

Supplementary Information

Silver complex of chloroquine: synthesis, characterization and structural properties.

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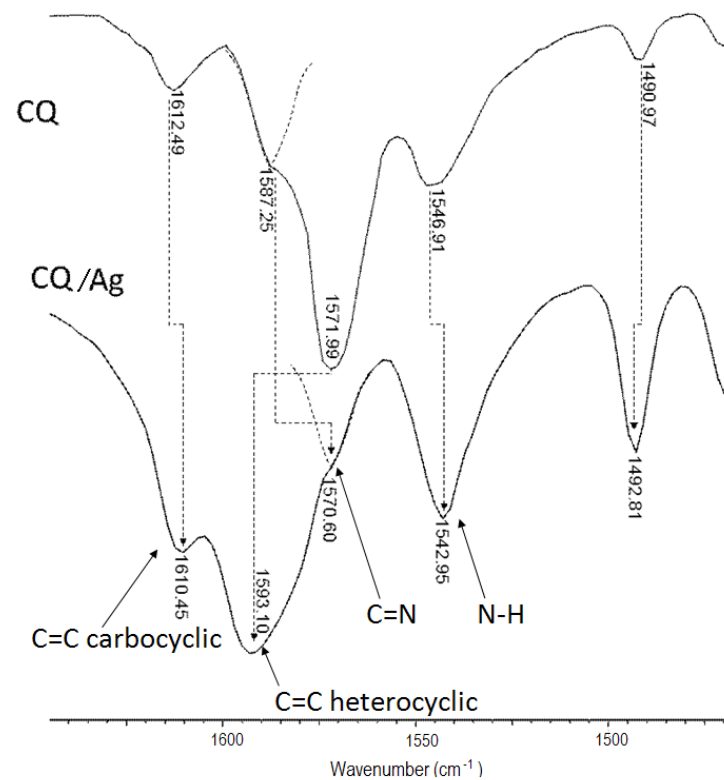
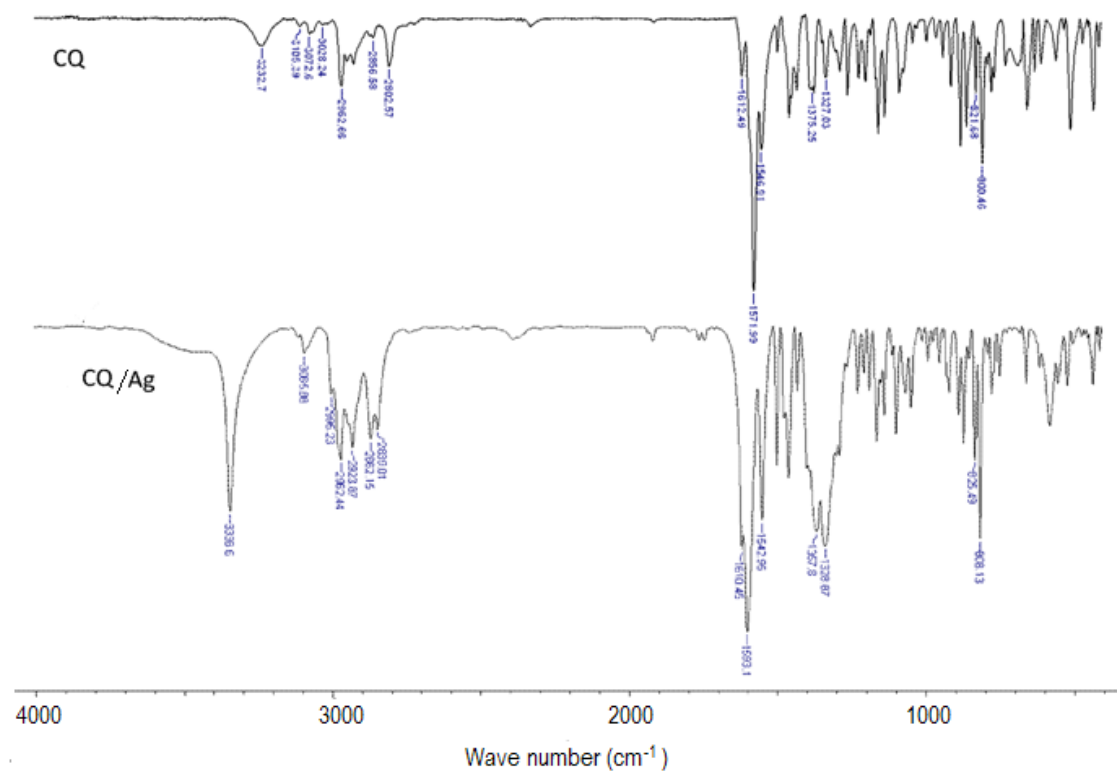
Topics

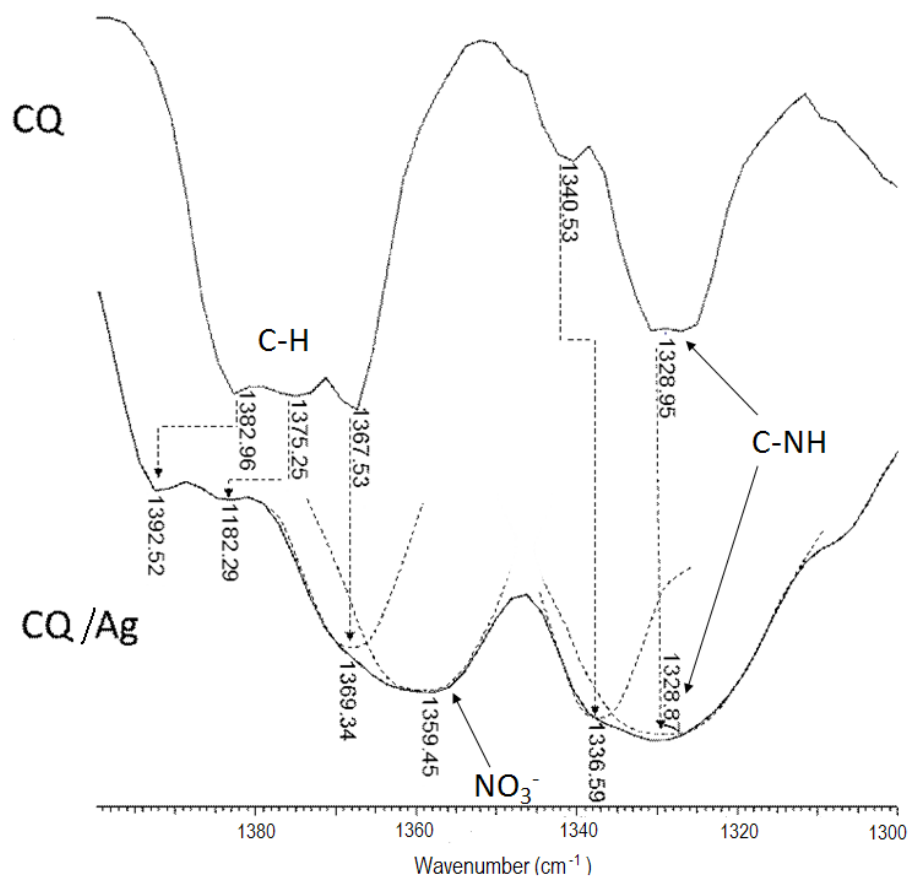
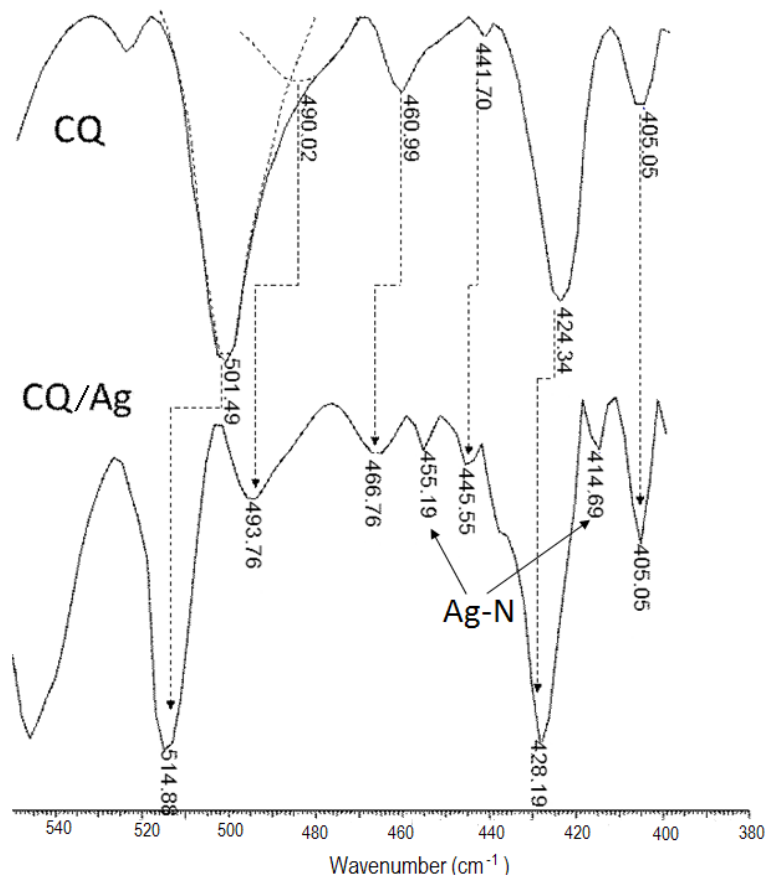
S1. FTIR spectra

S2. Computational results

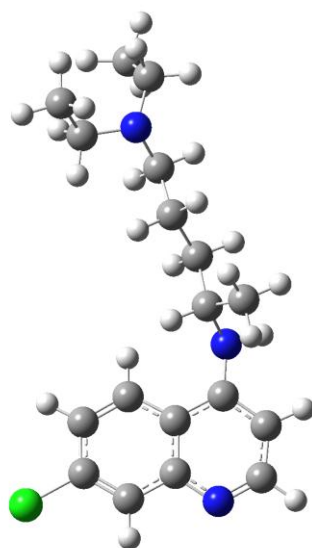
S3. ESI/FT-ICR Mass Spectrometry

S1. Comparative FTIR spectra of CQ and CQ/Ag complex



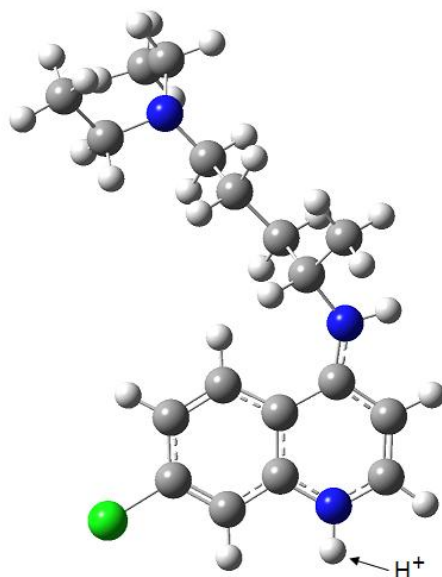


S2. Computational results: Structures optimized at B3LYP/6-31G(d)/LANL2DZ ECP level.
Interatomic distances in Å.



CQ

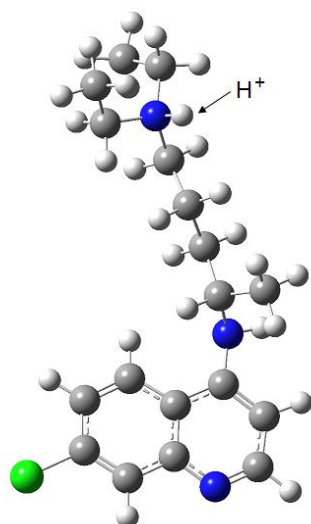
Protonated chloroquines



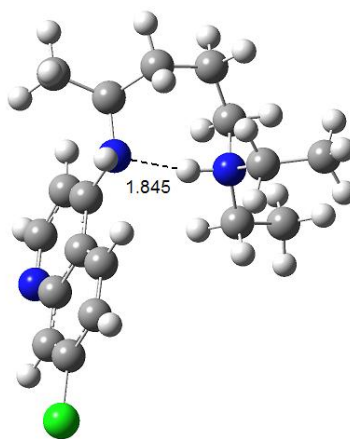
CQH⁺[1]: protonated on N sp²

PA= 1017.8 kJ·mol⁻¹

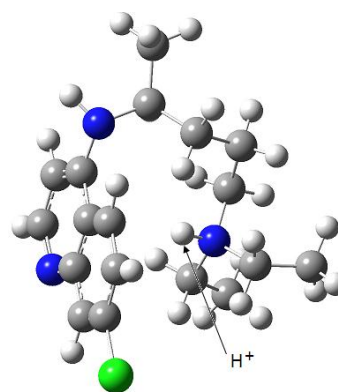
Protonated on N sp³:



CQH⁺[2a]
PA= 975.0 kJ·mol⁻¹

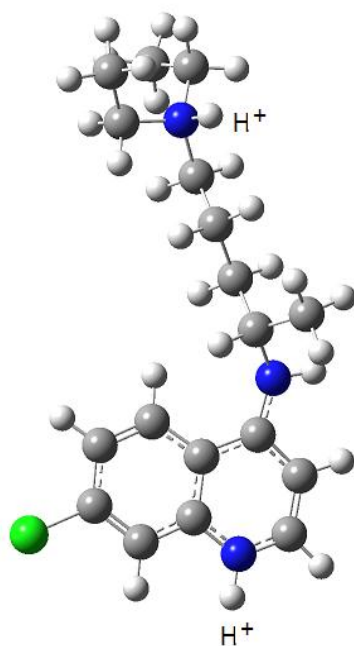


CQH⁺[2b]
PA= 1009.8 kJ·mol⁻¹

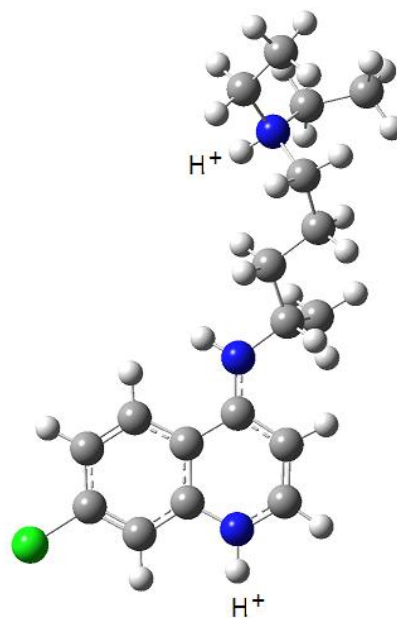


CQH⁺[2c]
PA= 968.0 kJ·mol⁻¹

Diprotonated chloroquinines

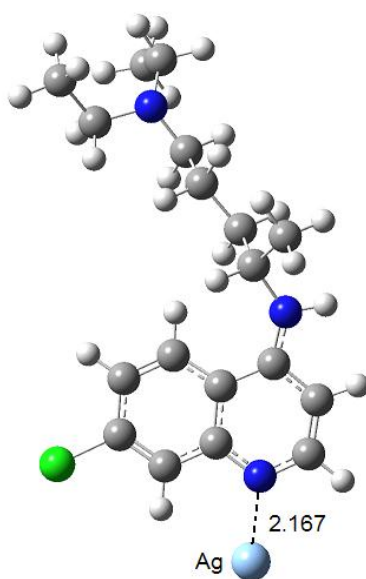


CQH₂²⁺[a]
 $\Delta_r H^0 = 795.4 \text{ kJ}\cdot\text{mol}^{-1}$



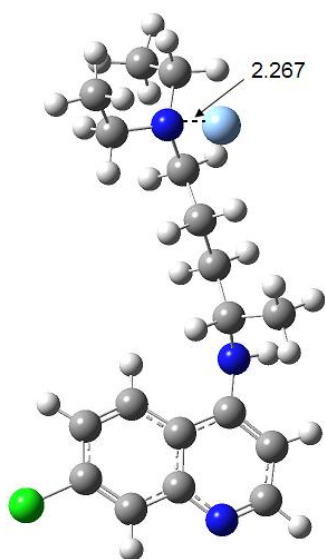
CQH₂²⁺[b]
 $\Delta_r H^0 = 797.4 \text{ kJ}\cdot\text{mol}^{-1}$

CQAg⁺ cations



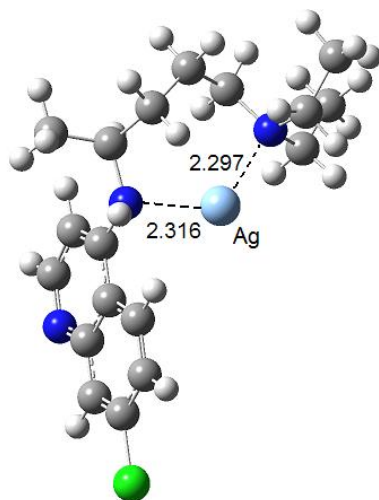
CQAg⁺[1]

$$\Delta_r H^0 = 207.6 \text{ kJ}\cdot\text{mol}^{-1}$$



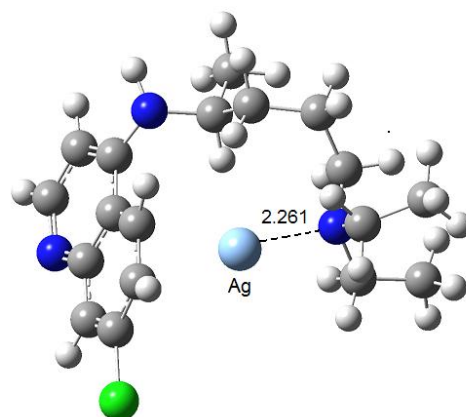
CQAg⁺[2a]

$$\Delta_r H^0 = 168.5 \text{ kJ}\cdot\text{mol}^{-1}$$



CQAg⁺[2b]

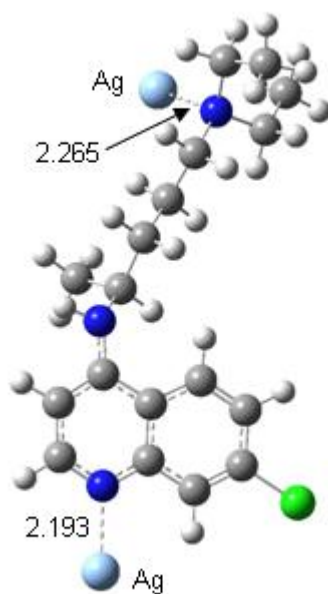
$$\Delta_r H^0 = 228.9 \text{ kJ}\cdot\text{mol}^{-1}$$



CQAg⁺[2c]

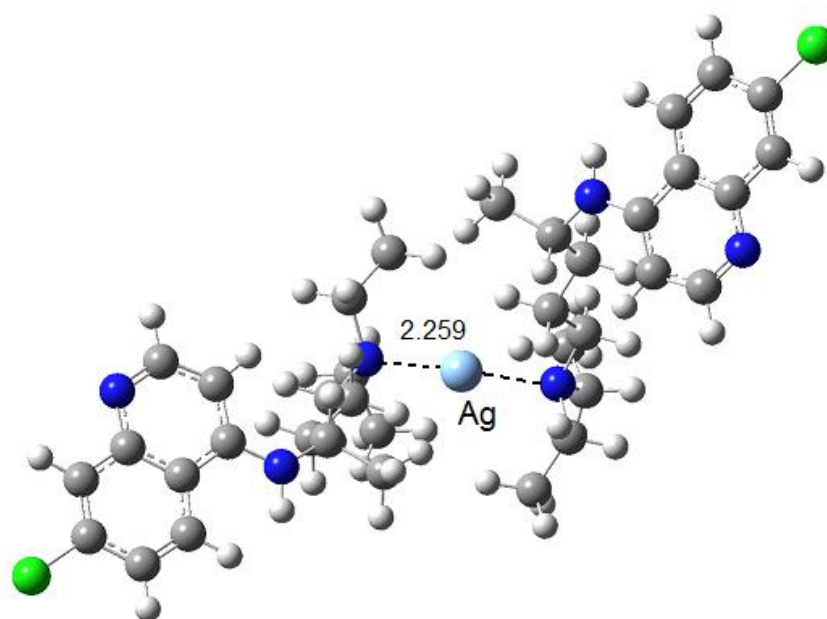
$$\Delta_r H^0 = 229.5 \text{ kJ}\cdot\text{mol}^{-1}$$

CQAg_2^{2+} dication

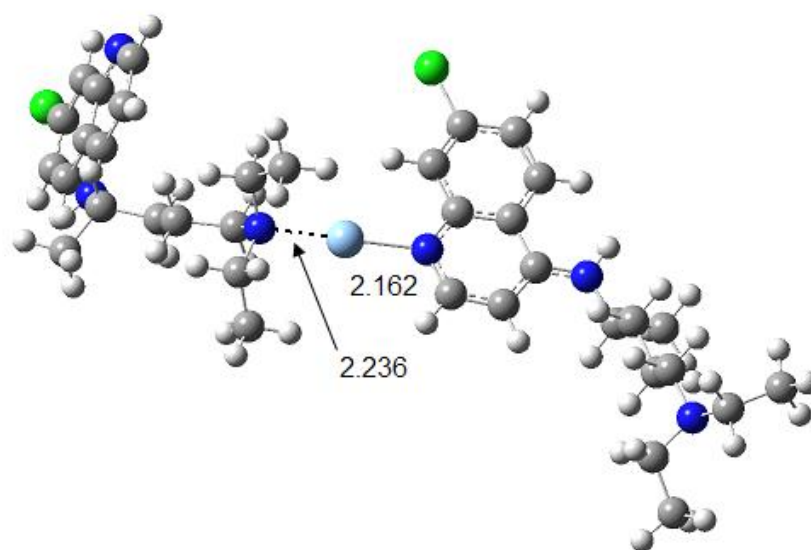


$$\Delta_r H^0 = 49.2 \text{ kJ}\cdot\text{mol}^{-1}$$

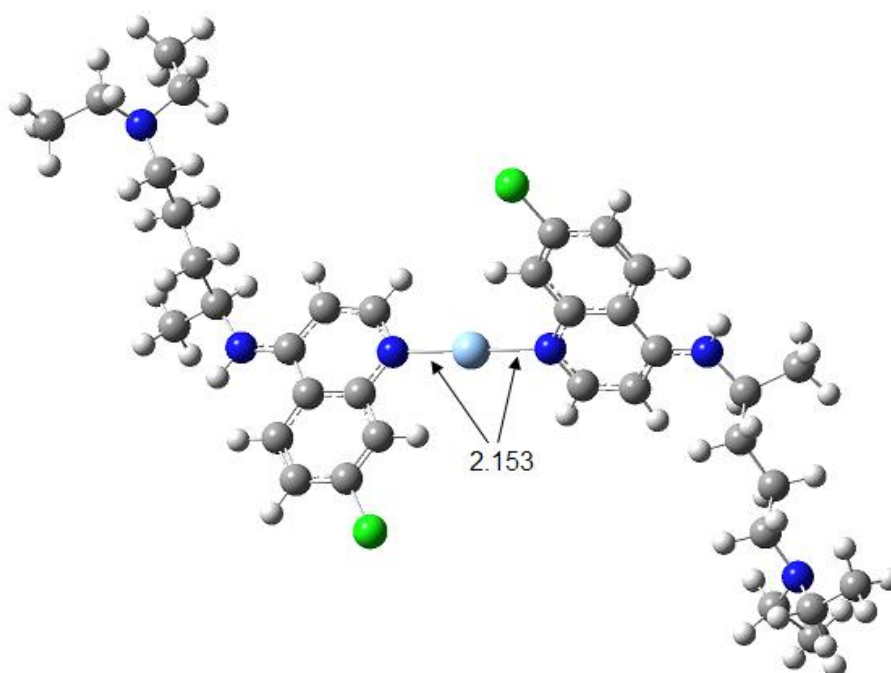
CQ_2Ag^+ cations



CQ_2Ag^+ “tail-tail” (N sp^3 –Ag–N sp^3 bonds),
 $\Delta_r H^0 = 307.2 \text{ kJ}\cdot\text{mol}^{-1}$



CQ₂Ag⁺ “head-tail” (N sp²–Ag–N sp³ bonds),
 $\Delta_r H^0 = 356.5 \text{ kJ}\cdot\text{mol}^{-1}$



CQ₂Ag⁺ “head-head” (N sp²–Ag–N sp² bonds),
 $\Delta_r H^0 = 407.1 \text{ kJ}\cdot\text{mol}^{-1}$

S3. ESI/FT-ICR Mass Spectrometry

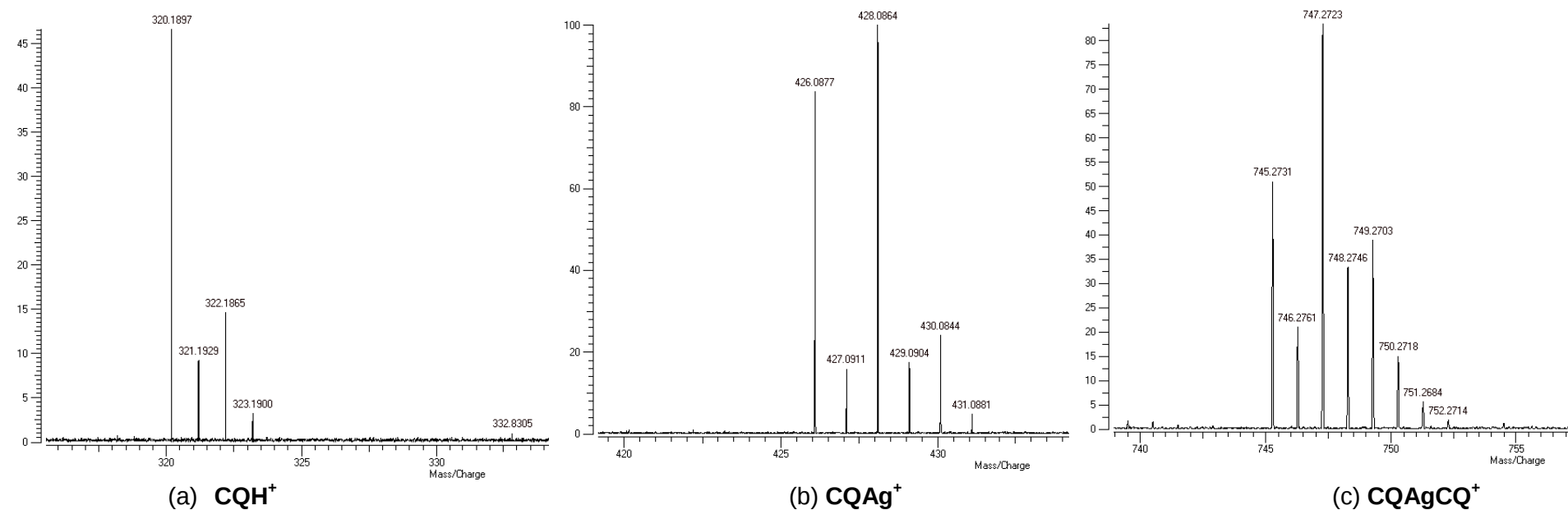


Figure S4a. ESI/FT-ICR spectrum of isolated cations of CQ/Ag complex, solved in DMSO (100 µg/mL), showing their isotopic mass distribution

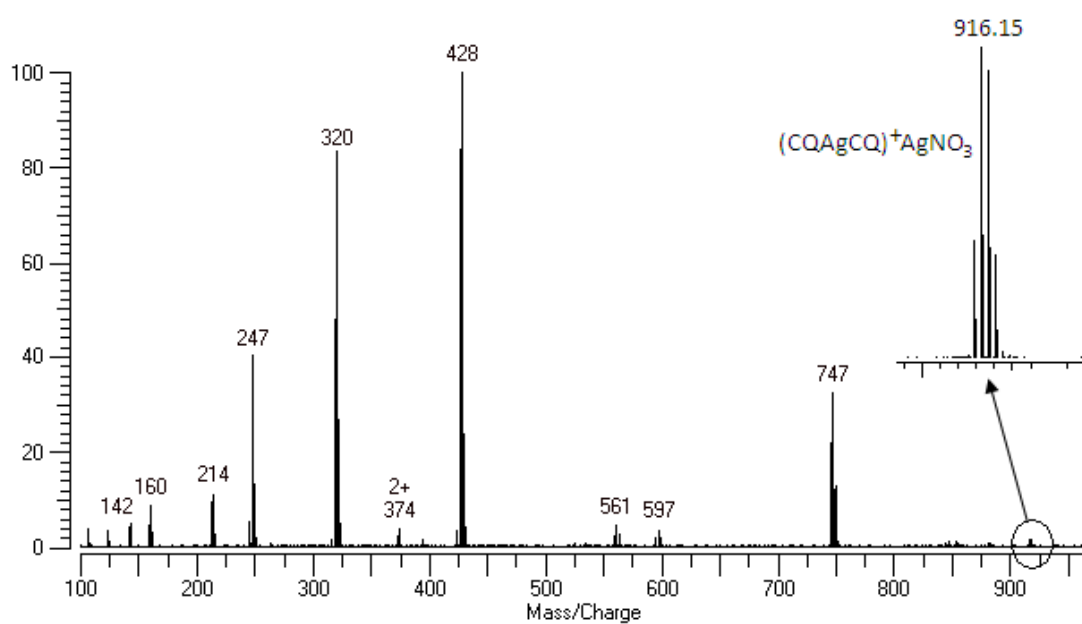


Figure S4b. ESI/FT-ICR spectrum of **CQ/Ag** complex, solved in DMSO:ACN (4:96, 100 $\mu\text{g/mL}$).