Supporting Information For:

Cyclometalated Ruthenium(II) Complexes with a Bis-Carbene CCC-Pincer Ligand

You-Ming Zhang,^{†,§} Jiang-Yang Shao,[†] Chang-Jiang Yao,[†] and Yu-Wu Zhong^{*,†,‡}

[†]Beijing National Laboratory for Molecular Sciences, CAS Key Laboratory of Photochemistry, Institute of Chemistry, Chinese Academy of Sciences, Beijing 100190, People's Republic of China;

[‡]State Key Laboratory of Coordination Chemistry, Nanjing University, Nanjing 210093, People's Republic of China;

[§]College of Chemistry, Xiangtan University, Xiangtan 411105, People's Republic of China

***Correspondence:**

Prof. Yu-Wu Zhong CAS Key laboratory of photochemistry Institute of Chemistry, Chinese Academy of Sciences Beijing 100190, China Email: <u>zhongyuwu@iccas.ac.cn</u> Website: <u>http://zhongyuwu.iccas.ac.cn/</u>

Experimental Section

Spectroscopic Measurement. All optical ultraviolet-visible (UV/vis) absorption spectra were obtained using a TU-1810DSPC spectrometer of Beijing Purkinje General Instrument Co. Ltd. at room temperature in acetonitrile, with a conventional 1.0 cm quartz cell. Emission spectra were recorded using a F-380 spectrofluorimeter of Tianjin Gangdong Sci & Tech Development Co. Ltd., with a red-sensitive photomultiplier tube R928F.

Electrochemical Measurement. All cyclic voltammetry (CV) measurements were taken using a CHI620D potentiostat with one-compartment electrochemical cell under an atmosphere of nitrogen. All measurements were carried out in 0.1 M of Bu₄NClO₄/acetonitrile at a scan rate of 100 mV/s. The working electrode was a glassy carbon with a diameter of 0.3 mm. The electrode was polished prior to use with 0.05 µm alumina and rinsed thoroughly with water and acetone. A large area platinum wire coil was used as the counter electrode. All potentials are referenced to a Ag/AgCl electrode in saturated aqueous NaCl without regard for the liquid junction potential.

Computational Methods. DFT and TDDFT calculations are carried out using the B3LYP exchange correlation functional¹ and implemented in the *Gaussian* 03 program package.² The electronic structures of complexes were determined using a general basis set with the Los Alamos effective core potential LanL2DZ basis set for ruthenium, and 6-31G* for other atoms in vacuum.³

Synthesis of Complex 2. To 15 mL of dry acetone were added Ru(daatpy)Cl₃ (67 mg, 0.1 mmol) and AgOTf (82 mg, 0.3 mmol). The mixture was refluxed for 2 h before cooling to room temperature. After removing the white AgCl precipitate by filtering, the filtrate was concentrated to dryness. To the residue were added the carbene precursor 1 (48.8 mg, 0.1 mmol), *t*-BuOK (56 mg, 0.5 mmol), 10 mL DMF, and 10 mL *t*-BuOH. The resulting system was refluxed for 24 h. After cooling to room temperature, the solvent was removed under reduced pressure. The residue was dissolved in proper amount of methanol. After adding an excess of KPF₆, the resulting precipitate was collected by filtering and washing with water and Et₂O. The obtained solid was subjected to flash column chromatography on silica gel using acetonitrile, saturated aqueous KNO₃ and water (200/1/20) as the eluent. The desired complex 2 was

obtained as a reddish violet solid (44 mg, 43%). ¹HNMR (400 MHz, CD₃CN): δ 0.61 (m, 10H), 0.77 (m, 4H), 2.82 (t, *J* = 8 Hz, 4H), 3.87 (s, 6H), 6.79 (s, 2H), 6.83 (t, *J* = 6.4 Hz, 2H), 7.09 (d, *J* = 8.8 Hz, 4H), 7.14 (d, *J* = 4.8 Hz, 2H), 7.28 (t, *J* = 7.6 Hz, 1H), 7.37 (d, *J* = 8.8 Hz, 4H), 7.48 (d, *J* = 7.6 Hz, 2H), 7.52 (t, *J* = 7.8 Hz, 2H), 7.79 (s, 2H), 7.95 (s, 2H), 8.01 (d, *J* = 8.4 Hz, 2H). MALDI-MS: 833.1 for [M - PF₆]⁺. Anal. Calcd for C₄₁H₄₂F₆N₉PRu·2.5H₂O: C, 54.85; H, 5.07; N, 10.62. Found: C, 54.60; H, 4.63; N, 10.62.

Synthesis of Complex 3. According the similar synthetic procedure for **2**, complex **3** was prepared from Ru(Mebip)Cl₃ and **1** as a reddish violet solid in a yield of 22%. ¹HNMR (400 MHz, CD₃CN): δ 0.52 (m, 14H), 2.66 (s, 4H), 4.38 (s, 6H), 6.08 (d, *J* = 8.4 Hz, 2H), 6.61 (d, *J* = 2.0 Hz, 2H), 6.86 (t, *J* = 7.6 Hz, 2H), 7.28 (t, *J* = 8.0 Hz, 2H), 7.50 (t, *J* = 8.4 Hz, 1H), 7.52 (d, *J* = 8.4 Hz, 2H), 7.59 (d, *J* = 7.6 Hz, 2H), 7.73 (d, *J* = 1.6 Hz, 2H), 8.03 (t, *J* = 8.0 Hz, 1H), 8.69 (d, *J* = 8.0 Hz, 2H). MALDI-MS: 762.1 for [M - PF₆]⁺. Anal. Calcd for C₄₁H₄₂F₆N₉PRu·3H₂O: C, 51.25; H, 5.03; N, 13.12. Found: C, 51.64; H, 4.83; N, 13.26.

Synthesis of Complex 4. According the similar synthetic procedure for **2**, complex **4** was prepared from Ru(ttpy)Cl₃ and **1** as a reddish violet solid in a yield of 49%. ¹HNMR (400 MHz, CD₃CN): δ 0.52 (m, 14H), 2.49 (s, 3H), 2.65 (m, 4H), 6.75 (d, *J* = 2.0 Hz, 2H), 6.95 (m, 2H), 7.28 (d, *J* = 5.6 Hz, 2H), 7.36 (t, *J* = 8.0 Hz, 1H), 7.52 (m, 4H), 7.70 (m, 2H), 7.80 (d, *J* = 2.0 Hz, 2H), 8.02 (d, *J* = 8.0 Hz, 2H), 8.58 (d, *J* = 8.0 Hz, 2H), 8.93 (s, 2H). MALDI-MS: 746.1 for [M - PF₆]⁺. Anal. Calcd for C₄₂H₄₂F₆N₇PRu·2H₂O: C, 54.42; H, 5.00; N, 10.58 Found: C, 54.04; H, 4.75; N, 10.33.

Synthesis of Complex 5. According the similar synthetic procedure for **2**, complex **5** was prepared from Ru(Me₃tcbtpy)Cl₃ and **1** as a green solid in a yield of 21%. ¹HNMR (400 MHz, CD₃CN): δ 0.45 (m, 14H), 2.44 (m, 4H), 3.92 (s, 6H), 4.15 (s, 3H), 6.73 (d, *J* = 2.0 Hz, 2H), 7.46 (t, *J* = 8.0 Hz, 1H), 7.49 (d, *J* = 8.0 Hz, 2H), 7.57 (m, 4H), 7.80 (d, *J* = 1.6 Hz, 2H), 9.06 (s, 2H), 9.37 (s, 2H). MALDI-MS: 830.0 for [M - PF₆]⁺. Anal. Calcd for C₄₁H₄₂F₆N₇O₆PRu·2H₂O: C, 48.71; H, 4.59; N, 9.70. Found: C, 49.08; H, 4.42; N, 9.43.



Figure S1. Cyclic voltammograms of (a-d) **2-5** in acetonitrile containing 0.1 M n Bu₄NClO₄ as the supporting electrolyte at a scan rate of 100 mV/s. (e): A comparison of the Ru^{II/III} processes of four complexes. The working electrode was a glassy carbon and the counter electrode was a platinum wire. The reference electrode was Ag/AgCl in saturated aqueous NaCl solution.

Table S1. Electrochemical data of 2-5 and related comp	lexes.
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	Γ <i>a</i>	F a	.
complex	$E_{1/2}$.	$E_{1/2}$.	ΔE^{v}
	anodic	cathodic	(eV)
2	$+0.33, +1.07, +1.33, +1.52^{c}$	-1.64	1.97
3	$+0.42, +1.26^{\circ}$	-1.67	2.09
4	$+0.51, +1.28^{c}$	-1.61	2.12
5	$+0.75, +1.59^{c}$	-1.22, -1.58	1.97
[Ru(Mebip)(dpb)] ⁺	$+0.43, +1.49^{c}$	-1.56	1.92
$[Ru(tpy)(dpb)]^+$	$+0.56, +1.60^{\circ}$	-1.51	2.07

^{*a*}The potential is reported as the $E_{1/2}$ value vs Ag/AgCl. ^{*b*}The electrochemical energy gap is determined by the potential difference between the first oxidation and first reduction wave. ^{*c*} $E_{p,anodic}$, irreversible.



Figure S2. Isodensity plots of frontiers orbitals of complex 2. All orbitals have been computed at an isovalue of 0.02.



Figure S3. Isodensity plots of frontiers orbitals of complex 3. All orbitals have been computed at an isovalue of 0.02.



Figure S4. Isodensity plots of frontiers orbitals of complex 4. All orbitals have been computed at an isovalue of 0.02.



Figure S5. Isodensity plots of frontiers orbitals of complex 5. All orbitals have been computed at an isovalue of 0.02.



Figure S6. Frontier orbital energy level alignment of complexes 2-5.



Figure S7. (a) UV/vis spectra of 2-5 in acetonitrile. (b-e) TDDFT-predicted low-energy excitations of 2-5, respectively.

complex	Sn	E/eV	E/nm	f	λ_{max}/nm	е Е	dominant transitions
					observed	$(\times 10^5 M^{-1} cm^{-1})$	(percent contribution ^b)
2	1	2.04	606.8	0.0233		, , , , , , , , , , , , , , , , , , , ,	HOMO → LUMO (89%)
	4	2.23	554.9	0.0337			HOMO-2 \rightarrow LUMO (42%)
							HOMO \rightarrow LUMO+1 (51%)
	5	2.33	351.1	0.0278			HOMO-2 \rightarrow LUMO+1 (91%)
	6	2.52	492.7	0.160	520	0.139	HOMO-2 \rightarrow LUMO (53%)
							HOMO \rightarrow LUMO+1 (32%)
	7	2.88	430.1	0.0092			HOMO-3 → LUMO (94%)
3	4	2.14	578.1	0.0004			HOMO-2 → LUMO+1 (23%)
							HOMO-1 \rightarrow LUMO (76%)
	5	2.24	553.6	0.056			HOMO-2 \rightarrow LUMO (94%)
	6	2.71	457.1	0.1255	498	0.142	HOMO-2 \rightarrow LUMO+1 (55%)
	7	3.26	380.1	0.0078			HOMO-3 → LUMO+1 (97%)
	9	3.36	369.4	0.0225			HOMO-3 \rightarrow LUMO (93%)
4	4	2.26	548.0	0.0493			HOMO-1 → LUMO (67%)
	5	2.29	541.1	0.0184			HOMO-2 \rightarrow LUMO (60%)
							HOMO-1 \rightarrow LUMO+1 (36%)
	6	2.63	472.3	0.1732	502	0.143	HOMO-2 \rightarrow LUMO (36%)
							HOMO-1 \rightarrow LUMO+1 (42%)
	9	3.23	383.3	0.0413			HOMO-1 \rightarrow LUMO+2 (94%)
	10	3.26	380.7	0.0117			HOMO-1 \rightarrow LUMO+3 (92%)
5	2	2.01	616.6	0.0119			HOMO-2 \rightarrow LUMO (53%)
							HOMO-1 \rightarrow LUMO (35%)
	4	2.24	552	0.0174	625	0.067	HOMO-2 \rightarrow LUMO (20%)
							HOMO-2 \rightarrow LUMO+1 (29%)
							HOMO-1 \rightarrow LUMO (26%)
							HOMO-1 \rightarrow LUMO+1 (20%)
	5	2.33	530.8	0.0874	577	0.073	HOMO-2 → LUMO+1 (34%)
							HOMO-1 \rightarrow LUMO+1 (56%)
	6	2.64	469.6	0.0656	508	0.084	HOMO-2 → LUMO+1 (27%)
							HOMO-1 → LUMO (17%)
							HOMO-1 \rightarrow LUMO+1 (14%)
	9	2.96	418.1	0.0976	424	0.175	HOMO-1 \rightarrow LUMO+2 (65%)
	10	2.97	417.8	0.0545			HOMO-2 → LUMO+2 (56%)
	11	3.00	413.4	0.0868			HOMO-2 → LUMO+2 (20%)
							HOMO-2 → LUMO+3 (38%)
							HOMO-1 → LUMO+3 (32%)
	13	3.12	397.6	0.0222			HOMO-3 \rightarrow LUMO (91%)
	14	3.17	390.9	0.0555			HOMO-2 \rightarrow LUMO+3 (44%)
	16	3.42	362.9	0.0436			HOMO-3 \rightarrow LUMO+3 (31%) HOMO-3 \rightarrow LUMO+1 (86%)

Table S2. Observed absorption data and calculated excitation energy (E), oscillator strength (f), dominant contributing
transitions and the associated percent contribution and assignment of complexes 2-5. ^{<i>a</i>}

^{*a*}Computed at the TDDFT/B3LYP/LANL2DZ/6-31G*/vacuum level of theory. ^{*b*}The actual percent contribution = $(\text{configuration coefficient})^2 \times 2 \times 100\%$.

Cartesian coordinates of DFT-optimized structure of complex 2:

Charge = 1;	multiplicity=1	l	
Ru	-0.00282600	-0.06759100	0.02095000
С	-4.24250300	0.84992600	-0.73185200
С	-1.17566800	2.67773400	-0.27911700
С	1.17708800	2.68978300	0.11536000
С	1.20946300	4.08440900	0.06445800
С	0.00387300	4.76234600	-0.16474700
С	-1.20383100	4.07208300	-0.33949300
С	4.23763700	0.89480000	0.71771100
Н	2.12999800	4.64658000	0.19620100
Н	0.00555900	5.84708800	-0.20779000
Н	-2.12274000	4.62464900	-0.51565900
Н	0 87748400	0 50875900	-5 13984900
C	0.74539900	-0.21474600	-4.34228800
Ĥ	0 97312500	-1 96955300	-5 58823600
Ĉ	0 79773500	-1 58778000	-4 58736000
Č	0 51939200	0 22674400	-3 04380900
н	0.47204500	1 28229400	-2 80148900
C	0.62122100	-2 46348200	-3 52135100
N	0.34579300	-0.61132600	-2 00288600
н	0.65968200	-3 53468400	-3 68497400
C	0.39502900	-1 96527400	-2 23392900
C	0.19938800	-2 83118400	-1.05095800
н	0.35189900	-4 78420200	-1.05075000
C II	0.33107700	-4.22249600	-1.03870300
N N	-0.35072200	-0.47982100	2 07716700
н	-0.33072200	1 46108400	2.07710700
C II	-0.19689/00	-2 75637300	2.75154000
C	0.01283800	-4 91722300	0.180/2000
C C	0.52570000	0.42341300	3 06170000
C	-0.32370900	1 91610700	2 20526500
C	-0.39030700	-1.81010700	2.39330300
C	-0.18883300	-4.14343900	1.330/3100
C	-0./3030900	0.00049100	4.36016100
	-0.02188500	-2.23030400	5.71250200
П	-0.88339000	0.83980/00	3.13334100
П	-0.55014200	-4.04/00200	2.29855200
C	-0.80014000	-1.28/9/200	4./1951500
Н	-0.65884700	-3.28899500	5.94463500
H	-0.9/500800	-1.60451/00	5.74298500
N N	-0.00186600	-2.12/12600	0.0881/200
N N	-2.28416200	1./9/15300	-0.432/1200
N	-3.26332300	-0.12064400	-0.53436300
N	2.28323300	1.82080000	0.33708200
N	3.256/4100	-0.085/1900	0.59031300
C	3.62258500	2.09451900	0.55833000
C	2.03143900	0.461/8800	0.35212100
C	-2.03560000	0.44029200	-0.34381200
C	-3.62390100	2.05666000	-0.66822800
Н	4.02006300	3.09664100	0.58529100
Н	5.27220900	0.65365100	0.90869300
Н	-4.01913900	3.05473600	-0.77126300
H	-5.27847400	0.59763900	-0.89943300
C	-3.53844200	-1.54988600	-0.52354500
H	-4.03501000	-1.84595700	-1.45289100
Н	-2.59225200	-2.08120700	-0.43522200
Н	-4.18219600	-1.81059800	0.32308100
C	3.52562200	-1.51208900	0.69667400
Н	2.58098200	-2.04603300	0.60760600

Н	4.20256500	-1.83473900	-0.10116700
Н	3.98164700	-1.74088100	1.66513200
С	-0.00044500	1.96012000	-0.05288500
С	-0.50604900	-7.01393100	1.35799900
С	0.30823000	-7.88569500	2.08404700
С	-1.85393900	-6.87395300	1.72723700
С	-0.20067800	-8.61130000	3.16297700
Н	1.34852000	-8.00814400	1.79774200
С	-2.36439200	-7.57763000	2.80900500
Н	-2.50049400	-6.21205400	1.15786800
С	-1.54302100	-8.45669900	3.53610400
Н	0.45451300	-9.28496100	3.70264600
Н	-3.40484200	-7.48062900	3.10210800
С	0.56626300	-7.08155700	-0.85120800
С	-0.22675300	-8.02464100	-1.50836700
С	1.91011200	-6.93699400	-1.23346500
С	0.29886300	-8.81584300	-2.53179500
Н	-1.26336600	-8.15143100	-1.21086300
С	2.43666900	-7.70632300	-2.26138600
Н	2.54121600	-6.21940400	-0.71668400
С	1.63654500	-8.65658000	-2.91942200
Н	-0.34023300	-9.54357600	-3.01788800
Н	3.47421700	-7.60588700	-2.56354900
Ν	0.02210000	-6.30263900	0.22661800
0	2.24986500	-9.36218700	-3.90343700
0	-2.14175800	-9.10250800	4.56885100
С	1.50478700	-10.35782900	-4.59442200
Н	0.64809400	-9.92174800	-5.12397900
Н	2.19329800	-10.79472800	-5.31882800
Н	1.15140700	-11.13972000	-3.91049900
С	-1.37548200	-10.02805200	5.33058200
Н	-0.53143800	-9.53483400	5.82948100
Н	-2.05600700	-10.42871400	6.08293400
Н	-1.00079000	-10.84798300	4.70501800

Cartesian coordinates of DFT-optimized structure of complex 3:

Charge = 1;	multiplicity=	1	
Ru	0.00452000	-0.11239100	0.02131400
С	-4.29540100	0.81891200	0.03763800
С	-1.19665000	2.64044300	-0.06207600
С	1.18775400	2.64724400	-0.08813000
С	1.21368300	4.04223500	-0.13574100
С	-0.01111000	4.72382200	-0.14555000
С	-1.23149700	4.03525500	-0.10908100
Č	4 29839900	0 84374000	-0.05769900
н	2 14516000	4 60078500	-0 16447800
Н	-0.01460700	5 80872500	-0 18196100
н	-2 16650800	4 58856000	-0 11748700
C C	0.00202500	2 80883300	1 07/20100
с ц	0.00292500	-2.89883300	1 05828400
II C	0.00709300	4.87900100	-1.93838400
C	0.01283300	-4.29800700	-1.04020300
C	0.03082900	-2.81/43000	1.30/43000
C	0.03144900	-4.93098000	0.18/39000
C	0.04026/00	-4.21590300	1.3/419300
H	0.05496100	-4./3361/00	2.32349400
N	0.01314/00	-2.18501100	0.0936/300
N	-2.315/3000	1.76497000	-0.01890800
N	-3.29639200	-0.15192/00	0.06180100
N	2.31249200	1.77820700	-0.069/2400
N	3.3058/100	-0.1329/500	-0.01265200
C	3.67086400	2.04650200	-0.09361400
С	2.06155900	0.42252600	-0.01899300
C	-2.05584800	0.41081000	0.02713300
C	-3.67585000	2.02530300	-0.01291900
Н	4.07072200	3.04715200	-0.13308300
Н	5.34908700	0.59700400	-0.06008300
Н	-4.08226800	3.02361200	-0.04363200
Н	-5.34441900	0.56605200	0.05877700
С	-3.56927600	-1.58015200	0.11333000
Н	-4.12696100	-1.89492700	-0.77475900
Н	-2.61935100	-2.11068300	0.14973700
Н	-4.15493400	-1.82321300	1.00568500
С	3.58876200	-1.55941800	0.03431200
Н	2.64266200	-2.09705300	0.06407500
Н	4.15215100	-1.86560600	-0.85306900
Н	4.17269300	-1.80282000	0.92781200
С	-0.00210000	1.91962000	-0.05053700
Н	0.03934300	-6.03532500	0.22461200
С	0.03780000	-1.84661200	2.39455800
С	0.03758500	0.14465100	3.29970200
С	0.05182300	-0.79545300	4.35452200
С	0.03387200	1.52227600	3.56180800
С	0.06255900	-0.39451200	5.69525800
С	0.04447100	1.91894600	4.89130300
Н	0.02315400	2.23516700	2.74444000
С	0.05854100	0.97361900	5.94111700
Н	0.07355000	-1.10557400	6.51503300
Н	0.04197000	2.97694700	5.13462800
Н	0.06652600	1.32398500	6.96878400
С	-0.01589500	-2.00512500	-2.22548900
С	-0.03229400	-0.08039800	-3.26480700
С	-0.04199000	-1.09050400	-4.25287500
С	-0.03831900	1.27587800	-3.62066800
С	-0.05760300	-0.78196400	-5.61804900

С	-0.05372400	1.58091800	-4.97403800
Н	-0.03074800	2.04337100	-2.85475200
С	-0.06320800	0.56611700	-5.95677200
Н	-0.06429500	-1.54728000	-6.38748300
Н	-0.05845400	2.61989800	-5.28881300
Н	-0.07489300	0.84551200	-7.00593700
Ν	0.02900500	-0.53500800	2.09877100
Ν	0.05176600	-2.05077000	3.75895800
Ν	-0.01670100	-0.67633700	-2.02014500
Ν	-0.03224400	-2.30203900	-3.57277100
С	0.06297900	-3.31600300	4.47918000
Н	-0.83043900	-3.90587700	4.25252000
Н	0.95909600	-3.89711200	4.24070000
Н	0.06900500	-3.10500700	5.54826000
С	-0.04254800	-3.61357800	-4.20465600
Н	0.86211900	-4.17767600	-3.95749100
Н	-0.92672500	-4.18583100	-3.90783900
Н	-0.07378600	-3.47620200	-5.28524500

Cartesian coordinates of DFT-optimized structure of complex **4**:

Charge = 1;	multiplicity=1	1	
Ru	0.00510100	-0.02744700	0.01944200
С	-4.30015700	0.89971500	0.02435400
С	-1.20077300	2.72425600	-0.07785000
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Ċ	-1 23861600	4 11861200	-0 13167600
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Ν	-0.01382200	-0.55867100	-2.03334900
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Н	0.06041800	-3.22302900	4.01134700
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Ν	0.01341000	-2.07120300	0.09029000
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Ν	-3.30217000	-0.07133900	0.05454800
Ν	2.31099500	1.86658900	-0.07061800
Ν	3.31196200	-0.04110200	-0.00032100
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Н	5.35212100	0.69650900	-0.03812300
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Cartesian coordinates of DFT-optimized structure of complex 5:

Charge = 1;	multiplicity=	1	
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C II	-3.6//99500	2.09816200	-0.00/42/00
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С	0.02121600	-6.32826800	0.26322700
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С	-0.07319200	-1.62143000	-8.35864100
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Н	0.08788900	-4.47610700	7.73666000
Н	0.98331000	-3.01951300	8.28907700

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