

Electronic Supporting Information

Unprecedented Silicon(II)→Calcium Complexes with N-Heterocyclic Silylenes

Burgert Blom,^{a*} Günter Klatt,^b Daniel Gallego,^a Gengwen Tan,^a and Matthias Driess^{a*}

^aDepartment of Chemistry: Metalorganics and Inorganic Materials, Sekr. C2
Technische Universität Berlin Strasse des 17. Juni 135, 10623 Berlin (Germany)
Fax: (+49)30-314-29732; and ^bPhysikalisch-Chemisches Institut der Universität
Heidelberg, Im Neuenheimer Feld 229, D-69120 Heidelberg (Germany). E-mail:
burgert.blom@tu-berlin.de; matthias.driess@tu-berlin.de

Electronic Supplementary Information

Contents:

1. Selected NMR and ESI-MS spectra	S2
2. Crystal data and refinement of compounds 3 , 6 , 7 and 10 .	S9
3. Computational Details (DFT) for compounds 6 and 7 .	S35
4. Z-Matrices and optimized structures of 6 and 7	S36
5. Bader Atoms in Molecules (AIM) Analysis	S41
6. References	S44

1. Selected NMR and ESI-MS spectra.

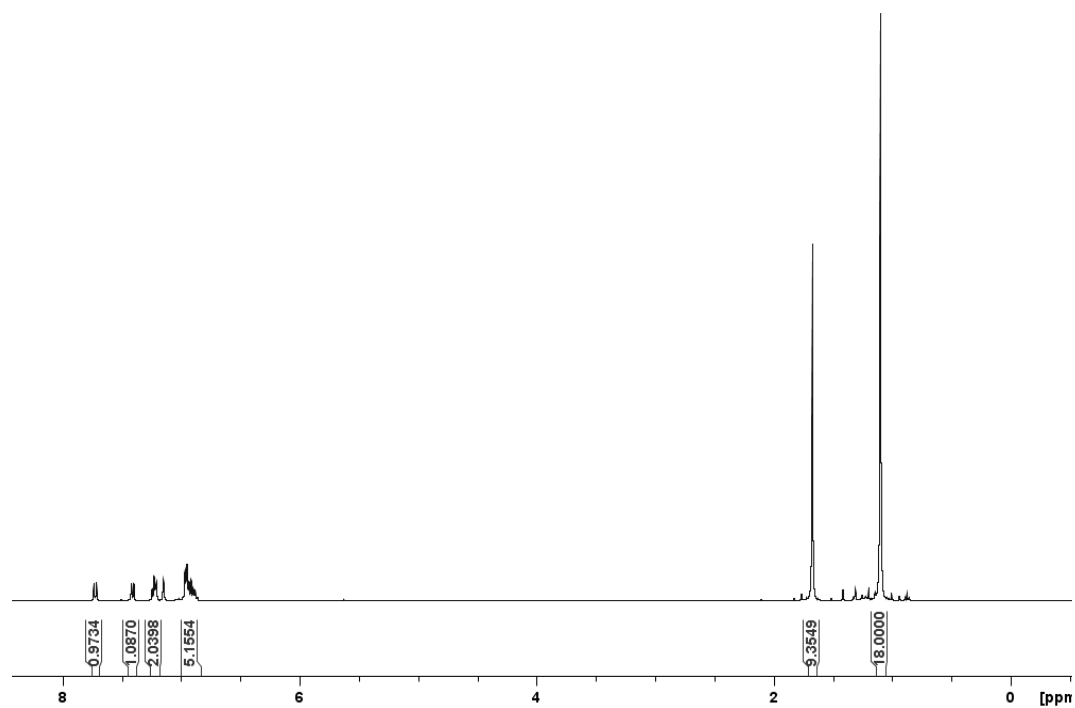


Fig. S1 ^1H NMR spectrum of the free NHSi **5** in C_6D_6 at 298 K.

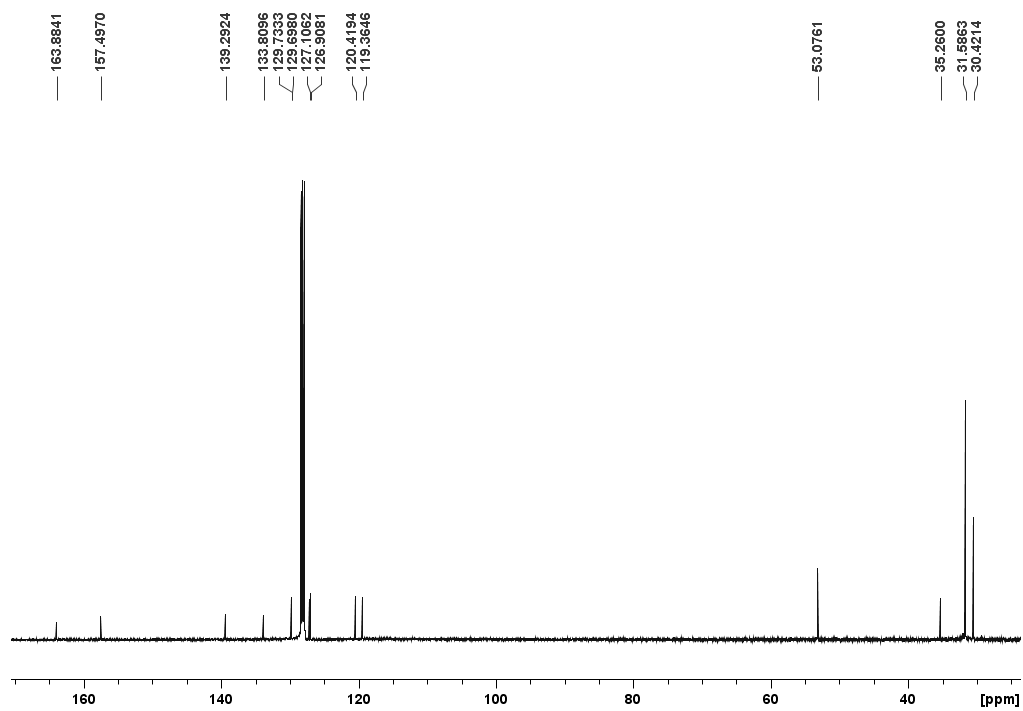


Fig. S2 $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of the free NHSi **5** in C_6D_6 at 298 K.

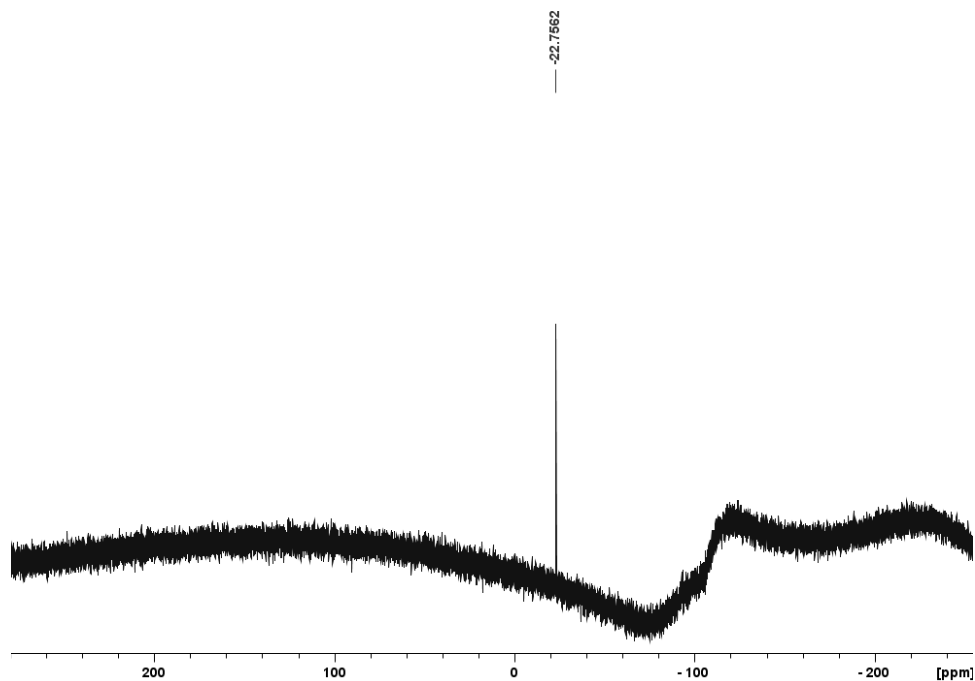


Fig. S3 $^{29}\text{Si}\{^1\text{H}\}$ NMR spectrum of the free NHSi **5** in C_6D_6 at 298 K.

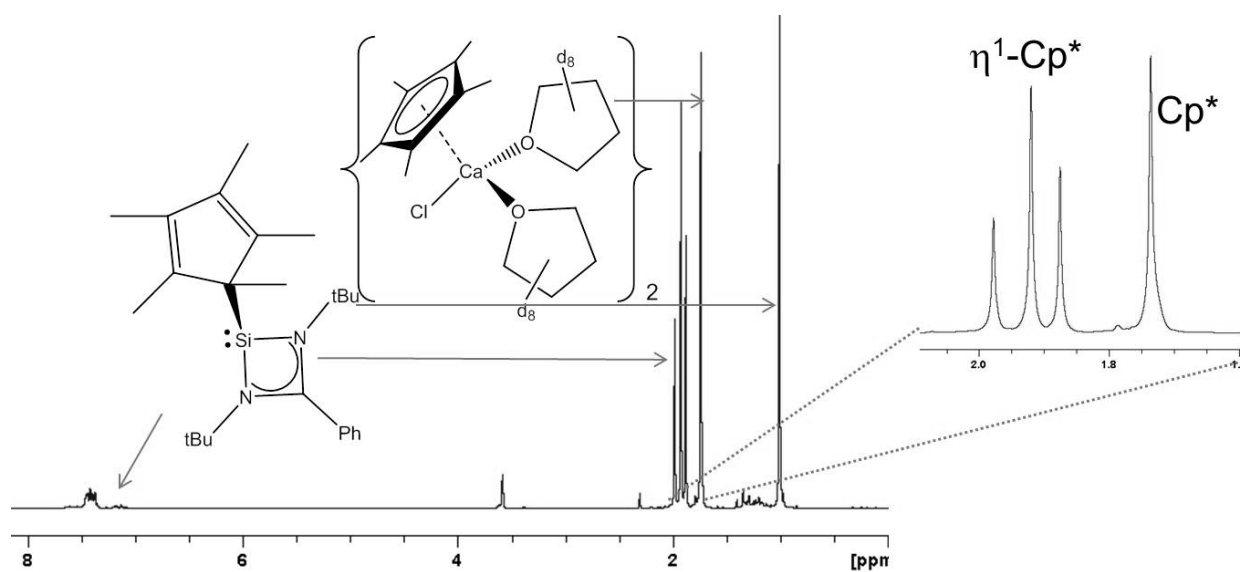


Fig. S4 ^1H NMR spectrum of the mixture of $[(\eta^5\text{-C}_5\text{Me}_5)\text{CaCl}(\text{thf-}d_8)_2]_2$ and $:\text{Si}(\text{Cp}^*)(\text{PhC}(\text{N}^t\text{Bu})_2)$ in $\text{THF-}d_8$ resulting from the reaction of compound **2** with chlorosilylene **1**.

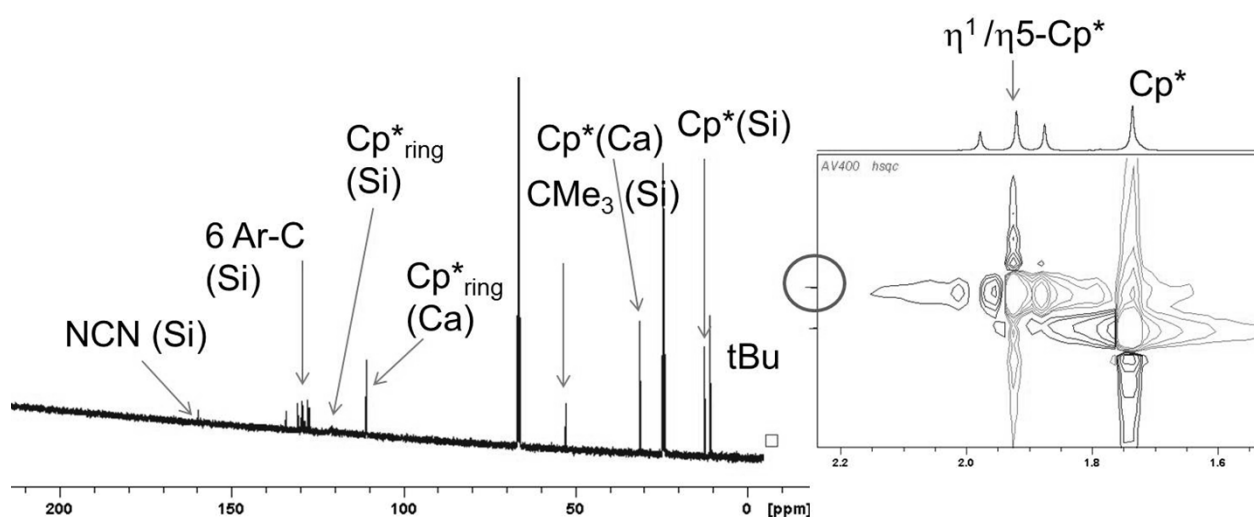


Fig. S5 $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of the mixture of $[(\eta^5\text{-C}_5\text{Me}_5)\text{CaCl}(\text{thf-d}_8)_2]_2$ and $:\text{Si}(\text{Cp}^*)(\text{PhC}(\text{N}^t\text{Bu})_2)$ in THF-d_8 resulting from the reaction of compound **2** with chlorosilylene **1**. The expansion on the right shows the HSQC spectrum where the Cp^* is coordinated to Si in a η^5 fashion due to only one cross peak for the CH_3 groups.

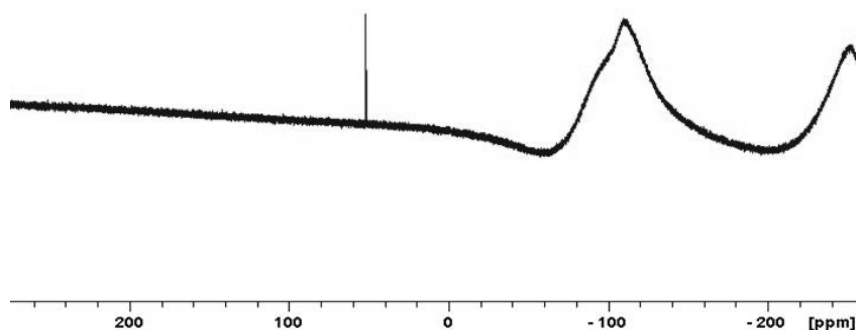


Fig. S6 $^{29}\text{Si}\{^1\text{H}\}$ NMR spectrum of $:\text{Si}(\text{Cp}^*)(\text{PhC}(\text{N}^t\text{Bu})_2)$ in THF-d_8 at 298 K. The resonance signal is at $\delta = 52.4$ ppm.

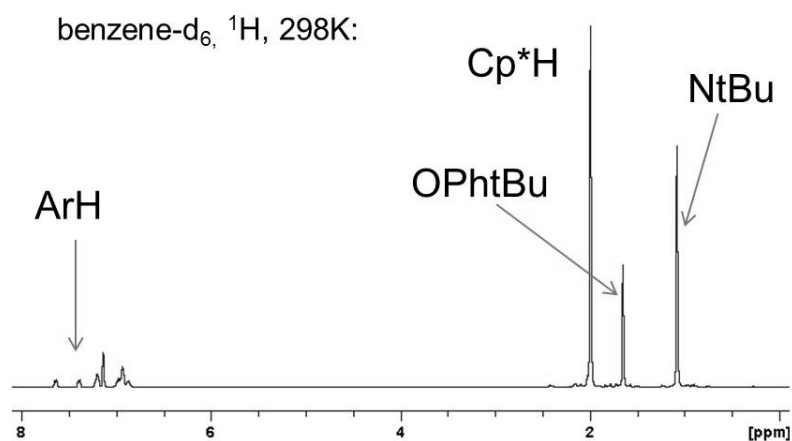


Fig. S7 ¹H NMR spectrum of $[(\eta^5\text{-C}_5\text{Me}_5)_2\text{Ca} \leftarrow \text{Si}(\text{O-C}_6\text{H}_4\text{-2-}^t\text{Bu})\{(\text{N}^t\text{Bu})_2\text{CPh}\}]$ (**6**) in C₆D₆ at 298 K.

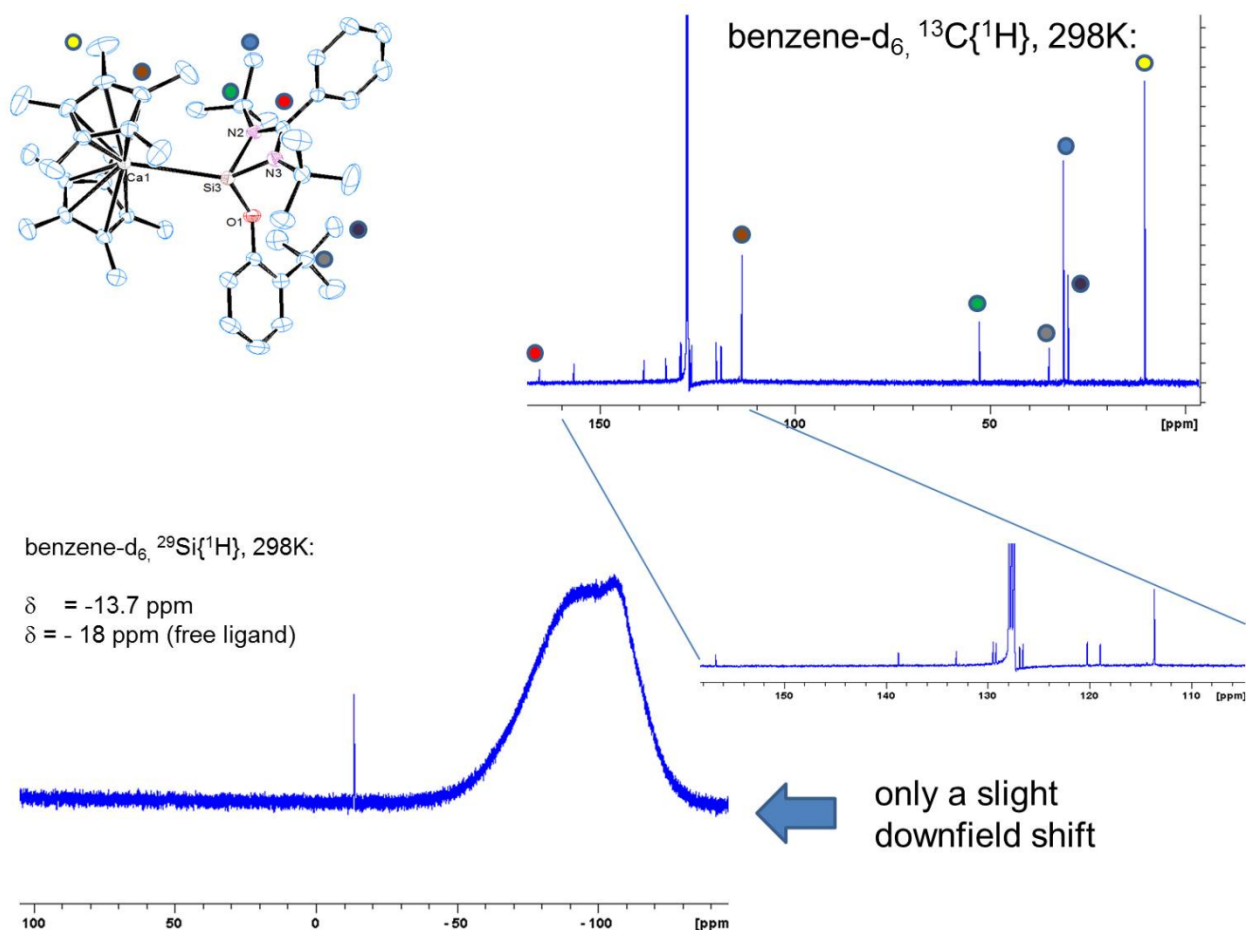


Fig. S8 ¹³C{¹H} (with partial assignment shown) and ²⁹Si NMR spectrum of $[(\eta^5\text{-C}_5\text{Me}_5)_2\text{Ca} \leftarrow \text{Si}(\text{O-C}_6\text{H}_4\text{-2-}^t\text{Bu})\{(\text{N}^t\text{Bu})_2\text{CPh}\}]$ (**6**) in C₆D₆ at 298 K. The resonance signal of the Si is only shifted by 4.3 ppm downfield relative to the free ligand.

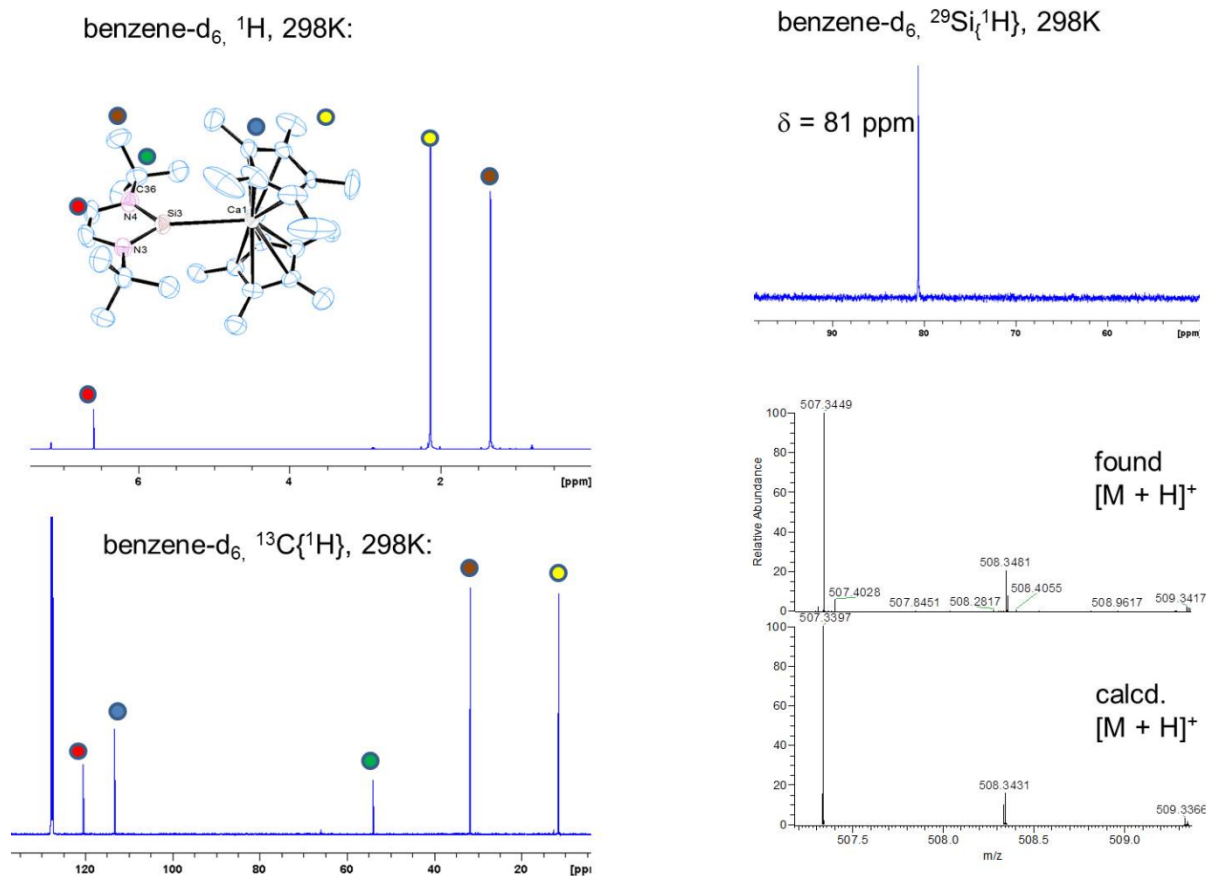


Fig. S9 ¹H, ¹³C{¹H} (with full assignment shown), ²⁹Si NMR spectrum; and ESI-MS of [M+H]⁺ of compound (7) in C₆D₆ at 298 K.

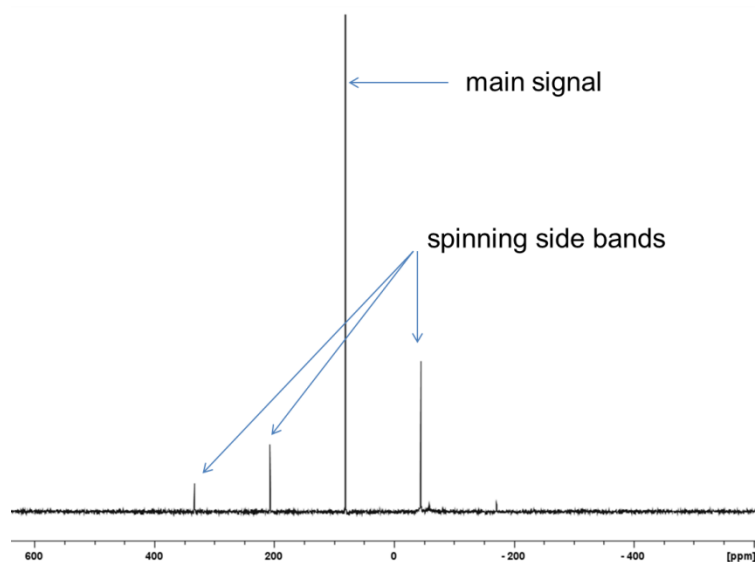


Fig. S10 ²⁹Si-MAS-NMR of compound 7 in the solid state. δ = 81.4 ppm (corresponds to chemical shift in solution).

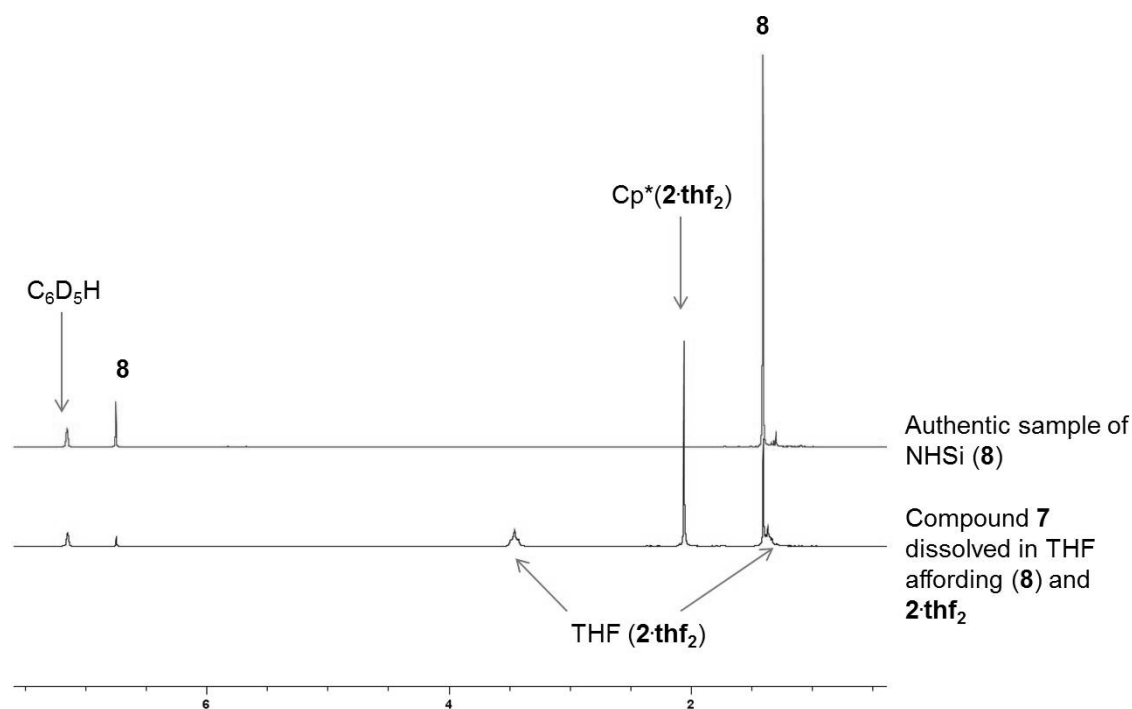


Fig. S11 Comparison of an authentic sample of NHSi **8** (top) and the NMR spectrum of compound **7** dissolved in THF, then dried *in vacuo* (both recorded in C_6D_6) (below) clearly showing free NHSi (**8**) and concomitant formation of 2-thf_2 . (The shifts match those reported in the literature. See: M. J. McCormick, S. C. Stockwell, C. E. H. Davies, T. P. Hanusa, J. C. Huffmann, *Organometallics* 1989, **8**, 2044.)

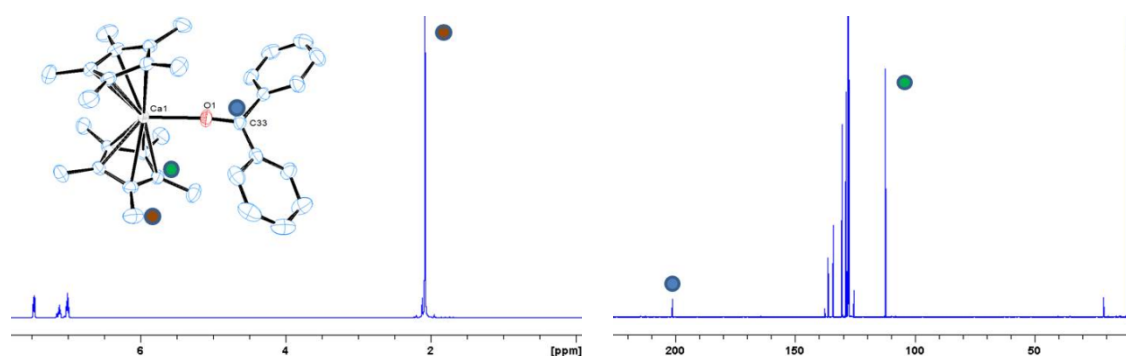


Fig. S12 ^1H NMR spectrum of isolated **10** in C_6D_6 at 298 K (left) and $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of **10** at 298 K (right) showing partial assignment.

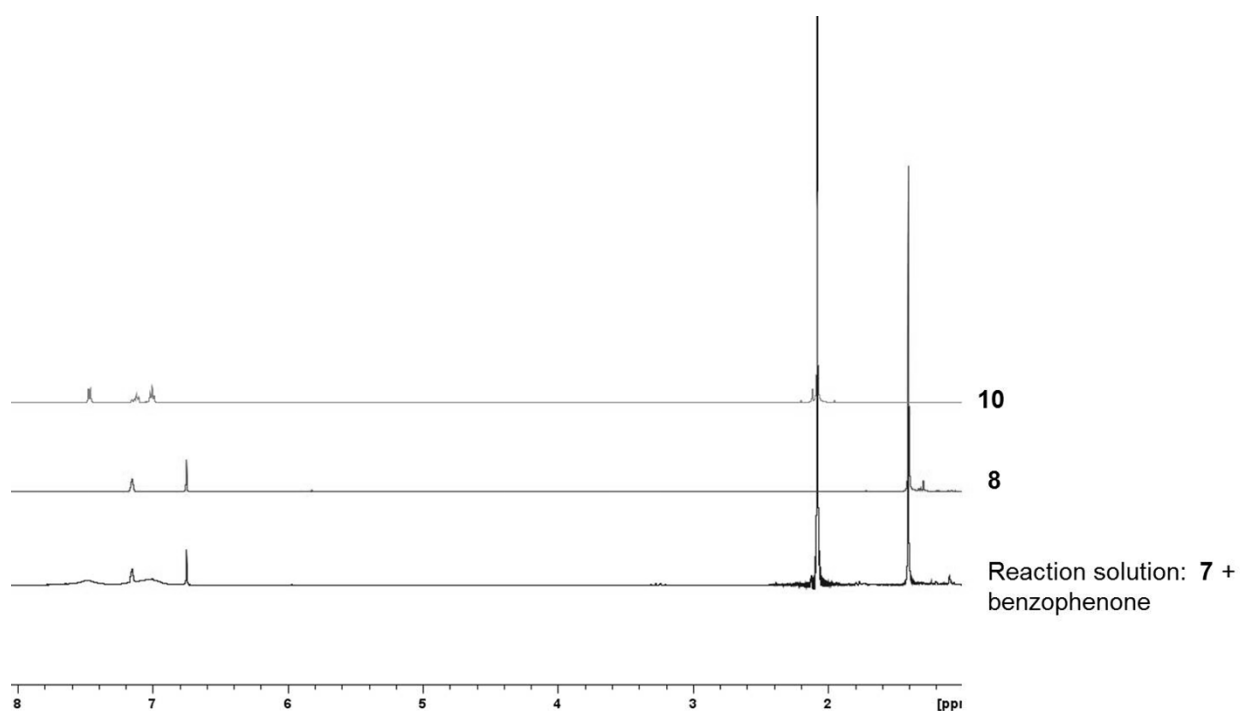


Fig. S13 Comparative ¹H NMR (in C₆D₆) spectra of the reaction solution of compound **7** with benzophenone (bottom); an authentic sample of NHSi **8** (middle) and isolated **10** (top).

2. Crystal data and refinement of compounds 3-5.

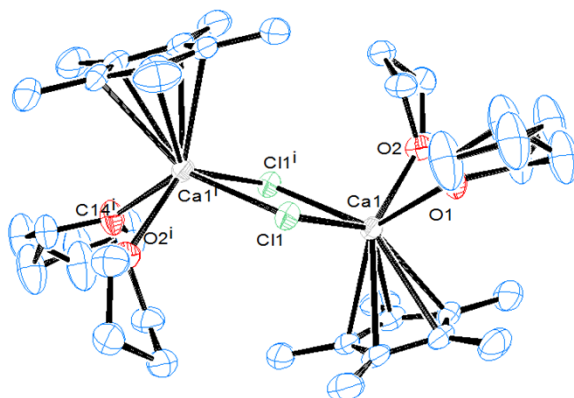


Fig. S14 Ortep representation of the molecular structure of dimer **3** in the solid state. Thermal ellipsoids are set at the 50 % probability interval and H atoms are omitted for clarity.

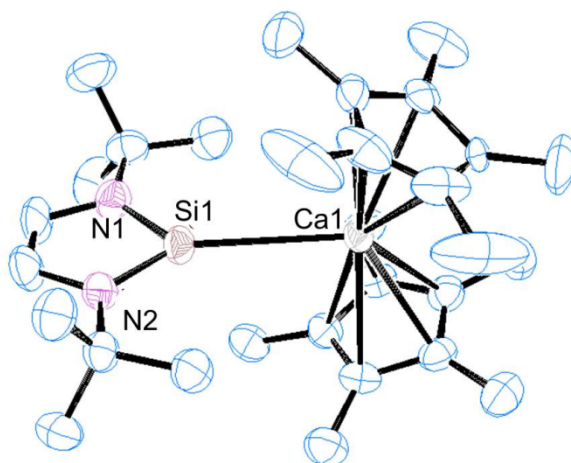


Fig. S15 Ortep representation of the molecular structure of compound **7** in the solid state. Thermal ellipsoids are set at the 50 % probability interval and H atoms are omitted for clarity. The crystal data is only of medium quality but the structural motif is definite.

Table 1s. Crystal Data and Structure Refinement for **3**, **6**, **7** and **10**.^a

	1/2·3	6	2·7	10·0.5 toluene
formula	C ₁₈ H ₃₁ CaClO ₂	C ₄₅ H ₆₆ CaN ₂ OSi	C ₆₀ H ₁₀₀ Ca ₂ N ₄ Si ₂	C _{36.50} H ₄₄ CaO
formula weight	354.96	719.17	1013.78	538.80
crystal system	Triclinic	Orthorhombic	Monoclinic	Monoclinic
space group	<i>P</i> -1	<i>P</i> 2 ₁ 2 ₁ 2 ₁	<i>P</i> 2 ₁ / <i>n</i>	<i>P</i> 2 ₁ / <i>c</i>
<i>a</i> /Å	9.0963(6)	10.74110(10)	19.1465(8)	9.91840(10)
<i>b</i> /Å	13.5005(9)	14.40170(10)	15.7739(9)	14.12280(10)
<i>c</i> /Å	17.6543(13)	27.9573(2)	21.0187(12)	23.6163(3)
<i>α</i> /deg	80.412(6)			
<i>β</i> /deg	76.114(6)		93.404(5)	101.9090(10)
<i>γ</i> /deg	70.938(6)			
<i>V</i> /Å ³	1980.1(2)	4324.72(6)	6336.8(6)	3236.87(6)
<i>Z</i>	4	4	4	4
ρ_{calcd} /g·cm ⁻³	1.191	1.105	1.063	1.106
μ /mm ⁻¹	3.997	1.758	2.190	1.843
<i>F</i> (000)	768	1568	2224	1164
crystal size/mm ³	0.15 x 0.09 x 0.08	0.43 x 0.25 x 0.12	0.21 x 0.16 x 0.13	0.39 x 0.32 x 0.11
θ range/deg	2.59–67.48	3.16–67.48	3.03–67.48	3.67–67.48
collected data	13210	27936	24037	21947
unique data	7128 (<i>R</i> _{int} = 0.0376)	7792 (<i>R</i> _{int} = 0.0304)	11406 (<i>R</i> _{int} = 0.0708)	5811 (<i>R</i> _{int} = 0.0226)
completeness to θ	100.0%	100.0%	100.0%	100.0%
data/restraints/parameters	7128 / 54 / 425	7792 / 0 / 470	11406 / 18 / 670	5811 / 92 / 390
GOF on <i>F</i> ²	1.094	1.034	1.923	1.128
final <i>R</i> indices [<i>I</i> > 2σ(<i>I</i>)]	<i>R</i> ₁ = 0.0497 <i>wR</i> ₂ = 0.1248	<i>R</i> ₁ = 0.0309 <i>wR</i> ₂ = 0.0853	<i>R</i> ₁ = 0.1855 <i>wR</i> ₂ = 0.5168	<i>R</i> ₁ = 0.0513 <i>wR</i> ₂ = 0.1442
<i>R</i> indices (all data)	<i>R</i> ₁ = 0.0648 <i>wR</i> ₂ = 0.1365	<i>R</i> ₁ = 0.0314 <i>wR</i> ₂ = 0.0857	<i>R</i> ₁ = 0.2158 <i>wR</i> ₂ = 0.5336	<i>R</i> ₁ = 0.0521 <i>wR</i> ₂ = 0.1445
Largest diff peak/hole (e·Å ⁻³)	0.598/−0.371	0.350/−0.274	2.262/−0.608	0.490/−0.372

^a All data were collected at 173(2) K using Mo K α (λ = 0.71073 Å) radiation. $R_1 = \sum(|F_o| - |F_c|)/\sum|F_o|$, $wR_2 = \{\sum[w(F_o^2 - F_c^2)^2/\sum[w(F_o^2)^2]]^{1/2}$, GOF = $\{\sum[w(F_o^2 - F_c^2)^2]/(N_o - N_p)\}^{1/2}$.

Table 2s. Selected bond (Å) and angles (°) for **3**.

Cl(2)-Ca(2)#1	2.7346(9)
Cl(2)-Ca(2)	2.7393(10)
Ca(2)-O(4)	2.417(2)
Ca(2)-O(3)	2.418(2)
Ca(2)-C(23)	2.664(3)
Ca(2)-C(19)	2.673(3)
Ca(2)-C(22)	2.678(3)
Ca(2)-C(21)	2.694(3)
Ca(2)-C(20)	2.700(3)
Ca(2)-Cl(2)#1	2.7346(9)
Ca(2)-Ca(2)#1	4.1570(12)
Cl(1)-Ca(1)	2.7298(10)
Cl(1)-Ca(1)#2	2.7545(9)
Ca(1)-O(2)	2.404(2)
Ca(1)-O(1)	2.435(2)
Ca(1)-C(5)	2.654(3)
Ca(1)-C(4)	2.666(3)
Ca(1)-C(1)	2.683(3)
Ca(1)-C(2)	2.693(3)
Ca(1)-C(3)	2.694(3)
Ca(1)-Cl(1)#2	2.7545(9)
Ca(1)-Ca(1)#2	4.1894(12)
O(1)-C(14)	1.437(5)
O(1)-C(11)	1.443(4)
O(2)-C(15)	1.445(4)
O(2)-C(18)	1.470(4)
C(1)-C(5)	1.410(5)
C(1)-C(2)	1.417(5)

C(1)-C(6)	1.502(5)
C(2)-C(3)	1.404(5)
C(2)-C(7)	1.516(5)
C(3)-C(4)	1.416(5)
C(3)-C(8)	1.511(5)
C(4)-C(5)	1.409(5)
C(4)-C(9)	1.512(5)
C(5)-C(10)	1.511(5)
C(11)-C(12A)	1.459(12)
C(11)-C(12)	1.538(11)
C(12)-C(13)	1.55(2)
C(13)-C(14)	1.395(17)
C(12A)-C(13A)	1.60(2)
C(13A)-C(14)	1.477(15)
C(15)-C(16)	1.511(5)
C(16)-C(17)	1.525(5)
C(17)-C(18)	1.497(5)
O(3)-C(29)	1.444(4)
O(3)-C(32)	1.454(4)
O(4)-C(33)	1.441(4)
O(4)-C(36)	1.443(4)
C(19)-C(20)	1.408(5)
C(19)-C(23)	1.414(5)
C(19)-C(24)	1.502(5)
C(20)-C(21)	1.418(5)
C(20)-C(25)	1.503(5)
C(21)-C(22)	1.407(5)
C(21)-C(26)	1.508(5)
C(22)-C(23)	1.411(5)
C(22)-C(27)	1.505(5)

C(23)-C(28)	1.505(5)
C(29)-C(30)	1.525(6)
C(30)-C(31)	1.481(6)
C(31)-C(32)	1.513(5)
C(33)-C(34)	1.525(5)
C(34)-C(35)	1.519(6)
C(35)-C(36)	1.513(5)
Ca(2)#1-Cl(2)-Ca(2)	98.83(3)
O(4)-Ca(2)-O(3)	74.46(8)
O(4)-Ca(2)-C(23)	109.16(10)
O(3)-Ca(2)-C(23)	139.76(10)
O(4)-Ca(2)-C(19)	87.46(9)
O(3)-Ca(2)-C(19)	114.01(10)
C(23)-Ca(2)-C(19)	30.72(11)
O(4)-Ca(2)-C(22)	137.44(9)
O(3)-Ca(2)-C(22)	123.85(10)
C(23)-Ca(2)-C(22)	30.62(10)
C(19)-Ca(2)-C(22)	50.44(10)
O(4)-Ca(2)-C(21)	128.01(10)
O(3)-Ca(2)-C(21)	94.65(10)
C(23)-Ca(2)-C(21)	50.44(11)
C(19)-Ca(2)-C(21)	50.27(10)
C(22)-Ca(2)-C(21)	30.38(11)
O(4)-Ca(2)-C(20)	97.69(10)
O(3)-Ca(2)-C(20)	89.30(9)
C(23)-Ca(2)-C(20)	50.53(11)
C(19)-Ca(2)-C(20)	30.38(11)
C(22)-Ca(2)-C(20)	50.34(10)
C(21)-Ca(2)-C(20)	30.48(11)

O(4)-Ca(2)-Cl(2)#1	130.52(6)
O(3)-Ca(2)-Cl(2)#1	80.44(6)
C(23)-Ca(2)-Cl(2)#1	117.42(8)
C(19)-Ca(2)-Cl(2)#1	141.98(8)
C(22)-Ca(2)-Cl(2)#1	91.89(7)
C(21)-Ca(2)-Cl(2)#1	95.53(8)
C(20)-Ca(2)-Cl(2)#1	124.17(8)
O(4)-Ca(2)-Cl(2)	84.13(6)
O(3)-Ca(2)-Cl(2)	130.93(6)
C(23)-Ca(2)-Cl(2)	88.80(8)
C(19)-Ca(2)-Cl(2)	108.40(9)
C(22)-Ca(2)-Cl(2)	101.81(8)
C(21)-Ca(2)-Cl(2)	132.18(8)
C(20)-Ca(2)-Cl(2)	137.75(8)
Cl(2)#1-Ca(2)-Cl(2)	81.17(3)
O(4)-Ca(2)-Ca(2)#1	111.10(6)
O(3)-Ca(2)-Ca(2)#1	108.81(6)
C(23)-Ca(2)-Ca(2)#1	106.81(8)
C(19)-Ca(2)-Ca(2)#1	136.57(9)
C(22)-Ca(2)-Ca(2)#1	99.01(7)
C(21)-Ca(2)-Ca(2)#1	120.39(8)
C(20)-Ca(2)-Ca(2)#1	149.01(8)
Cl(2)#1-Ca(2)-Ca(2)#1	40.630(19)
Cl(2)-Ca(2)-Ca(2)#1	40.544(19)
Ca(1)-Cl(1)-Ca(1)#2	99.61(3)
O(2)-Ca(1)-O(1)	74.71(8)
O(2)-Ca(1)-C(5)	126.31(10)
O(1)-Ca(1)-C(5)	134.71(10)
O(2)-Ca(1)-C(4)	96.84(10)
O(1)-Ca(1)-C(4)	125.82(10)

C(5)-Ca(1)-C(4)	30.71(11)
O(2)-Ca(1)-C(1)	141.86(9)
O(1)-Ca(1)-C(1)	106.42(10)
C(5)-Ca(1)-C(1)	30.63(11)
C(4)-Ca(1)-C(1)	50.60(11)
O(2)-Ca(1)-C(2)	115.72(10)
O(1)-Ca(1)-C(2)	84.79(10)
C(5)-Ca(1)-C(2)	50.42(10)
C(4)-Ca(1)-C(2)	50.30(10)
C(1)-Ca(1)-C(2)	30.56(11)
O(2)-Ca(1)-C(3)	91.49(9)
O(1)-Ca(1)-C(3)	95.29(10)
C(5)-Ca(1)-C(3)	50.55(10)
C(4)-Ca(1)-C(3)	30.62(11)
C(1)-Ca(1)-C(3)	50.40(10)
C(2)-Ca(1)-C(3)	30.21(11)
O(2)-Ca(1)-Cl(1)	127.95(6)
O(1)-Ca(1)-Cl(1)	85.14(6)
C(5)-Ca(1)-Cl(1)	101.99(8)
C(4)-Ca(1)-Cl(1)	132.68(9)
C(1)-Ca(1)-Cl(1)	89.61(8)
C(2)-Ca(1)-Cl(1)	109.37(8)
C(3)-Ca(1)-Cl(1)	138.53(8)
O(2)-Ca(1)-Cl(1)#2	80.90(6)
O(1)-Ca(1)-Cl(1)#2	134.49(7)
C(5)-Ca(1)-Cl(1)#2	90.62(7)
C(4)-Ca(1)-Cl(1)#2	94.47(8)
C(1)-Ca(1)-Cl(1)#2	116.31(8)
C(2)-Ca(1)-Cl(1)#2	140.70(8)
C(3)-Ca(1)-Cl(1)#2	123.49(8)

Cl(1)-Ca(1)-Cl(1)#2	80.39(3)
O(2)-Ca(1)-Ca(1)#2	107.26(6)
O(1)-Ca(1)-Ca(1)#2	113.93(6)
C(5)-Ca(1)-Ca(1)#2	98.19(7)
C(4)-Ca(1)-Ca(1)#2	119.53(8)
C(1)-Ca(1)-Ca(1)#2	106.68(8)
C(2)-Ca(1)-Ca(1)#2	136.46(8)
C(3)-Ca(1)-Ca(1)#2	148.33(8)
Cl(1)-Ca(1)-Ca(1)#2	40.410(19)
Cl(1)#2-Ca(1)-Ca(1)#2	39.975(19)
C(14)-O(1)-C(11)	108.3(3)
C(14)-O(1)-Ca(1)	125.2(2)
C(11)-O(1)-Ca(1)	126.5(2)
C(15)-O(2)-C(18)	108.2(3)
C(15)-O(2)-Ca(1)	128.1(2)
C(18)-O(2)-Ca(1)	110.54(19)
C(5)-C(1)-C(2)	107.4(3)
C(5)-C(1)-C(6)	127.2(3)
C(2)-C(1)-C(6)	125.4(3)
C(5)-C(1)-Ca(1)	73.57(18)
C(2)-C(1)-Ca(1)	75.13(19)
C(6)-C(1)-Ca(1)	118.5(2)
C(3)-C(2)-C(1)	108.5(3)
C(3)-C(2)-C(7)	126.0(3)
C(1)-C(2)-C(7)	125.4(3)
C(3)-C(2)-Ca(1)	74.91(18)
C(1)-C(2)-Ca(1)	74.31(18)
C(7)-C(2)-Ca(1)	119.8(2)
C(2)-C(3)-C(4)	107.8(3)
C(2)-C(3)-C(8)	126.6(3)

C(4)-C(3)-C(8)	125.4(4)
C(2)-C(3)-Ca(1)	74.88(18)
C(4)-C(3)-Ca(1)	73.60(18)
C(8)-C(3)-Ca(1)	122.0(2)
C(5)-C(4)-C(3)	107.9(3)
C(5)-C(4)-C(9)	125.5(4)
C(3)-C(4)-C(9)	126.5(4)
C(5)-C(4)-Ca(1)	74.19(19)
C(3)-C(4)-Ca(1)	75.77(19)
C(9)-C(4)-Ca(1)	118.7(2)
C(4)-C(5)-C(1)	108.4(3)
C(4)-C(5)-C(10)	125.0(3)
C(1)-C(5)-C(10)	126.6(4)
C(4)-C(5)-Ca(1)	75.10(19)
C(1)-C(5)-Ca(1)	75.80(19)
C(10)-C(5)-Ca(1)	116.6(2)
O(1)-C(11)-C(12A)	105.6(5)
O(1)-C(11)-C(12)	105.7(5)
C(12A)-C(11)-C(12)	29.9(7)
C(11)-C(12)-C(13)	95.2(8)
C(14)-C(13)-C(12)	106.1(13)
C(11)-C(12A)-C(13A)	106.7(11)
C(14)-C(13A)-C(12A)	97.3(9)
C(13)-C(14)-O(1)	105.6(7)
C(13)-C(14)-C(13A)	19.2(11)
O(1)-C(14)-C(13A)	112.0(7)
O(2)-C(15)-C(16)	106.4(3)
C(15)-C(16)-C(17)	101.5(3)
C(18)-C(17)-C(16)	102.2(3)
O(2)-C(18)-C(17)	105.8(3)

C(29)-O(3)-C(32)	109.1(3)
C(29)-O(3)-Ca(2)	127.6(2)
C(32)-O(3)-Ca(2)	110.2(2)
C(33)-O(4)-C(36)	104.3(2)
C(33)-O(4)-Ca(2)	124.6(2)
C(36)-O(4)-Ca(2)	131.0(2)
C(20)-C(19)-C(23)	108.5(3)
C(20)-C(19)-C(24)	127.1(4)
C(23)-C(19)-C(24)	124.0(4)
C(20)-C(19)-Ca(2)	75.86(19)
C(23)-C(19)-Ca(2)	74.31(19)
C(24)-C(19)-Ca(2)	121.6(2)
C(19)-C(20)-C(21)	107.5(3)
C(19)-C(20)-C(25)	126.5(4)
C(21)-C(20)-C(25)	125.8(4)
C(19)-C(20)-Ca(2)	73.75(19)
C(21)-C(20)-Ca(2)	74.53(19)
C(25)-C(20)-Ca(2)	121.1(2)
C(22)-C(21)-C(20)	108.1(3)
C(22)-C(21)-C(26)	125.5(4)
C(20)-C(21)-C(26)	126.2(4)
C(22)-C(21)-Ca(2)	74.19(19)
C(20)-C(21)-Ca(2)	74.99(19)
C(26)-C(21)-Ca(2)	120.7(3)
C(21)-C(22)-C(23)	108.2(3)
C(21)-C(22)-C(27)	125.1(4)
C(23)-C(22)-C(27)	126.5(4)
C(21)-C(22)-Ca(2)	75.43(19)
C(23)-C(22)-Ca(2)	74.17(19)
C(27)-C(22)-Ca(2)	119.9(2)

C(22)-C(23)-C(19)	107.6(3)
C(22)-C(23)-C(28)	126.8(4)
C(19)-C(23)-C(28)	125.4(4)
C(22)-C(23)-Ca(2)	75.21(19)
C(19)-C(23)-Ca(2)	74.97(19)
C(28)-C(23)-Ca(2)	119.4(2)
O(3)-C(29)-C(30)	104.3(3)
C(31)-C(30)-C(29)	102.4(4)
C(30)-C(31)-C(32)	103.5(3)
O(3)-C(32)-C(31)	105.9(3)
O(4)-C(33)-C(34)	105.5(3)
C(35)-C(34)-C(33)	104.3(3)
C(36)-C(35)-C(34)	104.0(3)
O(4)-C(36)-C(35)	103.5(3)

Symmetry transformations used to generate equivalent atoms:

#1 -x,-y+2,-z #2 -x+1,-y+1,-z+1

Table 3s. Selected bond lengths (Å) and angles (°) for **6**.

Ca(1)-C(19)	2.6466(15)
Ca(1)-C(18)	2.6538(15)
Ca(1)-C(30)	2.6621(16)
Ca(1)-C(29)	2.6674(16)
Ca(1)-C(26)	2.6673(17)
Ca(1)-C(28)	2.6828(15)
Ca(1)-C(20)	2.6859(15)
Ca(1)-C(17)	2.6889(16)
Ca(1)-C(27)	2.6920(16)
Ca(1)-C(16)	2.7022(16)

Ca(1)-Si(1)	3.2732(5)
Si(1)-O(1)	1.6893(12)
Si(1)-N(1)	1.8632(13)
Si(1)-N(2)	1.8831(14)
Si(1)-C(1)	2.3310(16)
N(1)-C(1)	1.334(2)
N(1)-C(12)	1.484(2)
N(2)-C(1)	1.334(2)
N(2)-C(8)	1.487(2)
O(1)-C(36)	1.3715(19)
C(1)-C(2)	1.490(2)
C(2)-C(7)	1.384(2)
C(2)-C(3)	1.387(2)
C(3)-C(4)	1.387(3)
C(4)-C(5)	1.383(3)
C(5)-C(6)	1.378(3)
C(6)-C(7)	1.388(2)
C(8)-C(10)	1.510(3)
C(8)-C(9)	1.516(2)
C(8)-C(11)	1.545(3)
C(12)-C(14)	1.517(3)
C(12)-C(13)	1.529(3)
C(12)-C(15)	1.529(2)
C(16)-C(17)	1.409(2)
C(16)-C(20)	1.421(2)
C(16)-C(21)	1.505(2)
C(17)-C(18)	1.416(2)
C(17)-C(22)	1.504(3)
C(18)-C(19)	1.420(2)
C(18)-C(23)	1.504(2)

C(19)-C(20)	1.421(2)
C(19)-C(24)	1.508(2)
C(20)-C(25)	1.505(2)
C(26)-C(30)	1.409(3)
C(26)-C(27)	1.414(3)
C(26)-C(31)	1.508(3)
C(27)-C(28)	1.414(2)
C(27)-C(32)	1.501(3)
C(28)-C(29)	1.413(2)
C(28)-C(33)	1.497(2)
C(29)-C(30)	1.409(2)
C(29)-C(34)	1.500(3)
C(30)-C(35)	1.508(3)
C(36)-C(37)	1.387(2)
C(36)-C(41)	1.407(2)
C(37)-C(38)	1.384(2)
C(38)-C(39)	1.374(3)
C(39)-C(40)	1.384(3)
C(40)-C(41)	1.399(2)
C(41)-C(42)	1.535(3)
C(42)-C(45)	1.536(3)
C(42)-C(43)	1.536(3)
C(42)-C(44)	1.542(3)
C(19)-Ca(1)-C(18)	31.07(5)
C(19)-Ca(1)-C(30)	165.16(5)
C(18)-Ca(1)-C(30)	145.39(5)
C(19)-Ca(1)-C(29)	134.50(5)
C(18)-Ca(1)-C(29)	125.32(5)
C(30)-Ca(1)-C(29)	30.67(5)

C(19)-Ca(1)-C(26)	152.67(6)
C(18)-Ca(1)-C(26)	121.61(6)
C(30)-Ca(1)-C(26)	30.66(6)
C(29)-Ca(1)-C(26)	50.63(5)
C(19)-Ca(1)-C(28)	117.36(5)
C(18)-Ca(1)-C(28)	97.42(5)
C(30)-Ca(1)-C(28)	50.60(5)
C(29)-Ca(1)-C(28)	30.62(5)
C(26)-Ca(1)-C(28)	50.56(5)
C(19)-Ca(1)-C(20)	30.89(5)
C(18)-Ca(1)-C(20)	50.88(5)
C(30)-Ca(1)-C(20)	162.23(5)
C(29)-Ca(1)-C(20)	161.13(5)
C(26)-Ca(1)-C(20)	148.17(5)
C(28)-Ca(1)-C(20)	147.17(5)
C(19)-Ca(1)-C(17)	50.95(5)
C(18)-Ca(1)-C(17)	30.73(5)
C(30)-Ca(1)-C(17)	136.29(5)
C(29)-Ca(1)-C(17)	138.41(5)
C(26)-Ca(1)-C(17)	105.66(6)
C(28)-Ca(1)-C(17)	108.12(5)
C(20)-Ca(1)-C(17)	50.48(5)
C(19)-Ca(1)-C(27)	125.16(5)
C(18)-Ca(1)-C(27)	95.56(5)
C(30)-Ca(1)-C(27)	50.53(5)
C(29)-Ca(1)-C(27)	50.49(5)
C(26)-Ca(1)-C(27)	30.59(6)
C(28)-Ca(1)-C(27)	30.51(5)
C(20)-Ca(1)-C(27)	141.39(5)
C(17)-Ca(1)-C(27)	91.02(5)

C(19)-Ca(1)-C(16)	50.85(5)
C(18)-Ca(1)-C(16)	50.59(5)
C(30)-Ca(1)-C(16)	143.28(6)
C(29)-Ca(1)-C(16)	166.25(5)
C(26)-Ca(1)-C(16)	117.96(6)
C(28)-Ca(1)-C(16)	138.03(5)
C(20)-Ca(1)-C(16)	30.58(5)
C(17)-Ca(1)-C(16)	30.31(5)
C(27)-Ca(1)-C(16)	115.78(5)
C(19)-Ca(1)-Si(1)	97.09(4)
C(18)-Ca(1)-Si(1)	128.16(4)
C(30)-Ca(1)-Si(1)	83.88(4)
C(29)-Ca(1)-Si(1)	88.01(4)
C(26)-Ca(1)-Si(1)	110.22(4)
C(28)-Ca(1)-Si(1)	117.45(4)
C(20)-Ca(1)-Si(1)	84.25(4)
C(17)-Ca(1)-Si(1)	133.39(4)
C(27)-Ca(1)-Si(1)	133.99(4)
C(16)-Ca(1)-Si(1)	104.43(4)
O(1)-Si(1)-N(1)	96.54(6)
O(1)-Si(1)-N(2)	98.79(6)
N(1)-Si(1)-N(2)	69.77(6)
O(1)-Si(1)-C(1)	100.20(6)
N(1)-Si(1)-C(1)	34.88(6)
N(2)-Si(1)-C(1)	34.91(6)
O(1)-Si(1)-Ca(1)	125.94(4)
N(1)-Si(1)-Ca(1)	110.44(4)
N(2)-Si(1)-Ca(1)	133.95(5)
C(1)-Si(1)-Ca(1)	128.71(4)
C(1)-N(1)-C(12)	132.02(14)

C(1)-N(1)-Si(1)	92.11(9)
C(12)-N(1)-Si(1)	134.42(11)
C(1)-N(2)-C(8)	130.13(14)
C(1)-N(2)-Si(1)	91.22(10)
C(8)-N(2)-Si(1)	138.06(11)
C(36)-O(1)-Si(1)	130.56(11)
N(2)-C(1)-N(1)	106.85(13)
N(2)-C(1)-C(2)	127.31(15)
N(1)-C(1)-C(2)	125.82(14)
N(2)-C(1)-Si(1)	53.87(8)
N(1)-C(1)-Si(1)	53.01(8)
C(2)-C(1)-Si(1)	178.78(12)
C(7)-C(2)-C(3)	120.50(16)
C(7)-C(2)-C(1)	121.54(15)
C(3)-C(2)-C(1)	117.93(16)
C(2)-C(3)-C(4)	119.26(19)
C(5)-C(4)-C(3)	120.40(19)
C(6)-C(5)-C(4)	120.03(18)
C(5)-C(6)-C(7)	120.17(19)
C(2)-C(7)-C(6)	119.64(17)
N(2)-C(8)-C(10)	112.07(16)
N(2)-C(8)-C(9)	106.42(13)
C(10)-C(8)-C(9)	110.14(18)
N(2)-C(8)-C(11)	109.92(15)
C(10)-C(8)-C(11)	110.8(2)
C(9)-C(8)-C(11)	107.29(16)
N(1)-C(12)-C(14)	108.71(14)
N(1)-C(12)-C(13)	106.15(14)
C(14)-C(12)-C(13)	109.19(18)
N(1)-C(12)-C(15)	114.38(14)

C(14)-C(12)-C(15)	110.21(17)
C(13)-C(12)-C(15)	108.03(15)
C(17)-C(16)-C(20)	108.13(15)
C(17)-C(16)-C(21)	124.85(16)
C(20)-C(16)-C(21)	126.42(16)
C(17)-C(16)-Ca(1)	74.32(9)
C(20)-C(16)-Ca(1)	74.07(9)
C(21)-C(16)-Ca(1)	124.55(11)
C(16)-C(17)-C(18)	108.20(15)
C(16)-C(17)-C(22)	124.48(16)
C(18)-C(17)-C(22)	126.41(16)
C(16)-C(17)-Ca(1)	75.37(9)
C(18)-C(17)-Ca(1)	73.26(9)
C(22)-C(17)-Ca(1)	125.97(12)
C(17)-C(18)-C(19)	108.08(14)
C(17)-C(18)-C(23)	126.34(16)
C(19)-C(18)-C(23)	125.37(16)
C(17)-C(18)-Ca(1)	76.01(9)
C(19)-C(18)-Ca(1)	74.19(9)
C(23)-C(18)-Ca(1)	120.07(11)
C(18)-C(19)-C(20)	107.74(14)
C(18)-C(19)-C(24)	125.60(15)
C(20)-C(19)-C(24)	126.58(15)
C(18)-C(19)-Ca(1)	74.74(9)
C(20)-C(19)-Ca(1)	76.08(8)
C(24)-C(19)-Ca(1)	117.86(10)
C(19)-C(20)-C(16)	107.85(14)
C(19)-C(20)-C(25)	126.58(16)
C(16)-C(20)-C(25)	124.93(15)
C(19)-C(20)-Ca(1)	73.03(8)

C(16)-C(20)-Ca(1)	75.34(9)
C(25)-C(20)-Ca(1)	124.71(10)
C(30)-C(26)-C(27)	108.13(15)
C(30)-C(26)-C(31)	125.9(2)
C(27)-C(26)-C(31)	125.7(2)
C(30)-C(26)-Ca(1)	74.47(9)
C(27)-C(26)-Ca(1)	75.67(10)
C(31)-C(26)-Ca(1)	120.31(13)
C(28)-C(27)-C(26)	107.78(15)
C(28)-C(27)-C(32)	126.03(19)
C(26)-C(27)-C(32)	125.4(2)
C(28)-C(27)-Ca(1)	74.39(9)
C(26)-C(27)-Ca(1)	73.74(9)
C(32)-C(27)-Ca(1)	125.48(13)
C(29)-C(28)-C(27)	107.93(15)
C(29)-C(28)-C(33)	125.69(17)
C(27)-C(28)-C(33)	125.82(17)
C(29)-C(28)-Ca(1)	74.09(9)
C(27)-C(28)-Ca(1)	75.10(9)
C(33)-C(28)-Ca(1)	123.49(11)
C(30)-C(29)-C(28)	108.08(15)
C(30)-C(29)-C(34)	127.09(17)
C(28)-C(29)-C(34)	124.76(16)
C(30)-C(29)-Ca(1)	74.46(9)
C(28)-C(29)-Ca(1)	75.30(9)
C(34)-C(29)-Ca(1)	118.74(11)
C(26)-C(30)-C(29)	108.07(15)
C(26)-C(30)-C(35)	126.75(19)
C(29)-C(30)-C(35)	124.92(19)
C(26)-C(30)-Ca(1)	74.87(10)

C(29)-C(30)-Ca(1)	74.87(9)
C(35)-C(30)-Ca(1)	120.96(11)
O(1)-C(36)-C(37)	120.21(15)
O(1)-C(36)-C(41)	118.66(15)
C(37)-C(36)-C(41)	121.13(15)
C(38)-C(37)-C(36)	121.14(17)
C(39)-C(38)-C(37)	119.02(18)
C(38)-C(39)-C(40)	119.91(17)
C(39)-C(40)-C(41)	122.91(19)
C(40)-C(41)-C(36)	115.88(17)
C(40)-C(41)-C(42)	121.66(17)
C(36)-C(41)-C(42)	122.44(15)
C(45)-C(42)-C(41)	110.79(17)
C(45)-C(42)-C(43)	109.86(18)
C(41)-C(42)-C(43)	109.27(16)
C(45)-C(42)-C(44)	106.99(19)
C(41)-C(42)-C(44)	112.02(17)
C(43)-C(42)-C(44)	107.8(2)

Symmetry transformations used to generate equivalent atoms:

Table 4s. Selected bond lengths (Å) and angles (°) for **10**.

Ca(1)-O(1)	2.2925(19)
Ca(1)-C(2)	2.632(3)
Ca(1)-C(3)	2.633(3)
Ca(1)-C(9)	2.634(3)
Ca(1)-C(10)	2.641(3)
Ca(1)-C(4)	2.644(3)
Ca(1)-C(1)	2.646(3)
Ca(1)-C(8)	2.652(3)

Ca(1)-C(5)	2.668(3)
Ca(1)-C(6)	2.670(3)
Ca(1)-C(7)	2.680(3)
O(1)-C(33)	1.238(3)
C(1)-C(2)	1.410(4)
C(1)-C(5)	1.415(4)
C(1)-C(11)	1.505(4)
C(2)-C(3)	1.413(4)
C(2)-C(12)	1.511(4)
C(3)-C(4)	1.418(4)
C(3)-C(13)	1.509(4)
C(4)-C(5)	1.418(4)
C(4)-C(14)	1.503(4)
C(5)-C(15)	1.511(4)
C(6)-C(7)	1.410(4)
C(6)-C(10)	1.417(4)
C(6)-C(19)	1.508(4)
C(7)-C(8)	1.418(4)
C(7)-C(18)	1.506(4)
C(8)-C(9)	1.413(4)
C(8)-C(17)	1.506(4)
C(9)-C(10)	1.406(4)
C(9)-C(16)	1.508(4)
C(10)-C(20)	1.511(4)
C(21)-C(22)	1.390(4)
C(21)-C(26)	1.391(4)
C(21)-C(33)	1.481(4)
C(22)-C(23)	1.375(5)
C(23)-C(24)	1.378(5)
C(24)-C(25)	1.388(5)

C(25)-C(26)	1.383(4)
C(27)-C(32)	1.389(4)
C(27)-C(28)	1.397(4)
C(27)-C(33)	1.477(4)
C(28)-C(29)	1.372(5)
C(29)-C(30)	1.381(5)
C(30)-C(31)	1.378(5)
C(31)-C(32)	1.387(4)
C(34)-C(39)	1.369(9)
C(34)-C(35)	1.380(10)
C(34)-C(40)	1.494(12)
C(35)-C(36)	1.359(10)
C(36)-C(37)	1.329(10)
C(37)-C(38)	1.388(11)
C(38)-C(39)	1.391(10)
O(1)-Ca(1)-C(2)	125.69(8)
O(1)-Ca(1)-C(3)	130.05(8)
C(2)-Ca(1)-C(3)	31.13(9)
O(1)-Ca(1)-C(9)	84.77(8)
C(2)-Ca(1)-C(9)	130.89(9)
C(3)-Ca(1)-C(9)	145.09(9)
O(1)-Ca(1)-C(10)	84.30(8)
C(2)-Ca(1)-C(10)	148.37(8)
C(3)-Ca(1)-C(10)	137.17(9)
C(9)-Ca(1)-C(10)	30.90(9)
O(1)-Ca(1)-C(4)	100.19(8)
C(2)-Ca(1)-C(4)	51.45(9)
C(3)-Ca(1)-C(4)	31.17(8)
C(9)-Ca(1)-C(4)	171.09(9)

C(10)-Ca(1)-C(4)	141.61(9)
O(1)-Ca(1)-C(1)	94.78(8)
C(2)-Ca(1)-C(1)	30.98(8)
C(3)-Ca(1)-C(1)	51.37(9)
C(9)-Ca(1)-C(1)	136.04(9)
C(10)-Ca(1)-C(1)	166.93(9)
C(4)-Ca(1)-C(1)	51.40(9)
O(1)-Ca(1)-C(8)	113.46(9)
C(2)-Ca(1)-C(8)	101.86(9)
C(3)-Ca(1)-C(8)	115.03(9)
C(9)-Ca(1)-C(8)	31.01(9)
C(10)-Ca(1)-C(8)	51.18(9)
C(4)-Ca(1)-C(8)	146.05(9)
C(1)-Ca(1)-C(8)	118.38(9)
O(1)-Ca(1)-C(5)	80.57(8)
C(2)-Ca(1)-C(5)	51.06(8)
C(3)-Ca(1)-C(5)	51.17(8)
C(9)-Ca(1)-C(5)	157.89(9)
C(10)-Ca(1)-C(5)	159.13(9)
C(4)-Ca(1)-C(5)	30.97(9)
C(1)-Ca(1)-C(5)	30.89(9)
C(8)-Ca(1)-C(5)	149.22(9)
O(1)-Ca(1)-C(6)	112.69(8)
C(2)-Ca(1)-C(6)	121.59(9)
C(3)-Ca(1)-C(6)	106.50(9)
C(9)-Ca(1)-C(6)	50.98(8)
C(10)-Ca(1)-C(6)	30.95(8)
C(4)-Ca(1)-C(6)	120.11(9)
C(1)-Ca(1)-C(6)	152.52(9)
C(8)-Ca(1)-C(6)	50.92(8)

C(5)-Ca(1)-C(6)	150.91(9)
O(1)-Ca(1)-C(7)	133.49(8)
C(2)-Ca(1)-C(7)	97.49(9)
C(3)-Ca(1)-C(7)	95.51(8)
C(9)-Ca(1)-C(7)	50.94(8)
C(10)-Ca(1)-C(7)	50.92(8)
C(4)-Ca(1)-C(7)	122.06(9)
C(1)-Ca(1)-C(7)	125.67(9)
C(8)-Ca(1)-C(7)	30.84(9)
C(5)-Ca(1)-C(7)	145.94(8)
C(6)-Ca(1)-C(7)	30.57(9)
C(33)-O(1)-Ca(1)	174.4(2)
C(2)-C(1)-C(5)	107.9(2)
C(2)-C(1)-C(11)	125.6(3)
C(5)-C(1)-C(11)	126.5(3)
C(2)-C(1)-Ca(1)	73.95(15)
C(5)-C(1)-Ca(1)	75.41(15)
C(11)-C(1)-Ca(1)	116.41(19)
C(1)-C(2)-C(3)	108.3(2)
C(1)-C(2)-C(12)	125.0(3)
C(3)-C(2)-C(12)	126.5(3)
C(1)-C(2)-Ca(1)	75.06(15)
C(3)-C(2)-Ca(1)	74.47(15)
C(12)-C(2)-Ca(1)	120.06(19)
C(2)-C(3)-C(4)	108.0(2)
C(2)-C(3)-C(13)	126.2(3)
C(4)-C(3)-C(13)	125.5(3)
C(2)-C(3)-Ca(1)	74.40(15)
C(4)-C(3)-Ca(1)	74.84(15)
C(13)-C(3)-Ca(1)	121.97(19)

C(3)-C(4)-C(5)	107.6(2)
C(3)-C(4)-C(14)	125.2(3)
C(5)-C(4)-C(14)	127.0(3)
C(3)-C(4)-Ca(1)	73.99(15)
C(5)-C(4)-Ca(1)	75.46(16)
C(14)-C(4)-Ca(1)	119.89(19)
C(1)-C(5)-C(4)	108.1(2)
C(1)-C(5)-C(15)	125.8(3)
C(4)-C(5)-C(15)	125.9(3)
C(1)-C(5)-Ca(1)	73.70(15)
C(4)-C(5)-Ca(1)	73.56(15)
C(15)-C(5)-Ca(1)	123.09(18)
C(7)-C(6)-C(10)	108.0(2)
C(7)-C(6)-C(19)	125.0(3)
C(10)-C(6)-C(19)	126.8(3)
C(7)-C(6)-Ca(1)	75.11(16)
C(10)-C(6)-Ca(1)	73.42(15)
C(19)-C(6)-Ca(1)	121.26(19)
C(6)-C(7)-C(8)	108.0(2)
C(6)-C(7)-C(18)	125.1(3)
C(8)-C(7)-C(18)	126.3(3)
C(6)-C(7)-Ca(1)	74.32(15)
C(8)-C(7)-Ca(1)	73.49(15)
C(18)-C(7)-Ca(1)	125.2(2)
C(9)-C(8)-C(7)	107.7(2)
C(9)-C(8)-C(17)	126.2(3)
C(7)-C(8)-C(17)	125.8(3)
C(9)-C(8)-Ca(1)	73.82(15)
C(7)-C(8)-Ca(1)	75.67(15)
C(17)-C(8)-Ca(1)	121.7(2)

C(10)-C(9)-C(8)	108.4(2)
C(10)-C(9)-C(16)	126.5(3)
C(8)-C(9)-C(16)	125.0(3)
C(10)-C(9)-Ca(1)	74.82(15)
C(8)-C(9)-Ca(1)	75.17(16)
C(16)-C(9)-Ca(1)	119.60(18)
C(9)-C(10)-C(6)	107.9(2)
C(9)-C(10)-C(20)	125.9(3)
C(6)-C(10)-C(20)	126.2(3)
C(9)-C(10)-Ca(1)	74.28(15)
C(6)-C(10)-Ca(1)	75.62(15)
C(20)-C(10)-Ca(1)	116.81(18)
C(22)-C(21)-C(26)	119.1(3)
C(22)-C(21)-C(33)	120.2(3)
C(26)-C(21)-C(33)	120.4(3)
C(23)-C(22)-C(21)	121.0(3)
C(22)-C(23)-C(24)	119.7(3)
C(23)-C(24)-C(25)	120.2(3)
C(26)-C(25)-C(24)	120.1(3)
C(25)-C(26)-C(21)	119.9(3)
C(32)-C(27)-C(28)	118.8(3)
C(32)-C(27)-C(33)	121.0(3)
C(28)-C(27)-C(33)	120.1(2)
C(29)-C(28)-C(27)	121.1(3)
C(28)-C(29)-C(30)	119.3(3)
C(31)-C(30)-C(29)	120.8(3)
C(30)-C(31)-C(32)	119.9(3)
C(31)-C(32)-C(27)	120.0(3)
O(1)-C(33)-C(27)	119.4(3)
O(1)-C(33)-C(21)	119.4(3)

C(27)-C(33)-C(21)	121.1(2)
C(39)-C(34)-C(35)	118.9(7)
C(39)-C(34)-C(40)	123.5(8)
C(35)-C(34)-C(40)	117.6(7)
C(36)-C(35)-C(34)	119.3(7)
C(37)-C(36)-C(35)	123.0(8)
C(36)-C(37)-C(38)	119.1(8)
C(37)-C(38)-C(39)	118.9(7)
C(34)-C(39)-C(38)	120.7(8)

Symmetry transformations used to generate equivalent atoms:

3. Computational Details

We have used density functional theory in the Kohn–Sham formulation as implemented in the Turbomole 6.4 program ¹ to locate minima on the potential energy surface of the systems studied. A recent implementation of Grimme’s D3 dispersion parameters ² in Turbomole has paved the way for calculations on large organometallic systems that include dispersive interactions. To account for these forces we optimised all stationary points at the BP86-D3 level of theory, which contains Grimme’s D3 dispersion parameters. The double ζ -quality basis set DZP ³ as well as the multipole-accelerated resolution of the identity approximation MARI-J with suitable fitting basis sets ⁴ were used in these computations. All stationary points were fully optimised. Molecular orbitals were drawn with Molden 4.8 ⁵ using ± 0.03 au isosurfaces for the orbital plots. A natural population analysis (NPA)⁶ and resulting Wiberg bond indices (WBI)⁷ were used for a detailed study of the electronic structure and bonding in some cases. Instead of the distances, WBIs describe the bond strengths, allowing the limitation of looking only at bond lengths to be overcome. The Gaussian 09 software package⁸ was used at this stage.

4. Z-Matrices and optimized structures

Structure 6

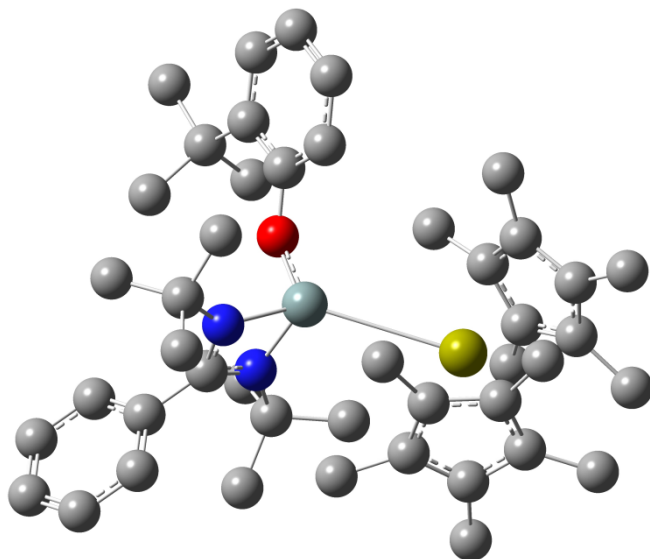


Fig. S16 Geometry optimized structure of compound **6**.

116

FINAL	HEAT OF FORMATION = -2905.301420		
Ca	-0.346715	-0.163854	-2.563001
O	0.747484	-1.232751	1.540727
Si	-0.249758	-0.232405	0.526279
N	0.460978	1.411368	1.176401
N	-1.358048	0.494390	1.925614
C	1.833041	-1.713366	-2.381196
C	-2.451616	0.011520	2.798653
C	0.673141	-2.558418	1.834038
C	-0.549111	1.561337	2.049309
C	0.789294	-2.498188	-2.979247
C	-1.619776	3.747929	2.719442
H	-2.236657	3.709094	1.810550
C	-0.322934	-4.763836	1.450631
H	-1.029022	-5.399954	0.895494
C	-1.859503	1.176156	-4.284708
C	1.373886	-0.831538	-4.481748
C	0.100978	2.757270	4.140560
H	0.802976	1.934400	4.342605
C	-2.725322	0.225117	-3.651202
C	1.525259	-3.085597	2.852328
C	-0.231570	-3.399091	1.148728
H	-0.855585	-2.960407	0.354733
C	-0.902347	4.889245	4.747558
H	-0.976527	5.745889	5.434556
C	0.508543	-1.955498	-4.276807
C	-0.710856	2.706021	2.988502

C	-2.874088	0.606915	-2.280326
C	2.185092	-0.678929	-3.312145
C	1.575201	2.289365	0.769211
C	0.131345	-3.728153	-2.406956
H	0.550595	-3.996864	-1.418737
H	0.263327	-4.609323	-3.072359
H	-0.969131	-3.609986	-2.270320
C	2.581340	-2.031828	-1.113271
H	3.549643	-2.537284	-1.327932
H	2.008659	-2.700053	-0.442562
H	2.828294	-1.125820	-0.521478
C	-1.710065	4.838448	3.598108
H	-2.413914	5.656068	3.381337
C	1.575201	-0.045039	-5.748128
H	2.555949	-0.286658	-6.217214
H	1.575902	1.052223	-5.571880
H	0.793534	-0.255855	-6.501806
C	2.576170	-2.202223	3.559959
C	1.394113	-4.461398	3.135676
H	2.025281	-4.909208	3.914669
C	-2.102700	1.796271	-2.060857
C	0.490545	-5.298482	2.458499
H	0.430878	-6.365671	2.719304
C	-1.460955	2.137576	-3.298243
C	1.831065	1.991891	-0.721253
H	2.014664	0.910029	-0.883166
H	2.724632	2.533063	-1.085564
H	0.967062	2.305290	-1.344030
C	3.348824	0.273227	-3.215419
H	4.142557	0.014533	-3.952521
H	3.823677	0.253652	-2.214860
H	3.061988	1.326489	-3.427110
C	-0.437234	-2.564658	-5.276533
H	-0.774622	-1.828279	-6.033017
H	-1.347336	-2.975865	-4.793945
H	0.036356	-3.405935	-5.832036
C	-3.764617	-0.090493	-1.289307
H	-3.619895	-1.193596	-1.292861
H	-4.844453	0.082458	-1.498969
H	-3.571846	0.258189	-0.256249
C	1.278722	3.793968	0.935482
H	0.326258	4.074152	0.444541
H	2.093949	4.373132	0.458067
H	1.224534	4.100847	1.996047
C	-2.123503	2.642877	-0.816818
H	-1.125209	3.048862	-0.557575
H	-2.469710	2.062873	0.060882
H	-2.804908	3.517526	-0.923612
C	3.609977	-1.736397	2.504132
H	4.135104	-2.606354	2.061212
H	4.368501	-1.069668	2.965633
H	3.115598	-1.188623	1.681941
C	-3.459950	-0.912209	-4.309900
H	-3.062952	-1.131809	-5.319207
H	-4.544639	-0.688844	-4.424946
H	-3.396725	-1.854490	-3.722649
C	-2.864428	-1.375130	2.268910
H	-3.147968	-1.327551	1.198988
H	-3.736610	-1.754269	2.836510

H	-2.042583	-2.108703	2.384465
C	1.909370	-0.972872	4.232341
H	1.383332	-0.344359	3.491608
H	2.679475	-0.352009	4.737218
H	1.178855	-1.297422	5.001643
C	-0.609956	3.354981	-3.557806
H	-1.143720	4.119169	-4.166485
H	0.323482	3.110255	-4.110629
H	-0.308651	3.854437	-2.614362
C	-3.664382	0.965097	2.745588
H	-3.418629	1.957142	3.168760
H	-4.503447	0.544446	3.335584
H	-4.012234	1.103576	1.702789
C	3.335865	-2.973455	4.660031
H	2.654992	-3.340019	5.455584
H	4.076635	-2.300737	5.138432
H	3.895041	-3.840380	4.253528
C	-1.579807	1.275563	-5.758524
H	-2.390977	1.827251	-6.285530
H	-1.502693	0.280831	-6.241117
H	-0.635953	1.814967	-5.967717
C	0.000108	3.846158	5.019863
H	0.630925	3.881344	5.920856
C	-1.947849	-0.140933	4.250301
H	-1.040483	-0.775477	4.273990
H	-2.727822	-0.620040	4.875634
H	-1.705185	0.839266	4.704066
C	2.818463	1.904225	1.599230
H	2.639443	2.096098	2.675825
H	3.701464	2.494600	1.281537
H	3.049605	0.829186	1.481149

Structure 7

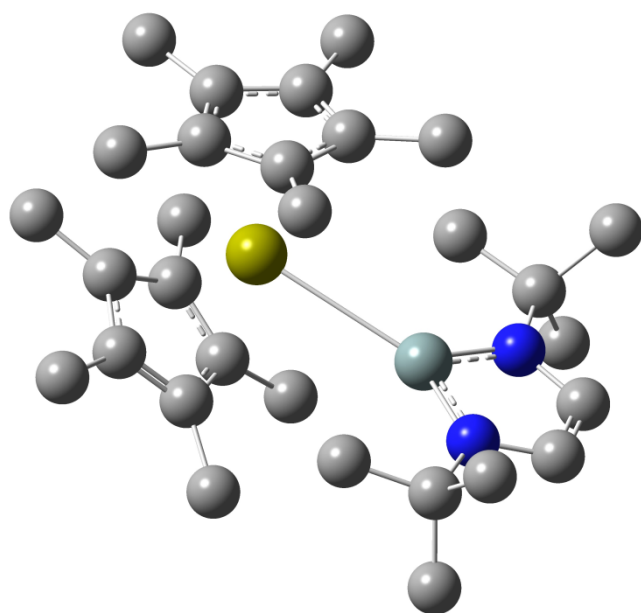


Fig. S17 Geometry optimized structure of compound 7. Calculated $d(\text{Ca-Si}) = 3.090 \text{ \AA}$.

FINAL HEAT OF FORMATION = -2249.201890

Ca	-0.942642	0.592913	-0.544730
Si	1.462311	-0.644233	0.793773
C	-3.520502	0.364441	-0.124212
C	-2.279081	-1.491356	0.503795
C	-3.085830	-0.938030	-0.544689
N	2.436905	-0.701667	2.290489
C	-0.875587	3.000229	-1.566587
C	-0.163600	1.315730	-2.997938
C	-1.279776	2.144824	-2.645409
C	2.368095	1.210583	-2.290912
H	2.809415	0.876114	-1.326087
H	3.019664	2.026196	-2.675789
H	2.465845	0.364645	-3.000836
C	-2.961374	0.623569	1.172081
C	0.938848	1.669518	-2.153020
C	-1.770772	-2.905155	0.609009
H	-0.741277	-2.953683	1.024921
H	-1.759505	-3.423128	-0.371437
H	-2.413097	-3.516988	1.282282
C	3.279137	-1.810087	2.331395
H	3.947401	-1.991872	3.182729
C	-0.175821	0.327233	-4.135502
H	-0.373435	0.825687	-5.110407
H	-0.967109	-0.446853	-4.017399
H	0.789562	-0.207446	-4.234515
C	3.181898	-2.615729	1.224728
H	3.763510	-3.532260	1.064612
C	-1.699058	4.082334	-0.918771
H	-1.506629	5.082477	-1.368306
H	-1.483713	4.176888	0.166647
H	-2.786531	3.889512	-1.016693
N	2.258865	-2.163856	0.286830
C	1.970633	-0.403096	4.705956
H	0.955827	-0.829625	4.581698
H	1.944930	0.324044	5.542245
H	2.657674	-1.223830	4.993205
C	2.425667	0.291020	3.404554
C	3.368048	-2.786990	-1.831639
H	3.623930	-1.725257	-2.019846
H	3.248755	-3.298790	-2.807731
H	4.219855	-3.258198	-1.301964
C	0.922036	-2.216028	-1.790940
H	-0.038294	-2.283768	-1.241652
H	0.783837	-2.720593	-2.766591
H	1.144957	-1.151489	-2.003203
C	-2.563710	2.197248	-3.427420
H	-2.397452	2.630219	-4.439744
H	-3.332246	2.818431	-2.930752
H	-3.005545	1.190672	-3.587557
C	-2.205860	-0.527311	1.565132
C	3.838519	0.890247	3.571883
H	4.582602	0.120275	3.857208
H	3.836123	1.665061	4.364545
H	4.172879	1.358427	2.624376
C	1.699731	-4.357831	-0.736809
H	2.513355	-4.900067	-0.215962
H	1.508550	-4.883413	-1.693632

H	0.785817	-4.421237	-0.114482
C	2.067153	-2.883612	-1.005792
C	1.439683	1.421781	3.056903
H	1.752189	1.966783	2.144068
H	1.396345	2.151807	3.888899
H	0.414656	1.031160	2.896539
C	0.491107	2.697785	-1.256476
C	1.357773	3.424936	-0.263007
H	2.209875	2.797121	0.071337
H	0.794761	3.732883	0.643792
H	1.796006	4.355607	-0.690716
C	-1.628705	-0.804027	2.927630
H	-0.643905	-1.315747	2.874822
H	-2.298933	-1.464986	3.523063
H	-1.490820	0.122259	3.521129
C	-4.539369	1.232480	-0.810371
H	-5.539544	1.128010	-0.332619
H	-4.662330	0.968856	-1.877718
H	-4.279280	2.310144	-0.764406
C	-3.458558	-1.621362	-1.835489
H	-2.694590	-2.365218	-2.144358
H	-3.560236	-0.898379	-2.671926
H	-4.426878	-2.166045	-1.765756
C	-3.218438	1.856684	1.998962
H	-4.172117	1.789245	2.569945
H	-3.289874	2.767873	1.369393
H	-2.412703	2.035291	2.741223

5. Bader Atoms in Molecules (AIM) Analysis

The Bader atoms in molecules (AIM) analysis was conducted on the geometry optimized complex **7**, using the freely available computer program Multiwfn.⁹ Only the relevant parameters at the (3, -1) bond critical point (BCP) for the Ca-Si bond are shown below.

```
===== CP 155, Type (3,-1) =====  
Position (Bohr): 0.21424610794047 0.12062972200135 0.10406095133013  
Density of all electrons: 0.1746582940E-01  
Density of Alpha electrons: 0.8732914700E-02  
Density of Beta electrons: 0.8732914700E-02  
Spin density of electrons: 0.0000000000E+00  
Lagrangian kinetic energy G(r): 0.1406783355E-01  
Hamiltonian kinetic energy K(r): -0.2345620396E-02  
Potential energy density V(r): -0.1172221315E-01  
Energy density (H): 0.2345620396E-02  
Laplacian of electron density: 0.6565381578E-01  
Electron localization function (ELF): 0.5437688397E-01  
Localized orbital locator (LOL): 0.1935287937E+00  
Local information entropy: 0.6078567283E-03  
Reduced density gradient (RDG): 0.4225128518E-15  
Sign(lambda2)*rho: -0.1746582940E-01  
Corr. hole for alpha, ref.: 0.00000 0.00000 0.00000 : -0.8305892971E-02  
Source function, ref.: 0.00000 0.00000 0.00000 : -0.1956867271E-01  
Wavefunction value for orbital 1 : 0.2667098916E-04  
Average local ionization energy: 0.5483134849E+00  
User defined real space function: 0.1000000000E+01  
ESP from nuclear charges: 0.4605267389E+02  
ESP from electrons: -0.4579494080E+02
```

Total ESP: 0.2577330870E+00 a.u. (0.7013274E+01 J/C, 0.1617172E+03 kcal/mol)

Note: Below information are for electron density

Components of gradient in x/y/z are:

-0.1153997687E-16 0.2281906760E-17 0.1401839528E-17

Norm of gradient is: 0.1184665855E-16

Components of Laplacian in x/y/z are:

0.5317552493E-01 0.3885135376E-02 0.8593155482E-02

Total: 0.6565381578E-01

Hessian matrix:

0.5317552493E-01 -0.3225423983E-01 0.3679313903E-01

-0.3225423983E-01 0.3885135376E-02 -0.1903850472E-01

0.3679313903E-01 -0.1903850472E-01 0.8593155482E-02

Eigenvalues of Hessian: $\lambda_3 = 0.9015781911E-01$ $\lambda_2 = -0.1295523378E-01$
 $\lambda_1 = -0.1154876955E-01$

Eigenvectors(columns) of Hessian:

0.7977608581E+00 -0.5369751486E-01 -0.6005782131E+00

-0.3981780537E+00 0.7010536496E+00 -0.5915893997E+00

0.4528044288E+00 0.7110839312E+00 0.5378919892E+00

Determinant of Hessian: 0.1348914326D-04

Ellipticity of electron density: 0.121785

eta index: 0.143695

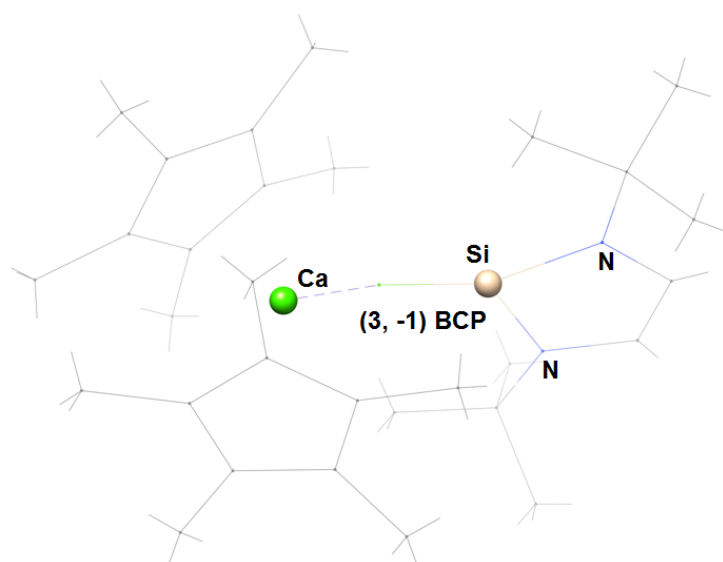


Fig. S18 Graphical representation of the (3,-1) BCP between the calcium and Si centres in compound **7**. (All other BCPs in the molecule have been removed for clarity)

6. References

- 1 TURBOMOLE V6.4 2012, a development of University of Karlsruhe and Forschungszentrum Karlsruhe GmbH, 1989-2007; TURBOMOLE GmbH, since 2007; available from <http://www.turbomole.com>
- 2 S. Grimme, J. Antony, S. Ehrlich, H. Krieg, *J. Chem. Phys.* 2010, **132**, 154104.
- 3 A. Schaefer, H. Horn, R. Ahlrichs, *J. Chem. Phys.* 1992, **97**, 2571.
- 4 (a) M. Sierka, A. Hogeckamp, R. Ahlrichs, *J. Chem. Phys.* 2003, **118**, 9136-9148; (b) K. Eichkorn, F. Weigend, O. Treutler, R. Ahlrichs, *Theor. Chem. Acc.* 1997, **97**, 119-124; (c) F. Weigend, *Phys. Chem. Chem. Phys.* 2006, **8**, 1057-1065; (d) A. Hellweg, C. Hättig, S. Höfener, W. Klopper, *Theor. Chem. Acc.* 2007, **117**, 587-597.
- 5 J. H. Schaftenaar, J. Noordik, *J. Comp.-Aided Mol. Des.* 2000, **14**, 123.
- 6 J. E. Carpenter, F. Weinhold, *J. Mol. Struct. (Theochem)* 1988, **169**, 41-62.
- 7 K. B. Wiberg, *Tetrahedron* 1968, **24**, 1083-1096.
- 8 Gaussian 09, Revision B.01, M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H. P. Hratchian, A. F. Izmaylov, J. Bloino, G. Zheng, J. L. Sonnenberg, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J. A. Montgomery, Jr., J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, T. Keith, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, N. Rega, J. M. Millam, M. Klene, J. E. Knox, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R. L. Martin, K. Morokuma, V. G. Zakrzewski, G. A. Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, O. Farkas, J. B. Foresman, J. V. Ortiz, J. Cioslowski, D. J. Fox, Gaussian, Inc., Wallingford CT, 2010.
- 9 T. Lu, F. Chen, *J. Comput. Chem.* 2012, **33**, 580.