

**Multi-coloration of polyurethane derivatives through click postfunctionalization,
electrochemical oxidation, and Ag⁺ ion complexation**

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1. ^1H NMR spectra

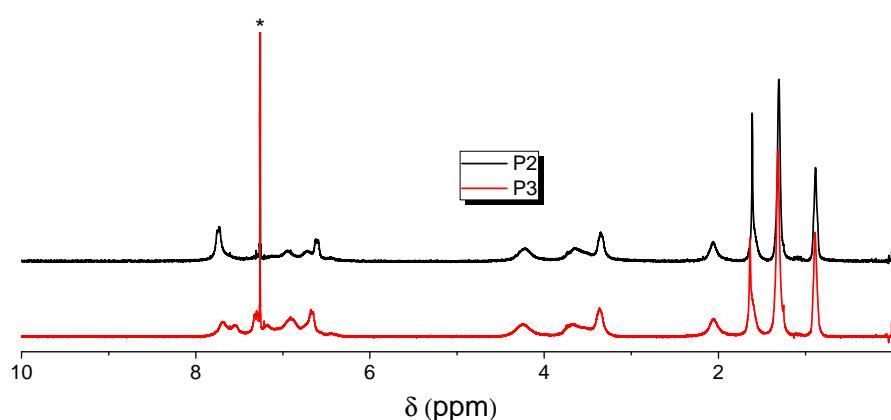


Fig. S1 ^1H NMR spectra of **P2** and **P3** in CDCl_3 at 20 °C.

2. Thermogravimetric analysis (TGA)

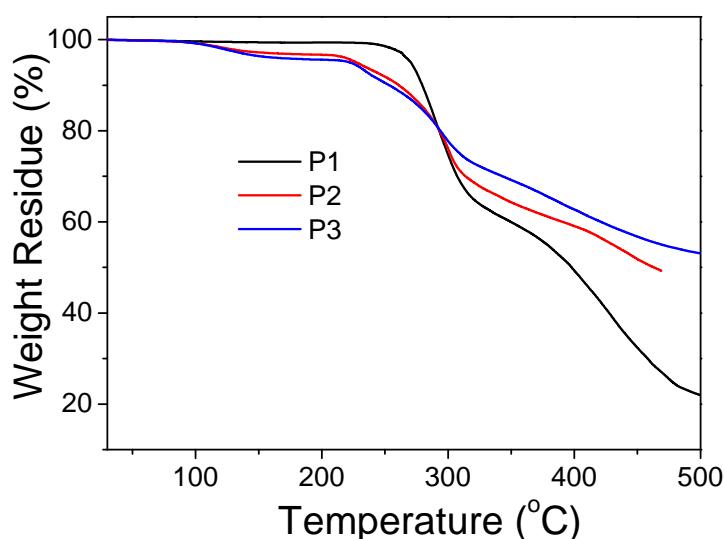


Fig. S2 Thermogravimetric analysis of **P1-P3** at the heating rate of 10 °C min⁻¹ under flowing nitrogen.

3. Electrochemical oxidation

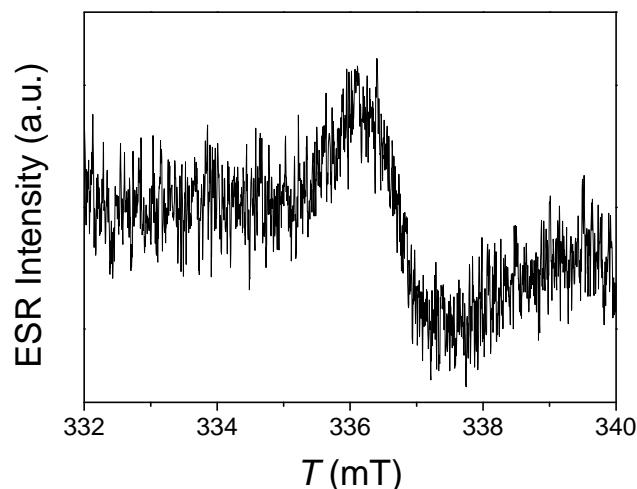


Fig. S3 ESR spectrum of the **P1** thin film on an ITO-coated glass plate by applying 0.25 V for 300 s. The spectrum was recorded one day after the oxidation. The ITO-coated glass plate did not show any ESR signals under the same conditions.

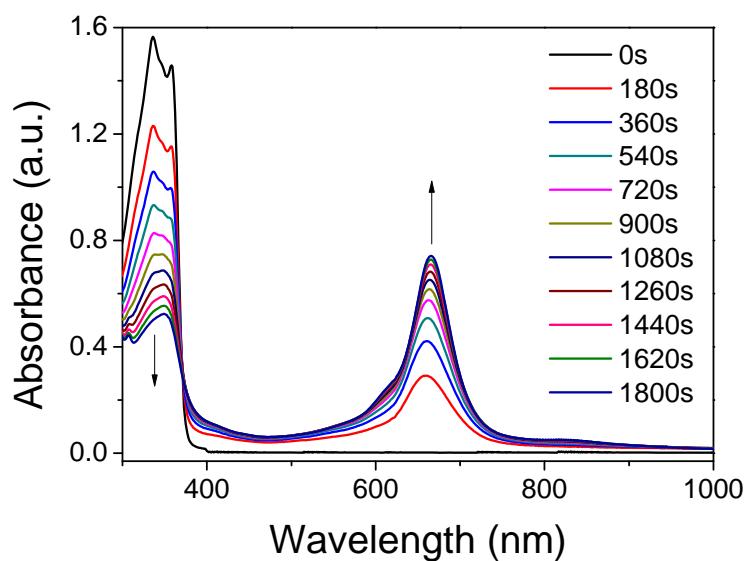


Fig. S4 UV-vis-near IR spectral change of **P1** solution in CHCl_3 containing 0.1 M $(n\text{C}_4\text{H}_9)_4\text{NClO}_4$ by applying a controlled potential of 0.25 V at 20 °C. Platinum mesh was used as a working electrode.

4. Colorimetry

Table S1 Summary of L*a*b* values for the polyurethane derivatives in CHCl₃ (300 μM repeat unit⁻¹)

	P1	P2	P3	P2+Ag ⁺	P3+Ag ⁺	P1 ^{+•}
L*	100	82.65	71.37	63.55	71.55	89.68
a*	0	23.13	-23.48	58.16	-7.86	-14.48
b*	0	76.52	31.72	13.54	50.51	-7.14

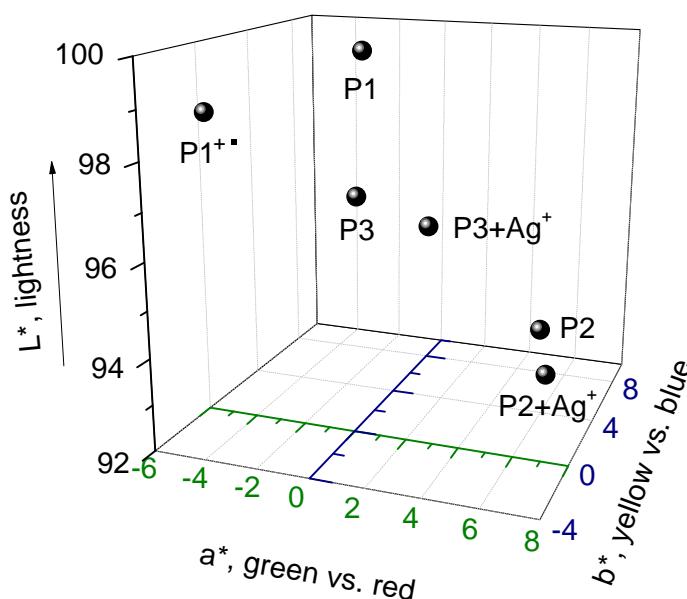


Fig. S5 Three-dimensional L*a*b* plots of values for the polyurethane thin films with the thickness of about 1 μm: **P1**, **P2**, **P3**, **P2** complex with Ag⁺ ions (**P2+Ag⁺**), **P3** complex with Ag⁺ ions (**P3+Ag⁺**), and electrochemically-oxidized state of **P1** by 0.25 V application for 0.5 h (**P1^{+•}**).

Table S2 Summary of L*a*b* values for the polyurethane thin films^a

	P1	P2	P3	P2+Ag⁺	P3+Ag⁺	P1^{+•}
L*	99.86	94.12	95.85	93.56	95.27	99.88
a*	-0.36	6.11	-3.19	6.79	0.01	-4.31
b*	1.81	3.05	7.98	1.27	8.10	-2.54

^a Film thickness of about 1 μm .