

Reaction Pathways for the Formation of $\text{Cu}_2\text{ZnSn}(\text{Se,S})_4$ Absorber Materials from Liquid-Phase Hydrazine-based Precursor Inks

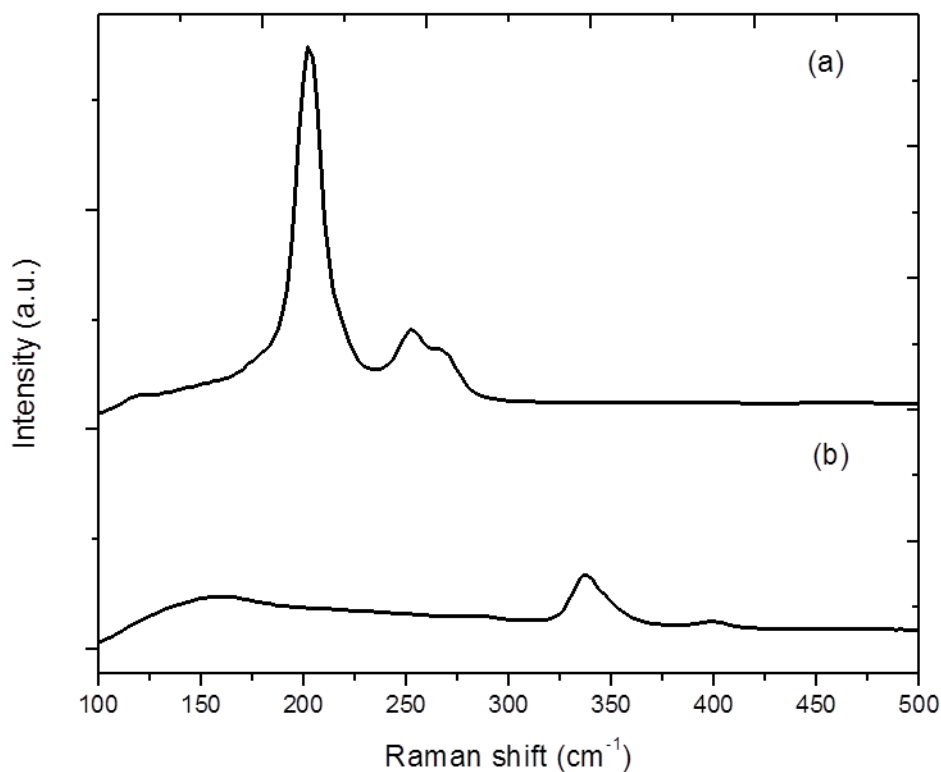
Wan-Ching Hsu ^{a,b}, Brion Bob ^{a,b}, Wenbing Yang ^{a,b}, Choong-Heui Chung ^{a,b}, Yang Yang ^{a,b}

^a California NanoSystems Institute, 570 Westwood Plaza, Los Angeles, CA 90095, United States

^b Department of Materials Science and Engineering, University of California Los Angeles, 410 Westwood Plaza, Los Angeles, CA 90095, United States

Tel (+)310-825-4052; E-mail: yangy@ucla.edu

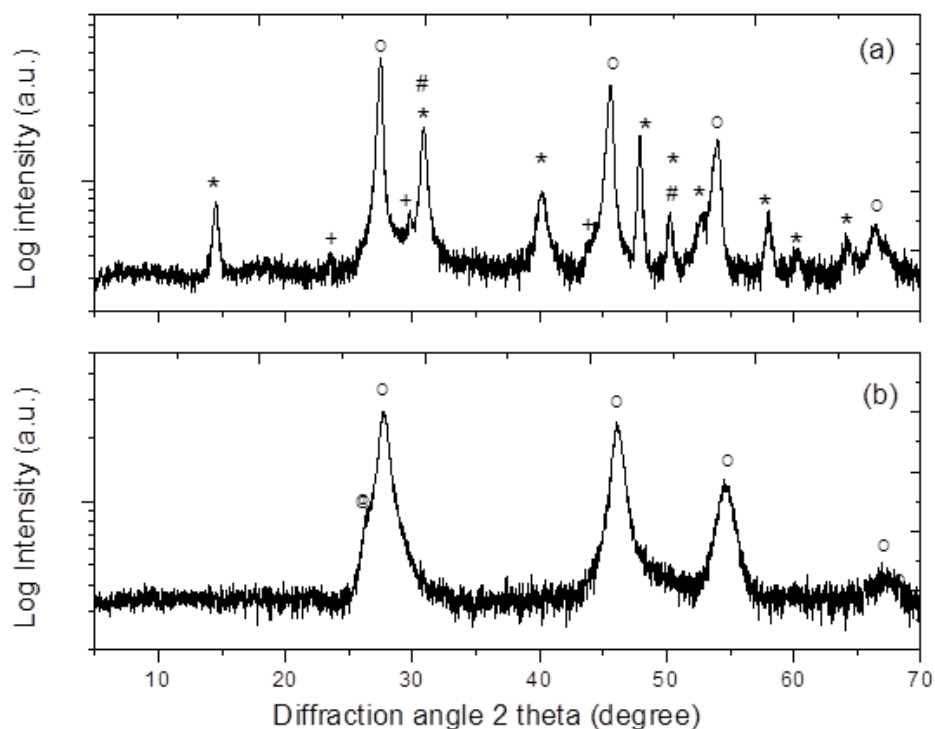
Electronic Supplementary Information



S. 1.

Raman spectra of (a) SnSe_2 -Se solution with $\text{SnSe}_2 : \text{Se} = 1 : 1$ and (b) Cu_2S -S solution with $\text{Cu}_2\text{S} : \text{S} = 1 :$

2.



S. 2.

Powder XRD pattern of the powder prepared by (a) drying $\text{Cu}_2\text{S-S}$ and $\text{SnSe}_2\text{-Se}$ mixed solution annealing at 200°C , and (b) filtering and drying CZTSe precursor solution. $\text{Cu}_2\text{Sn}(\text{Se,S})_3$ or kesterite is marked by “o” (reference code 01-089-2879); SnSe_2 is marked by “*” (reference code 00-023-0602); CuSe is marked by “#” (reference code 00-049-1457); Se is marked by “+” (reference code 00-042-1425); hexagonal ZnX is marked by “@”