

- Supplementary Information -

Light-emitting Ga-oxide nanocrystals in glass: a new paradigm for low-cost and robust UV-to-visible solar-blind converters and UV-emitters

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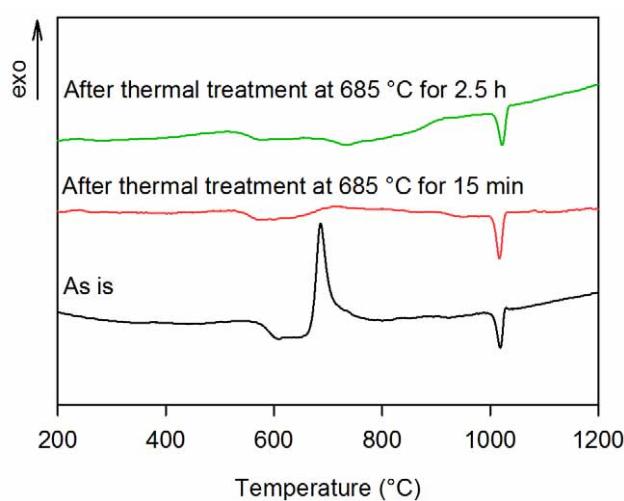


Figure S1. Differential-scanning-calorimetry data. Representative DSC curves before and after 15 min treatment and after prolonged treatment of 2.5 hours at the temperature of the exothermic peak of crystallization of the γ -Ga₂O₃ nanophase in 7.5Li₂O-2.5Na₂O-20Ga₂O₃-35GeO₂-35SiO₂ with 0.1 mol% NiO.

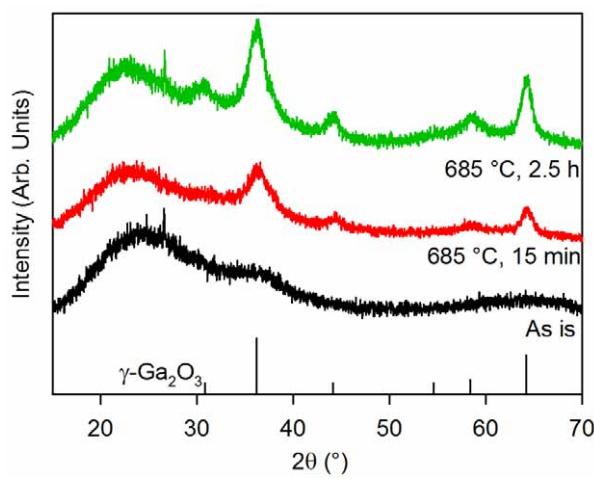


Figure S2. X-ray-diffraction data. Representative XRD patterns before and after 15 min treatment and after prolonged treatment of 2.5 hours at the temperature of the exothermic peak of γ -Ga₂O₃ crystallization in 7.5Li₂O-2.5Na₂O-20Ga₂O₃-35GeO₂-35SiO₂ with 0.1 mol% NiO. Analogous changes are observed in undoped and heavier doped (1 mol% NiO) glass after thermal treatment at the exothermic peak. The pattern of treated material comprises the broad amorphous halo from the glass matrix (with maximum between 20 and 30°) and reflections ascribable to γ -Ga₂O₃ nanocrystals (stick pattern according to JCPDS card No. 20-0426). The narrow weak reflection at about 27° is caused by quartz impurities introduced during sample preparation for XRD analysis using agate mortar.

Estimation of nanocrystal concentration n_{NC} from the mean NC radius R and the nanoparticle volume fraction f_V .

The volume occupied by all $\gamma\text{-Ga}_2\text{O}_3$ nanocrystals (NCs) s is equal to the mean NC volume $V_{NC}=4\pi R^3/3$ (approximated as a sphere) multiplied by the number N_{NC} of NCs. So, f_V may be written as $N_{NC}V_{NC}/V_{GC}$, where V_{GC} is the total volume occupied by the glassceramic sample. Since the concentration n_{NC} of NCs per unit volume is the ratio N_{NC}/V_{GC} between the total number of NCs in the sample and the total sample volume, we obtain that $n_{NC}=f_V/(4\pi R^3/3)$. On the other hand, the volume fraction f_V is also related to the molar fraction f through the relation $f_V=f(M_{\text{Ga}_2\text{O}_3}/M_{\text{GC}})(\rho_{\text{GC}}/\rho_{\text{Ga}_2\text{O}_3})$, where $M_{\text{Ga}_2\text{O}_3}$ and M_{GC} are the molar mass (in g/mol) of $\gamma\text{-Ga}_2\text{O}_3$ and glassceramic material, respectively, and $\rho_{\text{Ga}_2\text{O}_3}$ and ρ_{GC} the density (in g/cm³) of nanoparticle and material. Therefore, n_{NC} may also be estimated from f by using information on molar mass and density of nanoparticle and material. For instance, taking $f=0.2$ (supposing 20 mol% of crystallized nanoparticle in the material), $M_{\text{Ga}_2\text{O}_3}$ and M_{GC} 187.5 and 98.9 g/mol, respectively, $\rho_{\text{Ga}_2\text{O}_3}$ and ρ_{GC} 6.05 and 3.66 g/cm³, respectively, and $R=3$ nm, we have $f_V=0.23$ and $V_{NC}=1.13\times 10^{-19}$ cm³. Finally, from f_V and V_{NC} , we obtain $n_{NC}=2\times 10^{18}$ cm⁻³, i.e. 2 million NCs per μm^3 . From $(1/n_{NC})^{1/3}$ we can also give a rough estimation of the inter-nanoparticle mean distance, which in this example turns out to be about 8 nm.

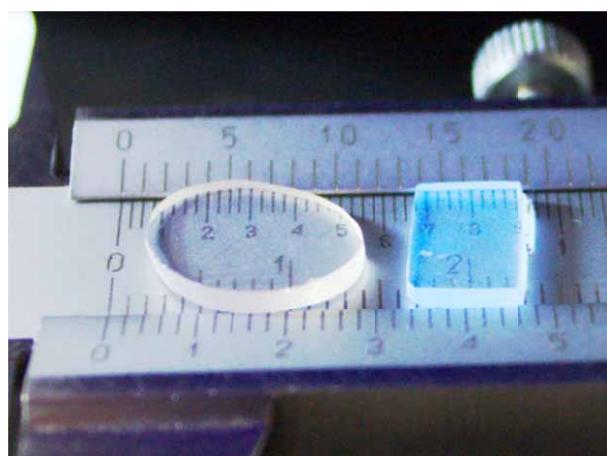


Figure S3. Fluorescence under UV-C illumination. Untreated (left) and treated nanostructured (right) samples of 7.5Li₂O-2.5Na₂O-20Ga₂O₃-35GeO₂-35SiO₂ glass under UV-C illumination of a mercury lamp.

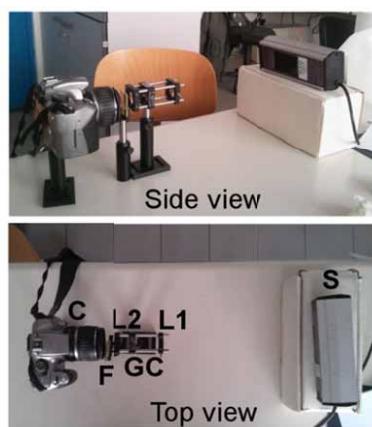


Figure S4. Lab-version of optics for UV-to-visible converter. The adopted experimental set-up to collect images in Figure 7 of the article consists of an UV source (S), a lens (L1) that focuses the image on the glassceramic plate (GC), a second lens (L2) that collects the light coming from the focus in GC, a UV cut-off filter (F) and a digital camera (C).