

## Supplementary Material - JAAS Paper Ref. B907122A

### A. Sample analyses and precision calculation:

Sample analysis by Q-ICP-MS was done following standard-sample-standard bracketing protocol. A total of five such replications were done for a single column eluted sample. The  $\delta^7\text{Li}$  of each replicate (individual analysis) was calculated using the following equation;

$$\delta^7\text{Li}(\%) = \left\{ \left[ \frac{\left( \frac{^7\text{Li}}{^6\text{Li}} \right)_{\text{Sample}}}{\left( \frac{^7\text{Li}}{^6\text{Li}} \right)_{\text{Standard-average}}} \right] - 1 \right\} \times 1000$$

Where,  $\left( \frac{^7\text{Li}}{^6\text{Li}} \right)_{\text{Standard-average}}$  = Average  $\left( \frac{^7\text{Li}}{^6\text{Li}} \right)$  Count ratio of L-SVEC Li standard run before and after each sample analysis.

The  $2\sigma$  uncertainty for samples were calculated using the following equation;

$$\sigma = \sqrt{\frac{\sum(x - \bar{x})^2}{n-1}}$$

Where  $n = 5$ ,  $x$  is the  $\delta^7\text{Li}$  value of an individual analysis, and  $\bar{x}$  is the mean of five analyses.

B. Matrix Na tolerance -  $\delta^7\text{Li}$  (‰) of Na doped L-SVEC Li standards in 2% HNO<sub>3</sub>

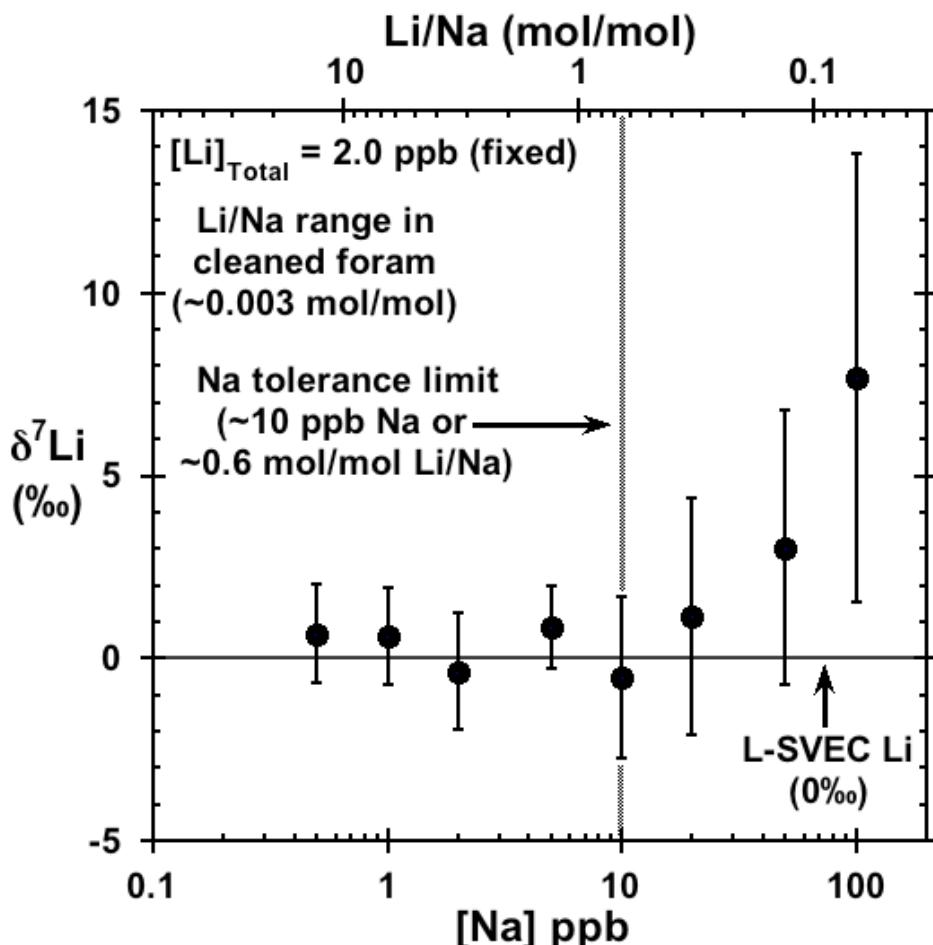


Figure B. Lithium isotope ratios of L-SVEC Li standard ( $[\text{Li}] = 2 \text{ ppb}$ , fixed) doped with increasing concentration of sodium (0.5 ppb to 100 ppb). Li/Na ratios (mol/mol) are plotted on the top Y-axis. The samples were analyzed using standard-sample-standard bracketing technique. Each symbol represents an average of five separate analyses with  $2\sigma$  external variation. The Na tolerance limit of the Q-ICP-MS for matrix-based sodium determined in this study is relatively high ( $> 0.6 \text{ mol/mol}$ , Li/Na). Moreover, tolerance limit of this ICP-MS method for total matrix sodium (10 ppb, Na) is well within the cumulative sodium blank from our chromatographic separation method (see text).

C. Matrix Ca tolerance -  $\delta^7\text{Li}$  (‰) of Ca doped L-SVEC Li standards in 2% HNO<sub>3</sub>

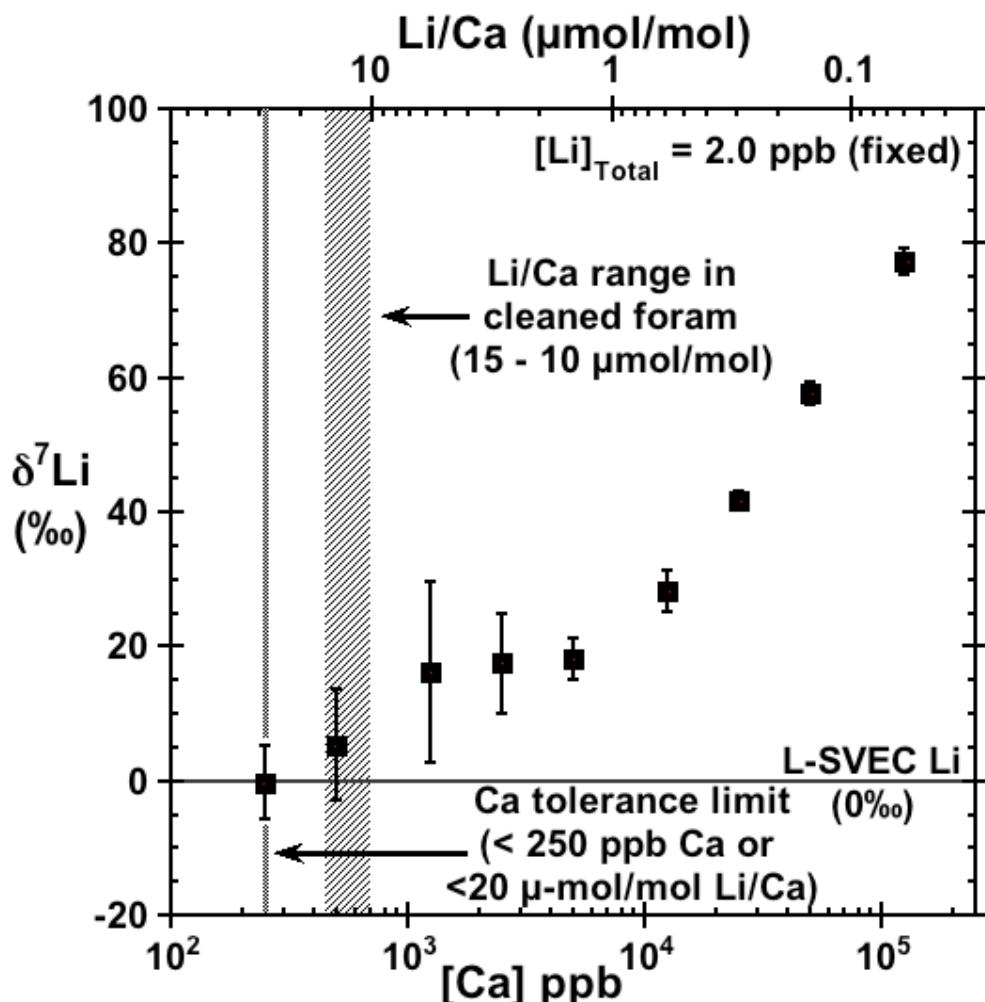


Figure C. Lithium isotope ratios of L-SVEC standard ( $[\text{Li}] = 2 \text{ ppb}$ , fixed) doped with increasing concentration of calcium (250 ppb to  $125 \times 10^3$  ppb). Li/Ca ratios ( $\mu\text{mol/mol}$ ) are plotted on the top Y-axis. The samples were analyzed using standard-sample-standard bracketing technique. Each symbol represents an average of five separate analyses with  $2\sigma$  external error. The tolerance limit of the Q-ICP-MS for matrix-based calcium determined in this study is relatively high (~20  $\mu\text{mol/mol}$ , Li/Ca). Moreover, tolerance

limit of this ICP-MS method for total matrix calcium (250 ppb, Ca) is well within the cumulative calcium blank from our chromatographic separation method (see text).

#### D. Dead-time correction:

Normally, for an Agilent 7500cs mass spectrometer, the detector dead-time is calibrated using In (indium, m/z = 115) or Er (erbium, m/z = 166) as the standard reference mass. The default dead-time value for reference mass 166 amu is 38 ns. However, the dead-time value of the entire mass spectrum (0 amu to 260 amu) can range from about 30 ns to 60 ns depending on detector age. The Agilent operating system uses a mass dependent (6.0 ns/ 260 amu) correction factor to determine the exact dead-time for other masses by applying a simple linear equation.

If the dead-time at reference mass (166 amu) is  $D-T_{166 \text{ amu}}$  ns then the dead time at mass M amu ( $D-T_{M \text{ amu}}$ ) is given by;

$$D-T_{M \text{ amu}} (\text{ns}) = D-T_{166 \text{ amu}} (\text{ns}) + [(6 \text{ ns}) \times (M \text{ amu} - 166 \text{ amu})/260 \text{ amu}]$$

Where, 6 ns/ 260 amu is the correction factor (CF). The correction factor is determined experimentally and is fixed for all Agilent 7500 series mass spectrometers. The mass dependency of the dead-time correction is due to non-linearity of pulse mode detection and detector baseline shift from the discriminator baseline at high ion count rates. By performing our dead-time calibration directly on 6 amu and 7 amu ( $^6\text{Li}$  and  $^7\text{Li}$ ) the mass dependency in dead-time calculation is avoided. Moreover, at 1 amu mass difference the dead-time variation between 6 amu and 7 amu is approximately 0.023 ns (6 ns/ 260 amu), which is insignificant (<0.01%) in comparison to the actual dead-time (~36 ns).