

Morphology Transitions of Supramolecular Hyperbranched Polymers Induced by Double Supramolecular Driving Forces

Yang Bai, Xiao-dong Fan*, Wei Tian*, Ting-ting Liu, Hao Yao,

Zhen Yang, Hai-tao Zhang, Wan-bin Zhang

The Key Laboratory of Space Applied Physics and Chemistry, Ministry of Education
and Shaanxi Key Laboratory of Macromolecular Science and Technology, School of
Science, Northwestern Polytechnical University, Xi'an, 710072, P. R. China

Content

- 1. Synthesis of alkynyl-terminated Ada, Ada-alk₂**
- 2. Synthesis of Ada-CD₂**
- 3. Characterization of Ada-CD₂**
- 4. Results and Discussion**

1 Synthesis of alkynyl-terminated Ada, Ada-alk₂

Ada-NH₂ (756 mg, 5 mmol), K₂CO₃ (5520 mg, 40mmol) and Propargylbromide (4462 mg, 30 mol) were dissolved in dry DMF (30mL). After being bubbled with nitrogen gas for 30 min, the reaction was conducted at 60 °C for 48 hours. After removing the insoluble salts by suction filtration, the filtrate was concentrated and then further purified by silica gel column chromatography using petroleum ether/ethyl acetate

(1:1 v/v) as the eluent. After removing the solvent by a rotary evaporator, the obtained residue was distilled under reduced pressure. Yield: 78%.. Yield: 75.3%. FT-IR (KBr): 3256cm⁻¹ (v, ≡C-H); 2103cm⁻¹ (v, C≡C). ¹H NMR (CDCl₃, TMS): δ= 1.52-2.15 (15H, Ada); 2.41(1H,-C≡CH); 3.32 (2H, -CH₂C≡CH).

2 Synthesis of Ada-CD₂

The synthesis of Ada-CD₂ was accomplished by click reaction between Ada-alk₂ and β-CD-N₃. At first, a mixture of β-CD-N₃ (877 mg, 0.76 mmol), Ada-alk₂ (78 mg, 0.34 mmol) and DMF (6 mL) was bubbled with nitrogen gas for 15 min. Cu(PPh₃)₃Br (130.2 mg, 0.14 mmol) was then added into the reaction system. After being bubbled with nitrogen gas for 30 min, the reaction was conducted at 60 °C for 18 hours, and then, Alk-wang resin (500 mg) was added and allowed this system to react for another 8 hours under nitrogen atmosphere. The mixture was exposed to air and diluted with DMF, followed by passing it through a neutral Al₂O₃ column, and then, subjected to rotary evaporation. The residue was precipitated into cold acetone and washed with acetone three times. Finally, the product was dried under vacuum at 30 °C for 1 day. Yield: 80%. FT-IR (KBr): 3370cm⁻¹ (v,O-H); 2870cm⁻¹ (v,C-H); 1033cm⁻¹ (v, C-O-C in β-CD). ¹H NMR (DMSO-d₆, TMS): δ=1.45-2.1 (15H, Ada); 3.95-3.2 (2,3,4,5,6-H in β-CD); 4.6-4.3 (12H, 6-OH); 4.98-

4.7 (14H, 1-H); 5.9–5.6 (28H, 2,3-OH); $\delta=8.02$ (2H, methine proton in 1,2,3-triazole).

3 Characterization of Ada-CD₂

Fourier Transform Infrared (FT-IR) spectra were recorded on a NICOLET iS10 IR spectrometer. The ¹H NMR spectra were obtained from a Bruker Avance 300 spectrometer (Bruker BioSpin, Switzerland) operating at 300 MHz (1H) in CDCl₃ or DMSO-d₆. The 2D ¹H NMR ROESY spectra were recorded on a Bruker-Avance III NMR spectrometer (400 MHz) with D₂O/DMSO-d₆ as solvent. The molecular structure parameters of the polymer synthesized were determined on a Q-TOF-MS, measurements were performed on a Waters-ACQUITY™ UPLC & Q-TOF-MS Premier with deionized water as the solvent.

4 Results and Discussion

The amphiphilic AB₂ monomer Ada-CD₂ was first synthesized to study its supramolecular behaviors in varied solvents according to the routes depicted in Scheme 1. To enhance the functionality of Ada-NH₂, two alkyne units were introduced into its ends by the alkylation reaction in the presence of K₂CO₃, which is denoted as Ada-alk₂. In the FT-IR spectrum of Ada-alk₂ (Fig. S1(a)), the appearance of the characteristic absorption peak of alkynyl at 3256 cm⁻¹, 2103 cm⁻¹ demonstrated that the amino group in adamantane have reacted with propargylbromide. That was also evidenced by the appearance of proton peaks a and d in the ¹H

NMR spectrum of Ada-alk₂ (Fig. S2-(II)). The peak integral ratio of proton **d** to **a** was calculated to be 1.94, and the signal of proton **a** in amino group disappeared, these results mentioned demonstrated that the alkylation reaction could be completely proceeded.

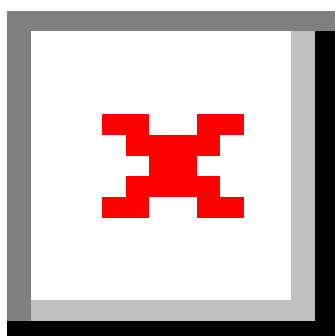


Figure. S1 FT-IR spectra of alkynyl-terminated Ada, Ada-alk₂ (a) and AB₂ type monomer, Ada-CD₂ (b).

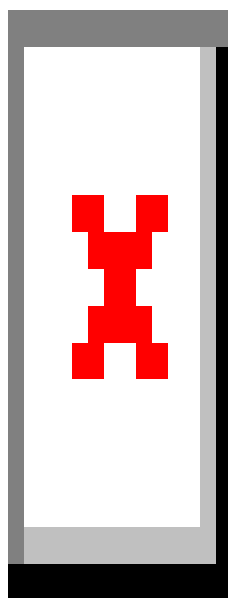


Figure. S2 ^1H NMR spectra of the Ada-NH₂(I),Ada-alk₂(II) and Ada-CD₂ (III).

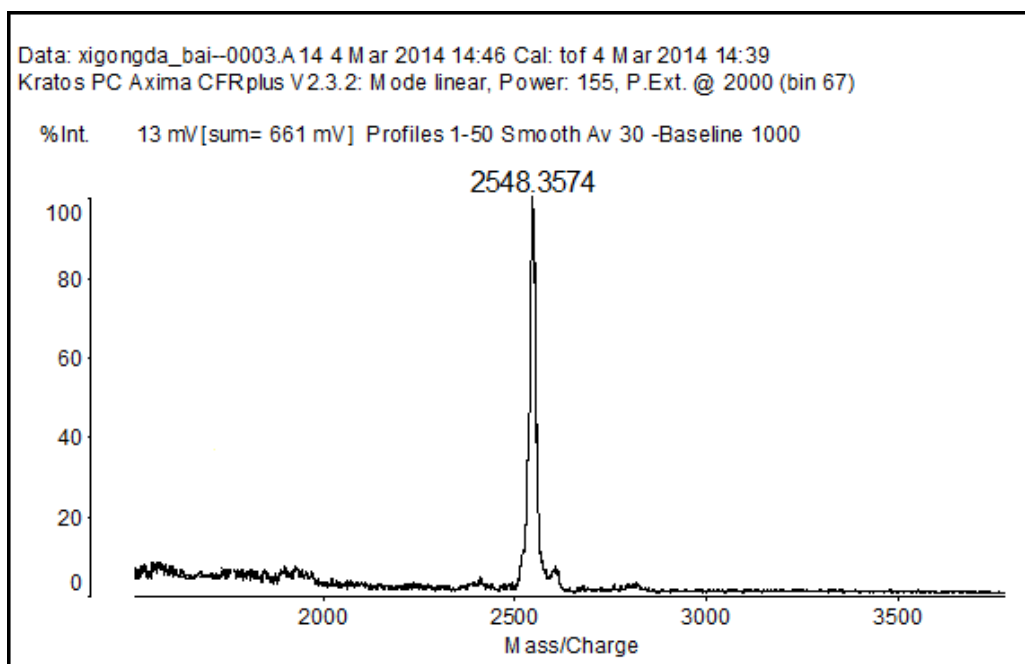


Figure. S3 MALDI-TOF-MS spectrum of Ada-CD₂

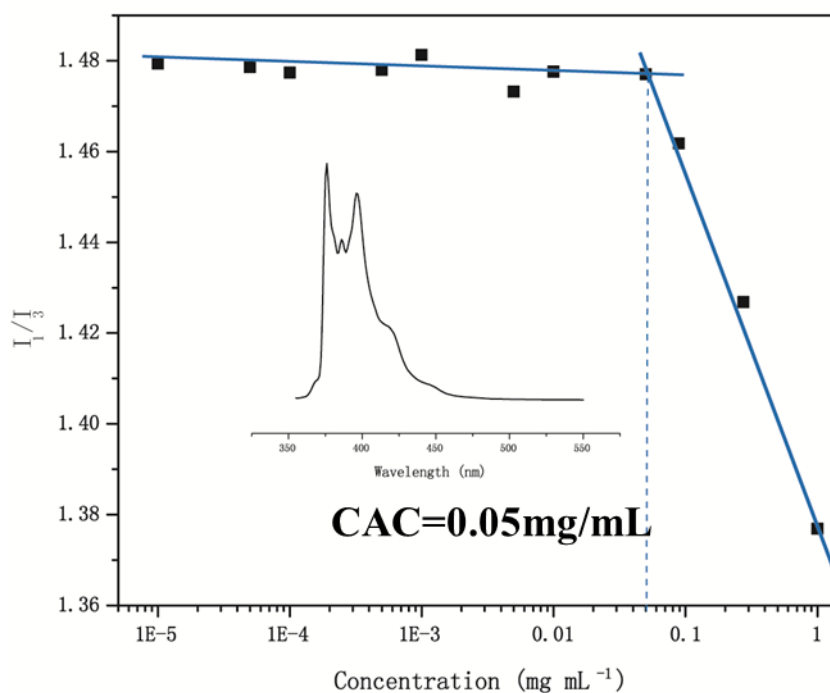


Figure. S4 Relationship between the fluorescence intensity ratio and the Ada-CD₂ concentration in DMF/H₂O (1:2) solutions after the addition of Ada-Na at 25 °C (inset: the pyrene fluorescence emission spectrum).

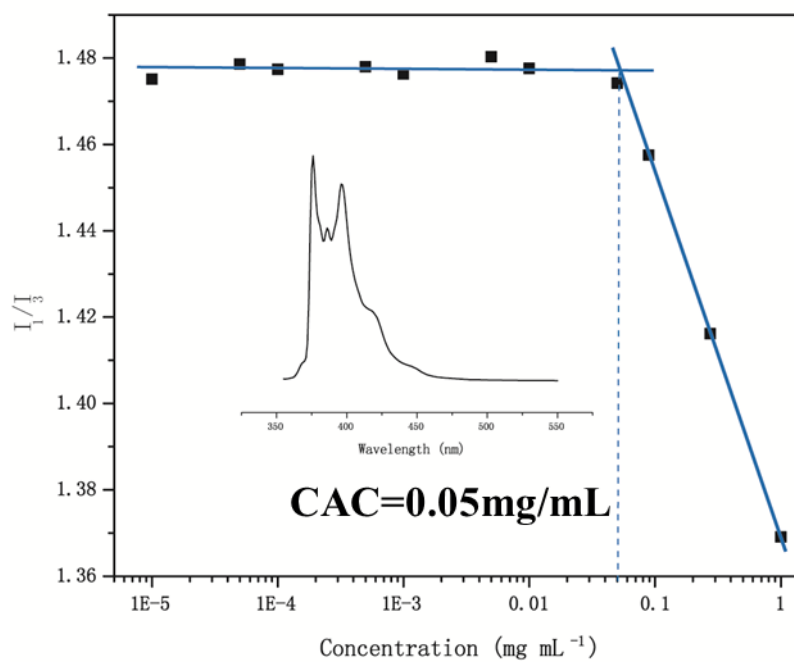


Figure. S5 Relationship between the fluorescence intensity ratio and the Ada-CD₂ concentration in DMF/H₂O (2:1) solutions after the addition of Ada-Na at 25 °C (inset: the pyrene fluorescence emission spectrum).