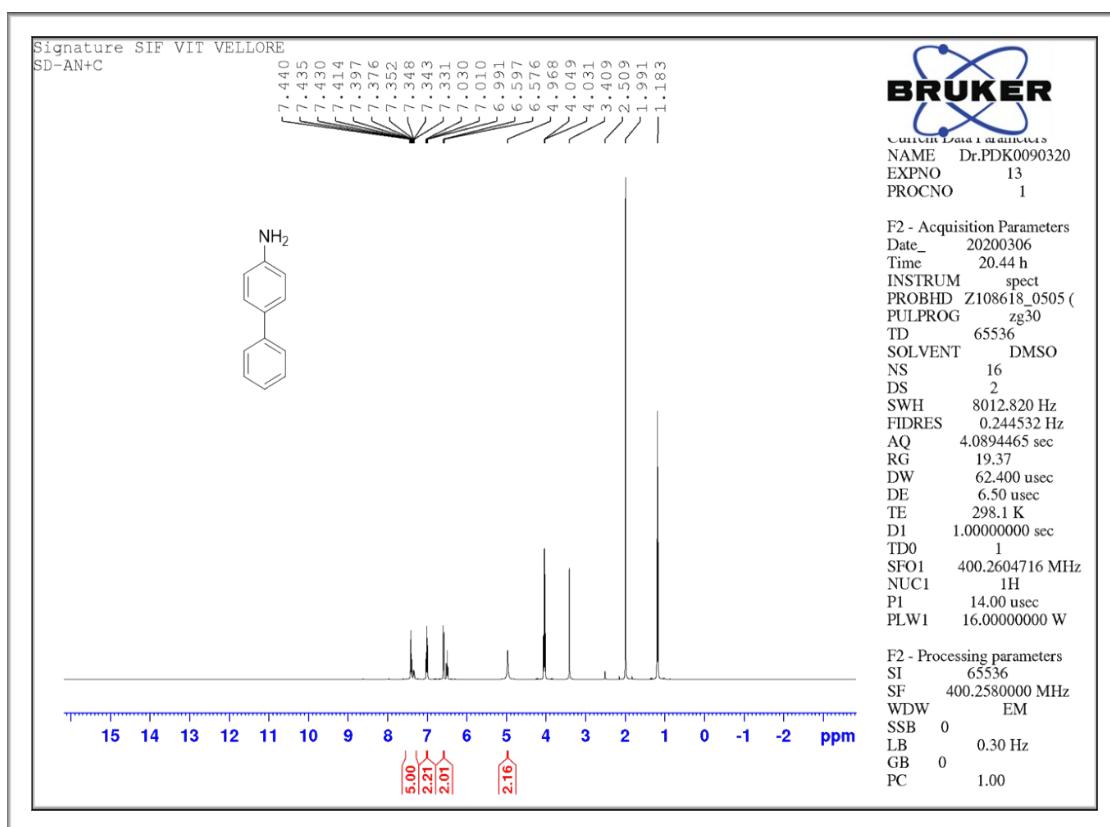
## C-H Bond Activation



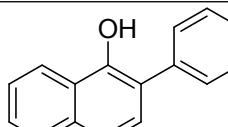
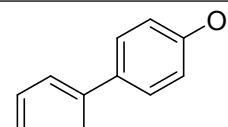
**Fig. S58**  $^1\text{H}$  NMR Spectrum of 2-phenylnaphthalen-1-ol



**Fig. S59**  $^1\text{H}$  NMR Spectrum of 1,1'-biphenyl-4-ol



**Fig. S60**  $^1\text{H}$  NMR Spectrum of 1,1'-biphenyl-4-amine

| Carbon–hydrogen activation Product  | bond | Conformation Data (NMR data)  |
|---|------|---|
|  |      | 400 MHz, DMSO-d6: $\delta$ 10.11 (s, 1H), 8.11 (d, 2H), 7.80 (d, 1H), 7.44 (t, 6H), 7.30 (t, 2H). |
|  |      | 400 MHz, DMSO-d6: $\delta$ 0.00 (s, 1H), 7.32-7.34 (m, 5H), 0.00 (d, 2H), 0.00 (d, 2H).           |

**Table S6.**  $^1\text{H}$  value CH between

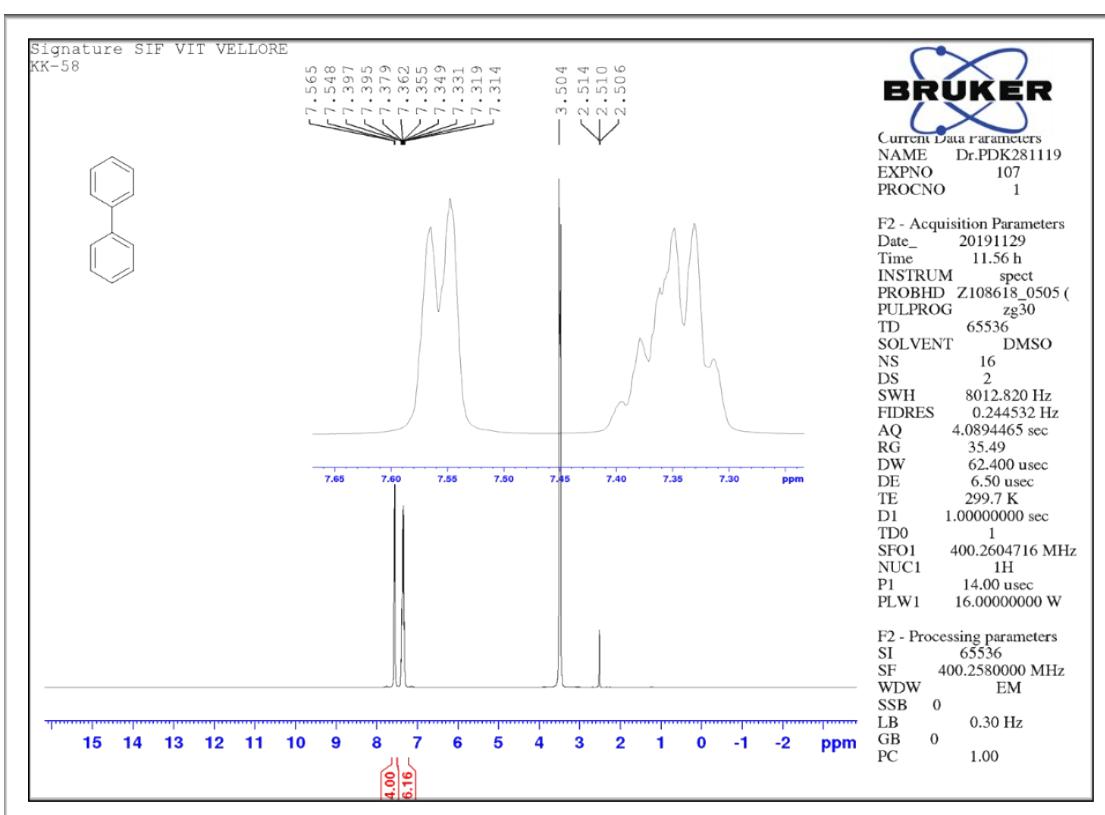
and  $^{13}\text{C}$  activation various Aryl

halide and phenolic compound under Photocatalytic Conditions.

**Table S7. Optimization of Carbon–hydrogen bond activation between 4-chloroaniline and 1-naphthol under Photocatalytic Conditions Catalysed by Photo-catalyst**

| Entry | Reagent | Catalyst (mg) | Time (h) | Yield (%) |
|-------|---------|---------------|----------|-----------|
| 1     | R1 +R2  | 5             | 25       | 49        |
| 2     | R1 +R2  | 10            | 25       | 57        |
| 3     | R1 +R2  | 15            | 25       | 69        |
| 4     | R1 +R2  | 20            | 25       | 77        |
| 5     | R1 +R2  | 25            | 25       | 77        |
| 6     | R1 +R2  | 20            | 5        | 54        |
| 7     | R1 +R2  | 20            | 10       | 61        |
| 8     | R1 +R2  | 20            | 15       | 67        |
| 9     | R1 +R2  | 20            | 20       | 76        |
| 10    | R1 +R2  | 20            | 25       | 77        |

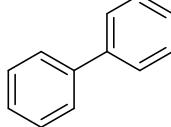
## Ullmann Coupling



**Fig. S61**  $^1\text{H}$  NMR Spectrum of 1,1-biphenyl

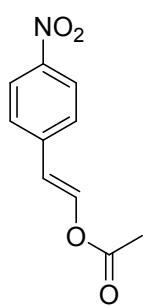
400 MHz, DMSO-d6:  $\delta$  7.55 (d, 4H), 7.31-7.33 (m, 6H)

**Table S8.  $^1\text{H}$  and  $^{13}\text{C}$  value for Ullmann Coupling**

| Ullmann Product   | Coupling | Conformation Data (NMR data)  |
|---|----------|---|
|  |          | 400 MHz, DMSO-d6: $\delta$ 7.55 (d, $J$ = 8.00 Hz, 4H), 7.31-7.33 (m, 6H) |

**Table 9. Optimization of Ullmann Coupling between 4-chloroaniline under Photocatalytic Conditions Catalysed by Photo-catalyst**

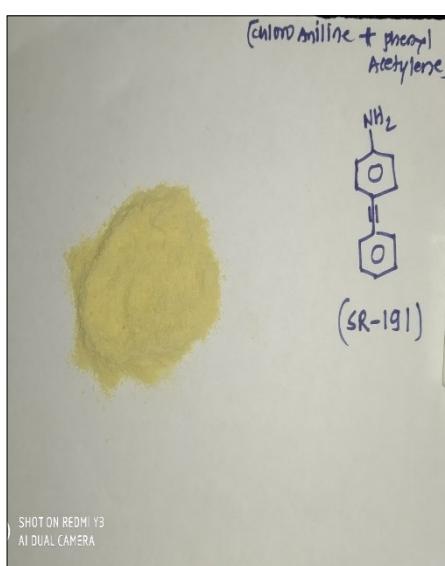
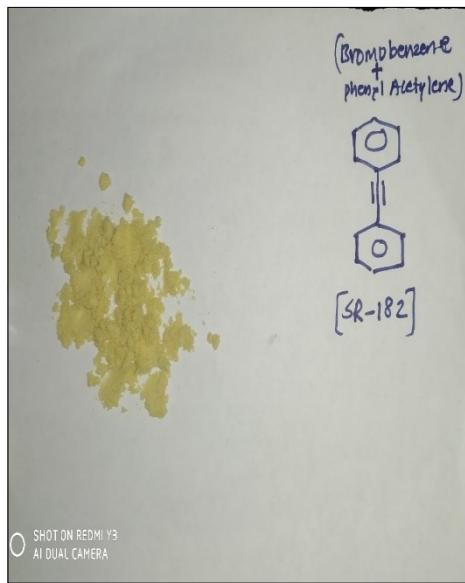
| Entry | Reagent | Catalyst (mg) | Time (h) | Yield (%) |
|-------|---------|---------------|----------|-----------|
| 1     | R1 +R2  | 5             | 24       | 57        |
| 3     | R1 +R2  | 10            | 24       | 74        |
| 4     | R1 +R2  | 15            | 24       | 88        |
| 5     | R1 +R2  | 20            | 24       | 91        |
| 6     | R1 +R2  | 25            | 24       | 91        |
| 7     | R1 +R2  | 20            | 8        | 74        |
| 8     | R1 +R2  | 20            | 12       | 81        |
| 9     | R1 +R2  | 20            | 16       | 91        |
| 10    | R1 +R2  | 20            | 24       | 91        |

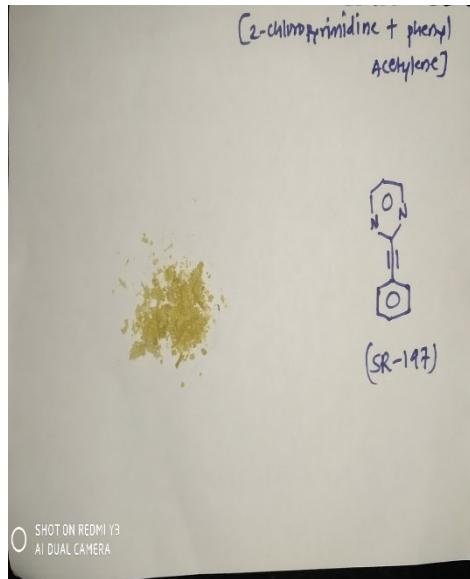
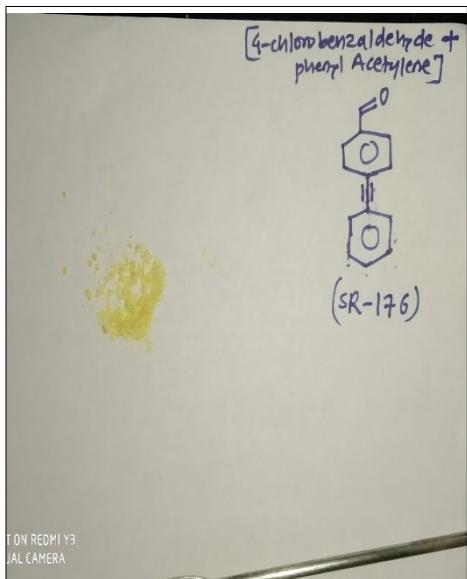


(*E*)-4-nitrostyryl acetate



### Other derivatives of heck coupling





SHOT ON REDMI Y3  
AI DUAL CAMERA



**During recrystallization of Heck coupling product**