

Supporting Information for

Structures and magnetism of $\{\text{Ni}_2\text{Na}_2\}$, $\{\text{Ni}_4\}$ and $\{\text{Ni}_6^{\text{II}}\text{Ni}^{\text{III}}\}$

2-Hydroxy-3-alkoxy-benzaldehyde clusters

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Table S1. Selected bond lengths [Å] and angles [°] for **1**

Ni1–O4	2.000(2)	O5– Ni1–Ni1 ⁱ	172.25(11)
Ni1–O1	2.011(2)	O4– Ni1–Ni1	96.38(10)
Ni1–O5	2.031(3)	O1– Ni1–Ni1	89.56(11)
Ni1–O3	2.062(3)	O5– Ni1–Ni1	93.64(12)
Ni1–Ni1 ⁱ	2.092(3)	O3– Ni1–Ni1	174.02(12)
Ni1–N1	2.112(3)	N1 ⁱ – Ni1–Ni1	79.37(14)
Ni1–Ni1 ⁱ	3.235(2)	O4 ⁱ –Na1–O1	97.84(11)
Ni1–Na1 ⁱ	3.485(2)	O4 ⁱ –Na1–O7	121.46(12)
Na1–O4 ⁱ	2.268(3)	O1–Na1–O7	133.91(12)
Na1–O1	2.292(3)	O4 ⁱ –Na1–O2	134.33(11)
Na1–O7	2.325 (4)	O1–Na1–O2	66.46(10)
Na1–O2	2.454(3)	O7–Na1–O2	96.14(12)
Na1–O6 ⁱ	2.527(3)	O4 ⁱ –Na1–O6 ⁱ	65.94(10)
Na1–Ni1 ⁱ	2.670(3)	O1–Na1–O6 ⁱ	135.49(11)
Ni1 ⁱ –Ni1–Ni1	100.63(14)	O7–Na1–O6 ⁱ	85.88(12)
O4–Ni1–O1	173.78(9)	O2–Na1–O6 ⁱ	94.78(10)
O4–Ni1–O5	89.86(11)	O4 ⁱ –Na1–N1	32.79(6)
O1–Ni1–O5	87.93(11)	O1–Na1–N1	82.18(7)
O4–Ni1–O3	87.23(11)	O2–Na1–N1	117.40(10)
O1– Ni1–O3	87.01(10)	O7–Na1–N1	144.99(19)
O5– Ni1–O3	91.13(11)	O6–Na1–N1	97.21(8)
O4– Ni1–Ni1 ⁱ	87.60(11)	N3–N2–N1	179.7(5)
O1– M1–Ni1 ⁱ	95.32(11)		

Symmetry codes: (i) $(i) - x + 1, -y + 1, -z + 2$.

Table S2. Selected bond lengths [Å] and angles [°] for **2**.

Ni1–O2	2.022(6)	O1 ⁱⁱ –Ni1–O4	93.0(2)
Ni1–O1	2.013(5)	O2 ⁱⁱ –Ni1–O4	98.0(2)
Ni1–O4 ⁱ	2.052(5)	O4 ⁱⁱ –Ni1–O4	80.5(3)
Ni1–O4 ⁱⁱ	2.061(5)	O4 ⁱ –Ni1–O4 ⁱⁱ	82.8(2)
Ni1–O4	2.045(5)	O1–Ni1–O5	94.8(2)

Ni1–O5	2.107(12)	O2–Ni1–O5	88.9(2)
Ni1–O5'	2.18(4)	O4–Ni1–O5 ⁱ	91.2(2)
Ni ⁱⁱⁱ –Ni	3.052–3.127	O4 ⁱ –Ni1–O5	89.9(2)
O1–Ni1–O2	90.6(2)	O4 ⁱⁱ –Ni1–O5	169.5(2)
O1–Ni1–O4	172.6(2)	O4 ⁱ –Ni1–O4	83.3(2)
O2–Ni1–O4	93.8(2)	Ni1–O4–Ni1 ⁱⁱⁱ	96.5(2)
O1 ⁱ –Ni1–O4	92.4(2)	Ni1 ⁱⁱ –O4–Ni1	96.5(2)
O2 ⁱ –Ni1–O4	176.9(2)	Ni1 ⁱⁱⁱ –O4–Ni1 ⁱⁱ	95.9(2)

Symmetry codes: (i) $-x + 1/2, -y + 3/2, z$; (ii) $-y + 1, x + 1/2, -z + 1$; (iii) $y - 1/2, -x + 1, -z + 1$.

Table S3. Selected bond lengths [Å] and angles [°] for **3**.

Ni1–O10	1.997(4)	O4 ^a –Ni2–O2	101.8(2)
Ni1–O10 ^a	1.997(4)	O6 ^a –Ni2–O2	95.2(3)
Ni1–O12 ^a	2.005(4)	O10–Ni2–O2	153.3(2)
Ni1–O12	2.005(4)	O12 ^a –Ni2–O2	95.9(2)
Ni1–O11 ^a	2.018(4)	O1–Ni2–O2	75.6(2)
Ni1–O11	2.018(4)	O1–Ni3–O7	176.7(2)
Ni2–O1	1.965(5)	O1–Ni3–O3	87.9(3)
Ni2–O4 ^a	1.970(5)	O7–Ni3–O3	94.2(3)
Ni2–O6 ^a	2.039(6)	O1–Ni3–O11	101.6(2)
Ni2–O10	2.078(5)	O7–Ni3–O11	80.5(2)
Ni2–O12 ^a	2.093(5)	O3–Ni3–O11	99.9(2)
Ni2–O2	2.183(6)	O1–Ni3–O10	78.7(2)
Ni3–O1	1.938(5)	O7–Ni3–O10	99.3(2)
Ni3–O7	1.957(5)	O3–Ni3–O10	166.1(3)
Ni3–O3	2.073(7)	O11–Ni3–O10	79.3(2)
Ni3–O11	2.084(4)	O1–Ni3–O8	100.9(2)
Ni3–O10	2.105(5)	O7–Ni3–O8	76.4(2)
Ni3–O8	2.181(5)	O3–Ni3–O8	92.6(3)
Ni4–O7	1.959(5)	O11–Ni3–O8	154.5(2)
Ni4–O4	1.979(5)	O10–Ni3–O8	93.7(2)
Ni4–O9	2.037(6)	O7–Ni4–O4	174.7(2)
Ni4–O12	2.080(4)	O7–Ni4–O9	88.3(2)
Ni4–O11	2.119(5)	O4–Ni4–O9	94.4(2)
Ni4–O5	2.151(5)	O7–Ni4–O12	103.8(2)
Ni1 ⁱⁱⁱ –Ni2	3.093(1)	O4–Ni4–O12	80.4(2)
Ni1 ⁱⁱⁱ –Ni3	3.111(1)	O9–Ni4–O12	97.0(2)

Ni1...Ni4	3.106(1)	O7–Ni4–O11	79.5(2)
Ni2...Ni3	3.114(1)	O4–Ni4–O11	98.3(2)
Ni2...Ni4 ^a	3.098(1)	O9–Ni4–O11	166.2(2)
Ni3...Ni4	3.103(2)	O12–Ni4–O11	79.8(2)
O10–Ni1–O12 ^a	83.6(2)	O7–Ni4–O5	99.2(2)
O10–Ni1–O12	96.4(2)	O4–Ni4–O5	76.1(2)
O10–Ni1–O11 ^a	96.0(2)	O9–Ni4–O5	96.2(2)
O10 ⁱ –Ni1–O11 ^a	84.0(2)	O12–Ni4–O5	153.8(2)
O12 ^a –Ni1–O11 ^a	83.7(2)	O11–Ni4–O5	92.2(2)
O12–Ni1–O11 ^a	96.3(2)	Ni3–O1–Ni2	105.8(2)
O1–Ni2–O4 ^a	176.7(2)	Ni4–O7–Ni3	104.4(2)
O1–Ni2–O6 ^a	95.0(2)	Ni1–O10–Ni2	98.8(2)
O4 ^a –Ni2–O6 ^a	87.2(2)	Ni1–O10–Ni3	98.5(2)
O1–Ni2–O10	79.0(2)	Ni2–O10–Ni3	96.3(2)
O4 ^a –Ni2–O10	103.3(2)	Ni1–O11–Ni3	98.7(2)
O6 ^a –Ni2–O10	95.0(2)	Ni1–O11–Ni4	97.2(2)
O1–Ni2–O12 ^a	98.1(2)	Ni3–O11–Ni4	95.1(2)
O4 ^a –Ni2–O12 ^a	80.0(2)	Ni1–O12–Ni4	98.9(2)
O6 ^a –Ni2–O12 ^a	164.6(2)	Ni1–O12–Ni2 ^a	97.8(2)
O10–Ni2–O12 ^a	79.7(2)	Ni4–O12–Ni2 ^a	95.9(2)

Symmetry codes: (a) $-x + 1/2, -y + 3/2, -z$.

Table S4 Coordination number (*N*), bond valence sum (*BVS*), expected atomic valence (*V*) and deviation from the expected atomic valence (ΔV) for the nickel cations.

	Ni1	Ni2	Ni3	Ni4
N	6	6	6	6
BVS	2.98	2.07	2.06	2.07
V	+3	+2	+2	+2
ΔV	0.02	0.07	0.06	0.07

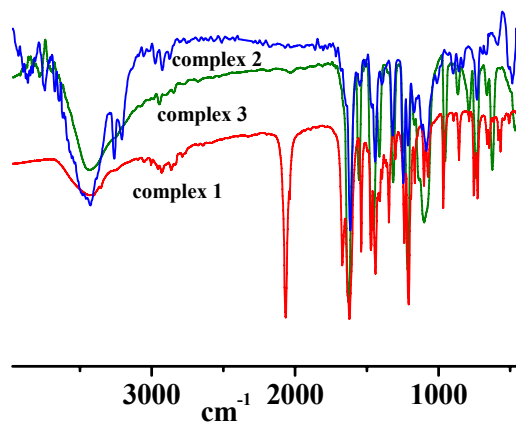
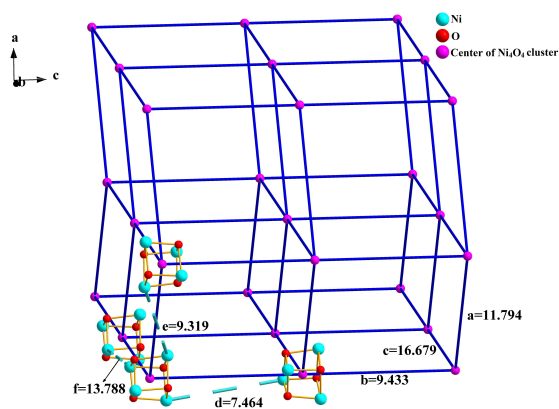


Fig. S1. IR contrast of 1-3.



Scheme S1. Illustration of correlation representing the distance (a,b,c) between the adjacent $\{\text{Ni}_4\text{O}_4\}$ center, along a,b,c axial, (e, f, d) the $\text{M}\cdots\text{M}$ distances of adjacent $\{\text{Ni}_4\text{O}_4\}$ in a, b, c axial.

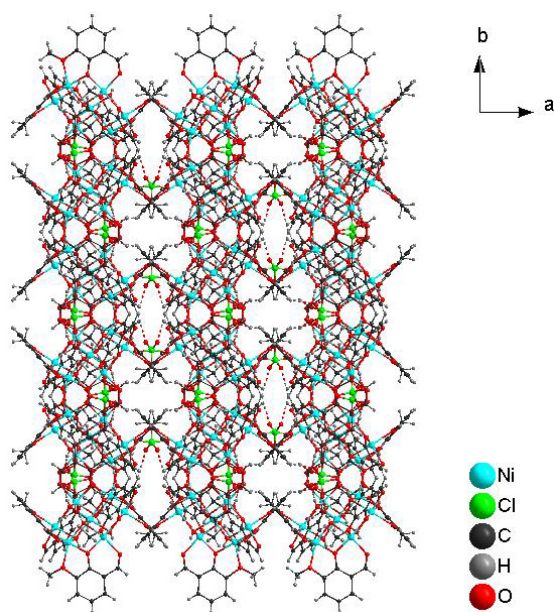


Fig. S2 Packing drawing of the 3.

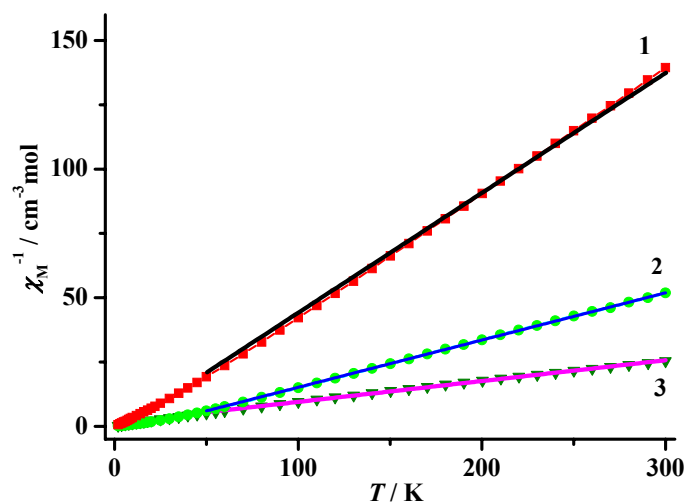


Fig. S3 Plot of χ_M^{-1} vs T for 1 – 3. The solid line represents a fit of data in the temperature range 50-300 K.

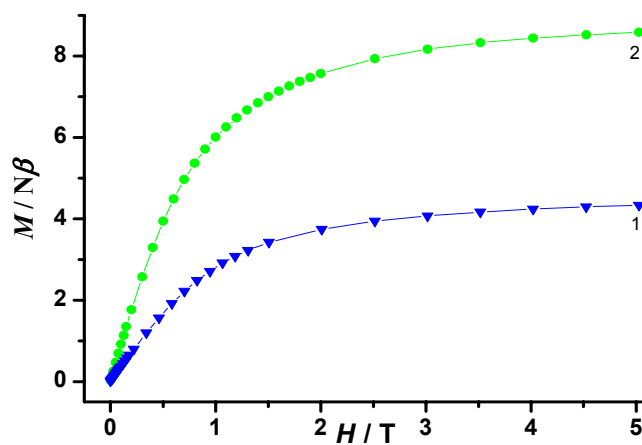


Fig. S4 Field dependence of magnetization at 2 K for 1 – 2.

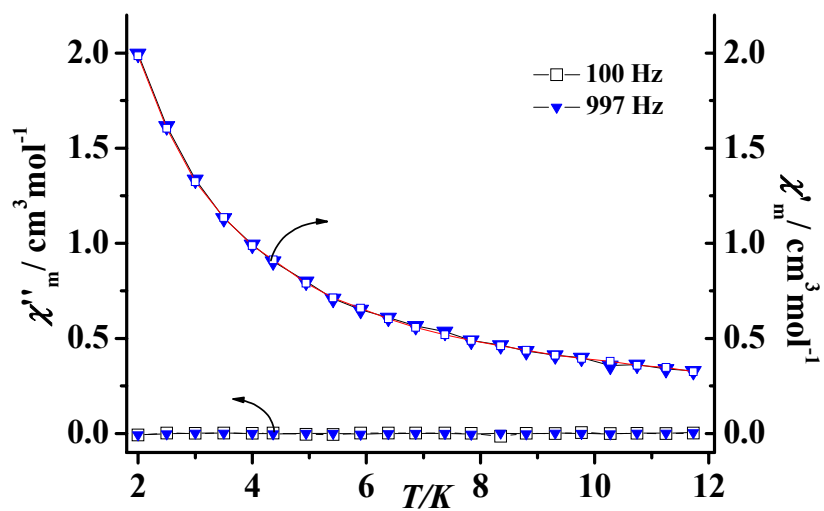


Fig. S5 In-phase and out-of-phase ac the range of 2-11.7 K for 1.

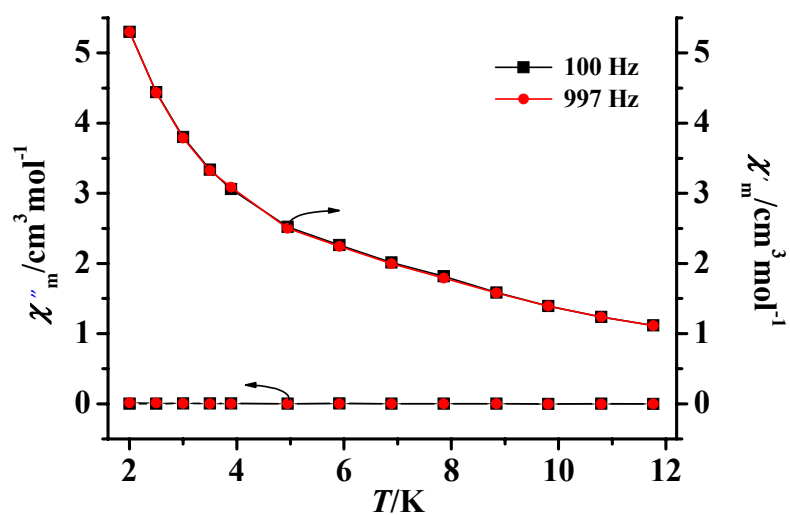


Fig. S6 In-phase and out-of-phase ac the range of 2-12 K for 2.