Electronic Supplementary Information for:

Tuning stoichiometry and supramolecular assembly in perfluorinated indazolato coinage metal complexes

Michael Kleinwächter,^{*a,b*} Laure Vendier,^{*a,b*} Chiara Dinoi,^{*a,b*} and Michel Etienne^{*a,b*}

^a CNRS, LCC (Laboratoire de Chimie de Coordination), BP 44099, 205 route de Narbonne, F-31077 Toulouse, France. E-mail: chiara.dinoi@lcc-toulouse.fr, michel.etienne@lcc-toulouse.fr ^b Université de Toulouse, UPS, INPT, LCC, F-31077 Toulouse, France.

- 1. Experimental details
- 2. Synthesis of {[3-(C₂F₅)IndF₄]Ag}₃ [1]
- 3. Synthesis of [(toluene)(1)(toluene)]
- 4. Synthesis of {Cu₅[3-(C₂F₅)IndF₄]₆}Et₃NH [2]
- 5. Crystal data and structure refinement for [1] (Table S1)
- 6. Crystal data and structure refinement for [(toluene)(1)(toluene)] (Table S2)
- 7. Crystal data and structure refinement for [2] (Table S3)
- 8.1 ORTEP view of [Cu(CH₃CN)₄]BF₄
- 8.2 Crystal data and structure refinement for [Cu(CH₃CN)₄]BF₄ (Table S4)

1. Experimental details

All reactions requiring inert atmosphere were carried out using Schlenk techniques or glovebox. The solvents were dried and distilled using the conventional methods, diethyl ether and tetrahydrofuran (Na/benzophenone), toluene (Na), dichloromethane, pentane and dimethoxyethane (CaH₂). The 3-pentafluoroethyl-4,5,6,7-tetrafluoro-1-H-indazole¹ and $[Cu(MeCN)_4][BF_4]^{2,3}$ were synthesized according to literature procedures. All other chemicals were used as purchased.

The NMR experiments were acquired at 298 K using a DPX300 Bruker and an AVANCE Bruker 400 MHz spectrometers. Elemental analysis and X-ray diffraction were carried out by the analytical service of our laboratory.

2. Synthesis of {[3-(C₂F₅)IndF₄]Ag}₃ [1]:

A suspension of Ag_2O (0.058 g, 0.25 mmol) and 3-pentafluoroethyl-4,5,6,7-tetrafluoro-1-Hindazole (0.152 g, 0.49 mmol) in dry toluene (10 ml) was heated to reflux for 4.5 h in a Schlenk flask covered with aluminum foil. The not yet cold solution was filtered through a bed of Celite to remove insoluble material. The filtrate was collected, and the solvent was removed under reduced pressure. The resulting solid was washed with pentane (2x5 ml) to yield {[3-(C₂F₅)IndF₄]Ag}₃ [1]-toluene as a white solid (0.200 g, 0.15 mmol, 90 % yield). X-ray quality crystals were obtained by slow evaporation of an ether solution.

¹⁹**F-NMR** (CD₂Cl₂, 282 MHz, 298 K) δ [ppm]: -84.7 (s, 3F, -CF₃), -112.2 (d, 2F, $J_{FF} = 24.1$ Hz, CF₂CF₃), -143.0 (m, 1F, F(4)), -157.1 (t, 1F, $J_{FF} = 17.5$ Hz, F(6)), -159.5 (m, 1F, F(7)), -162.8 (t, 1F, $J_{FF} = 17.9$ Hz, F(5)).

Anal. Calcd. for $\{[3-(C_2F_5)IndF_4]Ag\}_3$ toluene, $C_{34}H_8Ag_3F_{27}N_6$: C, 30.54; H, 0.60; N, 6.29. Found: C, 30.12, H, 0.83; N, 6.62.

3. Synthesis of [(toluene)(1)(toluene)]

[1]-toluene was dissolved in toluene. X-ray quality crystals of $[(toluene)(\{[3-(C_2F_5)IndF_4]Ag\}_3)(toluene)]$ were obtained by slow evaporation of the toluene solution in the Glove box.

4. Synthesis of {Cu₅[3-(C₂F₅)IndF₄]₆}Et₃NH [2]

 Et_3N (0.117 g, 1.15 mmol, 0.16 ml) was added to a suspension of $[Cu(MeCN)_4][BF_4]$ (0.2 g, 0.64 mmol) and 3-pentafluoroethyl-4,5,6,7-tetrafluoro-1-H-indazole (0.237 g, 0.77 mmol) in toluene (20 ml). The suspension turned yellow immediately. After stirring for 3 h at r.t., the

solvent was removed under vacuum. The resulting solid was then extracted with dry Et_2O (3x10 ml) and the extract filtered through a bed of Celite. After evaporation of the solvent, the solid was washed with dry pentane (2x5 ml) to yield { $Cu_5[3-(C_2F_5)IndF_4]_6$ }Et₃NH as an off-white solid (0.200 g, 0.09 mmol, 70% yield).

X-ray crystals were obtained by slow evaporation of an ether solution.

¹⁹**F-NMR** (acetonitrile-*d*₃, 376 MHz, 298 K) δ [ppm]: -84.1 (s, 3F, C*F*₃), -108.1 (d, 2F, J_{FF} = 20.2 Hz, C*F*₂CF₃), -146.7 (s, 1F, F(4)), -159.5 (s, 1F, F(6)), -165.5 (t, 1F, J_{FF} = 17.0 Hz, F(7)), -169.7 (t, 1F, J_{FF} = 17.0 Hz, F(5)).

Anal. Calcd. for C₆₀H₁₆Cu₅F₅₄N₁₃: C, 31.85; H, 0.71; N 8.05. Found: C, 32.02; H, 0.71; N, 8.19.

Empirical formula	Car Aga Far Na Ca Has O
Empirical formula	$C_2/Rg_3 \Gamma_2/R_6, C_4 \Pi_{10} O$
Formula weight	1519.00
Temperature	180(2) K
Wavelength	0.71073 A
Crystal system, space group	Triclinic, P -1
Unit cell dimensions	a = 12.6128(9) A alpha = 78.199(3) deg.
	b = 12.9362(5) A beta = 73.173(3) deg.
	c = 13.6226(7) A gamma = 63.132(4) deg.
Volume	1890.77(18) A^3
Z, Calculated density	2, 2.317 Mg/m^3
Absorption coefficient	1.712 mm^-1
F(000)	1260
Crystal size	0.2 x 0.12 x 0.03 mm
Theta range for data collection	3.38 to 26.37 deg.
Limiting indices	-15<=h<=15, -16<=k<=16, -17<=l<=17
Reflections collected / unique	31328 / 7710 [R(int) = 0.0432]
Completeness to theta $= 26.37$	99.7 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.8779 and 0.6954
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	7710 / 0 / 615
Goodness-of-fit on F ²	0.999
Final R indices [I>2sigma(I)]	R1 = 0.0310, wR2 = 0.0621
R indices (all data)	R1 = 0.0497, wR2 = 0.0687
Largest diff. peak and hole	0.803 and -0.500 e.A^-3

5. Crystal data and structure refinement for 1 (Table S1)

6. Crystal data and structure refinement for [(toluene)(1)(toluene)] (Table S2)

Empirical formula	C27 Ag3 F27 N6, 2(C7 H8)
Formula weight	1429.21
Temperature	180(2) K
Wavelength	0.71073 A
Crystal system, space group	Triclinic, P -1
Unit cell dimensions	a = 12.5640(5) A alpha = 74.837(3) deg.
	b = 12.8440(4) A beta = 75.239(4) deg.

	c = 16.4240(7) A gamma = $60.808(4) deg.$
Volume	2207.60(15) A^3
Z, Calculated density	2, 2.150 Mg/m^3
Absorption coefficient	1.474 mm^-1
F(000)	1376
Crystal size	0.15 x 0.12 x 0.1 mm
Theta range for data collection	3.36 to 26.37 deg.
Limiting indices	-15<=h<=15, -16<=k<=16, -20<=l<=20
Reflections collected / unique	26584 / 8997 [R(int) = 0.0549]
Completeness to theta $= 26.37$	99.8 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.987 and 0.895
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	8997 / 6 / 696
Goodness-of-fit on F ²	1.096
Final R indices [I>2sigma(I)]	R1 = 0.0685, wR2 = 0.1474
R indices (all data)	R1 = 0.1017, wR2 = 0.1608
Largest diff. peak and hole	2.126 and -0.886 e.A^-3

7. Crystal data and structure refinement for 2 (Table S3)

Empirical formula	$C_{54} Cu_5 F_{54} N_{12}, C_6 H_{16} N, C_4 H_{10} O$
Formula weight	2336.68
Temperature	180(2) K
Wavelength	0.71073 A
Crystal system, space group	Orthorhombic, P n a 21
Unit cell dimensions	a = 22.0065(12) A alpha = 90 deg.
	b = 13.1308(9) A beta = 90 deg.
	c = 27.0131(19) A gamma = 90 deg.
Volume	7805.8(9) A^3
Z, Calculated density	4, 1.988 Mg/m^3
Absorption coefficient	1.523 mm^-1
F(000)	4560
Crystal size	0.15 x 0.14 x 0.11 mm
Theta range for data collection	1.72 to 26.37 deg.
Limiting indices	-25<=h<=27, -16<=k<=16, -33<=l<=33
Reflections collected / unique	83406 / 15886 [R(int) = 0.0419]
Completeness to theta $= 26.37$	99.9 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.991 and 0.897
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	15886 / 30 / 1159
Goodness-of-fit on F ²	1.017
Final R indices [I>2sigma(I)]	R1 = 0.0479, $wR2 = 0.1226$
R indices (all data)	R1 = 0.0723, wR2 = 0.1390
Largest diff. peak and hole	1.198 and -0.667 e.A^-3

8.1. ORTEP view of [Cu(CH₃CN)₄]BF₄



8.2. Crystal data and structure refinement for [Cu(CH₃CN)₄]BF₄ (Table S4)

Empirical formula	C ₈ H ₁₂ Cu N ₄ , BF ₄
Formula weight	314.58
Temperature	100(2) K
Wavelength	0.71073 A
Crystal system, space group	Orthorhombic, P n a 21
Unit cell dimensions	a = 23.705(5) A alpha = 90 deg.
	b = 8.276(5) A beta = 90 deg.
	c = 20.188(5) A gamma = 90 deg.
Volume	3961(3) A^3
Z, Calculated density	12, 1.583 Mg/m^3
Absorption coefficient	1.688 mm^-1
F(000)	1896
Crystal size	0.2 x 0.15 x 0.08 mm
Theta range for data collection	1.99 to 36.37 deg.
Limiting indices	-37<=h<=39, -13<=k<=13, -32<=l<=25
Reflections collected / unique	95171 / 15924 [R(int) = 0.0213]
Completeness to theta $= 25.00$	99.8 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.985 and 0.876
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	15924 / 1 / 498
Goodness-of-fit on F ²	1.029
Final R indices [I>2sigma(I)]	R1 = 0.0238, wR2 = 0.0560
R indices (all data)	R1 = 0.0293, wR2 = 0.0575
Largest diff. peak and hole	0.472 and -0.306 e.A^-3

References:

- A. Caballero, E. Despagnet-Ayoub, M. M. Díaz-Requejo, A. Díaz-Rodríguez, M. E. González-Núñez, R. Mello, B. K. Muñoz, W.-S. Ojo, G. Asensio, and M. Etienne, *Science*, 2011, 332, 835.
- 2 G. J. Kubas, Inorg. Synth., 1990, 28, 68.
- 3 X. Tang, S. Woodward, and N. Krause, Eur. J. Inorg. Chem., 2009, 2836.