

**Unprecedented collateral sensitivity for cisplatin-resistant lung cancer cells
presented by new ruthenium organometallic compounds**

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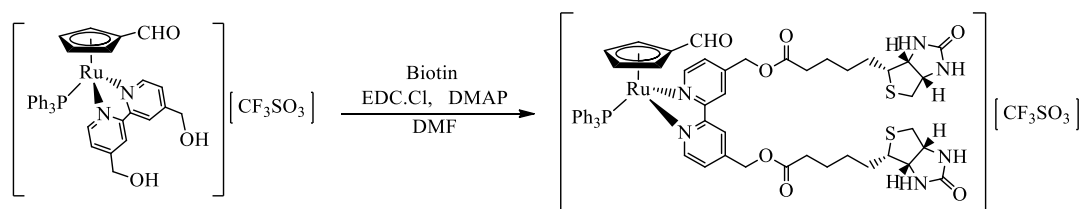
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Corresponding Authors

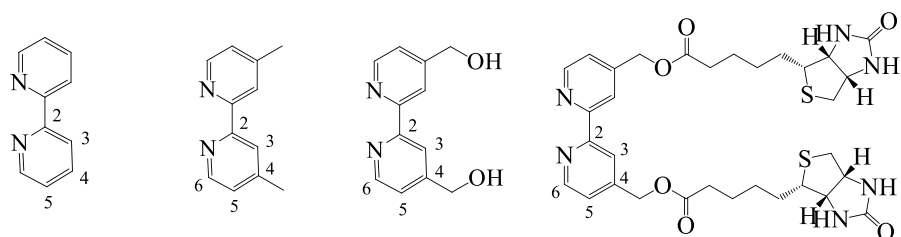
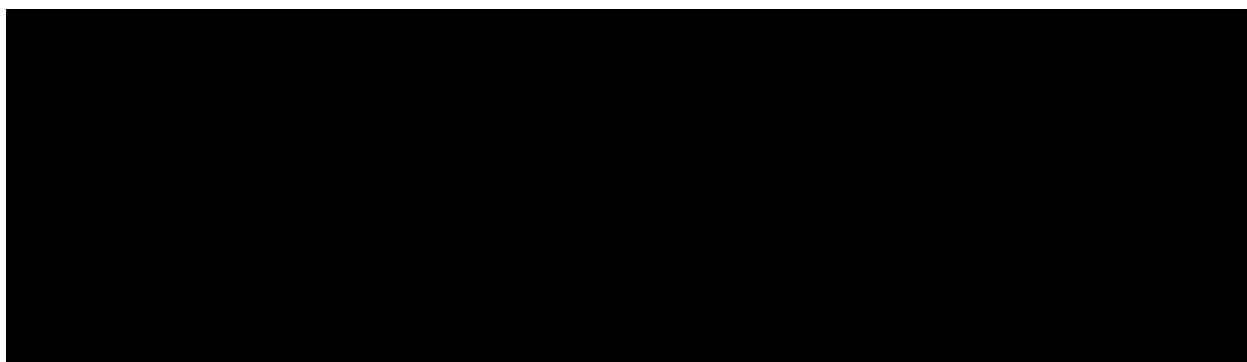
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Scheme S1. Synthesis of compound **4**.



Scheme S2. Synthesis route of compounds **1–3** and **5–8**; all compounds are numbered for NMR assignments. ** Compound **8** was not studied in this work.

S1. ESI mass spectra

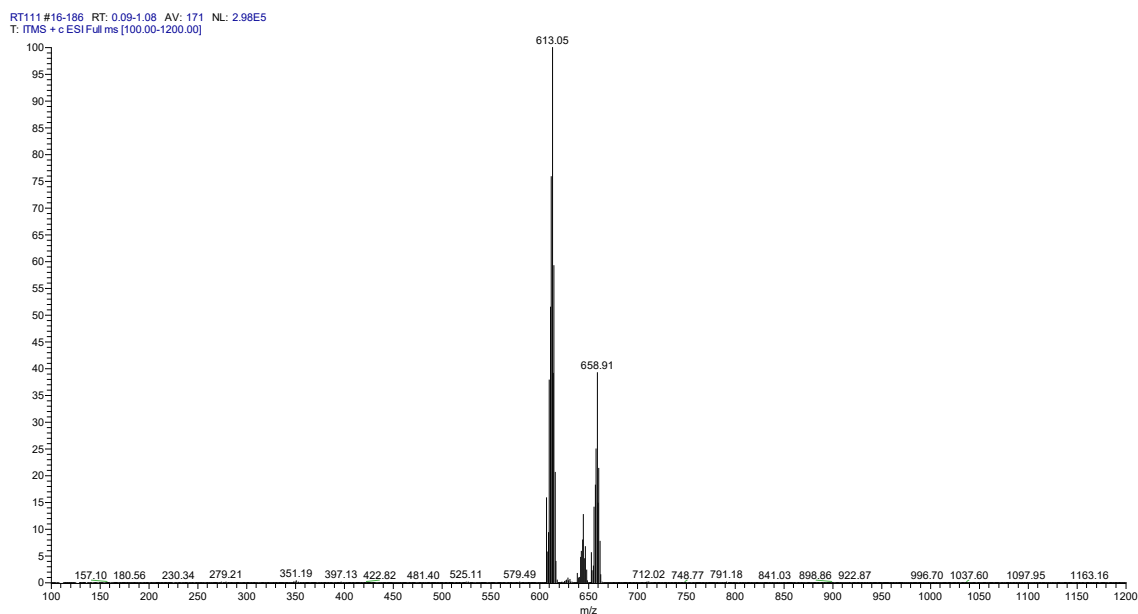


Figure S1. ESI-MS spectrum of complex **1** (positive detection mode).

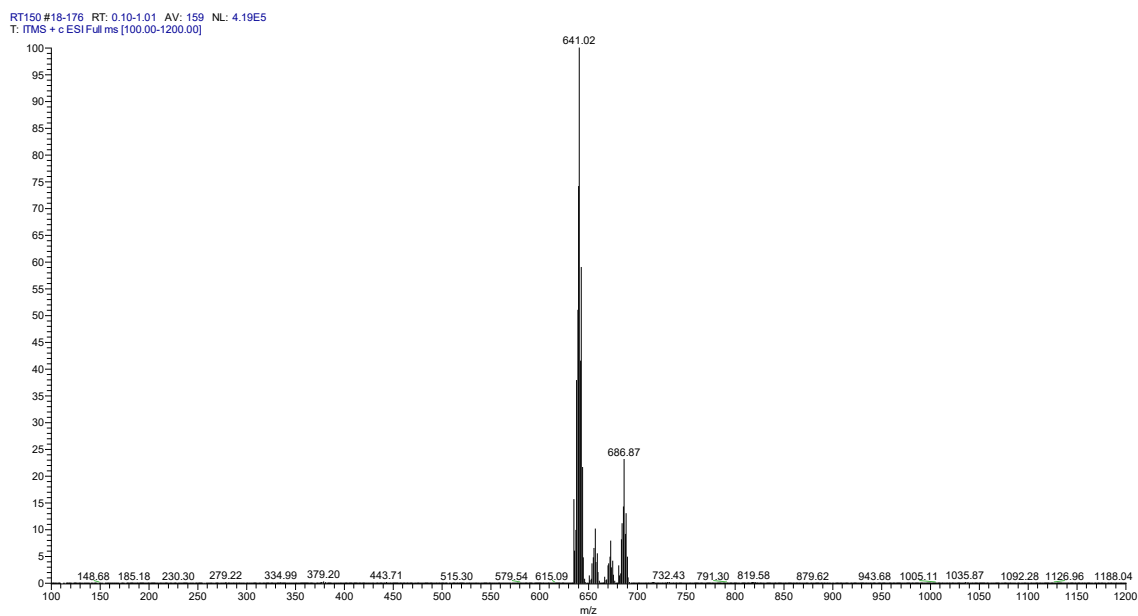


Figure S2. ESI-MS spectrum of complex **2** (positive detection mode).

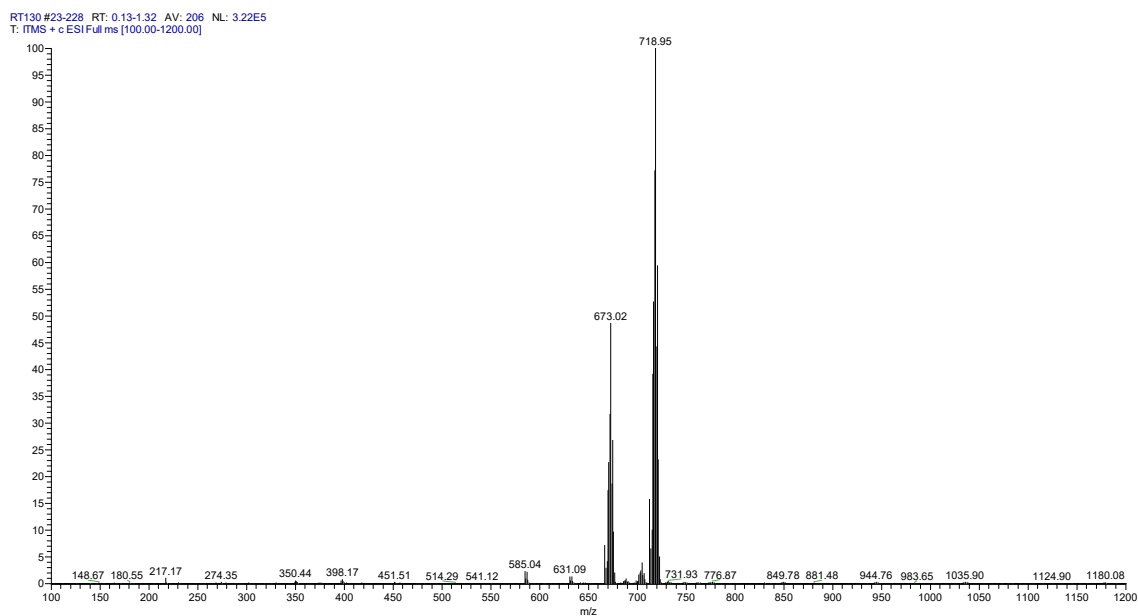


Figure S3. ESI-MS spectrum of complex **3** (positive detection mode).

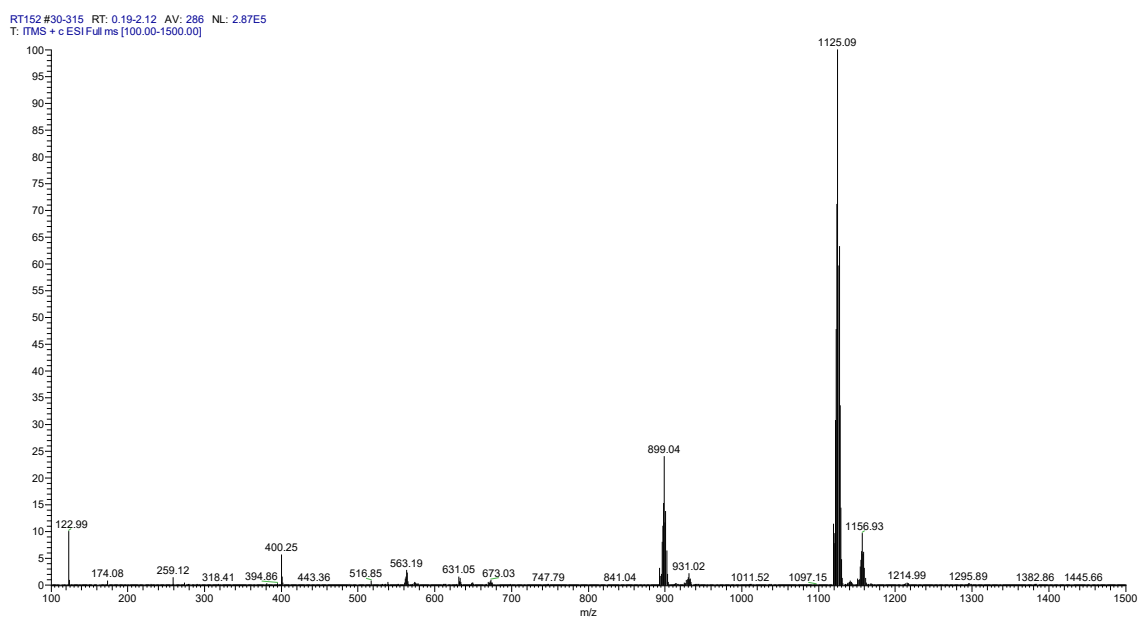


Figure S4. ESI-MS spectrum of complex **4** (positive detection mode).

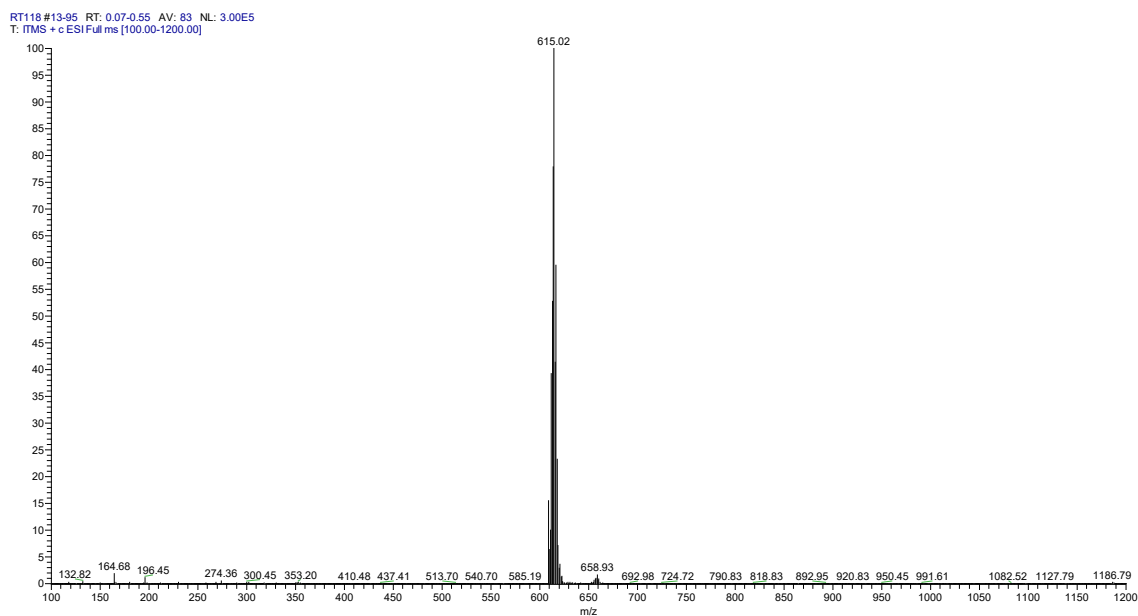


Figure S5. ESI-MS spectrum of complex **5** (positive detection mode).

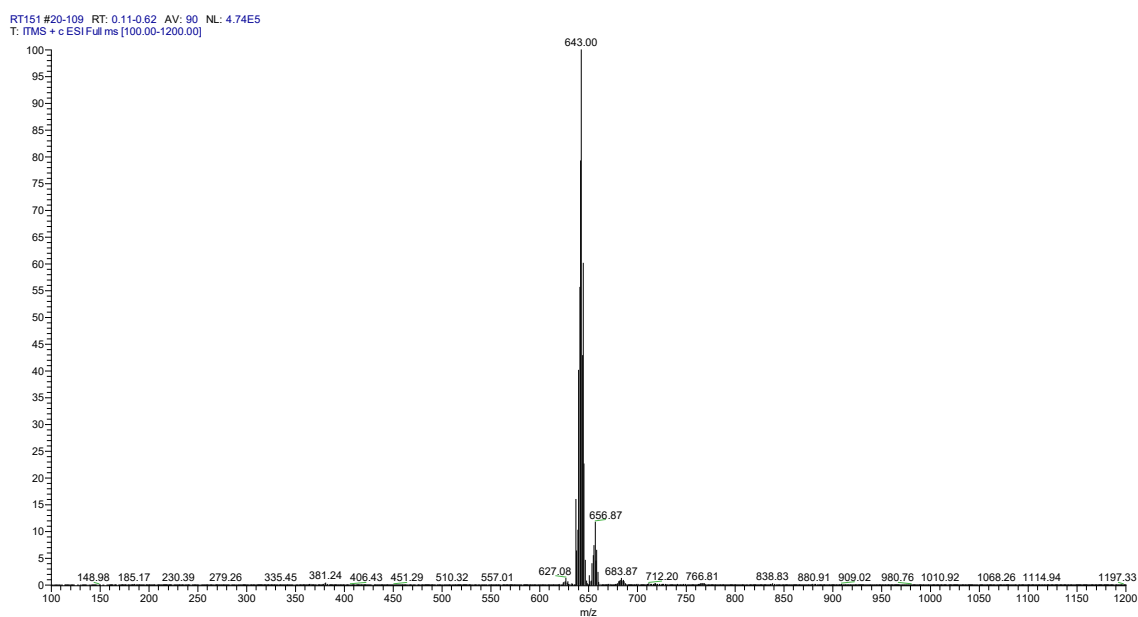


Figure S6. ESI-MS spectrum of complex **6** (positive detection mode).

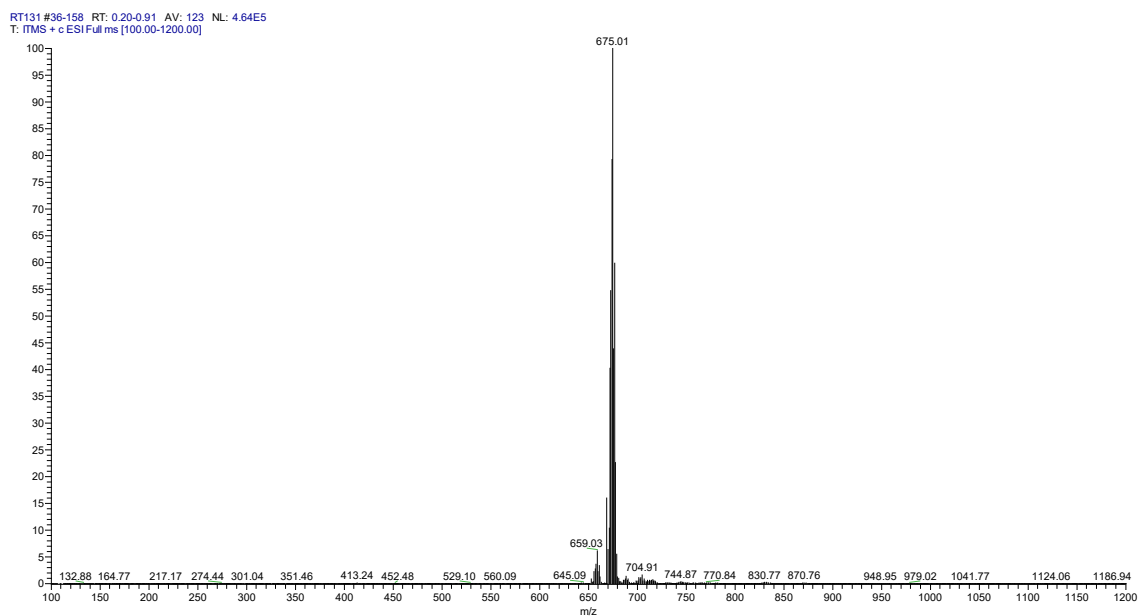


Figure S7. ESI-MS spectrum of complex **7** (positive detection mode).

S2. NMR spectra

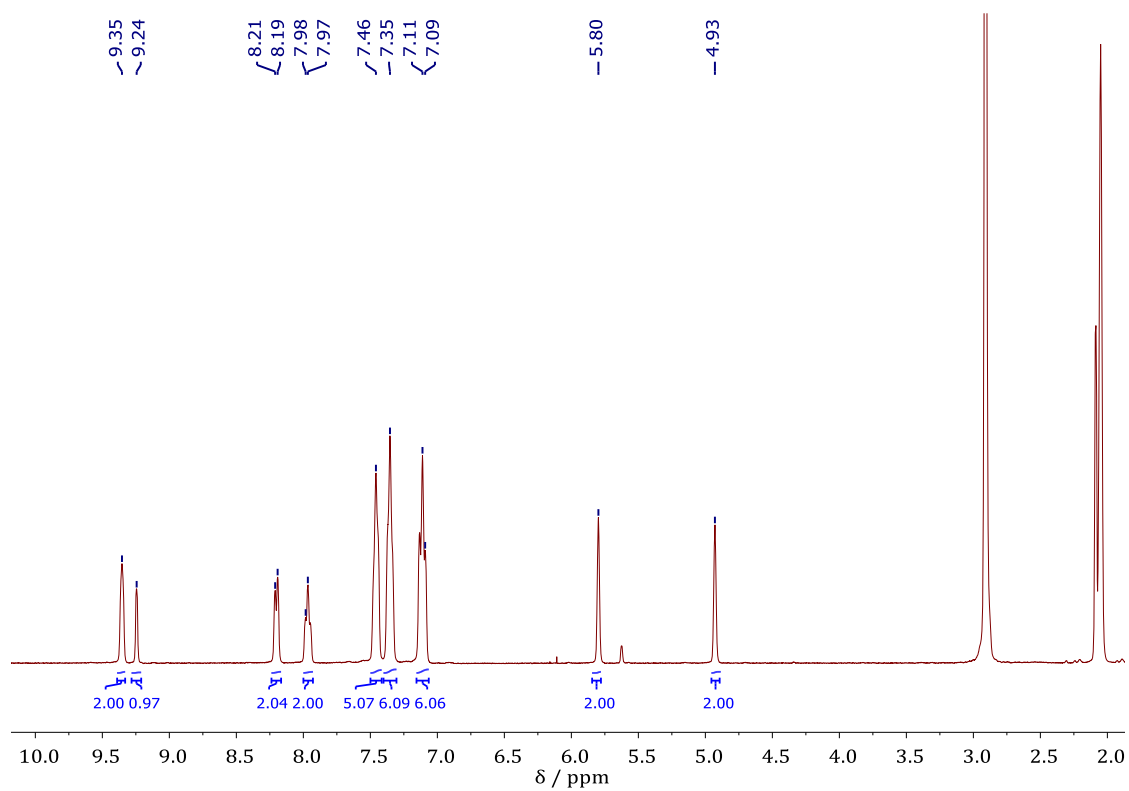


Figure S8. ¹H-NMR spectrum of complex **1** in acetone-*d*₆ at 298 K.

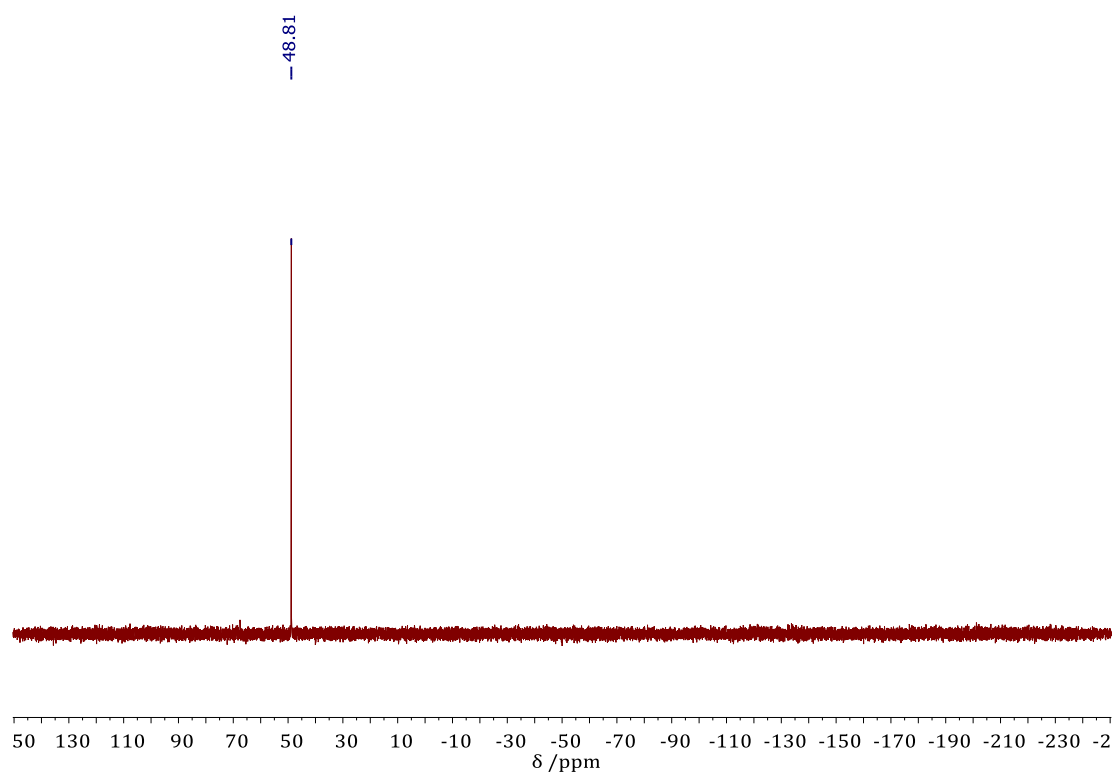


Figure S9. ³¹P{¹H}-NMR spectrum of complex **1** in acetone-*d*₆ at 298 K.

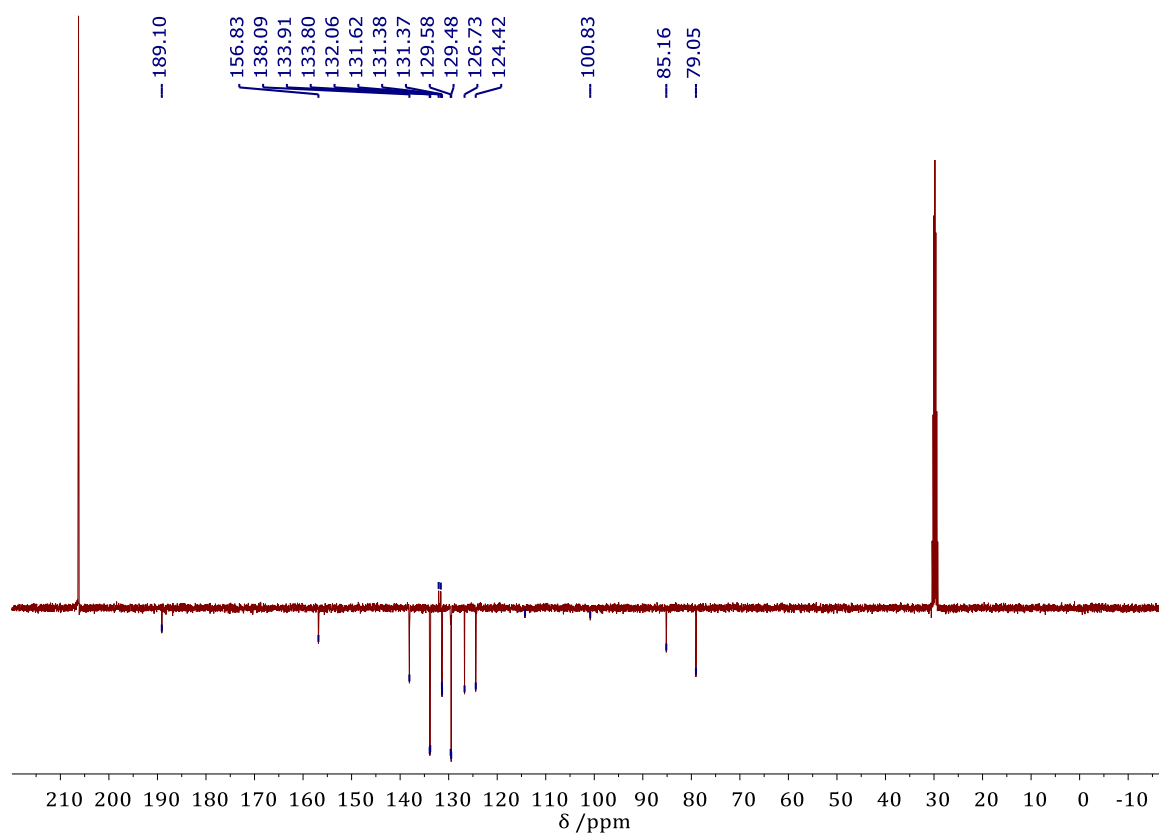


Figure S10. APT $^{13}\text{C}\{^1\text{H}\}$ -NMR spectrum of complex **1** in acetone- d_6 at 298 K.

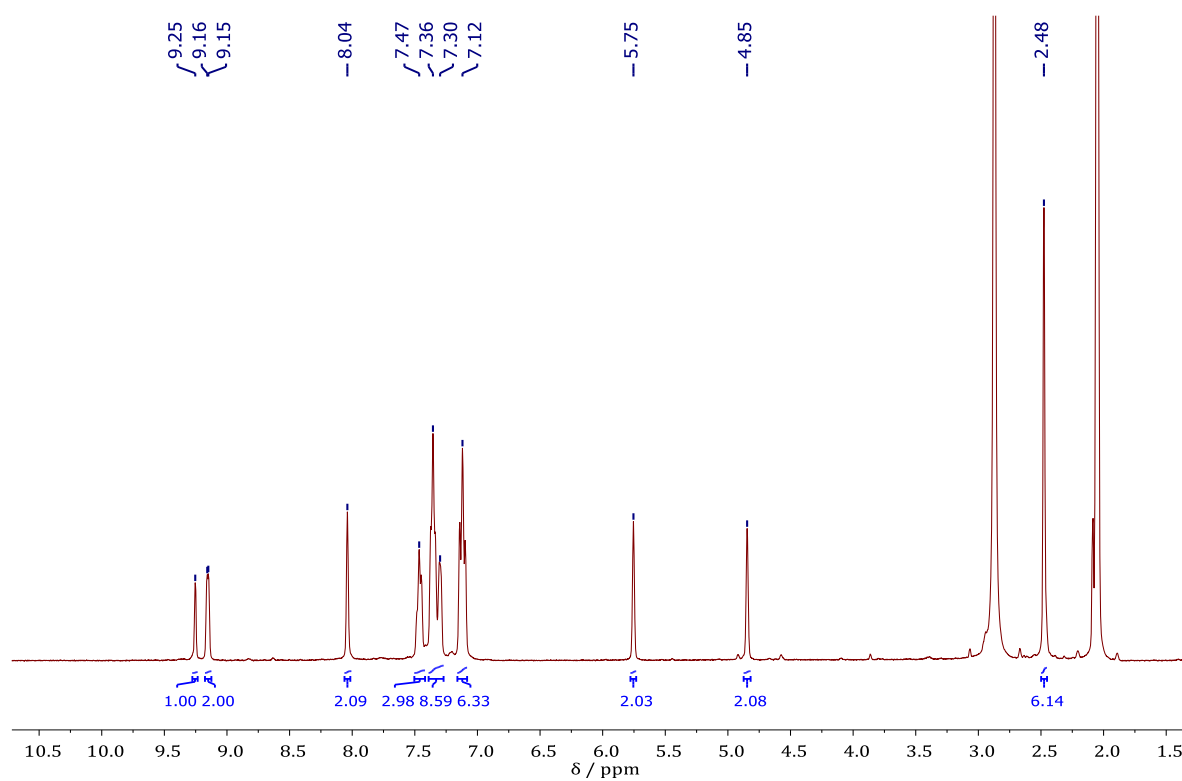


Figure S11. ^1H -NMR spectrum of complex **2** in acetone- d_6 at 298 K.

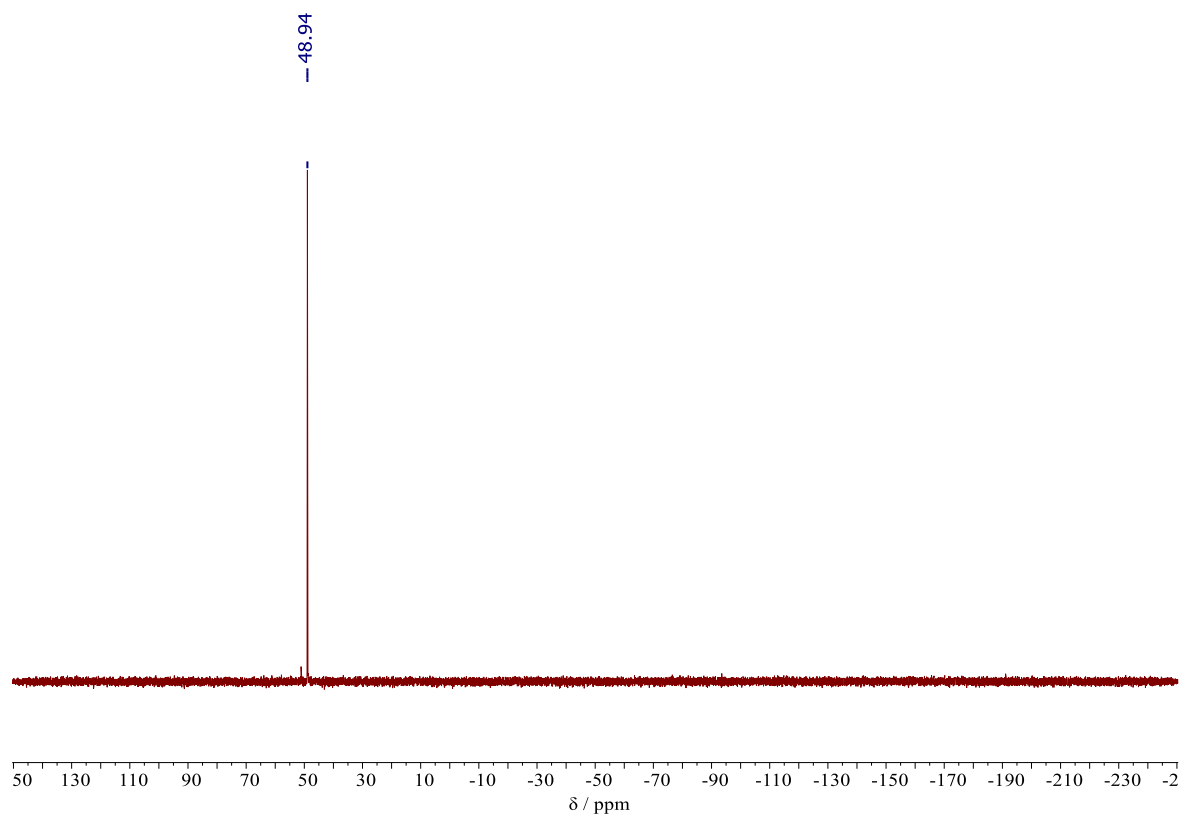


Figure S12. $^{31}\text{P}\{^1\text{H}\}$ -NMR spectrum of complex **2** in acetone- d_6 at 298 K.

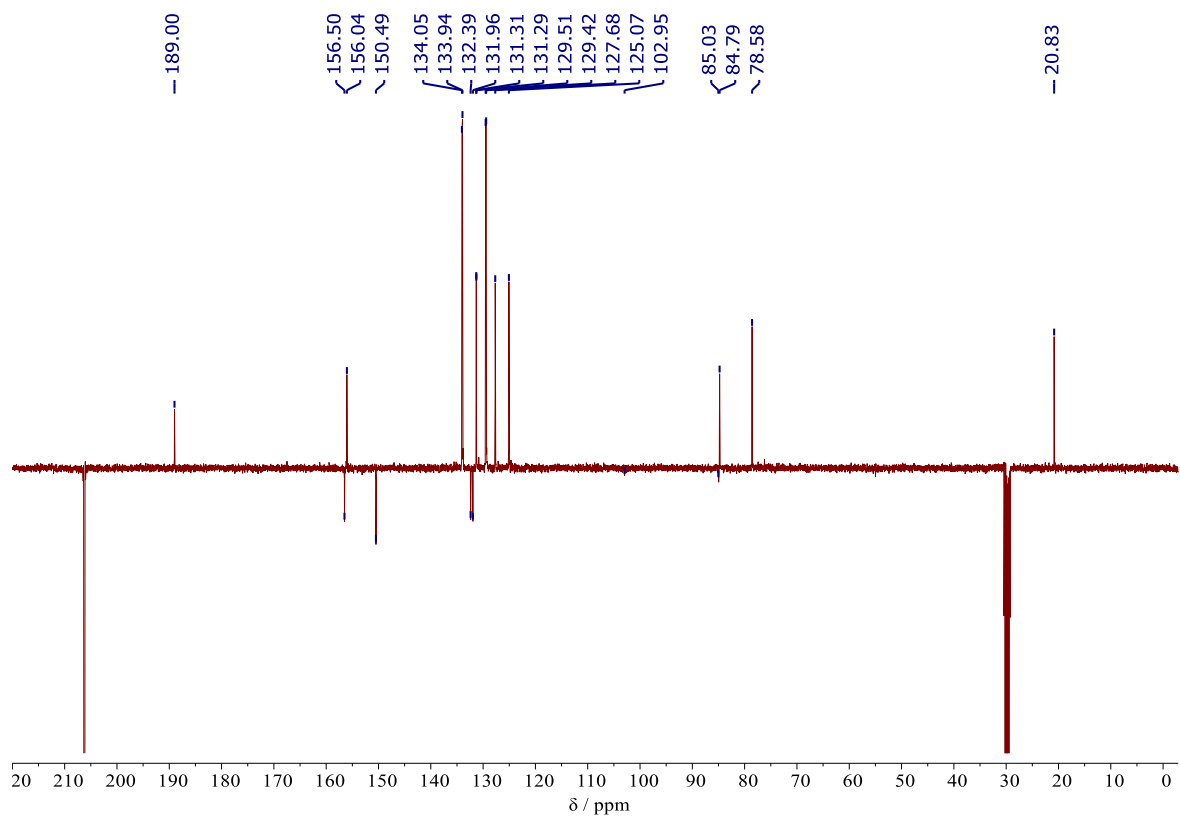


Figure S13. APT $^{13}\text{C}\{^1\text{H}\}$ -NMR spectrum of complex **2** in acetone- d_6 at 298 K.

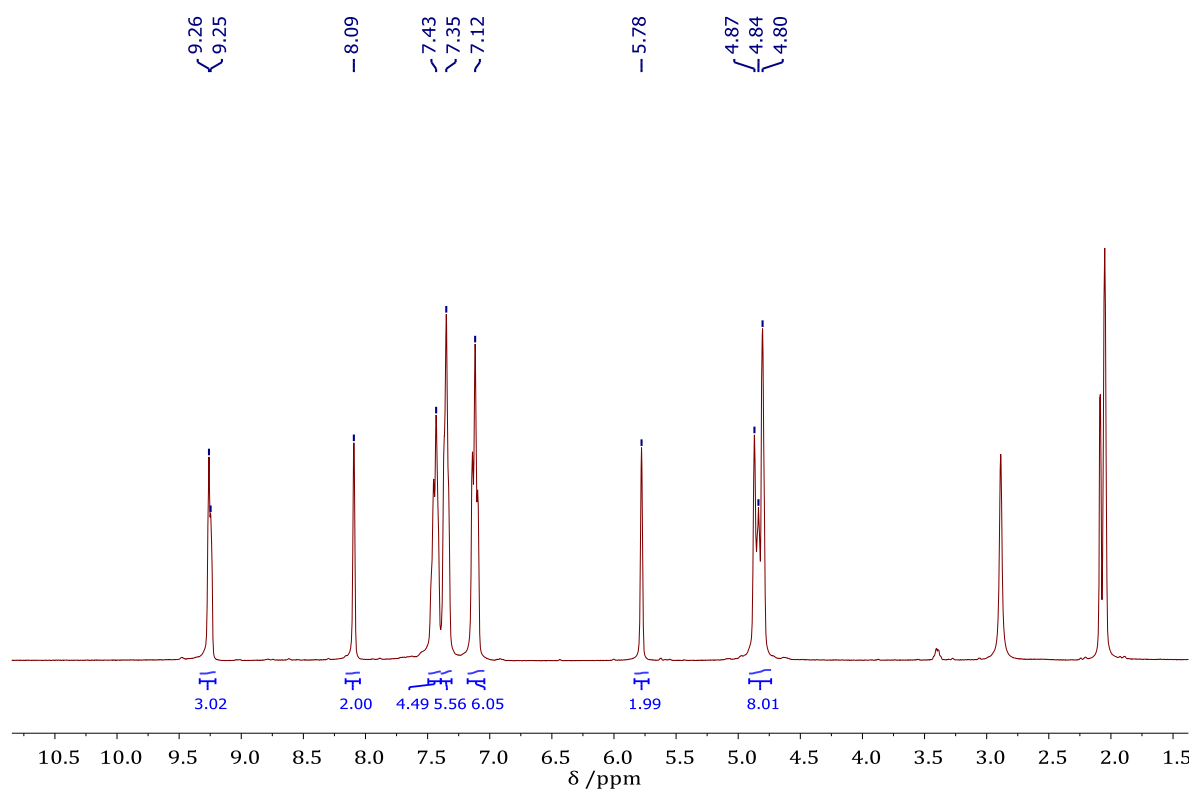


Figure S14. ^1H -NMR spectrum of complex **3** in acetone- d_6 at 298 K.

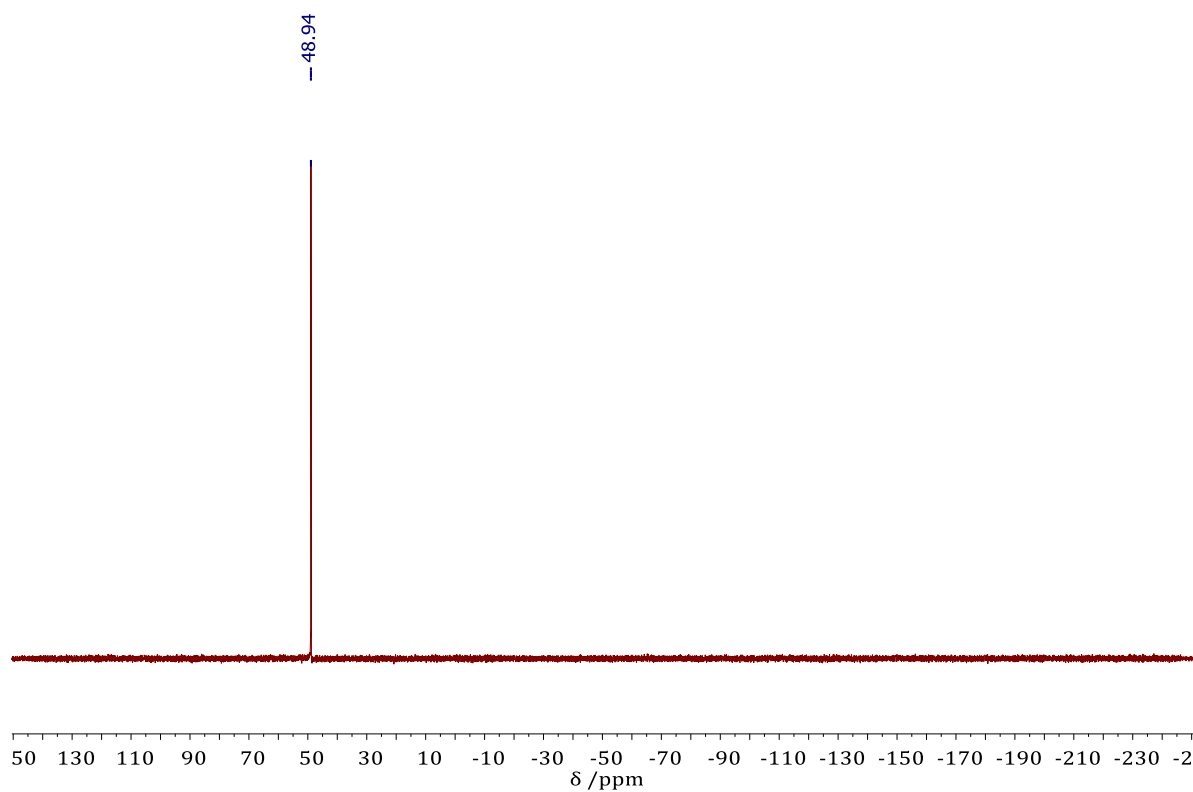


Figure S15. $^{31}\text{P}\{^1\text{H}\}$ -NMR spectrum of complex **3** in acetone- d_6 at 298 K.

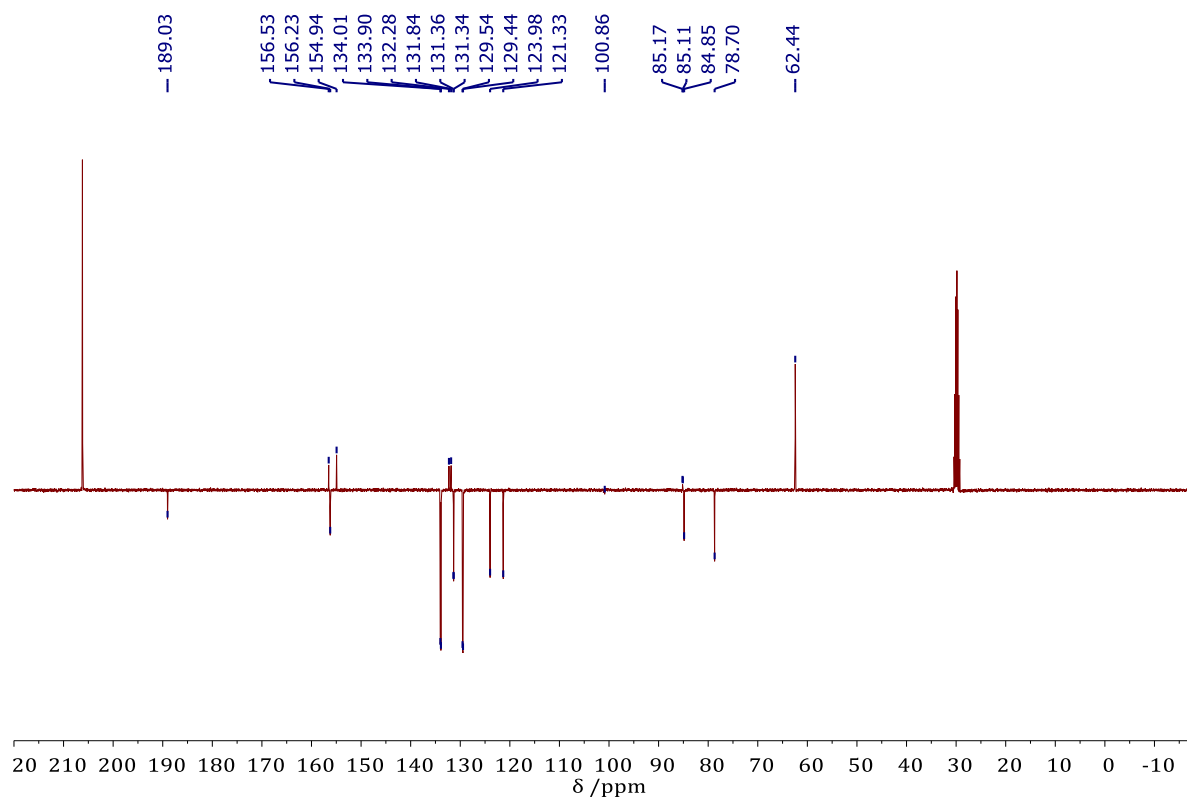


Figure S16. APT $^{13}\text{C}\{^1\text{H}\}$ -NMR spectrum of complex **3** in acetone- d_6 at 298 K.

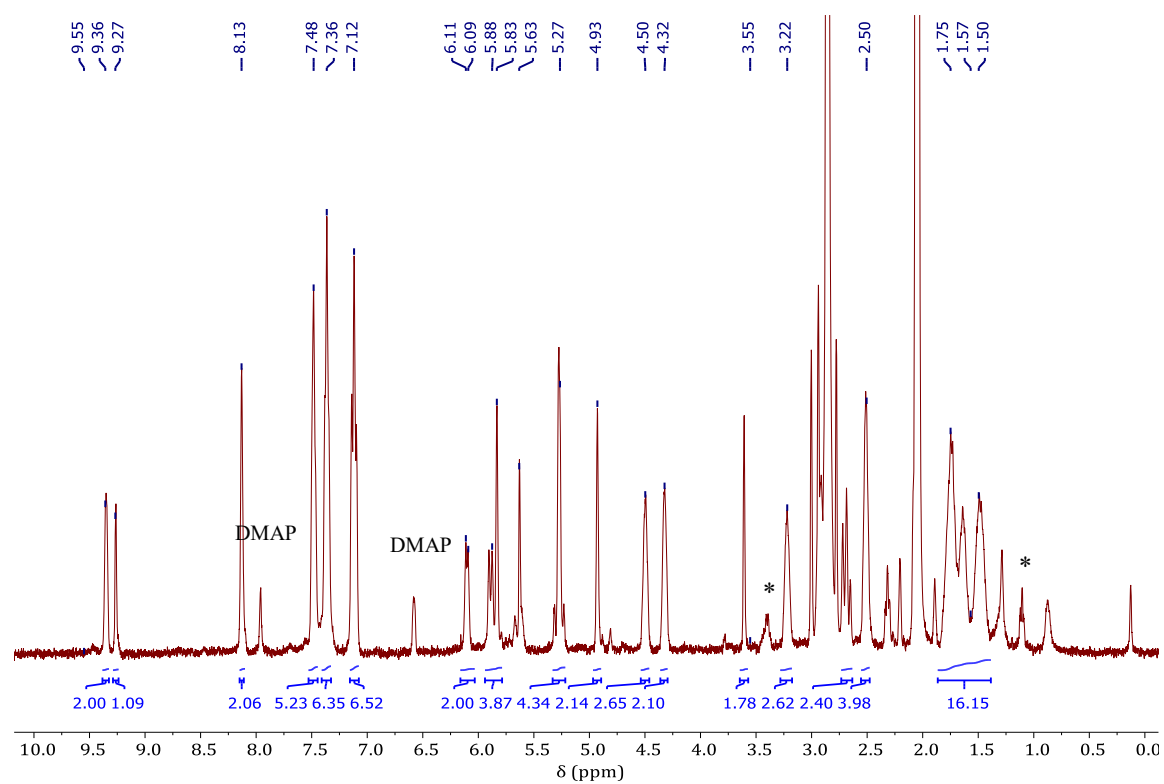


Figure S17. ^1H -NMR spectrum of complex **4** in acetone- d_6 at 298 K.

* residual diethyl ether

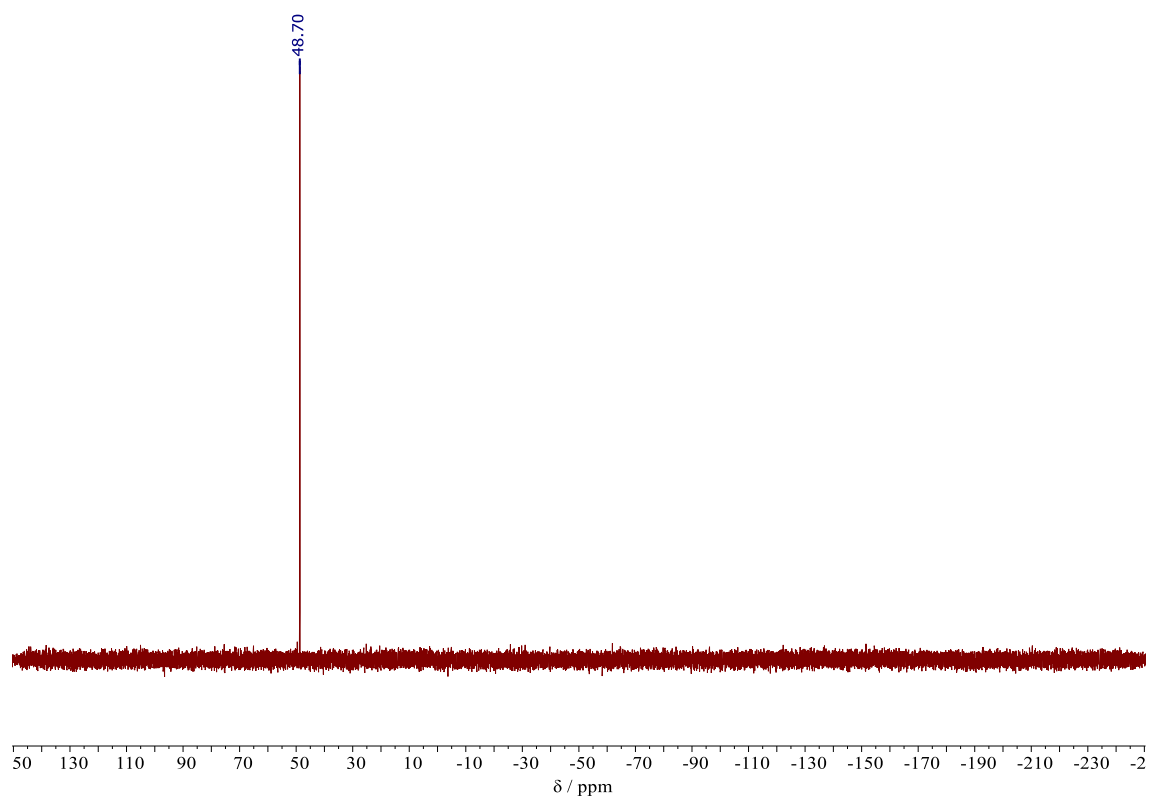


Figure S18. $^{31}\text{P}\{^1\text{H}\}$ -NMR spectrum of complex **4** in acetone- d_6 at 298 K.

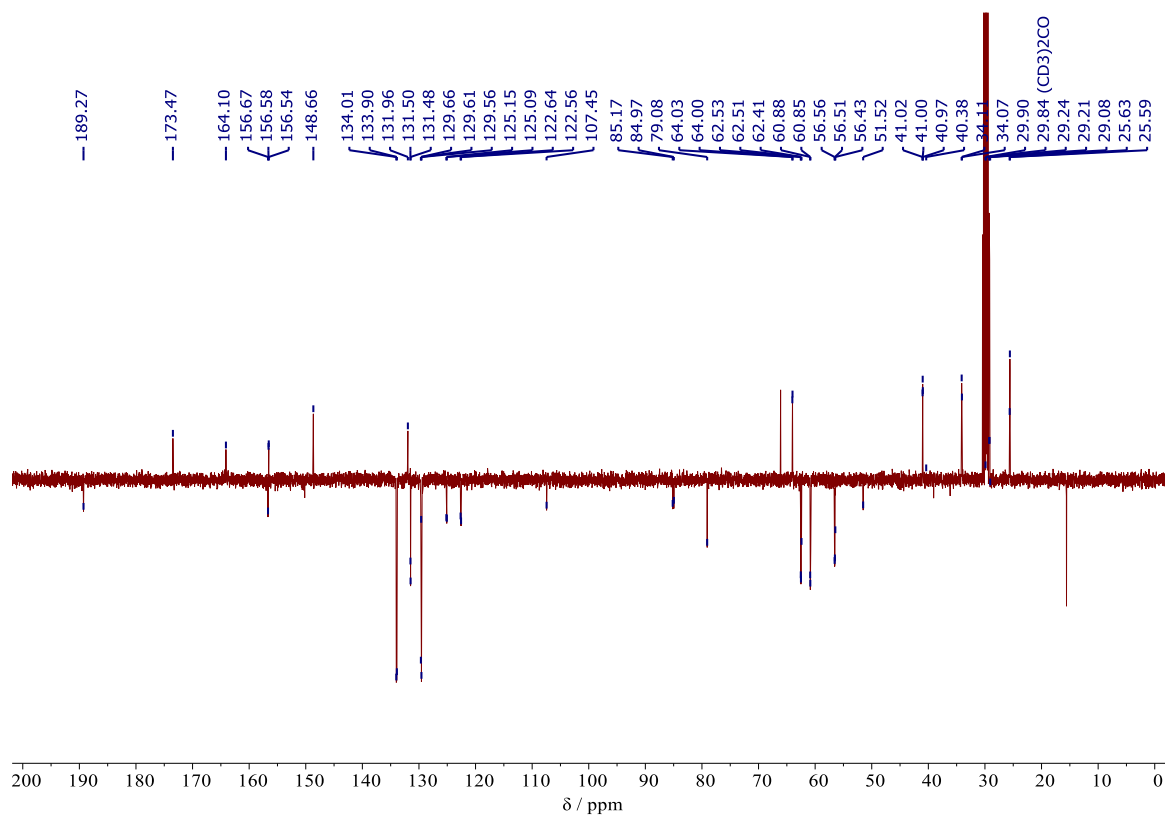


Figure S19. APT $^{13}\text{C}\{^1\text{H}\}$ -NMR spectrum of complex **4** in acetone- d_6 at 298 K.

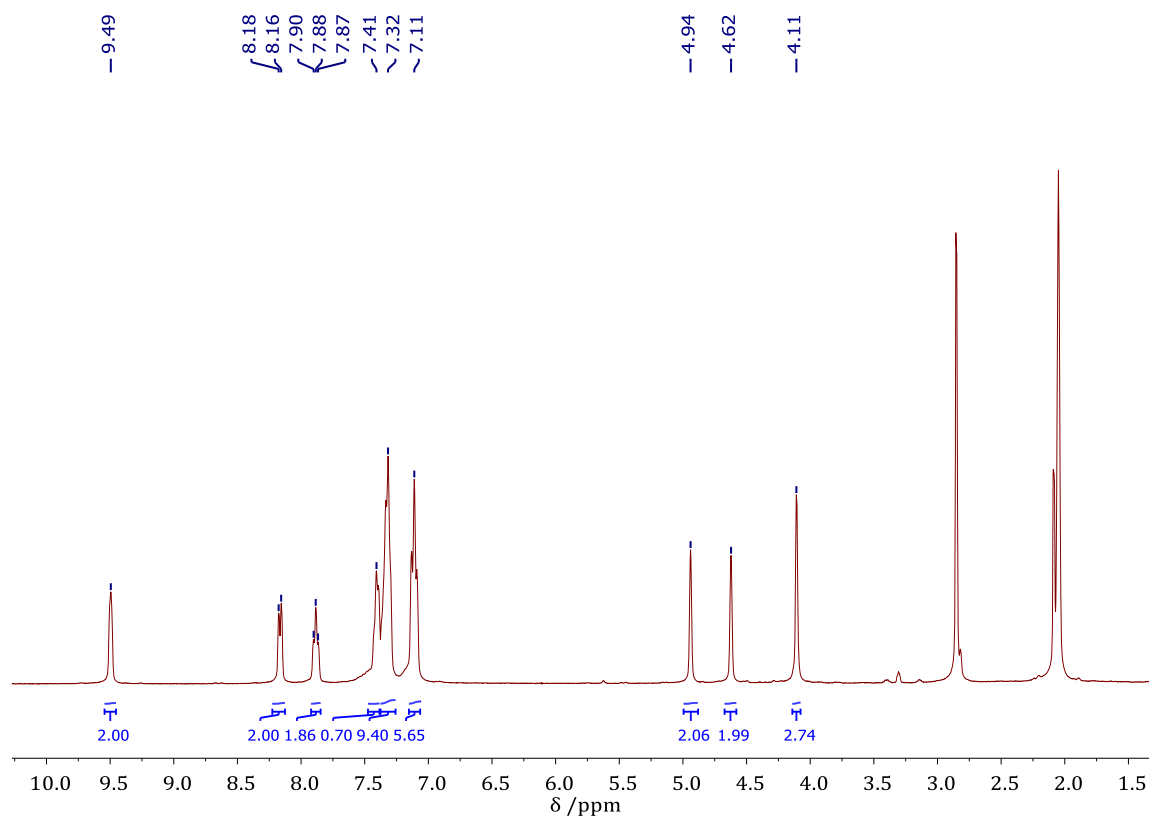


Figure S20. ^1H -NMR spectrum of complex **5** in acetone- d_6 at 298 K.

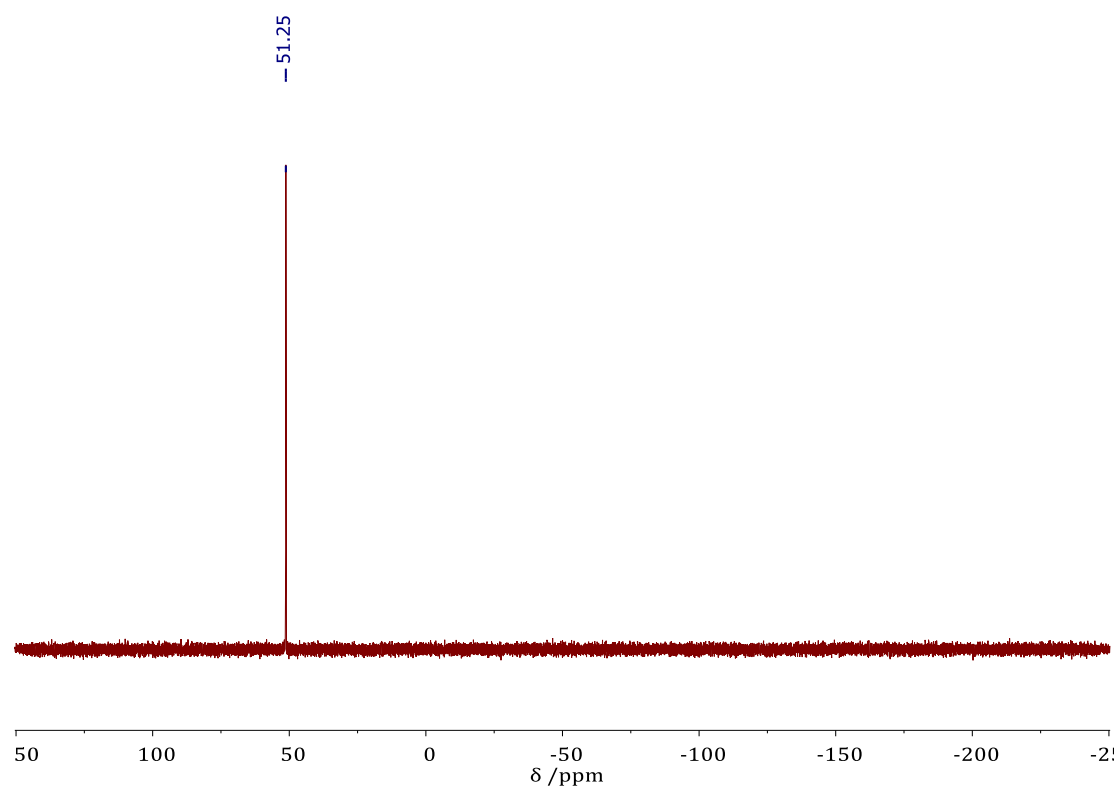


Figure S21. $^{31}\text{P}\{^1\text{H}\}$ -NMR spectrum of complex **5** in acetone- d_6 at 298 K.

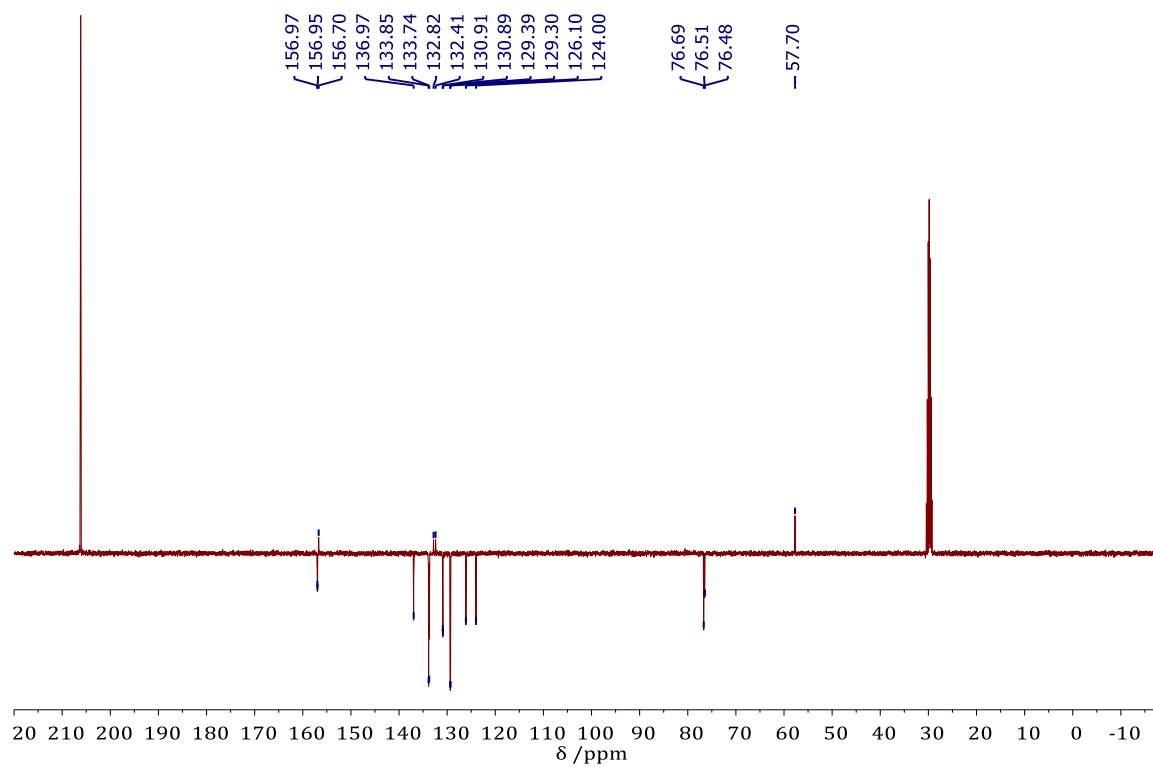


Figure S22. APT $^{13}\text{C}\{^1\text{H}\}$ -NMR spectrum of complex **5** in acetone- d_6 at 298 K.

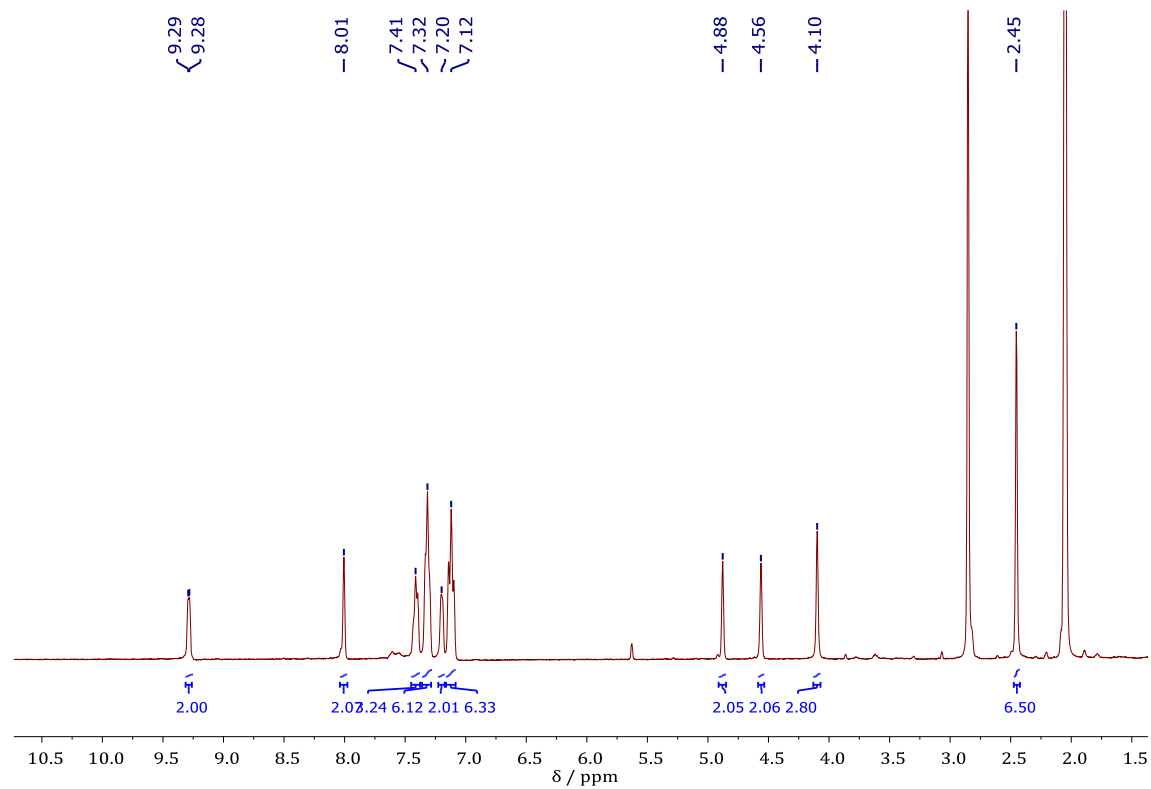


Figure S23. ^1H -NMR spectrum of complex **6** in acetone- d_6 at 298 K.

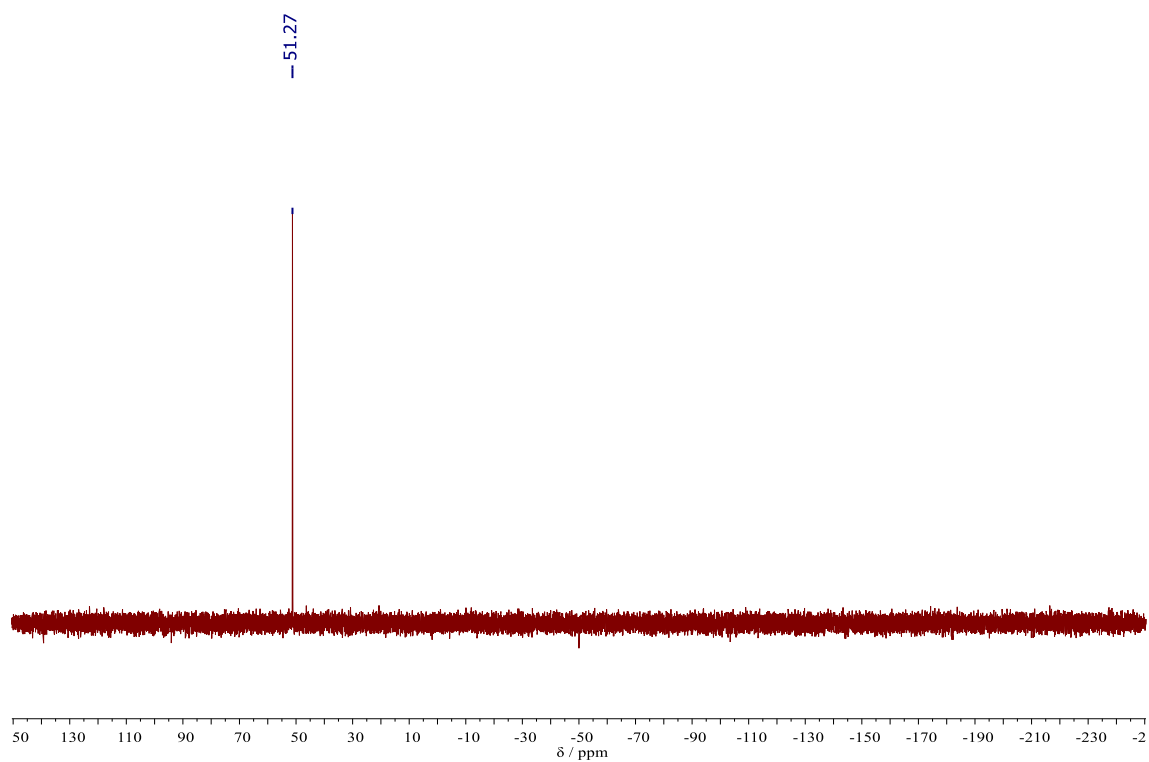


Figure S24. $^{31}\text{P}\{^1\text{H}\}$ -NMR spectrum of complex **6** in acetone- d_6 at 298 K.

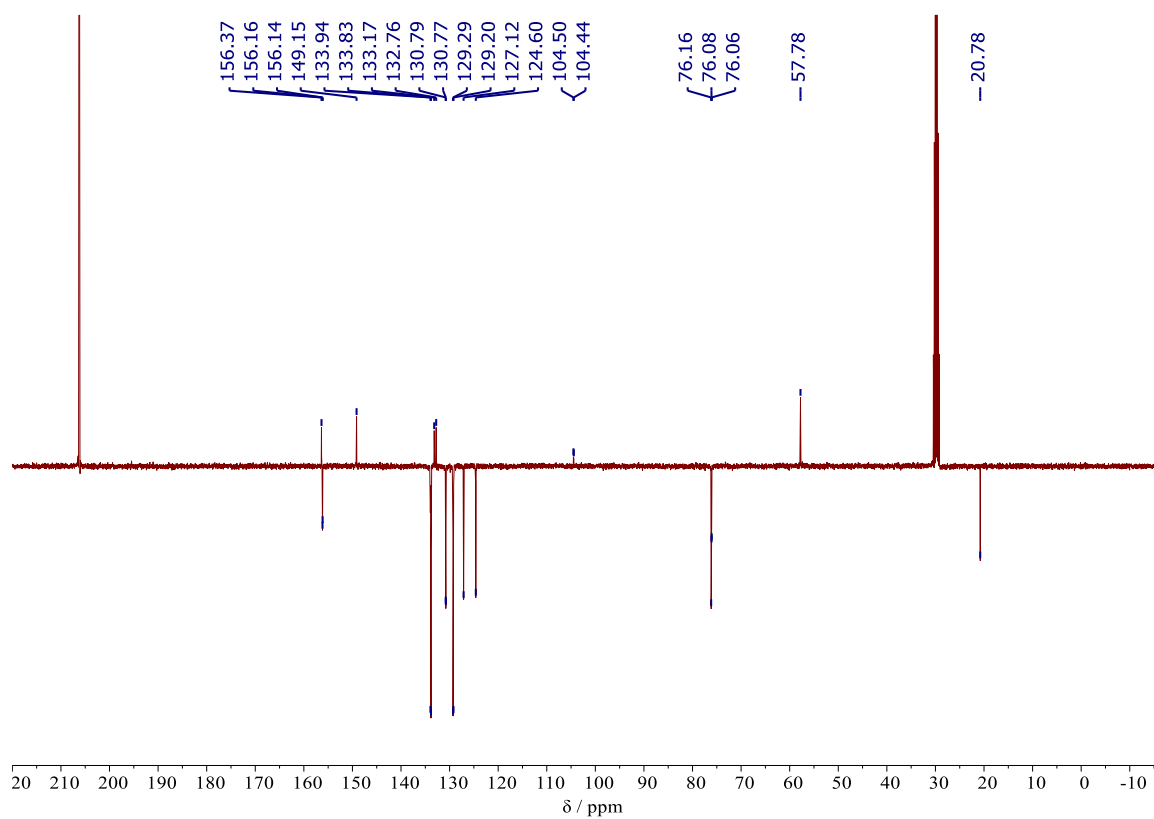


Figure S25. APT $^{13}\text{C}\{^1\text{H}\}$ -NMR spectrum of complex **6** in acetone- d_6 at 298 K.

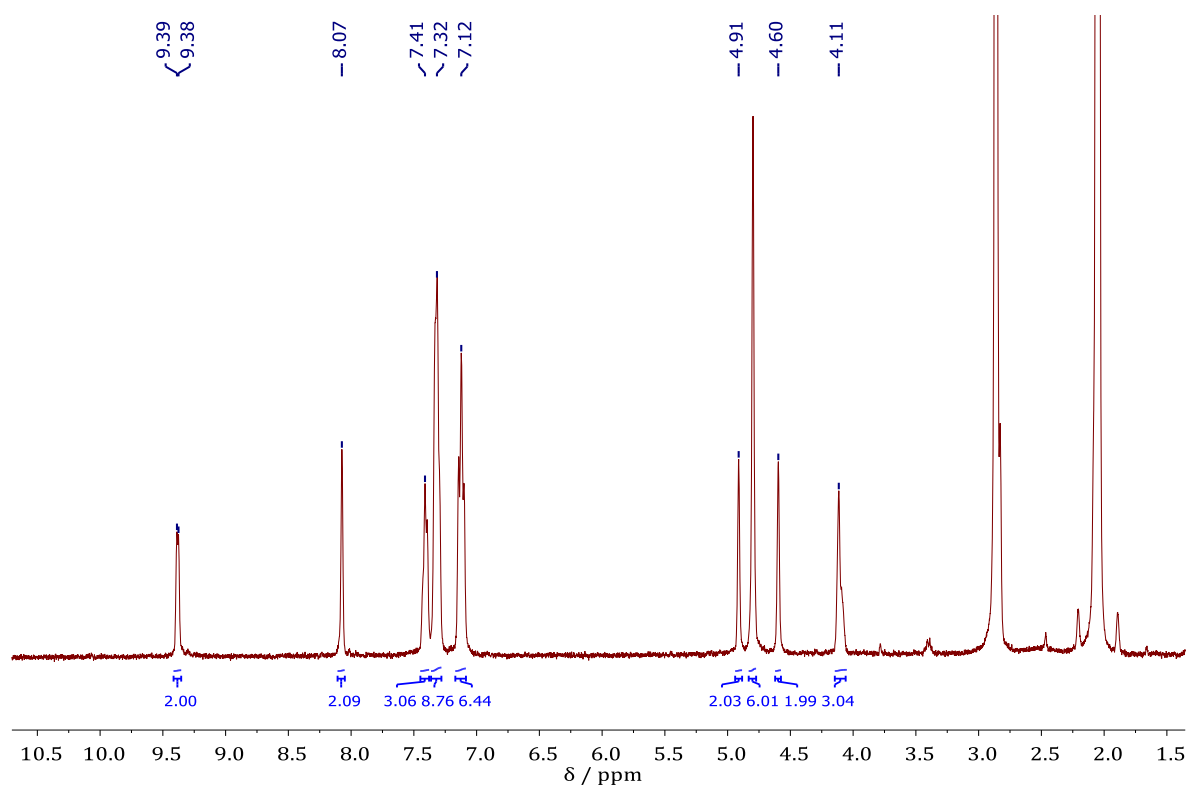


Figure S26. ^1H -NMR spectrum of complex **7** in acetone- d_6 at 298 K.

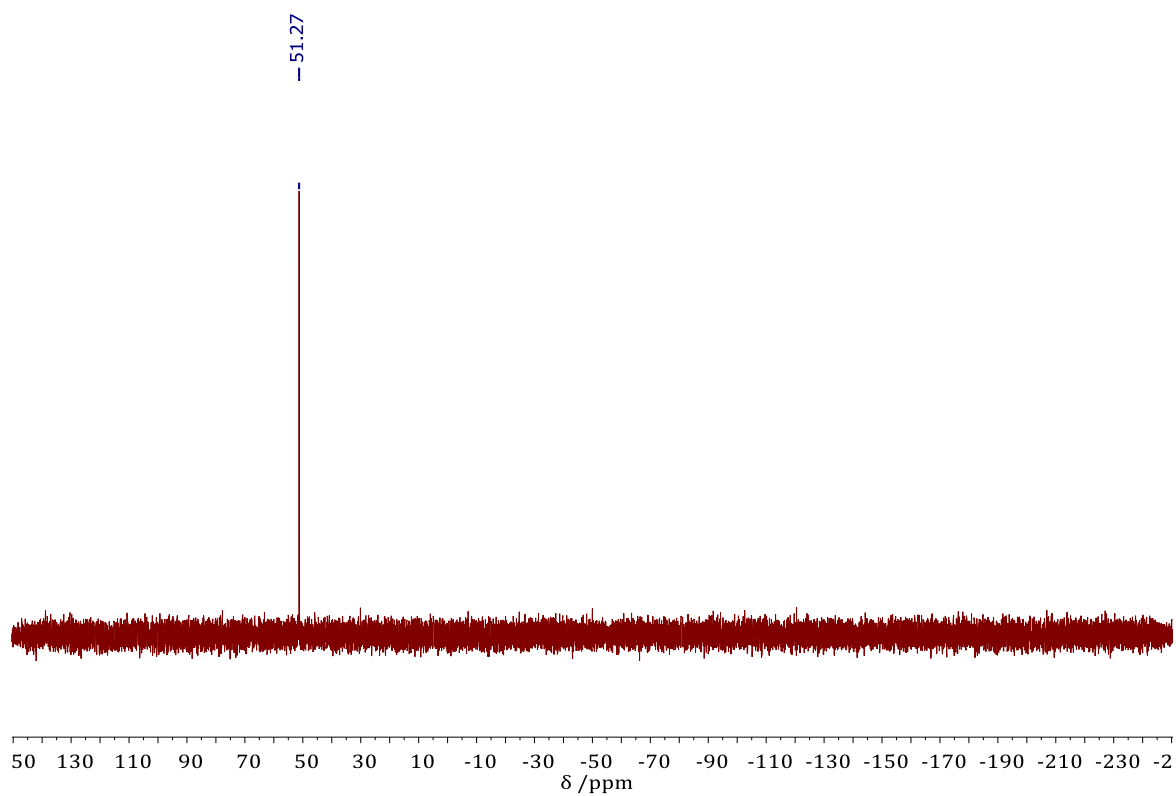


Figure S27. $^{31}\text{P}\{^1\text{H}\}$ -NMR spectrum of complex **7** in acetone- d_6 at 298 K.

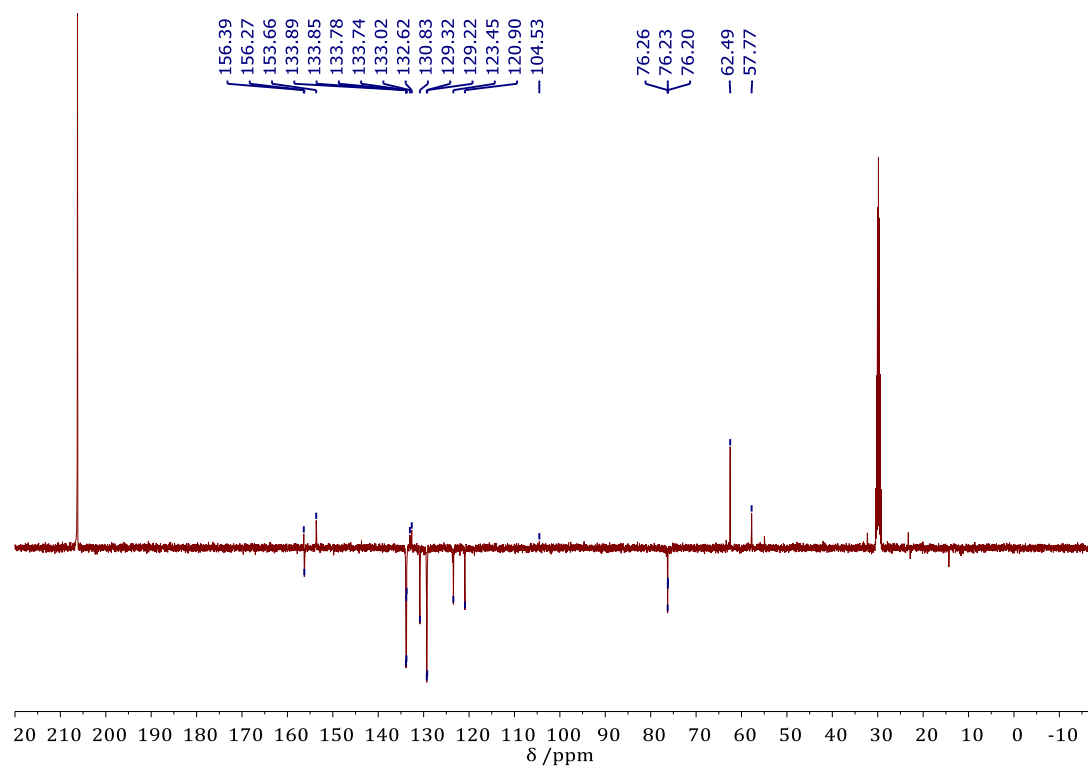


Figure S28. APT $^{13}\text{C}\{^1\text{H}\}$ -NMR spectrum of complex **7** in acetone- d_6 at 298 K.

S3. Electronic data

Table S1. Optical spectral data for compounds **1–7** in dichloromethane and dimethylsulfoxide. Measurements were performed at room temperature using 10^{-4} – 10^{-6} M solutions. (Sh = Shoulder).

Compound	λ_{max} /nm ($\epsilon \times 10^3$ / $\text{M}^{-1}\text{cm}^{-1}$)	
	Dichloromethane	Dimethylsulfoxide
1	289 (21.71), 357 (Sh), 409 (Sh), 457 (Sh)	289 (24.49), 362 (6.46), 415 (Sh)
2	290 (26.10), 366 (6.91), 412 (Sh)	289 (26.27), 371 (7.21), 412 (Sh)
3	277 (Sh), 289 (28.43), 379 (7.43), 451 (Sh)	290 (27.15), 366 (7.27), 416 (Sh)
4	291 (12.59), 345 (Sh), 416 (2.39), 479 (Sh)	293 (47.69), 358 (11.94), 414(Sh) 468 (Sh)
5	291 (24.89), 348 (Sh), 423 (4.38), 486 (Sh)	292 (16.59), 344 (Sh), 411 (2.93), 483 (Sh)
6	287 (29.20), 345 (Sh), 417 (5.28), 470 (Sh)	291 (4.11), 351 (Sh), 418 (6.86), 472 (Sh).
7	292 (8.16), 367 (Sh), 416 (1.75), 467 (Sh)	255 (Sh), 292 (5.67), 348 (Sh), 416 (1.02), 478 (Sh)

S4. X-ray crystallographic structure determination

Table S2. Crystallographic data and structural refinement details for compounds **1**, **2** and **3**.

	Compound 1	Compound 2	Compound 3
Empirical formula	C ₃₅ H ₂₈ F ₃ N ₂ O ₄ PRuS	C ₃₈ H ₃₄ Cl ₂ F ₃ N ₂ O ₄ PRuS	C ₃₇ H ₃₂ F ₃ N ₂ O ₆ PRuS
Formula weight	761.69	874.67	821.74
Temperature (K)	100(2)	100(2)	100(2)
Crystal system	Monoclinic	Triclinic	Monoclinic
space group	C 2/c	P-1	P 2 ₁ /c
a (Å)	31.2279(13)	10.807(8)	12.6767(4)
b (Å)	9.3748(4)	11.451(8)	14.7750(5)
c (Å)	21.6058(8)	16.422(12)	19.1689(6)
β (deg)	92.8700	α = 74.05(2); β = 72.092(18); γ = 77.223(17)	108.3440(10)
Volume (Å ³)	6317.3(4)	1838.(2)	3407.86(19)
Z	8	2	4
Calculated density (g cm ⁻³)	1.602	1.580	1.602
Absorption coefficient (mm ⁻¹)	0.674	0.731	0.636
Goodness-of-fit	1.060	1.109	1.093
R ₁ [I>2σ(I)]	0.0236	0.0321	0.0248
wR ₂ [I>2σ(I)]	0.0588	0.0637	0.0555

Table S3. Crystallographic data and structural refinement details for compounds **5**, **6** and **7**.

	Compound 5	Compound 6	Compound 7
Empirical formula	C ₃₅ H ₃₀ F ₃ N ₂ O ₄ PRuS	C ₃₇ H ₃₄ F ₃ N ₂ O ₄ PRuS	C ₃₇ H ₃₄ F ₃ N ₂ O ₆ PRuS
Formula weight (g mol ⁻¹)	763.71	791.76	823.76
Temperature (K)	100(2)	100(2)	100(2)
Crystal system	Monoclinic	Monoclinic	Monoclinic
space group	P 2 ₁ /n	P 2 ₁ /n	P 2 ₁ /c
a (Å)	12.995(8)	12.8677(4)	16.356(15)
b (Å)	13.774(9)	14.2849(5)	10.914(10)
c (Å)	18.311(12)	19.4057(6)	20.074(18)
β (deg)	105.75(2)	105.8660(10)	108.95(3)
Volume (Å ³)	3154.(4)	3431.14(19)	3389.(5)
Z	4	4	4
Calculated density (g cm ⁻³)	1.608	1.533	1.614
Absorption coefficient (mm ⁻¹)	0.676	0.624	0.639
Goodness-of-fit	1.051	1.115	1.156
R ₁ [I>2σ(I)]	0.0265	0.0390	0.0491
wR ₂ [I>2σ(I)]	0.0554	0.0915	0.1153

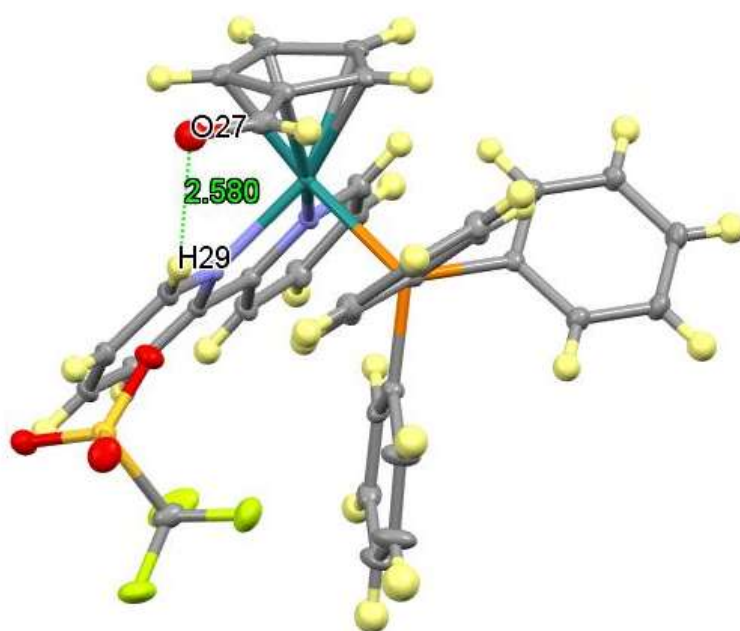
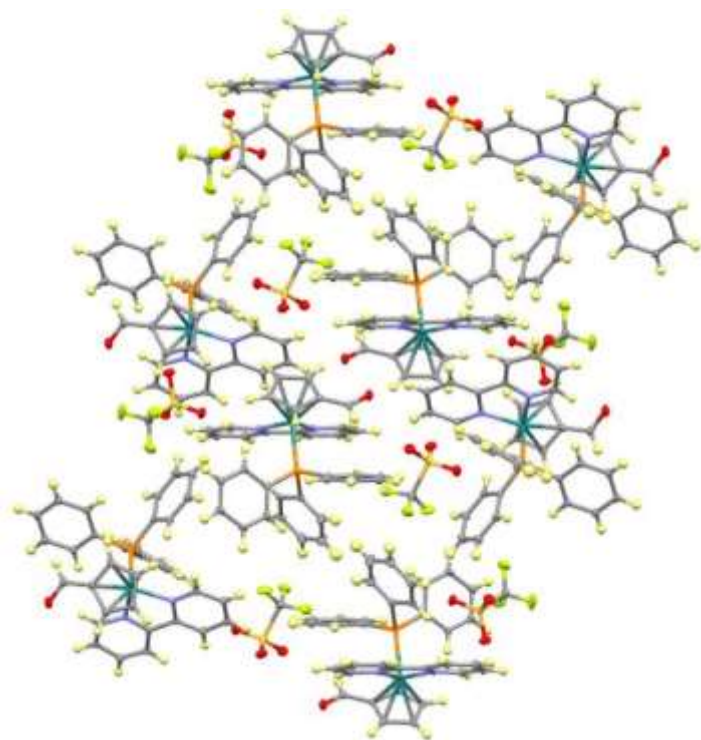
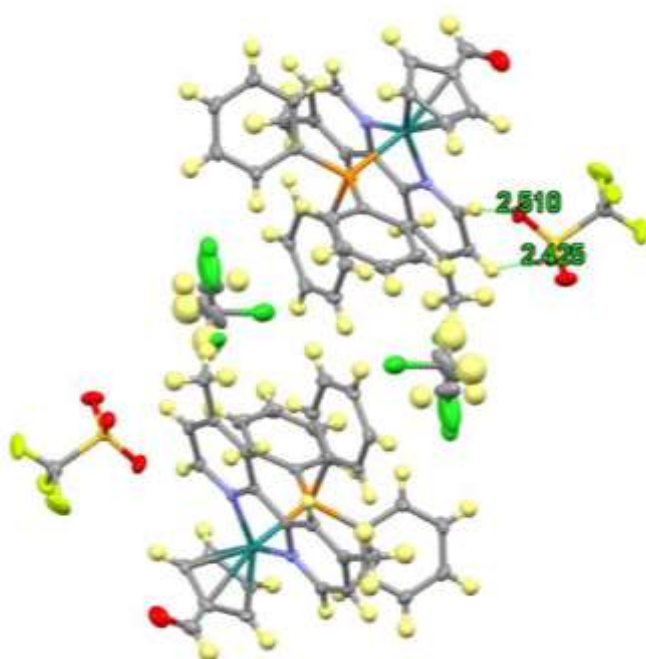


Figure S29. X-ray structure of **1** showing the intramolecular hydrogen bond.

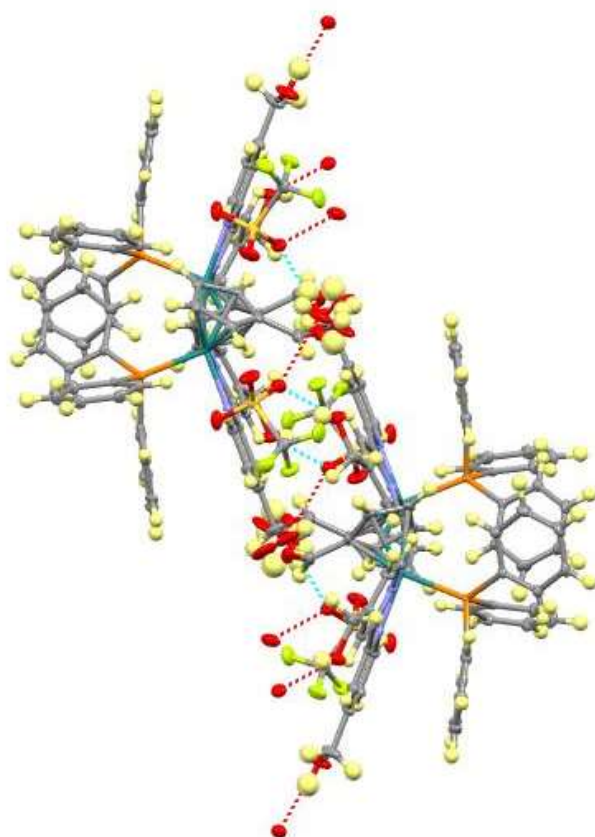
A)



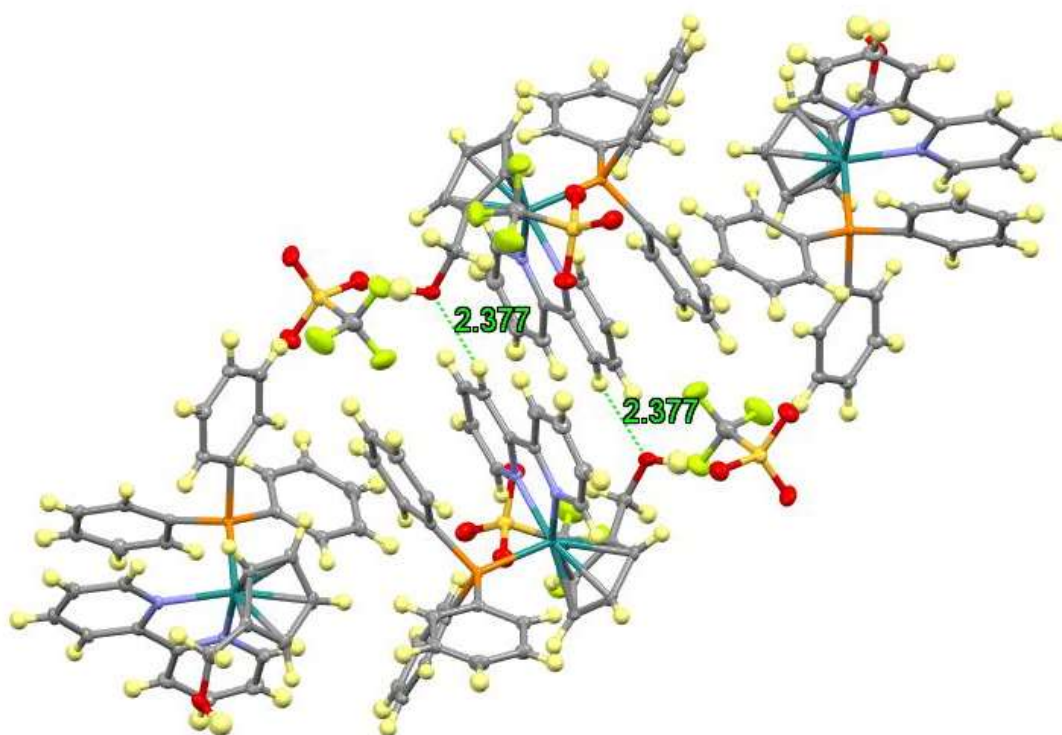
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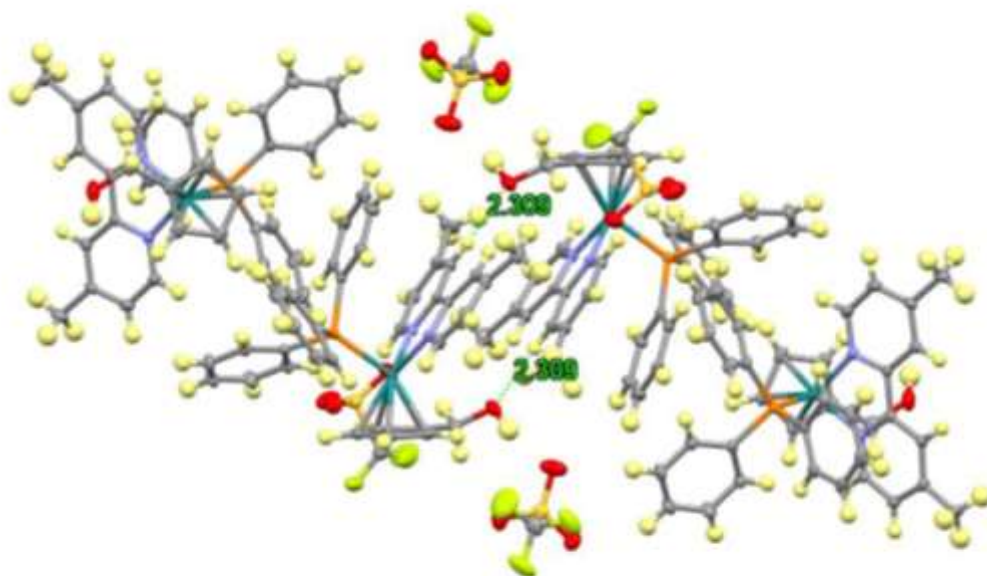
C)



D)



E)



F)

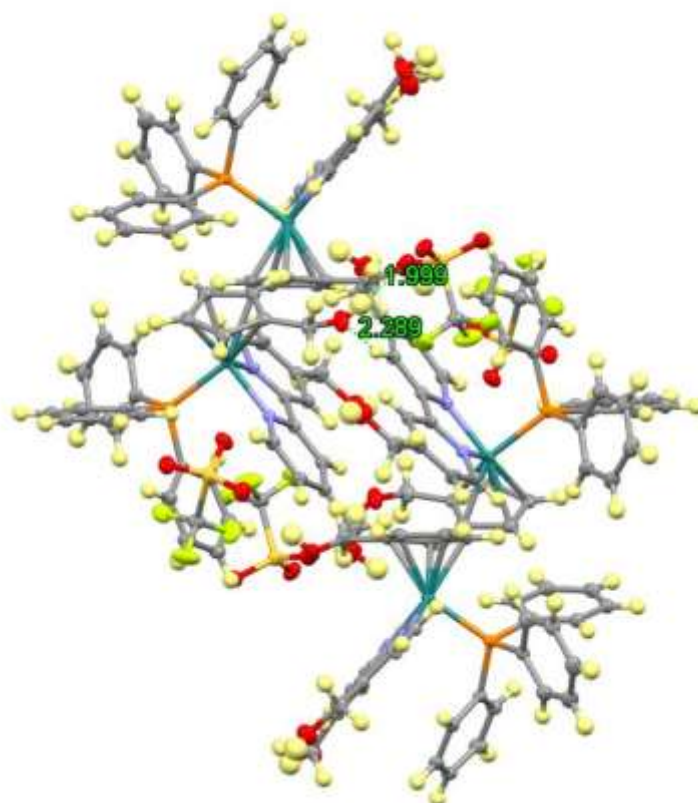
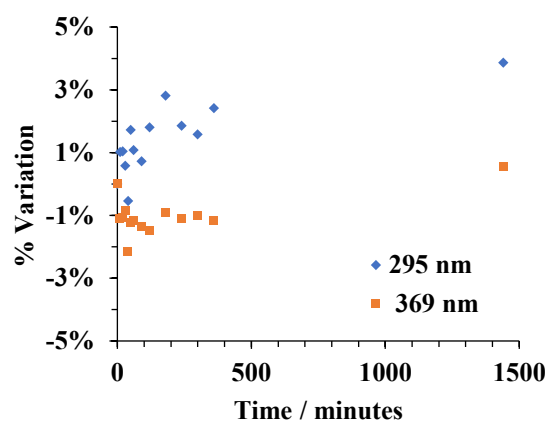
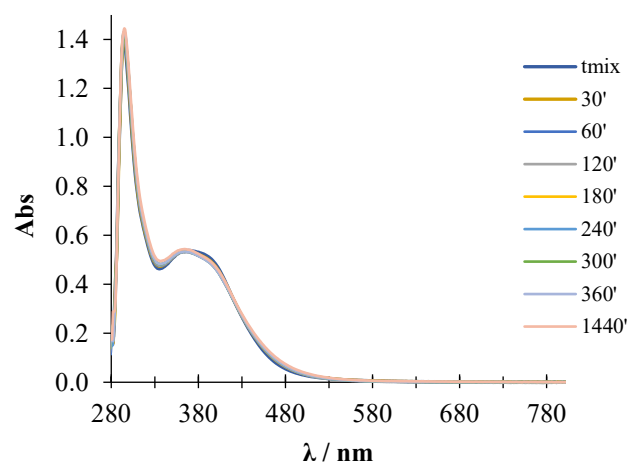


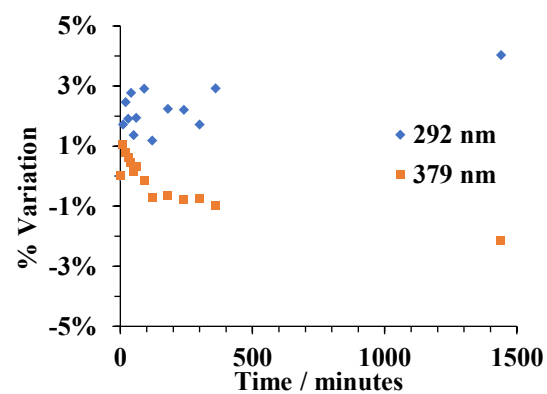
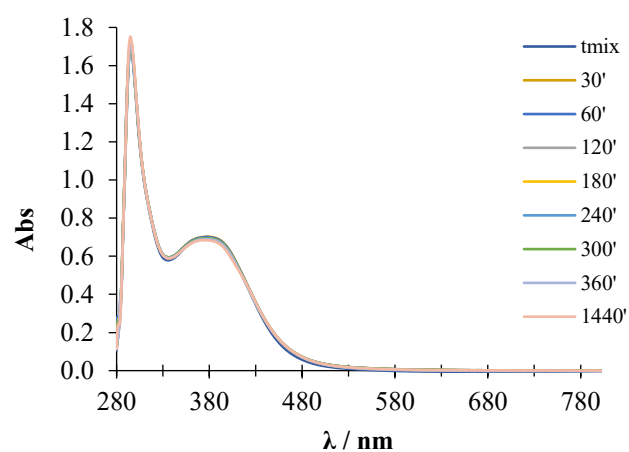
Figure S30. Packing diagram for complexes **1** (A), **2** (B), **3** (C), **5** (D), **6** (E) and **7** (F).

S5. Stability studies in aqueous solution

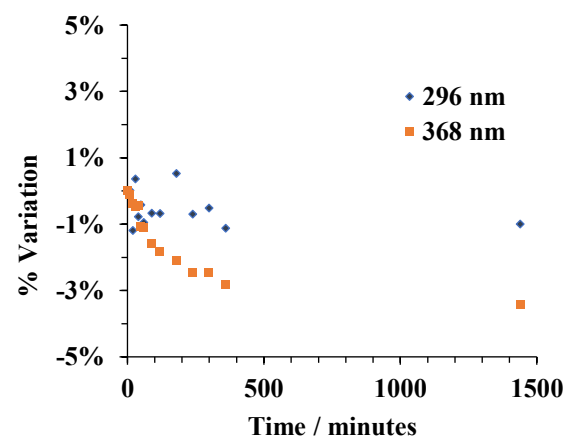
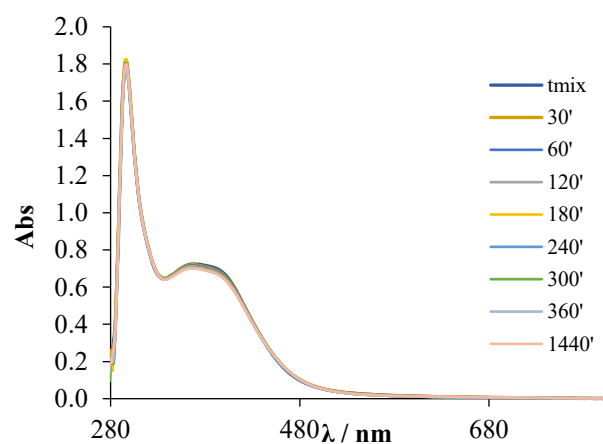
A) Compound 1



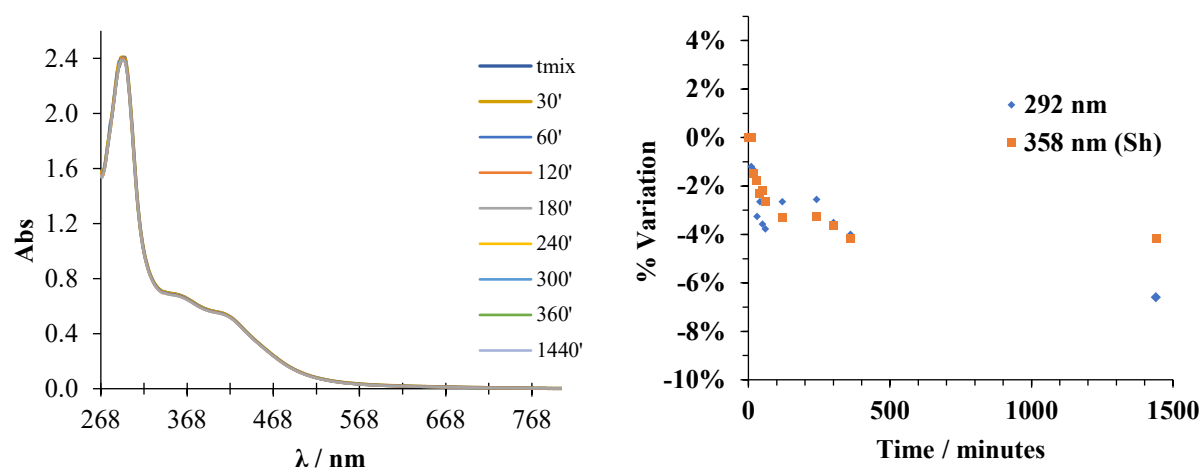
B) Compound 2



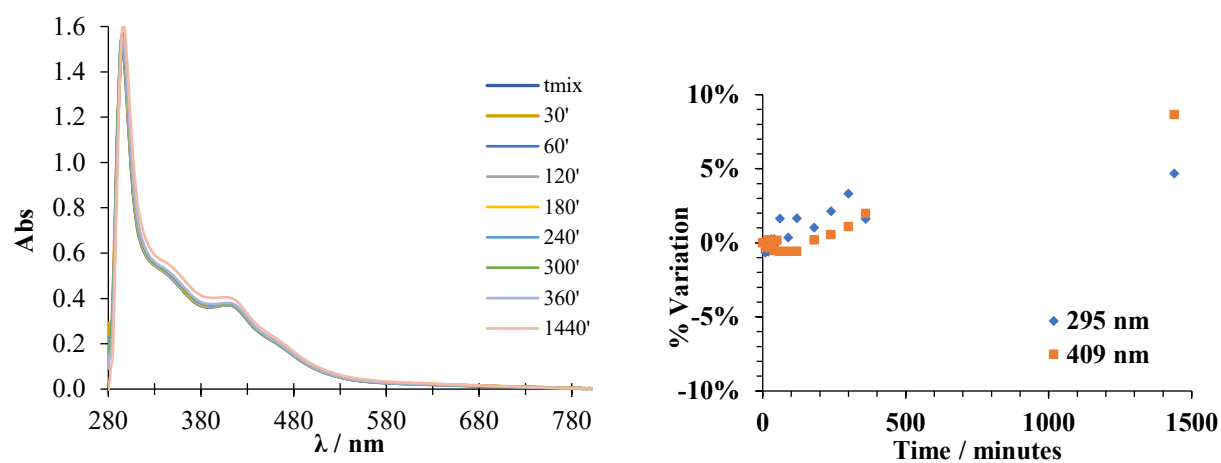
C) Compound 3



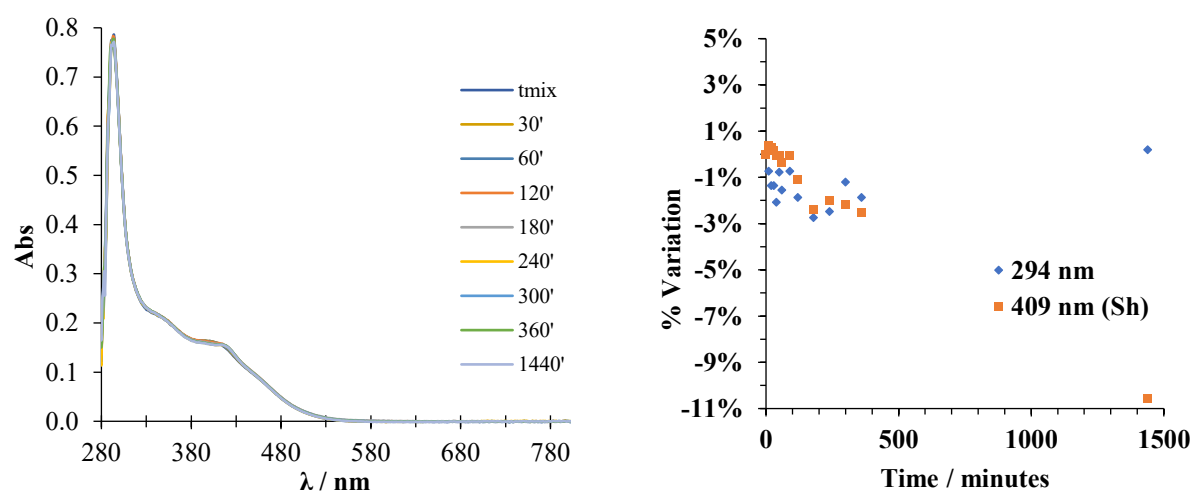
D) Compound 4



E) Compound 5



F) Compound 6



G) Compound 7

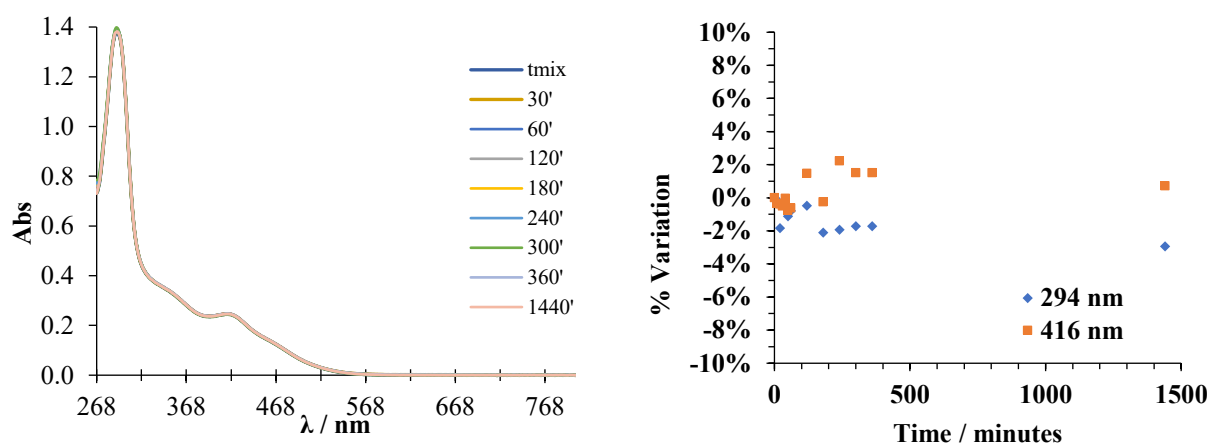
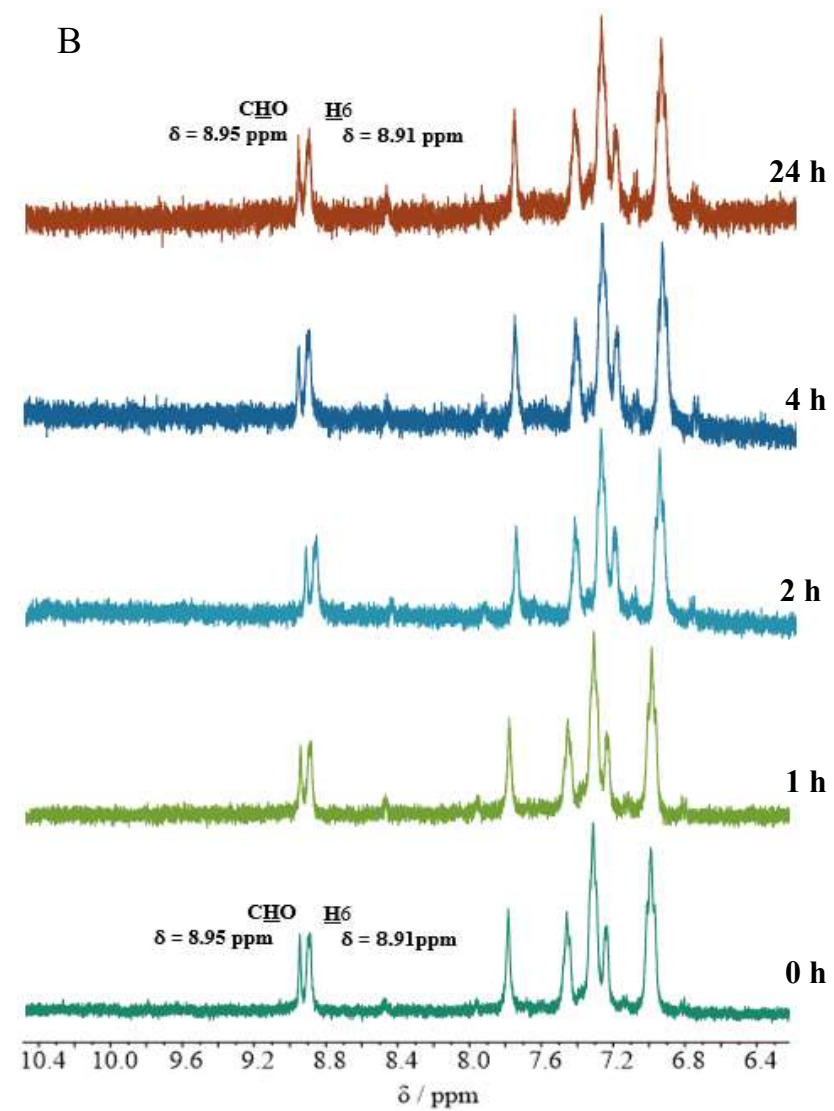
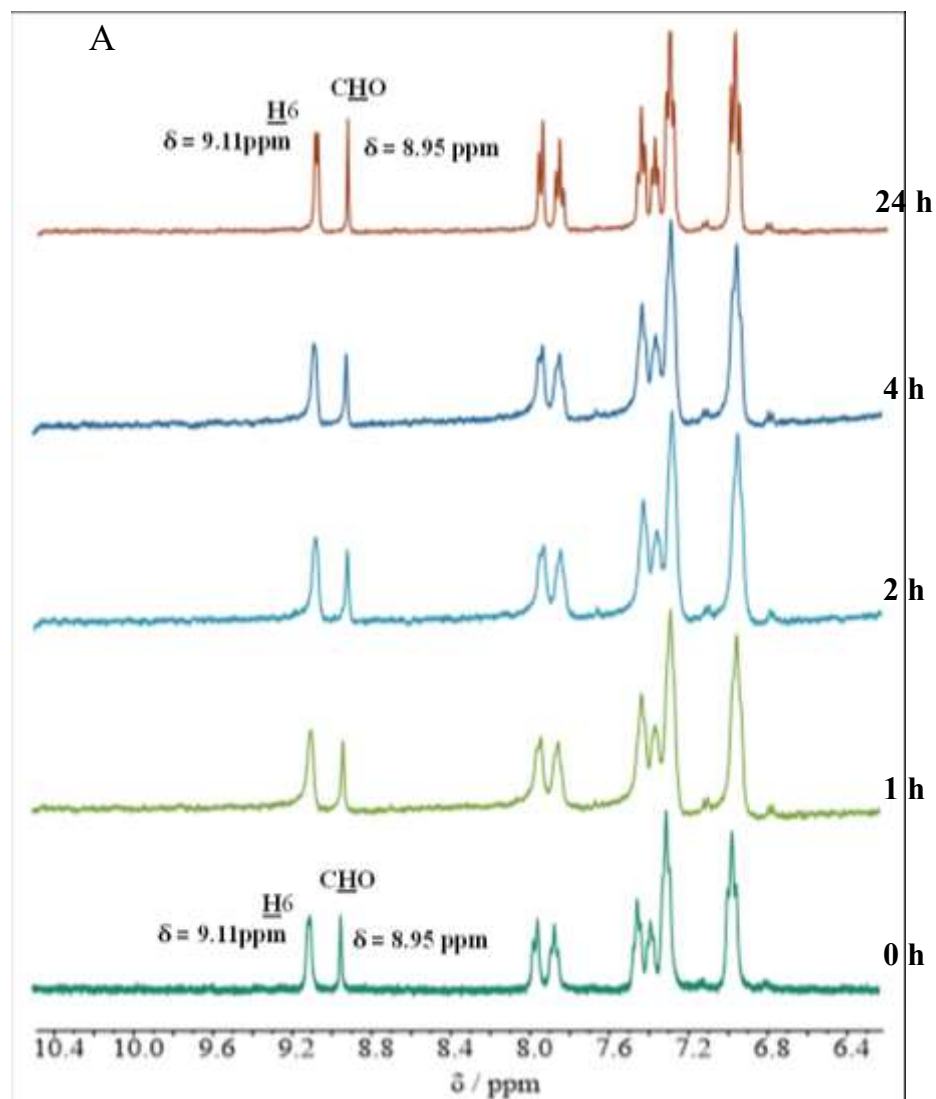


Figure S31. Stability studies in cellular media, 2% DMSO / 98 % DMEM for compounds **1** (A), **2** (B), **3** (C), **4** (D), **5** (E), **6** (F) and **7** (G). On the right are represented the UV-Vis spectra during the 24 h of the study and on the left the percentage of variation for fixed wavelengths along time (1 cm optical path; see experimental section for details).



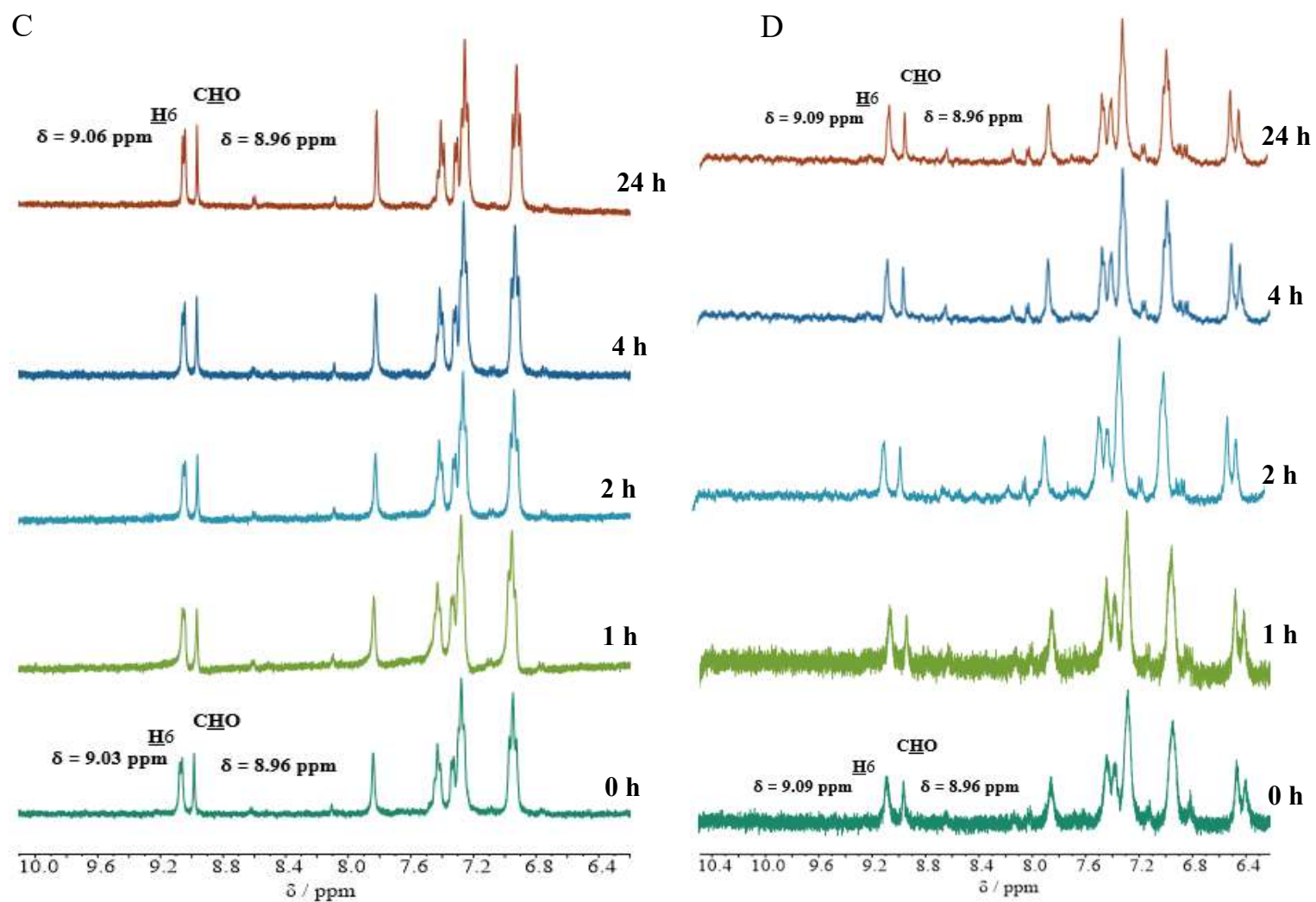


Figure S32. Stability studies monitored by ^1H -NMR in cellular media (35 % DMSO / 65 % DMEM) for compounds 1(A), 2(B), 3(C) and 4(D).

S6. P-gp and MRP1 expression in non-small cell lung cancer cell lines

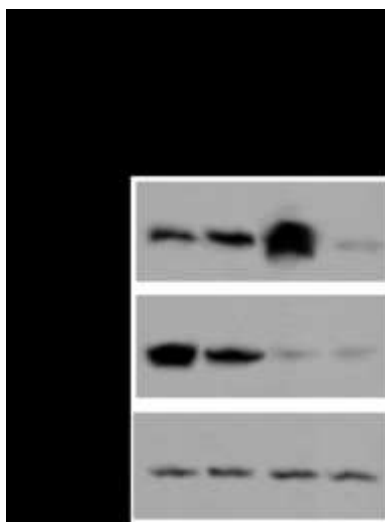


Figure S33. P-gp and MRP1 expression in non-small cell lung cancer cell lines, measured by immunoblotting. The figure is representative of 1 out of 3 experiments. Tubulin has been used to check the equal control of proteins.