Supplementary Material (ESI) for Molecular BioSystems This journal is (c) The Royal Society of Chemistry, 2009 Table 1ES1 Results of Scatchard analysis of MB and QNA binding to polynucleotides in CP

buffer at different salt concentration, PH 7.0 at 20°C^a

Polynucleotide	Drug	[Na ⁺] mM	n K	$L(\mathbf{X}\mathbf{M}^{-1})$
Poly(dG-dC). poly(dG-dC)	MB	10	3.76	1.89x10 ⁶
		20	3.96	8.38×10^5
		50	4.32	3.25×10^5
		100	4.80	1.58×10^{5}
	QNA	10	2.24	6.82x10 ⁶
		20	2.36	3.19×10^{6}
		50	2.44	1.03×10^{6}
		100	2.74	4.41×10^5
Poly(dG). poly(dC)	MB	10	3.52	1.12×10^{6}
		20	3.69	6.06×10^5
		50	3.96	2.29×10^5
		100	4.28	1.05×10^5
	QNA	10	2.39	2.45×10^{6}
		20	2.48	9.97×10^5
		50	2.61	4.25×10^5
		100	2.84	1.74×10^5
Poly(dA-dT).poly(dA-dT)	MB	10	3.36	4.40×10^{5}
		20	3.72	2.60×10^5
		50	3.90	9.28×10^4
		100	4.18	$4.20 \mathrm{x} 10^4$
	QNA	10	2.35	5.25x10 ⁶
		20	2.53	3.02×10^{6}
		50	2.61	8.86×10^{5}
		100	2.71	4.23×10^{5}
Poly(dA).poly(dT)	MB	10	3.36	1.03×10^{5}
		20	3.56	5.47×10^4
		50	4.37	1.76×10^4
		100	4.80	.89x10 ⁴
	QNA	10	3.63	9.55×10^{5}
		20	4.03	4.50×10^{5}
		50	4.70	1.46×10^5
		100	5.24	5.79×10^4

^aData presented from the average of four determinations in each.







FIGURE LEGENDS OF ESI

Fig. S1 of ESI. Log-log plot of the variation of the intrinsic binding *K* as a function of ionic strength $[Na^+]$ for the complexation of (Panel a) MB-Poly(dG-dC).poly(dG-dC) (\blacksquare), MB-Poly(dG).poly(dC) (\bullet),MB-Poly(dA-dT). poly(dA-dT) (\blacktriangle), and MB-Poly(dA).poly(dT) (∇) and (panel b) QNA-Poly(dG-dC).poly(dG-dC) (\blacksquare), QNA-Poly(dG).poly(dC) (\bullet),QNA-Poly(dA-dT). poly(dA-dT) (\bigstar) and QNA-Poly(dA).poly(dT) (∇), The solid lines represent the best fit to the experimental points.

Fig. S2 of ESI. The top panels (a), (b), represent the raw ITC data for the sequential injection of aliquots of the MB (panel a) 10 μ L each form a stock of 200 μ M for poly(dG).poly(dC).) was injected into the isothermal sample chamber containing 30 μ M poly(dG).poly(dC). but in case of poly(dA-dT).poly(dA-dT),(panel b) the reverse titration was performed by injecting poly(dA-dT).poly(dA-dT), 10 μ L each from a stock of 600 μ M into MB solution (10 μ M) in the calorimeter cell. The top panels (c), (d), represent the raw ITC data for the sequential injection of aliquots of the QNA, 10 μ L each form a stock of 400 μ M for poly(dG-dC).poly(dG-dC), poly(dG).poly(dC). and 200 μ M for poly(dA-dT).poly(dA-dT), were injected into the isothermal sample chamber containing 30 μ M poly(dG-dC).poly(dG-dC) and 50 μ M poly(dA-dT).poly(A-dT).poly(A-dT).poly(DNA of MB and QNA respectively). The data (closed squares) were fitted to a one-site model and the solid lines represent the best-fit data.

Fig. S3 of ESI. Temperature dependence of the thermodynamic binding parameters, $T\Delta S(\blacktriangle - \bigstar)$ $\Delta H (\bullet - \bullet)$, and $\Delta G (\blacksquare - \blacksquare)$ for binding of MB (panel a, b, c, d) poly(dG-dC).poly(dG-dC), poly(dG).poly(dC).), poly(dA-dT).poly(dA-dT), poly(dA).poly(dT) and QNA (panel e, f, g, h) to poly(dG-dC).poly(dG), poly(dG).poly(dC).), poly(dA-dT).poly(dA-dT),

poly(dA).poly(dT) respectively.