

## Electronic supplementary Information (ESI)

### Carbon Nanotubes to Outperform Metal Electrodes in Perovskite Solar Cells via Dopant Engineering and Hole-Selectivity Enhancement

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## Experimental Methods

**Aerosol CNT films Preparation.** CNTs were synthesised by an aerosol (floating catalyst) CVD method based on ferrocene vapour decomposition in a CO atmosphere. The catalyst precursor was vaporised by passing ambient temperature CO through a cartridge filled with ferrocene powder. The flow containing ferrocene vapour was then introduced into the high-temperature zone of a ceramic tube reactor through a water-cooled probe and mixed with additional CO. To obtain a stable growth of CNTs, a controlled amount of CO<sub>2</sub> was added together with the carbon source, CO. CNTs were directly collected downstream of the reactor by filtering the flow through a nitrocellulose membrane filter (Millipore Corp., USA; HAWP, 0.45 µm pore diameter). The time of collection determines the transparency and the conductivity of the CNT films.

**MAPbI<sub>3</sub> solution preparation.** CH<sub>3</sub>NH<sub>3</sub>I (Aldrich), PbI<sub>2</sub> (TCI), and anhydrous dimethyl sulfoxide (TCI) (molar ratio 1:1:1) were mixed in anhydrous *N,N*-dimethylformamide (TCI) with a concentration of 50 wt%. The solution was filtered through a 0.2 µm poly(tetrafluoroethylene) filter before use.

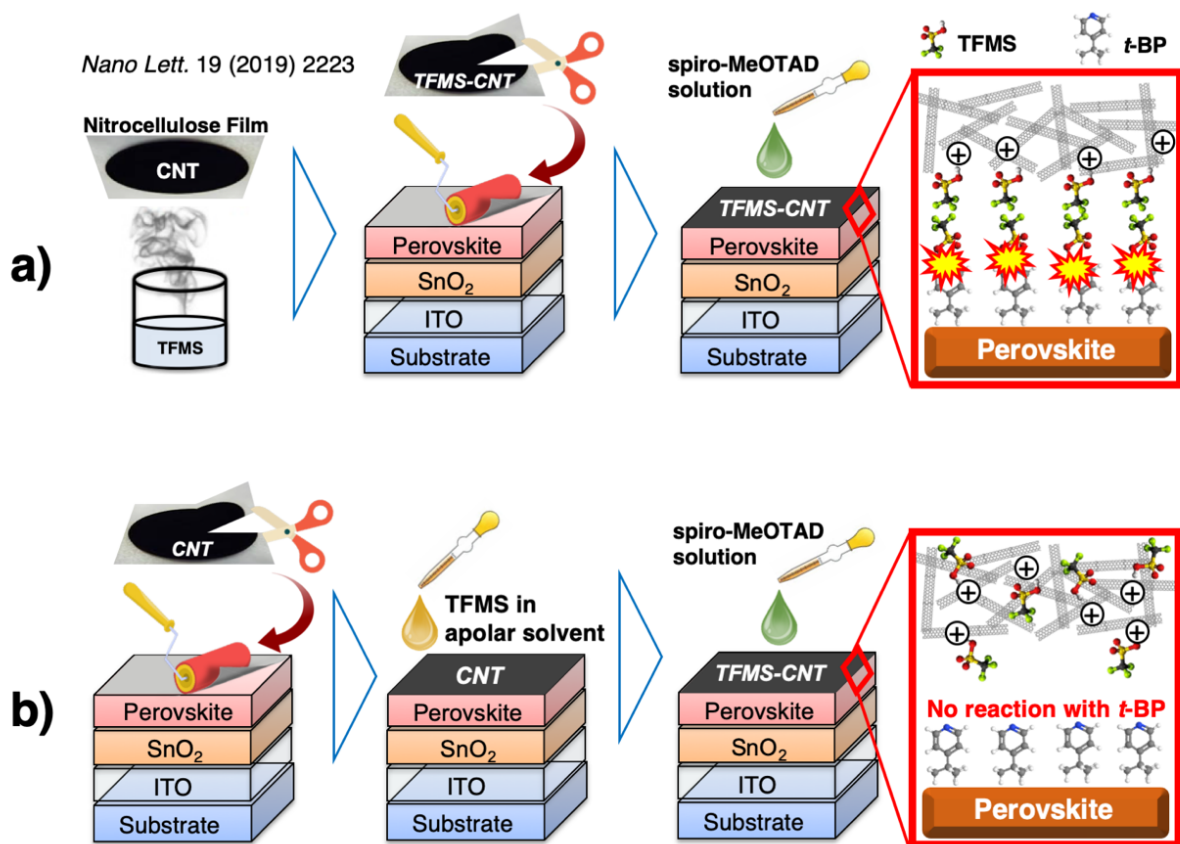
**Perovskite Solar Cell Fabrication.** ITO patterned glass substrates were cleaned and sonicated with detergent, distilled water, acetone, and isopropanol in an ultrasonic bath for 15 min, respectively. The cleaned substrates underwent the UV-ozone for wettability enhancement and removal of any organic contamination. 30 millimolar SnCl<sub>2</sub>·2H<sub>2</sub>O (Aldrich, >99.995%) solution in ethanol (anhydrous, Fujifilm Wako Pure Chemical Co.) as a precursor solution was used for deposition of a compact SnO<sub>2</sub> layer. The solution was filtered by a 0.2 µm syringe filter, followed by spin-coating on the cleaned substrate at 3000 rpm for 30 s. The spin-coated film was annealed at 165 °C for 30 min. After cooling down to room temperature, another cycle of the spin-coating

process was repeated, which was followed by annealing at 150 °C for 5 min and 190 °C for 1 h. The SnO<sub>2</sub>-coated ITO glass was treated with UV-ozone before spin-coating of the prepared perovskite solution. Then, 20 µL of perovskite precursor solution was spin-coated onto the SnO<sub>2</sub> layer at 4000 rpm for 20 s, with 0.15–0.30 mL of anhydrous diethyl ether slowly dripped onto the substrate 12 s after the start of the spin-coating process. Next, the film was annealed at 100 °C for 10 min to obtain a dense brown MAPbI<sub>3</sub> film.

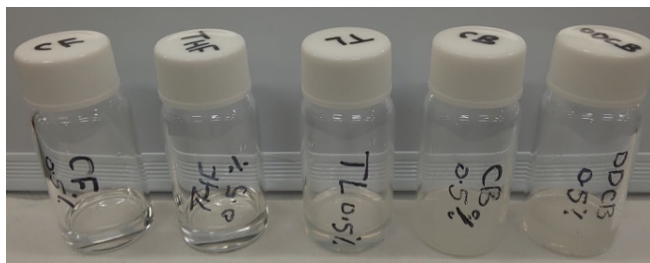
For the CNT-based devices, pre-cut CNT films of different transparencies (active area 5x2 mm<sup>2</sup>) were laminated on the top of the prepared MAPbI<sub>3</sub>/SnO<sub>2</sub>/ITO/glass substrates to room temperature by applying press-transfer method on the perovskite surface just after cooling. Before depositing the hole transporting layer (HTL), a 20 µL solution containing different concentrations of TFMS in ODCB (0.01, 0.015, 0.02, 0.025 and 0.05 wt%) was spin-coated on the CNT-laminated PSCs at 300 rpm for 30 s. Then, the HTL solution was prepared by dissolving 85.8 mg of spiro-MeOTAD (Merck) in 1 mL of chlorobenzene (anhydrous, 99.8%, Sigma-Aldrich) which was mixed with 33.8 µL of 4-tert-butylpyridine (96%, Aldrich) and 19.3 µL of Li-TFSI (99.95%, Aldrich, 520 mg mL<sup>-1</sup> in acetonitrile) solution. Different concentrations of HTL tested were by a factor of 0.9, 0.95, 1.0, 1.05, 1.085, 1.17, 1.25, and 1.4. The spiro-MeOTAD solution was spin-coated on the perovskite layer at 3000 rpm for 30 s by dropping 20 µL of the solution. The fabrication was finally completed by thermal evaporation of a 50-nm-thick film of gold electrode (3x3 mm<sup>2</sup> active area) for the reference devices only at a constant evaporation rate of 0.05 nm s<sup>-1</sup>.

**Characterisations.** The *J–V* curves were measured using a software-controlled source meter (Keithley 2400 Source-Meter) under dark conditions and the simulated sunlight irradiation of 1 sun (AM 1.5G; 100 mW cm<sup>-2</sup>) using a solar simulator (EMS- 35AAA, Ushio Spax Inc.) with a Ushio Xe short arc lamp 500. The source meter was calibrated using a silicon diode (BS-520BK,

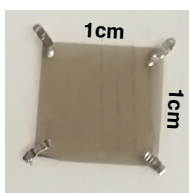
Bunkokeiki). By using Agilent 4156C analyser with a four-probe station, the sheet resistance of CNT films were measured (van der Pauw method). The SEM analysis of the CNTs and device cross-section were performed using an S-4800 (Hitachi) electron microscope. The SEM images were analysed by ImageJ software. An inVia Raman microscope (Renishaw) was employed for the vibrational spectra observation of CNT and doped-CNT electrodes with 532 nm laser wavelength. Shimadzu UV-3150 was used for the UV–Vis–NIR measurement. The PL measurements were performed using JASCO Spectrofluorometer (FP-8300). The valence band and the Fermi levels measurements were performed using Riken Keiki PYS-A AC-2 and Kelvin probe spectroscopy in air (ESA), respectively. A source measurement unit was used to record the current at each specific wavelength. Solartron SI1287 Electrochemical Interface and Solartron 1255B Frequency Response Analyser were used for the Impedance Measurement.



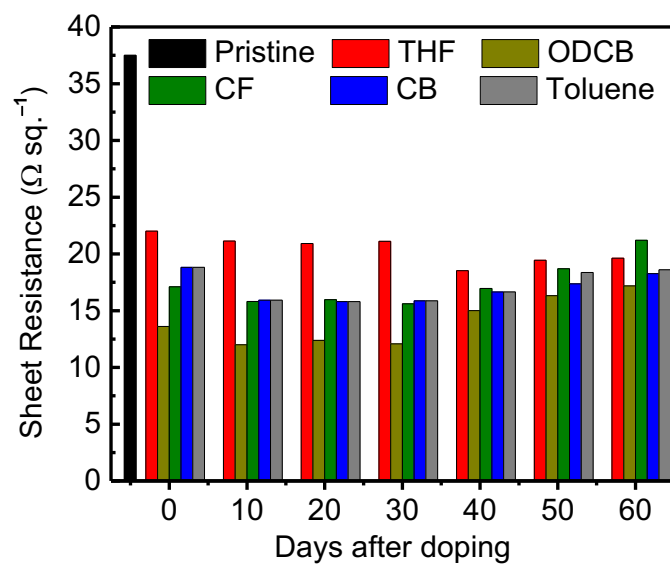
**Fig. S1.** Illustration of top CNT electrode *p*-doping and fabrications a) of our previous work (Reproduced with permission [17] Copyright 2017, American Chemical Society) and b) of this work.



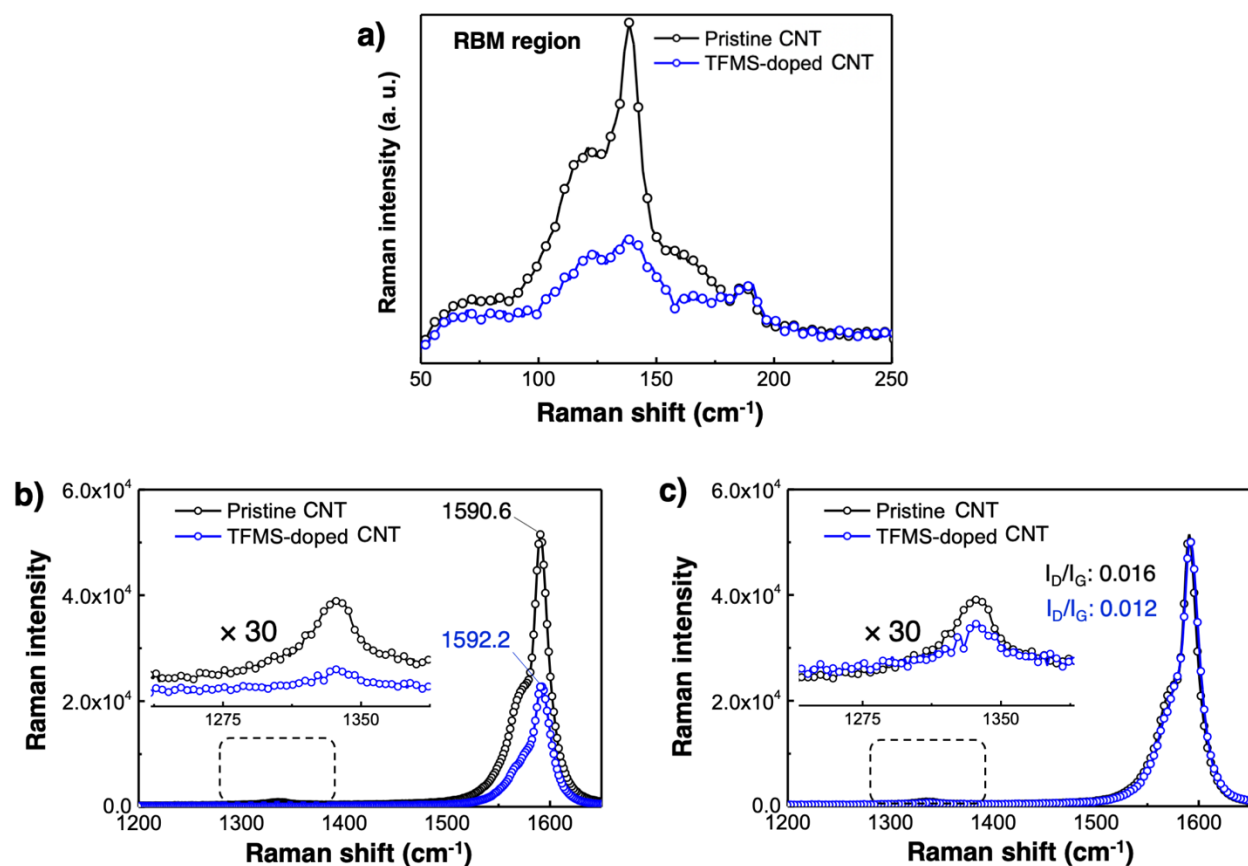
**Fig. S2.** TFMS dispersed in various non-polar solvent (CF, THF, toluene, CB and ODCB). The solution appears transparent or murky depending on solvents.



**Fig. S3.** Indium contact applied T60% CNT for the four probe Van der Pauw method.



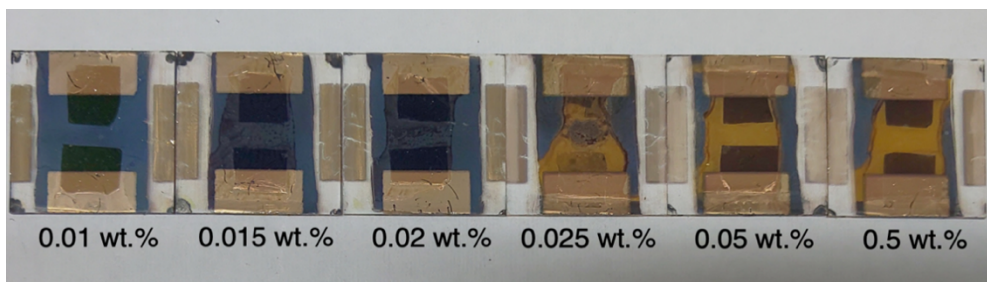
**Fig. S4.** A bar graph showing the sheet resistance changes upon TFMS doping in different apolar solvents and their durability.



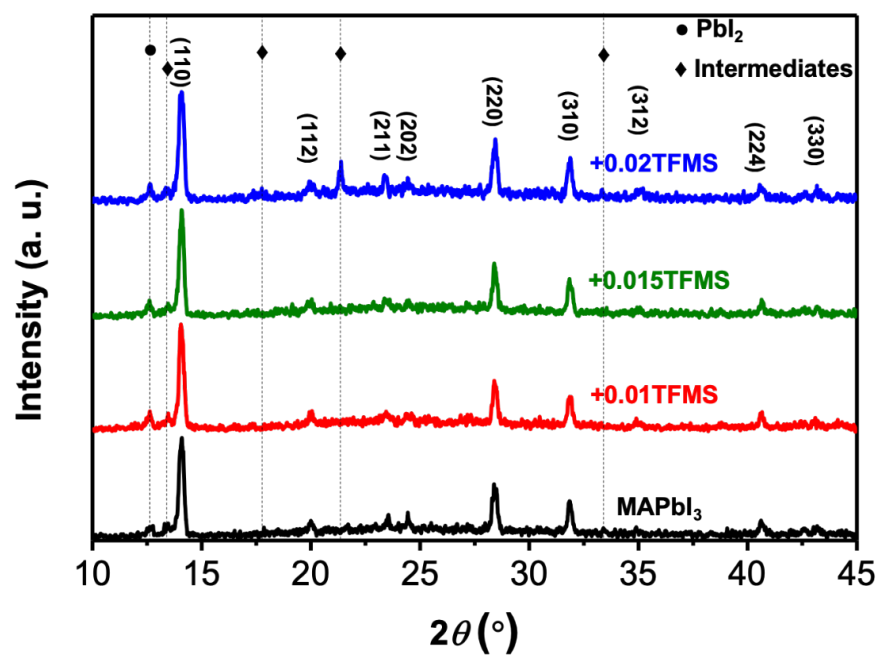
**Fig. S5.** a) RBM region, b) the G-band, and c) The G-to-D ratio of the Raman spectra of a pristine CNT film (black), and a TFMS-applied CNT film.

**Table S1.** Photovoltaic parameters of varying concentrations of TFMS-applied CNT-based PSCs under AM1.5G 1 sun illumination.

TFMS conc.	$J_{sc}$ (mA cm <sup>-2</sup> )	$V_{oc}$ (V)	FF	PCE (%) [average $\pm$ error]
0	19.8	0.97	0.59	11.3 [10.3 $\pm$ 0.73]
0.010 wt.%	20.3	1.00	0.60	12.1 [10.8 $\pm$ 0.90]
0.015 wt.%	20.6	1.03	0.62	13.2 [12.4 $\pm$ 0.60]
0.020 wt.%	15.9	0.87	0.49	6.9 [6.5 $\pm$ 0.35]
0.025 wt.%	15.5	0.80	0.40	5.0 [4.5 $\pm$ 0.63]
0.050 wt.%	4.6	0.66	0.33	1.1 [0.84 $\pm$ 0.28]



**Fig. S6.** Pictures of the CNT-PSCs upon various concentrations of TFMS application.

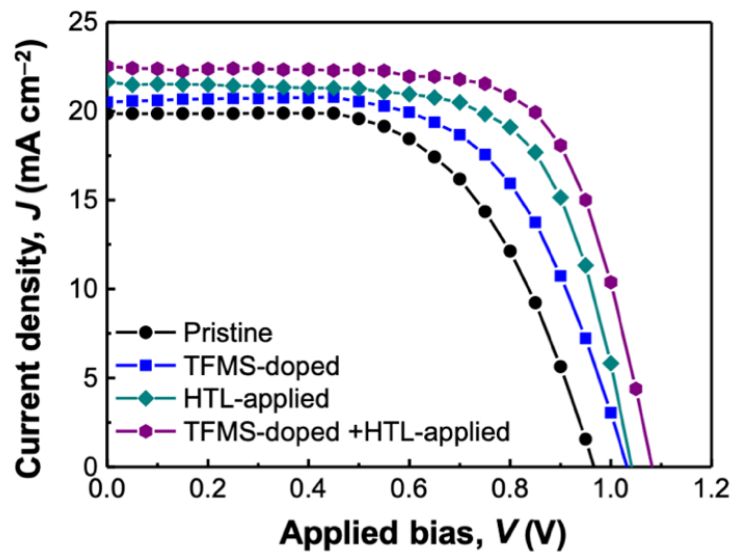


**Fig. S7.** XRD spectra of the perovskite films after various concentrations of TFMS doping.

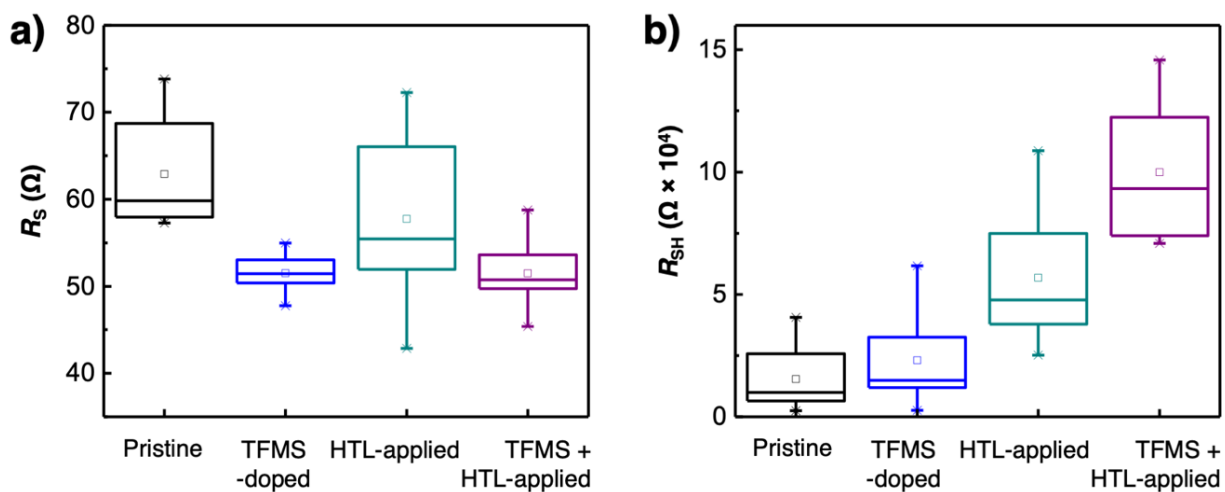


**Table S2.** Photovoltaic parameters of CNT-PSCs upon TFMS doping and spiro-MeOTAD application under AM1.5G 1 sun illumination.

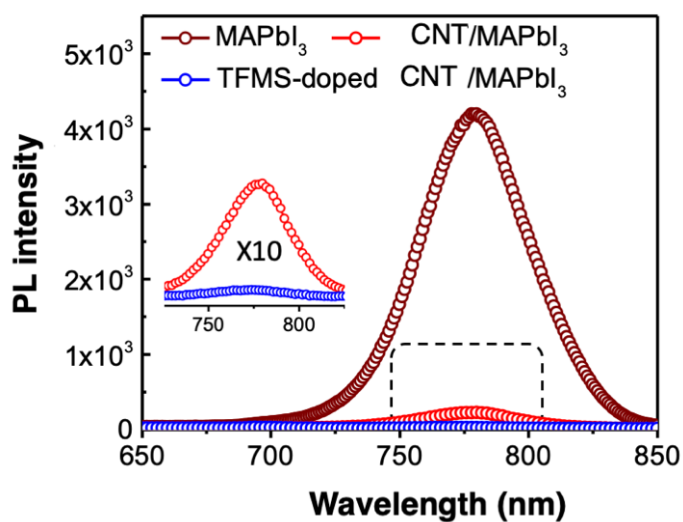
Device type	$J_{sc}$ (mA cm <sup>-2</sup> )	$V_{oc}$ (V)	FF	$R_s$ ( $\Omega$ )	$R_{sh}$ ( $\times 10^4 \Omega$ )	PCE
Pristine	19.8	0.97	0.59	62.9 $\pm$ 7.6	1.5 $\pm$ 1.3	11.4%
TFMS-doped	20.6	1.03	0.62	51.5 $\pm$ 2.4	2.3 $\pm$ 2.1	13.2%
HTL-applied	21.7	1.04	0.67	57.7 $\pm$ 9.8	5.6 $\pm$ 2.7	15.3%
TFMS-doped & HTL-applied	1.08	1.08	0.69	51.5 $\pm$ 4.7	10 $\pm$ 3.0	17.0%



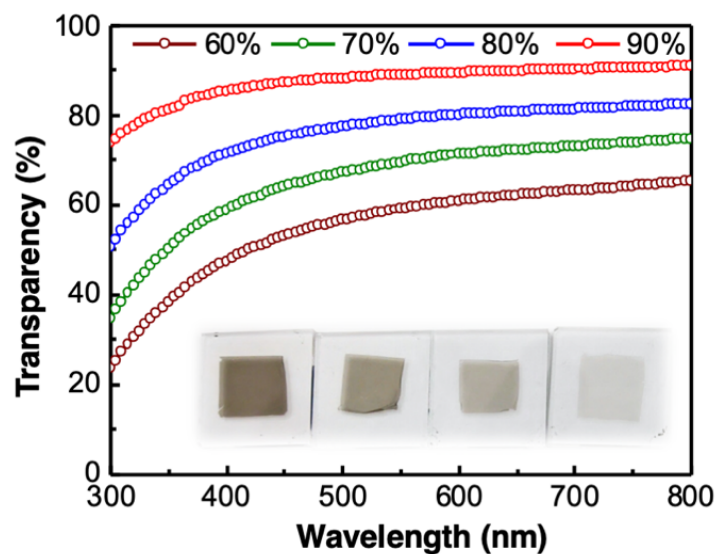
**Fig. S8.**  $J$ - $V$  curves of the CNT-PSC (black circles), the 0.15 wt.% TFMS-doped CNT-PSC (blue squares), the HTL-applied CNT-PSC (turquoise diamonds), and the 0.15 wt.% TFMS-doped HTL-applied CNT-PSC (purple circles).



**Fig. S9.** a)  $R_s$  and b)  $R_{sh}$  values of the CNT-PSC (black), the TFMS-doped CNT-PSC (blue), the HTL-applied CNT-PSC (turquoise), and the TFMS-doped HTL-applied CNT-PSC (purple).



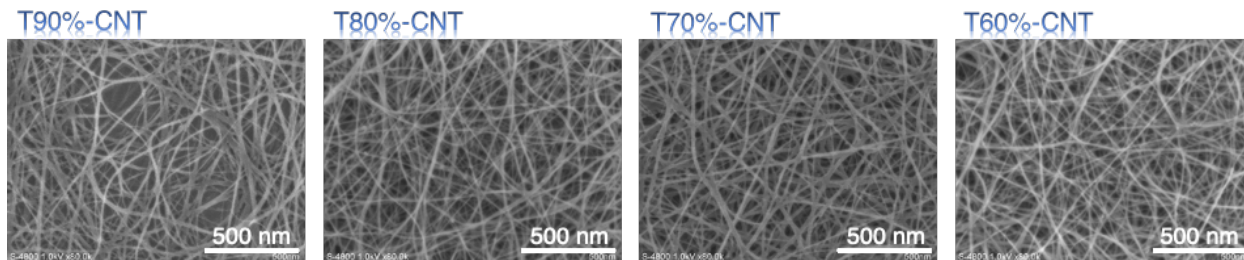
**Fig. S10.** PL spectra of the MAPbI<sub>3</sub> film (brown), CNT on the MAPbI<sub>3</sub> film (red), and TFMS-doped CNT on the MAPbI<sub>3</sub> film (blue).



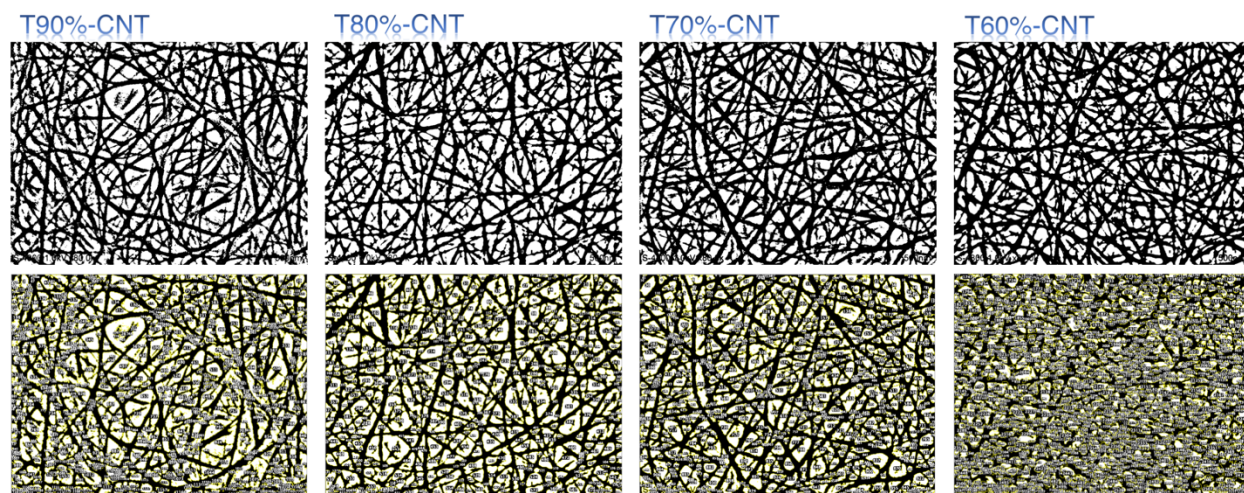
**Fig. S11.** Transmittance spectra of CNT films with different densities, namely, T90%-CNT (red squares), T80%-CNT (blue squares), T70%-CNT (green squares), and T60%-CNT (brown squares) and the photos as an inset.

**Table S3.** Sheet resistance of CNT films with different densities before and after doping.

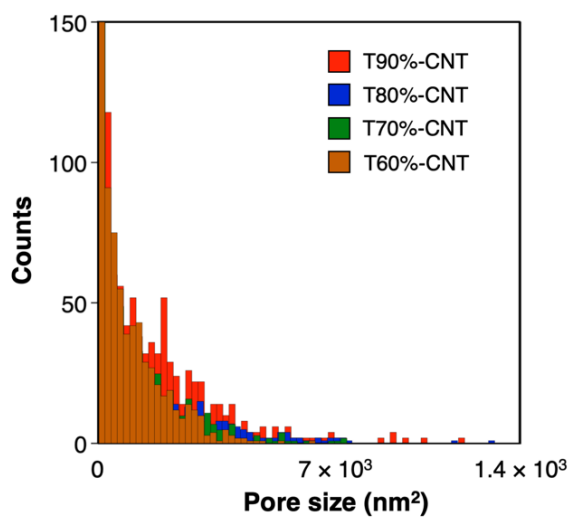
CNT type	Sheet Resistance ( $\Omega \text{ sq}^{-1}$ )	After 0.05 wt.%TFMS doping ( $\Omega \text{ sq}^{-1}$ )
T90%	100 $\pm$ 12.1	81 $\pm$ 14.0
T80%	80.9 $\pm$ 1.25	68.8 $\pm$ 2.74
T70%	57.4 $\pm$ 1.72	35.5 $\pm$ 3.98
T60%	36.5 $\pm$ 0.88	28.4 $\pm$ 1.82



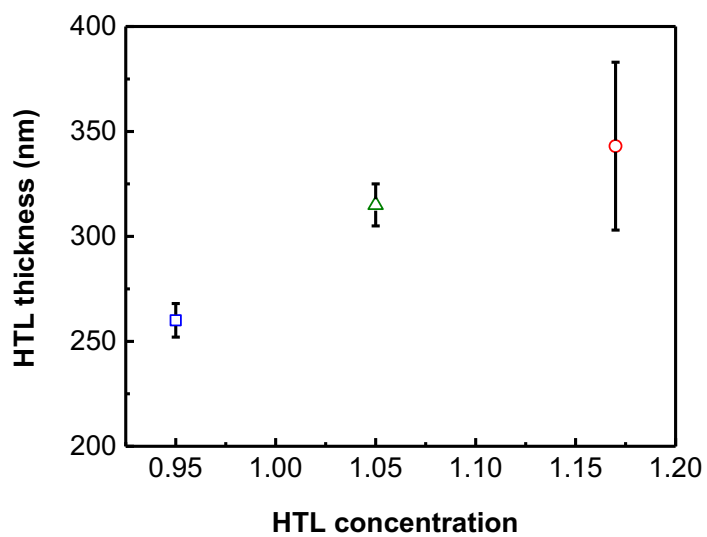
**Fig. S12.** SEM images of T90%-CNT, T80%-CNT, T70%-CNT and T60%-CNT.



**Fig. S13.** Processed SEM images of T90%-CNT, T80%-CNT, T70%-CNT and T60%-CNT.



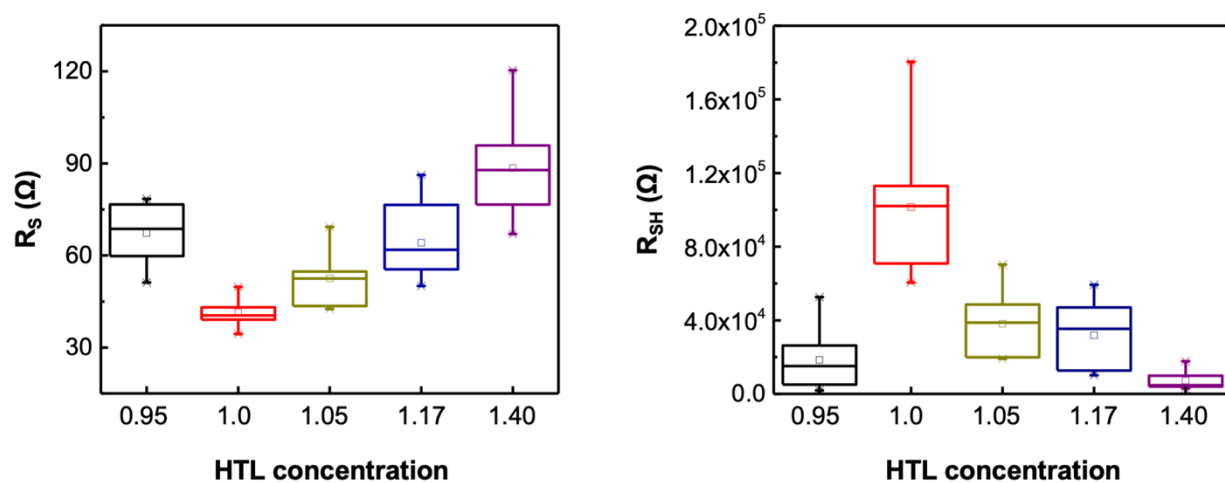
**Fig. S14.** Pore size distribution of different CNT density films as indicated.



**Fig. S15.** Thickness of spiro-MeOTAD on a perovskite film measured by a profilometer.

**Table S4.** Thickness of spiro-MeOTAD on a perovskite film measured by a profilometer.

HTL concentration	Thickness(nm)
x 0.95	260±8
x 1.05	315±10
x 1.17	343±40

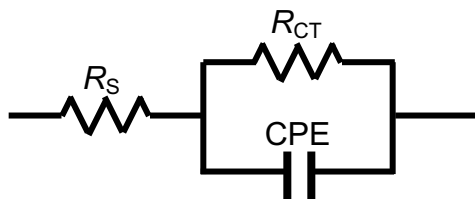


**Fig. S16.**  $R_s$  and  $R_{sh}$  of different spiro-MeOTAD concentrations in Au-based PSC devices.

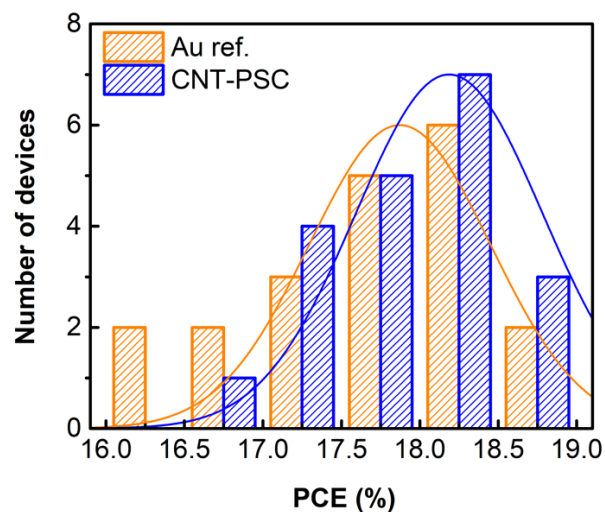
**Table S5.**  $R_s$  and  $R_{sh}$  of different spiro-MeOTAD concentrations in Au-based PSC devices.

HTL concentration	$R_s$ ( $\Omega$ )	$R_{sh}$ ( $\times 10^4 \Omega$ )
x 0.95	67.3 $\pm$ 10.1	1.8 $\pm$ 1.7
x 1.00	41.1 $\pm$ 4.4	10.1 $\pm$ 3.7
x 1.05	52.5 $\pm$ 8.8	3.8 $\pm$ 1.9
x 1.17	64.1 $\pm$ 12.8	3.2 $\pm$ 1.8
x 1.40	88.5 $\pm$ 16.8	0.7 $\pm$ 0.5

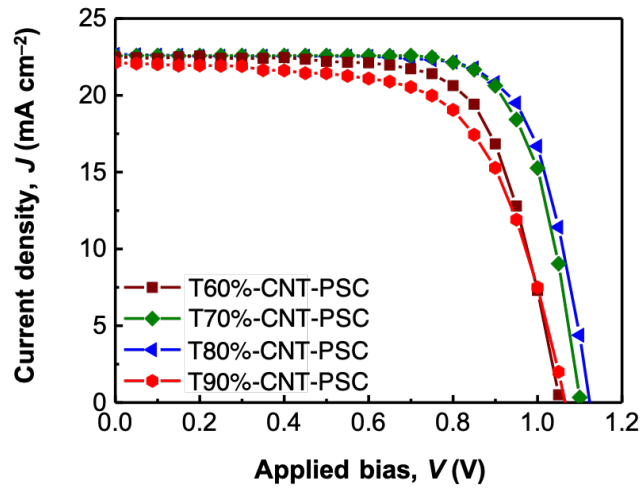
**Table S6.**  $R_S$  and  $R_{CT}$  values derived from the Nyquist plot of the EIS measurement.



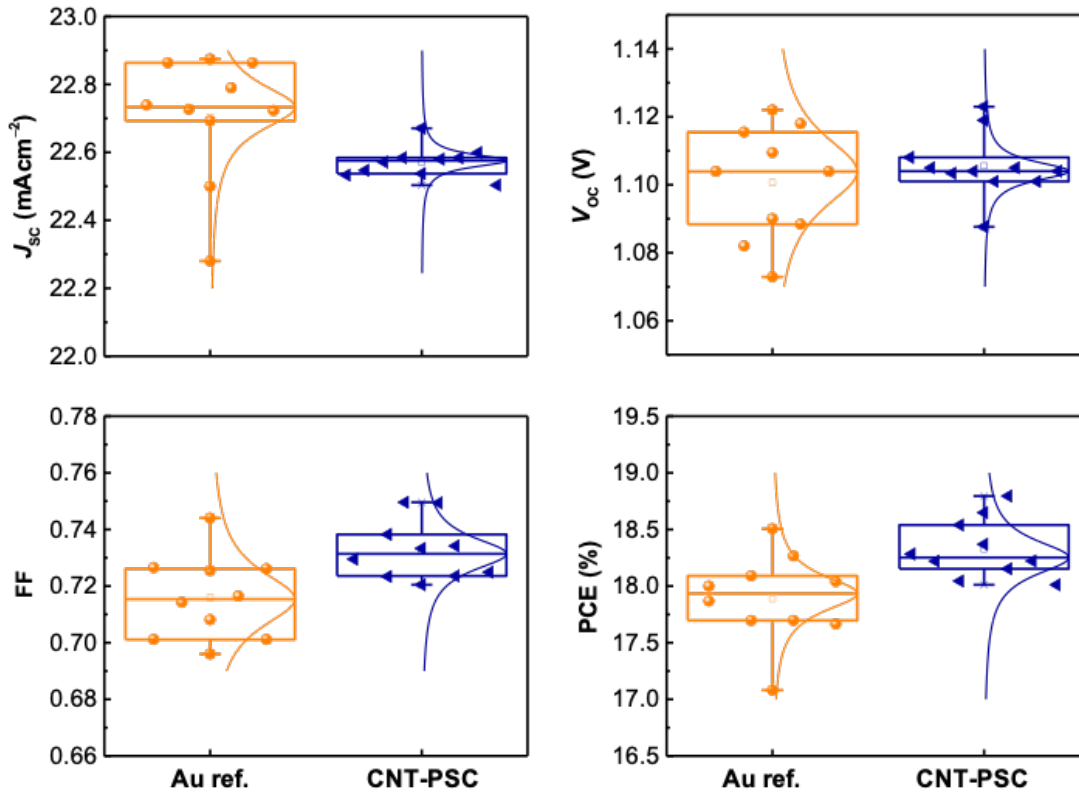
HTL conc.	$R_S$ ( $\Omega$ )	$R_{CT}$ ( $\Omega$ )
x0.95	32.5	106.2
x1.05	26.2	142.8
x1.17	23.5	157.1



**Fig. S17.** Statistical analysis of PCEs of the Au-based PSCs and the CNT-based PSCs collected from 20 devices each.

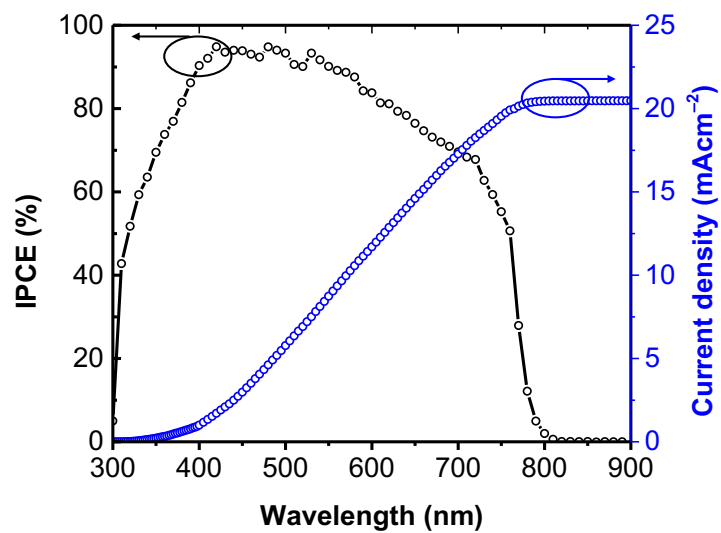


**Fig. S18.**  $J$ - $V$  curves of the champion T60%-CNT-PSC (brown square), T70%-CNT-PSC (green diamond), T80%-CNT-PSC (blue triangle) and T90%-CNT-PSC (red hexagon).

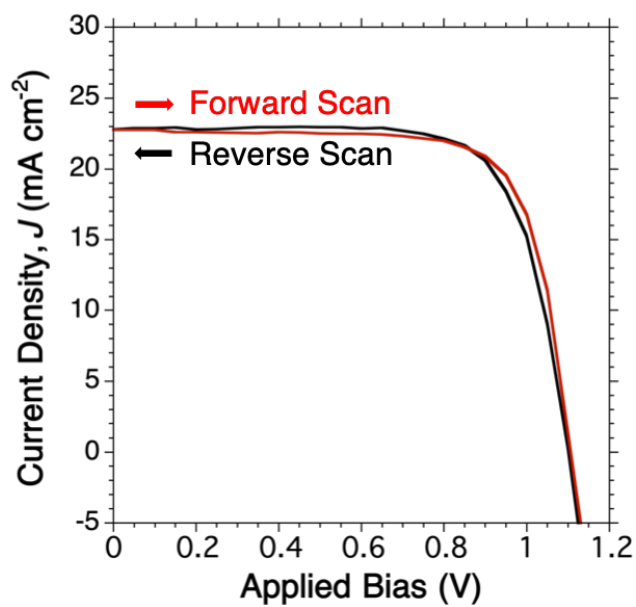


**Fig. S19.** Statistical analyses of the photovoltaic parameters from the best 10 devices with the device points (Fig. 5c).





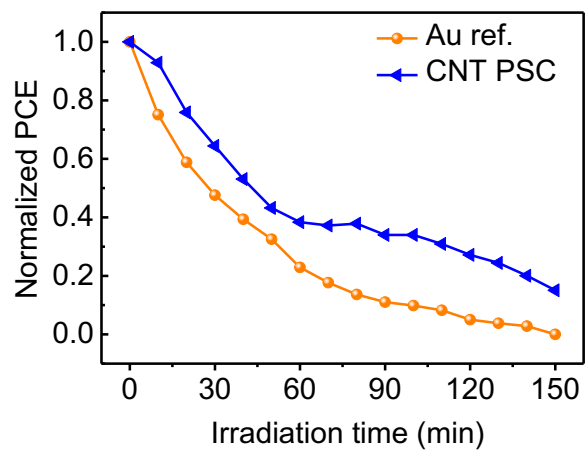
**Fig. S20.** IPCE (black line) and accumulative current density (red line) of the T80%-CNT-PSC.



**Fig. S21.**  $J$ - $V$  curves of the T80%-CNT-PSC with both scan directions.

**Table S7.** Reported PCEs of the CNT-based PSCs. SWNT stands for single-walled CNT, DWNT stands for double-walled CNT, CSCNT stands for cross-stacked CNT and MWNT stands for multi-walled CNT. The citation numbers are referenced to the reference list of the main text.

	Year	CNT Type	Electrode Type	PCE (%)	Control (%)	Citation
n/a	2020	SWNT	Top Electrode	18.8	18.4	This work
a	2019	DWNT	Bottom Electrode	17.2	n/a	[42]
b	2018	SWNT	Top Electrode	17.56	n/a	[17]
c	2018	SWNT	Top Electrode	15.3~17	n/a	[13]
d	2018	CSCNT(MWNT)	Top Electrode	11.9	n/a	[48]
e	2018	SWNT	Top Electrode	11.8	n/a	[47]
f	2017	SWNT	Top Electrode	16.6	18.4	[12]
g	2017	SWNT	Bottom Electrode	15.3	17.7	[29]
h	2017	B-doped MWNT	Top Electrode	14.6	n/a	[49]
i	2017	CSCNT (MWNT)	Top Electrode	14.3	16 (rigid)	[50]
j	2017	SWNT/NiO	Top Electrode	12.7	n/a	[51]
k	2016	SWNT	Top Electrode	15.5	18.8	[52]
l	2016	C/SWNT	Top Electrode	14.7	n/a	[53]
m	2016	CSCNT (MWNT)	Top Electrode	10.54	14.76	[54]
n	2015	MWNT	Top Electrode	12.67	n/a	[55]
o	2015	SWNT	Bottom Electrode	6.32	9.5	[56]
p	2015	Twisted MWNT fiber	Top Electrode	3.03	2.8	[57]



**Fig. S22.** Device stability data of the unencapsulated Au-based PSC (orange circles) and the unencapsulated TFMS-treated T80%-CNT-PSC (blue triangles) under constant illumination of one sun.