

Synthesis of Mesoporous Metal Complex-Silica Materials and their Use as Solvent-free Catalysts

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Supporting information

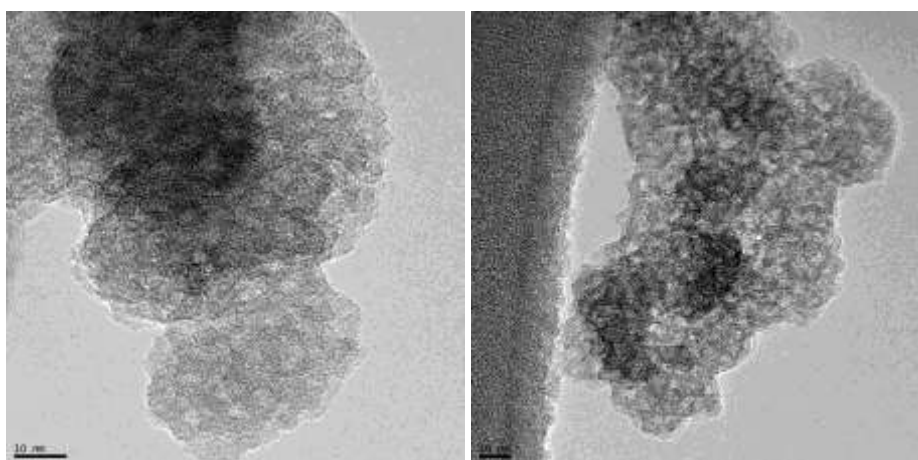
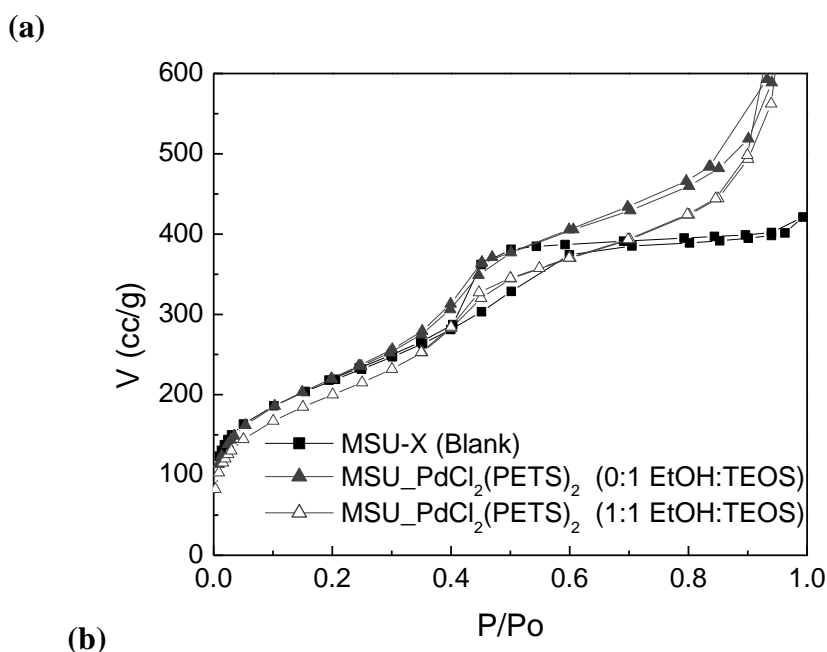


Figure S1. (a) N₂ isotherms for catalyst MSU-PdCl₂(PETS)₂ prepared with 1% Pd:SiO₂ with (- Δ -) and without (- \blacktriangle -) ethanol in their synthesis. (b) Representative TEM images of these samples prepared with (left) and without (right) ethanol in the synthesis. The scale bars represent 10 nm

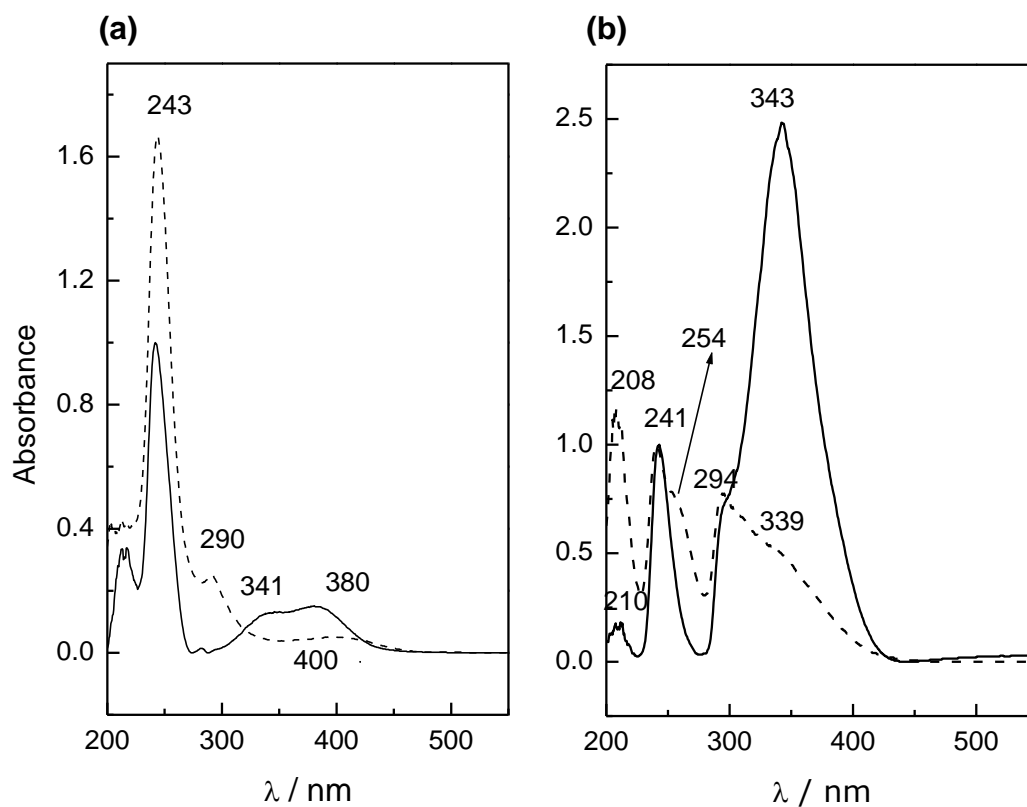


Figure S2. DRUV spectra of (a) pressed pure powders of the MSU-PdCl₂(APTS)₂ material (solid line) and the monomer complex [trans-PdCl₂(APTS)₂] **1** (dotted line) and (b) pressed pure powders of the MSU-PdCl₂(PPETS)₂ material (solid line) and the monomer complex [trans-PdCl₂(PPETS)₂] **3** (dotted line).

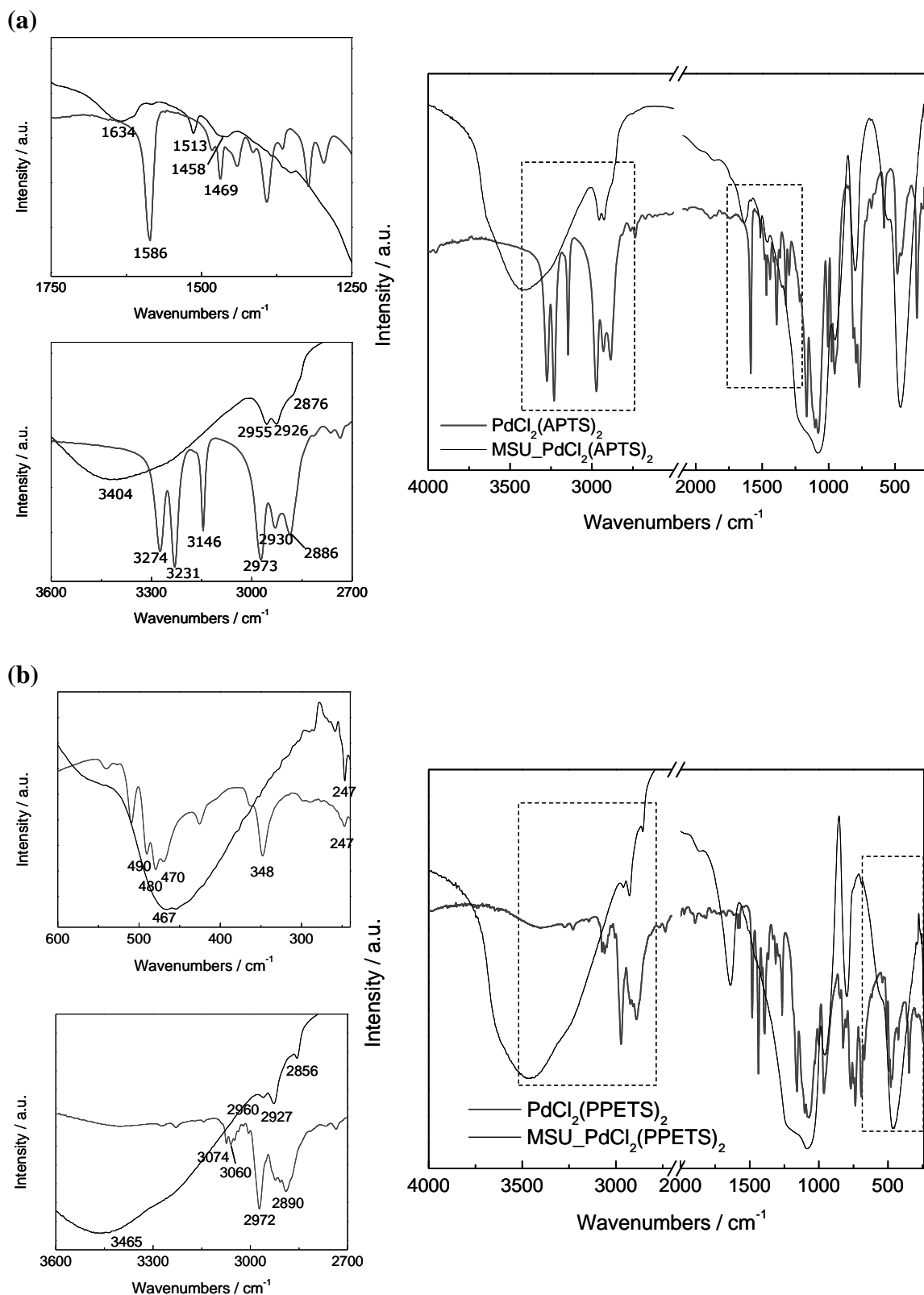


Figure S3. FT-IR spectra of (a) the MSU-PdCl₂(APTS)₂ silica material (up) and the monomer complex [trans-PdCl₂(APTS)₂] **1** (down) and (b) the MSU-PdCl₂(PPETS)₂ silica material (up) and the monomer complex [trans-PdCl₂(PPETS)₂] **3** (down). For clarity, two magnifications of the marked areas in the spectra are shown in the left.

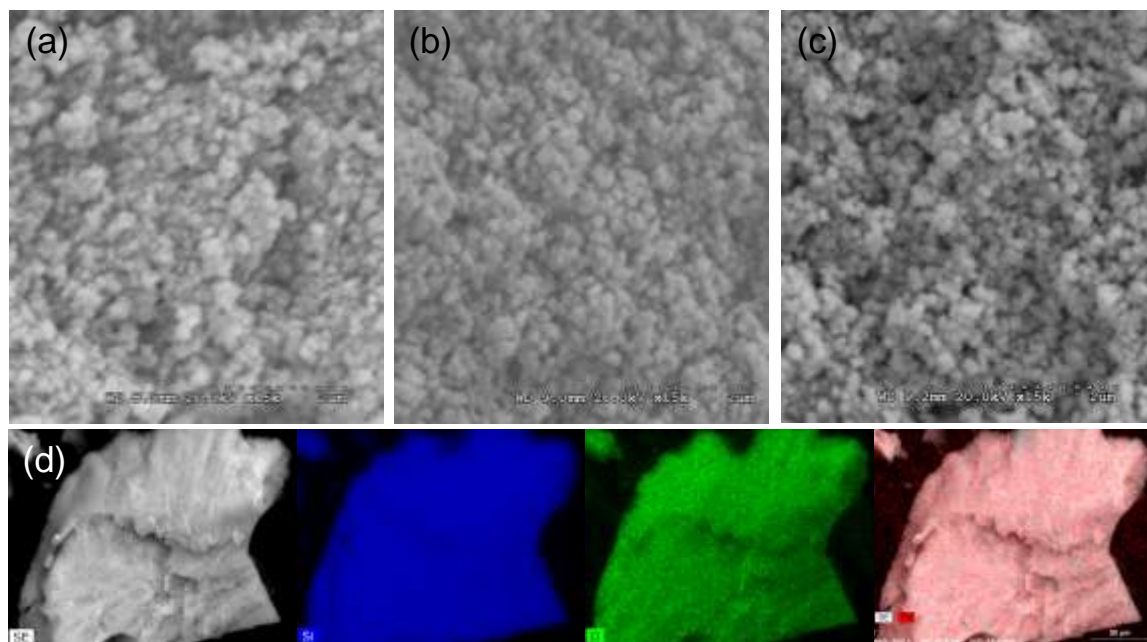


Figure S4. SEM images of the (a) MSU_PdCl₂(APTS)₂ material prepared from amine ligand and 0.65 wt% loading of Pd, (b) MSU_PdCl₂(PETS)₂ material prepared from pyridine ligand and 0.75 wt% loading of Pd and (c) MSU_PdCl₂(PPETS)₂ material prepared from phosphine ligand and 0.70 wt% loading of Pd. (d) X-ray elemental mapping images for a representative sample of MSU_PdCl₂(PETS)₂ (0.75 wt% Pd:SiO₂), blue, green and red dots represent silicon, oxygen and palladium atoms, respectively.