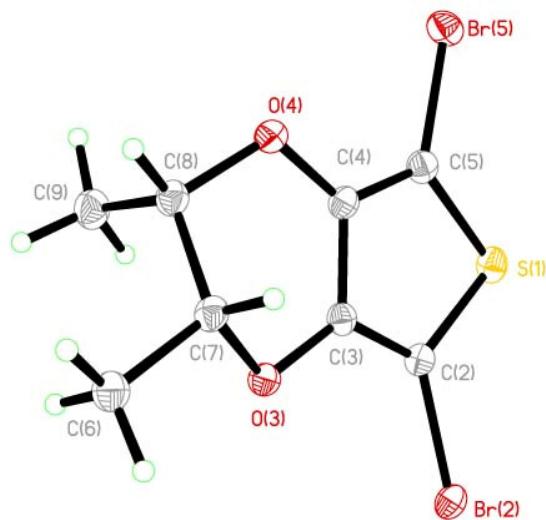


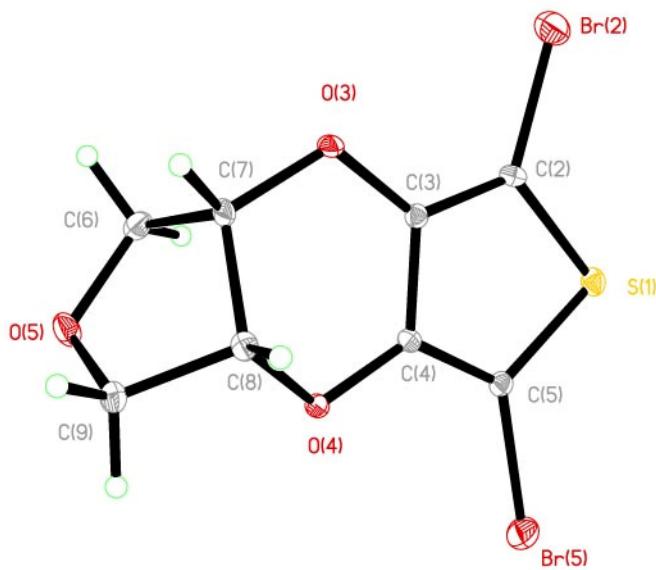
Electronic Supporting Information for the paper

**Towards crystal engineering of solid-state polymerization in dibromothiophenes**

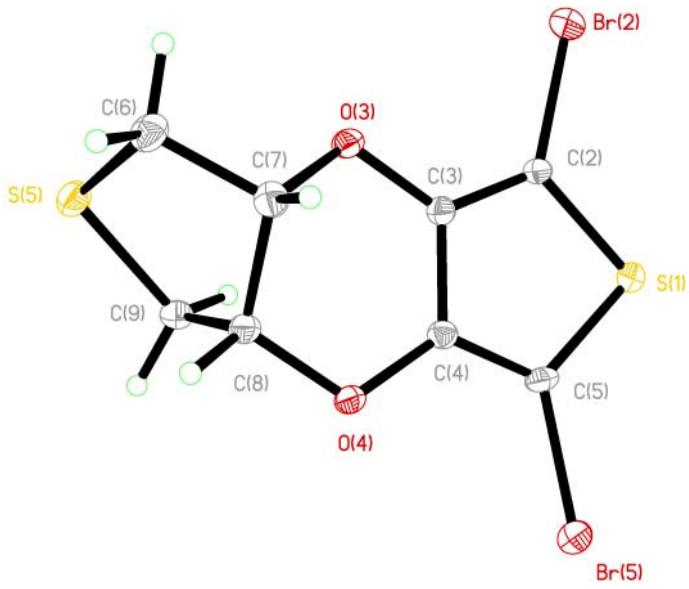
*M. Lepeltier, J. Hiltz, T. Lockwood, F. Bélanger-Gariépy, D. F. Perepichka\**



**Figure S1.** Molecular structure of **4a**



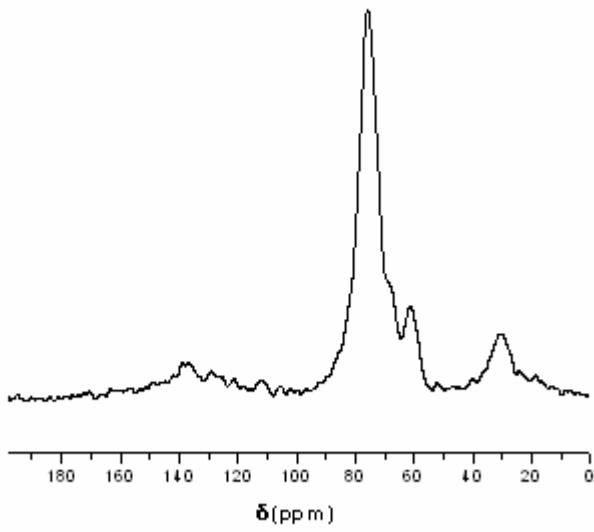
**Figure S2.** Molecular structure of **4b**



**Figure S3.** Molecular structure of **4c**

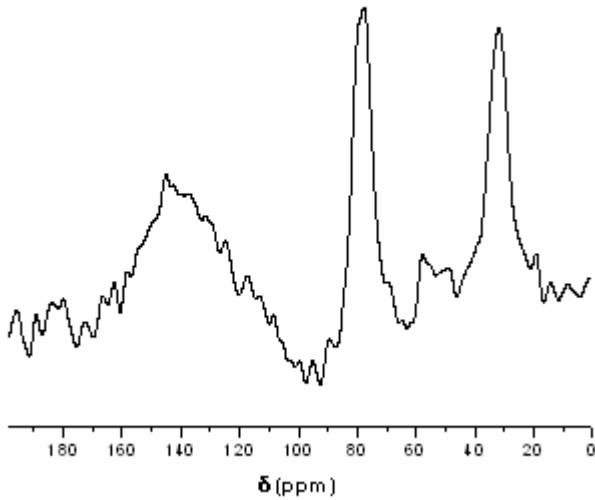
**Table S1.** List of X-ray crystallographic data for **4a**, **4b**, **4c**

	<b>4a</b>	<b>4b</b>	<b>4c</b>
Empirical formula	C8 H8 Br2 O2 S	C8 H6 Br2 O3 S	C8 H6 Br2 O2 S2
Formula weight	328.02	342.01	358.07
Temperature (K)	150	150	150
Crystal system	Orthorhombic	Orthorhombic	Orthorhombic
Space group	Pbca	Pbca	Pbca
a (Å)	8.7321(3)	7.3511(2)	9.3142(3)
b (Å)	14.2816(4)	12.1366(4)	11.3466(4)
c (Å)	16.6582(5)	21.9075(7)	19.4858(7)
Volume (Å <sup>3</sup> )	2077.42(11)	1954.53(10)Å <sup>3</sup>	2059.35(12)Å <sup>3</sup>
Z	8	8	8
D <sub>calc</sub> (g/cm <sup>-3</sup> )	2.098	2.325	2.310
μ(CuKα)(cm <sup>-1</sup> )	11.515	12.363	13.546
F(000)	1264	1312	1376
Crystal size (mm)	0.08 x 0.08 x 0.03	0.08 x 0.08 x 0.07	0.20 x 0.06 x 0.04
Collected reflections	42435	23675	24995
Unique reflections	2026 [R <sub>int</sub> = 0.035]	1916 [R <sub>int</sub> = 0.042]	2010 [R <sub>int</sub> = 0.069]
Observed data [I>2σ(I)]	1949	1809	1940
Goodness-of-fit	1.186	1.182	1.229
R <sub>1</sub>	0.0226	0.0471	0.0492
wR2	0.0638	0.1644	0.1418
final max/min Δρ (eÅ <sup>-3</sup> )	0.38/-0.79	1.15/-2.12	1.46/-1.28



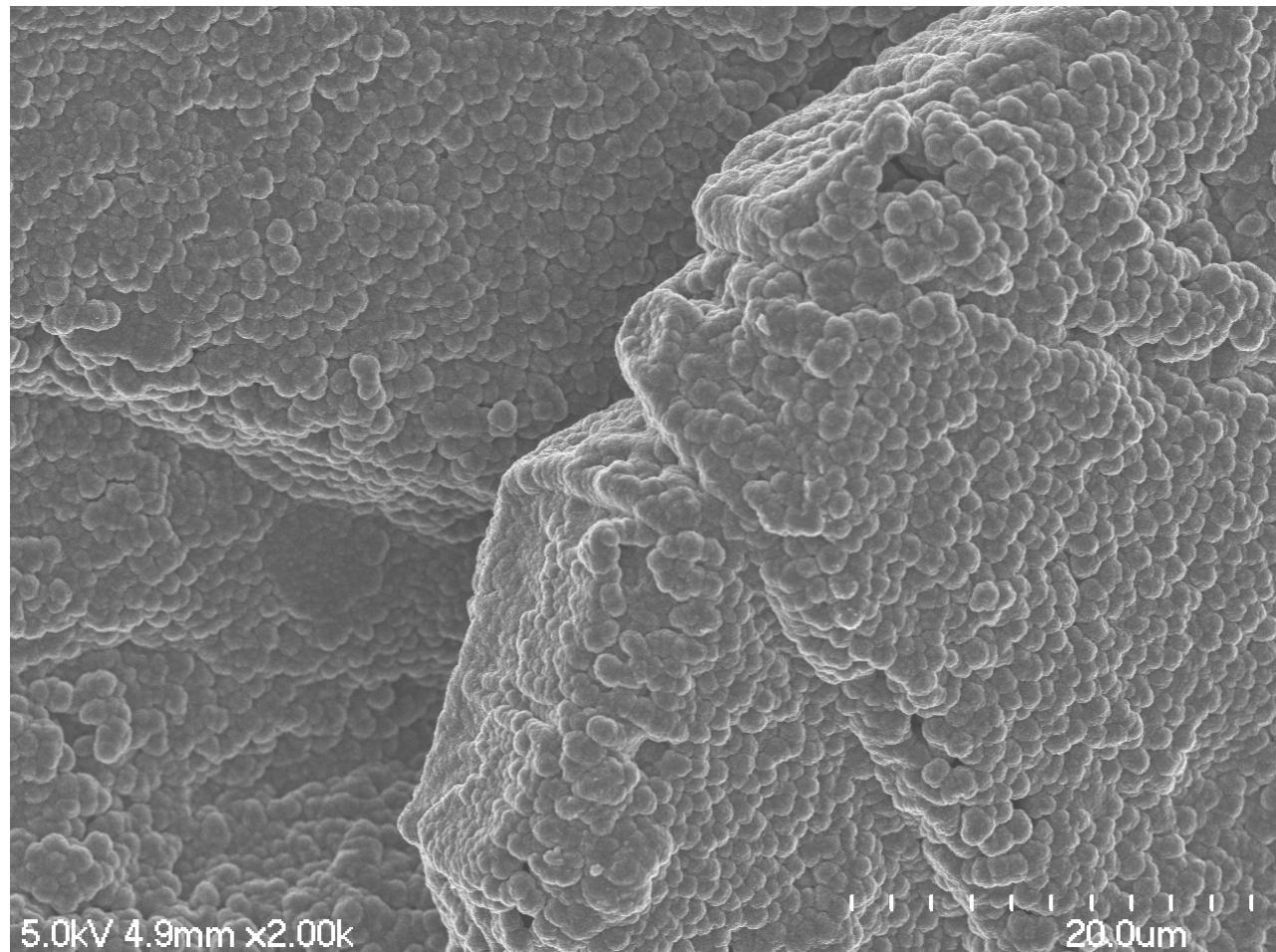
**Figure S4.** CP-MAS  $^{13}\text{C}$  NMR spectrum of poly-4b.

Explanation: There is a strong paramagnetic broadening of aromatic carbons (due to unusually facile doping by air); however, the ring-opening of the furan moiety by bromine is evident: the peak at 62ppm ( $\text{OCH}(\text{CH})\text{CH}_2\text{O}$ ) is reduced in intensity (relative to that of  $\text{OCH}(\text{CH})\text{CH}_2\text{O}$  at 78ppm) and an extra peak appears at  $\sim$ 30 ppm ( $\text{CH}_2\text{-Br}$ ).



**Figure S5.** CP-MAS  $^{13}\text{C}$  NMR spectrum of SSP C.

Explanation: There is a strong paramagnetic broadening of aromatic carbons (due to facile doping by air); however, the bromination of the tetrahydrothiophene moiety is evident from an extra peak appears at  $\sim$ 55 ppm ( $\text{CH}_2\text{S=O}$ ).



**Figure S6.** FE-SEM image of polymer A prepared by SSP of crystalline **4c**, showing substantial morphological changes to the crystal surface.