

## Supporting information

# Facile preparation and cell imaging applications of fluorescent organic nanoparticles that combine AIE dye and ring-opening polymerization

Xiqi Zhang\*<sup>a</sup>, Xiaoyong Zhang<sup>a</sup>, Bin Yang<sup>a</sup>, Junfeng Hui<sup>a</sup>, Meiyong Liu<sup>b</sup>, Zhenguo Chi<sup>c</sup>, Siwei Liu<sup>c</sup>, Jiarui Xu<sup>c</sup>, Yen Wei\*<sup>a</sup>

<sup>a</sup> Department of Chemistry and the Tsinghua Center for Frontier Polymer Research, Tsinghua University, Beijing, 100084, P. R. China. <sup>b</sup> Beijing National Laboratory for Molecular Sciences (BNLMS), Key Laboratory of Organic Solids, Laboratory of New Materials, Institute of Chemistry, Chinese Academy of Sciences, Beijing 100190, China. <sup>c</sup> PCFM Lab, DSAPM Lab and KLGHEI of Environment and Energy Chemistry, FCM Institute, State Key Laboratory of Optoelectronic Materials and Technologies, School of Chemistry and Chemical Engineering, Sun Yat-sen University, Guangzhou 510275, China.

[sychy Zhang@126.com](mailto:sychy Zhang@126.com); [weiyen@tsinghua.edu.cn](mailto:weiyen@tsinghua.edu.cn)

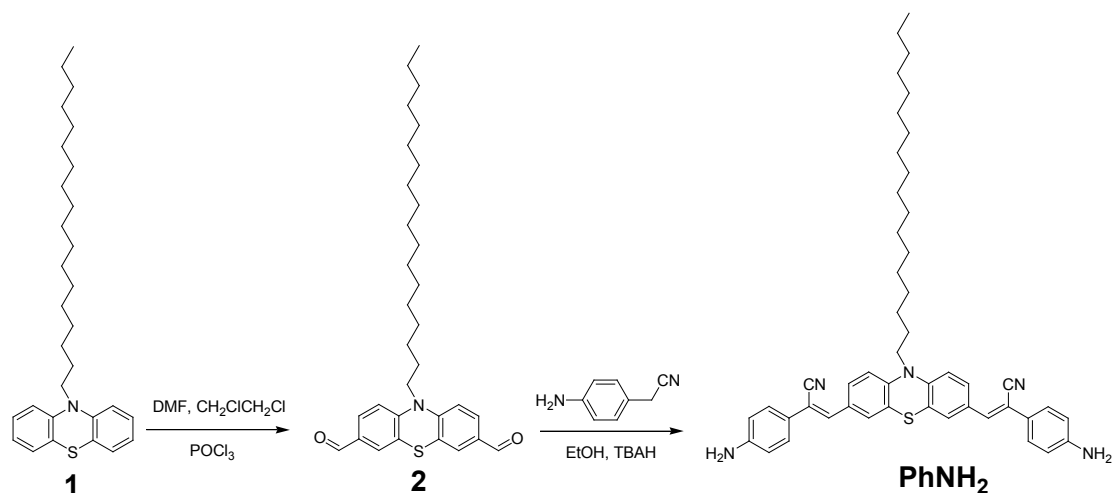
## 1. Experimental

### 1.1 Materials and measurements

Phenothiazine, 1-bromooctadecane, N,N-dimethylformamide (DMF), 1,2-dichloroethane, phosphoryl chloride, 4-aminobenzyl cyanide, tetrabutylammonium hydroxide (0.8M in methanol), N,N-dimethylacetamide (DMAc), 4,4'-Oxydiphthalic anhydride purchased from Alfa Aesar were used as received. All other agents and solvents were purchased from commercial sources and used directly without further purification. Tetrahydrofuran (THF) was distilled from sodium/benzophenone. Ultra-pure water was used in the experiments.

<sup>1</sup>H NMR spectra were measured on a JEOL 400 MHz spectrometer [CDCl<sub>3</sub> or d<sub>6</sub>-DMSO as solvent and tetramethylsilane (TMS) as the internal standard]. HRMS was obtained on Shimadzu LCMS-IT-TOF high resolution mass spectrometry. UV-Visible absorption spectra were recorded on UV/Vis/NIR Perkin-Elmer lambda750 spectrometer (Waltham, MA, USA) using quartz cuvettes of 1 cm path length. Fluorescence spectra were measured on a PE LS-55 spectrometer with a slit width of 3 nm for both excitation and emission. The FT-IR spectra were obtained in a transmission mode on a Perkin-Elmer Spectrum 100 spectrometer (Waltham, MA, USA). Typically, 8 scans at a resolution of 1 cm<sup>-1</sup> were accumulated to obtain one spectrum. Transmission electron microscopy (TEM) images were recorded on a JEM-1200EX microscope operated at 100 kV, the TEM specimens were made by placing a drop of the nanoparticles suspension on a carbon-coated copper grid. The size distribution and zeta potential measurement of **RO-OA** FONs in phosphate buffer solution (PBS) were determined using a zeta Plus apparatus (ZetaPlus, Brookhaven Instruments, Holtsville, NY). Gel permeation chromatography (GPC) analyses of polymers were performed using DMF as the eluent. The GPC system was a Shimadzu LC-20AD pump system comprising of an auto injector, a MZ-Gel SDplus 10.0 mm guard column (50×8.0 mm, 10<sup>2</sup> Å) followed by a MZ-Gel SDplus 5.0 μm bead-size columns (50-10<sup>6</sup> Å, linear) and a Shimadzu RID-10A refractive index detector. The system was calibrated with narrow molecular weight distribution polystyrene standards ranging from 200 to 10<sup>6</sup> g mol<sup>-1</sup>.

### 1.2 Synthesis and characterization of PhNH<sub>2</sub>



Scheme S1 Synthetic route of **PhNH<sub>2</sub>**.

The intermediate 10-octadecyl-10H-phenothiazine (**1**) was synthesized according to the literature.<sup>1</sup>

In a 500 mL flask, POCl<sub>3</sub> (24.5 g, 0.160 mol) and DMF (7.8 g, 0.106 mol) were added to CH<sub>2</sub>ClCH<sub>2</sub>Cl (20 mL) at 0 °C. After stirring the reaction mixture for 30 minutes, the intermediate **1** (6.0 g, 13.3 mmol) was added. The reaction mixture was heated at 90 °C overnight. The crude product was added dropwise into 600 mL cool water and extracted with ethyl acetate. The pure 10-octadecyl-10H-phenothiazine-3,7-dicarbaldehyde (**2**) was isolated by flash column chromatography using a 1:1 mixture of dichloromethane and petroleum ether as an eluent (2.5 g, yield 37%). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ: 0.87 (t, 3H, J = 6.8 Hz), 1.20-1.34 (m, 28H), 1.38-1.48 (m, 2H), 1.82 (quint, 2H, J = 7.6 Hz), 3.92 (t, 2H, J = 7.2 Hz), 6.94 (d, 2H, J = 8.4 Hz), 7.58 (d, 2H, J = 2.0 Hz), 7.66 (dd, 2H, J<sub>1</sub> = 8.4 Hz, J<sub>2</sub> = 1.6 Hz), 9.81 (s, 2H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ (ppm): 189.97, 149.03, 132.09, 130.21, 128.61, 124.54, 115.66, 48.60, 32.01, 29.79, 29.76, 29.75, 29.68, 29.57, 29.45, 26.79, 22.78, 14.22. HRMS calcd. for C<sub>32</sub>H<sub>45</sub>NO<sub>2</sub>S, [M+H]<sup>+</sup>: 508.3244, found 508.3239.

A solution of **2** (0.81 g, 1.6 mmol) and 2-(4-aminophenyl)acetonitrile (0.84 g, 3.2 mmol) in ethanol (20 mL) was stirred at room temperature. Then tetrabutyl ammonium hydroxide solution (0.8 M, 10 drops) was added and the mixture was heated to reflux for 2 h precipitating a red solid. The reaction mixture was cooled to room temperature and filtered, washed with ethanol for several times obtaining a dark red solid **PhNH<sub>2</sub>** (0.97 g, yield 83%). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ (ppm): 0.87 (t, 3H, J = 6.8 Hz), 1.21-1.32 (m, 28H), 1.39-1.46 (m, 2H), 1.81 (quint, 2H, J = 7.2 Hz), 3.84 (t, 2H, J = 8.0 Hz), 6.69 (d, 4H, J = 8.4 Hz), 6.85 (d, 2H, J = 8.4 Hz), 7.17 (s, 2H),

7.38-7.50 (m, 6H), 7.76 (dd, 2H,  $J_1 = 6.4$  Hz,  $J_2 = 2.4$  Hz);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm): 147.27, 145.27, 136.99, 129.10, 129.03, 128.28, 128.14, 127.12, 124.84, 123.80, 118.73, 115.34, 115.20, 109.33, 48.06, 32.03, 29.81, 29.79, 29.77, 29.74, 29.68, 29.64, 29.47, 29.33, 26.97, 26.75, 22.80, 14.25; HRMS calcd. for  $\text{C}_{48}\text{H}_{57}\text{N}_5\text{S}$ ,  $[\text{M}+\text{H}]^+$ : 736.4407, found 736.4411.

### 1.3 Preparation of RO-OA FONs

$\text{PhNH}_2$  (37 mg, 0.05 mmol), 4,4'-oxydiphthalic anhydride (17 mg, 0.055 mmol) were dissolved in 10 mL DMAc. The above mixture was stirred under air atmosphere at room temperature for 2h. Then the reaction of polymerization was stopped and dialyzed against tap water for 24 h and ethanol for 6 h using 3500 Da Mw cutoff dialysis membranes. Finally, this solution in dialysis bag was carried out by freeze-drying to obtain the product.

### 1.4 Cytotoxicity of RO-OA FONs

Cell morphology was observed to examine the effects of **RO-OA** FONs to A549 cells. Briefly, cells were seeded in 6-well microplates at a density of  $1 \times 10^5$  cells  $\text{mL}^{-1}$  in 2 mL of respective media containing 10% fetal bovine serum (FBS). After cell attachment, plates were washed with PBS and cells were treated with complete cell culture medium, or different concentrations of **RO-OA** FONs prepared in 10% FBS containing media for 24 h. Then all samples were washed with PBS three times to remove the uninternalized nanoparticles. The morphology of cells was observed by using an optical microscopy (Leica, Germany), the overall magnification was  $\times 100$ .

The cell viability of **RO-OA** FONs on A549 cells was evaluated by cell counting kit-8 (CCK-8) assay. Briefly, cells were seeded in 96-well microplates at a density of  $5 \times 10^4$  cells  $\text{mL}^{-1}$  in 160  $\mu\text{L}$  of respective media containing 10% FBS. After 24 h of cell attachment, the cells were incubated with 10, 20, 40, 80, 120  $\mu\text{g mL}^{-1}$  **RO-OA** FONs for 8 and 24 h. Then nanoparticles were removed and cells were washed with PBS three times. 10  $\mu\text{L}$  of CCK-8 dye and 100  $\mu\text{L}$  of DMEM cell culture medium were added to each well and incubated for 2 h at 37 °C. Plates were then analyzed with a microplate reader (VictorIII, Perkin-Elmer). Measurements of formazan dye absorbance were carried out at 450 nm, with the reference wavelength at 620 nm. The values were proportional to the number of live cells. The percent reduction of CCK-8 dye was compared to controls (cells not exposure to **RO-OA** FONs), which represented 100% CCK-8 reduction. Three replicate wells were used per microplate, and the experiment was repeated three times. Cell survival was expressed as absorbance relative to that of untreated controls. Results are presented

as mean  $\pm$  standard deviation (SD).

### 1.5 Confocal microscopic imaging of cells using RO-OA FONs

A549 cells were cultured in Dulbecco's modified eagle medium (DMEM) supplemented with 10% heat-inactivated FBS, 2 mM glutamine, 100 U mL<sup>-1</sup> penicillin, and 100  $\mu$ g mL<sup>-1</sup> of streptomycin. Cell culture was maintained at 37 °C in a humidified condition of 95% air and 5% CO<sub>2</sub> in culture medium. Culture medium was changed every three days for maintaining the exponential growth of the cells. On the day prior to treatment, cells were seeded in a glass bottom dish with a density of  $1 \times 10^5$  cells per dish. On the day of treatment, the cells were incubated with RO-OA FONs at a final concentration of 20  $\mu$ g mL<sup>-1</sup> for 3 h at 37 °C. Afterward, the cells were washed three times with PBS to remove the RO-OA FONs and then fixed with 4% paraformaldehyde for 10 min at room temperature. Cell images were taken with a confocal laser scanning microscope (CLSM) Zesis 710 3-channel (Zesis, Germany) with the excitation wavelength of 543 nm.

## 2. Results

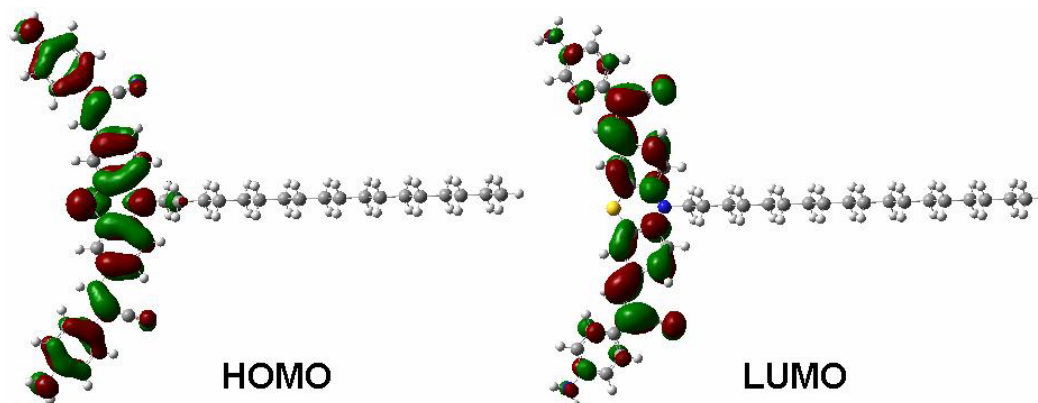


Fig. S1 Calculated spatial electron distributions of HOMO and LUMO of PhNH<sub>2</sub>.

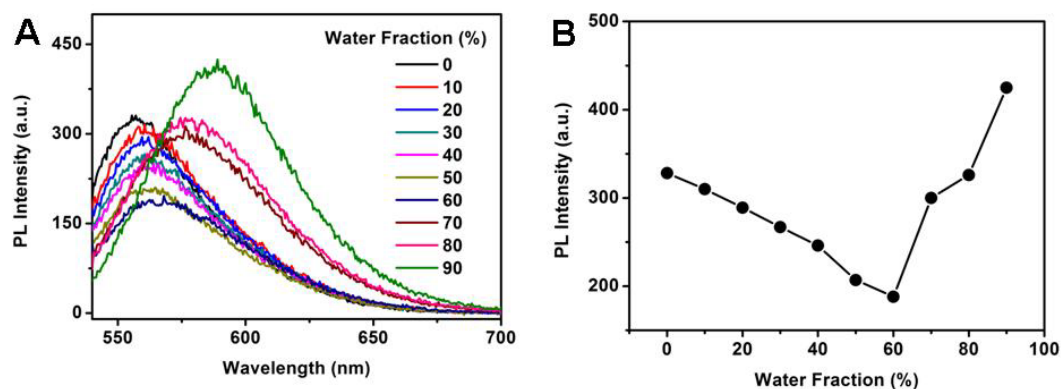
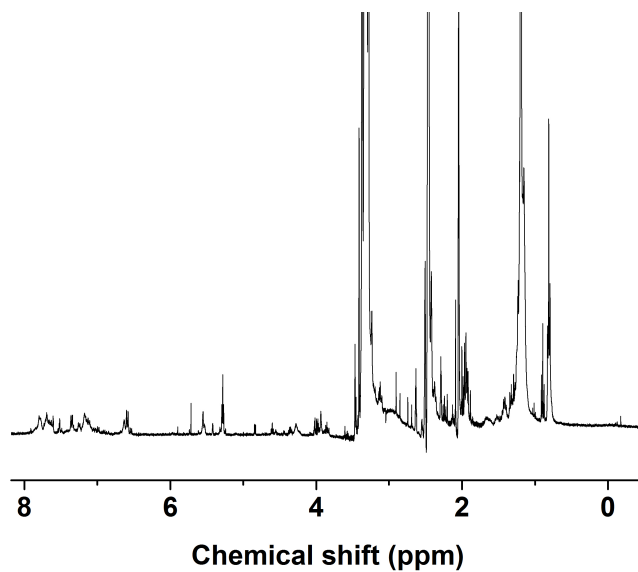
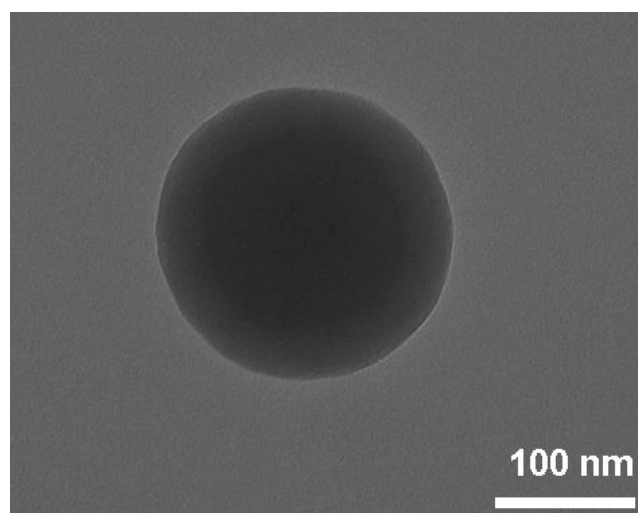


Fig. S2 The AIE characteristic of PhNH<sub>2</sub> ( $\lambda_{\text{Ex}} = 530$  nm): (A) PL spectra of PhNH<sub>2</sub> in

THF–water mixtures with different water fractions; (B) the changes in PL peak intensity of the compound in different water fraction mixtures.



**Fig. S3** <sup>1</sup>H NMR spectrum of **RO-OA** dissolved in d<sub>6</sub>-DMSO.



**Fig. S4** TEM image of **RO-OA** FONs dispersed in water, scale bar = 100 nm.

#### Reference

1. J. Y. Bae and L. Kevan, *Microporous Mesoporous Mater.*, 2001, **50**, 1-12.