

## Supporting information

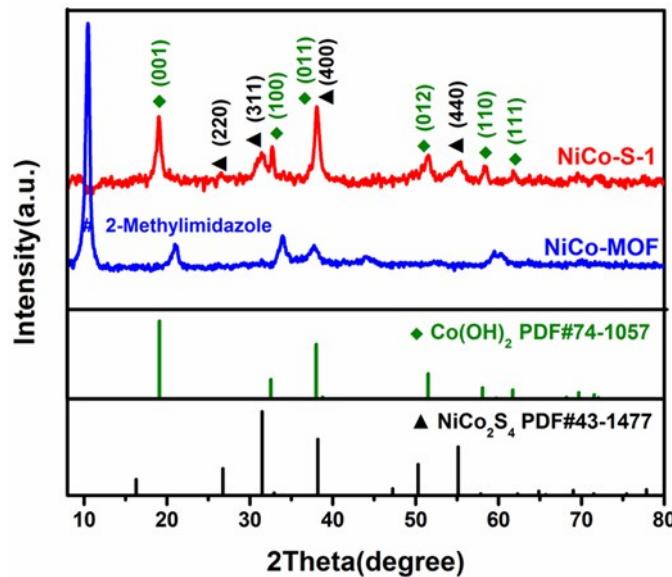
### Construction of self-supported hierarchical bimetallic sulfide nanosheet arrays for supercapacitors with ultrahigh specific capacitance

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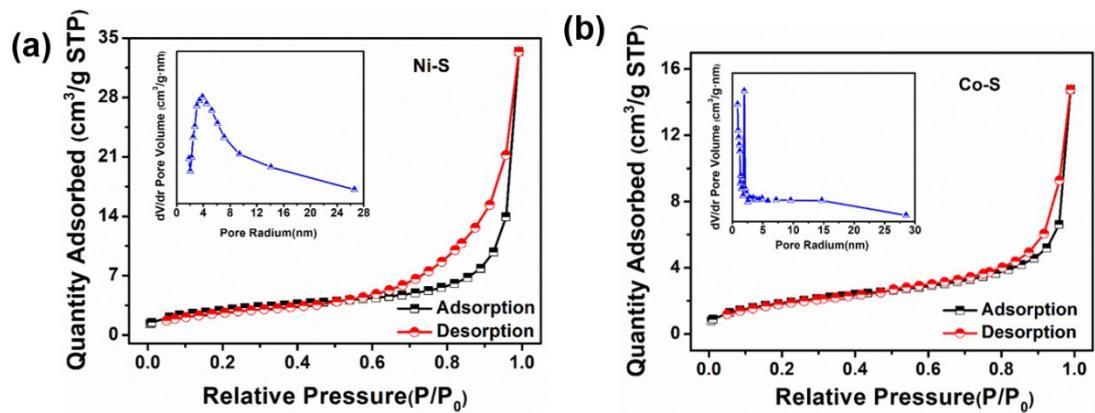
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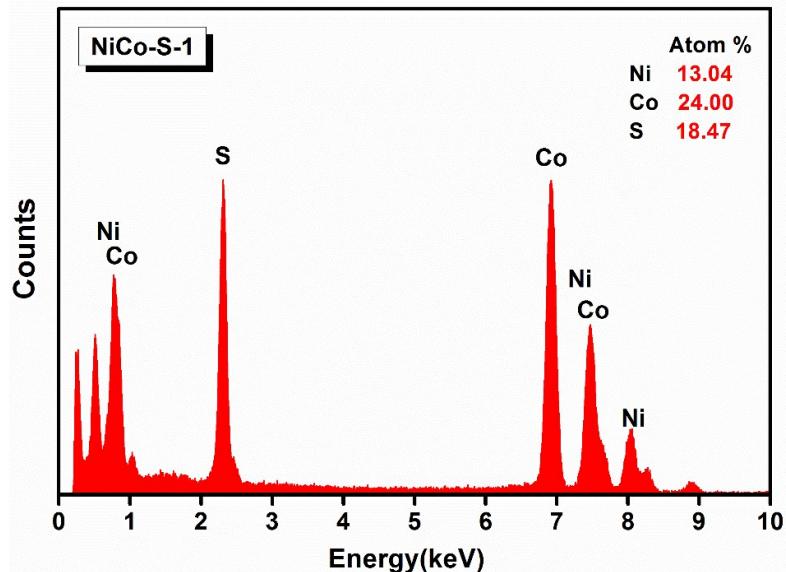
E-mail: [zhenghj@zjut.edu.cn](mailto:zhenghj@zjut.edu.cn) (H. Zheng)



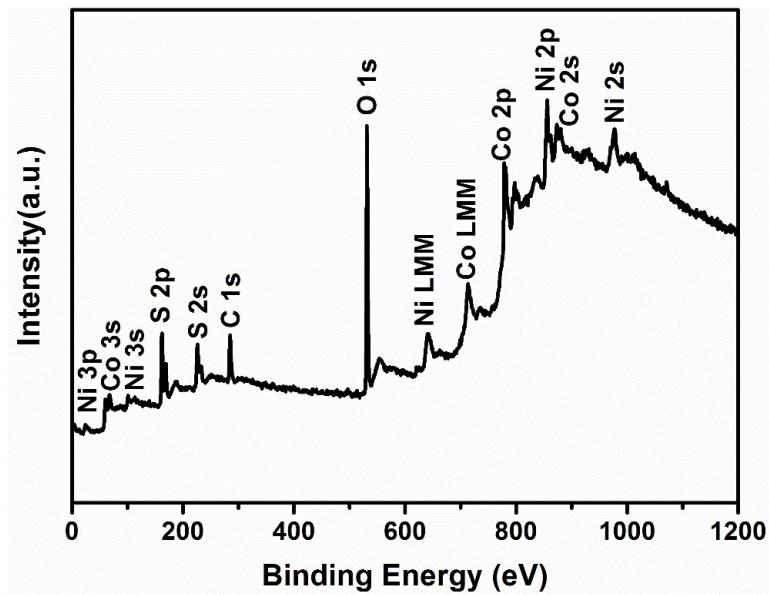
**Fig. S1.** XRD patterns of samples without Ni foam.



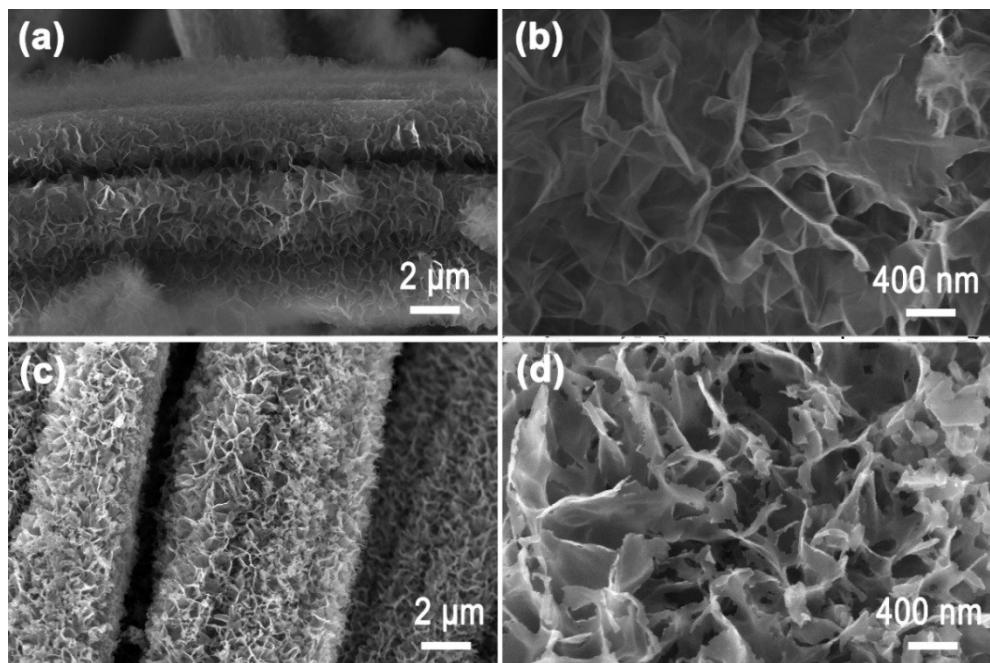
**Fig. S2.** N<sub>2</sub> adsorption-desorption isotherms and the corresponding pore size distribution of (a) Ni-S and (b) Co-S.



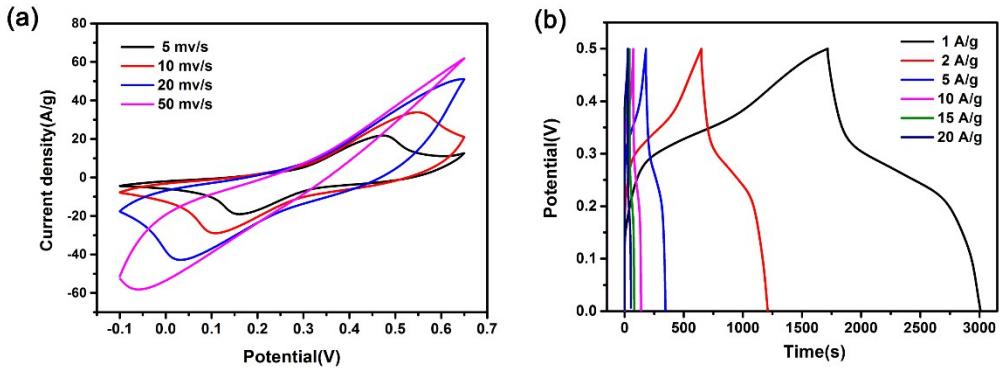
**Fig. S3.** EDS spectrum of the NiCo-S-1 sample.



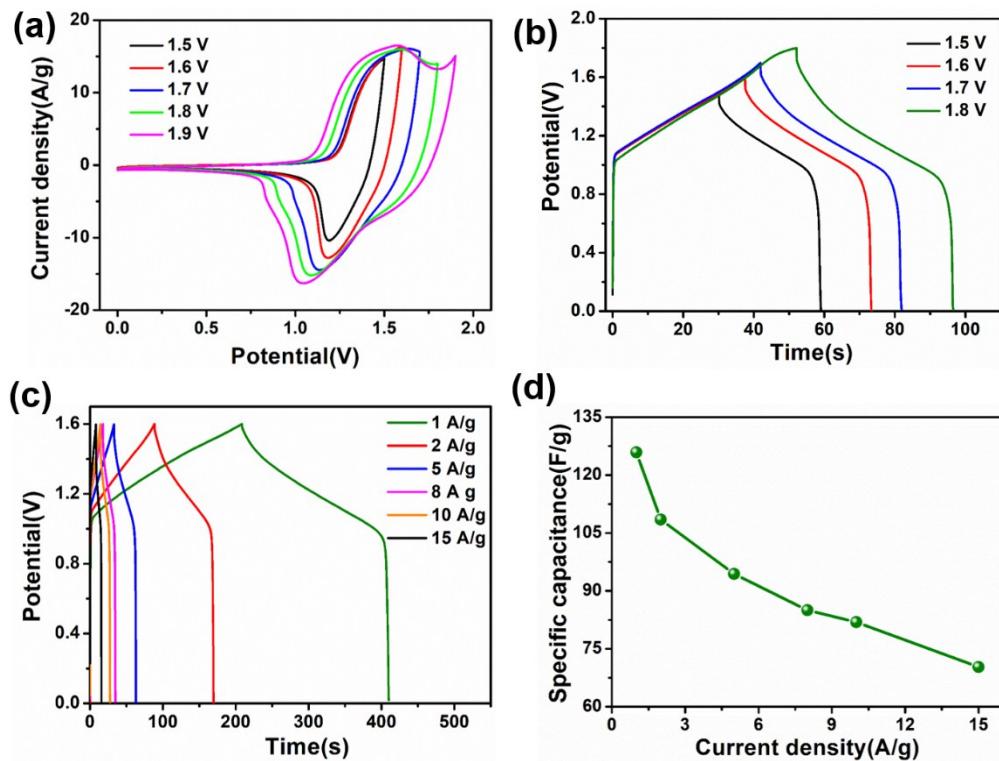
**Fig. S4.** The XPS full-scan spectrum of NiCo-S-1.



**Fig. S5.** SEM images of samples (a,b) NiCo-MOF@CC and (c,d) NiCo-S@CC.



**Fig. S6.** (a) CV curves at different densities and (b) GCD curves at different current densities of NiCo-S-1 nanosheet arrays on carbon cloth.



**Fig. S7.** The electrochemical performance of NiCo-S-1//AC asymmetric supercapacitor. (a) CV curves within different potential windows at scan rate of 50 mV s<sup>-1</sup>, (b) GCD curves within different potential windows at 5 A g<sup>-1</sup>, (c) GCD curves at different current densities, (d) specific capacitance at different current densities.

**Table S1.** Summary of the electrochemical performance on the related electrodes for supercapacitors in literatures.

Electrode materials	Electrolyte	Specific capacitance	Energy density, kg <sup>-1</sup>	Power density, kW	ref.
CuCo <sub>2</sub> S <sub>4</sub> -HNN//AC	3 M KOH	2163 F g <sup>-1</sup> @ 6 mA cm <sup>-2</sup>	44.1 W h kg <sup>-1</sup> , 0.80	0.80 kW <sup>1</sup>	
Zn-Co-S//AC	1 M KOH	2354.3 F g <sup>-1</sup> @ 0.5 A g <sup>-1</sup>	31.9 W h kg <sup>-1</sup> , 0.85	0.85 kW <sup>2</sup>	
Ni-Co-S//AC	1 M KOH	1406.9 F g <sup>-1</sup> , 0.5 A g <sup>-1</sup>	24.8 W h kg <sup>-1</sup> , 0.850	0.850 kW <sup>3</sup>	
Ni-Co-S//AC	6 M KOH	2392 F g <sup>-1</sup> , 1 A g <sup>-1</sup>	30.1 W h kg <sup>-1</sup> , 0.800	0.800 kW <sup>4</sup>	
NiCo <sub>2</sub> S <sub>4</sub> //AC	3 M KOH	1956 F g <sup>-1</sup> , 1 A g <sup>-1</sup>	27.5 W h kg <sup>-1</sup> , 0.747	0.747 kW <sup>5</sup>	
NiV <sub>2</sub> S <sub>4</sub> //AC	6 M KOH	639 C g <sup>-1</sup> , 2 mA cm <sup>-2</sup>	45.1 W h kg <sup>-1</sup> , 0.240	0.240 kW <sup>6</sup>	
Ni-MOF //AC	3 M KOH	1057 F g <sup>-1</sup> , 1 A g <sup>-1</sup>	21.05 W h kg <sup>-1</sup> , 6.03	6.03 kW <sup>7</sup>	
Ni-Co-S//AC	3 M KOH	1377.5 F g <sup>-1</sup> , 1 A g <sup>-1</sup>	36.9 W h kg <sup>-1</sup> , 1.066	1.066 kW <sup>8</sup>	
NiCo <sub>2</sub> S <sub>4</sub> //AC	3 M KOH	3724 F g <sup>-1</sup> , 1 A g <sup>-1</sup>	44.76 W h kg <sup>-1</sup> , 0.80	0.80 kW <sup>this work</sup>	

**Table S2.** The atomic content of Ni, Co, S in the sample NiCo-S-1.

Element	Ni	Co	S
Content(%)	12.56	29.86	22.05

1. S. E. Moosavifard, S. Fani and M. Rahamanian, *Chem. Commun.*, 2016, **52**, 4517-4520.
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3. M. K. Wu, C. Chen, J. J. Zhou, F. Y. Yi, K. Tao and L. Han, *J. Alloy. Comp.*, 2018, **734**, 1-8.
4. W. Zhao, Y. Zheng, L. Cui, D. Jia, D. Wei, R. Zheng, C. Barrow, W. Yang and J. Liu, *Chem. Eng. J.*, 2019, **371**, 461-469.
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7. P. Du, Y. Dong, C. Liu, W. Wei, D. Liu and P. Liu, *J. Colloid Interface Sci.*, 2018, **518**, 57-68.
8. C. Chen, M. K. Wu, K. Tao, J. J. Zhou, Y. L. Li, X. Han and L. Han, *Dalton T.*, 2018, **47**, 5639-5645.