Supporting Information

New-type nanoscale coordination particles: toward modification free detection of hydrogen sulfide gas

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Fig. S1[†] Particle size distribution of the as-prepared NCPs determined by dynamic light scattering.



Fig. S2[†] Energy-dispersive X-ray spectrum (EDX) of the obtained NCPs.



Fig. S3[†] Fluorescence spectra of the interest that were taken at different reaction times (from left to right: 30 s, 1 mins, 5 mins). All of the reaction agents were pre-heated to 80°C, before mixing together, and then samples were taken out at different reaction times to monitor their fluorescence changes).



Fig. S4⁺ Fluorescence spectral shifts of NCPs obtained at different reaction temperatures. (a) 60 °C, (b) 50°C,
(c) 40°C, (d) 30°C.



Fig. S5[†] Fluorescence spectral changes by varying the amount of Vc, while keep other conditions unchanged (form bottom to top the Vc/Cu (I) molar ratios were increased as follows: 2.5:1; 5:1; 10:1; 15:1; 30:1). Inset showed the corresponding photographs taken under a UV hand lamp.



Fig. S6[†] Visual color changes of the NCPs after treated with 600 ppm H_2S (left: under the daylight; right: under the UV hand lamp).

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Fig. S7⁺ Selectivity of the NCPs for H₂S over the representative gases (the concentrations of the detected gases were all 600 ppm).



Fig. S8^{\dagger} (a) XPS spectral in the Cu 2p region for NCPs treated with excess H₂S. (b) The corresponding XRD pattern.



Fig. S9[†] Optical images of the fluorescent text written on a filter paper by using the NCPs inks (left) and the corresponding optical images after treated with 80 ppm H₂S (right).



Fig. S10^{\dagger} Selectivity of the NCPs for H₂S over the representative gases (the concentration of H₂S was 8 ppm, while other interference gases were all 80 ppm).



Fig. S11[†] X-ray diffraction (XRD) patterns of the as-prepared NCPs. No sharp diffraction peaks was observed, indicating the amorphous nature of the as-prepared NCPs.