Supporting Information

Double-Long-Chain Imidazolium Ionic Liquids and Ionic Liquid Crystals

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Experimental Section

General Methods. N-(Trimethylsilyl)imidazole, NaBF₄, NaClO₄, NH₄PF₆, lithium trifluoromethanesulfonate and lithium bis(trifluoromethanesulfonyl) amide were purchased from ACROS and used without further purification. 1-Chlorodecane, 1-chlorotetradecane, 1-chlorotetradecane, and 1-chlorooctadecane were purchased from Fluka and used as received. 1,3-didecylimidazolium chloride, 1,3-didodecylimidazolium chloride, 1,3-didodecylimidazolium chloride, 1,3-didotecylimidazolium chloride were synthesized according to literature procedure.¹ 1,3-didodecylimidazolium tetrafluoroborate, [C₁₂C₁₂IM][BF₄], was synthesized according to our previously reported method.²

¹H NMR spectra were recorded on JEOL 400 MHz and 270 MHz instruments, operating at respective frequencies of 399.782 MHz and 269.714 MHz with a probe temperature of 23 °C. ¹³C NMR spectra were recorded on JEOL 400 MHz and 270 MHz instruments operating at respective frequencies of 100.525 MHz and 67.82 MHz with a probe temperature of 23 °C. Chemical shifts were reported relative to the peak for SiMe₄ using ¹H (residual) chemical shifts of the solvent as a secondary standard and are reported in ppm. Elemental analysis results for C, H, and N were obtained from the Analytical Laboratories at the Friedrich-Alexander-University Erlangen-Nürnberg (Erlangen, Germany).

Preparation of $[C_{12}C_{12}IM][PF_6]$. To a solution (10 mL) of 1,3-didodecylimidazolium chloride (0.974 g, 2.21 mmol) in acetonitrile was added NH₄PF₆ (1.1 g, 6.64 mmol). The mixture was stirred for 2 h, followed by removal of the solvent. Dichloromethane (10 mL) was then added to the solids and the resulting mixture was filtered. The filtrate was evaporated to dryness to give 1,3-didodecylimidazolium hexafluorophosphate. Yield: 1.168 g (96 %). ¹H NMR (270 MHz, DMSO-*d*₆): δ 9.15 (s, 1H, imidazole C(2)*H*), 7.75 (d, *J* = 1.2 Hz, 2H, imidazole C*H*=C*H*), 4.11 (t, *J* = 7.0 Hz, 4H, NC*H*₂), 1.74 (quintet, *J* = 6.8 Hz, NCH₂C*H*₂), 1.19 (m, 36H, (C*H*₂)_n), 0.81 (t, *J* = 6.5 Hz, C*H*₃). ¹³C{¹H} NMR (67.5 MHz, DMSO-*d*₆): δ 136.46, 123.01, 49.37, 31.84, 29.74, 29.57, 29.46, 29.40, 29.26, 28.85, 25.97, 22.63, 14.44.

Anal. Calcd for C₂₇H₅₃N₂PF₆: C, 58.89; H, 9.70; N, 5.09. Found: C, 59.18; H, 10.08; N, 5.08.

Preparation of $[C_{12}C_{12}IM][CF_3SO_3]$. Lithium trifluoromethanesulfonate (467 mg, 2.99 mmol) was added to a solution of 1,3-didodecylimidazolium chloride (1.2 g, 2.72 mmol) in 10 mL dichloromethane and stirred for 2 days. The suspension was filtered to remove precipitated lithium chloride salt and the organic phase was washed with small amounts of water (ca. 30 mL) until no precipitation of AgCl was observed in the aqueous phase on addition of a concentrated AgNO₃ solution. The organic phase was then washed with water to ensure complete removal of the lithium chloride salt. The solvent

was removed *in vacuo* and the resulting ionic liquid was dried at 70 °C in vacuo for 24 h. Yield: 1.22 g (81%). ¹H NMR (270 MHz, DMSO- d_6): δ 9.18 (s, 1H, imidazole C(2)*H*), 7.79 (s, 2H, imidazole C*H*=C*H*), 4.14 (t, *J* = 7.0 Hz, 4H, NC*H*₂), 1.74 (quintet, *J* = 6.2 Hz, NCH₂C*H*₂), 1.22 (m, 36H, (C*H*₂)_n), 0.84 (t, *J* = 6.2 Hz, C*H*₃). ¹³C{¹H} NMR (67.5 MHz, DMSO- d_6): δ 135.90, 122.48, 48.83, 31.28, 29.18, 29.00, 28.90, 28.84, 28.70, 28.29, 25.41, 22.08, 13.93.

Anal. Calcd for C₂₈H₅₃N₂F₃O₃S: C, 60.62; H, 9.63; N, 5.05. Found: C, 60.93; H, 9.77; N, 5.06.

Preparation of [C₁₂C₁₂**IM]**[(CF₃SO₂)₂N]. This compound was synthesized analogously to [C₁₂C₁₂IM][CF₃SO₃] described above: reaction of 1,3-didodecylimidazolium chloride (533 mg, 1.21 mmol) and lithium bis(trifluoromethanesulfonyl) amide (382 mg, 1.33 mmol) generates [C₁₂C₁₂IM] [(CF₃SO₂)₂N]. Yield 647 mg (78 %). ¹H NMR (270 MHz, DMSO-*d*₆): δ 9.18 (s, 1H, imidazole C(2)*H*), 7.78 (s, 2H, imidazole *CH*=*CH*), 4.14 (t, *J* = 6.8 Hz, 4H, NC*H*₂), 1.77 (quintet, *J* = 6.8 Hz, NCH₂C*H*₂), 1.22 (m, 36H, (*CH*₂)_n), 0.84 (t, *J* = 6.2 Hz, *CH*₃). ¹³C{¹H} NMR (67.5 MHz, DMSO-*d*₆): δ 136.44, 122.99, 49.36, 31.82, 29.72, 29.56, 29.43, 29.37, 29.24, 28.82, 25.94, 22.61, 14.41.

Anal. Calcd for C₂₉H₅₃N₃F₆O₄S₂: C, 50.78; H, 7.79; N, 6.13. Found: C, 51.01; H, 8.07; N, 6.20.

Preparation of $[C_{10}C_{10}IM][BF_4]$. To a solution of 1,3-decylimidazolium chloride (1.2 g, 3.12 mmol) in acetone (10 mL) was added NaBF₄ (0.41 g, 3.74 mmol). The mixture was stirred for 2 days, followed by complete removal of the solvent *in vacuo*. Dicloromethane (10 mL) was added to the solids followed by filtration of the suspension. The filtrate was evaporated to dryness to give 1,3-didecylimidazolium tetrafluoroborate. Yield: 1.27 g (93 %). ¹H NMR (400 MHz, DMSO-*d*₆): δ 9.17 (s, 1H, imidazole C(2)*H*), 7.79 (s, 2H, imidazole C*H*=C*H*), 4.16 (t, *J* = 7.0 Hz, 4H, NC*H*₂), 1.78 (quintet, *J* = 7.1 Hz, NCH₂C*H*₂), 1.23 (m, 28H, (C*H*₂)_n), 0.85 (t, *J* = 6.8 Hz, C*H*₃). ¹³C{¹H} NMR (100 MHz, DMSO-*d*₆): δ 136.48, 123.03, 49.43, 31.87, 29.82, 29.46, 29.44, 29.26, 28.91, 26.02, 22.67, 14.47.

Anal. Calcd for C₂₃H₄₅N₂BF₄: C, 63.30; H, 10.39; N, 6.42. Found: C, 63.54; H, 10.60; N, 6.44.

Preparation of $[C_{10}C_{10}IM][ClO_4]$. NaClO₄ (0.52 g, 4.24 mmol) was added to a solution of 1,3-didecylimidazolium chloride (1.09 g, 2.83 mmol) in 10 mL dichloromethane and stirred for 2 days. The suspension was filtered to remove the precipitated sodium chloride salt and the organic phase was washed with small volumes of water (ca. 30 mL) until no precipitation of AgCl occurred in the aqueous phase on addition of a concentrated AgNO₃ solution. The organic phase was then washed with water to ensure complete removal of the chloride salt. The solvent was removed *in vacuo* and the resulting ionic liquid was dried at 100 °C *in vacuo* for 24 h. Yield: 1.12 g (88 %). ¹H NMR (400 MHz, DMSO-*d*₆): δ 9.20 (s, 1H, imidazole C(2)*H*), 7.79 (d, *J* = 1.2 Hz, 2H, imidazole C*H*=C*H*), 4.16 (t, *J* = 7.2 Hz, 4H, NC*H*₂), 1.78 (quintet, *J* = 7.2 Hz, NCH₂C*H*₂), 1.23 (m, 28H, (C*H*₂)_n), 0.85 (t, *J* = 6.8 Hz, C*H*₃). ¹³C{¹H} NMR (100 MHz, DMSO-*d*₆): δ 136.51, 123.05, 49.44, 31.86, 29.80, 29.45, 29.42, 29.25, 28.89, 26.02, 22.66, 14.49.

Anal. Calcd for C₂₃H₄₅N₂ClO₄: C, 61.52; H, 10.10; N, 6.24. Found: C, 61.78; H, 10.08; N, 6.54.

Preparation of $[C_{14}C_{14}IM][BF_4]$. Procedure analogous to the preparation of $[C_{10}C_{10}IM][BF_4]$. ¹H NMR (400 MHz, CDCl₃): δ 8.86 (s, 1H, imidazole C(2)*H*), 7.35 (d, *J* = 1.6 Hz, 2H, imidazole C*H*=C*H*), 4.18 (t, *J* = 7.4 Hz, 4H, NC*H*₂), 1.86 (quintet, *J* = 6.8 Hz, NCH₂C*H*₂), 1.23 (m, 44H, (C*H*₂)_n), 0.86 (t, *J* = 6.8 Hz, C*H*₃). ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 135.87, 122.34, 50.22, 31.99, 30.19, 29.77, 29.74, 29.73, 29.70, 29.61, 29.48, 29.43, 29.04, 26.29, 22.75, 14.17.

Anal. Calcd for C₃₁H₆₁N₂BF₄: C, 67.87; H, 11.21; N, 5.11. Found: C, 68.17; H, 11.24; N, 5.16.

Preparation of $[C_{14}C_{14}IM][ClO_4]$. Procedure analogous to the preparation of $[C_{10}C_{10}IM][ClO_4]$. ¹H NMR (400 MHz, CDCl₃): δ 8.99 (s, 1H, imidazole C(2)*H*), 7.32

(d, J = 3.6 Hz, 2H, imidazole C*H*=C*H*), 4.21 (t, J = 7.4 Hz, 4H, NC*H*₂), 1.88 (quintet, J = 6.8 Hz, NCH₂C*H*₂), 1.24 (m, 44H, (C*H*₂)_n), 0.87 (t, J = 6.8 Hz, C*H*₃). ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 136.22, 122.15, 50.40, 32.00, 30.18, 29.77, 29.73, 29.70, 29.60, 29.47, 29.44, 29.04, 26.32, 22.76, 14.19.

Anal. Calcd for C₃₁H₆₁N₂ClO₄: C, 66.34; H, 10.95; N, 4.99. Found: C, 66.65; H, 11.13; N, 4.98.

Preparation of $[C_{16}C_{16}IM][BF_4]$. Procedure analogous to the preparation of $[C_{10}C_{10}IM][BF_4]$. ¹H NMR (400 MHz, CDCl₃): δ 8.93 (s, 1H, imidazole C(2)*H*), 7.31 (d, J = 3.6 Hz, 2H, imidazole C*H*=C*H*), 4.19 (t, J = 7.4 Hz, 4H, NC*H*₂), 1.87 (quintet, J = 6.6 Hz, NCH₂C*H*₂), 1.24 (m, 52H, (C*H*₂)_n), 0.87 (t, J = 6.8 Hz, C*H*₃). ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 136.09, 122.20, 50.26, 32.00, 30.18, 29.78, 29.75, 29.74, 29.71, 29.61, 29.48, 29.44, 29.04, 26.30, 22.76, 14.18.

Anal. Calcd for C₃₅H₆₉N₂BF₄: C, 69.51; H, 11.50; N, 4.63. Found: C, 69.30; H, 11.50; N, 4.71.

Preparation of $[C_{16}C_{16}IM][ClO_4]$. Procedure analogous to the preparation of $[C_{10}C_{10}IM][ClO_4]$. ¹H NMR (400 MHz, CDCl₃): δ 9.09 (s, 1H, imidazole C(2)*H*), 7.28 (d, *J* = 1.6 Hz, 2H, imidazole C*H*=C*H*), 4.22 (t, *J* = 7.6 Hz, 4H, NC*H*₂), 1.89 (quintet, *J* = 7.0 Hz, NCH₂C*H*₂), 1.24 (m, 52H, (C*H*₂)_n), 0.87 (t, *J* = 6.8 Hz, C*H*₃). ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 136.43, 122.04, 50.43, 32.01, 30.19, 29.80, 29.79, 29.78, 29.75, 29.71, 29.60, 29.48, 29.45, 29.04, 26.33, 22.77, 14.20.

Anal. Calcd for C₃₅H₆₉N₂ClO₄: C, 68.09; H, 11.26; N, 4.54. Found: C, 68.30; H, 11.43; N, 4.82.

Preparation of $[C_{18}C_{18}IM][BF_4]$. Procedure analogous to the preparation of $[C_{10}C_{10}IM][BF_4]$. ¹H NMR (400 MHz, CDCl₃): δ 9.28 (s, 1H, imidazole C(2)*H*), 7.27 (d, J = 1.6 Hz, 2H, imidazole C*H*=C*H*), 4.22 (t, J = 7.4 Hz, 4H, NC*H*₂), 1.88 (quintet, J = 7.2 Hz, NCH₂C*H*₂), 1.24 (m, 60H, (C*H*₂)_n), 0.87 (t, J = 6.8 Hz, C*H*₃). ¹³C{¹H} NMR (100

MHz, CDCl₃): δ 136.72, 121.88, 50.31, 32.01, 30.22, 29.79(4), 29.78(5), 39.75(4), 29.74(6), 29.71, 29.60, 29.48, 29.45, 29.05, 26.31, 22.77, 14.19.

Anal. Calcd for C₃₉H₇₇N₂BF₄: C, 70.88; H, 11.74; N, 4.24. Found: C, 71.14; H, 11.83; N, 4.50.

Preparation of $[C_{18}C_{18}IM][ClO_4]$. Procedure analogous to the preparation of $[C_{10}C_{10}IM][ClO_4]$. ¹H NMR (400 MHz, CDCl₃): δ 9.13 (s, 1H, imidazole C(2)*H*), 7.28 (d, *J* = 7.6 Hz, 2H, imidazole C*H*=C*H*), 4.22 (t, *J* = 7.4 Hz, 4H, NC*H*₂), 1.89 (quintet, *J* = 6.8 Hz, NCH₂C*H*₂), 1.24 (m, 60H, (C*H*₂)_n), 0.87 (t, *J* = 6.8 Hz, C*H*₃). ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 136.49, 122.00, 50.43, 32.00, 30.17, 29.78, 29.74(4), 29.73(5), 29.70, 29.60, 29.47, 29.43, 29.03, 26.33, 22.76, 14.17.

Anal. Calcd for C₃₉H₇₇N₂ClO₄: C, 69.55; H, 11.52; N, 4.16. Found: C, 69.82; H, 11.82; N, 4.06.

Characterization of Physical-Chemical Properties. Differential scanning calorimetry (DSC) measurements were performed on a Netzsch DSC 204 instrument with a heating rate of 5 K min⁻¹ in the temperature range of 273 K and 423 K. Polarized optical microscopy (POM) was carried out using a Nikon Eclipse 50i microscope equipped with a Nikon DsFi 1 digital camera and a Linkam PE95-LinkPad heating table.



Figure S1. DSC for a) $[C_{16}C_{16}IM][BF_4]$ and b) $[C_{16}C_{16}IM][ClO_4]$. 2nd heating cycle (bottom), 2nd cooling cycle (top).



Figure S2. POM images for a) $[C_{10}C_{10}IM][BF_4]$ at 7.3 °C (200 x magnification) and b) $[C_{10}C_{10}IM][CIO_4]$ at 6.3 °C (200 x magnification).



Figure S3. POM image of $[C_{10}C_{10}IM][BF_4]$ at 17.7 °C (200 x magnification). Liquid crystalline phase re-establishes after several minutes when cooling to 17.7 °C.



Figure S4. POM image for $[C_{14}C_{14}IM][BF_4]$ at 70 °C (200 x magnification).



a)

b)

Figure S5. POM image for a) $[C_{18}C_{18}IM][BF_4]$ at 120 °C (200 x magnification) and b) $[C_{18}C_{18}IM][CIO_4]$ at 120 °C.



Figure S6. POM image for the mixture of $[C_{10}C_{10}IM][BF_4]$ and $[C_{12}C_{12}IM][BF_4]$ with the molar ratio 1:3 at 40 °C.

References

- 1. K. J. Harlow, A. F. Hill, T. Welton, *Synthesis*, **1996**, *06*, 697.
- 2. X. Wang, F. W. Heinemann, M. Yang, B. U. Melcher, M. Fekete, A.-V. Mudring, P. Wasserscheid, K. Meyer, *Chem. Commun.*, **2009**, *47*, 7405.