

SUPPLEMENTARY INFORMATION

Inactivation of TiO₂ nano-powders for the preparation of photo-stable sunscreens via carbon-based surface modification.

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Summary

1. Aspect of the pristine and the carbon modified samples
2. Simulation of the PBN /O₂^{•-} adduct recorded after removal of oxygen.
3. Effect of the modification with ethylene glycol on the surface reactivity of anatase
4. Effect of the O₂ pressure during the heating process on the reactivity of TiO₂ modified with ethylene glycol

1. Aspect of the pristine and the carbon modified samples

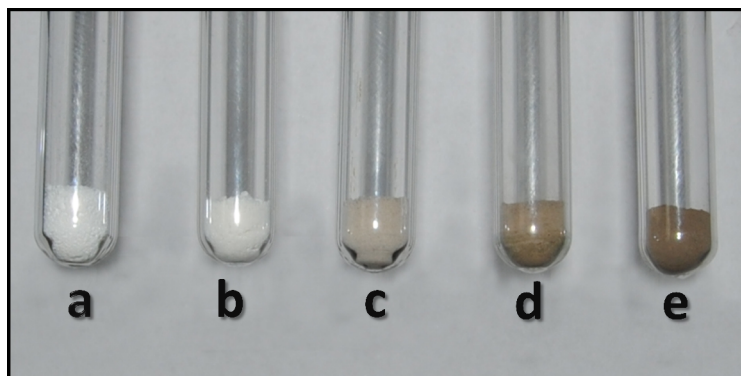


Figure S1. a) $T_{A/R}$; b) $T_{A/R-E}$; c) $T_{A/R-EG}$; d) $T_{A/R-GA}$; e) $T_{A/R-CA}$.

2. Simulation of the PBN /O₂^{•-} adduct recorded after removal of oxygen.

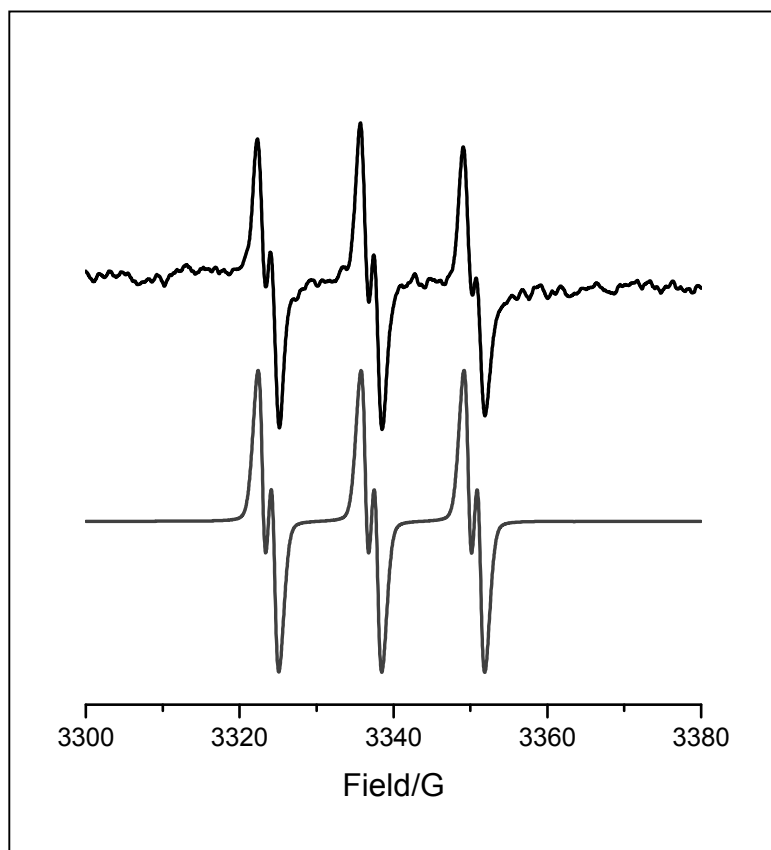


Figure S2. EPR spectra recorded on 28 mg of pristine TiO₂ suspended in a solution 20mM PBN in cyclohexane after 60 minutes of stirring in aerobic conditions. Dissolved oxygen was removed previous EPR measurement. Black line: experimental, gray line: simulation. Calculated hyperfine constants: $a^N=13.4$ G , $a^H=1.6$ G.

3. Effect of the modification with ethylene glycol on the surface reactivity of anatase

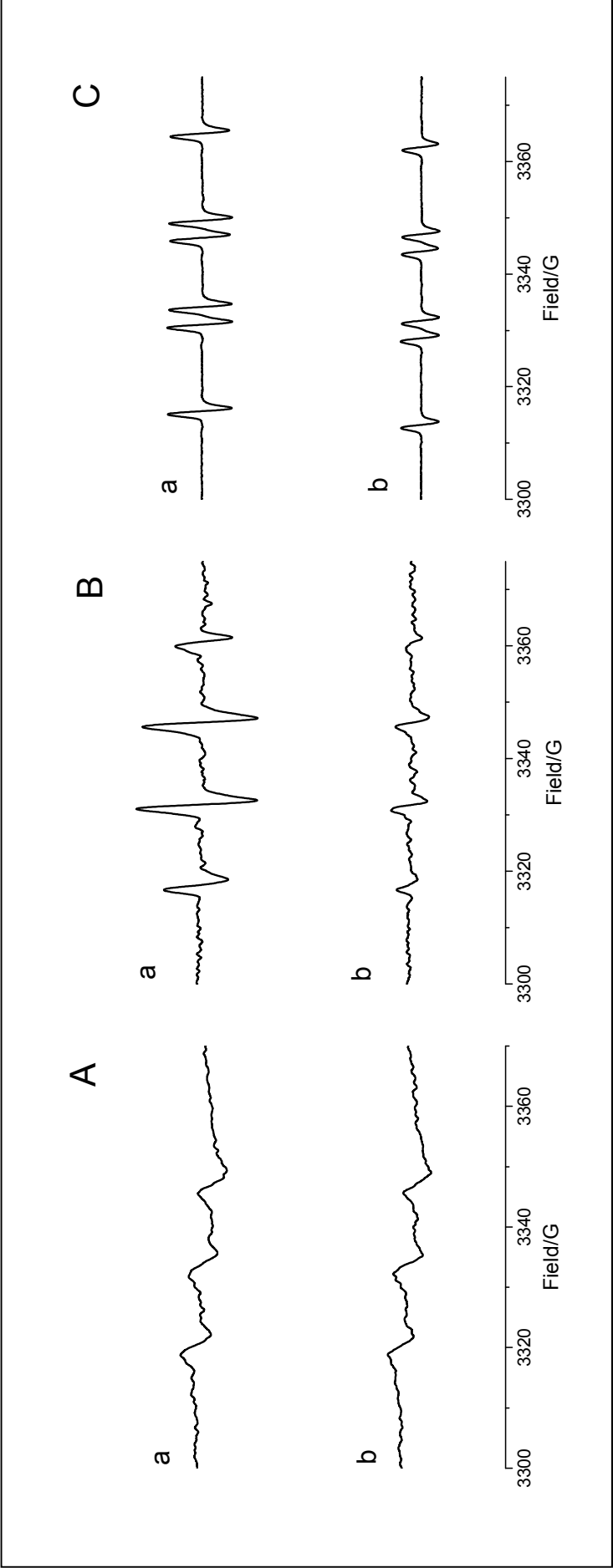


Figure S3. A) superoxide radicals; B) hydroxyl radicals; C) carboxylate radicals. a) T_A ; b) T_A-EG ;

4. Effect of the O₂ pressure during the heating process on the reactivity of TiO₂ modified with ethylene glycol

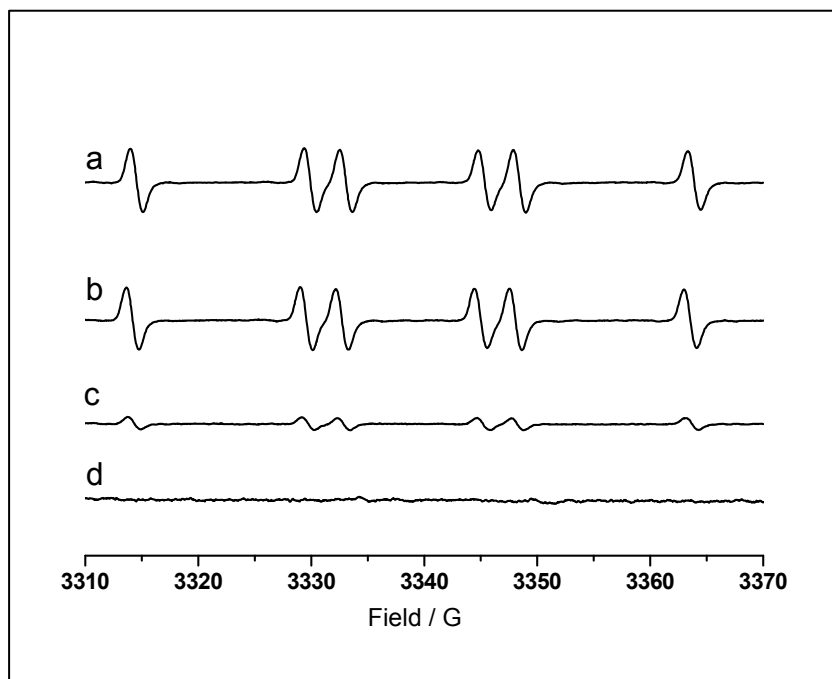


Figure S4. a) T_{A/R}, b) T_{A/R} EG heated at 300 °C in air, c) T_{A/R} EG heated under O₂ (partial pressure: 30 mbar), T_{A/R} EG heated under O₂ (partial pressure: 20 mbar),