## **Electronic Supplementary Information**

## Suzuki-Miyaura Cross Coupling Reaction from a Low-Leaching and Highly Recyclable Gold-Supported Palladium Material and Two Types of Microwave Equipment

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#### General

<sup>1</sup>H-NMR spectra were recorded in CDCl<sub>3</sub> at 25 °C unless otherwise noted, at 400 or 500 MHz, with TMS as an internal standard. <sup>13</sup>C-NMR spectra were recorded in CDCl<sub>3</sub> at 25 °C unless otherwise noted, at 100 or 125 MHz. Mass spectra were obtained using a JEOL JMS-700TZ. ICP-mass spectra were obtained on an AGILRNT ICP-MS 7500CS. HPLC was carried out using a Mighysil RP-18 (Kanto Chemical Co. Inc) and HPLC spectra were detected by a Shimadzu SPD-10AV (254 nm) and recorded by a Shimadzu CR-8A. Column chromatography were performed with silica gel 60N (spherical, neutral, 63-210 mm, Kanto Chemical Co., Inc.) unless otherwise stated. Microwave instrument were performed with EYELA MWO-1000S and EYELA NWS-1000 for single-mode and multi-mode, respectively.

#### Preparation of Sulfur-modified Au-supported Pd material SAPd

To a stirring ice cooled 98% H<sub>2</sub>SO<sub>4</sub> (4.7 g) was added Na<sub>2</sub>S<sub>2</sub>O<sub>8</sub> (4.0 g) in a small portions and then the crushed ice (13.0 g) and water (4.0 g) were added to the above solution while the temperature was kept below 15 °C. When all the salt dissolved to a homogeneous solution, the Au (100 mesh-14 × 12 mm<sup>2</sup>, 100.7 mg) was placed in the above solution (3.0 mL) for 5 min and then washed first by H<sub>2</sub>O (3.0 mL × 10) and then with EtOH (3.0 mL × 6). The resulted Au-mesh was placed in a round bottom flask and dried for 10 min under reduced pressure (*ca*. 6 mm Hg). The resulting sulfur-modified Au was placed in a solution of Pd(OAc)<sub>2</sub> (5.3 mg, 0.023 mmol) in xylene (3.0 mL × 50) and, after vacuum drying, it was placed in xylene (3.0 mL) and heated at 135 °C for 12 h. Finally, it was rinsed with xylene (3.0 mL × 50) and dried under vacuum for 10 min to give sulfur-modified Au-supported Pd material SAPd (100.8 mg, immobilized Pd: 57 ± 16 µg) and only this SAPd was used throughout this research.

## Typical experimental procedure of microwave assisted flowing for Suzuki-Miyaura coupling using iodobenzene 1a catalyzed by SAPd;

To a reaction tube (suitable for microwave), iodobenzene (**1a**, 102.0 mg, 0.50 mmol), SAPd and EtOH (2.0 mL) were added and then the tube was fixed in the single-mode microwave chamber. The tube was irradiated in a single-mode microwave conditions, MW (S): temp: 80 °C, time: 60 min, power: 200 W and after completing the irradiation, the SAPd was removed from the tube, washed with xylene ( $6 \times 3.0$  mL) and kept for next cycle. The irradiated solution was transferred to reaction vessel, which was previously charged with 4-chlorophenylboronic acid (**2a**, 117.3 mg, 0.75 mmol) and K<sub>2</sub>CO<sub>3</sub> (138.2 mg, 1.0 mmol). The SAPd containing tube was washed with EtOH ( $2 \times 1$  mL) and the washing solvents were also poured into the above reaction vessel. The resulting reaction vessel was then irradiated in

a multi-mode microwave conditions, MW (M): temp: 82 °C, time: 60 min, power: 500 W. The reaction mixture was then cooled to room temperature and was poured into water (10.0 mL) and the organic layer was extracted with AcOEt (3×20 mL). The combined organic extracts were washed with brine (3×25 mL) and dried over Na<sub>2</sub>SO<sub>4</sub>. Concentration at reduced pressure gave yellowish oil, which was chromatographed on silicagel with hexane–AcOEt (98:2 v/v) as eluent to give the 4-chlorobiphenyl (**3a**, 93.6 mg, 99%) as a white solid. The above reaction condition was maintained as for 2<sup>nd</sup> cycle and this procedure was repeated for a total 10 cycles.

# Typical experimental procedure of microwave assisted flowing for Suzuki-Miyaura coupling using bromobenzene 1b catalyzed by SAPd;

To a reaction tube (suitable for microwave), bromobenzene (1b, 78.5 mg, 0.50 mmol), SAPd and DMF (2.0 mL) were added and then the tube was fixed in the single-mode microwave chamber. The tube was irradiated in a single-mode microwave conditions, MW (S): temp: 90 °C, time: 50 min, power: 300 W and after completing the irradiation, the SAPd was removed from the tube, washed with xylene (6  $\times$  3.0 mL) and kept for next cycle. The irradiated solution was transferred to reaction vessel, which was previously charged with 4-chlorophenylboronic acid (2a, 117.3 mg, 0.75 mmol) and K<sub>2</sub>CO<sub>3</sub> (138.2 mg, 1.0 mmol). The SAPd containing tube was washed with toluene  $(2 \times 1 \text{ mL})$  and the washing solvents along with H<sub>2</sub>O (1.0 mL) were poured into the above reaction vessel. The resulting reaction vessel was then irradiated in a multi-mode microwave conditions, MW (M): temp: 104 °C, time: 60 min, power: 500 W. The reaction mixture was then cooled to room temperature and was poured into water (10.0 mL) and the organic layer was extracted with AcOEt ( $3 \times 20.0$ mL). The combined organic extracts were washed with brine  $(3 \times 25.0 \text{ mL})$  and dried over Na<sub>2</sub>SO<sub>4</sub>. Concentration at reduced pressure gave yellowish oil, which was chromatographed on silicagel with hexane-AcOEt (98:2 v/v) as eluent to give the 4-chlorobiphenyl (3a, 93.8 mg, 99%) as a white solid. The above reaction condition was maintained as for 2<sup>nd</sup> cycle and this procedure was repeated for a total 10 cycles.

Amount of leached-Pd (ng) in the reaction mixtures <sup><math>c, d, e</math></sup>										Pd on SAPd		
											(µg)	
Reactions	$1^{st}$	$2^{nd}$	3 <sup>rd</sup>	$4^{\text{th}}$	$5^{th}$	$6^{th}$	$7^{\rm th}$	$8^{th}$	$9^{\text{th}}$	$10^{\text{th}}$	Before	After
											use	use
1a + 2a	$170 \pm$	$144 \pm$	$107 \pm$	$120 \pm$	$103 \pm$	$65 \pm$	$86 \pm$	$58 \pm$	$56 \pm$	$62 \pm$	81	93
	181	143	84	113	53	23	62	22	24	26	± 27	$\pm 30$
	(0.05)	(0.04)	(0.03)	(0.03)	(0.03)	(0.02)	(0.02)	(0.02)	(0.02)	(0.02)		
1b + 2a	$487 \pm$	1016±	1321±	$667 \pm$	$862 \pm$	1010±	723±	711±	533±	360±	81	49
	153	821	970	274	445	709	496	602	367	204	± 23	± 13
	(0.11)	(0.22)	(0.29)	(0.15)	(0.19)	(0.22)	(0.16)	(0.16)	(0.12)	(0.08)		

**Table S1.** Amount of Pd in the reaction mixtures of **1a** with  $2a^{a}$ , **1b** with  $2a^{b}$  and on SAPd itself

<sup>*a*</sup> Reaction conditions: **1a** (0.50 mmol), **2a** (0.75 mmol), K<sub>2</sub>CO<sub>3</sub> (1.00 mmol), single-mode microwave settings, MW (S): solvent: EtOH (2.0 mL), temp: 80 °C, time: 60 min, power: 200 W; multi-mode microwave settings, MW (M): solvent: EtOH (2.0 mL), temp: 82 °C, time: 60 min, power: 500 W. <sup>*b*</sup> **1b** (0.50 mmol), **2a** (0.75 mmol), K<sub>2</sub>CO<sub>3</sub> (1.00 mmol), single-mode microwave settings, MW (S): solvent: DMF (2.0 mL), temp: 90 °C, time: 50 min, power: 300 W; multi-mode microwave settings, MW (M): solvent: toluene/H<sub>2</sub>O (3:1, 2.0 mL), temp: 104 °C, time: 60 min, power: 500 W. <sup>*c*</sup> The entire reaction mixture was acidified and subjected directly to ICP-MS measurement. <sup>*d*</sup> The standard deviation was calculated from 4 sets of samples. <sup>*e*</sup> Numbers in parentheses indicate the amount of leached Pd in ppm.

٨٣	SAPd, solve	nt (2 mL)	tion		(	HO) <sub>2</sub> B	–Ar′	2a-	-2c			• ′	
Al-		) Solu	ιοπ <u>–</u> K <sub>2</sub>	CO <sub>3</sub> , s	olven	t (2 mL	.) + M	W (S	) solv	ent	Aı	r—Ar	
1c	-1j		MW (M)									3b-3h	
Entry		Ar'-B(OH) <sub>2</sub>		Yields of $3(\%)^b$									Average
-	Ar-X	Ar'	$1^{st}$	2 <sup>nd</sup>	3 <sup>rd</sup>	4 <sup>th</sup>	5 <sup>th</sup>	6 <sup>th</sup>	7 <sup>th</sup>	8 <sup>th</sup>	9 <sup>th</sup>	$10^{\text{th}}$	yields %)
1	A-MeOC, H, Br	4-ClC2H4	98	99	98	99	98	97	98	98	97	98	98
1	( <b>1</b> c)	( <b>2</b> 9)	70	))	70	"	70	)1	70	70	)1	70	70
2	$(\mathbf{IC})$ 4-MeC <sub>6</sub> H <sub>4</sub> Br	(2a) 4-ClC <sub>6</sub> H <sub>4</sub>	95	93	95	94	93	95	95	95	93	94	94
	( <b>1d</b> )	( <b>2a</b> )											
3	$4-NO_2C_6H_4Br$	$4-ClC_6H_4$	98	99	99	>99	98	98	99	99	99	98	99
	( <b>1e</b> )	( <b>2a</b> )											
4	4-CNC <sub>6</sub> H <sub>4</sub> Br	$4-ClC_6H_4$	99	>99	98	99	97	98	99	99	98	99	99
	( <b>1f</b> )	( <b>2a</b> )											
5	$C_6H_5Br$	$C_6H_5$	97	98	97	97	96	97	93	95	94	95	96
	( <b>1b</b> )	( <b>2b</b> )											
6	$C_6H_5Br$	$4-MeC_6H_4$	99	98	96	98	94	98	96	99	98	97	98
	( <b>1b</b> )	( <b>2c</b> )											
7	4-MeOC <sub>6</sub> H <sub>4</sub> I	$4-ClC_6H_4$	97	98	98	97	96	96	97	97	98	97	97
	( <b>1g</b> )	( <b>2a</b> )											
8	$2-MeC_6H_4I$	$4-ClC_6H_4$	96	95	95	97	95	96	95	95	95	95	95
	( <b>1h</b> )	( <b>2a</b> )											
9	$4-AcC_6H_4I$	$4-ClC_6H_4$	>99	>99	98	99	99	98	99	97	98	99	99
	( <b>1i</b> )	( <b>2a</b> )											
10	$4-NO_2C_6H_4I$	$4-ClC_6H_4$	99	>99	99	95	98	98	99	99	98	99	99
	( <b>1j</b> )	( <b>2a</b> )											
11	$C_6H_5I$	$C_6H_5$	95	95	96	96	96	95	96	96	95	95	96
	( <b>1a</b> )	( <b>2b</b> )											
12	C <sub>6</sub> H <sub>5</sub> I	$4-\text{MeC}_6\text{H}_4$	96	97	96	96	96	95	96	94	95	96	96
	( <b>1a</b> )	( <b>2c</b> )											

Table S2. Suzuki-Miyaura cross-coupling reactions of various substrates using SAPd<sup>a</sup>

<sup>*a*</sup> Reaction conditions for arylbromides: arylbromides (0.50 mmol), boronic acids (0.75 mmol),  $K_2CO_3$  (1 mmol), single-mode microwave settings, MW (S): solvent: DMF (2.0 mL), temp: 90 °C, time: 50 min, power: 300 W, multi-mode microwave settings, MW (M): solvent: toluene/H<sub>2</sub>O (3:1, 2 mL), temp: 102 °C, time: 60 min, power: 500 W; Reaction conditions for aryliodides: aryliodides (0.50 mmol), boronic acids (0.75 mmol),  $K_2CO_3$  (1 mmol),

single-mode microwave settings, MW (S): solvent: EtOH (2.0 mL), temp: 80 °C, time: 60 min, power: 200 W, multi-mode microwave settings, MW (M): solvent: EtOH (2.0 mL), temp: 82 °C, time: 60 min, power: 500 W. <sup>*b*</sup> The isolated yields.

## 4-Chlorobiphenyl (3a)<sup>1, 2</sup>



From arylbromide method: average yield: 99%.

From aryliodide method: average yield: 99%.

White solid; mp 78-78.5 °C (MeOH) (lit.<sup>2</sup> 78.5 °C, benzene). <sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>):  $\delta$  7.54 (2H, d, J = 7.5 Hz), 7.51 (2H, d, J = 7.5 Hz), 7.45–7.39 (4H, m), 7.37–7.34 (1H, m); <sup>13</sup>C–NMR (125 MHz, CDCl<sub>3</sub>):  $\delta$  139.94, 139.62, 133.33, 128.88, 128.86, 128.36, 127.56, 126.95; LRMS (EI) *m*/*z* 188 (100%, M<sup>+</sup>).

#### 4-Chloro-4'-methoxybiphenyl (3b)<sup>1,2</sup>



From arylbromide method: By following the same procedure described for **3a**, biaryl **3b** was prepared from 4-bromoanisole (**1c**) and 4-chlorophenylboronic acid (**2a**): average yield: 98%. From aryliodide method: By following the same procedure described for **3a**, biaryl **3b** was prepared from 4-iodoanisole (**1g**) and 4-chlorophenylboronic acid (**2a**): average yield: 97%. White solid; mp 115-115.5 °C (lit.<sup>2</sup> 116 °C, EtOH). <sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>):  $\delta$  7.46 (4H, t, *J* = 8.5 Hz), 6.96 (2H, d, *J* = 8.5 Hz), 3.83 (3H, s); <sup>13</sup>C–NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  159.32, 139.22, 132.62, 132.43, 128.80, 127.97, 127.90, 114.27, 55.32; LRMS (EI) *m/z* 218 (100%, M<sup>+</sup>).

#### 4-Chloro-4'-methylbiphenyl (3c)<sup>3,4</sup>



By following the same procedure described for **3a**, biaryl **3c** was prepared from 4-bromotoluene (**1d**) and 4-chlorophenylboronic acid (**2a**): average yield: 94%.

White solid. mp 122-122.5 °C (lit. <sup>4</sup> 122 °C). <sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>):  $\delta$  7.49 (2H, d, J = 8.2 Hz), 7.44 (2H, d, J = 8.2 Hz), 7.38 (2H, d, J = 8.4 Hz), 7.24 (2H, d, J = 8.0 Hz), 2.39 (3H, s); <sup>13</sup>C–NMR (125 MHz, CDCl<sub>3</sub>):  $\delta$  139.55, 137.42, 137.07, 132.99, 129.58, 129.02, 128.82, 128.20, 126.79, 21.10; LRMS (EI) *m*/*z* 202 (90%, M<sup>+</sup>).

4-Chloro-4'-nitrobiphenyl (3d)<sup>2,5</sup>



From arylbromide method: By following the same procedure described for **3a**, biaryl **3d** was prepared from 4-bromonitrobenzene (**1e**) and 4-chlorophenylboronic acid (**2a**): average yield: 99%.

From aryliodide method: By following the same procedure described for **3a**, biaryl **3d** was prepared from 4-iodonitrobenzene (**1j**) and 4-chlorophenylboronic acid (**2a**): average yield: 99%.

Pale yellow solid. mp 145.5-146 °C (lit. <sup>2</sup> 146 °C, EtOH). <sup>1</sup>H-NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  8.29 (2H, d, J = 8.4 Hz), 7.70 (2H, d, J = 8.4 Hz), 7.56 (2H, d, J = 8.4 Hz), 7.46 (2H, t, J = 8.4 Hz); <sup>13</sup>C–NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  147.14, 146.22, 137.10, 135.17, 129.31, 128.57, 127.60, 124.15; LRMS (EI) *m/z* 233 (100%, M<sup>+</sup>).

## 4'-Chlorobiphenyl-4-carbonitrile (3e)<sup>6,7</sup>



By following the same procedure described for **3a**, biaryl **3e** was prepared from 4-bromobenzonitrile (**1f**) and 4-chlorophenylboronic acid (**2a**): average yield: 99%.

White solid. mp 125-126 °C (lit. <sup>7</sup> 124-126 °C). <sup>1</sup>H-NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  7.72 (2H, d, J = 8.0 Hz), 7.64 (2H, d, J = 7.6 Hz), 7.52 (2H, d, J = 8.4 Hz), 7.44 (2H, d, J = 8.4 Hz); <sup>13</sup>C–NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  144.24, 137.46, 134.84, 132.60, 129.22, 128.39, 127.47, 118.69, 111.14; LRMS (EI) *m*/*z* 213 (100%, M<sup>+</sup>).

Biphenyl (3f)<sup>1,8</sup>



From arylbromide method: By following the same procedure described for **3a**, biaryl **3f** was prepared from bromobenzene (**1b**) and phenylboronic acid (**2b**): average yield: 96%.

From aryliodide method: By following the same procedure described for **3a**, biaryl **3f** was prepared from iodobenzene (**1a**) and phenylboronic acid (**2b**): average yield: 96%.

White solid. mp 70.0-70.5 °C (lit. <sup>8</sup> 70-71 °C, MeOH). <sup>1</sup>H-NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  7.59 (4H, d, J = 7.6 Hz), 7.43 (4H, t, J = 7.6 Hz), 7.35–7.31(2H, m); <sup>13</sup>C–NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  141.20, 128.73, 127.22, 127.14; LRMS (EI) *m*/*z* 154 (100%, M<sup>+</sup>).

## 4-Methylbiphenyl (3g)<sup>1,9</sup>



From arylbromide method: By following the same procedure described for **3a**, biaryl **3g** was prepared from bromobenzene (**1b**) and 4-methylphenylboronic acid (**2c**): average yield: 98%. From aryliodide method: By following the same procedure described for **3a**, biaryl **3g** was prepared from iodobenzene (**1a**) and 4-methylphenylboronic acid (**2c**): average yield: 96%. White solid. mp 48.5-49 °C (lit. <sup>9</sup> 49-50 °C, EtOH). <sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>):  $\delta$  7.58 (2H, d, *J* = 8.6 Hz), 7.49 (2H, d, *J* = 8.0 Hz), 7.44–7.41(2H, m), 7.33–7.31(1H, m), 7.26–7.24 (2H, m), 2.39 (1H, s); <sup>13</sup>C–NMR (125 MHz, CDCl<sub>3</sub>):  $\delta$  141.14, 138.33, 137.01, 129.46, 128.70, 126.98, 126.96, 21.10; LRMS (EI) *m/z* 168 (100%, M<sup>+</sup>).

## 4'-chloro-2-methylbiphenyl (3h)<sup>10</sup>



By following the same procedure described for **3a**, biaryl **3h** was prepared from *o*-iodotoluene (**1h**) and 4-chlorophenylboronic acid (**2a**): average yield: 95%.

Light yellow oil. <sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>):  $\delta$  7.36 (2H, d, J = 8.5 Hz), 7.25–7.20 (5H, m), 7.17 (1H, d, J = 7.0 Hz), 2.24 (3H, s); <sup>13</sup>C–NMR (125 MHz, CDCl<sub>3</sub>):  $\delta$  140.61, 140.30, 135.21, 132.78, 130.48, 130.40, 129.62, 128.23, 127.55, 125.86, 20.37; LRMS (EI) *m/z* 202 (100%, M<sup>+</sup>).

## 1-(4'-Chlorobiphenyl-4-yl)ethanone (3i)<sup>3,11</sup>



By following the same procedure described for **3a**, biaryl **3i** was prepared from 4-acetyliodobenzene (**1i**) and 4-chlorophenylboronic acid (**2a**): average yield: 99%.

White solid. mp 104-104.5 °C (lit 103-104 °C, MeOH). <sup>1</sup>H-NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  8.01 (2H, d, *J* = 8.4 Hz), 7.62 (2H, d, *J* = 8.0 Hz), 7.53 (2H, d, *J* = 8.8 Hz), 7.42 (2H, d, *J* = 8.4 Hz), 2.62 (3H, s); <sup>13</sup>C–NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  197.54, 14.31, 138.16, 135.97, 134.35, 129.06, 128.93, 128.42, 126.96, 26.61; LRMS (EI) *m*/*z* 230 (50%, M<sup>+</sup>).

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