

Supporting Information

Alkyl Chain Length and Anion Effects of AIE-active α -Cyanostilbene-containing Triphenylamine Derivatives on Optical and Electrochemical Properties

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Table S4. Crystal data and experimental details for **diOMe-TPA-Py-2-Br**. S23

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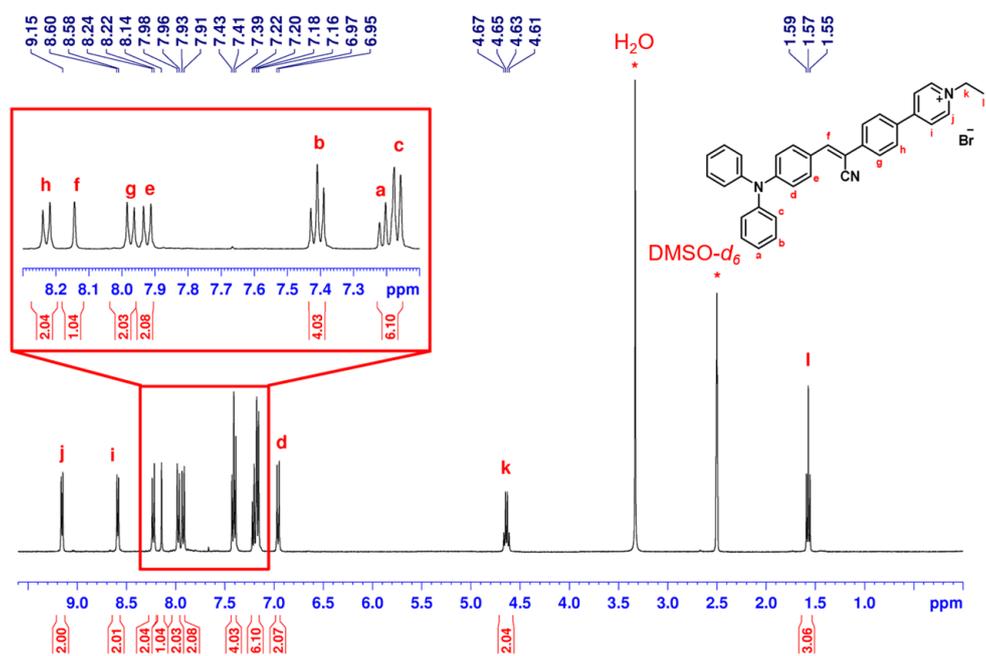


Figure S1. $^1\text{H-NMR}$ spectrum of TPA-Py-2-Br in $\text{DMSO-}d_6$.

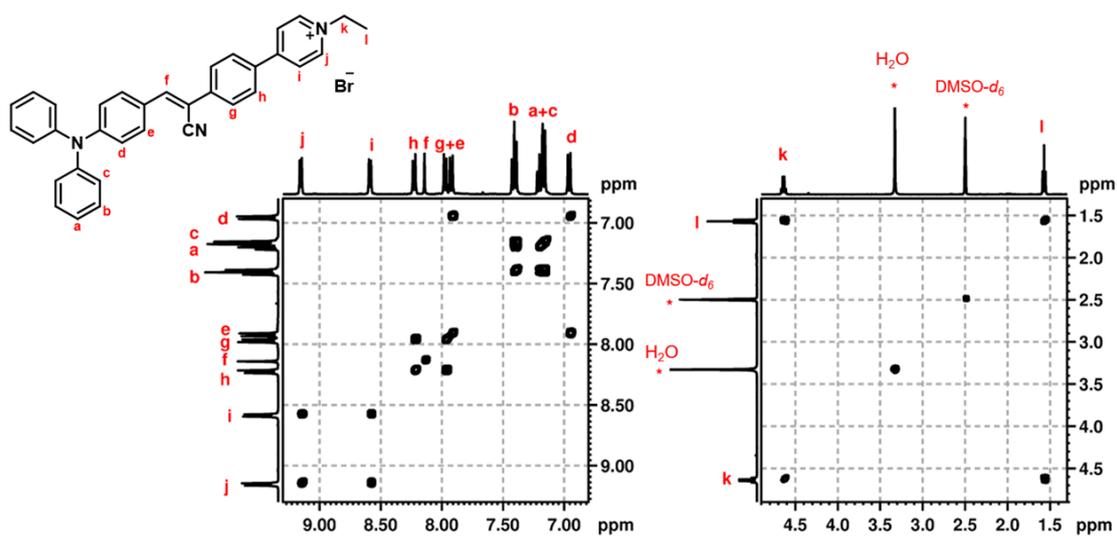


Figure S2. $^1\text{H-}^1\text{H}$ COSY spectrum of TPA-Py-2-Br in $\text{DMSO-}d_6$.

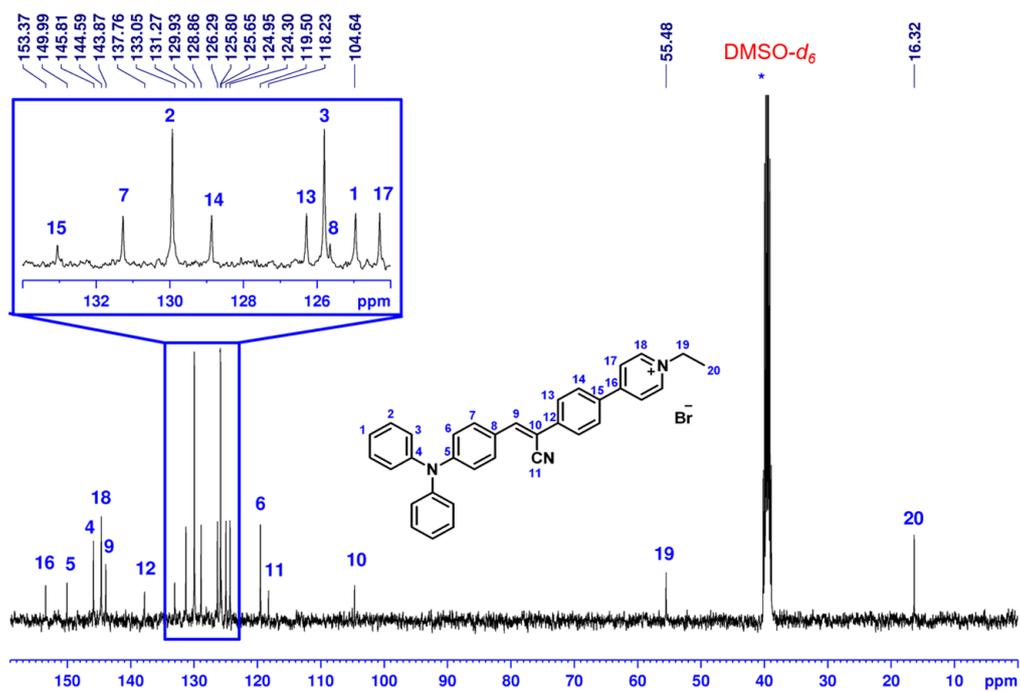


Figure S3. ^{13}C -NMR spectrum of TPA-Py-2-Br in $\text{DMSO-}d_6$.

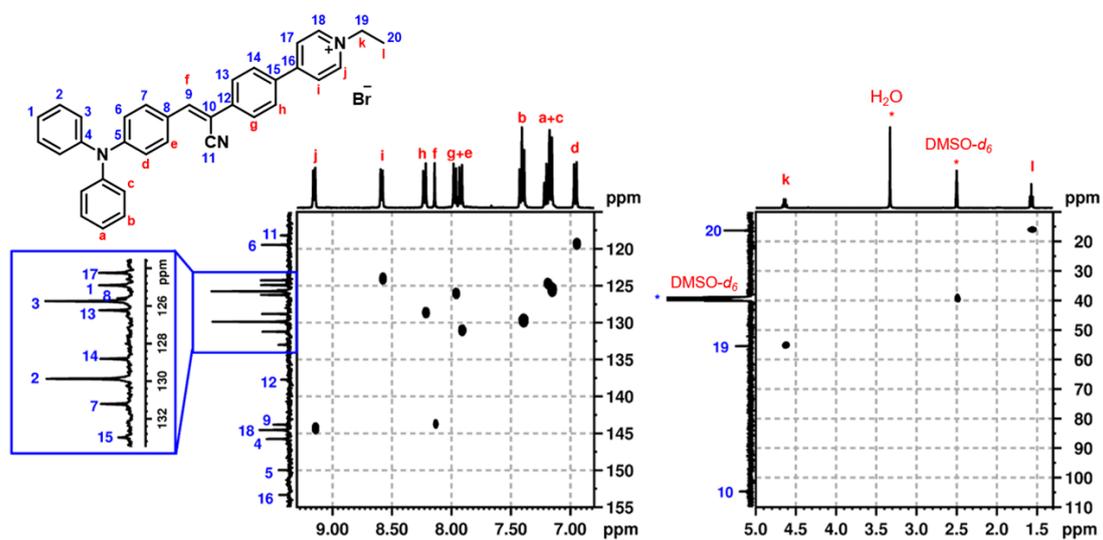


Figure S4. ^1H - ^{13}C HSQC spectrum of TPA-Py-2-Br in $\text{DMSO-}d_6$.

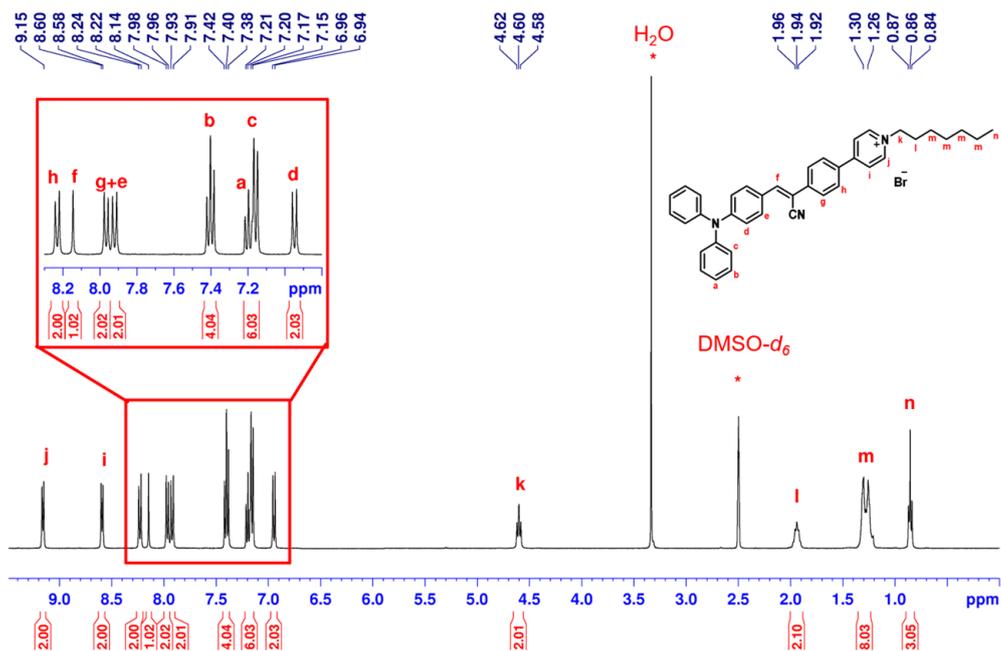


Figure S5. $^1\text{H-NMR}$ spectrum of TPA-Py-7-Br in $\text{DMSO-}d_6$.

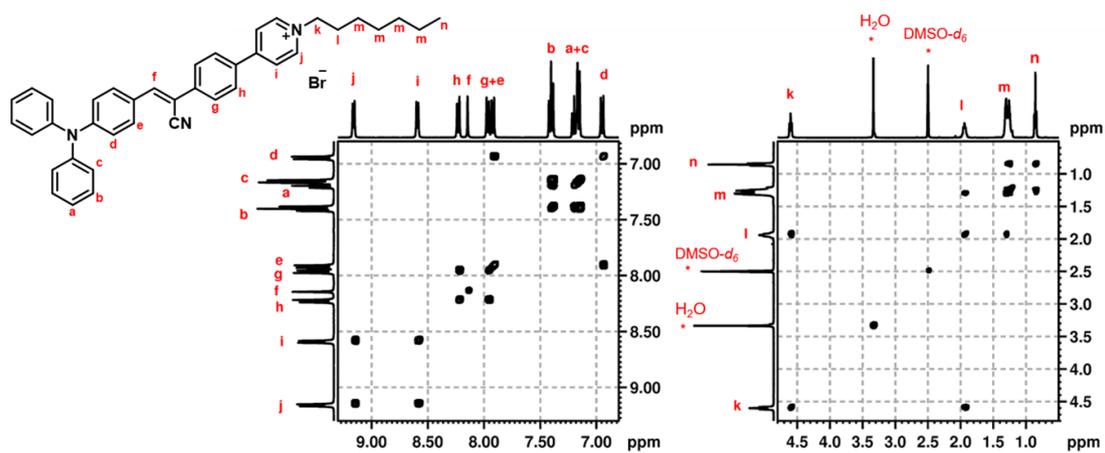


Figure S6. $^1\text{H-}^1\text{H}$ COSY spectrum of TPA-Py-7-Br in $\text{DMSO-}d_6$.

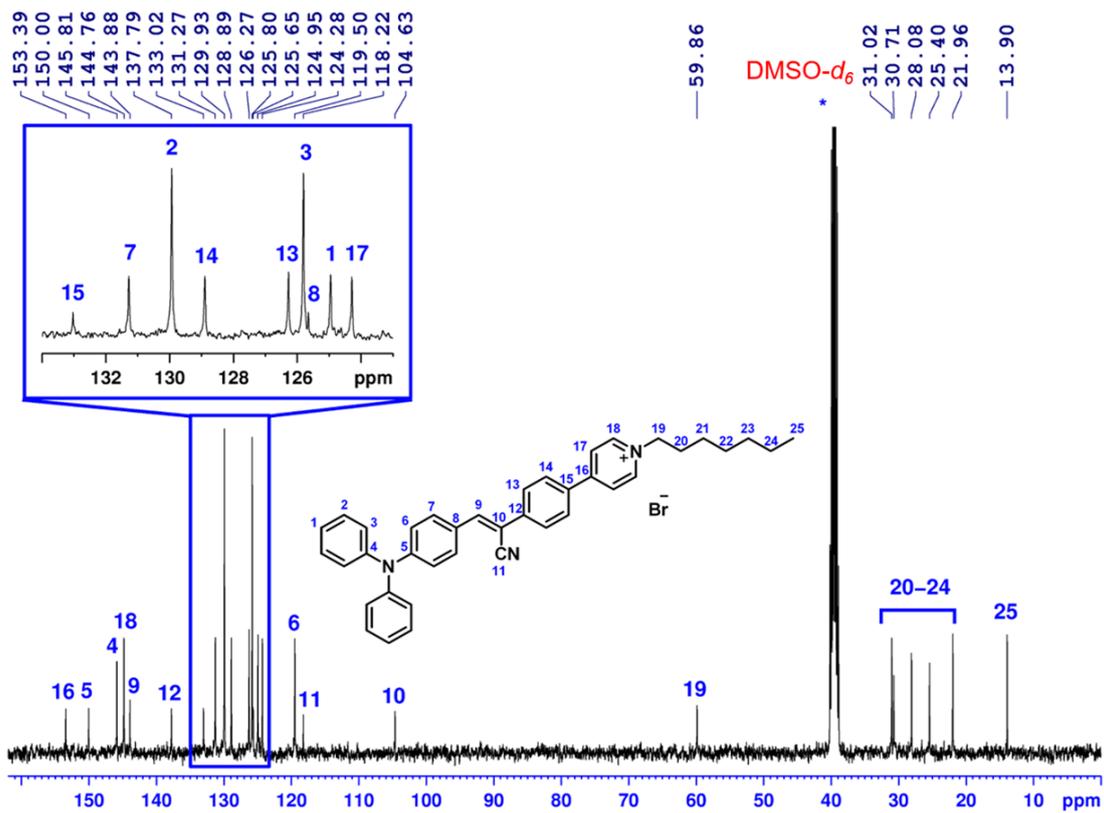


Figure S7. ^{13}C -NMR spectrum of TPA-Py-7-Br in $\text{DMSO-}d_6$.

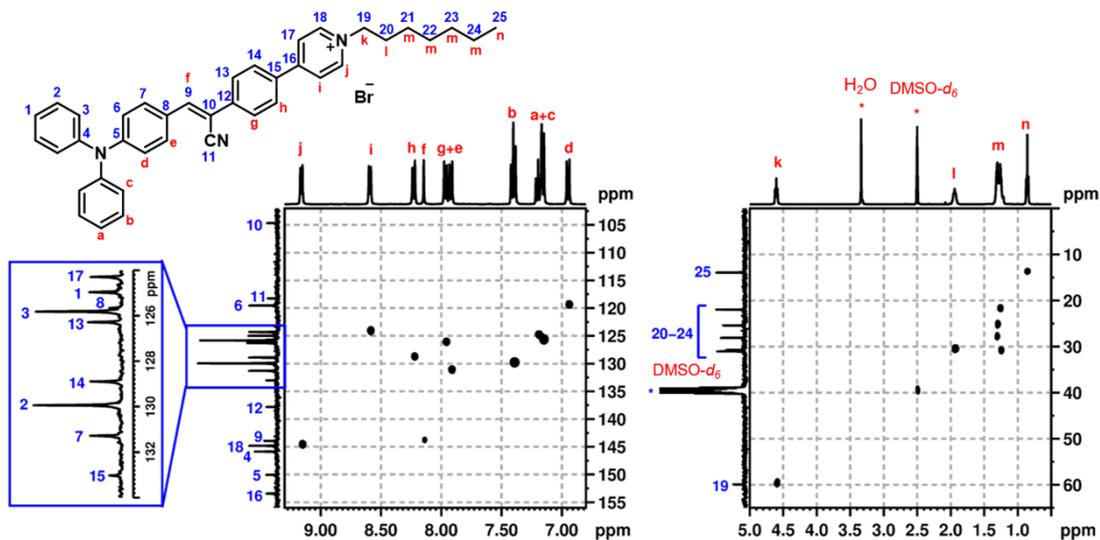


Figure S8. ^1H - ^{13}C HSQC spectrum of TPA-Py-7-Br in $\text{DMSO-}d_6$.

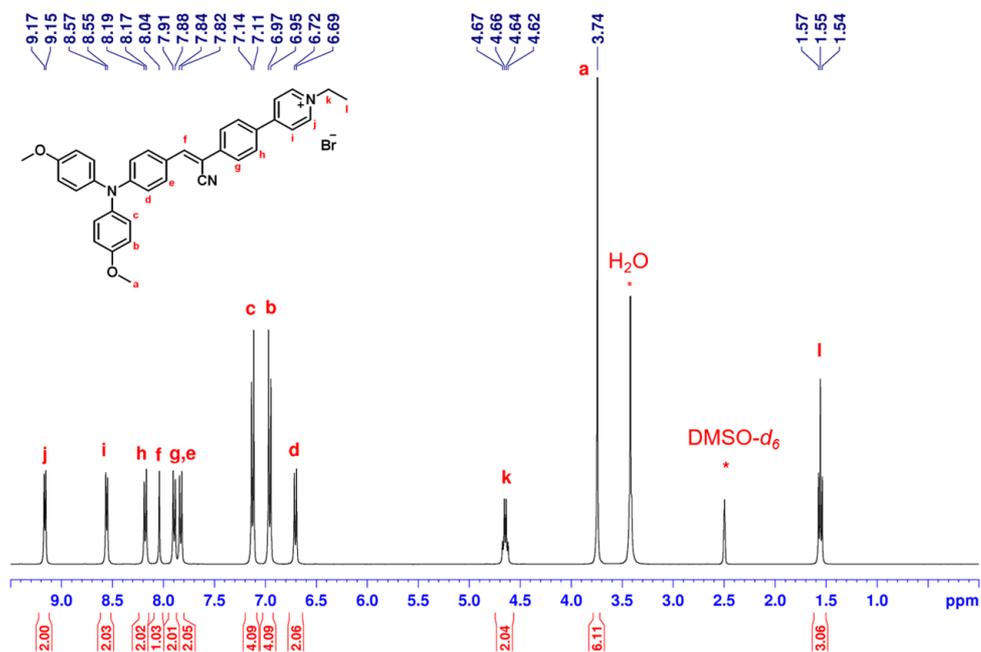


Figure S9. $^1\text{H-NMR}$ spectrum of diOMe-TPA-Py-2-Br in $\text{DMSO-}d_6$.

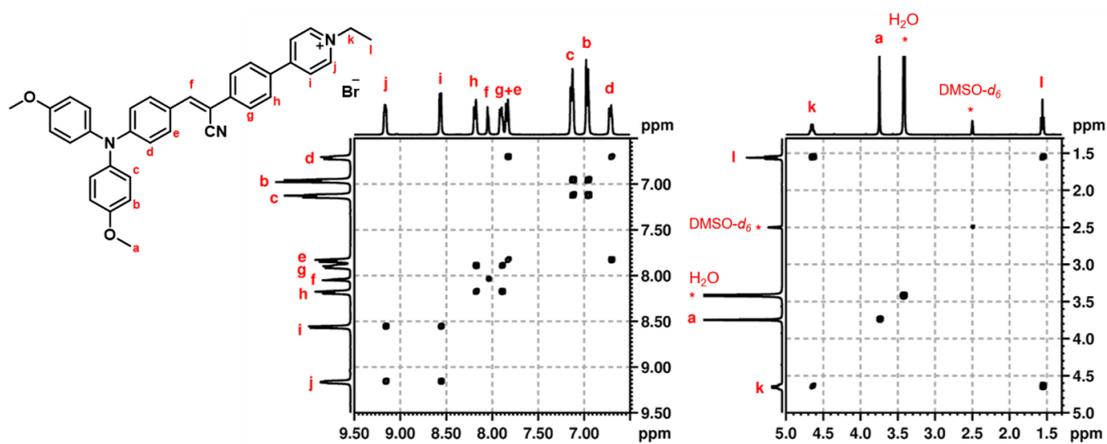


Figure S10. $^1\text{H-}^1\text{H}$ COSY spectrum of diOMe-TPA-Py-2-Br in $\text{DMSO-}d_6$.

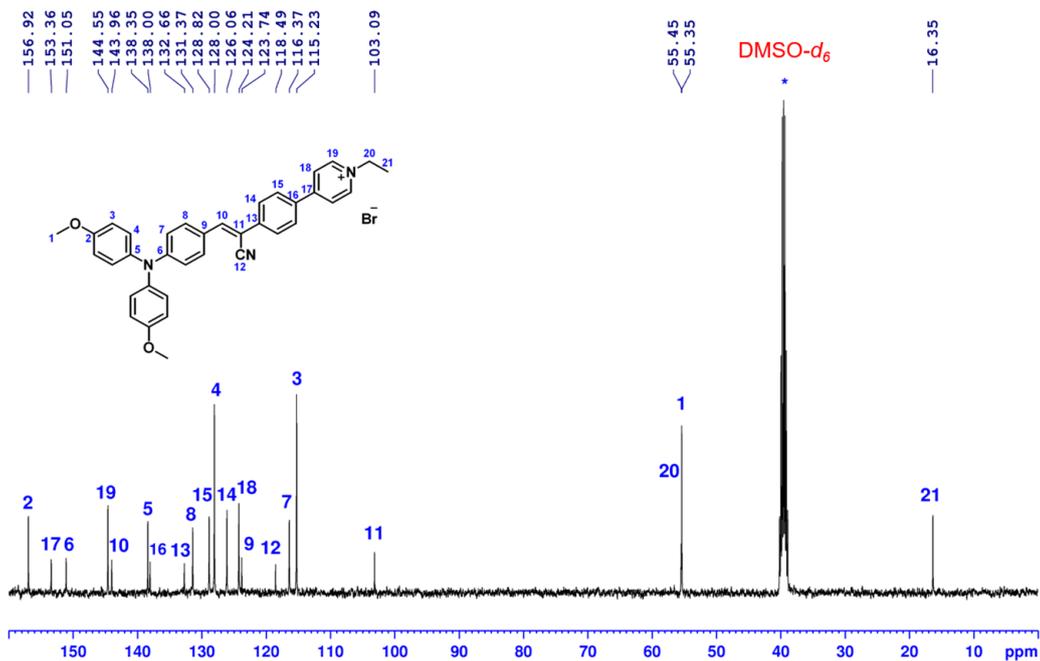


Figure S11. ^{13}C -NMR spectrum of diOMe-TPA-Py-2-Br in $\text{DMSO-}d_6$.

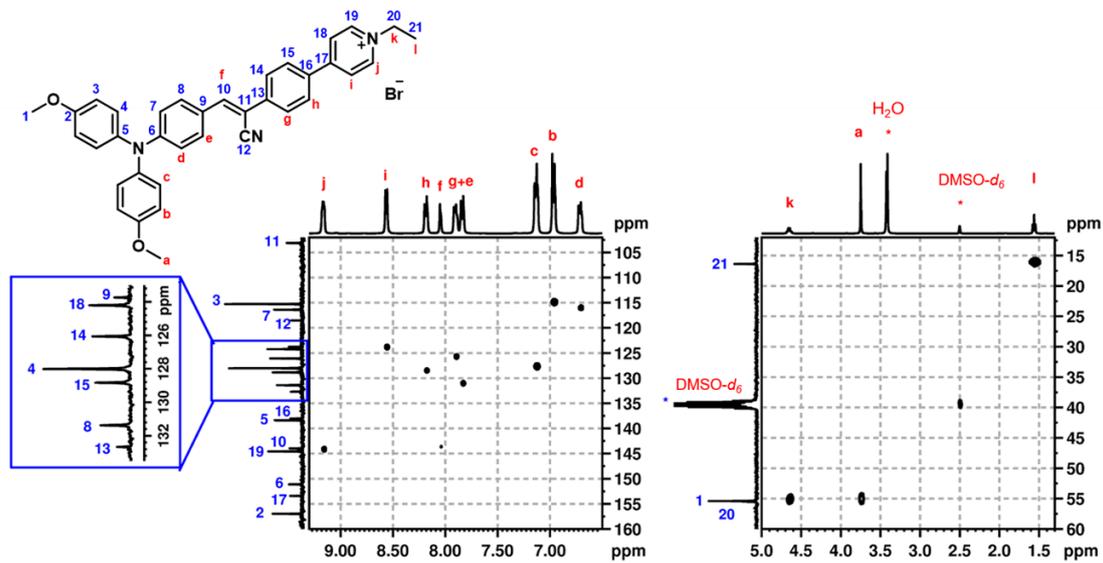


Figure S12. ^1H - ^{13}C HSQC spectrum of diOMe-TPA-Py-2-Br in $\text{DMSO-}d_6$.

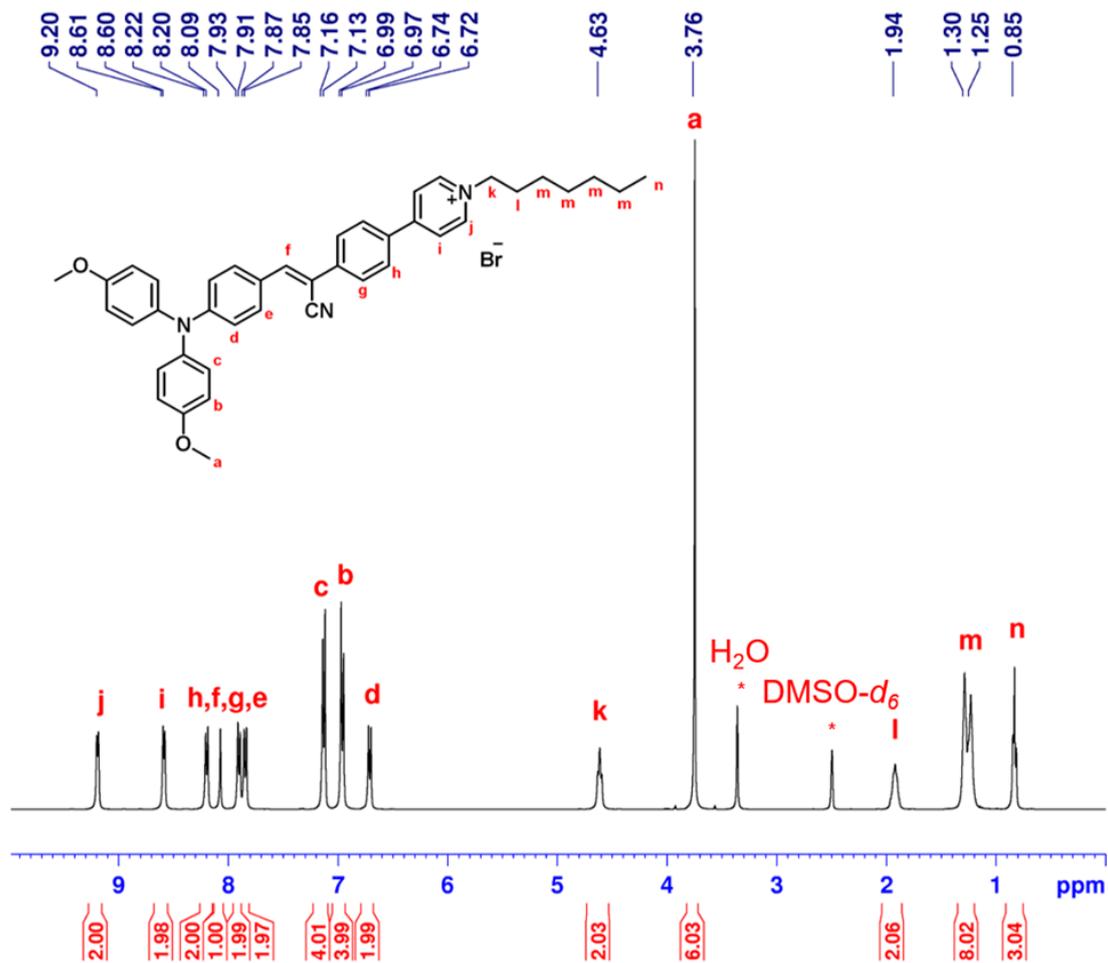


Figure S13. ¹H-NMR spectrum of diOMe-TPA-Py-7-Br in DMSO-*d*₆.

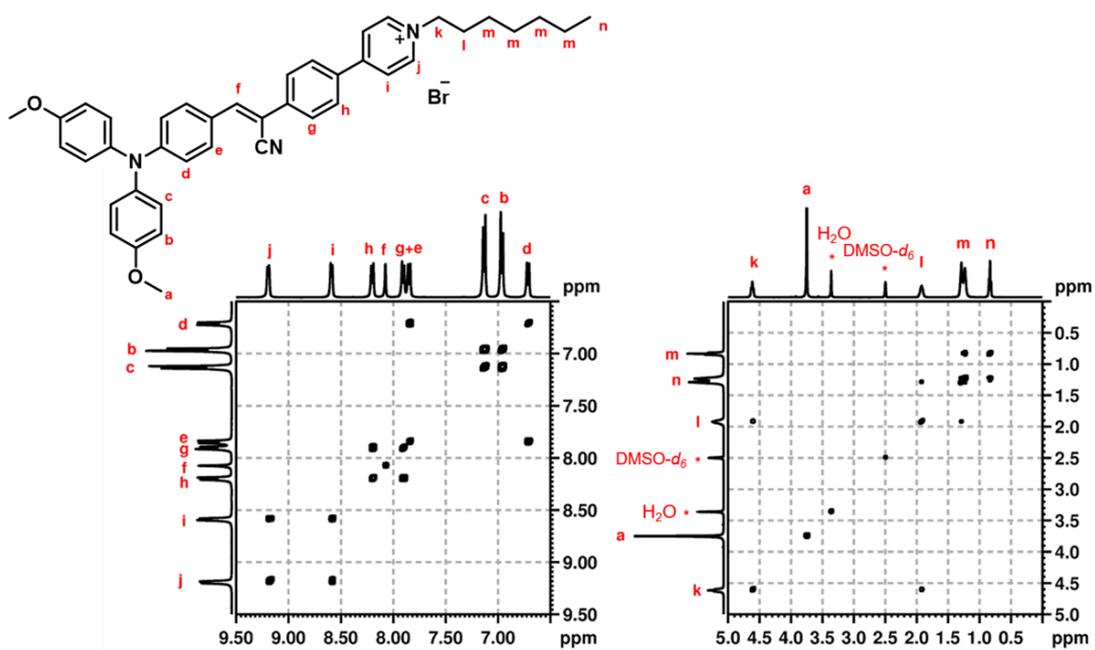


Figure S14. ¹H-¹H COSY spectrum of diOMe-TPA-Py-7-Br in DMSO-*d*₆.

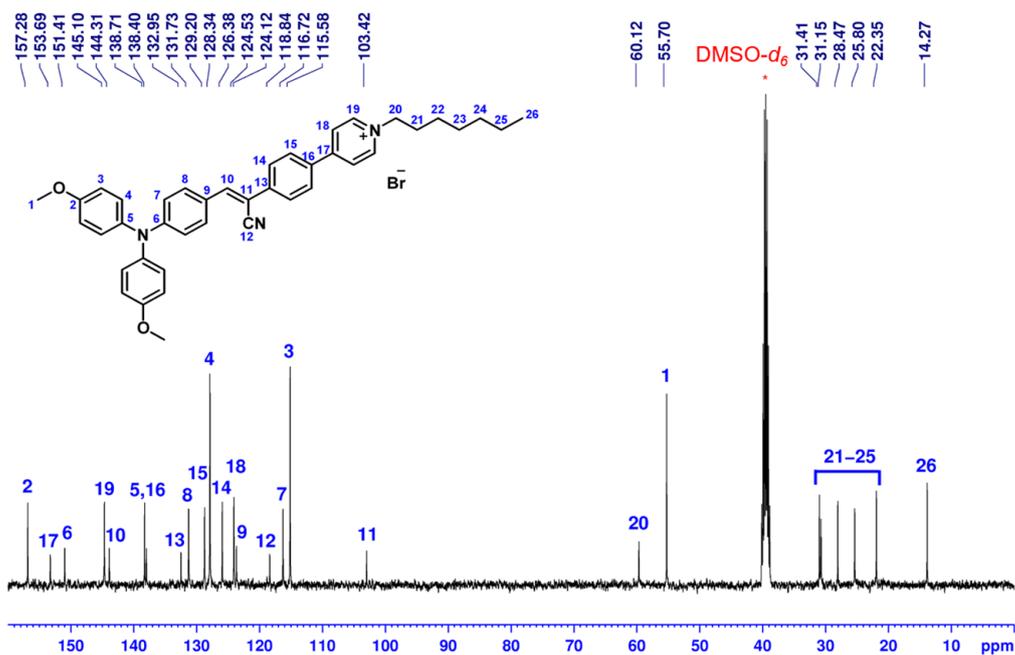


Figure S15. ^{13}C -NMR spectrum of diOMe-TPA-Py-7-Br in $\text{DMSO-}d_6$.

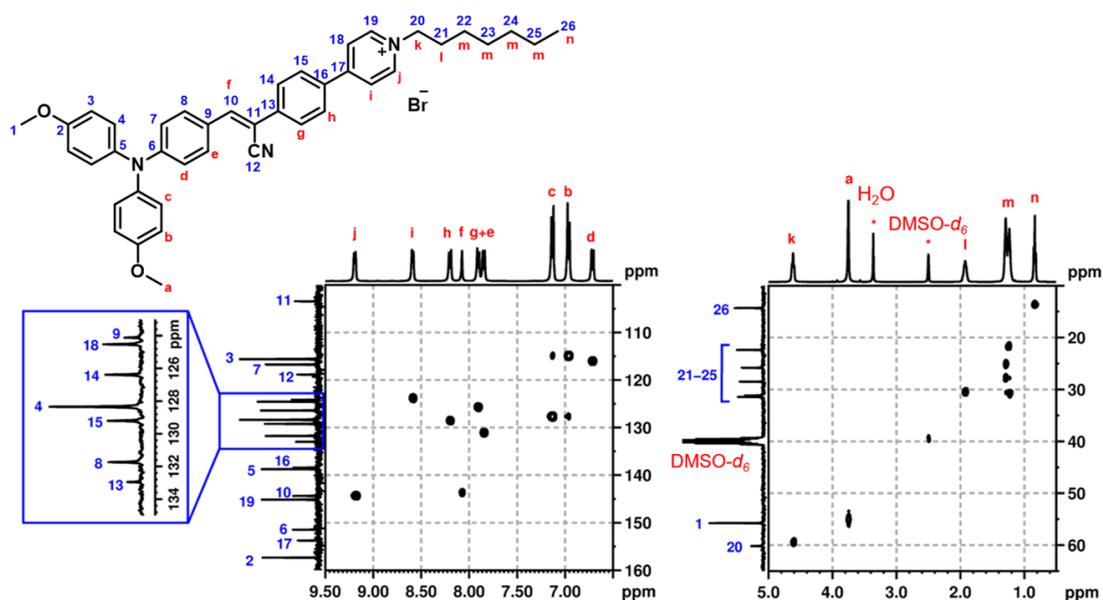


Figure S16. ^1H - ^{13}C HSQC spectrum of diOMe-TPA-Py-7-Br in $\text{DMSO-}d_6$.

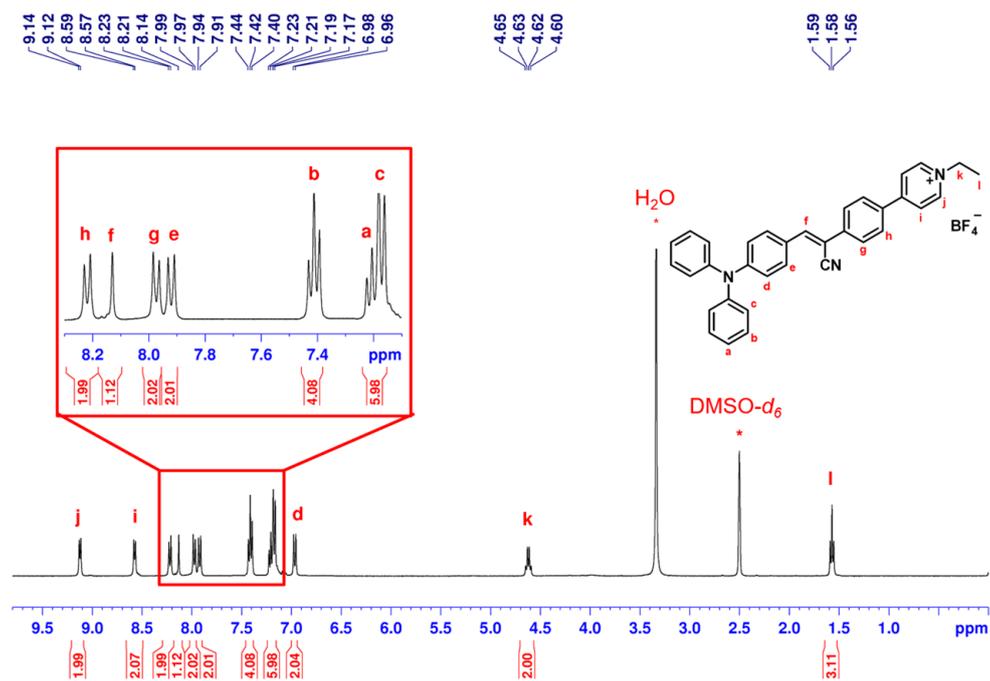


Figure S17. ¹H-NMR spectrum of TPA-Py-2-BF₄ in DMSO-*d*₆.

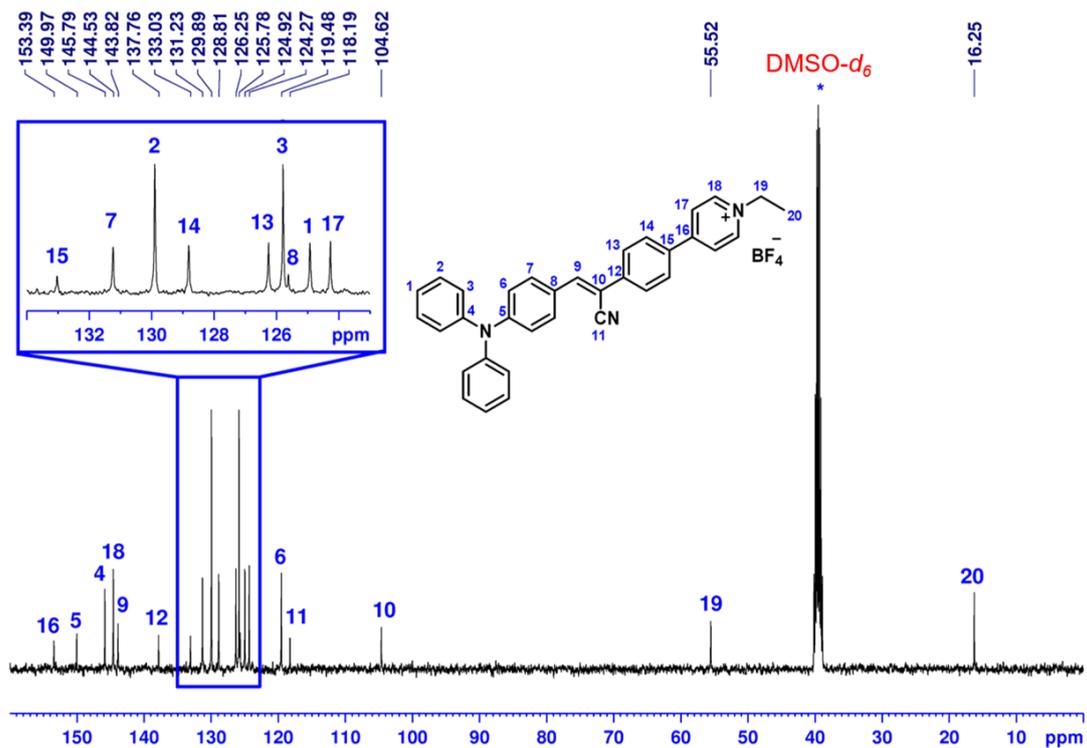


Figure S18. ¹³C-NMR spectrum of TPA-Py-2-BF₄ in DMSO-*d*₆.

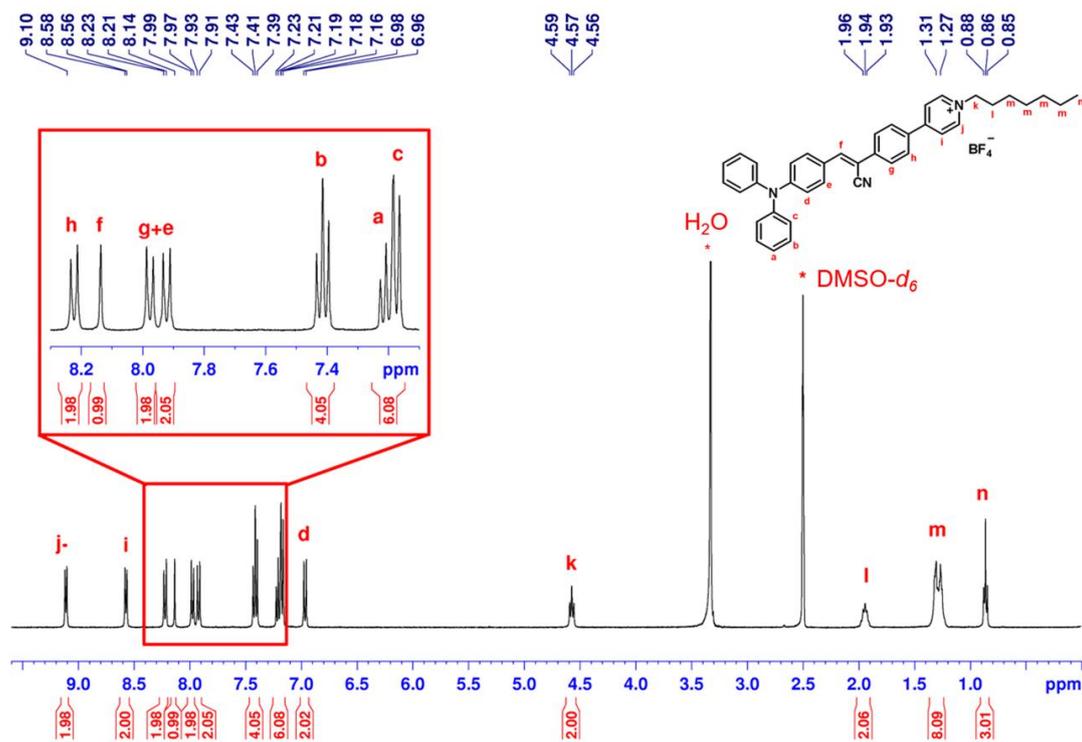


Figure S19. ¹H-NMR spectrum of TPA-Py-7-BF₄ in DMSO-d₆.

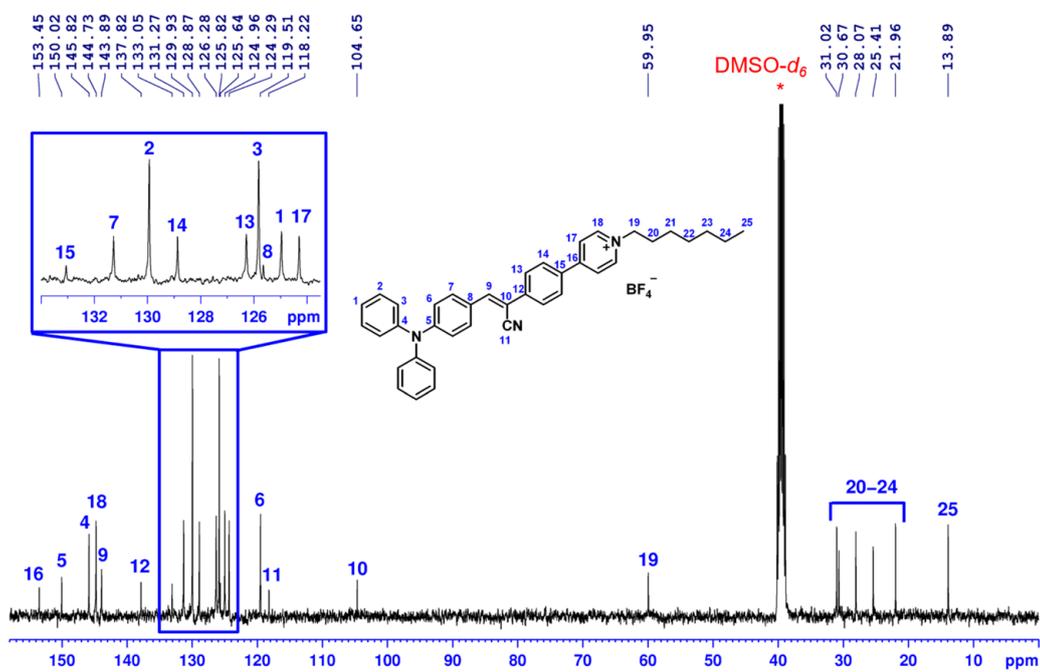


Figure S20. ¹³C-NMR spectrum of TPA-Py-7-BF₄ in DMSO-d₆.

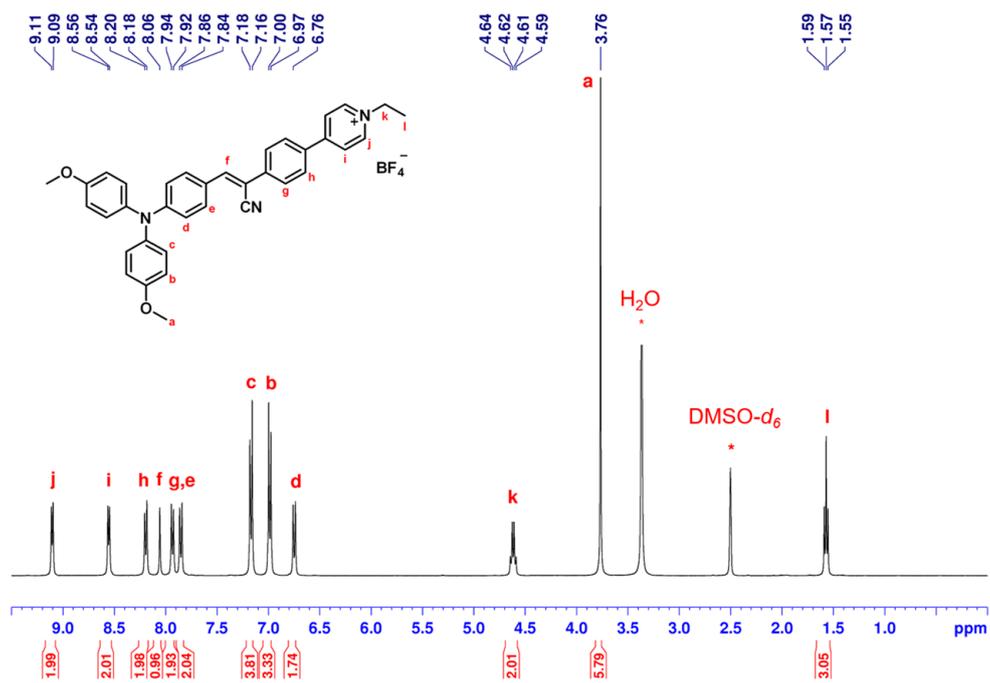


Figure S21. ¹H-NMR spectrum of diOMe-TPA-Py-2-BF₄ in DMSO-d₆.

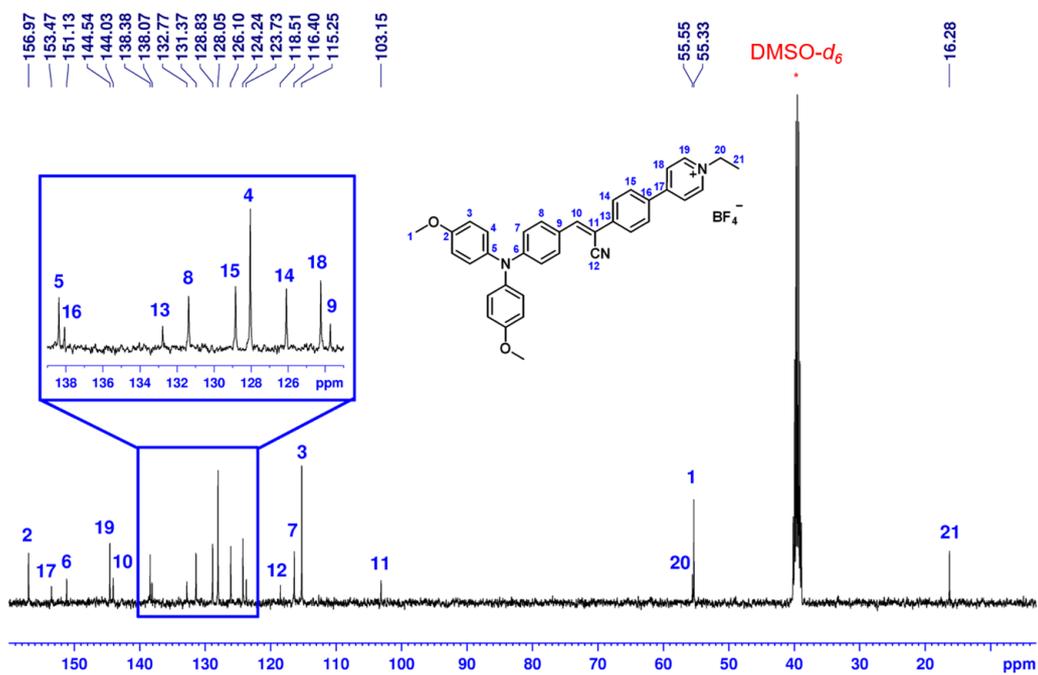


Figure S22. ¹³C-NMR spectrum of diOMe-TPA-Py-2-BF₄ in DMSO-d₆.

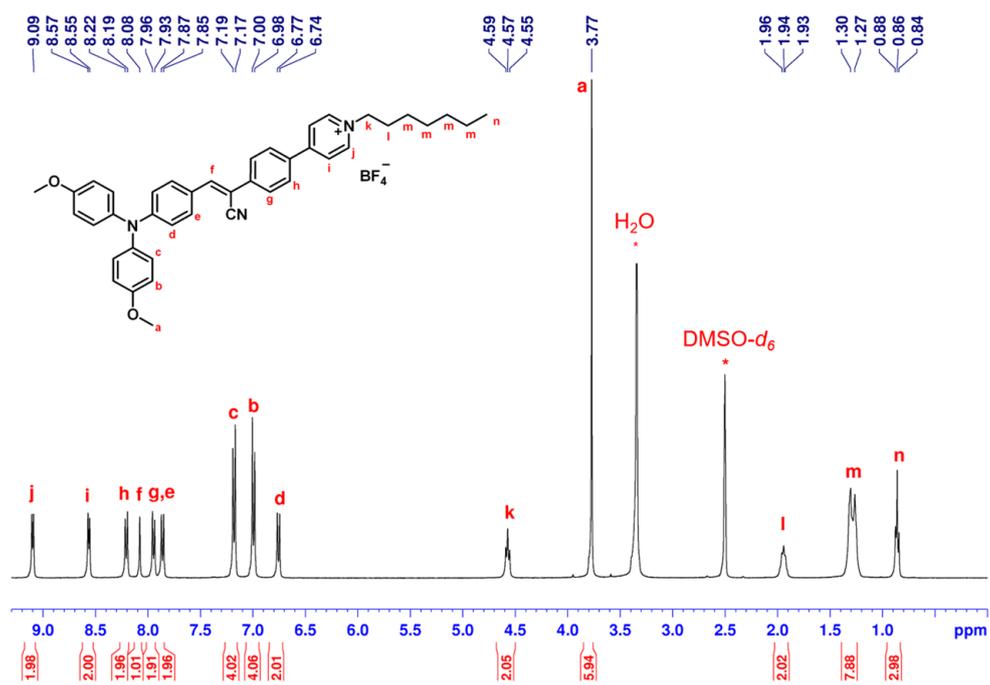


Figure S23. ¹H-NMR spectrum of diOMe-TPA-Py-7-BF₄ in DMSO-d₆.

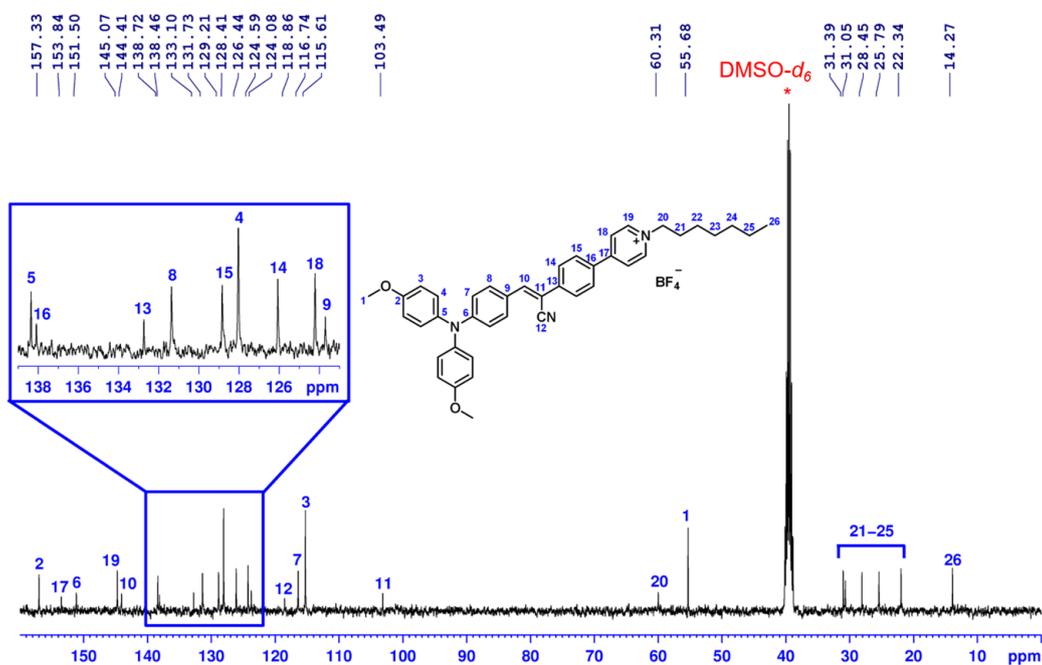


Figure S24. ¹³C-NMR spectrum of diOMe-TPA-Py-7-BF₄ in DMSO-d₆.

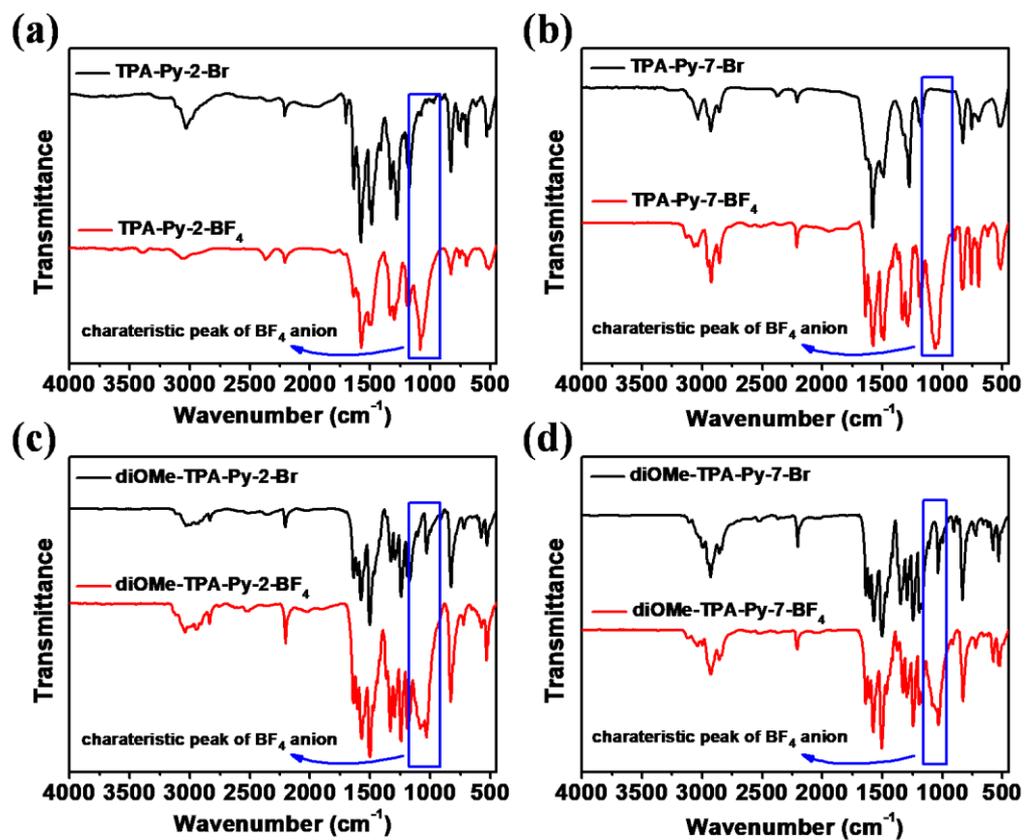


Figure S25. FTIR spectra of (a) TPA-Py-2-Br (black) and TPA-Py-2- BF_4 (red), (b) TPA-Py-7-Br (black) and TPA-Py-7- BF_4 (red), (c) diOMe-TPA-Py-2-Br (black) and diOMe-TPA-Py-2- BF_4 (red) and (d) diOMe-TPA-Py-7-Br (black) and diOMe-TPA-Py-7- BF_4 (red).

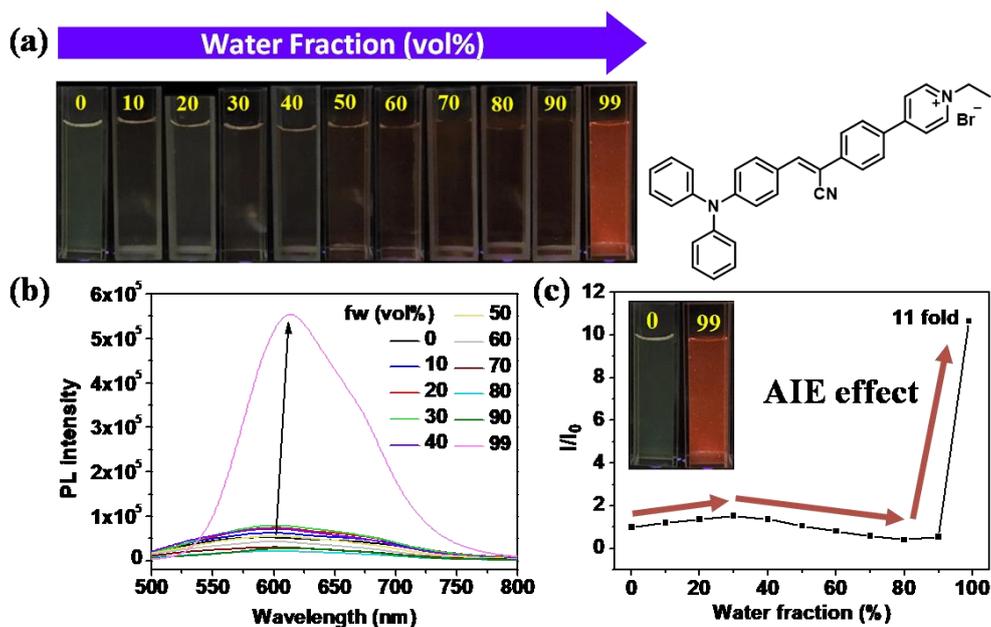


Figure S26. (a) Photos taken under UV light of **TPA-Py-2-Br** in different water/DMSO fractions, (b) PL spectra of **TPA-Py-2-Br** in water/DMSO solutions with different water fractions (f_w), (c) Plot of relative emission intensity (I/I_0) versus different compositions of aqueous mixtures of **TPA-Py-2-Br**. I_0 = Emission intensity in pure DMSO solution (10 μ M). λ_{ex} : 449 nm.

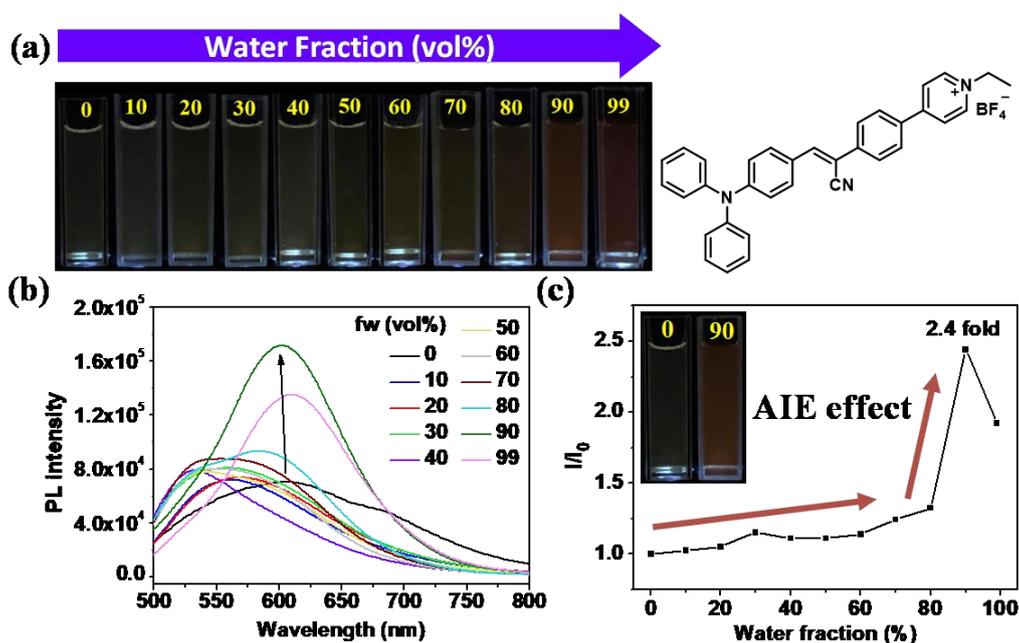


Figure S27. (a) Photos taken under UV light of **TPA-Py-2-BF₄** in different water/DMSO fractions, (b) PL spectra of **TPA-Py-2-BF₄** in water/DMSO solutions with different water fractions (f_w), (c) Plot of relative emission intensity (I/I_0) versus different compositions of aqueous mixtures of **TPA-Py-2-BF₄**. I_0 = Emission intensity in pure DMSO solution (10 μ M). λ_{ex} : 449 nm.

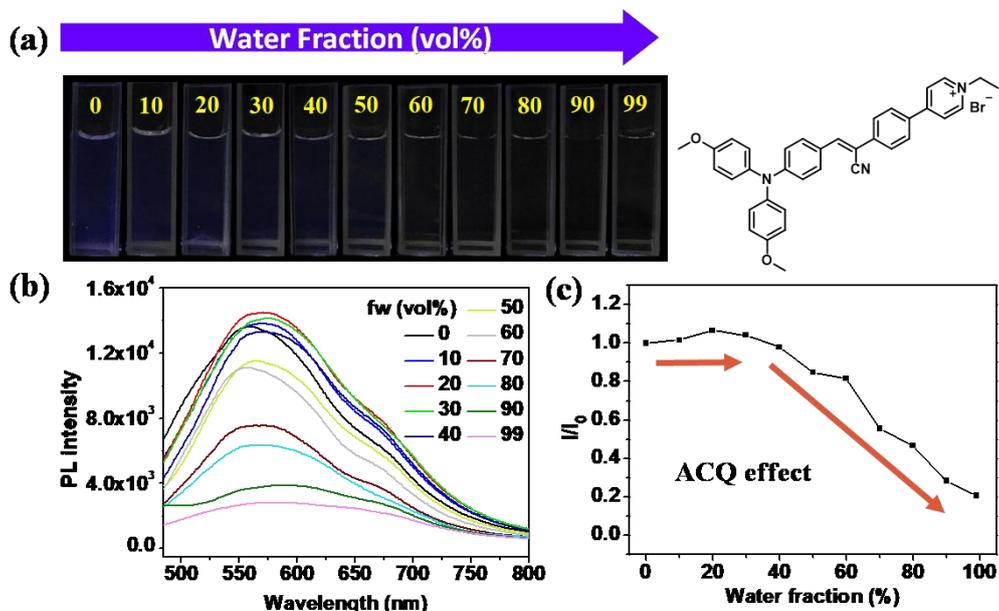


Figure S28. (a) Photos taken under UV light of **diOMe-TPA-Py-2-Br** in different water/DMSO fractions, (b) PL spectra of **diOMe-TPA-Py-2-Br** in water/DMSO solutions with different water fractions (f_w), (c) Plot of relative emission intensity (I/I_0) versus different compositions of aqueous mixtures of **diOMe-TPA-Py-2-Br**. I_0 = Emission intensity in pure DMSO solution (10 μ M). λ_{ex} : 463 nm.

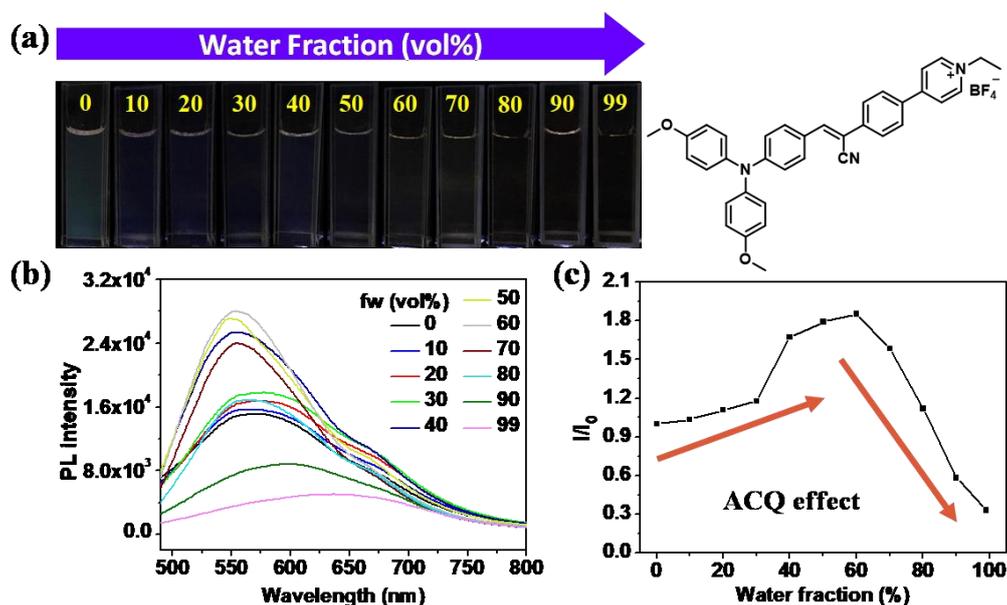


Figure S29. (a) Photos taken under UV light of **diOMe-TPA-Py-2-BF₄** in different water/DMSO fractions, (b) PL spectra of **diOMe-TPA-Py-2-BF₄** in water/DMSO solutions with different water fractions (f_w), (c) Plot of relative emission intensity (I/I_0) versus different compositions of aqueous mixtures of **diOMe-TPA-Py-2-BF₄**. I_0 = Emission intensity in pure DMSO solution (10 μ M). λ_{ex} : 463 nm.

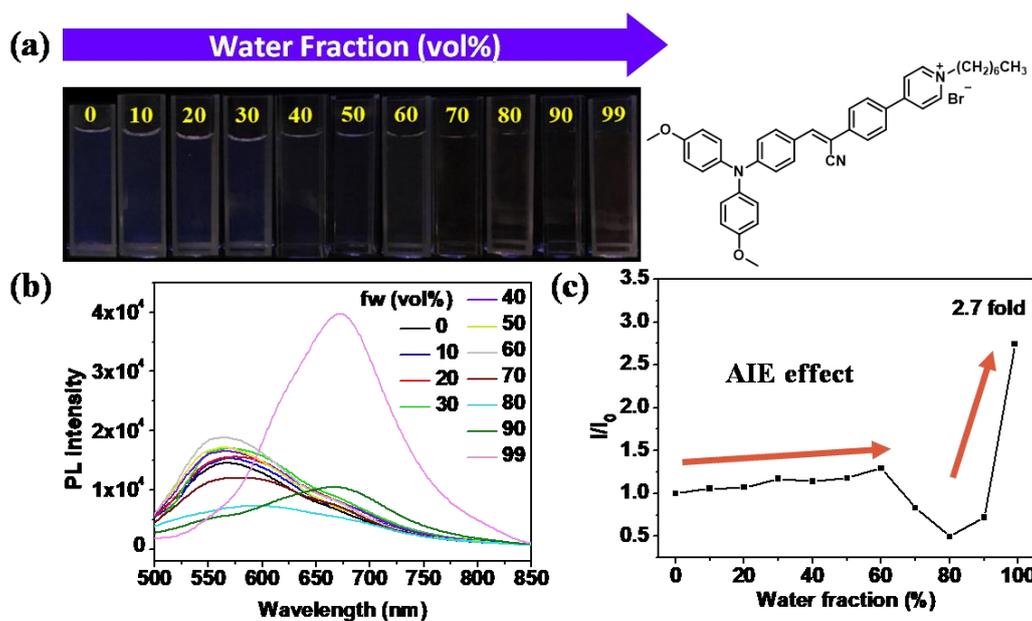


Figure 30. (a) Photos taken under UV light of **diOMe-TPA-Py-7-Br** in different water/DMSO fractions, (b) PL spectra of **diOMe-TPA-Py-7-Br** in water/DMSO solutions with different water fractions (f_w), (c) Plot of relative emission intensity (I/I_0) versus different compositions of aqueous mixtures of **diOMe-TPA-Py-7-Br**. I_0 = Emission intensity in pure DMSO solution (10 μ M). λ_{ex} : 466 nm.

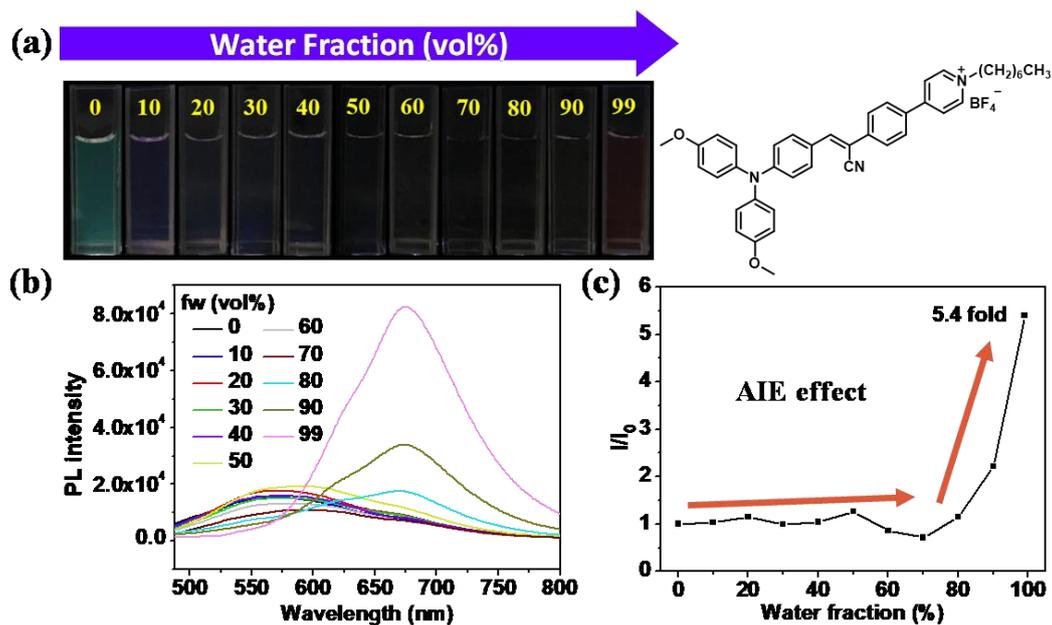


Figure S31. (a) Photos taken under UV light of **diOMe-TPA-Py-7-BF₄** in different water/DMSO fractions, (b) PL spectra of **diOMe-TPA-Py-7-BF₄** in water/DMSO solutions with different water fractions (f_w), (c) Plot of relative emission intensity (I/I_0) versus different compositions of aqueous mixtures of **diOMe-TPA-Py-7-BF₄**. I_0 = Emission intensity in pure DMSO solution (10 μ M). λ_{ex} : 466 nm.

Table S1. Crystal data and experimental details for **TPA-Py-2-BF₄**.

Compound	TPA-Py-2-BF ₄
Empirical formula	C ₃₄ H ₂₈ B F ₄ N ₃
Formula weight	565.40
Crystal system	Monoclinic
Space group	P 1 21/n 1
Unit cell dimensions	a = 13.1118(7) Å b = 15.2813(7) Å c = 13.9655(7) Å α = 90° β = 91.365(5)° γ = 90°
Volume	2797.4(2) Å ³
Z	4
F(000)	1176
Density (calculated)	1.342 Mg/m ³
Wavelength	0.71073 Å
Cell parameters reflections used	4441
Theta range for Cell parameters	3.3730 to 28.9110°
Absorption coefficient	0.097 mm ⁻¹
Temperature	150(2) K
Crystal size	0.25 x 0.20 x 0.20 mm ³
Diffractometer	Xcalibur, Atlas, Gemini
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	1.00000 and 0.97323
No. of measured reflections	11865
No. of independent reflections	6141 [R(int) = 0.0292]
No. of observed [I>2 _{sigma} (I)]	4280
Completeness to theta = 27.50°	95.7 %
Theta range for data collection	3.38 to 27.50°
Final R indices [I>2 _{sigma} (I)]	R1 = 0.0500, wR2 = 0.1366
R indices (all data)	R1 = 0.0805, wR2 = 0.1664
Goodness-of-fit on F ²	0.987
No. of reflections	6141
No. of parameters	379
No. of restraints	0
Largest diff. peak and hole	0.297 and -0.229 e.Å ⁻³

Table S2. Crystal data and experimental details for **TPA-Py-7-Br**.

Compound	TPA-Py-7-Br
Empirical formula	C ₃₉ H ₃₈ Br N ₃
Formula weight	628.63
Crystal system	Triclinic
Space group	P-1
Unit cell dimensions	a = 9.5749(6) Å b = 9.7795(4) Å c = 35.3728(18) Å $\alpha = 84.999(4)^\circ$ $\beta = 89.387(5)^\circ$ $\gamma = 82.412(4)^\circ$
Volume	3270.7(3) Å ³
Z	4
F(000)	1312
Density (calculated)	1.277 Mg/m ³
Wavelength	0.71073 Å
Cell parameters reflections used	3679
Theta range for Cell parameters	3.0880 to 27.1700°
Absorption coefficient	1.288 mm ⁻¹
Temperature	150(2) K
Crystal size	0.25 x 0.20 x 0.15 mm ³
Diffractometer	Xcalibur, Atlas, Gemini
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	1.00000 and 0.76124
No. of measured reflections	19095
No. of independent reflections	13990 [R(int) = 0.0557]
No. of observed [I>2 σ (I)]	8199
Completeness to theta = 27.50°	93.0 %
Theta range for data collection	2.96 to 27.50°
Final R indices [I>2 σ (I)]	R1 = 0.0899, wR2 = 0.2015
R indices (all data)	R1 = 0.1544, wR2 = 0.2542
Goodness-of-fit on F ²	1.092
No. of reflections	13990
No. of parameters	794
No. of restraints	0
Largest diff. peak and hole	2.042 and -0.756 e.Å ⁻³

Table S3. Crystal data and experimental details for **TPA-Py-7-BF₄**.

Compound	TPA-Py-7-BF ₄
Empirical formula	C ₃₉ H ₃₈ B F ₄ N ₃
Formula weight	635.53
Crystal system	Triclinic
Space group	P-1
Unit cell dimensions	a = 10.0994(2) Å b = 18.0030(8) Å c = 19.7095(6) Å α = 102.494(3)° β = 94.039(2)° γ = 103.958(2)°
Volume	3366.85(19) Å ³
Z	4
F(000)	1336
Density (calculated)	1.254 Mg/m ³
Wavelength	0.71073 Å
Cell parameters reflections used	6901
Theta range for Cell parameters	3.0570 to 28.4760°
Absorption coefficient	0.088 mm ⁻¹
Temperature	150(2) K
Crystal size	0.25 x 0.20 x 0.15 mm ³
Diffractometer	Xcalibur, Atlas, Gemini
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	1.00000 and 0.98348
No. of measured reflections	23237
No. of independent reflections	14560 [R(int) = 0.0304]
No. of observed [I>2 _{sigma} (I)]	8682
Completeness to theta = 27.50°	94.1 %
Theta range for data collection	3.00 to 27.50°
Final R indices [I>2 _{sigma} (I)]	R1 = 0.0764, wR2 = 0.1754
R indices (all data)	R1 = 0.1327, wR2 = 0.2218
Goodness-of-fit on F ²	1.003
No. of reflections	14560
No. of parameters	847
No. of restraints	0
Largest diff. peak and hole	1.172 and -0.621 e.Å ⁻³

Table S4. Crystal data and experimental details for **diOMe-TPA-Py-2-Br**.

Compound	diOMe-TPA-Py-2-Br
Empirical formula	C ₃₆ H ₃₆ Br N ₃ O ₄
Formula weight	654.59
Crystal system	Monoclinic
Space group	P 1 21/n 1
Unit cell dimensions	a = 14.8377(4) Å b = 10.3068(2) Å c = 21.2595(5) Å $\alpha = 90^\circ$ $\beta = 104.015(2)^\circ$ $\gamma = 90^\circ$
Volume	3154.42(13) Å ³
Z	4
F(000)	1360
Density (calculated)	1.378 Mg/m ³
Wavelength	0.71073 Å
Cell parameters reflections used	7664
Theta range for Cell parameters	3.5560 to 28.5120°
Absorption coefficient	1.347 mm ⁻¹
Temperature	150(2) K
Crystal size	0.30 x 0.25 x 0.20 mm ³
Diffractometer	Xcalibur, Atlas, Gemini
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	1.00000 and 0.87930
No. of measured reflections	22809
No. of independent reflections	7092 [R(int) = 0.0315]
No. of observed [I>2 σ (I)]	5435
Completeness to theta = 27.50°	97.8 %
Theta range for data collection	2.83 to 27.50°
Final R indices [I>2 σ (I)]	R1 = 0.0368, wR2 = 0.0827
R indices (all data)	R1 = 0.0584, wR2 = 0.0934
Goodness-of-fit on F ²	0.999
No. of reflections	7092
No. of parameters	409
No. of restraints	0
Largest diff. peak and hole	0.331 and -0.443 e.Å ⁻³

Table S5. Crystal data and experimental details for **diOMe-TPA-Py-7-Br**.

Compound	diOMe-TPA-Py-7-Br
Empirical formula	C41 H42 Br N3 O2
Formula weight	688.69
Crystal system	Monoclinic
Space group	P 1 21/c 1
Unit cell dimensions	a = 14.1460(3) Å b = 10.3836(2) Å c = 25.5561(4) Å $\alpha = 90^\circ$ $\beta = 105.051(2)^\circ$ $\gamma = 90^\circ$
Volume	3625.07(12) Å ³
Z	4
F(000)	1440
Density (calculated)	1.262 Mg/m ³
Wavelength	0.71073 Å
Cell parameters reflections used	5835
Theta range for Cell parameters	2.9840 to 27.4910°
Absorption coefficient	1.172 mm ⁻¹
Temperature	200(2) K
Crystal size	0.30 x 0.20 x 0.15 mm ³
Diffractometer	Xcalibur, Atlas, Gemini
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	1.00000 and 0.89149
No. of measured reflections	17089
No. of independent reflections	7982 [R(int) = 0.0299]
No. of observed [I>2 σ (I)]	5980
Completeness to theta = 27.50°	95.7 %
Theta range for data collection	2.98 to 27.50°
Final R indices [I>2 σ (I)]	R1 = 0.0423, wR2 = 0.0959
R indices (all data)	R1 = 0.0661, wR2 = 0.1098
Goodness-of-fit on F ²	1.012
No. of reflections	7982
No. of parameters	424
No. of restraints	0
Largest diff. peak and hole	0.356 and -0.451 e.Å ⁻³