Supporting Information

Geometry-Dependent Divergence in Gold-Catalyzed Redox Cascade Cyclization of *o-*Alkynylaryl Ketoximes and Nitrones Leading to Isoindoles

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Table S1. Screening of Solvents in the Conversion of (Z)-1e into 3e

entry ^a	solvent	conversion (%) ^b	yield (%) ^b
1	DMF	39	20
2	CH ₃ CN	49	22
3	CH_3NO_2	>95	70
4	CHCl ₃	>95	65
5	MeOH	>95	58
6	ClCH ₂ CH ₂ Cl	>95	64
7	THF	83	53
8	toluene	93	65

^aConditions: Au(IMes)OTf (5 mol %) at 70 °C in solvents (sealed tube); after 4 h, the reaction was quenched with Et₃N. ^bCrude NMR yield.

Table S2. Screening of Ligands in the Conversion of (Z)-1e into 3e

entry ^a	ligand	conversion (%) ^a	yield (%) ^a
1	PPh ₃	74	53
2	$P(o-toly)_3$	28	13
3	^t Bu ₂ P(o-biphenyl)	>95	62
4	^t Cy ₂ P(<i>o</i> -biphenyl)	>95	53
5	$P(C_6F_5)_3$	31	29
6	$P^{t}Bu_{3}$	79	54
7	IPr	>95	69

^aConditions: Au(L)OTf (5 mol %) at 70 °C in CH₂Cl₂ (sealed tube) (reaction time: 0.5-1.5 h)

X-Ray Crystallography Data for 3e:

For each determination a suitable crystal was mounted on a Bruker SMART diffractometer equipped with a graphite monochromated Mo_{Ka} ($\lambda = 0.71073$ Å) radiation source and a CCD detector and 45 frames of two-dimensional diffraction images were collected and processed to deduce the cell parameters and orientation matrix. The frame data were processed to give structure factors using the program SAINT. [1] The intensity data were corrected for Lorentz and polarization effects. The structures were solved by a combination of direct and difference Fourier methods provided by the program package SHELXTL, [2] and refined using a full matrix least square against F^2 for all data. All non-H atoms were refined anisotropically. All hydrogen atoms were included in calculated positions with isotropic thermal parameters 1.2 times those of attached atoms. (Reference: [1] Bruker, *SMART*, *SAINT* and *XPREP*: Area Detector Control and Integration Software Ver. 5.054; Bruker Analytical X-ray Instruments: Madison, Wisconsin, 2001. [2] Bruker, *SHELXTL*: Structure Determination Programs, Ver. 6.12; Bruker Analytical X-ray Instruments: Madison, Wisconsin, 2000.)

Crystallographic data for compound **3e**: Triclinic, space group P-1, a = 12.3395(5) Å, b = 16.6385(9) Å, c = 16.9325(7) Å, $\alpha = 102.905(3)$ °, $\beta = 101.820(2)$ °, $\gamma = 111.464(2)$ ° V = 2992.1(2) Å³, T = 146 K, Z = 8, $Mo_{K\alpha}$ radiation, graphite monochromator, scan width 0.3° in ω , 10 s irradiation time per measurement reflections, 12311 measurements, 24963 measured reflections, 12311 independent reflections, 766 parameters, GOF = 1.068, $R_1(I > 2\sigma(I)) = 0.0715$, $wR_2 = 0.1878$.

Table S3. Selected bond lengths $[\mathring{A}]$ and angles $[^{\circ}]$ for **3e**.

O(1)-C(15)	1.236(3)	
N(1)-C(1)	1.337(3)	
N(1)-C(8)	1.381(3)	
C(1)-C(2)	1.417(3)	
C(1)-C(9)	1.467(3)	
C(2)-C(3)	1.410(3)	
C(2)-C(7)	1.434(3)	
C(3)-C(4)	1.369(3)	
C(4)-C(5)	1.413(3)	
C(5)-C(6)	1.365(3)	
C(6)-C(7)	1.417(3)	
C(7)-C(8)	1.411(3)	
C(8)-C(15)	1.440(3)	
C(9)-C(10)	1.389(3)	
C(9)-C(14)	1.402(3)	
C(10)-C(11)	1.387(3)	
C(11)-C(12)	1.383(4)	
C(12)-C(13)	1.381(4)	
C(13)-C(14)	1.386(3)	
C(15)-C(16)	1.515(3)	
C(16)-C(17)	1.518(3)	
C(17)-C(18)	1.524(3)	
C(18)-C(19)	1.524(3)	
C(1)-N(1)-C(8)	112.26(18)	
N(1)-C(1)-C(2)	107.47(18)	
N(1)-C(1)-C(9)	123.65(19)	
C(2)-C(1)-C(9)	128.8(2)	
C(3)-C(2)-C(1)	131.9(2)	
C(3)-C(2)-C(7)	121.21(19)	
C(1)-C(2)-C(7)	106.85(19)	
C(4)-C(3)-C(2)	118.5(2)	
C(3)-C(4)-C(5)	120.8(2)	
C(6)-C(5)-C(4)	121.9(2)	

C(5)-C(6)-C(7)	119.3(2)	
C(8)-C(7)-C(6)	134.8(2)	
C(8)-C(7)-C(2)	106.96(18)	
C(6)-C(7)-C(2)	118.24(19)	
N(1)-C(8)-C(7)	106.45(18)	
N(1)-C(8)-C(15)	119.66(19)	
C(7)-C(8)-C(15)	133.9(2)	
C(10)-C(9)-C(14)	118.8(2)	
C(10)-C(9)-C(1)	121.1(2)	
C(14)-C(9)-C(1)	120.1(2)	
C(11)-C(10)-C(9)	120.4(2)	
C(12)-C(11)-C(10)	120.6(2)	
C(13)-C(12)-C(11)	119.6(2)	
C(12)-C(13)-C(14)	120.4(3)	
C(13)-C(14)-C(9)	120.3(2)	
O(1)-C(15)-C(8)	121.8(2)	
O(1)-C(15)-C(16)	121.2(2)	
C(8)-C(15)-C(16)	117.06(19)	
C(15)-C(16)-C(17)	115.31(18)	
C(16)-C(17)-C(18)	111.35(18)	
C(19)-C(18)-C(17)	113.7(2)	































































































































































































