

## Electronic Supplementary Information for:

# Two-step hydrothermal synthesis of submicron $\text{Li}_{1+x}\text{Ni}_{0.5}\text{Mn}_{1.5}\text{O}_{4-\delta}$ for lithium-ion battery cathodes ( $x = 0.02$ , $\delta = 0.12$ )

Xiaoguang Hao, Mark H. Austin and Bart M. Bartlett

University of Michigan Department of Chemistry  
930 N. University Avenue  
Ann Arbor, MI 48109-1055 USA  
Tele: +1 (734) 615-9279

## Table of Contents

<i>Index</i>	<i>Page</i>
Table S1. Refinement parameters for LNMO synthesized in air	S2
Table S2. Atomic coordinates and isotropic thermal parameters for LNMO synthesized in air.	S2
Fig. S1. XRD pattern of the NMO intermediate phases	S3
Fig. S2. EDX elemental map of the NMO intermediate	S3
Fig. S3. EDX elemental map of the LNMO product	S4
Fig. S4. XRD pattern of the one-pot synthesis LNMO product	S5
Fig. S5. SEM image of the one-pot synthesis LNMO product	S5
Fig. S6. XRD pattern of the products of the LNMO reaction performed under $\text{O}_2$	S6
Fig. S7. XRD pattern of LNMO synthesized under $\text{N}_2$	S7
Table S3. Refinement parameters for LNMO synthesized under $\text{N}_2$	S7
Table S4. Atomic coordinates and isotropic thermal parameters for LNMO synthesized under $\text{N}_2$	S7
Fig. S8. FTIR spectrum of NMO intermediate and LNMO product	S8
Fig. S9. $\text{N}_2$ sorption isotherms of the LNMO product	S8
Fig. S10. View of the spinel crystal structure along the [1 1 0] direction	S9
Fig. S11. C(1s) and K(2p) XP spectra of NMO and LNMO	S9
Fig. S12. Voltage-dependence of $D_{\text{Li}}$ in the LNMO product	S10
Fig. S13. Nyquist plots from EIS data collected from LNMO	S11
Fig. S14. Voltage profiles for LNMO cycles 1000 times between varying voltage limits	S12
Fig. S15. XRD pattern for a freshly prepared LNMO electrode	S13
Table S5. Refinement parameters for a freshly prepared LNMO electrode	S13
Table S6. Atomic coordinates and isotropic thermal parameters for a freshly prepared LNMO electrode	S13
Fig. S16. XRD pattern for an LNMO electrode after 60 cycles between 3.4 – 5.0 V	S14

*Table of Contents Continued*

Table S7. Refinement parameters for an LNMO electrode after 60 cycles between 3.4 – 5.0 V	S14
Table S8. Atomic coordinates and isotropic thermal parameters for an LNMO electrode after 60 cycles between 3.4 – 5.0 V	S14
Fig. S17. XRD pattern for an LNMO electrode with a 4.4 V upper voltage cut-off	S15
Table S9. Refinement parameters for an LNMO electrode with a 4.4 V upper voltage cut-off	S15
Table S10. Atomic coordinates and isotropic thermal parameters for an LNMO electrode with a 4.4 V upper voltage cut-off freshly prepared LNMO electrode	S15
Fig. S18. XRD pattern for an LNMO electrode with a 4.4 V lower voltage cut-off	S16
Table S11. Refinement parameters for an LNMO electrode with a 4.4 V lower voltage cut-off	S16
Table S12. Atomic coordinates and isotropic thermal parameters for an LNMO electrode with a 4.4 V lower voltage cut-off freshly prepared LNMO electrode	S16
Fig. S19. XRD pattern for an LNMO electrode after 1000 cycles with varying electrochemical windows.	S17
Table S13. Refinement parameters for an LNMO electrode after 1000 cycles with varying electrochemical windows	S17
Table S14. Atomic coordinates and isotropic thermal parameters for an LNMO electrode after 1000 cycles with varying electrochemical windows	S17

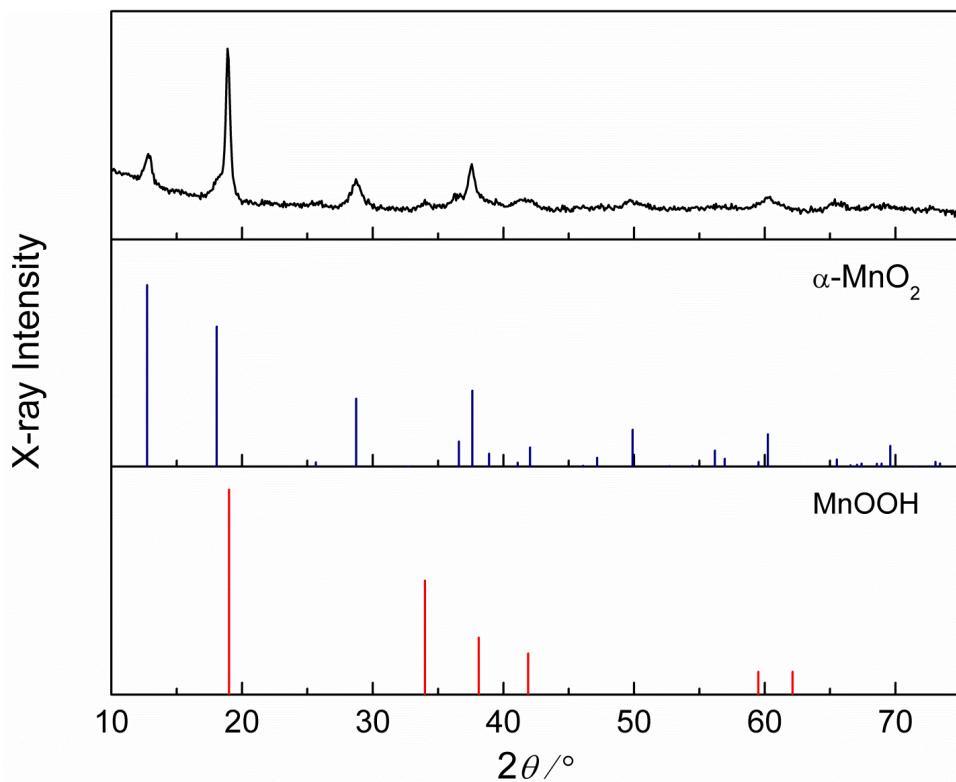
**Table S1.** Refinement parameters for LNMO synthesized in air. (The experimental XRD pattern is shown as Fig. 1 in the text).

Space Group	$Fd\bar{3}m$
Scale	$5.460 \times 10^{-4}$
Lattice parameter, $a$ ( $\text{\AA}$ )	8.183
Unit Cell Mass ( $\text{g}\cdot\text{mol}^{-1}$ )	1450.260
Unit Cell Volume ( $\text{\AA}^3$ )	548.141
Crystallite Size, Lorentzian (nm)	120.8
Crystal Density ( $\text{g}\cdot\text{cm}^3$ ), calculated	4.393
Crystal Linear Absorption Coefficient ( $\text{cm}^{-1}$ )	592.397
Wt%-Rietveld	86.398
$R_{\text{exp}}/ R_{\text{exp}}'$ <sup>(a)</sup>	0.93/1.69
$R_{\text{wp}}/ R_{\text{wp}}'$	1.85/3.34
$R_p/ R_p'$	1.23/2.40
$R_{\text{Bragg}}$	3.607
$GOF$	1.98
$DW_d$	0.76

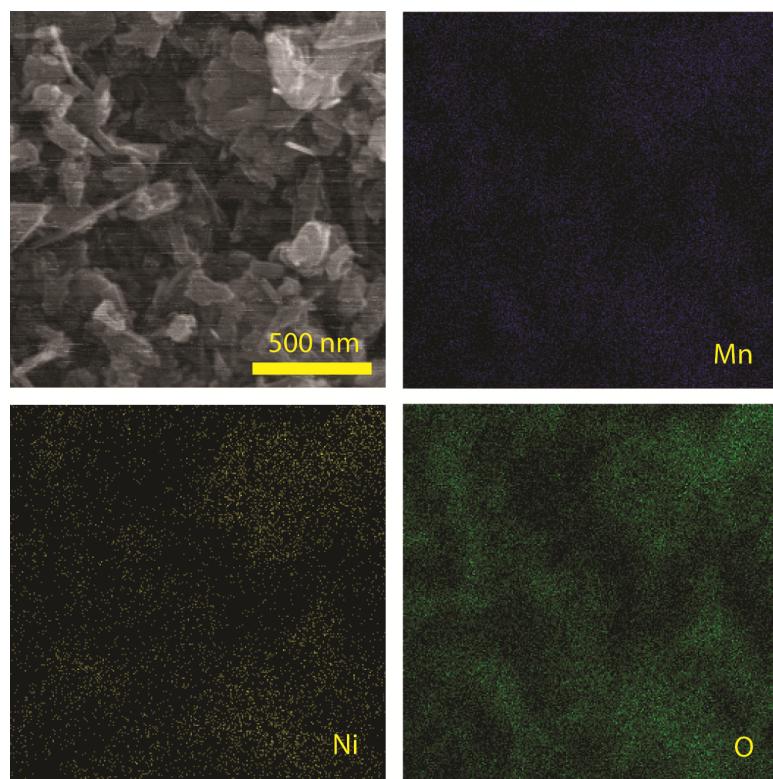
(a)-Primed parameters are background corrected.

**Table S2.** Atomic coordinates and isotropic thermal parameters for LNMO synthesized in air.

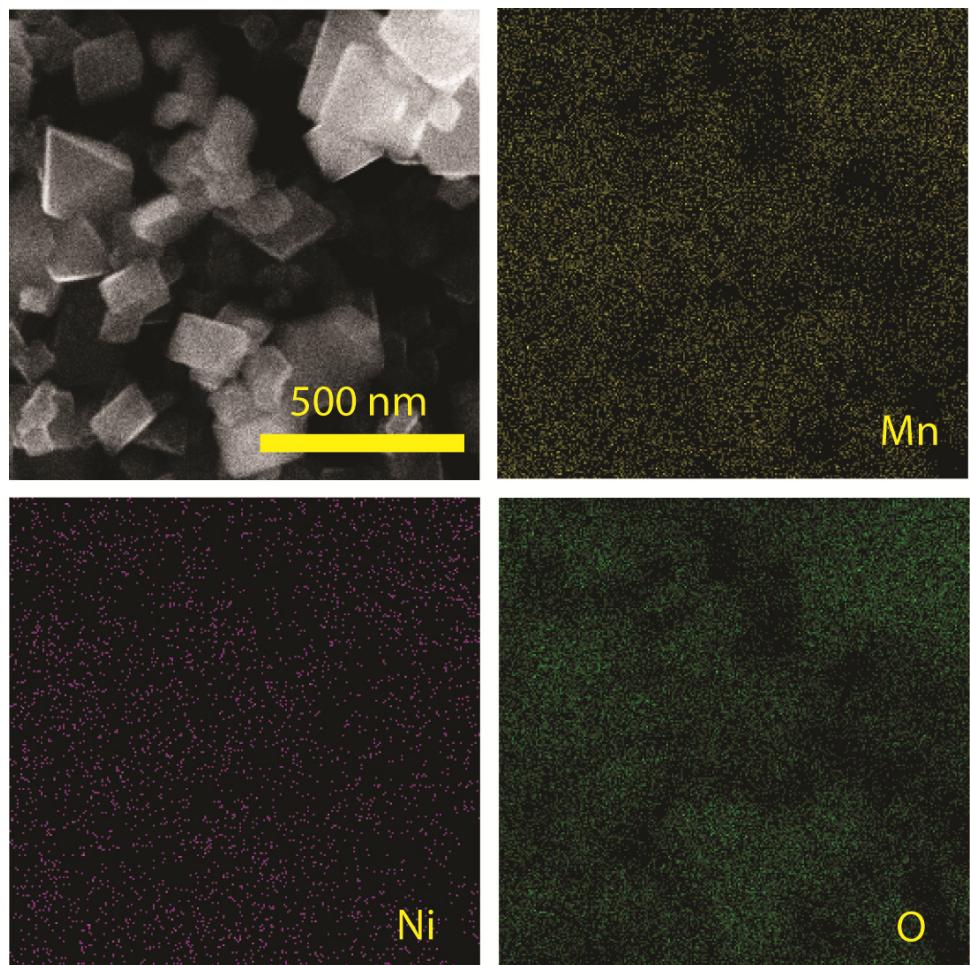
Atom	Wyckoff Site	x	y	z	SOF	$B_{\text{eq}}$
Li	8a	0	0	0	1	2.924
Mn	16d	0.6250	0.6250	0.6250	0.75	1.758
Ni	16d	0.6250	0.6250	0.6250	0.25	0.9867
O	32e	0.38723	0.38723	0.38723	1	-0.06318



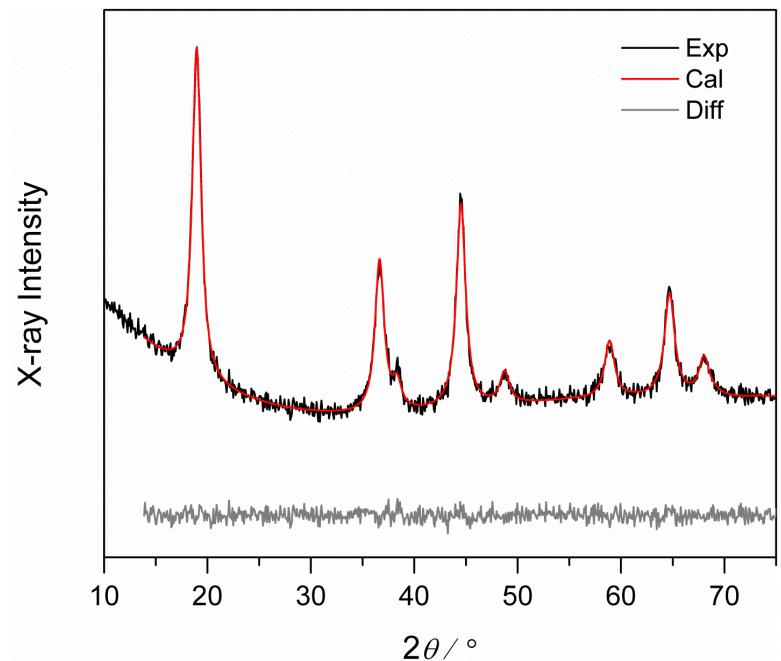
**Fig. S1.** Indexed XRD pattern of the NMO intermediate ( $H_{0.4}K_{0.08}Ni_{0.5}Mn_{1.5}O_{3.305}$ )



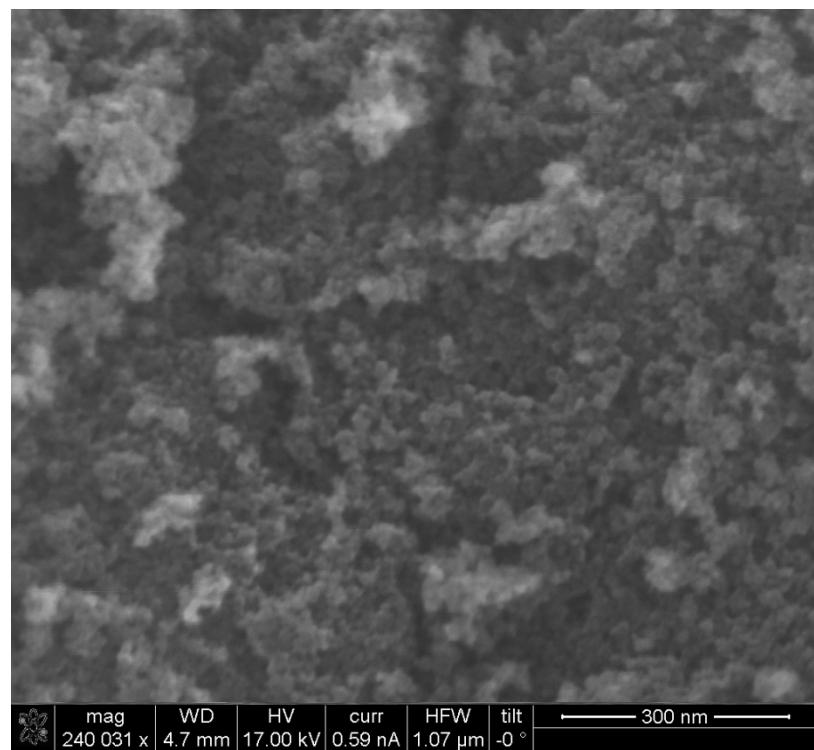
**Fig. S2** EDX elemental map of the NMO intermediate



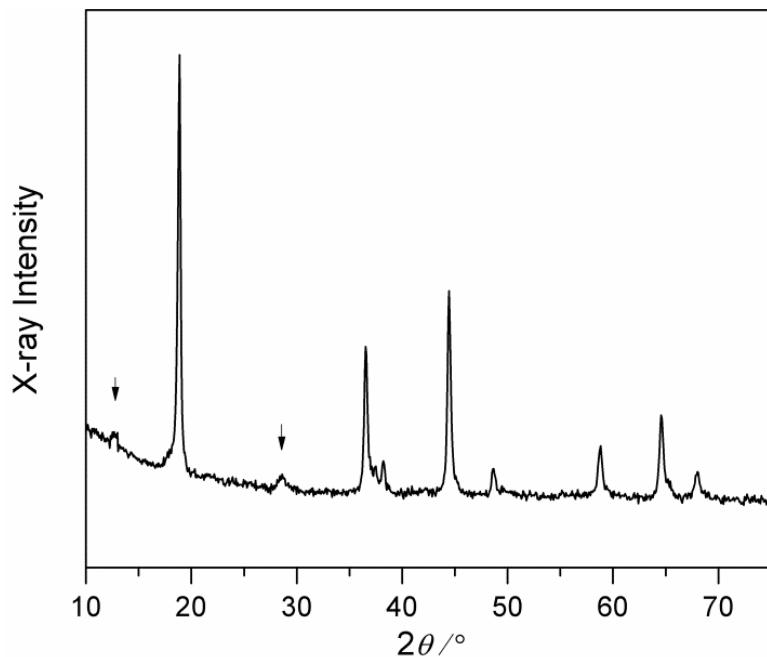
**Fig. S3.** EDX elemental map of the LNMO product



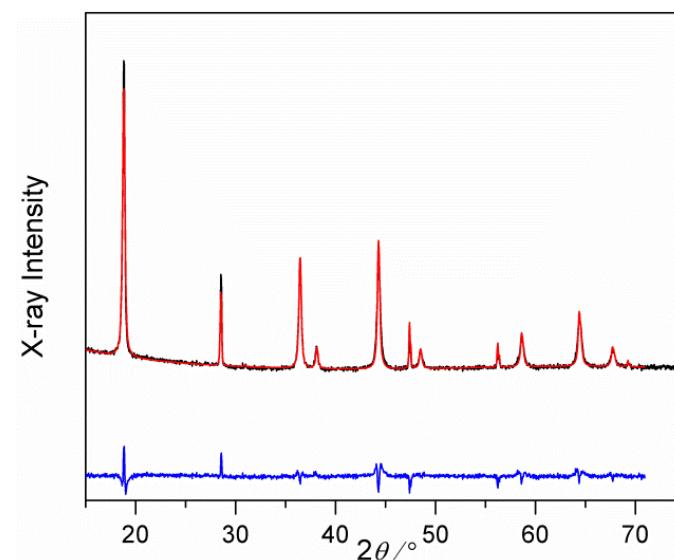
**Fig. S4.** XRD pattern of the one-pot synthesis LNMO product.



**Fig. S5.** SEM image of the one-pot synthesis LNMO product.



**Fig. S6.** XRD pattern of the products of the LNMO reaction performed under O<sub>2</sub>. The arrows point out reflections for the  $\alpha$ -MnO<sub>2</sub> impurity phase. Since the reaction product is not phase-pure, no Rietveld refinement was carried out.



**Fig. S7.** XRD data (black), calculated Rietveld refinement (red), and difference pattern (blue) for LNMO with second step prepared in N<sub>2</sub>.

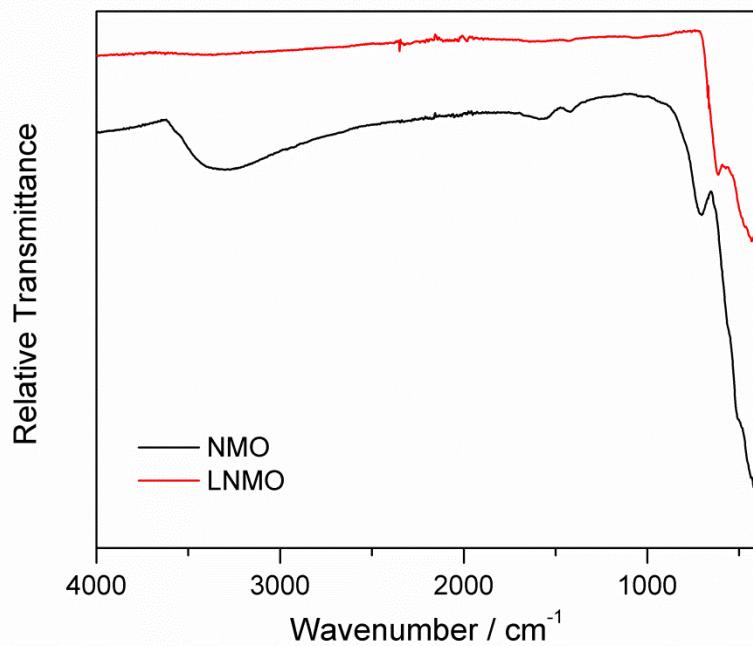
**Table S3.** Refinement parameters for LNMO synthesized under N<sub>2</sub>

Space Group	<i>Fd</i> $\bar{3}m$
Scale	$5.328 \times 10^{-4}$
Lattice parameter, <i>a</i> (Å)	8.196
Unit Cell Mass (g•mol <sup>-1</sup> )	1450.260
Unit Cell Volume (Å <sup>3</sup> )	549.649
Crystallite Size, Lorentzian (nm)	111.9
Crystal Density (g•cm <sup>-3</sup> ), calculated	4.381
Crystal Linear Absorption Coefficient (cm <sup>-1</sup> )	590.771
Wt%-Rietveld	86.675
<i>R</i> <sub>exp</sub> / <i>R</i> <sub>exp'</sub> <sup>(a)</sup>	0.91/3.04
<i>R</i> <sub>wp</sub> / <i>R</i> <sub>wp'</sub>	1.94/6.49
<i>R</i> <sub>p</sub> / <i>R</i> <sub>p'</sub>	1.26/4.70
<i>R</i> <sub>Bragg</sub>	0.413
<i>GOF</i>	2.14
<i>DW</i> <sub>d</sub>	0.69

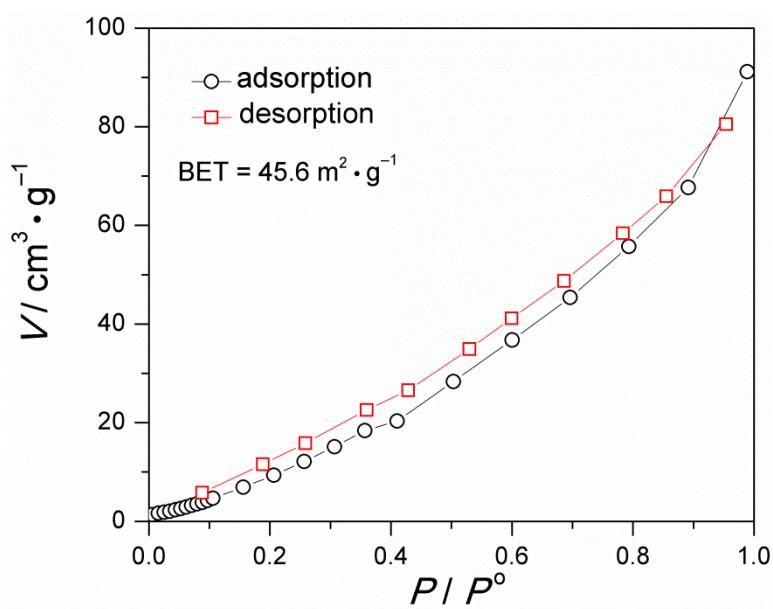
(a)-Primed parameters are background corrected.

**Table S4.** Atomic coordinates and isotropic thermal parameters for LNMO synthesized under N<sub>2</sub>

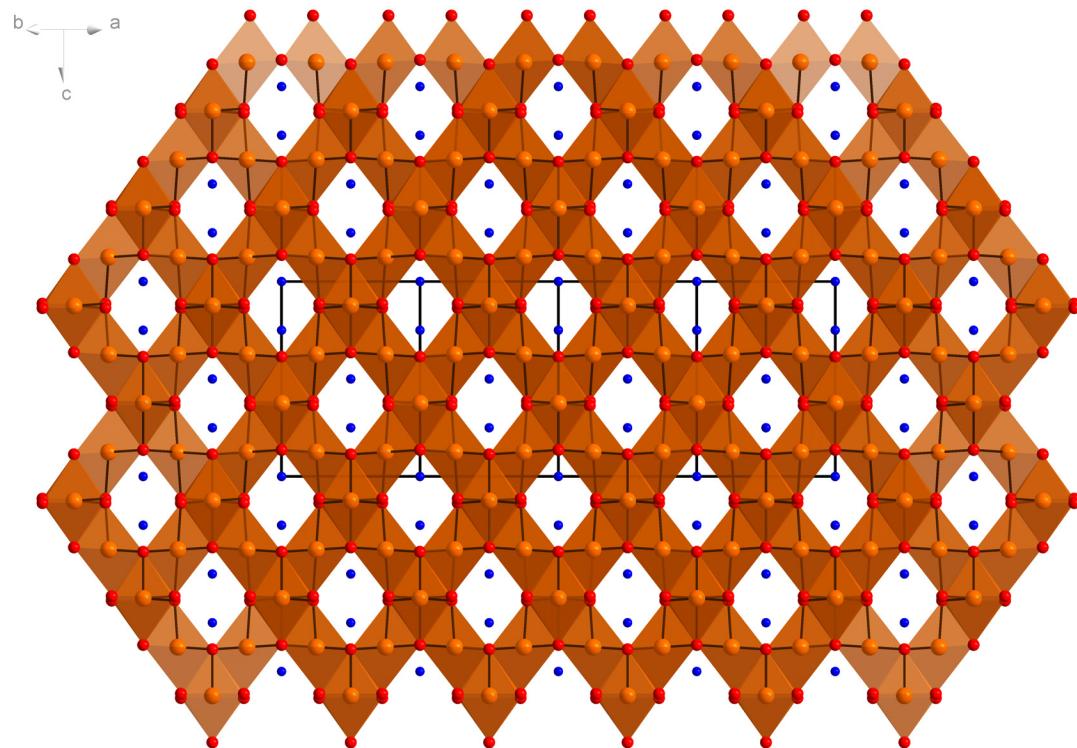
Atom	Wyckoff Site	<i>x</i>	<i>y</i>	<i>z</i>	SOF	<i>B</i> <sub>eq</sub>
Li	8 <i>a</i>	0	0	0	1	2.916
Mn	16 <i>d</i>	0.6250	0.6250	0.6250	0.75	1.962
Ni	16 <i>d</i>	0.6250	0.6250	0.6250	0.25	0.9938
O	32 <i>e</i>	0.38749	0.38749	0.38749	1	$-5.697 \times 10^{-5}$



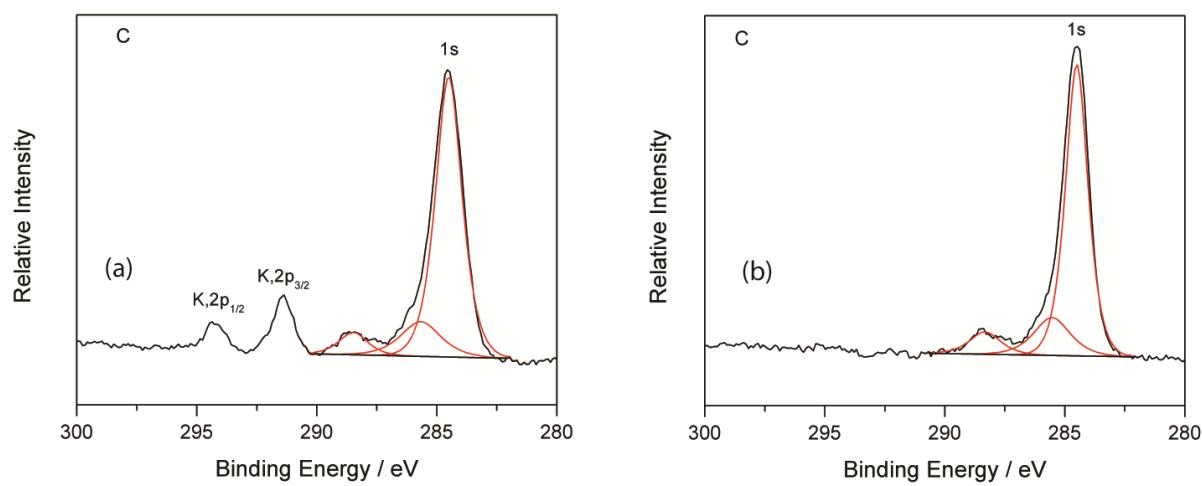
**Fig. S8.** FTIR spectra of the NMO intermediate and LNMO product.



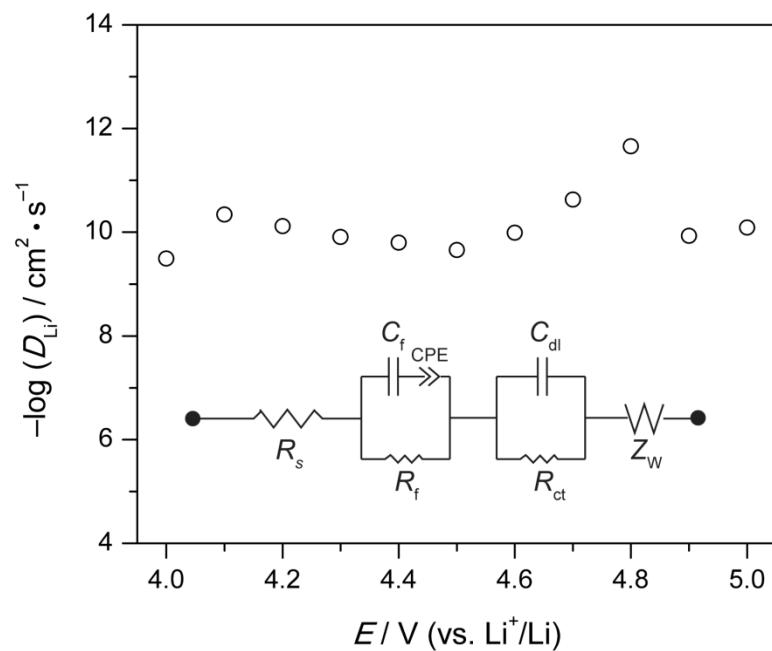
**Fig. S9.** N<sub>2</sub> sorption isotherm for the final LNMO product synthesized in the two-step hydrothermal method in air.



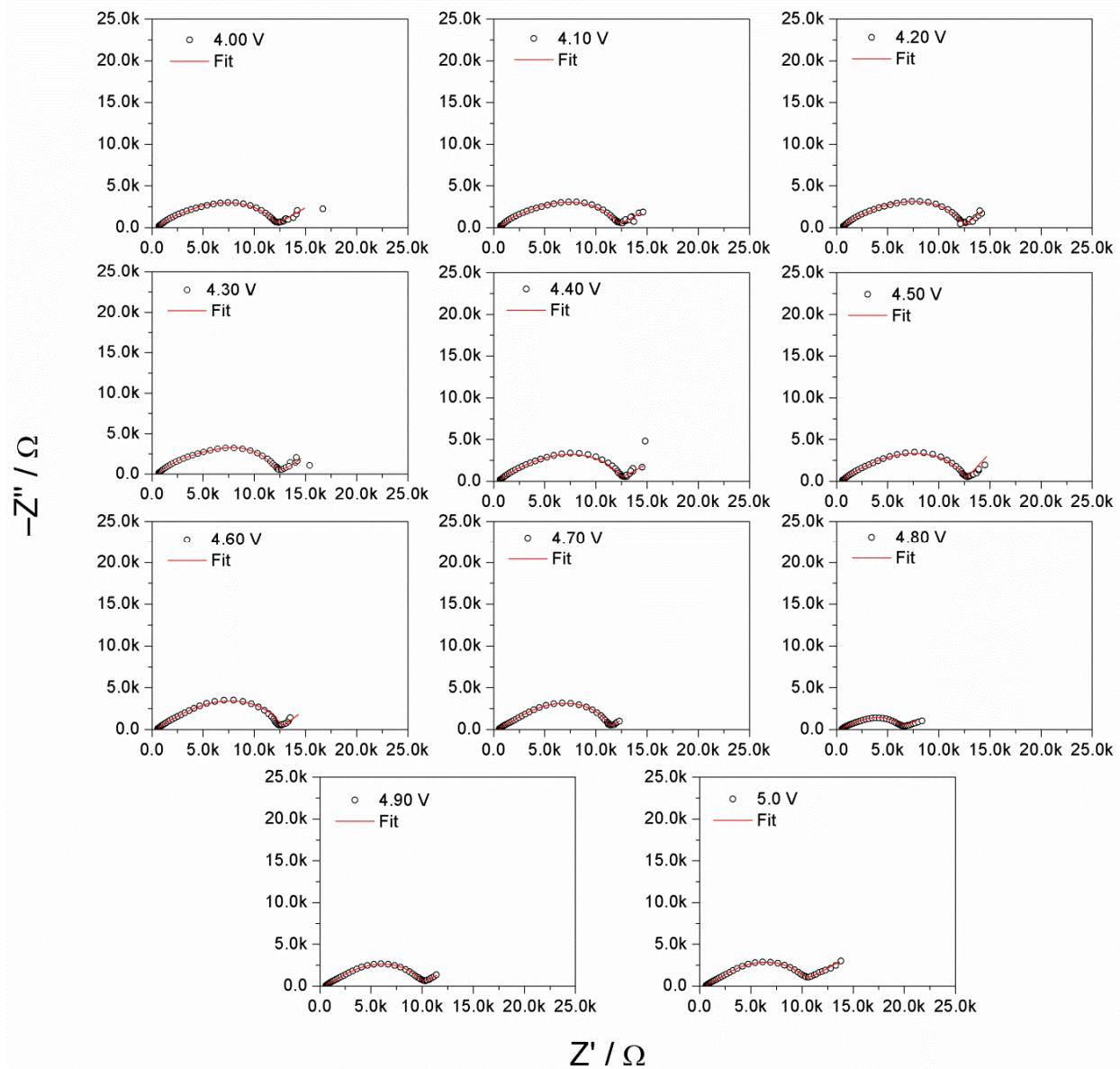
**Fig. S10.** View of the spinel crystal structure along the [1 1 0] direction. Blue, orange, and red spheres represent lithium, manganese/nickel, and oxygen atoms respectively.



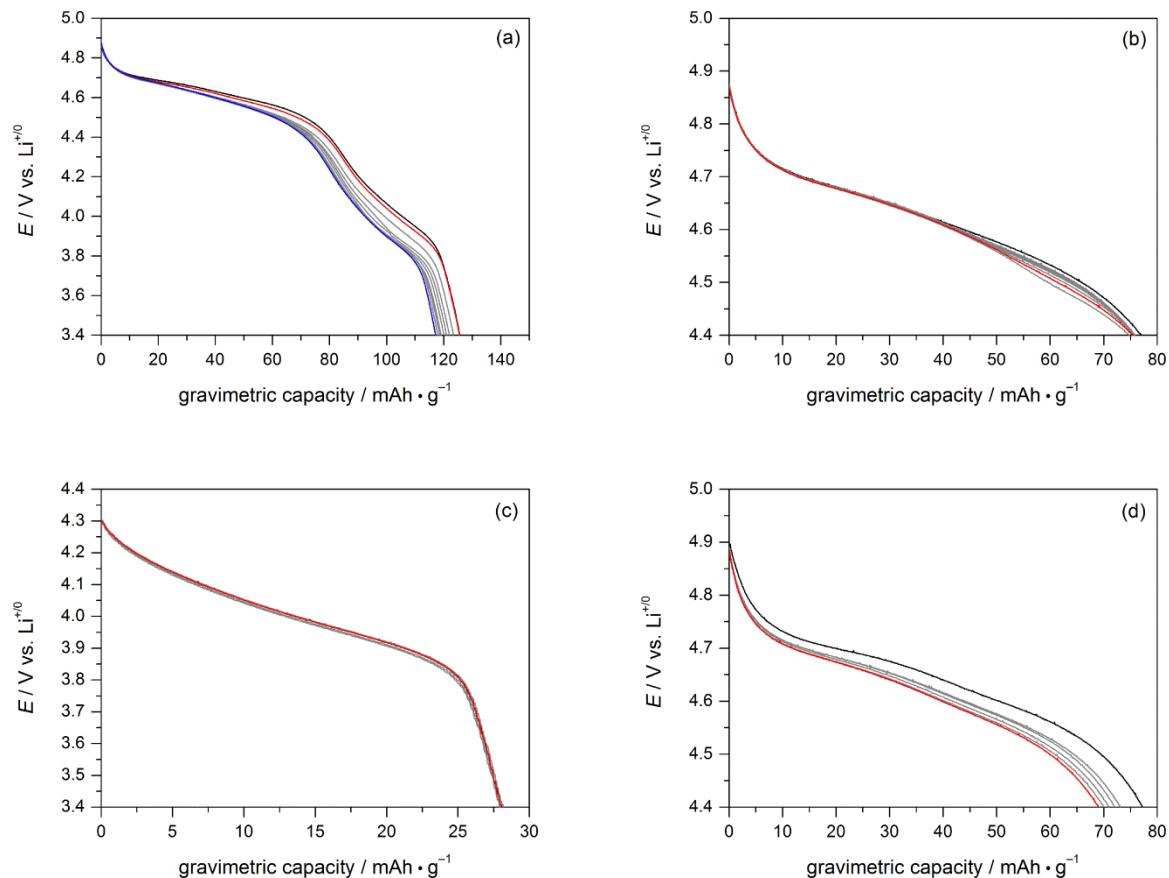
**Fig. S11.** C(1s) XP spectra for NMO (a) and LNMO (b). NMO also shows K(2p) features.



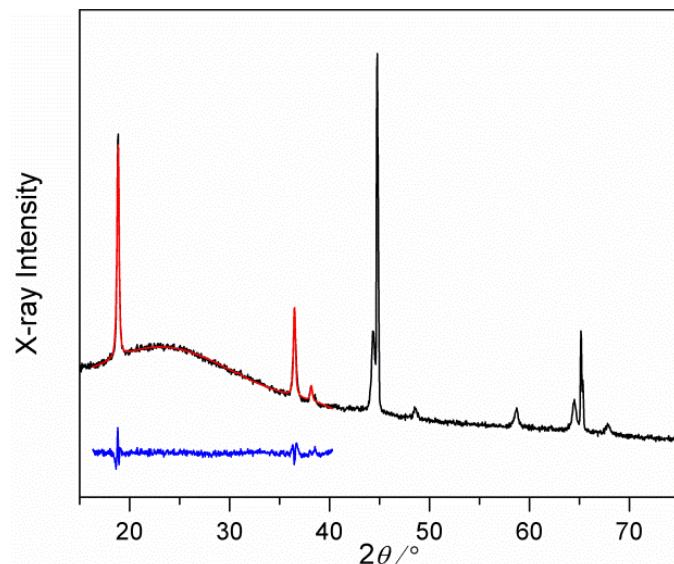
**Fig. S12** Voltage-dependence of the  $\text{Li}^+$  diffusion constant ( $D_{\text{Li}}$ ) determined from electrochemical impedance spectroscopy. The data are fit to the equivalent circuit illustrated in the inset.



**Fig S13.** Nyquist plots for determining  $D_{\text{Li}}$  of LNMO as a function of potential.



**Fig. S14.** Voltage profiles for the 1000 cycle LNMO cell. a) Cycles 1 – 10 performed between the voltage limits 3.4 – 5.0 V. Black, red, and blue represent cycles one, two, and ten respectively. Others are shown in gray. b) Cycles 11 – 121 performed between the voltage limits 4.4 – 5.0 V. Black and red represent cycles 11 and 121 respectively. Every 10<sup>th</sup> cycle in between is shown in gray. c) Cycles 122 – 622 performed between the voltage limits 3.4 – 4.4 V. Black and red present cycles 122 and 622 respectively with every 100<sup>th</sup> cycle in between shown in gray. d) Cycles 623 – 1123 performed between the voltage limits 4.4 – 5.0 V. Black and red represent cycles 623 and 1123 respectively with every 100<sup>th</sup> cycle in between shown in gray.



**Fig. S15.** XRD data (black), calculated Rietveld refinement (red), and difference pattern (blue) for a freshly prepared LNMO electrode.

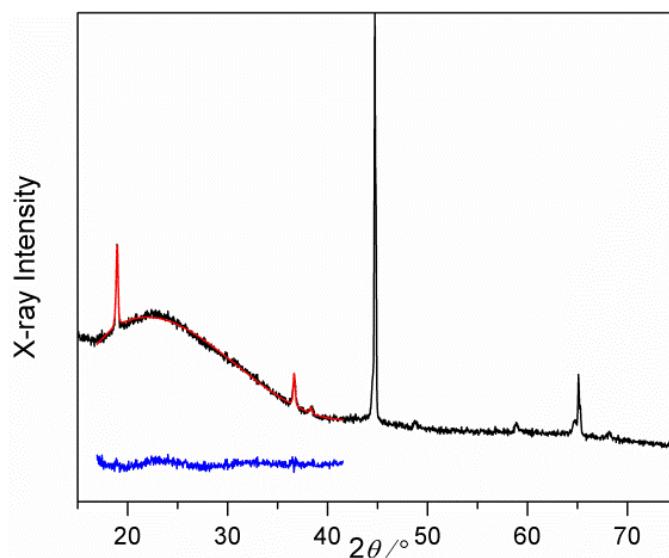
**Table S5.** Refinement parameters for a freshly prepared LNMO electrode.

Space Group	$Fd\bar{3}m$
Scale	$4.603 \times 10^{-5}$
Lattice parameter, $a$ (Å)	8.185
Unit Cell Mass (g•mol <sup>-1</sup> )	1450.260
Unit Cell Volume (Å <sup>3</sup> )	548.485
Crystallite Size, Lorentzian (nm)	68.8
Crystal Density (g•cm <sup>3</sup> ), calculated	4.391
Crystal Linear Absorption Coefficient (cm <sup>-1</sup> )	592.025
Wt%-Rietveld	100
$R_{\text{exp}}/R_{\text{exp}}'$ <sup>(a)</sup>	1.28/2.50
$R_{\text{wp}}/R_{\text{wp}}'$	1.93/3.78
$R_p/R_p'$	1.39/2.79
$R_{\text{Bragg}}$	0.166
$GoF$	1.51
$DW_d$	1.01

(a)-Primed parameters are background corrected.

**Table S6.** Atomic coordinates and isotropic thermal parameters for a freshly prepared LNMO electrode.

Atom	Wyckoff Site	x	y	z	SOF	B <sub>eq</sub>
Li	8a	0	0	0	1	-3.857
Mn	16d	0.6250	0.6250	0.6250	0.75	16.04
Ni	16d	0.6250	0.6250	0.6250	0.25	6.058
O	32e	0.37415	0.37415	0.37415	1	13.79



**Fig. S16.** XRD data (black), calculated Rietveld refinement (red), and difference pattern (blue) for an LNMO electrode after 60 cycles between 3.4 – 5.0 V.

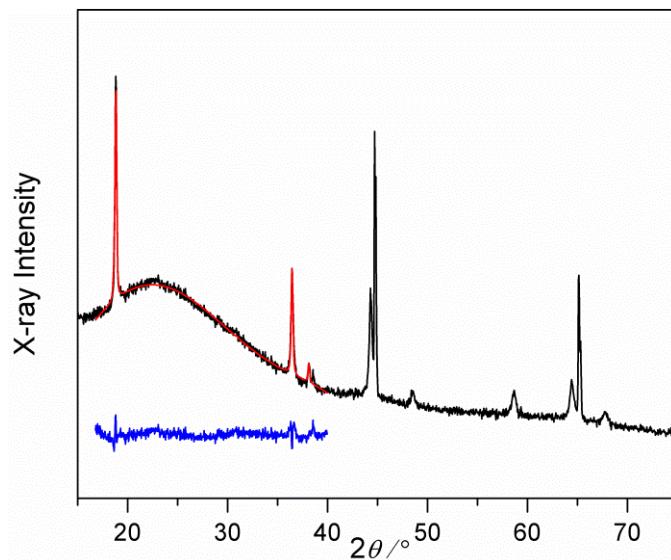
**Table S7.** Refinement parameters an LNMO electrode after 60 cycles between 3.4 – 5.0 V.

Space Group	$Fd\bar{3}m$
Scale	$1.187 \times 10^{-4}$
Lattice parameter, $a$ (Å)	8.157
Unit Cell Mass (g•mol <sup>-1</sup> )	1450.260
Unit Cell Volume (Å <sup>3</sup> )	542.886
Crystallite Size, Lorentzian (nm)	102.6
Crystal Density (g•cm <sup>3</sup> ), calculated	4.436
Crystal Linear Absorption Coefficient (cm <sup>-1</sup> )	598.131
Wt%-Rietveld	100
$R_{\text{exp}}/R_{\text{exp}}'$ <sup>(a)</sup>	1.60/3.15
$R_{\text{wp}}/R_{\text{wp}}'$	2.02/3.97
$R_p/R_p'$	1.62/3.22
$R_{\text{Bragg}}$	0.058
$GOF$	1.26
$DW_d$	1.31

(a)-Primed parameters are background corrected.

**Table S8.** Atomic coordinates and isotropic thermal parameters for an LNMO electrode after 60 cycles between 3.4 – 5.0 V.

Atom	Wyckoff Site	x	y	z	SOF	$B_{\text{eq}}$
Li	8a	0	0	0	1	7.31
Mn	16d	0.6250	0.6250	0.6250	0.75	18.71
Ni	16d	0.6250	0.6250	0.6250	0.25	17.79
O	32e	0.36358	0.36358	0.36358	1	20



**Fig. S17.** XRD data (black), calculated Rietveld refinement (red), and difference pattern (blue) for an LNMO electrode after 10 cycles 3.4 – 5.0 V and 50 cycles 3.4 - 4.4 V charge/discharge.

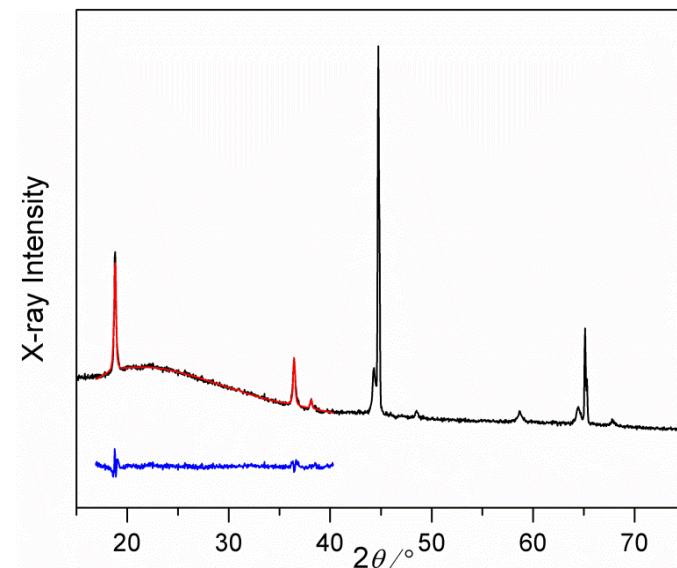
**Table S9.** Refinement parameters for LNMO after 10 cycles 3.4 – 5.0 V and 50 cycles 3.4 - 4.4 V charge/discharge.

Space Group	$Fd\bar{3}m$
Scale	$2.687 \times 10^{-4}$
Lattice parameter, $a$ ( $\text{\AA}$ )	8.194
Unit Cell Mass ( $\text{g}\cdot\text{mol}^{-1}$ )	1450.260
Unit Cell Volume ( $\text{\AA}^3$ )	550.261
Crystallite Size, Lorentzian (nm)	110.0
Crystal Density ( $\text{g}\cdot\text{cm}^{-3}$ ), calculated	4.376
Crystal Linear Absorption Coefficient ( $\text{cm}^{-1}$ )	590.114
Wt%-Rietveld	100
$R_{\text{exp}}/R_{\text{exp}}'$ <sup>(a)</sup>	1.45/2.85
$R_{\text{wp}}/R_{\text{wp}}'$	2.01/3.95
$R_{\text{p}}/R_{\text{p}}'$	1.55/3.10
$R_{\text{Bragg}}$	1.459
$GOF$	1.38
$DW_d$	1.21

(a)-Primed parameters are background corrected.

**Table S10.** Atomic coordinates and isotropic thermal parameters for LNMO after 10 cycles 3.4 – 5.0 V and 50 cycles 3.4 - 4.4 V charge/discharge.

Atom	Wyckoff Site	x	y	z	SOF	$B_{\text{eq}}$
Li	8a	0	0	0	1	13.38
Mn	16d	0.6250	0.6250	0.6250	0.75	19.53
Ni	16d	0.6250	0.6250	0.6250	0.25	1
O	32e	0.36880	0.36880	0.36880	1	-7.117



**Fig. S18.** XRD data (black), calculated Rietveld refinement (red), and difference pattern (blue) for an LNMO electrode after 10 cycles 3.4 - 5 V and 50 cycles 4.4 - 5 V charge/discharge.

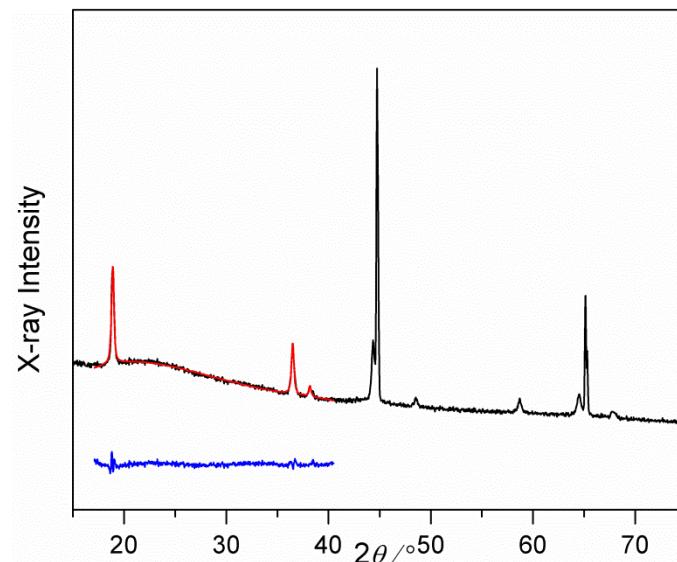
**Table S11.** Refinement parameters for an LNMO electrode after 10 cycles 3.4 - 5 V and 50 cycles 4.4 - 5 V charge/discharge.

Space Group	<i>Fd</i> $\bar{3}m$
Scale	$3.976 \times 10^{-4}$
Lattice parameter, <i>a</i> ( $\text{\AA}$ )	8.189
Unit Cell Mass ( $\text{g}\cdot\text{mol}^{-1}$ )	1450.260
Unit Cell Volume ( $\text{\AA}^3$ )	549.322
Crystallite Size, Lorentzian (nm)	83.3
Crystal Density ( $\text{g}\cdot\text{cm}^{-3}$ ), calculated	4.384
Crystal Linear Absorption Coefficient ( $\text{cm}^{-1}$ )	591.123
Wt%-Rietveld	100
$R_{\text{exp}}/R_{\text{exp}}'$ <sup>(a)</sup>	1.39/2.70
$R_{\text{wp}}/R_{\text{wp}}'$	2.25/4.36
$R_p/R_p'$	1.64/3.21
$R_{\text{Bragg}}$	0.529
$GOF$	1.62
$DW_d$	1.06

(a)-Primed parameters are background corrected.

**Table S12.** Atomic coordinates and isotropic thermal parameters for an LNMO electrode after 10 cycles 3.4 - 5 V and 50 cycles 4.4 - 5 V charge/discharge.

Atom	Wyckoff Site	<i>x</i>	<i>y</i>	<i>z</i>	SOF	<i>B</i> <sub>eq</sub>
Li	8 <i>a</i>	0	0	0	1	-9.146
Mn	16 <i>d</i>	0.6250	0.6250	0.6250	0.75	16.04
Ni	16 <i>d</i>	0.6250	0.6250	0.6250	0.25	19.59
O	32 <i>e</i>	0.37019	0.37019	0.37019	1	19.93



**Fig. S19.** XRD data (black), calculated Rietveld refinement (red), and difference pattern (blue) for an LNMO electrode after 1000 cycles with different electrochemical windows.

**Table S13.** Refinement parameters for an LNMO electrode after 1000 cycles with different electrochemical windows.

Space Group	$Fd\bar{3}m$
Scale	$1.281 \times 10^{-4}$
Lattice parameter, $a$ ( $\text{\AA}$ )	8.209
Unit Cell Mass ( $\text{g}\cdot\text{mol}^{-1}$ )	2347.295
Unit Cell Volume ( $\text{\AA}^3$ )	553.229
Crystallite Size, Lorentzian (nm)	92.5
Crystal Density ( $\text{g}\cdot\text{cm}^{-3}$ ), calculated	4.396
Crystal Linear Absorption Coefficient ( $\text{cm}^{-1}$ )	680.508
Wt%-Rietveld	100
$R_{\text{exp}}/R_{\text{exp}}'$ <sup>(a)</sup>	1.41/2.71
$R_{\text{wp}}/R_{\text{wp}}'$	1.98/3.80
$R_p/R_p'$	1.51/2.95
$R_{\text{Bragg}}$	1.158
$GOF$	1.40
$DW_d$	1.06

(a)-Primed parameters are background corrected.

**Table S14.** Atomic coordinates and isotropic thermal parameters for an LNMO electrode after 1000 cycles with different electrochemical windows.

Atom	Wyckoff Site	x	y	z	SOF	$B_{\text{eq}}$
Li	8a	0	0	0	1	19.92
Mn	16d	0.6250	0.6250	0.6250	0.75	19.68
Ni	16d	0.6250	0.6250	0.6250	0.25	12.18
O	32e	0.34704	0.34704	0.34704	1	-0.8975