

Electronic Supplementary Information

Self-assembled 3D Pt/TiO₂ architecture for high-performance photocatalytic hydrogen production

Haiyan Li,^a Hongwen Yu,^{*a} Lei Sun,^a Jiali Zhai^a and Xuerong Han^{*b}

^aKey Laboratory of Wetland Ecology and Environment, Northeast Institute of Geography and Agroecology, Chinese Academy of Sciences, Changchun 130102, China

^bFaculty of Advanced Life Science, Hokkaido University, Sapporo 001-0021, Japan

*To whom correspondence should be addressed. E-mails: yuhw@neigae.ac.cn, hxr@mail.sci.hokudai.ac.jp



Fig. S1 The photograph of the experimental setup of the photocatalytic H₂ production.

The photacatalytic on-line analysis system mainly includes the following sections: a Pyrex reaction cell, a closed gas circulation and evacuation system, Xe lamp, gas chromatography, and computer.

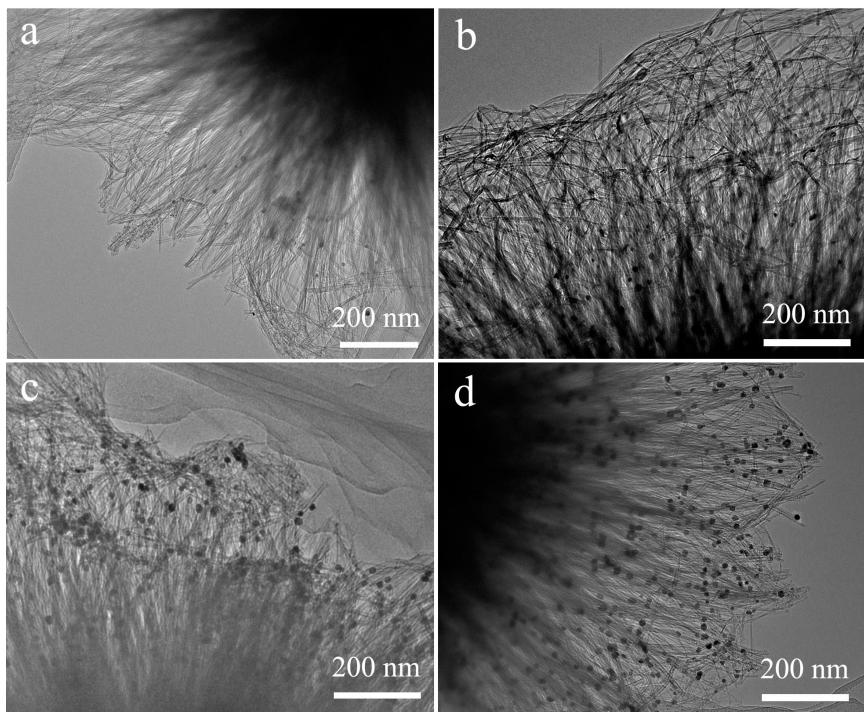


Fig. S2 TEM images of the as-prepared Pt/TiO₂ architecture with different amounts of Pt. (a) 0.2 wt %, (b) 0.3 wt %, (c) 0.5 wt % and (d) 0.8 wt %.

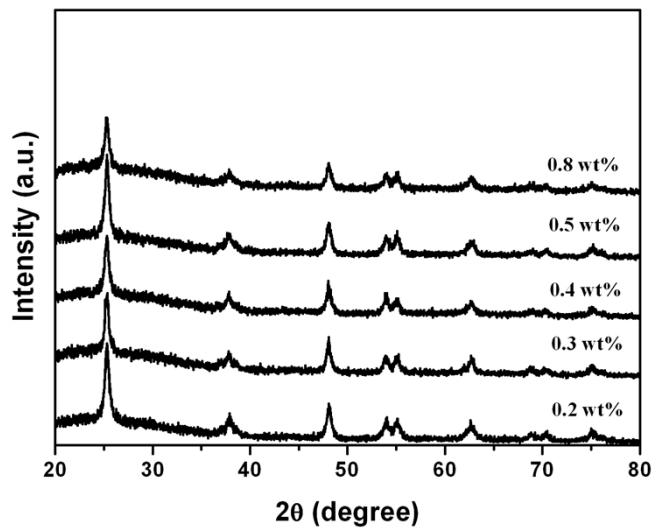


Fig. S3 XRD patterns of the 3D Pt/TiO₂ architecture with different Pt loading amounts.

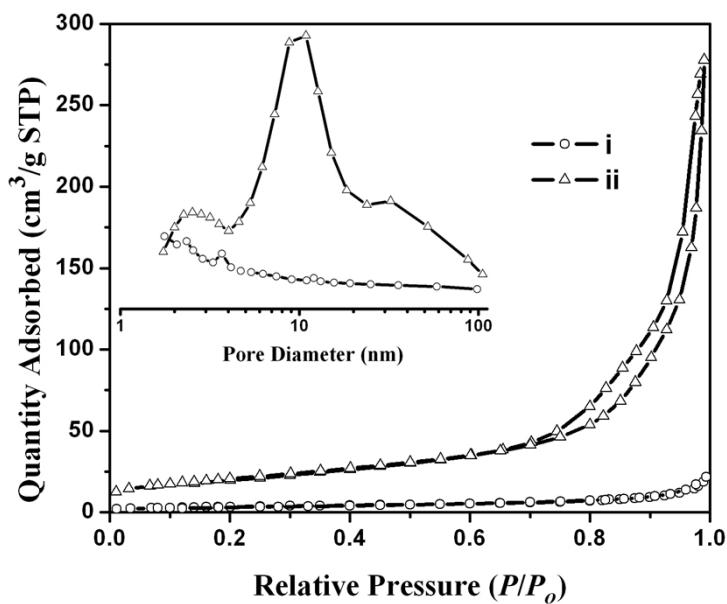


Fig. S4 Nitrogen adsorption-desorption isotherms and the corresponding pore size distribution curves (inset) of the TiO_2 nanowire (i) and 3D Pt/ TiO_2 architecture (ii).

Table S1. Structural parameters of the 3D Pt/TiO₂ architecture with different Pt loading amounts.

Sample	^a wt % (Pt/TiO ₂)	<i>S</i> _{BET} ^b (m ² /g)	<i>APS</i> ^c (nm)	<i>V</i> _p ^d (cm ³ /g)	<i>R</i> ^e (μmol/h)
TiO ₂ nanowire	0	12.2	7.38	0.022	3.7
0.2 wt % Pt/TiO ₂	0.19	86.6	13.6	0.25	380
0.3 wt % Pt/TiO ₂	0.28	86.1	14.4	0.26	523
0.4 wt % Pt/TiO ₂	0.36	87.1	13.6	0.25	667
0.5 wt % Pt/TiO ₂	0.41	89.2	13.1	0.25	571
0.8 wt % Pt/TiO ₂	0.59	88.8	12.4	0.24	537

^a wt % determined by ICP-OES data.

^b*S*_{BET} denotes specific surface area.

^cAPS denotes average pore size.

^d*V*_p denotes pore volume.

^e*R* denotes the H₂-production rate of the photocatalyst samples.

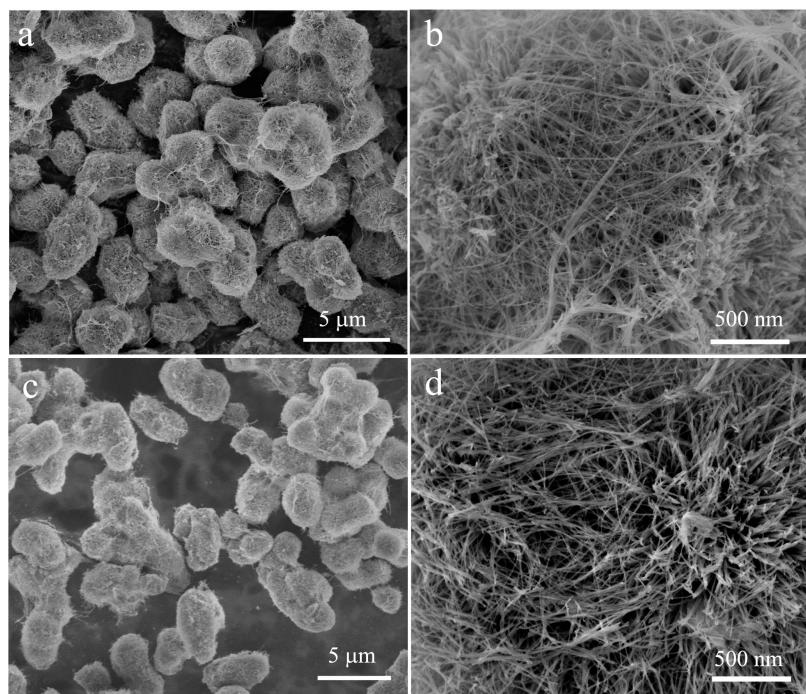


Fig. S5 SEM images of the 0.4 wt % Pt/TiO₂ architecture obtained with different reaction times.
(a,b) 5 h; (c,d) 10 h.

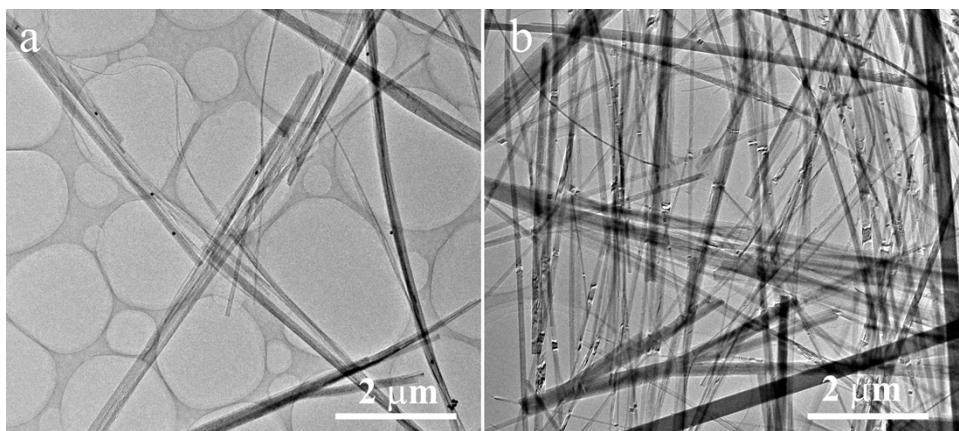


Fig. S6 TEM images of the products obtained under different reaction conditions. (a) $\text{Ti}(\text{OC}_4\text{H}_9)_4 + \text{ET} + \text{NaOH} + \text{H}_2\text{PtCl}_6$, (b) $\text{Ti}(\text{OC}_4\text{H}_9)_4 + \text{EG} + \text{NaOH}$.

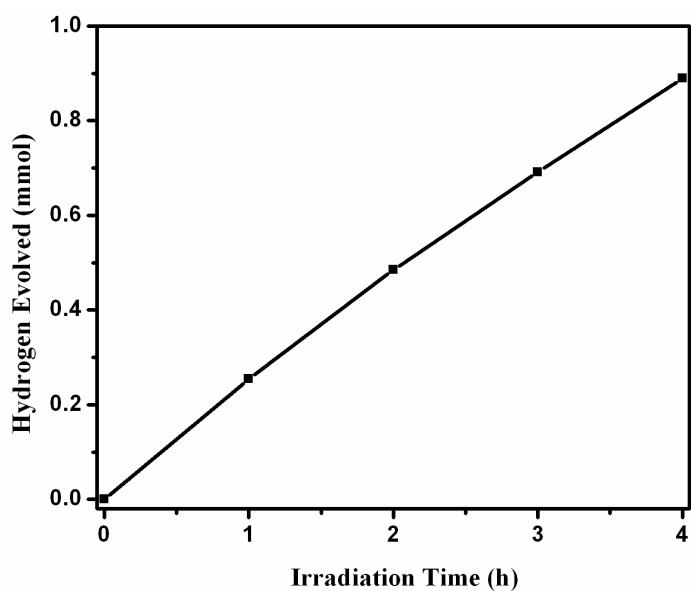


Fig. S7 Time-dependent hydrogen evolution over 0.4 wt % Pt/P25 photocatalyst. Reaction conditions: catalyst, 50 mg; 20 vol % CH₃OH solution, 50 ml; light source, 300 W Xe lamp with a 365 nm filter.

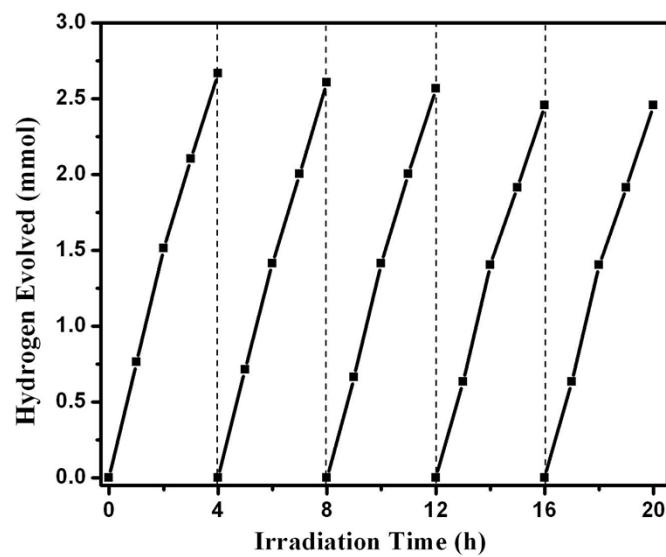


Fig. S8 Cycling performance of 0.4 wt % 3D Pt/TiO₂ architecture.

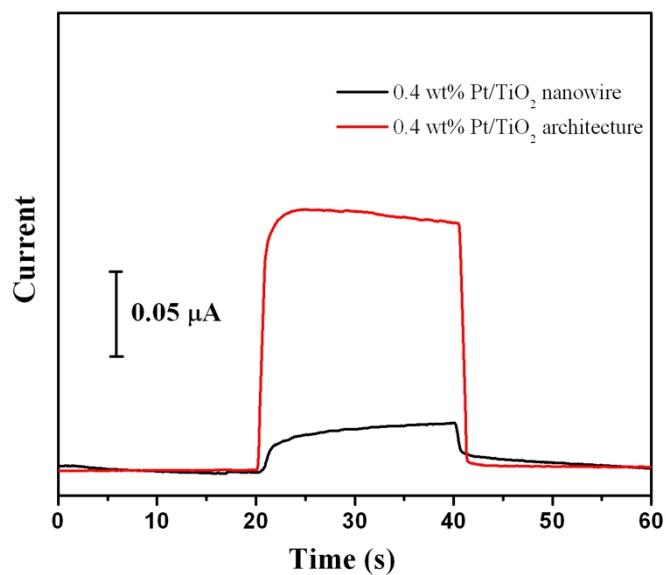


Fig. S9 Transient photocurrent responses of 0.4 wt % 3D Pt/TiO₂ architecture and nanowire irradiation with UV light for 20 s under an electrode potential of 0 V versus Ag/AgCl.

Movie S1: Hydrogen evolution over 20 mg of 0.4 wt % 3D Pt/TiO₂ photocatalyst in 20 vol % CH₃OH solution irradiated under a 300 W Xe lamp with a 365 nm filter (light intensity was about 20 mW/cm²).