

Figure S1. Calculated phonon band structures for TMB ($\text{TM}=\text{TM}=\text{Co, Cr, Fe, Mn, Sc, Ti}$) monolayer. Similar to phonon spectra of the Ge monolayer¹ and borophene², the tiny imaginary frequency nearing the Γ point is also found, although the Ge monolayer and borophene have been experimentally realized on the surface of metal substrate.²⁻⁷

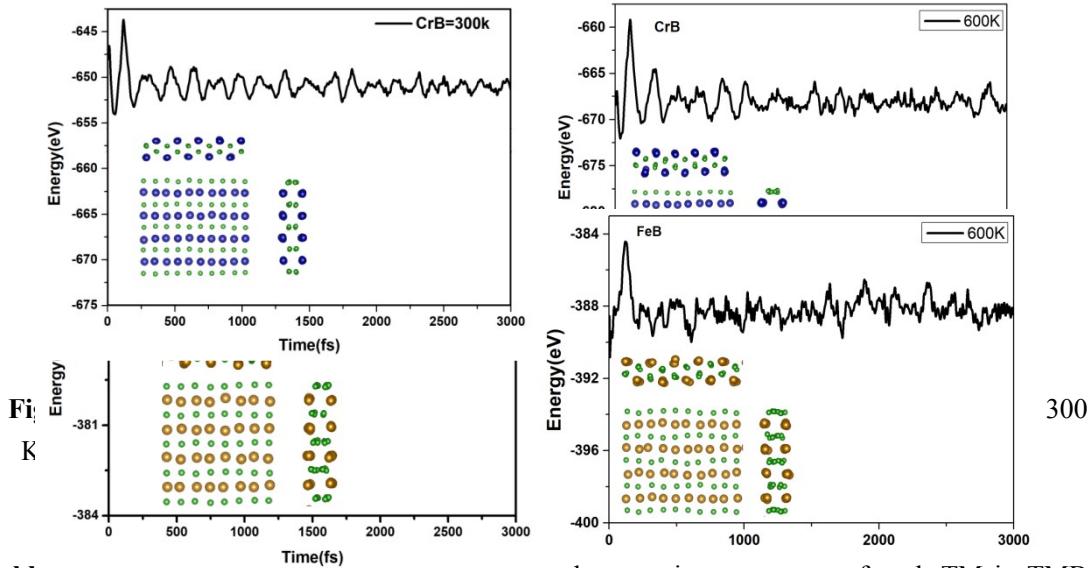
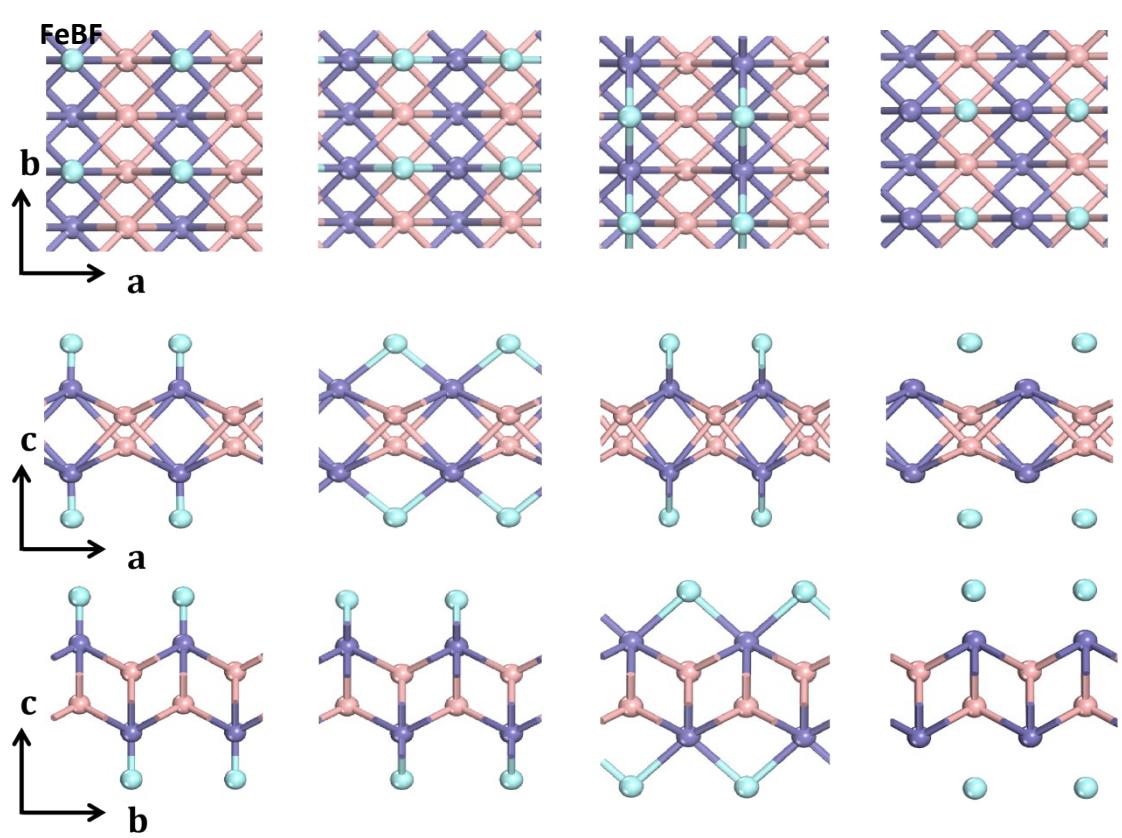
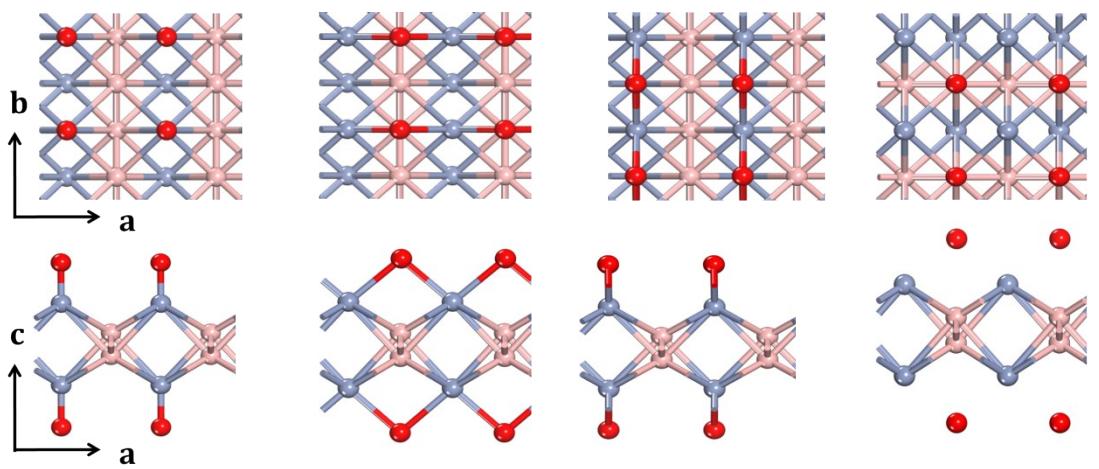


Table S1. The details of the lattice parameters and magnetic momentum of each TM in TMB (TM=Co, Cr, Fe, Mn, Sc, Ti) monolayers. The results are obtained with GGA+U calculation.

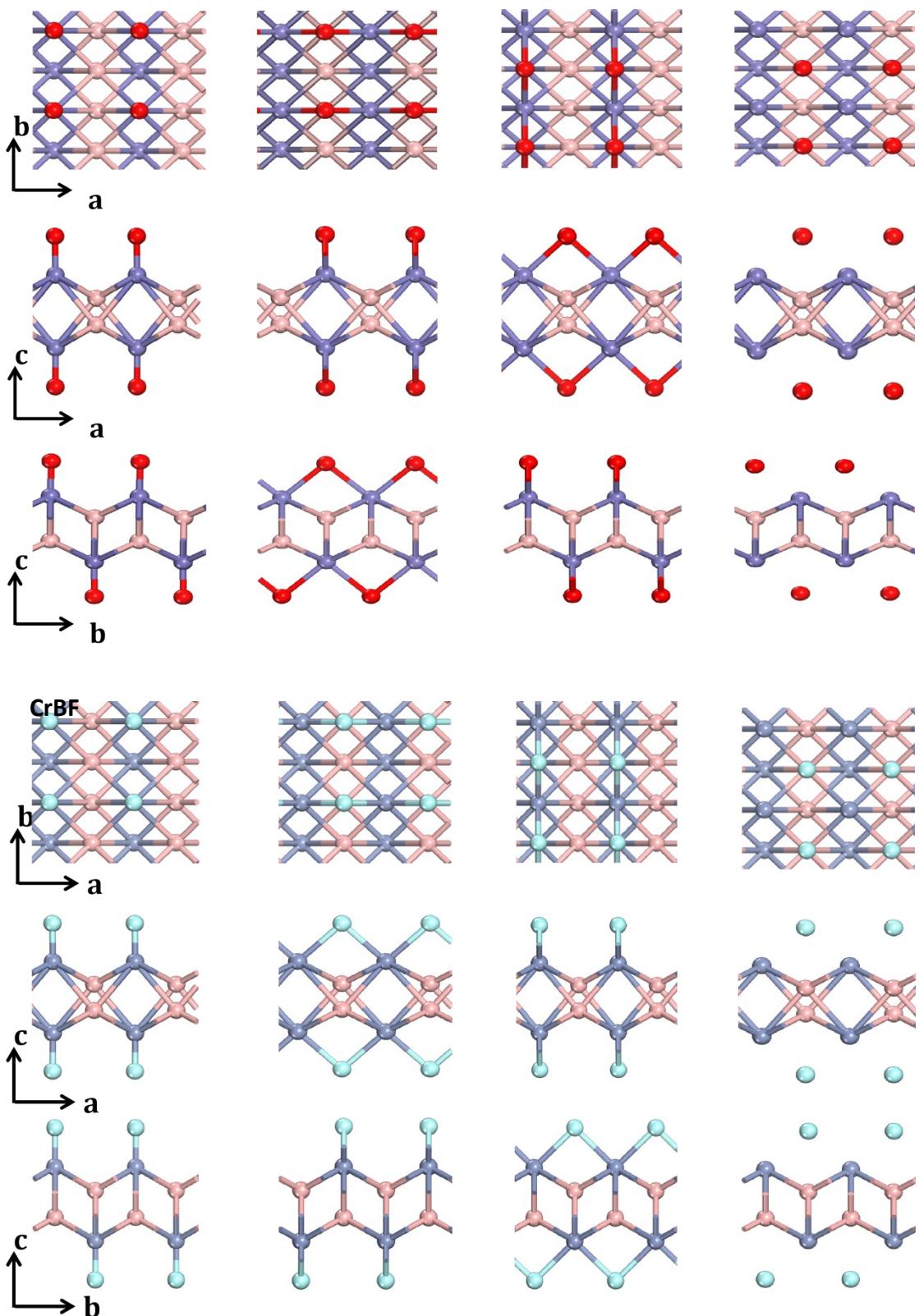
2D monolayer	a(Å)	b(Å)	M _{TM} (μ _B)
CoB	2.59	2.90	0.48
CrB	2.93	2.87	3.50
FeB	2.77	2.82	2.70
MnB	2.90	2.87	3.70
ScB	3.30	3.09	0.09
TiB	3.10	2.96	0.53
VB	3.18	2.91	0.004

CrBO

P1 P2 P3 P4



FeBO



FeBOH

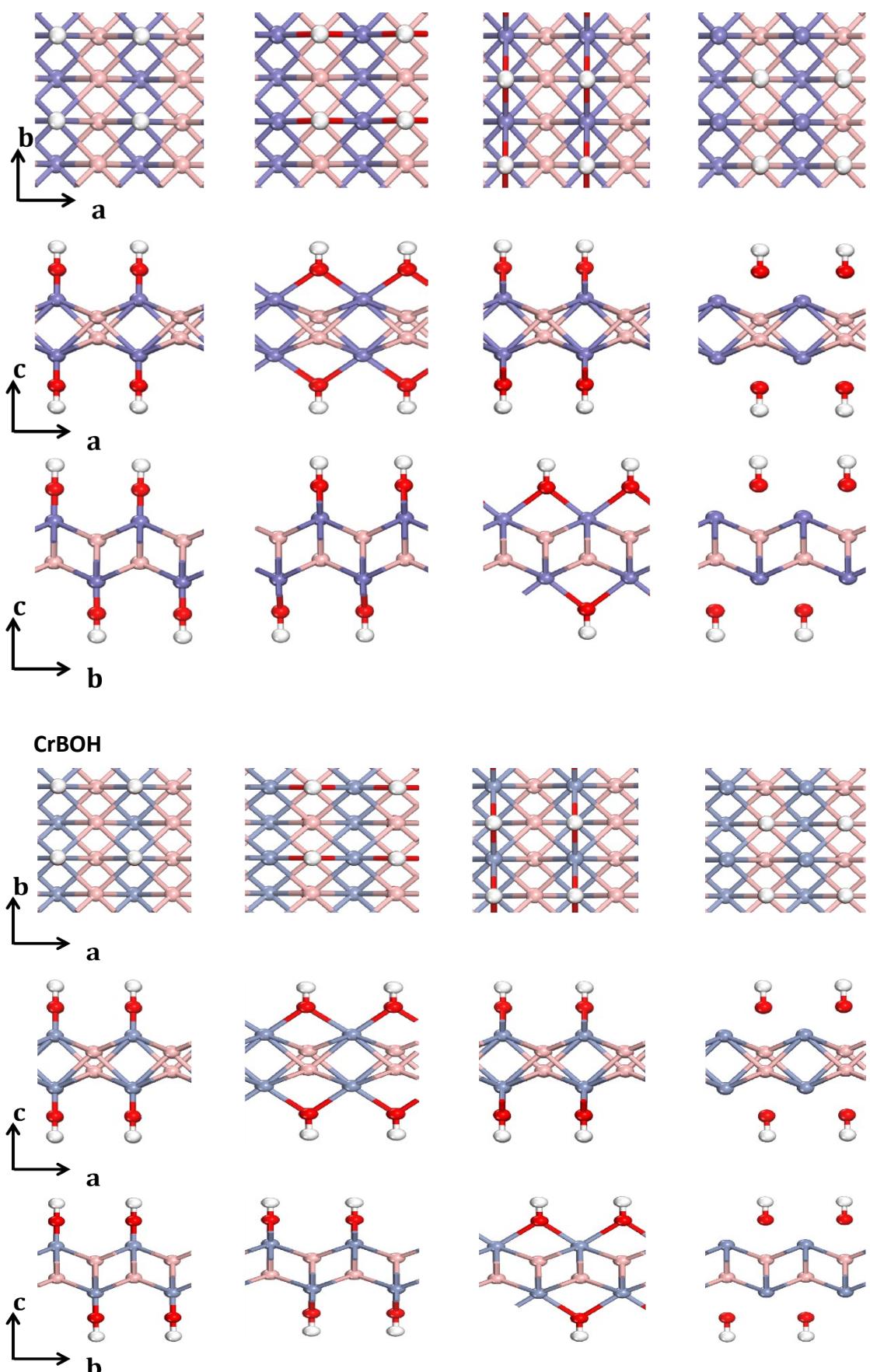


Figure S3. In this work, we considered four possible positions (P1, P2, P3 and P4) for F, O and OH functional groups adsorbing on the 2D TMB monolayer as shown in Figure S2.

Table S2. The adsorption energies (E_f) of the TMBX(X=F,O,OH) monolayer in four forms respectively. The adsorption energies is described as: $E_f = (E_{TMBX} - E_{TMB} - nE_X) / n$

E_f(eV)	CrBF	CrBOH	CrBO	FeBO	FeBF	FeBOH
P1	-4.61	-3.85	-4.65	-3.54	-3.64	-3.68
P2	-4.41	-3.87	-4.95	-3.74	-3.65	-3.70
P3	-4.69	-4.20	-5.28	-3.66	-3.64	-3.63
P4	-3.62	-3.22	-4.42	-3.37	-2.96	-3.23

Table S3. The total energies of the unit cell of TMBX(X=F,O,OH) in four forms respectively.

Total Energy(eV)	CrBF	CrBOH	CrBO	FeBO	FeBF	FeBOH
P1	-43.08	-54.44	-39.60	-32.32	-39.11	-52.01
P2	-42.67	-54.48	-40.16	-32.56	-39.12	-52.04
P3	-43.24	-55.12	-40.80	-32.55	-39.11	-51.90
P4	-41.09	-53.13	-39.10	-31.97	-37.75	-51.10

References:

1. S. Cahangirov, M. Topsakal, E. Aktürk, H.Şahin and S. Ciraci, *Phys. Rev. Lett.*, 2009, **102**, 236804.
2. A. J. Mannix, X. F. Zhou, B. Kiraly, J. D. Wood, D. Alducin, B. D. Myers, X. Liu, B. L. Fisher, U. Santiago, J. R. Guest, M. J. Yacaman, A. Ponce, A. R. Oganov, M. C. Hersam and N. P. Guisinger, *Science*, 2015, **350**, 1513-1516.
3. M. Derivaz, D. Dentel, R. Stephan, M. -C. Hanf, A. Mehdaoui, P. Sonnet and C. Pirri, *Nano Lett.*, 2015, **15**, 2510-2516.
4. L. Li, S. -Z. Lu, J. Pan, Z. Qin, Y. -Q. Wang, Y. Wang, G. -Y. Cao, S. Du and H. -J. Gao, *Adv. Mater.*, 2014, **26**, 4820-4824.
5. M. E. Dávila and G. L. Lay, *Sci. Rep.*, 2016, **6**, 20714.
6. B. Feng, J. Zhang, Q. Zhong, W. Li, S. Li, H. Li, P. Cheng, S. Meng, L. Chen and K. Wu, *Nat. Chem.*, 2016, **8**, 563-568.
7. Z. Zhang, E. S. Penev and B. I. Yakobson, *Nat. Chem.*, 2016, **8**, 525-527.