

Supplementary information

Fabrication of Large Scale Single Crystal Bismuth Telluride (Bi₂Te₃) Nanosheet Arrays by Single Step Electrolysis Process

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Calcualtion of the bond energies of Te⁽¹⁾-Bi, Te⁽²⁾-Bi, and Te⁽¹⁾- Te⁽¹⁾ in Bi₂Te₃ :

According to the calculation results of Kaviany and co-workers, the calculated bond energies of Te⁽¹⁾-Bi, Te⁽²⁾-Bi, and Te⁽¹⁾- Te⁽¹⁾ in Bi₂Te₃ can be obtained as shown in below.

For Te⁽¹⁾ atom: bonds with three Bi and three Te⁽¹⁾ atoms

$$\Rightarrow 3 \times 0.974 + 3 \times 0.0691 = 3.1293 \text{ eV}$$

For Te⁽²⁾ atom: bonds with six Bi atoms

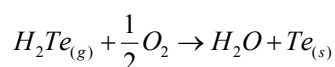
$$\Rightarrow 6 \times 0.5801 = 3.4806 \text{ eV}$$

For Bi atom: bonds with three Te⁽¹⁾ and three Te⁽²⁾ atoms

$$\Rightarrow 3 \times 0.974 + 3 \times 0.5801 = 4.6623 \text{ eV}$$

Existence of H₂Te gas during the electrolysis process

In order to prove the existence of H₂Te gas during the electrolysis process, we design an experiment as shown in Figure S1 (a). A Si substrate was suspended upon the electrolyte while doing the electrolysis process and the tellurium was formed by the decomposition of H₂Te gas^[1]. Figure S1 (b) shows the schematics of formation of H₂Te derived Te, and the reaction process can be visualized as below:



The H_2Te derived Te was distinguished by Raman spectrum. In the Raman spectrum of H_2Te derived Te, three peaks are located at 92.4, 121.6, and 141.1 cm^{-1} , corresponding to optical modes of E^l , A_1 , and E^u , respectively^[2], as shown in Figure S1(d). These three peaks are consistent with the results from pure Te ingot. The SEM image of H_2Te derived Te is shown in Figure S1(c) and reveals the blade shape.

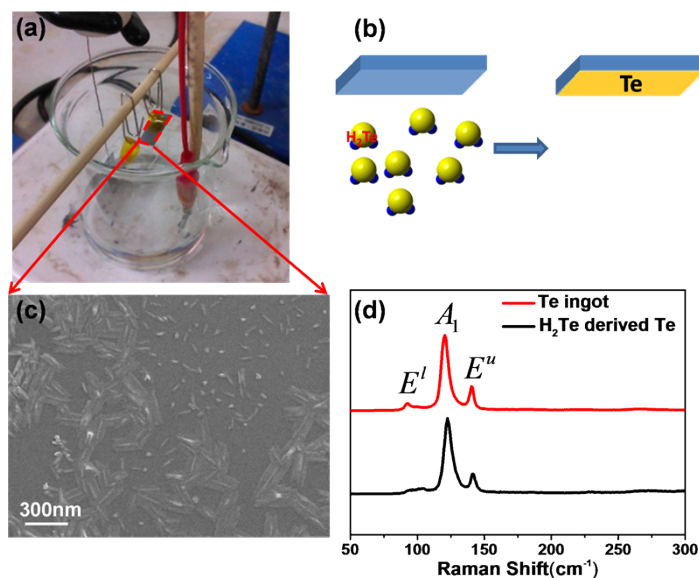


Figure S1. (a) Suspended Si substrate upon the electrolyte. (b) Schematics of H_2Te derived Te. (c) SEM image of H_2Te derived Te. (d) Raman spectrum of Te ingot and H_2Te derived Te.

REFERENCES

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- [2] B. H. Torrie, *Solid State Communications* **1970**, 8, 1899-1901.