

Supporting Information

Arbitrary deformable and high-strength electroactive polymer/ MXene anti-exfoliative composite films assembled into high performance flexible all-solid-state supercapacitors

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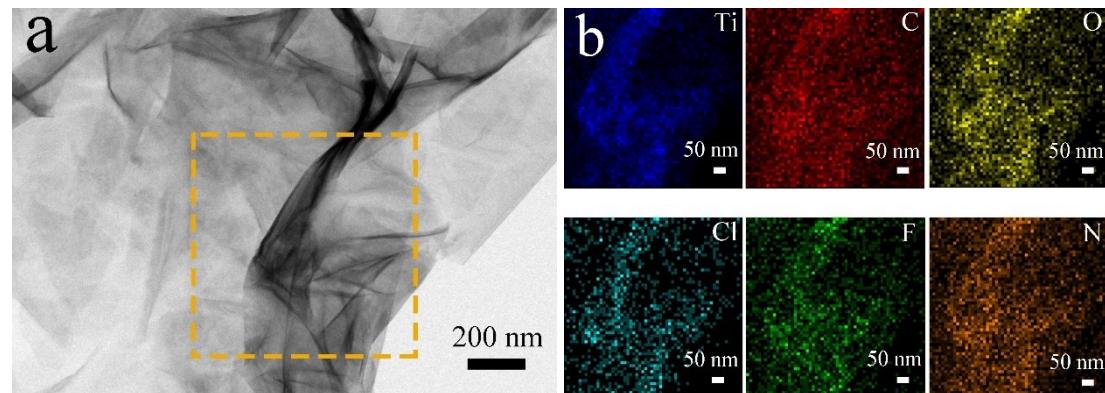


Fig. S1 (a) STEM image and (b) Elemental mapping images of i-PANI@Ti₃C₂T_x.

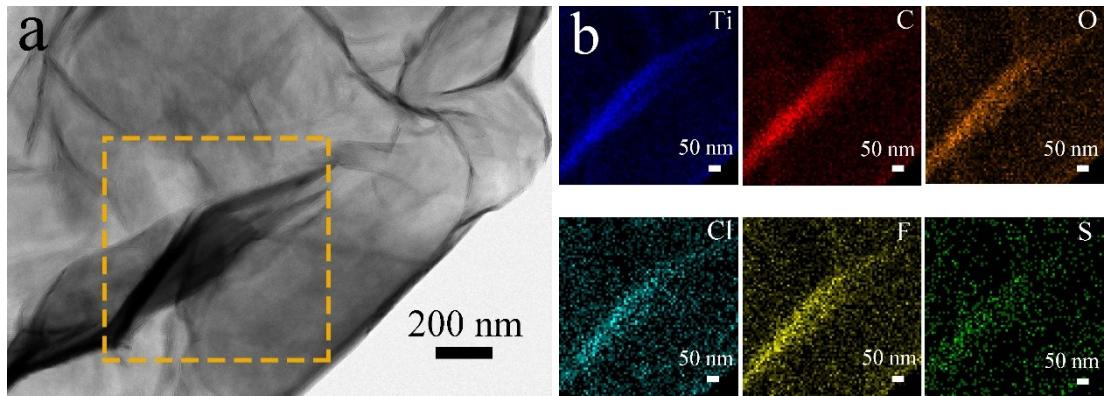


Fig. S2 (a) STEM image and (b) Elemental mapping images of Lig@ $\text{Ti}_3\text{C}_2\text{T}_x$.

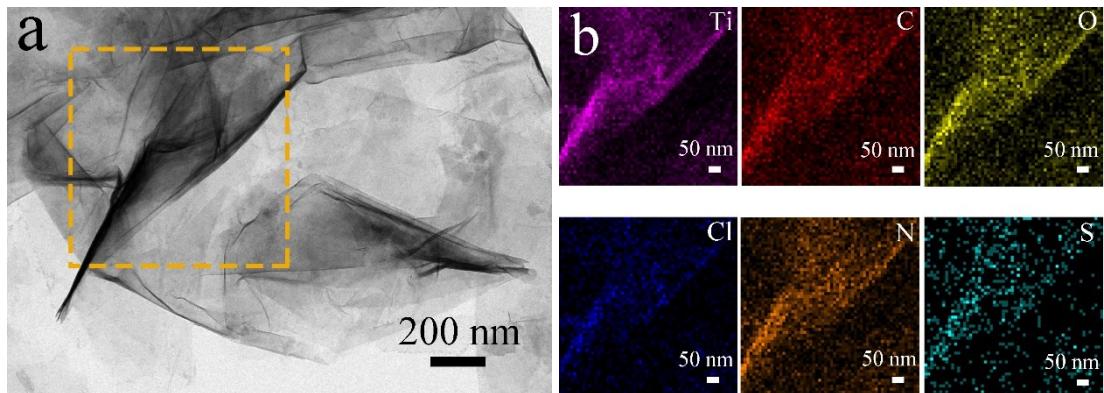
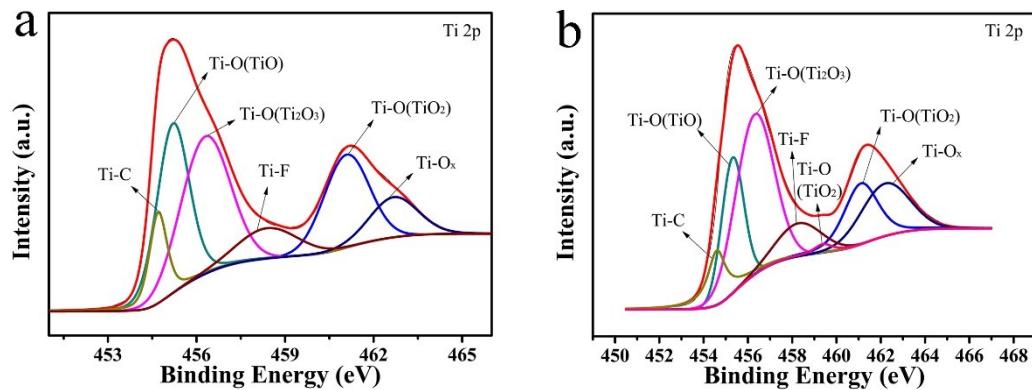


Fig. S3 (a) STEM image and (b) Elemental mapping images of Lig@ $\text{Ti}_3\text{C}_2\text{T}_x$ /i-PANI@ $\text{Ti}_3\text{C}_2\text{T}_x(5/5)$.



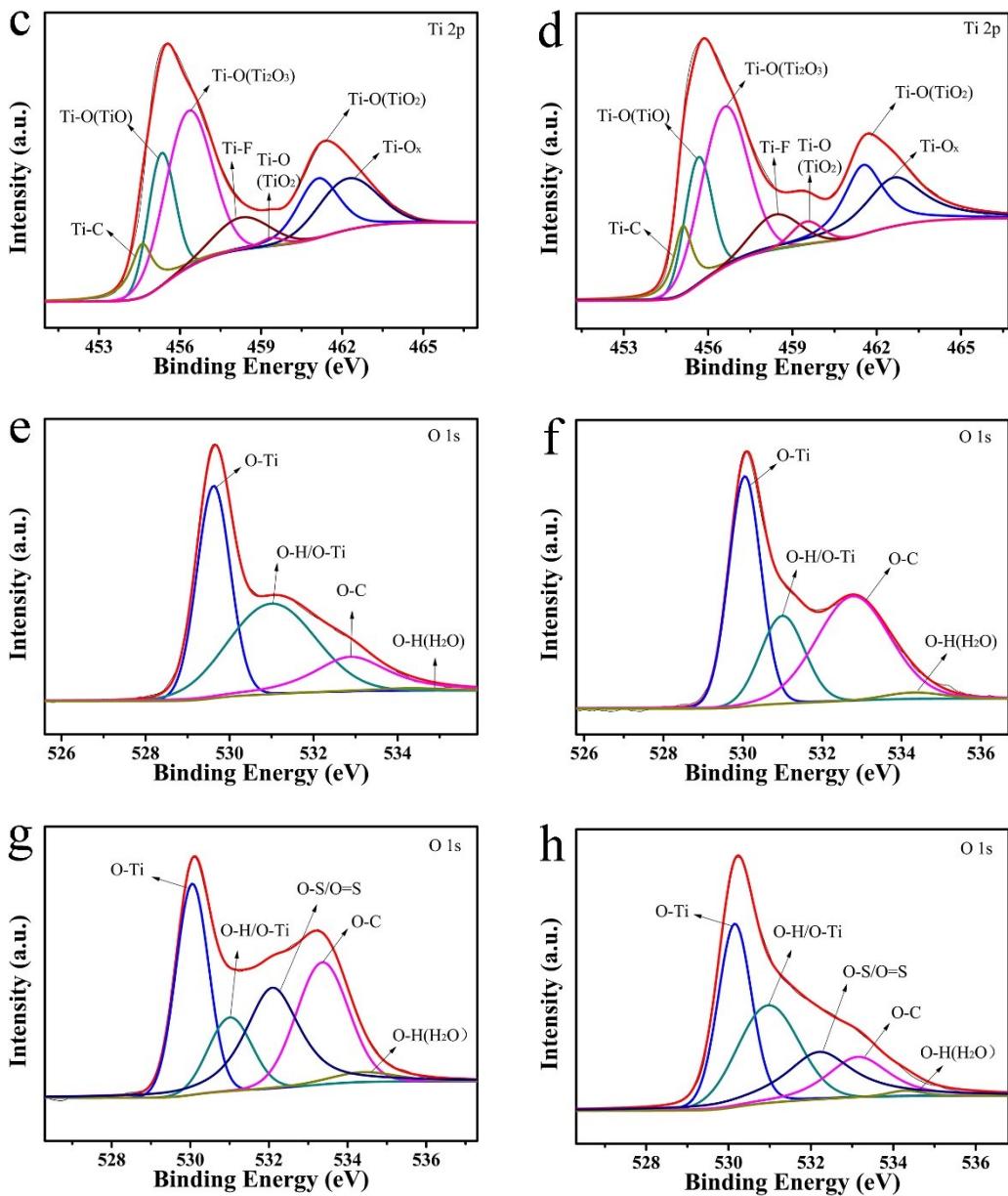


Fig. S4 The Ti 2p spectra of (a) $\text{Ti}_3\text{C}_2\text{T}_x$, (b) i-PANI@ $\text{Ti}_3\text{C}_2\text{T}_x$, (c) Lig@ $\text{Ti}_3\text{C}_2\text{T}_x$ and (d) Lig@ $\text{Ti}_3\text{C}_2\text{T}_x$ /i-PANI@ $\text{Ti}_3\text{C}_2\text{T}_x$ (5/5). The O 1s spectra of (e) $\text{Ti}_3\text{C}_2\text{T}_x$, (f) i-PANI@ $\text{Ti}_3\text{C}_2\text{T}_x$, (g) Lig@ $\text{Ti}_3\text{C}_2\text{T}_x$ and (h) Lig@ $\text{Ti}_3\text{C}_2\text{T}_x$ /i-PANI@ $\text{Ti}_3\text{C}_2\text{T}_x$ (5/5).

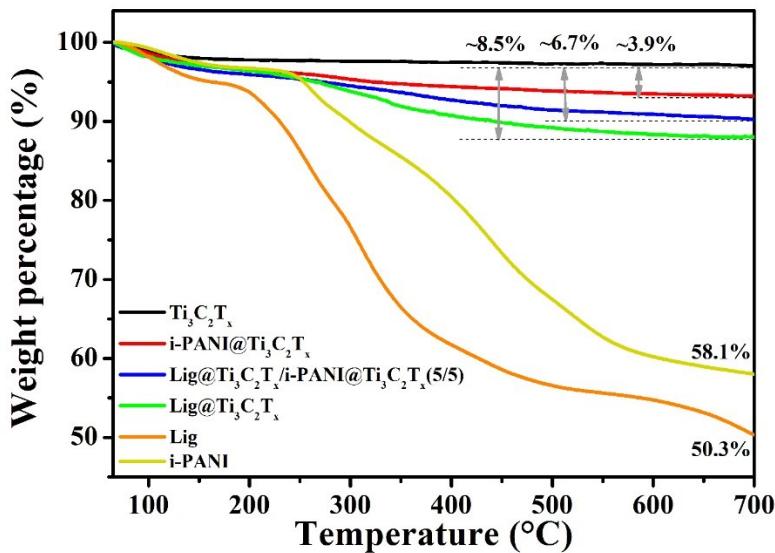
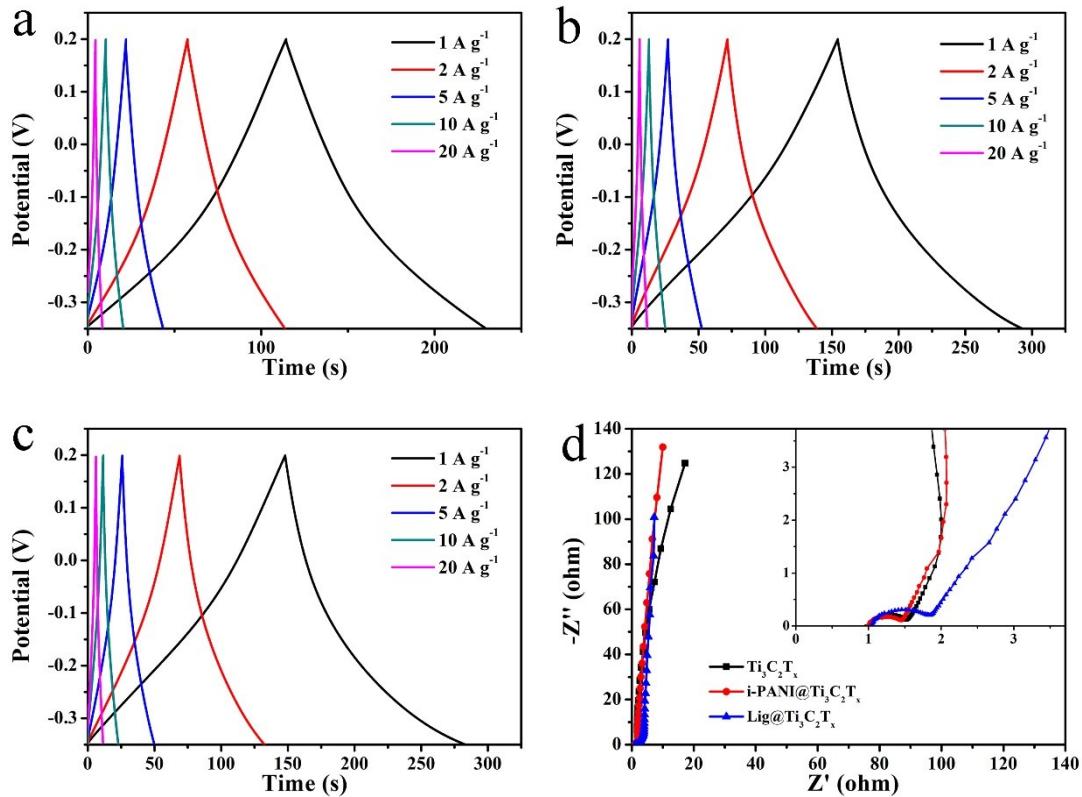


Fig. S5 The TGA curves of $\text{Ti}_3\text{C}_2\text{T}_x$, i-PANI (prepared using PPD as initiator and FeCl_3 as oxidant with a molar ratio of aniline:PPD=10:1 and aniline: FeCl_3 =1:1), Lig, i-PANI@ $\text{Ti}_3\text{C}_2\text{T}_x$, Lig@ $\text{Ti}_3\text{C}_2\text{T}_x$ and Lig@ $\text{Ti}_3\text{C}_2\text{T}_x$ /i-PANI@ $\text{Ti}_3\text{C}_2\text{T}_x$ (5/5).



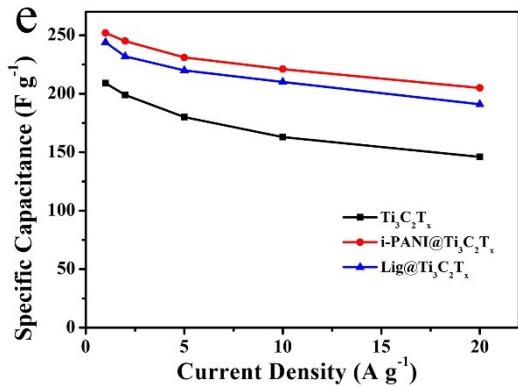


Fig. S6 Electrochemical properties of $\text{Ti}_3\text{C}_2\text{T}_x$, i-PANI@ $\text{Ti}_3\text{C}_2\text{T}_x$ and Lig@ $\text{Ti}_3\text{C}_2\text{T}_x$ films in three-electrode system. GCD curves of (a) $\text{Ti}_3\text{C}_2\text{T}_x$, (b) i-PANI@ $\text{Ti}_3\text{C}_2\text{T}_x$ and (c) Lig@ $\text{Ti}_3\text{C}_2\text{T}_x$ films in a current density range of 1 to 20 A g^{-1} , (d) Nyquist plots in a frequency range of 100 kHz to 0.01 Hz and (e) Specific capacitance at various current densities.

The electrochemical performance of i-PANI@ $\text{Ti}_3\text{C}_2\text{T}_x$ film in ASSSs with different mass ratio of aniline to $\text{Ti}_3\text{C}_2\text{T}_x$ is shown in Fig. S4. The introduction of i-PANI into $\text{Ti}_3\text{C}_2\text{T}_x$ largely enhances the specific capacitance of i-PANI@ $\text{Ti}_3\text{C}_2\text{T}_x$ composite films, attributed to the spacer effect and pseudocapacitance contribution of i-PANI. With increasing the mass ratio of i-PANI to $\text{Ti}_3\text{C}_2\text{T}_x$, the specific capacitance of i-PANI@ $\text{Ti}_3\text{C}_2\text{T}_x$ composite films increases. While excessive amount of i-PANI may decrease the electrochemical-active surface of $\text{Ti}_3\text{C}_2\text{T}_x$, leading to decreased specific capacitance. Therefore, the most appropriate mass ratio of aniline to $\text{Ti}_3\text{C}_2\text{T}_x$ is 2:1 (aniline:PPD = 10:1).

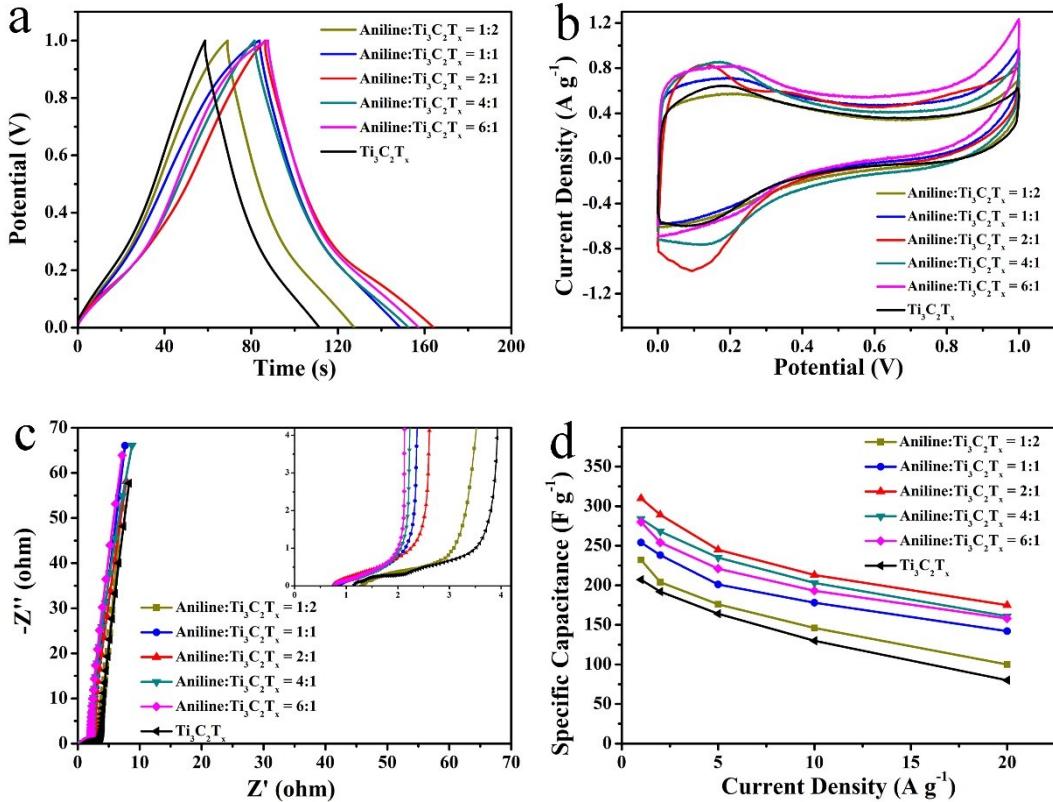


Fig. S7 Electrochemical properties of i-PANI@Ti₃C₂T_x assembled ASSSs with different mass ratio of aniline to Ti₃C₂T_x. (a) GCD curves at a current density of 1 A g⁻¹, (b) CV curves at a scan rate of 5 mV s⁻¹, (c) Nyquist plots in a frequency range of 100 kHz to 0.01 Hz, (d) Specific capacitance of i-PANI@Ti₃C₂T_x at various current densities

The influence of PPD is also investigated. The specific capacitance and rate capability of i-PANI@Ti₃C₂T_x are superior to that of PANI@Ti₃C₂T_x (Fig. S5). The PPD initiator may reduce oxidation potential of polymerization and adjust the structure of PANI (increasing the content of quinone), resulting in a higher rate and degree of polymerization. Therefore, the introduction of PPD can improve the electrochemical performance of i-PANI@Ti₃C₂T_x.

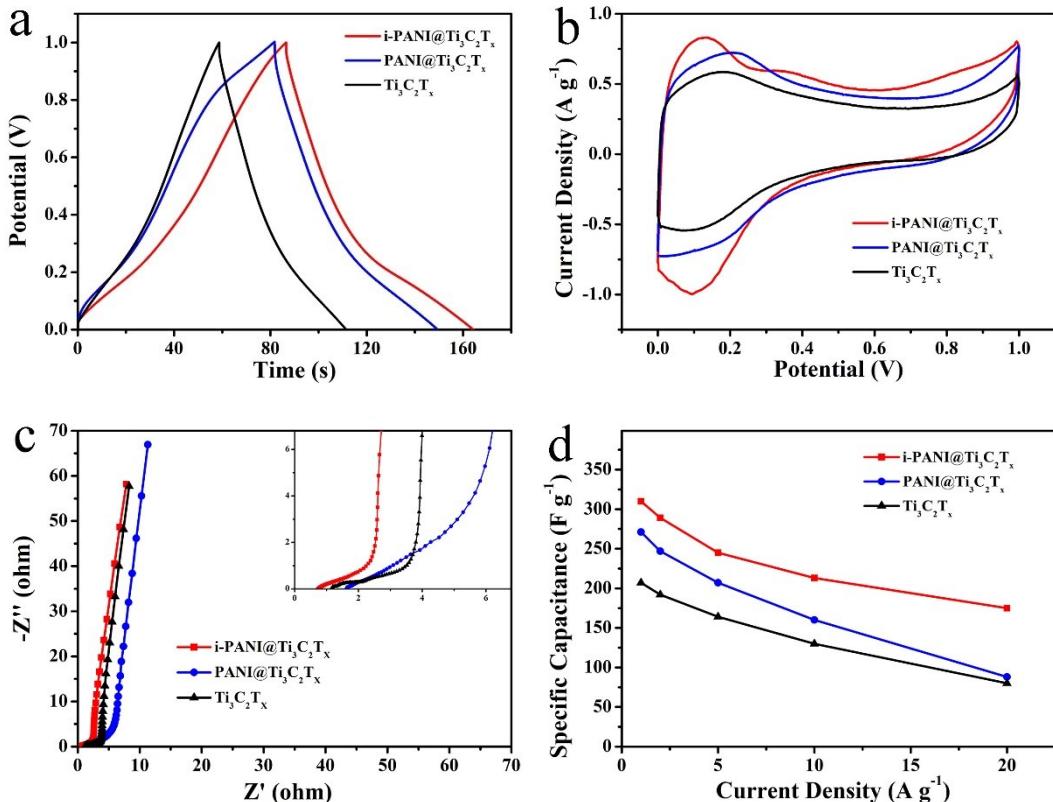


Fig. S8 Electrochemical properties of Ti₃C₂T_x, PANI@Ti₃C₂T_x and i-PANI@Ti₃C₂T_x assembled ASSSs (aniline:Ti₃C₂T_x = 2:1, aniline:PPD = 10:1) film. (a) GCD curves at a current density of 1 A g⁻¹, (b) CV curves at a scan rate of 5 mV s⁻¹, (c) Nyquist plots of the samples in a frequency range of 100 kHz to 0.01 Hz, (d) Specific capacitance at various current densities.

The electrochemical performance of Lig@Ti₃C₂T_x assembled ASSSs with different mass ratio of Lig to Ti₃C₂T_x is shown in Fig. S6. The optimized mass ratio of Lig to Ti₃C₂T_x is 2:1. Lig can effectively enhance the capacitance through redox reaction and promote the ion transfer by increasing the interlayer space. While a further increase of Lig would lead to the dramatic decrease of specific capacitance

and rate capability. It is suggested that the insulative nature of Lig and the MXene sheets wrapped by excessive Lig would lead to a deterioration of layer structure and decrease of active sites.

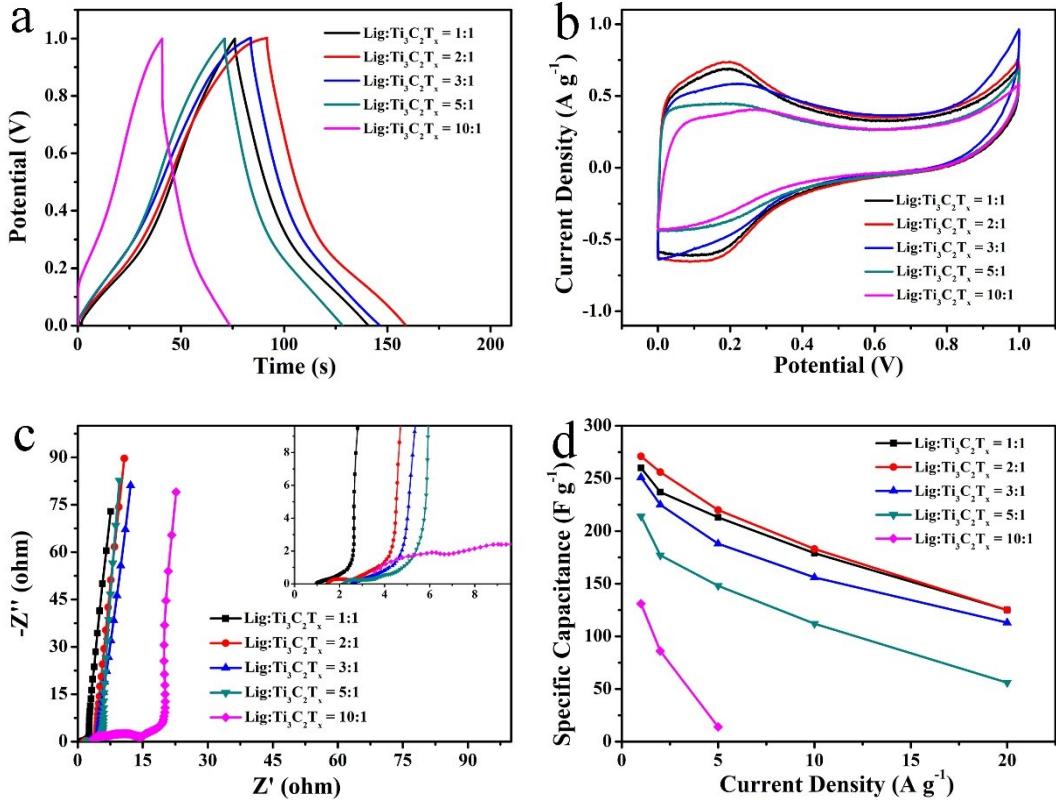


Fig. S9 Electrochemical properties of Lig@ $\text{Ti}_3\text{C}_2\text{T}_x$ assembled ASSSs with different mass ration of Lig to $\text{Ti}_3\text{C}_2\text{T}_x$. (a) GCD curves at a current density of 1 A g^{-1} , (b) CV curves at a scan rate of 5 mV s^{-1} , (c) Nyquist plots in a frequency range of 100 kHz to 0.01 Hz , (d) Specific capacitance at various current densities.

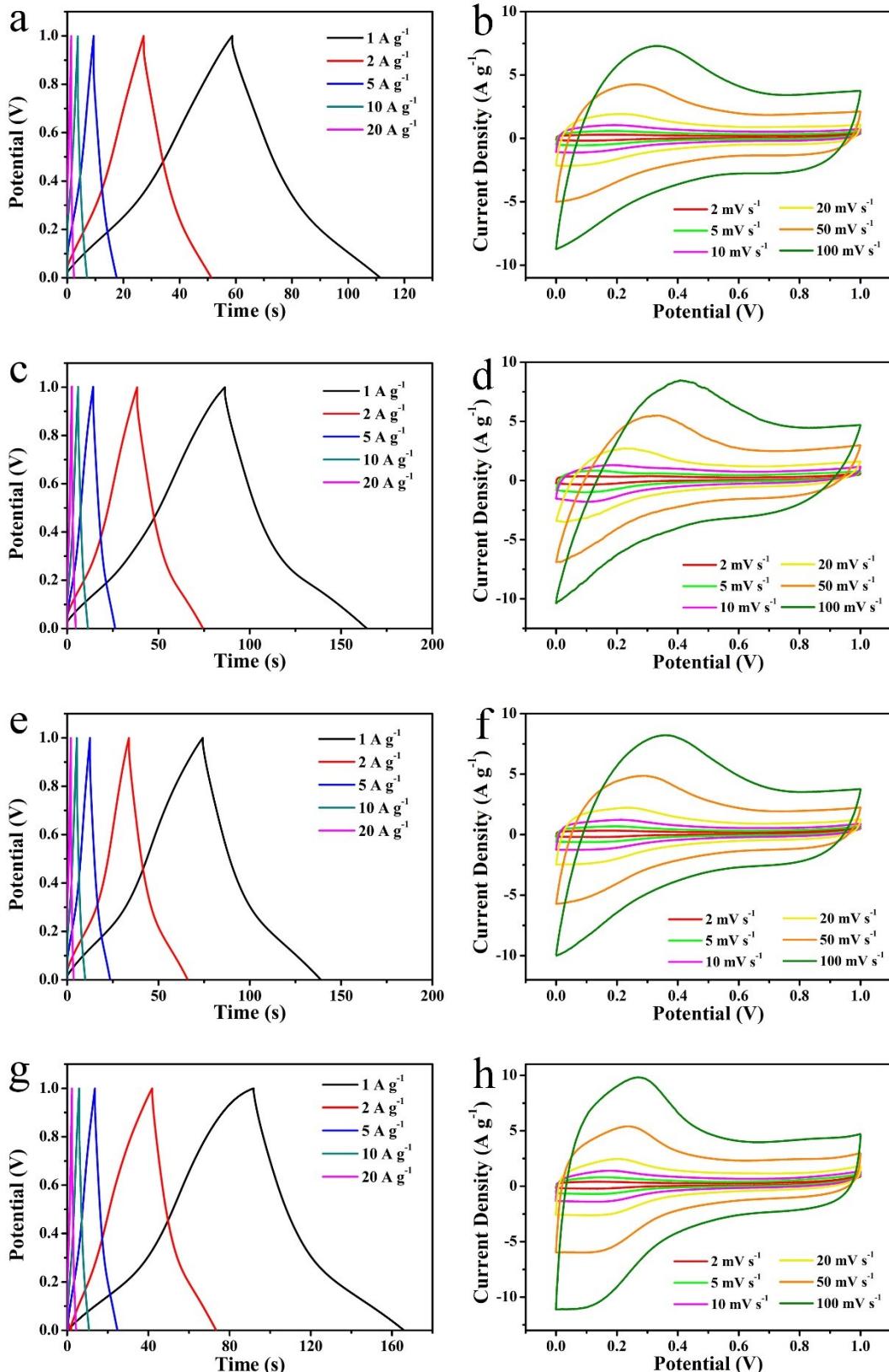


Fig. S10 Electrochemical properties of ASSSs. GCD curves of (a) $\text{Ti}_3\text{C}_2\text{T}_x$, (c) $\text{Lig}@\text{Ti}_3\text{C}_2\text{T}_x$, (e) i-PANI@ $\text{Ti}_3\text{C}_2\text{T}_x$ and (g) $\text{Lig}@\text{Ti}_3\text{C}_2\text{T}_x/\text{i-PANI}@\text{Ti}_3\text{C}_2\text{T}_x(5/5)$

films in a current density range of 1 to 20 A g⁻¹ and CV curves of (b) Ti₃C₂T_x, (d) Lig@Ti₃C₂T_x, (f) i-PANI@Ti₃C₂T_x and (h) Lig@Ti₃C₂T_x/i-PANI@Ti₃C₂T_x(5/5) films at scan rates ranging from 2 to 100 mV s⁻¹.

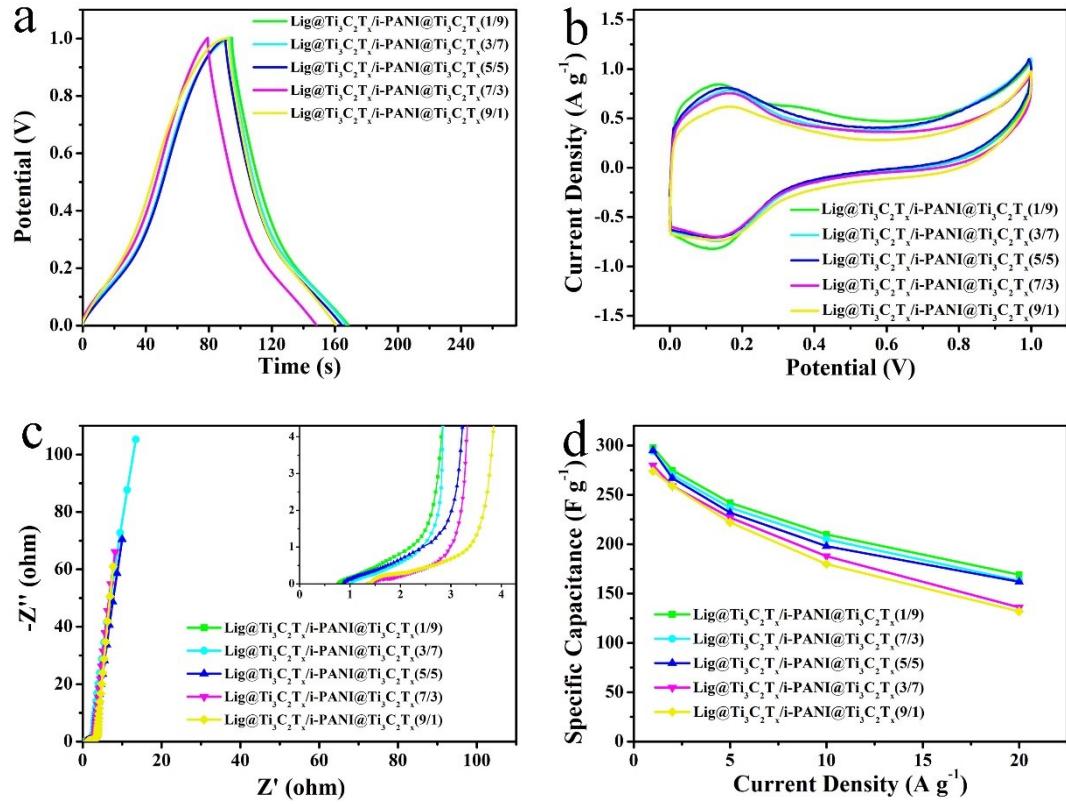


Fig. S11 Electrochemical properties of Lig@Ti₃C₂T_x/i-PANI@Ti₃C₂T_x assembled ASSSs with different mass ratio of Lig@Ti₃C₂T_x to i-PANI@Ti₃C₂T_x. (a) CV curves at a scan rate of 5 mV s⁻¹, (b) GCD curves at a current density of 1 A g⁻¹, (c) Nyquist plots of the samples in a frequency range of 100 kHz to 0.01 Hz and (d) Specific capacitance in various current densities.

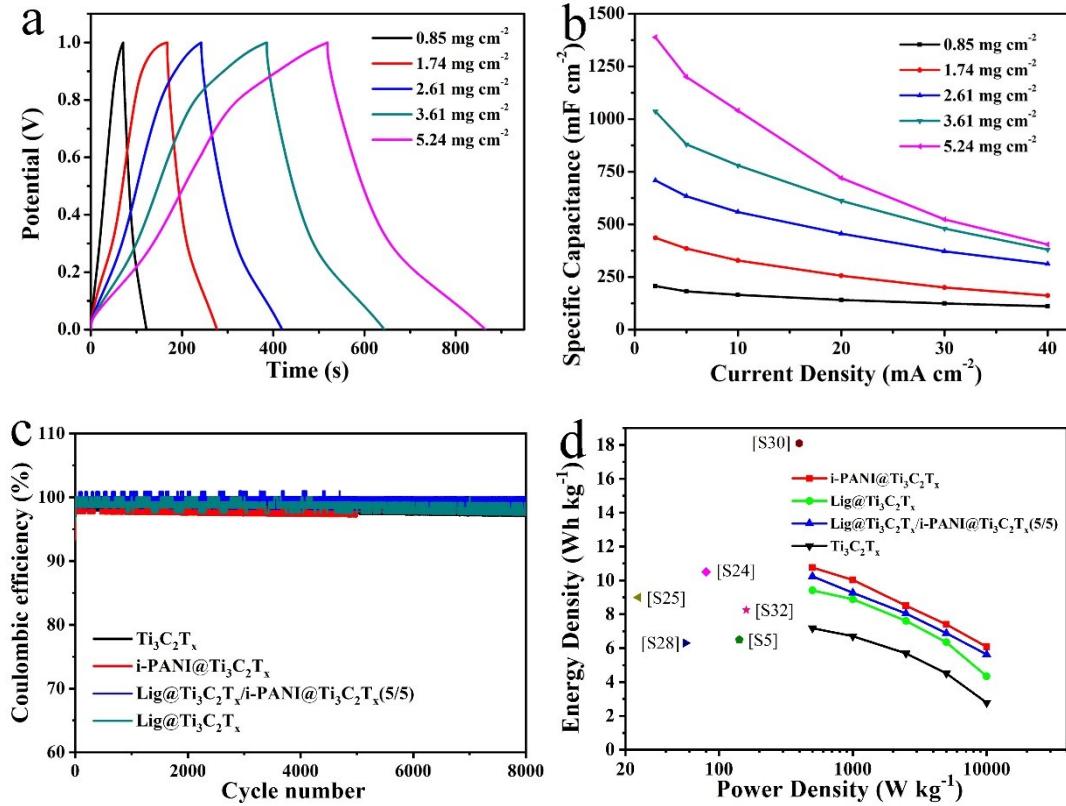


Fig. S12 Electrochemical properties of ASSSs. (a) GCD curves at a current density of 2 mA cm⁻² and (b) areal capacitance (current density ranging from 2 to 40 mA cm⁻²) of Lig@Ti₃C₂T_x/i-PANI@Ti₃C₂T_x(5/5) electrodes with different mass-loadings. (c) Coulombic efficiency during 8000 cycles of cycling test at a current density of 10 A g⁻¹. (d) Ragone plot of energy density comparing to those reported in the literature.

Table S1. The tensile strength of some composites.

Composite films	Tensile strength (MPa)	Ref.
Ti ₃ C ₂ T _x (90 wt%)/PVA film	30	S1

Ti ₃ C ₂ T _x (80 wt%)/PVA film	25	S1
Ti ₃ C ₂ T _x /rGO-90 film	12.9	S2
Ti ₃ C ₂ T _x /rGO-70 film	12.3	S2
CNTs (3 mg)/Ti ₃ C ₂ T _x (30 mg)/CNFs (8 mg) composite paper	98	S3
The MWCNT/Ti ₃ C ₂ T _x /PCL composite membrane	3	S4
Ti ₃ C ₂ T _x /PEDOT:PSS (mass ratios of 7:3) fibers	58.1	S5
PPy/RGO/CNT/bacterial cellulose film	57.7	S6
Ti ₃ C ₂ T _x /BC (mass ratios of 0.75:1) film	70	S7
i-PANI@Ti ₃ C ₂ T _x film	33.2	This work
Lig@Ti ₃ C ₂ T _x film	75.4	This work
Lig@Ti ₃ C ₂ T _x /i-PANI@Ti ₃ C ₂ T _x (5/5) film	53.7	This work

Table S2. The thermal diffusion coefficient, specific heat capacity, density and thermal conductivity of Ti₃C₂T_x, Lig@Ti₃C₂T_x, i-PANI@Ti₃C₂T_x, and Lig@Ti₃C₂T_x/i-PANI@Ti₃C₂T_x(5/5) at 25 °C.

MXene	Lig@Ti ₃ C ₂ T _x	i-PANI@Ti ₃ C ₂ T _x	PANI@Ti ₃ C ₂ T _x (5/5)	Lig@Ti ₃ C ₂ T _x /i-PANI@Ti ₃ C ₂ T _x (5/5)
Thermal diffusion coefficient (mm ² s ⁻¹)	0.47	0.42	0.32	0.33

Specific heat capacity (J g ⁻¹ °C ⁻¹)	1.182	1.265	1.112	1.283
Density (g cm ⁻³)	3.21	3.01	2.81	2.98
Thermal conductivity (W m ⁻¹ k ⁻¹)	1.78	1.60	1.01	1.26

Table S3. The through-plane thermal conductivity of some composite films.

Composites film	Through-plane thermal conductivity (W m ⁻¹ K ⁻¹)	Ref.□
hBN (hydroxylated boron nitride nanosheets)@PANI	2.1	S8
rGO-cellulose nanocrystals	4.596	S9
BN (boron nitride)/PDPA(poly(diallyl dimethyl ammonium chloride)	1.0	S10
Silicon rubber/graphene nanoplatelets/BN	0.80	S11
DOPO-g-GO (DOPO: 9, 10-dihydro-9-oxa-10-phosphaphenanthrene-10-oxide)	1.28	S12
30%-GO@PANI	0.42	S13
rGO-CNT	0.164	S14
15%-RGO@PANI	0.6	S15
hBN/polymethyl-vinyl siloxane rubber	1.11	S16

rGO-CNT	0.061	S17
PANI/graphene	19 $\mu\text{W m}^{-1} \text{K}^{-2}$	S18
50%-RGO/CNF	0.13	S19
PI (polyimide)/GO	0.32	S20
HDPE (high density polyethylene)/BN/MWCNTs/graphite	1.45	S21
PI/SiC/GO	0.577	S22
Ti ₃ C ₂ T _x /PVDF film	0.767	S23
i-PANI@Ti ₃ C ₂ T _x	1.01	This work
Lig@Ti ₃ C ₂ T _x	1.60	This work
Lig@Ti ₃ C ₂ T _x /i-PANI@Ti ₃ C ₂ T _x (5/5)	1.26	This work

Table S4. The capacitance of some supercapacitor electrodes.

Electrode material	Electrolyte	Specific capacitance	Ref.
Ti ₃ C ₂ T _x /rGO-5 wt% film	3 M H ₂ SO ₄	80.3 F g ⁻¹ at 2 mV s ⁻¹	two electrode system S24
MnO _x -Ti ₃ C ₂ T _x film	1 M Li ₂ SO ₄	602.0 F cm ⁻³ at 2 mV s ⁻¹	two electrode system S25
Nitrogen doped reduced graphene oxide foams (NrGFs)	6 M KOH	260 F g ⁻¹ at 0.1 A g ⁻¹	two electrode system S26
Ti ₃ C ₂ T _x /CuS composites//Ti ₃ C ₂ T _x MXene	1 M KOH	49.3 F g ⁻¹ at 1 A g ⁻¹	two electrode system S27
Ti ₃ C ₂ T _x / PEDOT:PSS //rGO film	1 M H ₂ SO ₄	117 F cm ⁻³ at 1.5 mA cm ⁻²	asymmetric, two electrode system S29
d-Ti ₃ C ₂ T _x /Ni foam//b-Ti ₃ C ₂ T _x film	6M KOH	51.1 F g ⁻¹ at 0.5 A g ⁻¹	asymmetric, two electrode system S30
N-Ti ₃ C ₂ T _x -300 film	3 M H ₂ SO ₄	\sim 150 F g ⁻¹ at 2mV s ⁻¹ 295 F cm ⁻³ at 2mV s ⁻¹	two electrode system S31
Active carbon/Co ₃ O ₄ -	6M KOH	55 F g ⁻¹	asymmetric, S32

doped 3D $\text{Ti}_3\text{C}_2\text{T}_x/\text{rGO}$ hybrid film		at 0.5 A g ⁻¹	two electrode system	
d- $\text{Ti}_3\text{C}_2\text{T}_x$ film under 40 MPa	1 M EMIMBF ₄ /AN	244.5 F cm ⁻³ at 2 mV s ⁻¹	two electrode system	S33
$\text{Ti}_3\text{C}_2\text{T}_x/\text{MnO}_2$ nano wires film	PVA/LiCl gel	1025 F cm ⁻³ at 1 A cm ⁻³	flexible SCs device	S34
$\text{MnO}_2/\text{Ti}_3\text{C}_2\text{T}_x$ film	PVA/H ₂ SO ₄ gel	130.5 F g ⁻¹ at 0.2 A g ⁻¹	flexible SCs device	S35
Metal porphyrin frameworks/ $\text{Ti}_3\text{C}_2\text{T}_x$ hybrid film	PVA/H ₂ SO ₄ gel	408 mF cm ⁻² at 0.5 mA cm ⁻²	flexible SCs device	S36
$\text{Ti}_3\text{C}_2\text{T}_x/\text{CNT}$ fibers	PVA/LiCl gel	22.7 F cm ⁻³ at 0.1 A cm ⁻³	solid-state fibriform SCs	S37
$\text{Ti}_3\text{C}_2\text{T}_x$ film	1 M Et ₄ NBF ₄ /ACN	54 F g ⁻¹ (220 F cm ⁻³) at 2 mV s ⁻¹	organic electrolyte asymmetric device	S38
$\text{Ti}_3\text{C}_2\text{T}_x/\text{rGO}/\text{Ti}_3\text{C}_2\text{T}_x$ film	1 M Et ₄ NBF ₄ /ACN	48 F g ⁻¹ (78 F cm ⁻³) at 2 mV s ⁻¹	organic electrolyte asymmetric device	S38
N-doping cotton-derived carbon frameworks/graphene aerogel	PVA/KOH gel	~200 F g ⁻¹ at 0.1 A g ⁻¹	all-solid-state SCs	S39
PANI/orderly nanotube array film	PVA/H ₂ SO ₄ gel	237.5 mF cm ⁻² at 10 mV s ⁻¹	flexible all- solid-state SCs	S40
Fe ₃ O ₄ /carbon nanotube/polyaniline ternary film	PVA/H ₂ SO ₄ gel	201 F g ⁻¹ at 20 mV s ⁻¹	all-solid-state SCs	S41
CNT/PANI composite film	PAAm/LiCl hydrogel	99.3 F cm ⁻³ at 0.1 mA cm ⁻³	all-solid-state SCs	S42
$\text{Ti}_3\text{C}_2\text{T}_x$	PVA/H ₂ SO ₄ gel	27.29 mF cm ⁻² at 2 mV s ⁻¹	Micro-SCs	S43
$\text{Ti}_3\text{C}_2\text{T}_x/\text{rGO}$	PVA/H ₂ SO ₄ gel	80 F cm ⁻³ at 2 mV s ⁻¹	Micro-SCs	S44
Ascorbate/ $\text{Ti}_3\text{C}_2\text{T}_x$	PVA/H ₂ SO ₄ gel	720.7 F cm ⁻³ at 1 A g ⁻¹	Micro-SCs	S45
Hierarchical $\text{Ti}_3\text{C}_2\text{T}_x$	PVA/H ₃ PO ₄ gel	485 F cm ⁻³ at 1 A cm ⁻³	all-solid-state SCs	S46
PEDOT:PSS/ $\text{Ti}_3\text{C}_2\text{T}_x$ (PEDOT:poly(3,4- ethylenedioxythiophene)	PVA/H ₂ SO ₄ gel	361.4 F cm ⁻³ at 2 mV s ⁻¹	fiber-shaped SCs	S5

(PPS: poly(styrenesulfonate))					
Ti ₃ C ₂ T _x /rGO-90 fibers	PVA/H ₃ PO ₄ gel	586.4 F cm ⁻³ at 10 mV s ⁻¹	fiber based all-solid-state symmetric SCs	S2	
Ti ₃ C ₂ T _x /GO fibers	PVA/H ₂ SO ₄ gel	256 F cm ⁻³ at 0.1 A cm ⁻³	fiber SCs device	S47	
Ti ₃ C ₂ T _x film	PVA/H ₂ SO ₄ gel	183 F cm ⁻³ at 0.25 mA cm ⁻²	flexible solid- state Micro- SCs	S48	
Electrochemically exfoliated graphene/ Ti ₃ C ₂ T _x film	PVA/H ₃ PO ₄ gel	216 F cm ⁻³ at 0.1 A cm ⁻³	flexible all- solid-state SCs	S49	
PANI/bacterial cellulose//Ti ₃ C ₂ T _x /bacteria 1 cellulose	PVA/H ₂ SO ₄ gel	585 mF cm ⁻² at 3 mA cm ⁻²	flexible all- solid-state SCs	S50	
RuO ₂ //Ti ₃ C ₂ T _x	PVA/H ₃ PO ₄ gel	60 mF cm ⁻² at 5 mV s ⁻¹	all-solid-state SCs	S51	
PANI@ANF film	PVA/H ₂ SO ₄ gel	138 F g ⁻¹ at 0.5 A g ⁻¹	all-solid-state SCs	S52	
i-PANI@Ti ₃ C ₂ T _x film	PVA/H ₂ SO ₄ gel	310 F g ⁻¹ (1001 F cm ⁻³) at 1 A g ⁻¹	flexible all- solid-state SCs	This work	
Lig@Ti ₃ C ₂ T _x film	PVA/H ₂ SO ₄ gel	271 F g ⁻¹ (881 F cm ⁻³) at 1 A g ⁻¹	flexible all- solid-state SCs	This work	
Lig@Ti ₃ C ₂ T _x /i- PANI@Ti ₃ C ₂ T _x (5/5) film	PVA/H ₂ SO ₄ gel	295 F g ⁻¹ (959 F cm ⁻³) at 1 A g ⁻¹	flexible all- solid-state SCs	This work	

Table S5. The comparison of energy density of previously reported supercapacitor and the as-prepared flexible ASSSs.

Electrode material	Electrolyte	Gravimetric power density (Wh kg ⁻¹)	Volume power density (Wh L ⁻³)	Ref.
Ti ₃ C ₂ T _x /rGO-5 wt% film	3 M H ₂ SO ₄	10.5 at 80.3 W kg ⁻¹	32.6 at 200 W L ⁻¹	two electrode system S24
MnO _x -Ti ₃ C ₂ T _x film	1 M Li ₂ SO ₄	/	13.64 at 2 mV s ⁻¹	two electrode system S25

Nitrogen doped reduced graphene oxide foams (NrGFs)	6 M KOH	9 at 25 W kg ⁻¹	/	two electrode system	S26
Ti ₃ C ₂ T _x /CuS composites//Ti ₃ C ₂ T _x MXene	1 M KOH	15.4 at 750.2 W kg ⁻¹	/	asymmetri c, two electrode system	S27
Fe(OH) ₃ /Ti ₃ C ₂ T _x film	3 M H ₂ SO ₄	6.3 at 56 W kg ⁻¹	20.7 at 184.8 W L ⁻¹	two electrode system	S28
Ti ₃ C ₂ T _x /PEDOT:PSS//r GO film	1 M H ₂ SO ₄	/	23 at 7659 W L ⁻¹	asymmetri c, two electrode system	S29
d-Ti ₃ C ₂ T _x /Ni foam//b- Ti ₃ C ₂ T _x film	6M KOH	18.1 at 397.8 W kg ⁻¹	/	asymmetri c, two electrode system	S30
N-Ti ₃ C ₂ T _x -300 film	3 M H ₂ SO ₄	/	21.0 at 151.3 W L ⁻¹	two electrode system	S31
Active carbon/Co ₃ O ₄ - doped 3D Ti ₃ C ₂ T _x /rGO hybrid film	6M KOH	8.25 at 159.94 W kg ⁻¹	/	two electrode system	S32
d-Ti ₃ C ₂ T _x film under 40 MPa	1 M EMIMBF ₄ /AN	/	41 at 108 W L ⁻¹ (27 at 500 W L ⁻¹) ¹⁾	two electrode system	S33
Ti ₃ C ₂ T _x /MnO ₂ nano wires film	PVA/LiCl gel	/	56.94 at 500 W L ⁻¹	flexible SCs device	S34
MnO ₂ /Ti ₃ C ₂ T _x film	PVA/H ₂ SO ₄ gel	/	0.7 μWh cm ⁻² at 80.0 μW cm ⁻²	flexible SCs device	S35
Metal porphyrin frameworks/Ti ₃ C ₂ T _x hybrid film	PVA/H ₂ SO ₄ gel	/	20.4 μWh cm ⁻² at 152.2 μW cm ⁻² ²⁾	flexible SCs device	S36
Ti ₃ C ₂ T _x /CNT fibers	PVA/LiCl gel	/	2.55 at 45.9 W L ⁻¹	solid-state fibriform SCs	S37
Ti ₃ C ₂ T _x /MWCNT film	1 M Et ₄ NBF ₄ /ACN	3 at 60 W kg ⁻¹	/	organic electrolyte symmetric device	S38
Ti ₃ C ₂ T _x /rGO//Ti ₃ C ₂ T _x film	1 M Et ₄ NBF ₄ /ACN	8 at 55 W kg ⁻¹	15 at 90 W L ⁻¹	organic electrolyte	S38

					asymmetric device
N-doping cotton-derived carbon frameworks/graphene aerogel film	PVA/KOH gel	20 at 4000 W kg ⁻¹	/	all-solid-state SCs	S39
PANI/orderly nanotube array film	PVA/H ₂ SO ₄ gel	/	24.31 at 2.74 W L ⁻¹	flexible all-solid-state SCs	S40
Fe ₃ O ₄ /carbon nanotube/polyaniline ternary film	PVA/H ₂ SO ₄ gel	28.0 at 5.3 kW kg ⁻¹	/	all-solid-state SCs	S41
CNT/PANI composite film	PAAm/LiCl hydrogel	/	8.8 at 370 W L ⁻¹	all-solid-state SCs	S42
Ti ₃ C ₂ T _x	PVA/H ₂ SO ₄ gel	/	6.1 at 200 W L ⁻¹	Micro-SCs	S43
Ti ₃ C ₂ T _x /rGO	PVA/H ₂ SO ₄ gel	/	8.6 at 200 W L ⁻¹	Micro-SCs	S44
Ascorbate/Ti ₃ C ₂ T _x	PVA/H ₂ SO ₄ gel	/	100.2 at 1900 W L ⁻¹	Micro-SCs	S45
Hierarchical Ti ₃ C ₂ T _x	PVA/H ₃ PO ₄ gel	/	9.6 at 2800 W L ⁻¹	all-solid-state SCs	S46
PEDOT:PSS/Ti ₃ C ₂ T _x	PVA/H ₂ SO ₄ gel	6.49 at 142.16 W L ⁻¹	7.13 at 142.16 W L ⁻¹	fiber-shaped SCs	S5
Ti ₃ C ₂ T _x /rGO-90 fibers	PVA/H ₃ PO ₄ gel	/	13.03 at 590 W L ⁻¹	fiber based all-solid-state symmetric SCs	S2
Ti ₃ C ₂ T _x /GO fibers	PVA/H ₂ SO ₄ gel	/	5.1 at 200 W L ⁻¹	fiber SCs device	S47
Ti ₃ C ₂ T _x film	PVA/H ₂ SO ₄ gel	/	12.4 at 87.5 W cm ⁻²	flexible solid-state Micro-SCs	S48
Electrochemically exfoliated graphene/Ti ₃ C ₂ T _x film	PVA/H ₃ PO ₄ gel	/	3.4 at 200 W L ⁻¹	flexible all-solid-state SCs	S49
PANI/bacterial cellulose//Ti ₃ C ₂ T _x /bacterial cellulose	PVA/H ₂ SO ₄ gel	/	159 µWh cm ⁻² at 34.4 mW cm ⁻²	flexible all-solid-state SCs	S50
RuO ₂ //Ti ₃ C ₂ T _x	PVA/H ₃ PO ₄ gel	/	37 µW h cm ⁻² at 40 mW cm ⁻²	solid-state device	S51
NiCoS/CC//AC	2 M KOH	40 at 379 W	/	asymmetric	S53

		kg^{-1}			c, two electrode system	
i-PANI@Ti ₃ C ₂ T _x film	PVA/H ₂ SO ₄ gel	10.76 at 500 W kg ⁻¹	34.8 at 1615 W L ⁻¹	flexible all-solid-state SCs	This work	
Lig@Ti ₃ C ₂ T _x film	PVA/H ₂ SO ₄ gel	9.41 at 500 W kg ⁻¹	30.6 at 1625 W L ⁻¹	flexible all-solid-state SCs	This work	
Lig@Ti ₃ C ₂ T _x /i-PANI@Ti ₃ C ₂ T _x (5/5) film	PVA/H ₂ SO ₄ gel	10.24 at 500 W kg ⁻¹	33.3 at 1625 W L ⁻¹	flexible all-solid-state SCs	This work	

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