

Supporting Information

**Catalysis and molecular magnetism of dinuclear iron(III) complexes with
N-(2-pyridylmethyl)-iminodiethanol/-ate**

Jong Won Shin, Jeong Mi Bae, Cheal Kim, and Kil Sik Min

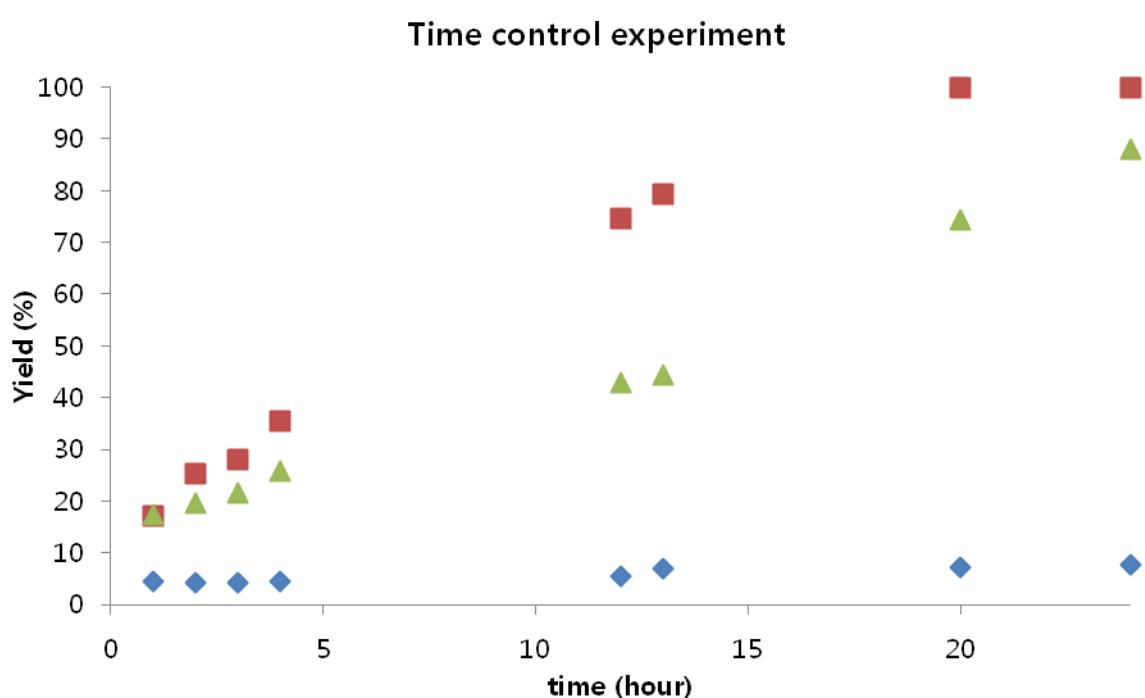


Fig. S1. Time-course for the **1**-catalyzed oxidation of cyclohexanol with *t*-BuOOH in CH₃CN. Reaction conditions: catalyst; 1.2×10^{-3} mmol, cyclohexanol; 0.30 mmol, *t*-BuOOH; 0.03 mmol, dodecane; 0.01 mmol, solvent; CH₃CN (1 mL).

◆ without catalyst, ▲ $[(pmide)Fe(N_3)_2]$, ■ $[(Hpmide)Fe(NO_3)_2(NO_3)_2]$

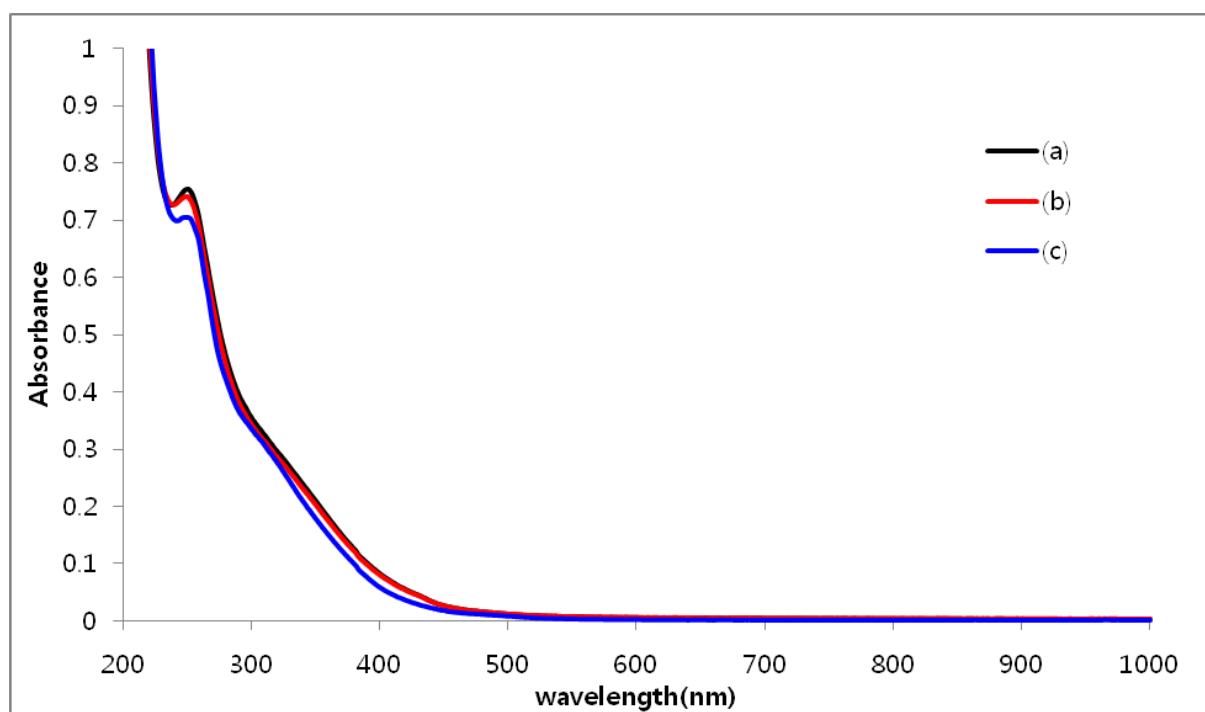


Fig. S2. UV-vis spectra of (a) **1** (1.2×10^{-3} mmol) with 250 equiv of cyclohexanol (black line), (b) immediately after the addition of *t*-BuOOH (25 equiv) (red line), and (c) 24 h after the addition of *t*-BuOOH (25 equiv) (blue line) in CH₃CN at room temperature. All UV-vis spectra were taken after 30 times dilution.

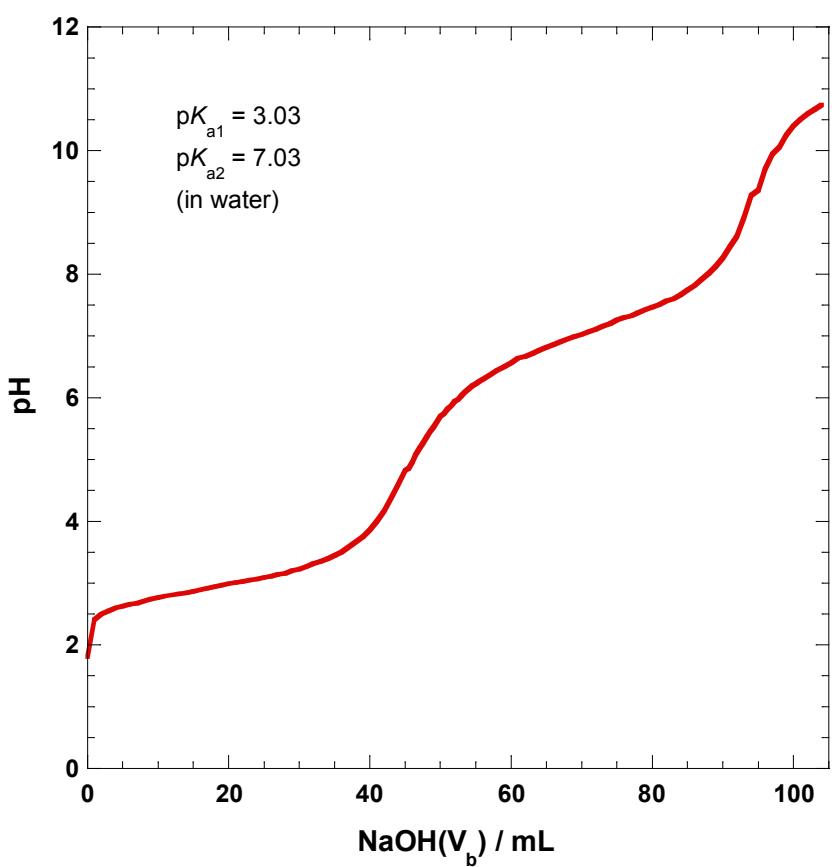


Fig. S3. Titration curve for the reaction of 0.025 M $[(\text{Hpmide})\text{Fe}(\text{NO}_3)]_2(\text{NO}_3)_2 \cdot 2\text{CH}_3\text{OH}$ (10 mL) and 0.01 M NaOH.