Supporting Information

Synthesis and Characterization of Perovskite PbTiO3 Nanoparticles with Solution Processibility

Jun Wang, Xinchang Pang, Mufit Akinc, and Zhiqun Lin*

Department of Materials Science and Engineering, Iowa State University, Ames, IA 50011

E-mail: zqlin@iastate.edu

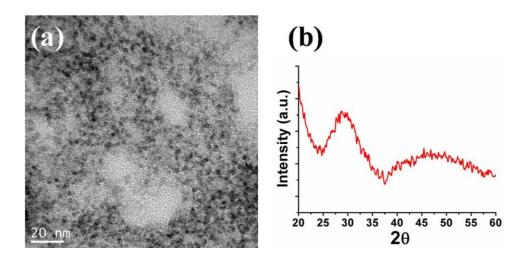


Fig. S1 Characterization of PbTiO₃ nanoparticles obtained after heating at 200 $^{\circ}$ C for 48 hr with the presence of oleic acid. (a) TEM image; and (b) XRD pattern.

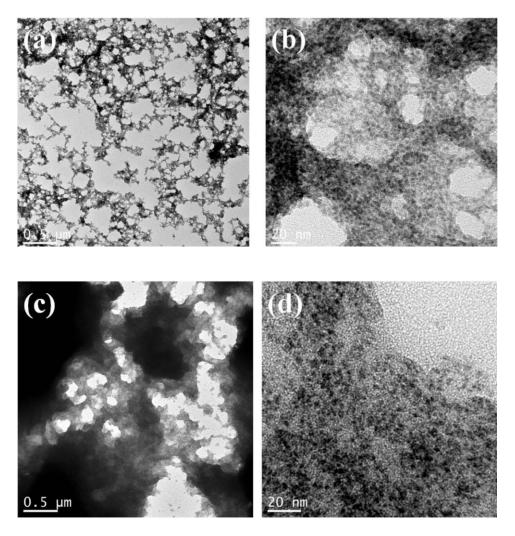


Fig. S2 TEM images of PbTiO₃ nanoparticles obtained after heating at 200°C for 2 hr *without* the addition of oleic acid. (a) and (b): nanoparticles aggregated in ethanol; (c) and (d): nanoparticles aggregated in toluene.

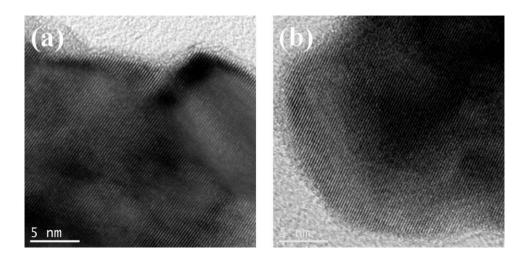
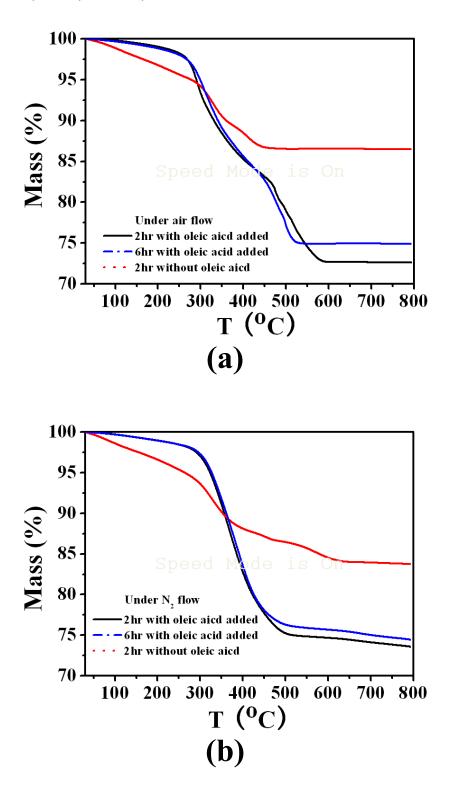


Fig. S3 High resolution TEM images of PbTiO₃ nanoparticles annealed at 500°C for 10 min. (a) particles obtained after heating at 200°C for 2 hr *with* the oleic acid utilized as surface capping ligand; (b) particles obtained after heating at 200°C for 2 hr *without* the addition of oleic acid. To prepare samples for TEM characterization, annealed PbTiO₃ nanoparticles were dispersed in ethanol by ultrasonication, followed by drop casting the suspension on carbon coated copper TEM grid and allowed to dry in air.



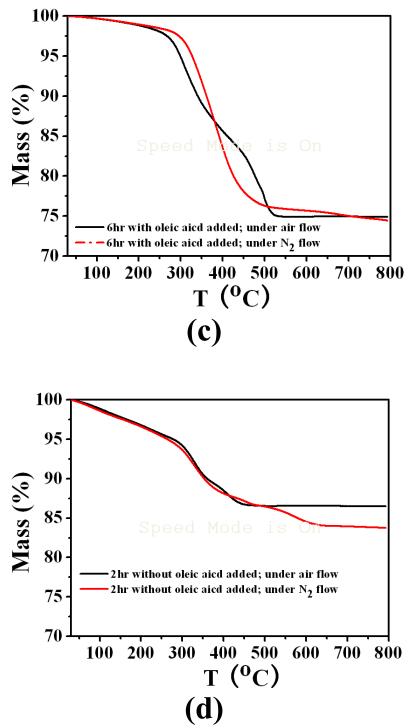


Fig. S4 TGA measurements on fresh PbTiO₃ nanoparticles under air flow (back curve) and N_2 flow (red curve). (a) TGA curves of particles under air flow; (b) TGA curves of particles under N_2 flow; (c) TGA curves of particles obtained after heating at 200°C for 6 hr *with* the oleic acid used as surface capping ligand; and (d) TGA curves of particles obtained after heating at 200°C for 2 hr *without* the addition of oleic acid.