

ARTICLE

Supporting Information

**Potassium modified layered $\text{Ln}_2\text{O}_2\text{CO}_3$ (Ln: La, Nd, Sm, Eu) materials:
Efficient and stable heterogeneous catalysts for biofuel production**

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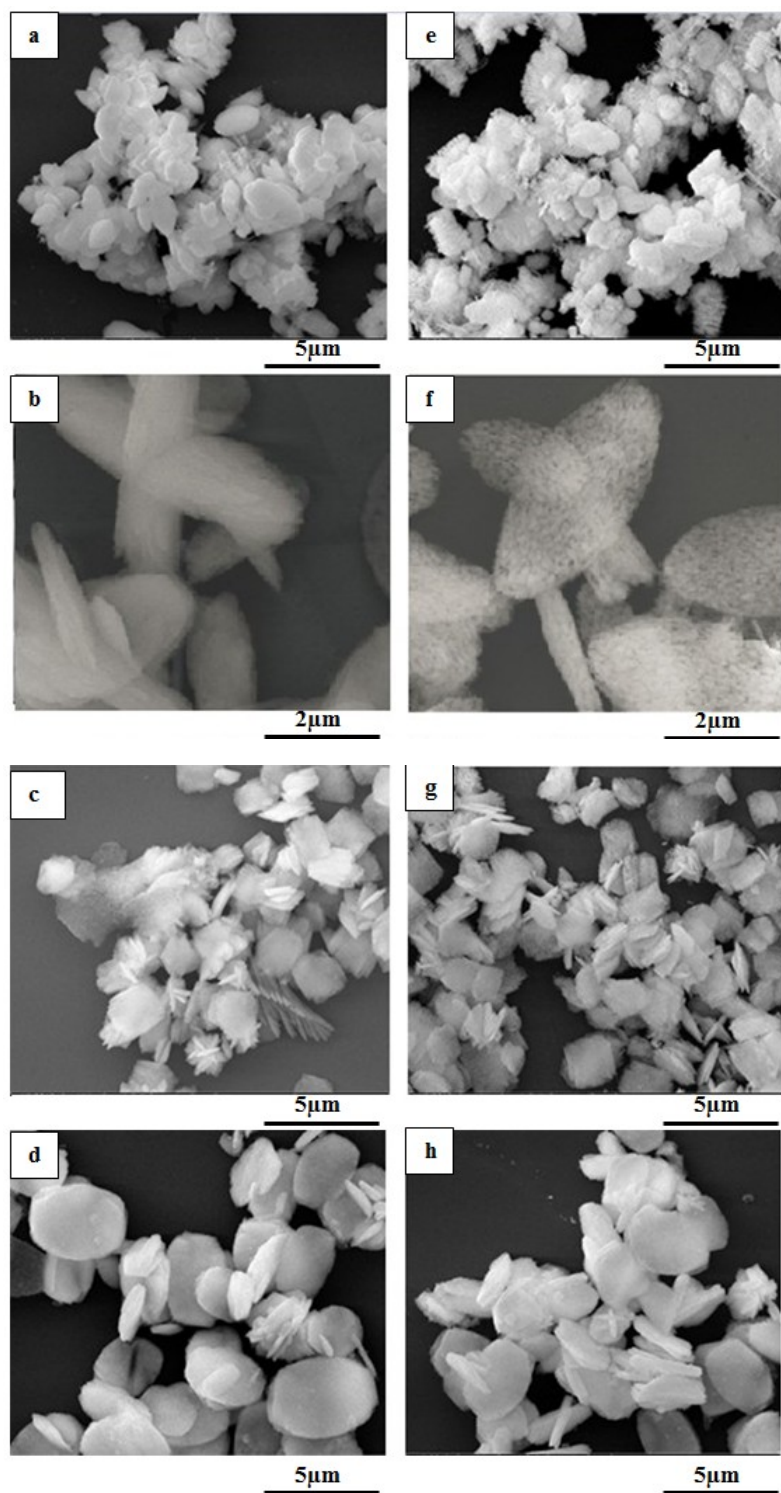


Fig. S1. SEM images of $\text{Ln}_2\text{O}(\text{CO}_3)_2$ (before calcination), Ln: (a) La, (b): Nd, (c) Sm, and (d) Eu, and $\text{Ln}_2\text{O}_2\text{CO}_3$ (after calcination), Ln : (e) La, (f) Nd, (g) Sm, and (h) Eu.

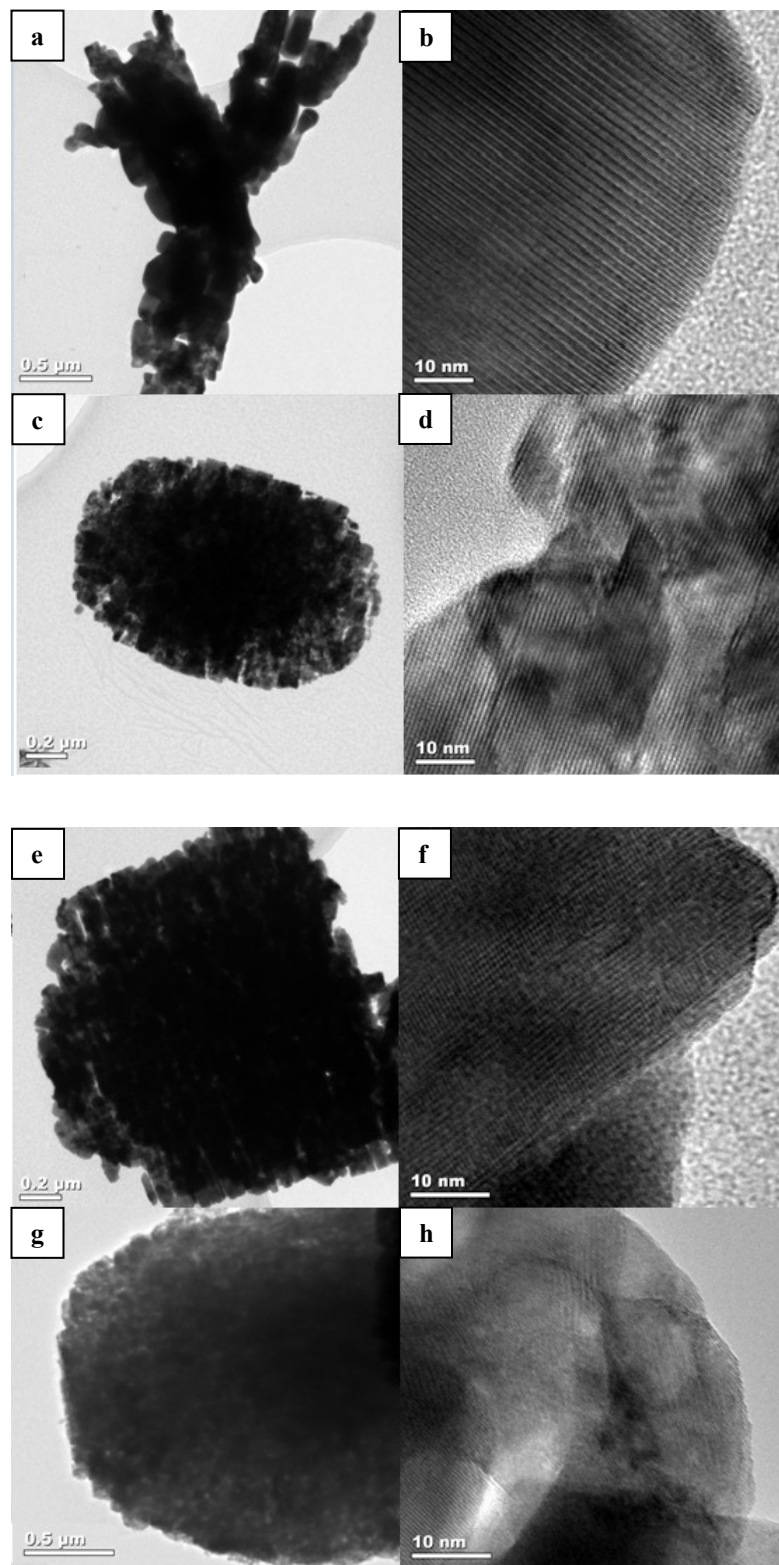


Fig. S2 TEM images of $\text{Ln}_2\text{O}_2\text{CO}_3$, Ln: (a) (b) La, (c) (d) Nd, (e) (f) Sm, and (g) (h) Eu.

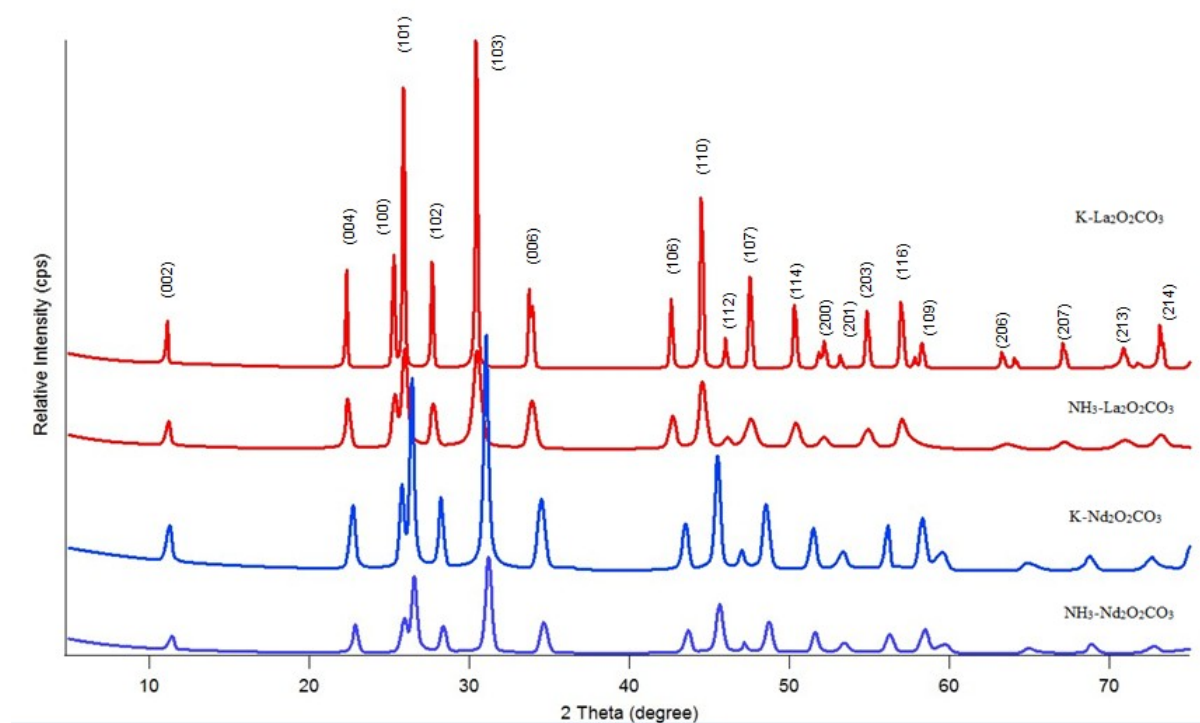


Fig. S3 XRD patterns of synthesized layered $\text{Ln}_2\text{O}_2\text{CO}_3$ (Ln: La and Nd) materials by using NH_3 and $(\text{NH}_4)_2\text{CO}_3$ (labeled as $\text{NH}_3\text{-Ln}_2\text{O}_2\text{CO}_3$) and KOH and K_2CO_3 (labeled as $\text{K-Ln}_2\text{O}_2\text{CO}_3$) as basic starting materials.

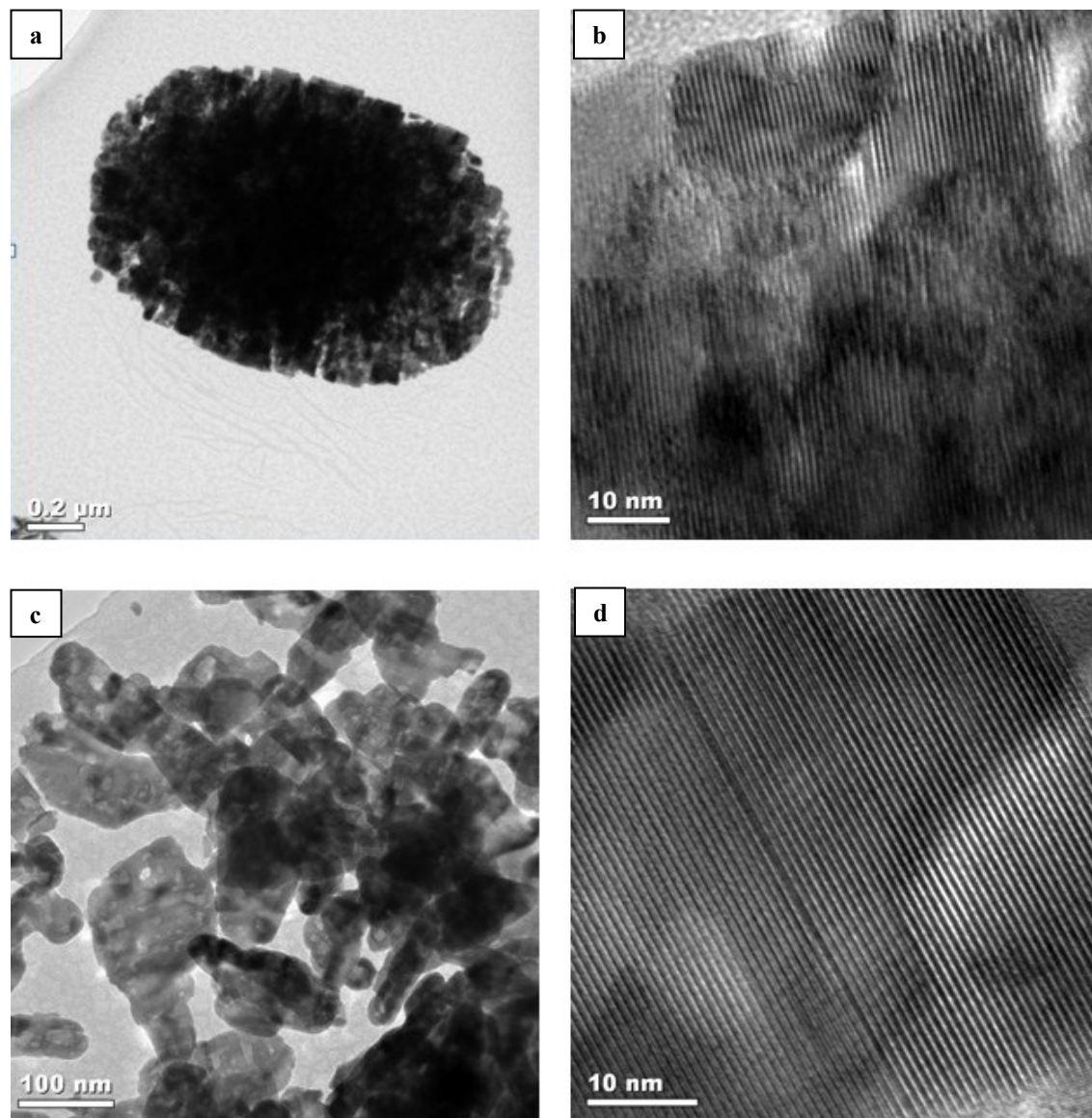


Fig. S4 TEM pictures of (a), (b) potassium contained base synthesized $\text{Nd}_2\text{O}_2\text{CO}_3$ material and (c), (d) ammonium contained base synthesized $\text{Nd}_2\text{O}_2\text{CO}_3$ material.

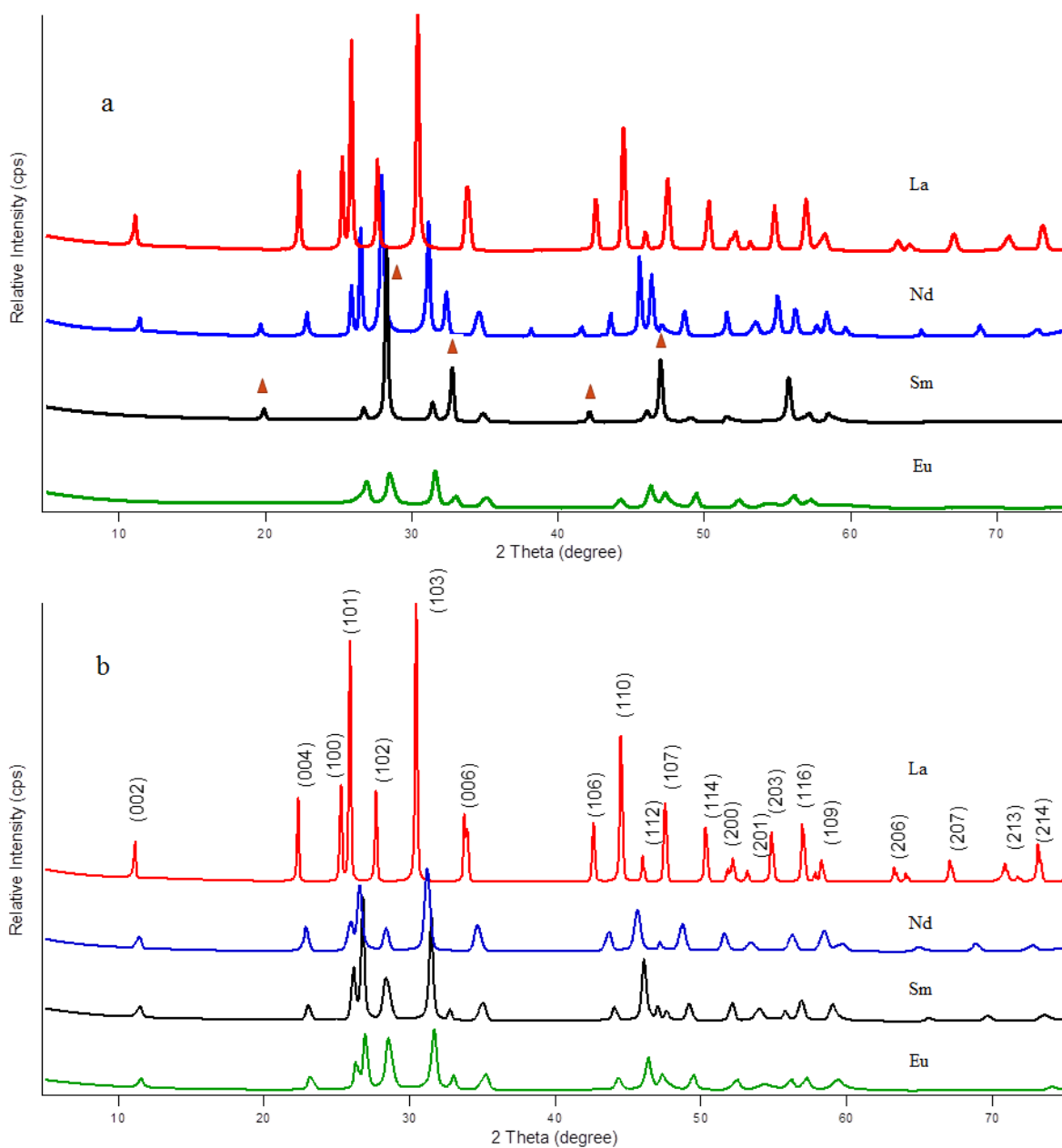


Fig. S5 XRD patterns of (a) synthesized $\text{Ln}_2\text{O}(\text{CO}_3)_2$ ($\text{Ln} = \text{La}, \text{Nd}, \text{Sm}, \text{Eu}$) solid materials after 4th cycle of biodiesel reactions, \blacktriangle the peaks from Ln_2O_3 (b) synthesized layered rare earth oxycarbonate materials $\text{Ln}_2\text{O}_2\text{CO}_3$ ($\text{Ln} = \text{La}, \text{Nd}, \text{Sm}, \text{Eu}$) before reactions.

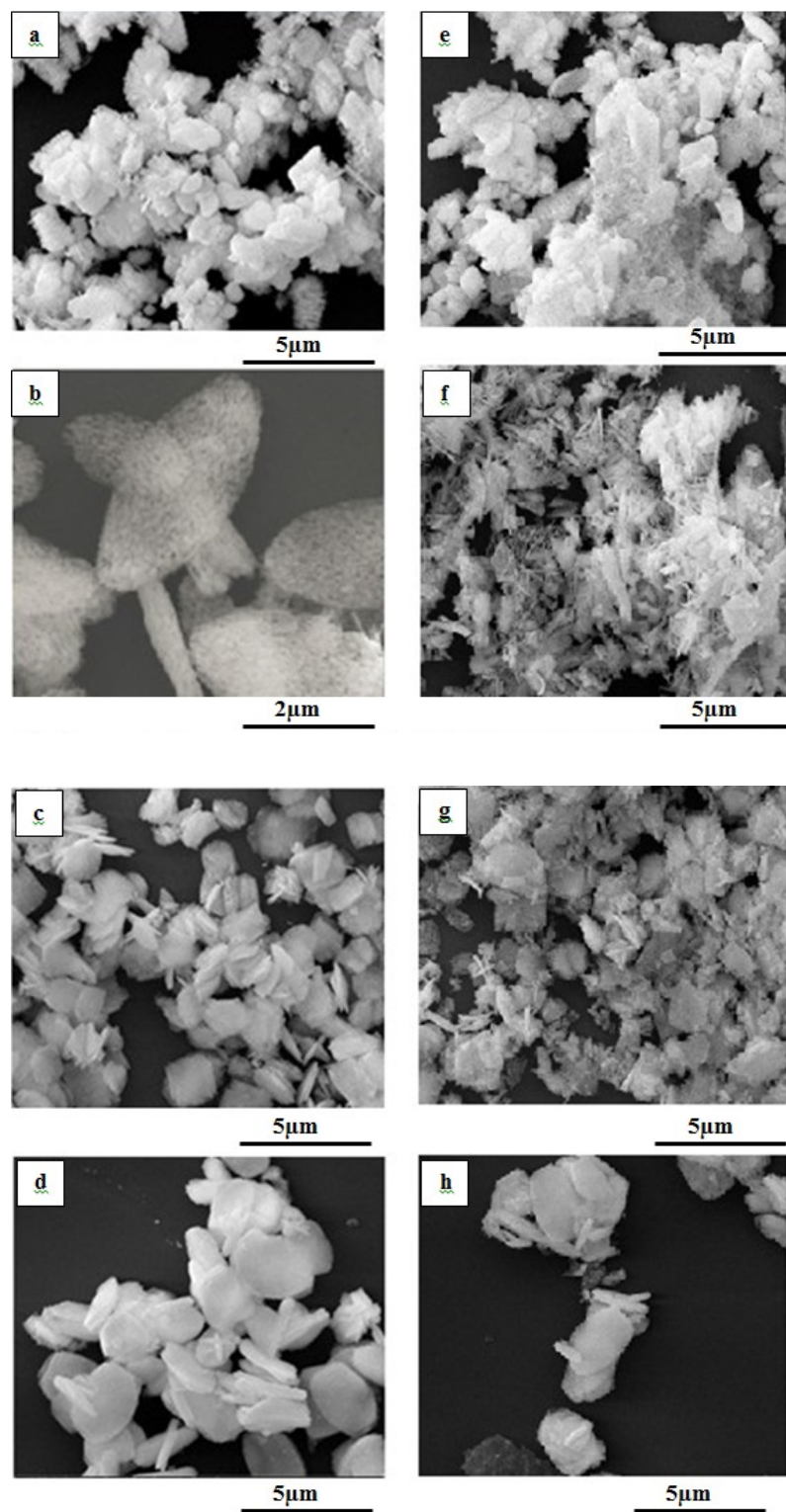


Fig. S6 SEM images of $\text{Ln}_2\text{O}_2\text{CO}$ before reactions (a) La, (b): Nd, (c) Sm, and (d) Eu and after reactions, (e) La, (f) Nd, (g) Sm, and (h) Eu.

$$\text{Yield \% of FAME} = \frac{\text{FAME}}{\text{FAME} + \text{MG} + \text{DG} + \text{TG}} \times 100\%$$

Equation S1 The equation to calculate yield of FAME for biodiesel reactions. (FAME: fatty acid methyl esters; MG: monoglycerides; DG: diglycerides; TG: triglycerides.)

Table S1. BET surface area and basic strength tests by indicators and TPD-CO₂.

Sample	BET Surface Area (g/m ²)	Basic Strength (H ₊)	Basic Sites Amount (< 200 °C mmol/g)
La ₂ O ₂ CO ₃	11	+9.3	0.15
Nd ₂ O ₂ CO ₃	19	+9.3	0.33
Sm ₂ O ₂ CO ₃	15	+9.3	0.15
Eu ₂ O ₂ CO ₃	13	+9.3	0.21

Table S2. Biodiesel yields of potassium and ammonium based $\text{Ln}_2\text{O}_2\text{CO}_3$ materials.

Sample	Starting Base Mixture	FAME Yield
$\text{La}_2\text{O}_2\text{CO}_3$	KOH and K_2CO_3	78 %
	NH_3 and $(\text{NH}_4)_2\text{CO}_3$	24 %
$\text{Nd}_2\text{O}_2\text{CO}_3$	KOH and K_2CO_3	99 %
	NH_3 and $(\text{NH}_4)_2\text{CO}_3$	25 %

Table S3. The potassium amount in fresh and recycled $\text{Ln}_2\text{O}_2\text{CO}_3$ materials.

Sample	K^+ Amount in $\text{Ln}_2\text{O}_2\text{CO}_3$ Materials (wt. %)	K^+ Amount in Recycled $\text{Ln}_2\text{O}_2\text{CO}_3$ Materials (wt. %)	K^+ Amount in $\text{Ln}_2\text{O}_2\text{CO}_3$ Materials (Mole. %)
$\text{La}_2\text{O}_2\text{CO}_3$	1.35	0.77	11.5
$\text{Nd}_2\text{O}_2\text{CO}_3$	1.45	1.12	12.5
$\text{Sm}_2\text{O}_2\text{CO}_3$	1.10	0.47	10.0
$\text{Eu}_2\text{O}_2\text{CO}_3$	1.03	0.42	9.5

Table S4. The potassium and rare earth metal amount in biodiesel.

Sample	K ⁺ Amount in Biodiesel Product (ppm)	Ln ³⁺ Amount in Biodiesel Product (ppm)
La ₂ O ₂ CO ₃	1.4	ND
Nd ₂ O ₂ CO ₃	ND	ND
Sm ₂ O ₂ CO ₃	2.5	ND
Eu ₂ O ₂ CO ₃	4.5	ND

ND: Not Detected