<Electronic Supplementary Information>

Halogen effects on photoluminescence and catalytic properties: a series of spatially arranged trimetallic zinc(II) complexes

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Table S1 Weak Inter- and Intra-molecular Interactions (Å) for the Compounds

	$[Zn_3Cl_6L(MeOH)_3]$	$[Zn_3Br_6L(MeOH)_3]$	$[Zn_3I_6L(MeOH)_3]$
π···π	3.6839(5)	3.83(4)	3.954(4)
	3.6940(5)	3.845(4)	4.011(4)
	3.7041(5)	3.86(4)	4.011(4)
Intermolecular interaction NH···O=C	2.206	2.230	2.358
	2.214	2.318	2.494
	2.245	2.385	2.693
Intramolecular interaction MeOH···O=C	1.776	1.809	1.735
	1.785	1.854	1.908

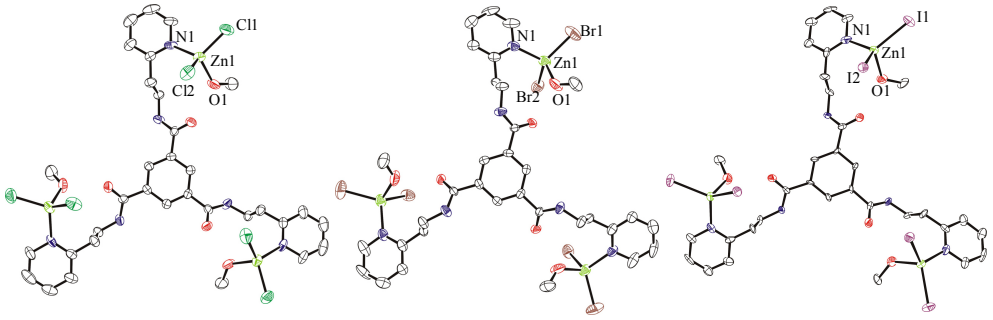

Fig. S1 ORTEP drawings of $[Zn_3Cl_6L(MeOH)_3]$ (left), $[Zn_3Br_6L(MeOH)_3]$ (middle), and $[Zn_3I_6L(MeOH)_3]$ (right).

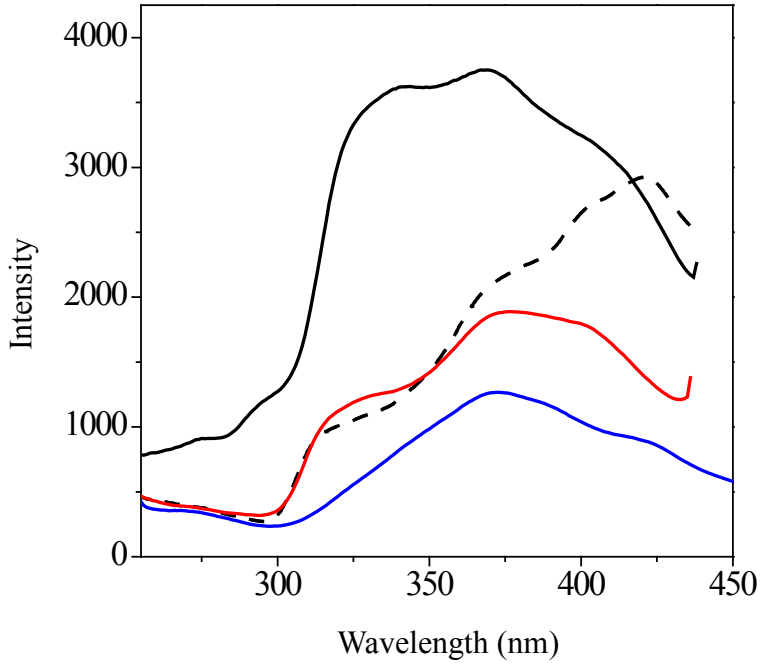

Fig. S2 Solid-state excitation spectra of L (dashed line), $[Zn_3Cl_6L(MeOH)_3]$ (solid black line), $[Zn_3Br_6L(MeOH)_3]$ (solid red line), and $[Zn_3I_6L(MeOH)_3]$ (solid blue line) at room temperature.

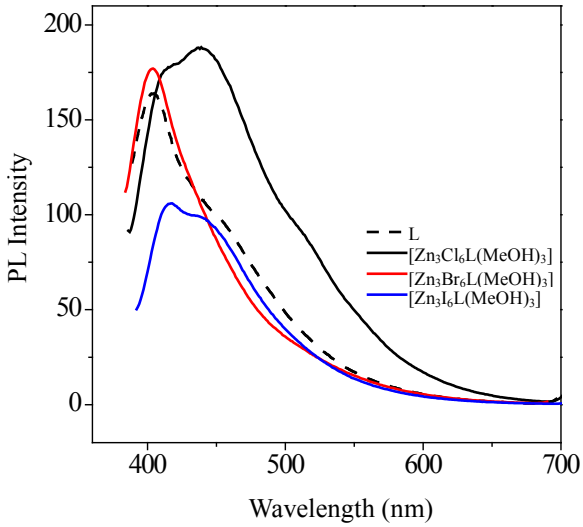

Fig. S3 Photoluminescence spectra of L (dashed line), $[Zn_3Cl_6L(MeOH)_3]$ (solid black line), $[Zn_3Br_6L(MeOH)_3]$ (solid red line), and $[Zn_3I_6L(MeOH)_3]$ (solid blue line) at room temperature in Me_2SO.

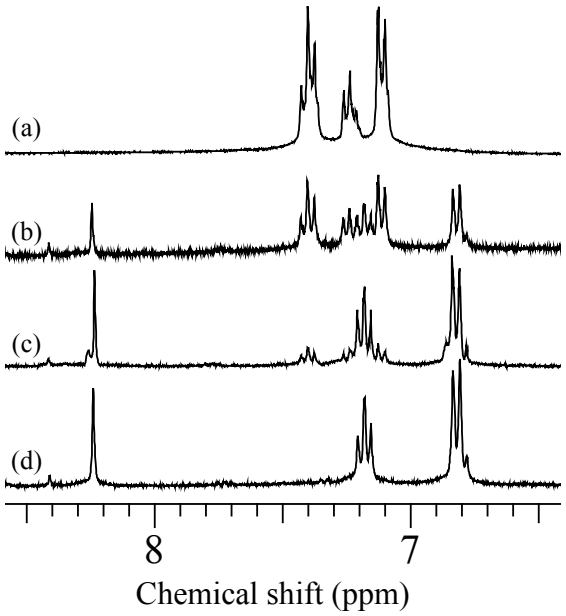

Fig. S4 Partial ^{1}H NMR spectra (Me_2CO-d_6) showing the catalytic effects for transesterification of phenyl acetate using $[Zn_3I_6L(MeOH)_3]$ catalyst in a mixture of MeOH and CH_3CN (*v*/*v*=1:1) at 50 ºC for 2 h (b), 4 h (c), and 6 h (d). (a) shows the resonance of the authentic phenyl acetate.

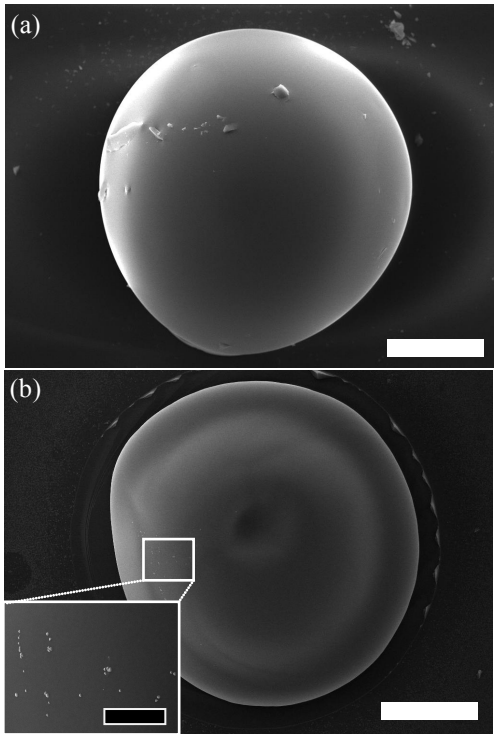

Fig. S5 SEM images of calcined $[Zn_3I_6L(MeOH)_3]$ at 200 °C (a) and 400 °C (b). White bar = 200 μm, black bar = 20 μm.