Near-Infrared Light Controlled Photocatalytic Activity of Carbon Quantum Dots for Highly Selective Oxidation Reaction

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Supporting Information

1. Figures

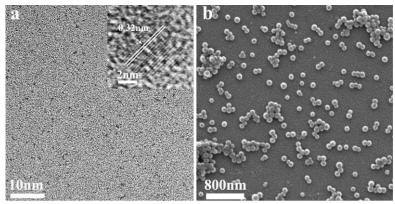


Figure S1 (a) TEM image of CQDs, inset is the HRTEM image; (b) SEM image of CNPs.

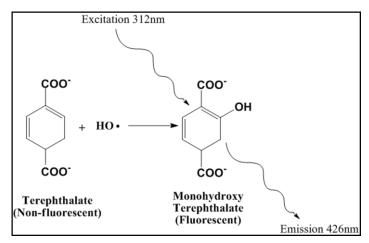


Figure S2. Reaction between hydroxyl radicals and terephthalate yielding one intensely fluorescent mono-hydroxylated isomer.

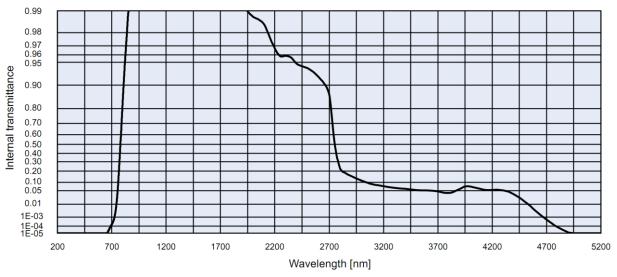


Figure S3 The wavelength distribution of light after filtered by Schott optical filter RG780.

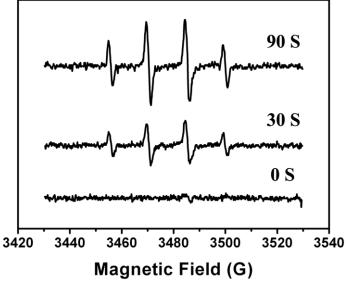


Figure S4. The ESR signals of the DMPO-•OH (aqueous solution) adducts for CQDs/H₂O₂ system under NIR light irradiation. [CQDs]=4 mg/50 mL; [H₂O₂]= 10mM; [DMPO]=40mM.

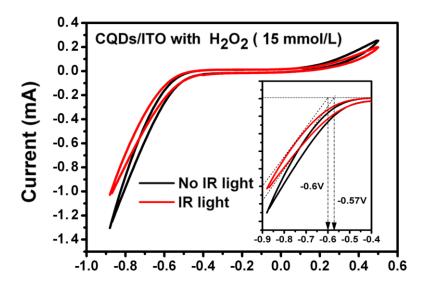


Figure S5 Cyclic voltammograms of the CQDs/ITO electrode in 0.05 M (pH 7.4) phosphate buffer and in the presence of 15 mM of H_2O_2 under NIR light irradiation or not.

2.Table

Table S1 Selective oxidation of alcohols to aldehydes and ketones without NIR light irradiation and CQDs catalyst. [a]

Entry	Reactants	Products	Conv. [%]
1	CH₂OH	СНО	21
2	CI CH ₂ OH	СНО	18
3	CH ₂ OH	СНО	22
4	H ₃ C CH ₂ OH	н ₃ С СНО	20
5	O ₂ N CH ₂ OH	O ₂ N CHO	17
6	H ₃ OC CH ₂ OH	H ₃ OC CHO	25
7	OH C-CH ₃	о С-сн ₃	16
8	0 C-O-CH ₂ OH	о Ё-о-сно	21

[[]a] 10 mmol alcohol, H_2O_2 (10 mmol, 30 wt% in water), no catalyst, 60° C, 12 h without NIR light irradiation. H_2O_2 was added continuously over 12 h.

Table S2. Selective photocatalytic oxidation of benzyl alcohol and its derivatives into corresponding aldehydes in the presence of CQDs and H_2O_2 under irradiation with NIR light after three cycles.^[a]

Entry	Reactants	Products	Conv. [%]	Sel. [%]
1	CH ₂ OH	СНО	90	100
2	CI CH ₂ OH	CICHO	84	>99
3	F CH ₂ OH	F CHO	89	>98
4	H ₃ C CH ₂ OH	H ₃ C CHO	87	>98
5	O ₂ N CH ₂ OH	O ₂ N CHO	86	>99
6	H ₃ OC CH ₂ OH	H ₃ OC CHO	86	>99
7	OH C-CH ₃	O C-CH ₃	87	>96
8	O C-O-CH ₂ OH	C-O-CHO	87	>99

 $^{^{[}a]}$ 10 mmol alcohol, H_2O_2 (10 mmol, 30 wt% in water), 8 mg CQDs catalyst, 60°C, followed by NIR irradiation for 12 h. H_2O_2 was added continuously over 12 h.