

Electrochemical methodology for determination of imidazolium ionic liquids (solids at room temperature) properties: Influence of the temperature

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Supplementary Information

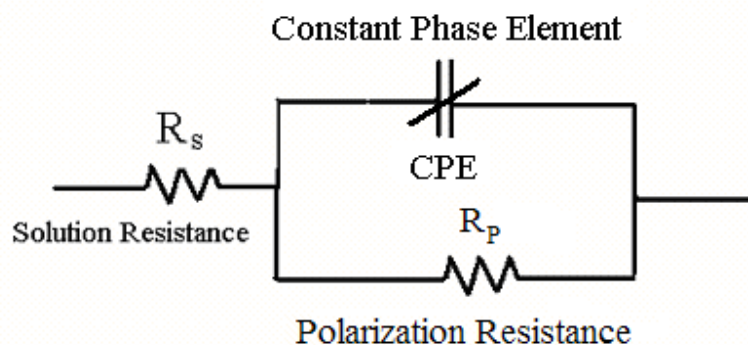


Figure S1. Proposed equivalent circuit for the platinum/IL interface.

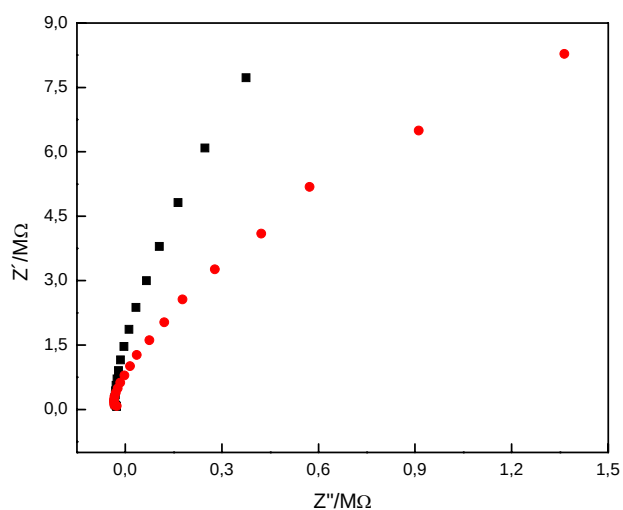


Figure S2. Nyquist diagram of **3** [C₂O₂MIm][Cl] at 32 °C under a pure argon atmosphere (black points) and pure oxygen atmosphere (red points).

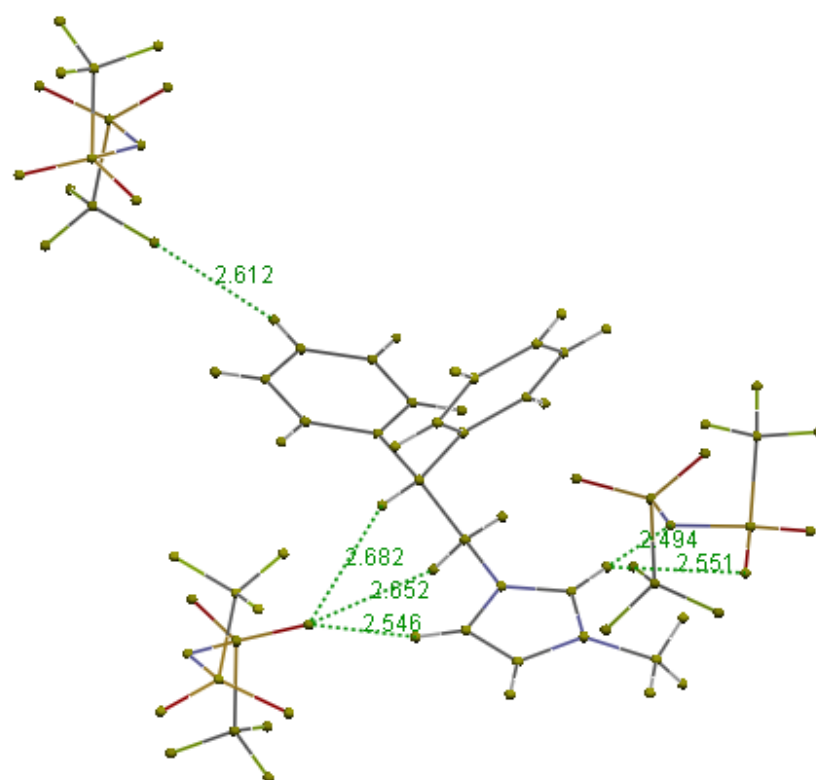


Figure S3. Mercury view of **2a** [Ph₂C₂MIm][NTf₂] showing supramolecular interactions at room temperature.

Table S1. Supramolecular interactions in the solid state of **2a** [Ph₂C₂MIm][NTf₂] at room temperature.

D	H	A	d D-H (Å) ^a	d H-A (Å) ^a	d D-A (Å) ^a	Angles DHA (°)
C15	H15	F6A ^b	0.930	2.612	3.389	141.42
C2	H2	N3 ^c	0.930	2.494	2.494	155.23
C2	H2	O12 ^c	0.930	2.551	2.551	138.38
C4	H4	O11 ^d	0.929	2.546	3.255	133.37
C1	H1A	O11 ^d	0.971	2.652	3.258	120.84
C5	H5	O11 ^d	0.980	2.682	3.325	123.42

^a d = distance. ^b Symmetry transformation used to generate equivalent atoms: 1.5-x, 1/2+y,

1.5-z. ^c Symmetry transformation used to generate equivalent atoms: 1-x, -y, 1-z.

^d Symmetry transformation used to generate equivalent atoms: 0.5+x, 0.5-y, -0.5+z.

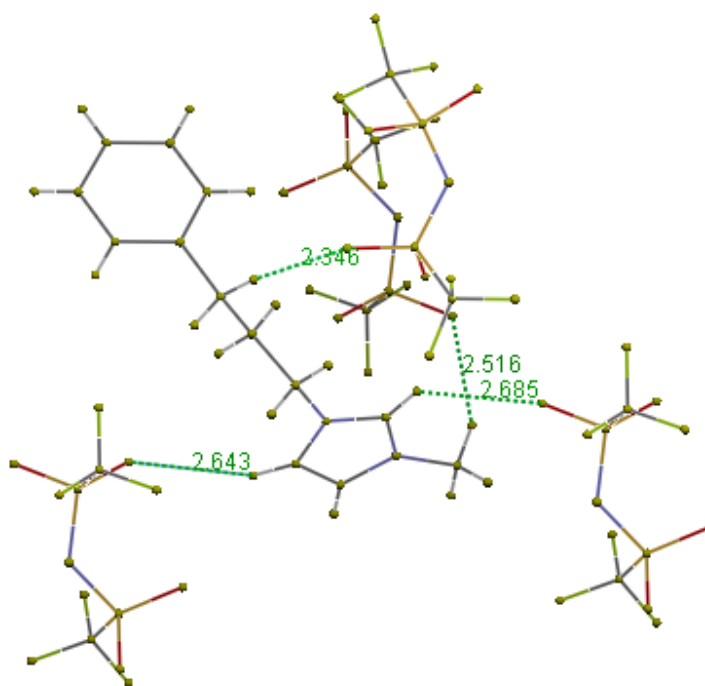


Figure S4. Mercury view of **2b** [PhC₃MIm][NTf₂] showing supramolecular interactions at room temperature.

Table S2. Supramolecular interactions in the solid state of **2b** [PhC₃MIm][NTf₂] at room temperature.

D	H	A	d D-H (Å) ^a	d H-A (Å) ^a	d D-A (Å) ^a	Angles DHA (°)
C6	H6C	O18 ^b	0.960	2.516	3.392	151.74
C2	H2	O18 ^c	0.930	2.685	3.492	133.24
C9	H9A	O25 ^d	0.970	2.346	3.302	168.94
C4	H4	O19 ^b	0.930	2.643	3.520	157.45

^a d = distance. ^b Symmetry transformation used to generate equivalent atoms: -x, 0.5+y,

0.5-z. ^c Symmetry transformation used to generate equivalent atoms: -x, 0.5+y, 0.5-z.

^d Symmetry transformation used to generate equivalent atoms: x,y,z.

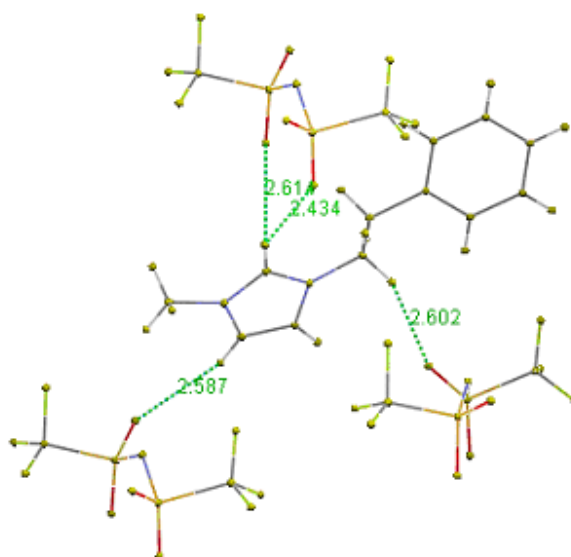


Figure S5. Mercury view of **2c** [PhC₂Mim][NTf₂] showing supramolecular interactions at room temperature.

Table S3. Supramolecular interactions in the solid state of **2c** [PhC₂Mim][NTf₂] at room temperature.

D	H	A	d D-H (Å) ^a	d H-A (Å) ^a	d D-A (Å) ^a	Angles DHA (°)
C2	H2A	O22 ^b	0.970	2.602	3.408	140.75
C5	H5	O12 ^c	0.930	2.587	3.475	159.70
C3	H3	O11 ^d	0.930	2.614	3.781	135.61
C3	H3	O21 ^d	0.930	2.434	3.623	141.05

^a d = distance. ^b Symmetry transformation used to generate equivalent atoms: 0.5 -x, 0.5+y,

0.5-z. ^c Symmetry transformation used to generate equivalent atoms: -x, 2-y, -z.

^d Symmetry transformation used to generate equivalent atoms: 1-x, 2-y, -z.

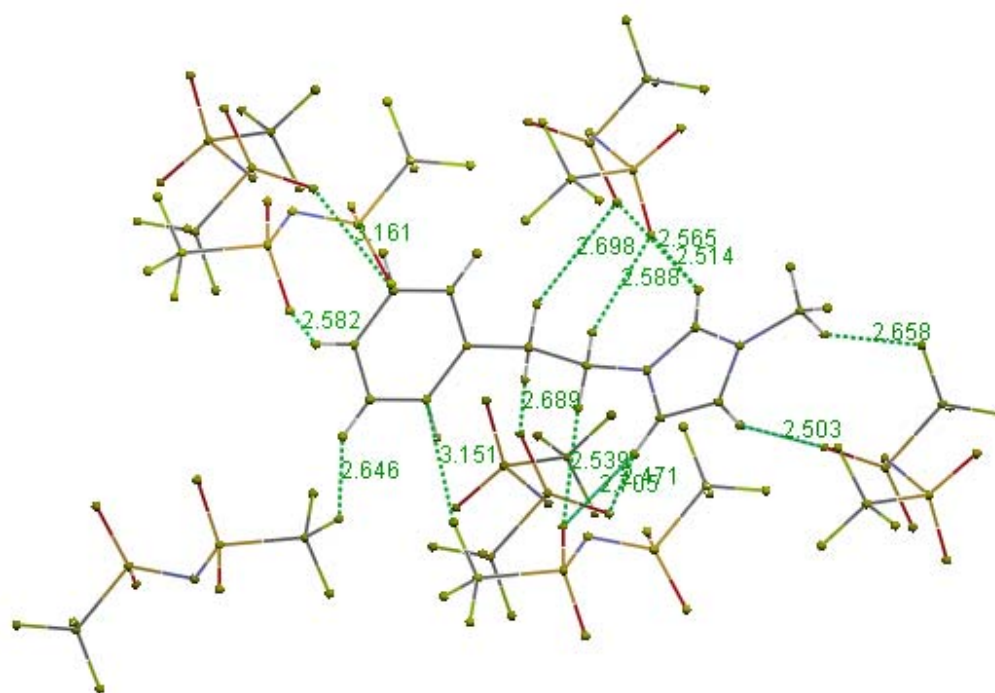


Figure S6. Mercury view of **2c** [PhC₂Mim][NTf₂] showing supramolecular interactions at -100 °C.

Table S4. Supramolecular interactions in the solid state of **2c** [PhC₂Mim][NTf₂] at -100 °C.

D	H	A	d D-H (Å) ^a	d H-A (Å) ^a	d D-A (Å) ^a	Angles DHA (°)
C9	H9	O4 ^b	0.950	2.471	3.209	134.53
C10	H10	O4 ^c	0.950	2.503	3.400	157.41
C12	H12A	F4 ^c	0.981	2.658	3.553	151.89
C7	H7B	O3 ^b	0.991	2.689	3.643	161.93
C9	H9	O1 ^d	0.950	2.705	3.359	126.66
C3	H3	O2 ^e	0.950	2.582	3.316	134.32
C4	H4	F2 ^f	0.950	2.646	3.236	120.67
C7	H7A	O3 ^g	0.990	2.698	3.495	137.73
C8	H8A	O2 ^g	0.989	2.588	3.499	153.16
C11	H11	O2 ^g	0.950	2.514	3.333	144.50
C11	H11	O3 ^g	0.950	2.565	3.042	111.29
C8	H8B	O1 ^d	0.990	2.539	3.344	138.39
C2	O4	S2 ^h	3.161	1.427	4.442	148.55
C5	F1	C13 ^d	3.151	1.329	4.055	124.52

^a d = distance. ^b Symmetry transformation used to generate equivalent atoms: 1-x, 1-y, 1-z.

^c Symmetry transformation used to generate equivalent atoms: 1+x, y, z. ^d Symmetry

transformation used to generate equivalent atoms: 0.5+x, 0.5-y, -0.5+z. ^e Symmetry

transformation used to generate equivalent atoms: -0.5+x, 0.5-y, -0.5+z. ^f Symmetry

transformation used to generate equivalent atoms: 0.5-x, 0.5+y, 0.5-z. ^g Symmetry

transformation used to generate equivalent atoms: x, y, z. ^h Symmetry transformation used

to generate equivalent atoms: -x, 1-y, 1-z.

Table S5. Crystallographic data for imidazolium ILs **2a-c**.

	2a (25 °C)	2b (25 °C)	2c (25°C)	2c (-100 °C)
	[Ph ₂ C ₂ Mim][NTf ₂]	[PhC ₃ Mim][NTf ₂]	[PhC ₂ Mim][NTf ₂]	[PhC ₂ Mim][NTf ₂]
formula	C ₂₀ H ₁₉ F ₆ N ₃ O ₄ S ₂	C ₁₅ H ₁₇ F ₆ N ₃ O ₄ S ₂	C ₁₄ H ₁₅ F ₆ N ₃ O ₄ S ₂	C ₁₄ H ₁₅ F ₆ N ₃ O ₄ S ₂
M	543.5	481.44	467.41	467.41
crystal system	monoclinic	monoclinic	monoclinic	monoclinic
space group	<i>P</i> 21/n	<i>P</i> 21/c	<i>P</i> 21/n	<i>P</i> 21/n
a [Å]	10.0301(9)	11.1807(2)	9.9583(4)	9.7783(9)
b [Å]	18.862(2)	14.8585(3)	15.4739(6)	15.2038(16)
c [Å]	12.979(1)	12.8594(3)	13.1044(5)	12.8926(14)
β [°C]	93.289(6)	105.50	92.897(6)	92.951(6)
Z	4	4	4	4
V [Å ³]	2451.3(4)	2058.66(7)	2016.73(17)	1914.2(3)
ρ _{calcd} [g/cm ³]	1.47	1.553	1.539	1.622
T [K]	173	173	173	173
θ range [°C]	3-30	2-29	2-25	2-30
μ [mm]	0.294	0.339	0.056	0.362
reflections	29841	36755	17169	23008
measured				
final R1,	0.078, 0.2326	0.092, 0.295	0.1012, 0.3344	0.0432, 0.1127
ωR2[I>2 σ (I)]				