

## Supplementary Information for:

# The Equilibrium Potentials of Ni-Ln alloys over the whole Composition Range in Phase Diagram– Experiment and Prediction

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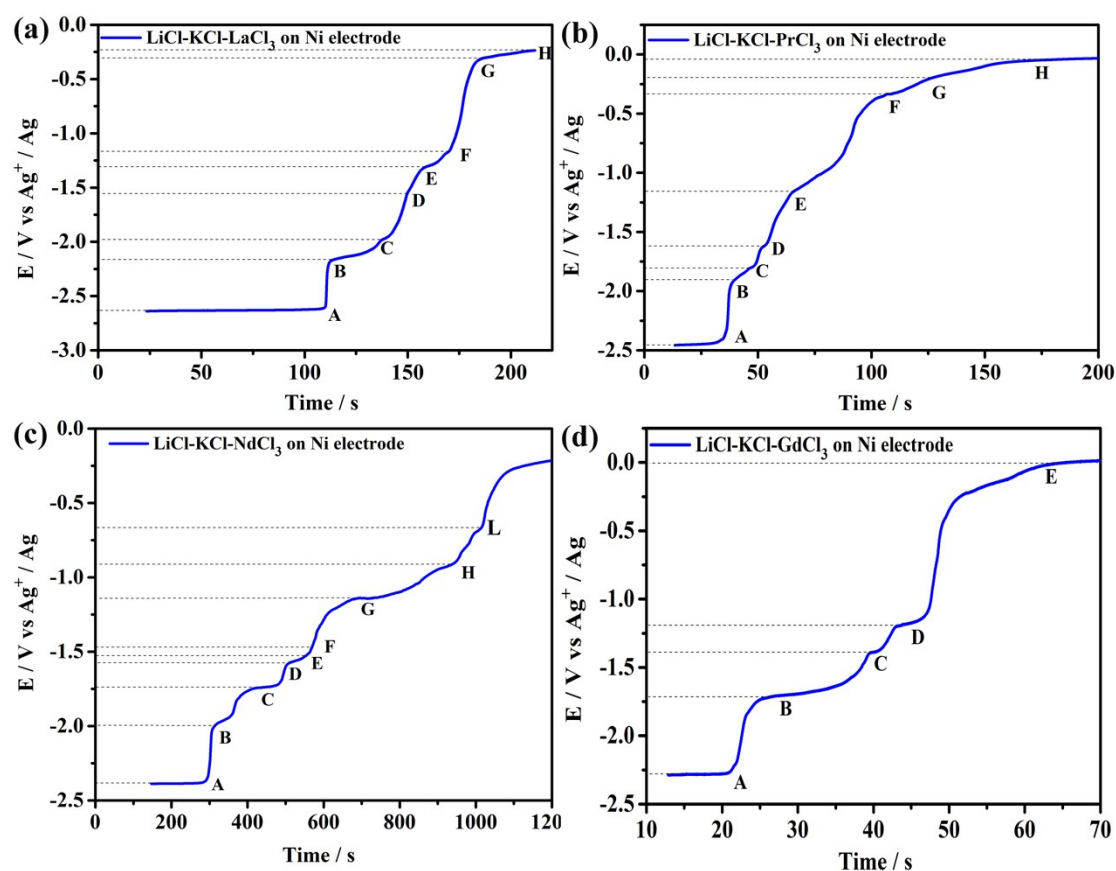
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**Fig. S1** (a) Open circuit chronopotentiometry curves obtained on a Ni electrode after potentiostatic electrolysis at -2.5 V (vs.  $Ag/Ag^+$ ) for 10 s in  $LiCl-KCl-LaCl_3$  (2.0 wt.%) melt. Temperature: 873

K. (b) Open circuit chronopotentiometry curves obtained on a Ni electrode after potentiostatic electrolysis at -2.5 V (vs. Ag/Ag<sup>+</sup>) for 10 s in LiCl-KCl-PrCl<sub>3</sub> (2.0 wt.%) melt. Temperature: 873

K. (c) Open circuit chronopotentiometry curves obtained on a Ni electrode after potentiostatic electrolysis at -2.5 V (vs. Ag/Ag<sup>+</sup>) for 10 s in LiCl-KCl-NdCl<sub>3</sub> (2.0 wt.%) melt. Temperature: 873

K. (d) Open circuit chronopotentiometry curves obtained on a Ni electrode after potentiostatic electrolysis at -2.5 V (vs. Ag/Ag<sup>+</sup>) for 10 s in LiCl-KCl-GdCl<sub>3</sub> (2.0 wt.%) melt. Temperature: 873

K.

**Table S1** Summarizes the relationship between the types of alloy compounds with different La atom percentages and the equilibrium potential.

773K	La	La <sub>3</sub> Ni	La <sub>7</sub> Ni <sub>3</sub>	LaNi	La <sub>2</sub> Ni <sub>3</sub>	LaNi <sub>2</sub>	LaNi <sub>5</sub>
Composition/%	100	75	70	50	40	33.3	16.8
Experimental value /V	-2.633	-2.16	-1.978	-1.554	-1.307	-1.165	-0.305

**Table S2** Summarizes the relationship between the types of alloy compounds with different Nd atom percentages and the equilibrium potential.

773K	Nd	Nd <sub>7</sub> Ni <sub>3</sub>	NdNi	NdNi <sub>2</sub>	NdNi <sub>3</sub>	Nd <sub>2</sub> Ni <sub>7</sub>	NdNi <sub>5</sub>	Nd <sub>2</sub> Ni <sub>17</sub>
Composition/%	100	70	50	33.3	25	22.2	16.66	10.526
Experimental value /V	-1.951	-1.737	-1.6	-1.36	-1.256	-1.125	-0.892	-0.684

**Table S3** Summarizes the relationship between the types of alloy compounds with different Pr atom percentages and the equilibrium potential.

773K	Pr	Pr <sub>3</sub> Ni	Pr <sub>7</sub> Ni <sub>3</sub>	PrNi	Pr <sub>2</sub> Ni <sub>7</sub>	PrNi <sub>5</sub>
Composition/%	100	75	70	50	22.3	16.8

Experimental	-2.449	-1.941	-1.804	-1.619	-0.332	-0.2
value /V						

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**Table S4** Summarizes the relationship between the types of alloy compounds with different Gd atom percentages and the equilibrium potential.

773K	Gd	Gd <sub>3</sub> Ni	Gd <sub>3</sub> Ni <sub>2</sub>	GdNi	Gd <sub>2</sub> Ni <sub>7</sub>
Composition/%	100	75	60	50	22.3
Experimental	-2.276	-1.71	-1.385	-1.196	-0.12
value /V					

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