

Sulfated Tungstate : A New Solid Heterogeneous Catalyst For Amide Synthesis

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SUPPORTING INFORMATION

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A) Experimental

a) Regents:

$\text{Na}_2\text{WO}_4 \cdot 2\text{H}_2\text{O}$ was purchased from West Coast Laboratories, India. Other chemicals were purchased from Spectrochem Pvt. Ltd. India and were used without further purification.

b) Catalyst preparation:

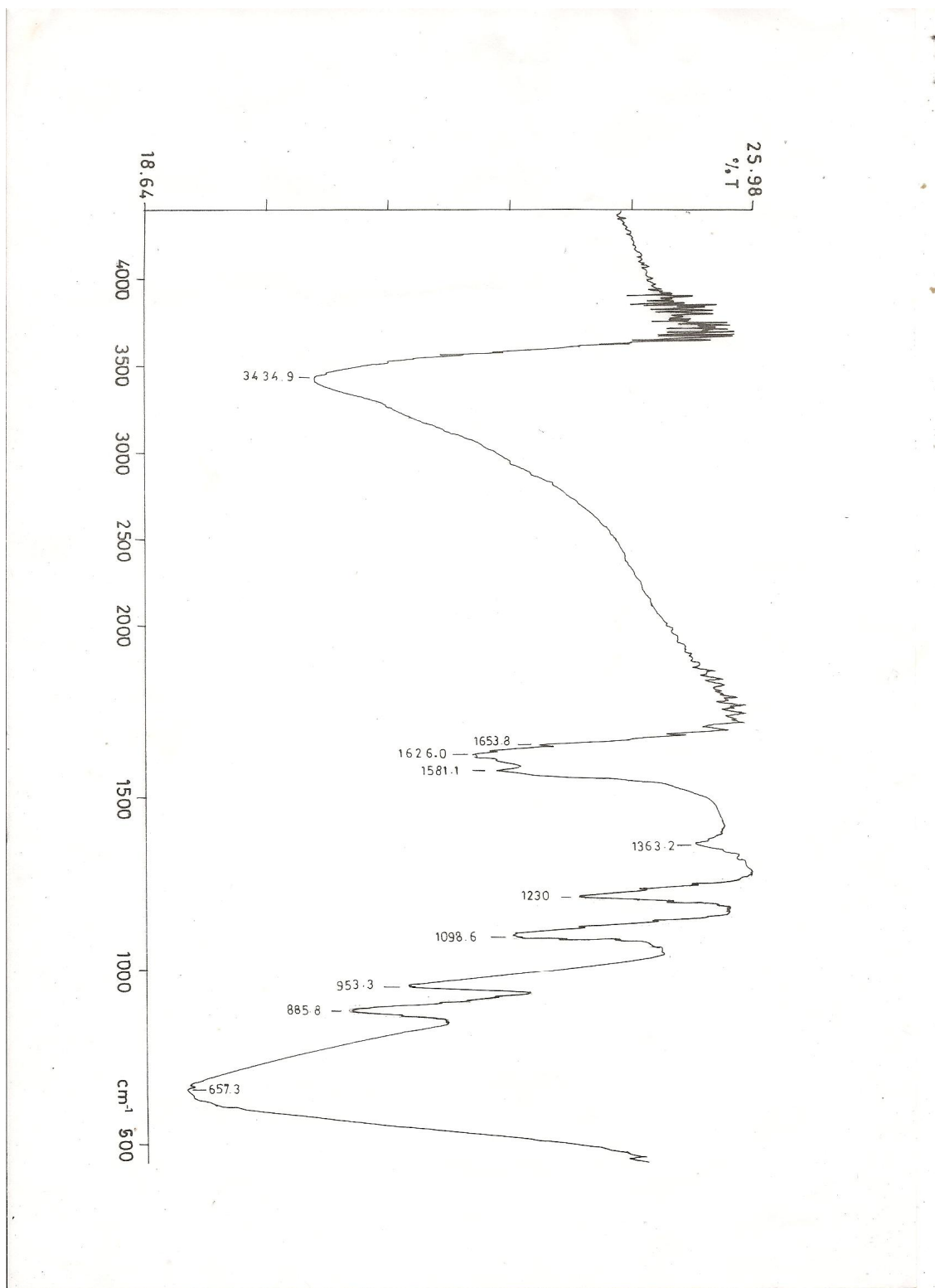
The catalyst was prepared by adding anhydrous sodium tungstate (0.1 mol) gradually to a stirred solution of chlorosulfonic acid (0.2 mol) in chloroform (150ml) contained in a 250ml round bottom flask placed in an ice bath. After completion of addition the mixture was stirred further for 1 h. A yellowish-white solid was obtained. It was filtered and repeatedly washed with deionized water until a neutral filtrate was obtained. The absence of chlorine ion was detected by AgNO_3 test. Then catalyst was dried in an oven for 2 h at 100 °C.

c) General procedure for amide synthesis:

To a solution of benzoic acid (2g, 16.39 mmol) in toluene (20ml) was added the catalyst (0.64g, 18 % (w/w)). To this mixture was added benzylamine (1.59g, 14.9 mmol) in one portion. The reaction mixture was refluxed for 12 hour and water was collected azeotropically in the Dean-Stark trap. After 12 hour the reaction mixture was allowed to cool at about 50-60 °C and filtered through a Buchner funnel and catalyst was recovered by washing with acetone and water. The filtrate was distilled off under reduced pressure to remove toluene. The residue obtained was dissolved in ethyl acetate (30ml) and wash with 10% (w/v) NaHCO_3 (10ml) and 5% (w/v) HCl (10ml). The organic layer was dried over Na_2SO_4 and concentrated under reduced pressure to afford amide as a white solid with purity of 98% (HPLC)

B) Characterization data of Catalyst:

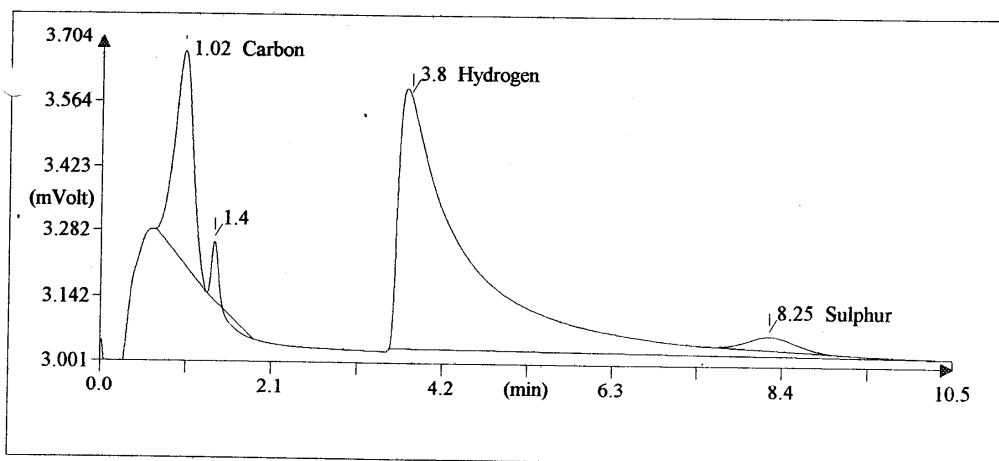
1. IR of Catalyst of the catalyst :



3) Elemental Analysis the catalyst :

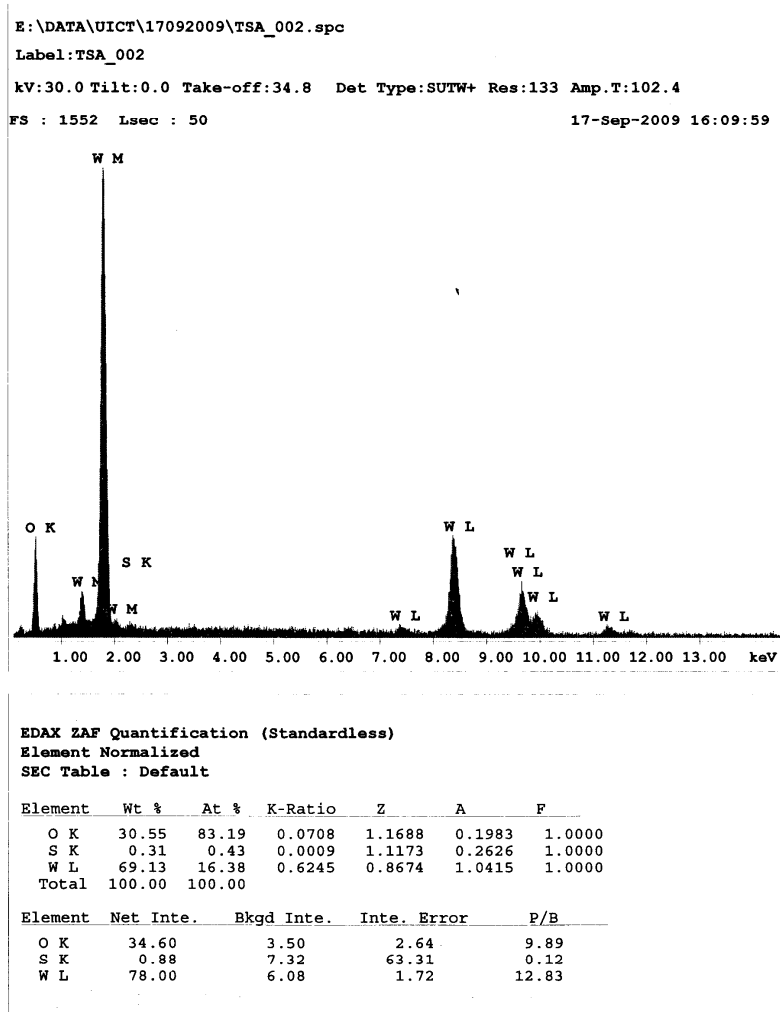
Title: SAIF-IIT, Powai, Mumbai

Operator ID: SMD
Company name: ThermoFinnigan
Method filename: C:\Eager 300 for EA1112\data\Sys_data_example\chn-2009\14-09-2009-CHNS.mth
Method name: NCHS
Analysed: 09/14/2009 19:04
Printed: 09-16-2009 11:16
Elemental Analyser method:
Sampler method:
Sample ID: 14-09-2009-035-P-12-CHN-97 (# 35)
Analysis type: UnkNown
Chromatogram filename: 14-09-2009-035-P-12-CHN-97.dat
Calibration method: K Factors
Sample weight: 2.281
Protein factor: 6.25



Peak Number (#)	Retention Time (min)	Area (.1* μ V*sec)	Element %	Compon
1	1.017	73336	0.000	Carbon
2	1.400	3723	0.000	
3	3.800	335642	0.825	Hydrog Sulphu
4	8.250	11806	0.294	
		424507	1.119	

4) EDAX Report of the catalyst



5) Particle size analysis report

Size Statistics Report by Intensity

v2.0



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Sample Details

Sample Name: Sawant Sir 1

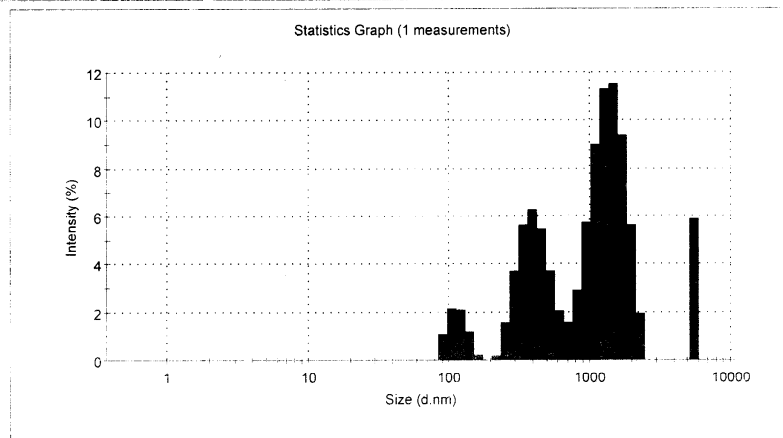
File Name: AATV_Nanosuspension fine tuned.dts

SOP Name: Atovaquone Nanosuspension.sop

Measurement Date and Time: Friday, September 18, 2009 3:39:24 PM

Z-Average (nm): 1428.832 Derived Count Rate (kcps): 2663791.63214...
Standard Deviation: 0 Standard Deviation: 0
%Std Deviation: 0 %Std Deviation: 0
Variance: 0 Variance: 0

Size d.nm	Mean Intensity %	Std Dev Intensity %	Size d.nm	Mean Intensity %	Std Dev Intensity %	Size d.nm	Mean Intensity %	Std Dev Intensity %	Size d.nm	Mean Intensity %	Std Dev Intensity %
0.4000	0.0		5.615	0.0		78.82	0.0		1106	9.0	
0.4632	0.0		6.503	0.0		91.28	1.1		1281	11.3	
0.5365	0.0		7.531	0.0		105.7	2.1		1494	11.5	
0.6213	0.0		8.721	0.0		122.4	2.1		1718	9.4	
0.7195	0.0		10.10	0.0		141.8	1.2		1990	5.6	
0.8332	0.0		11.70	0.0		164.2	0.2		2305	2.0	
0.9649	0.0		13.54	0.0		190.1	0.0		2669	0.0	
1.117	0.0		15.69	0.0		220.2	0.2		3091	0.0	
1.294	0.0		18.17	0.0		255.0	1.5		3580	0.0	
1.499	0.0		21.04	0.0		295.3	3.7		4145	0.0	
1.736	0.0		24.36	0.0		342.0	5.6		4801	0.0	
2.010	0.0		28.21	0.0		396.1	6.3		5560	5.9	
2.328	0.0		32.67	0.0		458.7	5.5		6439	0.0	
2.696	0.0		37.84	0.0		531.2	3.7		7456	0.0	
3.122	0.0		43.82	0.0		615.1	2.0		8635	0.0	
3.615	0.0		50.75	0.0		712.4	1.6		1.000e4	0.0	
4.187	0.0		58.77	0.0		825.0	2.9				
4.849	0.0		68.06	0.0		955.4	5.7				



Size Distribution Report by Intensity

v2.0



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Sample Details

Sample Name: Sawant Sir 1

SOP Name: Atovaquone Nanosuspension.sop

General Notes:

File Name: AATV_Nanosuspension fi...

Dispersant Name: Water

Record Number: 2895

Dispersant RI: 1.330

Material RI: 1.53

Viscosity (cP): 0.8872

Material Absorbion: 0.01

Measurement Date and Time: Friday, September 18, 2009 3...

System

Temperature (°C): 25.0

Duration Used (s): 50

Count Rate (kcps): 326.4

Measurement Position (mm): 0.65

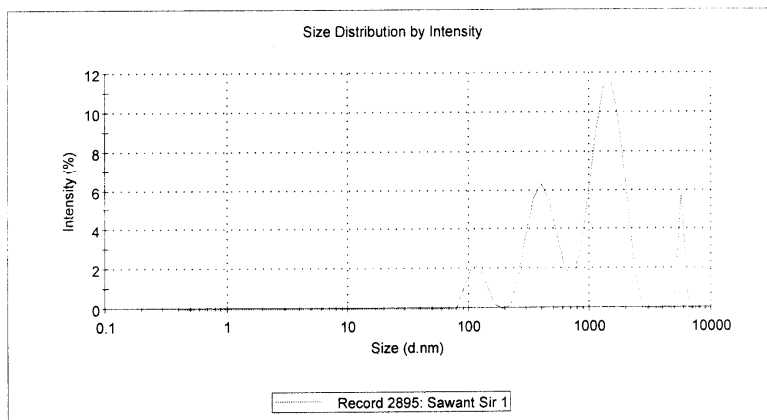
Cell Description: Disposable sizing cuvette

Attenuator: 3

Results

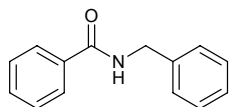
		Diam. (nm)	% Intensity	Width (nm)
Z-Average (d.nm): 1429	Peak 1:	1396	58.1	376.4
PdI: 0.877	Peak 2:	424.4	29.6	116.5
Intercept: 0.807	Peak 3:	116.8	6.6	18.24

Result quality : Refer to quality report



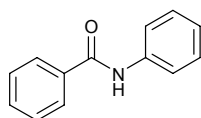
C) Amide Analytical Data

Entry 1) *N*-Benzylbenzamide



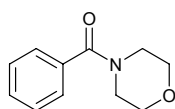
m.p. 129-130 °C (lit.,¹ 128-130 °C); $\nu_{\max}(\text{KBr})/\text{cm}^{-1}$ 3445 (NH), 1655 (C=O); $\delta_{\text{H}}(300 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 4.65 (2 H, d, $J=6$), 6.42 (1 H, br s), 7.25-7.56 (8 H, m) and 7.79 (2 H, m); $\delta_{\text{C}}(75 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 44.0, 127.1, 127.5, 127.6, 127.8, 128.5, 128.6, 134.3, 138.1 and 167.3.

Entry 2) *N*-Phenylbenzamide



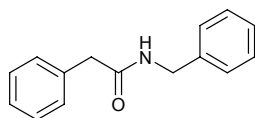
m.p. 165-166 °C (lit.,² 162 °C); $\nu_{\max}(\text{KBr})/\text{cm}^{-1}$ 3343 (NH) and 1653 (C=O); $\delta_{\text{H}}(300 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 7.13-7.86 (10 H, m); $\delta_{\text{C}}(75 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 165.79, 137.88, 134.95, 131.79, 129.05, 128.73, 127.00, 124.53 and 120.20.

Entry 3) Morpholino (phenyl)methanone



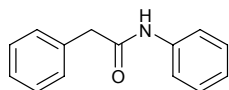
m.p. 72 °C (lit.,³ 72-74 °C); $\nu_{\max}(\text{KBr})/\text{cm}^{-1}$ 1660 (C=O); $\delta_{\text{H}}(300 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 3.34 (4 H, t), 1.5 (6 H, m), 7.95 (2 H, m), 7.44 (2 H, m) and 7.51 (1 H, m); $\delta_{\text{C}}(75 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 46.1, 66.8, 127.2, 128.6, 129.2, 135.5 and 168.9.

Entry 4) *N*-Benzyl-2-phenylacetamide



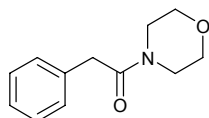
m.p. 117 °C (lit.,⁴ 117-119 °C); $\nu_{\max}(\text{KBr})/\text{cm}^{-1}$ 3271 (NH) and 1630 (C=O); $\delta_{\text{H}}(300 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 3.62 (2 H, s), 4.4 (2 H, d, $J=6\text{Hz}$), 5.9 (1 H, br s) and 7.15-7.55 (10 H, m); $\delta_{\text{C}}(75 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 43.4, 43.6, 127.2, 127.3, 127.4, 128.5, 128.9, 129.3, 134.8, 138.1 and 170.9.

Entry 5) 2,N-Diphenylacetamide



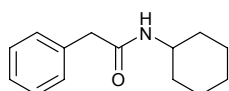
m.p. 118-119 °C (lit.,⁵ 118 °C); $\nu_{\max}(\text{KBr})/\text{cm}^{-1}$ 3254 (NH) and 1655 (C=O); $\delta_{\text{H}}(300 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 7.25-7.39 (10 H, m), 3.49 (2 H, s) and 6.2 (1 H, br s); $\delta_{\text{C}}(75 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 169.12, 137.85, 134.41, 129.47, 129.16, 128.89, 127.61, 124.41, 119.81 and 44.76.

Entry 6) 4-(phenylacetyl)morpholine



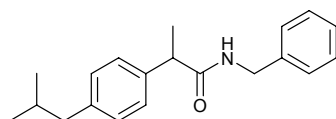
m.p. 62-64 °C (lit.,⁶ 64 °C); $\nu_{\max}(\text{KBr})/\text{cm}^{-1}$ 1655 (C=O); $\delta_{\text{H}}(300 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 7.31 (2 H, t), 7.21-7.25 (3H, m), 3.72 (2 H, m), 3.63 (4 H, m) and 3.40-3.47 (4 H, m); $\delta_{\text{C}}(75 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 45.6, 66.3, 38.6, 135.7, 129.7, 129.3, 127.7 and 160.5.

Entry 7) N-Cyclohexyl-2-phenylacetamide



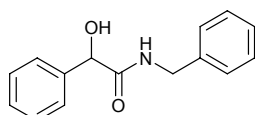
m.p. 124-128 °C (lit.,⁷ 125-126 °C); $\nu_{\max}(\text{KBr})/\text{cm}^{-1}$ 3345 (NH), 1645 (C=O); $\delta_{\text{H}}(300 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 0.92-1.14 (3 H, m), 1.23-1.36 (2 H, m), 1.48-1.63 (3 H, m), 1.78-1.84 (2 H, m), 3.52 (2 H, s), 3.65-3.76 (1 H, m), 5.22 (1 H, br s) and 7.2-7.38 (5 H, m); $\delta_{\text{C}}(75 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 24.6, 25.4, 32.9, 44.0, 48.1, 127.2, 128.9, 129.3, 135.1 and 169.9.

Entry 8) N-Benzyl-2-(4-isobutylphenyl)propanamide



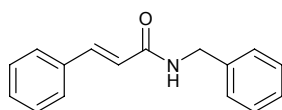
m.p. Yellow Oil (lit.,⁸ Yellow Oil); $\nu_{\max}(\text{Neat})/\text{cm}^{-1}$ 3280 (NH) and 1646 (C=O); $\delta_{\text{H}}(300 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 0.89 (6 H, d, $J=6\text{Hz}$), 1.53 (3 H, d, $J=6\text{Hz}$), 1.84 (1 H, m), 2.5 (3 H, d, $J=8\text{Hz}$), 3.64 (1 H, quartet, $J=8\text{Hz}$), 4.34 (2 H, d, $J=6\text{Hz}$), 6.4 (1 H, br s) and 7.05-7.4 (9 H, m); $\delta_{\text{C}}(75 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 18.27, 22.24, 30.01, 43.26, 44.86, 46.61, 127.28, 127.4, 127.43, 128.59, 129.64, 138.54, 140.69 and 174.77.

Entry 9) N-Benzyl-2-hydroxy-2-phenylethanamide



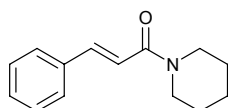
m.p. 96-99 °C (lit.,⁹ 96 °C); $\nu_{\max}(\text{KBr})/\text{cm}^{-1}$ 3250 (NH), 1641 (C=O); $\delta_{\text{H}}(300 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 3.6-3.65 (1 H, br s, OH exchangeable with D₂O), 4.43 (2 H, d, $J=5.78$), 5.07 (1 H, d, $J=3.49$), 6.52 (1 H, br s) and 7.16-7.43 (10 H, m); $\delta_{\text{C}}(75 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 43.06, 43.14, 73.93, 74.0, 126.59, 127.44, 127.48, 128.33, 128.55, 128.59, 137.63, 139.52 and 172.54.

Entry 10) (E)-N-Benzylcinnamamide



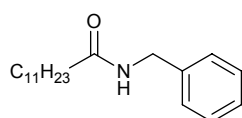
m.p. 106-108 °C (lit.,¹⁰ 108 °C); $\nu_{\max}(\text{KBr})/\text{cm}^{-1}$ 3263 (NH), 1652 (C=O) and 1615 (C=C); $\delta_{\text{H}}(300 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 7.67 (1 H, d, $J=15.3$), 7.47 (2 H, m), 7.32 (8 H, m), 6.44 (1 H, d, $J=15.3$), 6.21 (1 H, s) and 4.54 (2 H, d, $J=4.8$); $\delta_{\text{C}}(75 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 165.8, 141.3, 138.2, 134.7, 129.6, 128.7, 128.6, 127.9, 127.8, 127.5, 120.4 and 43.8.

Entry 11) (E)-3-phenyl-1-(piperidine-1-yl)prop-2-en-1-one



m.p. 110-112 °C (lit.,¹¹ 114 °C); $\nu_{\max}(\text{KBr})/\text{cm}^{-1}$ 1620 (C=C), 1645 (C=O); $\delta_{\text{H}}(300 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 1.30-1.78 (6 H, m), 3.32-3.65 (4 H, m), 6.74 (1 H, d, $J=16$), 7.01-7.4 (5 H, m) and 7.48 (1 H, d, $J=16$); $\delta_{\text{C}}(75 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 44.8, 25.4, 25.6, 166.4, 144, 118.9, 135.2, 126.4, 128.7 and 128.

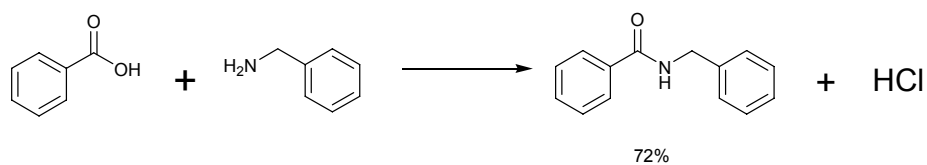
Entry 12) N-Benzyl laurylamide



m.p. 82-83 °C (lit.,¹² 84 °C); ν_{max} (KBr)/cm⁻¹ 3250 (NH), 1641 (C=O); δ_{H} (300 MHz; CDCl₃; Me₄Si) 0.88 (3 H, t), 1.51 (2 H, sextet), 1.21 (16 H, m), 1.41 (2 H, t), 3.85(2 H, d) and 7.15-7.33 (5 H, m); δ_{C} (75 MHz; CDCl₃; Me₄Si) 14.13, 22.69, 25.78, 29.24, 29.62, 25.9, 35.8, 175.0, 57.4, 142.10, 128.2, 128.3 and 126.7.

D) Green Metrics

- 1) Calculations for the synthesis of *N*-Benzylbenzamide via thionyl chloride, producing an acid chloride.¹³



Input		Output	
Benzoyl Chloride	3.5g	Crude <i>N</i> -benzylbenzamide	3.8g
Benzyl amine	2.7g		
Acetone	26.4 (33ml)	Aqueous waste	62g
NaHCO ₃ (aq)	12g		
Brine (aq)	50g	Organic Solvent Waste	2.7g
Total	94.6g	Total	64.7

$$\text{E-Factor,} \quad \left(\frac{64.7 \text{ g of waste produced}}{3.8 \text{ g of crude product}} \right) = 17$$

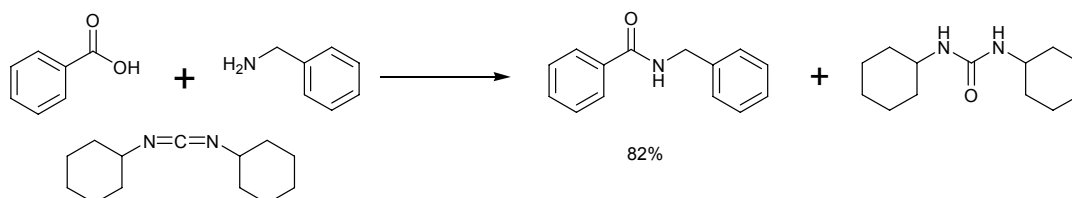
$$\text{Mass Intensity,} \quad \left(\frac{94.6 \text{ g of raw material used}}{3.8 \text{ g of crude product}} \right) = 25$$

$$\text{Atom Economy,} \quad \left(\frac{211}{140.5 + 107} \right) 100 = 85.2\%$$

Assumptions

1. 90% of organic solvents are recovered.
2. The formation of acyl chloride and use of thionyl chloride is not accounted for in calculations.

2) Calculations for the synthesis of *N*-Benzylbenzamide using DCC as activating agent.¹⁴



Input		Output	
Benzoic acid	1.5g	Crude <i>N</i> -benzylbenzamide	2g
Benzyl amine	1.3g		
DCC	2.7g	Aqueous waste	36g
DCM	92.7g (70ml)		
HCl (aq)	18g	Organic Solvent Waste	9.3g
KHCO ₃ (aq)	18g		

Total	134.2g	Total	45.3g
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$$\text{E-Factor, } \left(\frac{45.3 \text{ g of waste produced}}{2 \text{ g of crude product}} \right) = 22.6$$

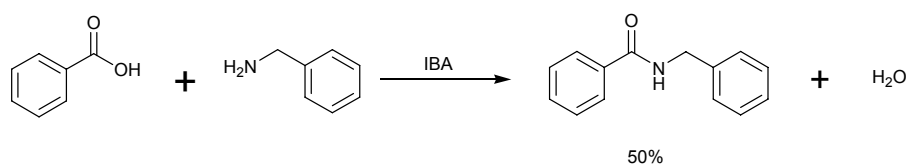
$$\text{Mass Intensity, } \left(\frac{134.2 \text{ g of raw material used}}{2 \text{ g of crude product}} \right) = 67$$

$$\text{Atom Economy, } \left(\frac{211}{206 + 107 + 122} \right) 100 = 48.5$$

Assumptions

1. 90% of organic solvents were recovered.
2. Calculations did not take into account of recrystallisation of the product.
3. Calculations did not account for the synthesis of DCC.

3) Calculation for the synthesis of *N*-Benzylbenzamide using ortho-*N,N*-Diisopropylbenzylaminoboronic acid (IBA).¹⁵



Input		Output	
Benzoic acid	0.6g	Crude <i>N</i> -benzylbenzamide	0.53g
Benzyl amine	0.54g		
IBA	0.12g		
Fluorobenzene	51.2g (50ml)		

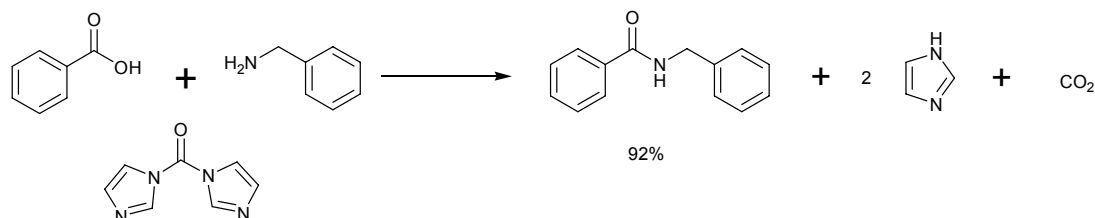
DCM	33.1g (25ml)	Aqueous waste	125g
Brine (aq)	25g		
HCl (aq)	25g		
Brine (aq)	25g		
NaOH (aq)	25g	Organic Solvent Waste	8.4g
Brine (aq)	25g		
Total	210.6g	Total	133.4g

$$\text{E-Factor, } \left(\frac{133.4\text{g of waste produced}}{0.5\text{g of crude product}} \right) = 251.7$$

$$\text{Mass Intensity, } \left(\frac{210.5\text{g of raw material used}}{0.53\text{g of crude product}} \right) = 397.3$$

$$\text{Atom Economy, } \left(\frac{211}{107 + 122} \right) 100 = 92.13\%$$

1. Calculations did not account for the synthesis of catalyst.
2. 90%recovery of organic solvents.
- 4) Calculations for the synthesis of *N*-Benzylbenzamide using *N,N'*-Carbonyldiimidazole as an activating agent.¹⁶



Input		Output	
Benzoic acid	1.2g	Crude <i>N</i> -	1.9g

Benzyl amine	1g	benzylbenzamide	
THF	17.7g (20ml)	Aqueous waste	60g
CDI	1.62g		
NaOH (aq)	10g	Organic Solvent Waste	1.7g
HCl (aq)	50g		
Total	63.9g	Total	61.7g

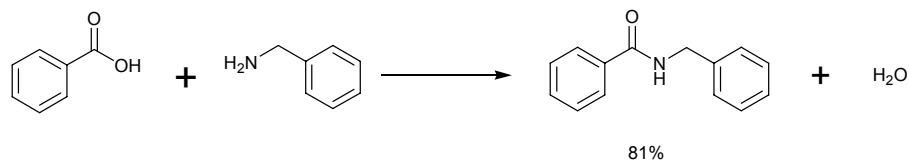
$$\text{E-Factor, } \left(\frac{61.7\text{g of waste produced}}{1.9\text{g of crude product}} \right) = 31.8$$

$$\text{Mass Intensity, } \left(\frac{63.9\text{g of raw material used}}{1.9\text{g of crude product}} \right) = 32.9$$

$$\text{Atom Economy, } \left(\frac{211}{122 + 107 + 162} \right) 100 = 53.9\%$$

Assumptions

1. 90% of organic solvents were recovered.
2. Calculations did not take into account of recrystallisation of the product.
3. Calculations did not account for the synthesis of CDI.
- 5) Calculations for the synthesis of *N*-Benzylbenzamide using Sulfated Tungstate (Catalyst).



Input		Output	
Benzoic acid	2g	Crude <i>N</i> -	2.8g

Benzyl amine	1.7g	benzylbenzamide	
Catalyst	0.64g	Aqueous waste	4.4g
Toluene	17.2g (20ml)		
EtOAc	27g (30ml)	Organic Solvent Waste	20
NaHCO ₃ (aq)	10g		
HCl (aq)	10g		
Total	68.5g	Total	24.4g

$$\text{E-Factor, } \left(\frac{24.4\text{g of waste produced}}{2.8\text{g of crude product}} \right) = 8.7$$

$$\text{Mass Intensity, } \left(\frac{68.5\text{g of raw material used}}{2.8\text{g of crude product}} \right) = 24.4$$

$$\text{Atom Economy, } \left(\frac{211}{122 + 107} \right) 100 = 92.13\%$$

Assumptions

1. 90% recovery of organic solvents.
2. Calculations did not account for the synthesis of the catalyst.

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