

Self-assembly of New Fluorescent Pd(II) and Pt(II) 2,7-Diazapyrenium-based Metallocycles. Study of its inclusion complexes and [3]catenanes.

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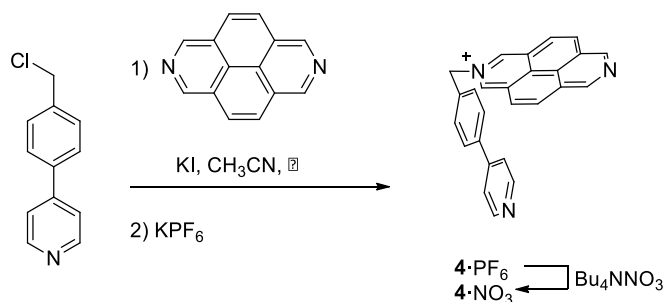
General Methods

2,7-diazapyrene,¹ ligand **1**·2PF₆,² Pd and Pt complexes,³ 4-(4'-cloromethylphenyl)pyridine,⁴ metallocycle **3**·2PF₆,² cyclophanes BPPC34C10 (**8**)⁵ and DN38C10 (**9**)⁶ and guests **11**,⁷ **13**,⁸ **15**,⁵ **17**,⁹ **19**¹⁰ and **21**¹¹ were prepared according to literature procedures. All other reagents used were commercial grade chemicals from freshly opened containers. Milli-Q water was purified with a Millipore Gradient A10 apparatus. Merck 60 F₂₅₄ foils were used for thin layer chromatography, and Merck 60 (230-400 mesh) silica gel was used for flash chromatography. Proton and carbon nuclear magnetic resonance spectra were recorded on a Bruker Avance 300 or a Bruker Avance 500 spectrometer equipped with a dual cryoprobe for ¹H and ¹³C, using the deuterated solvent as lock and the residual protiated solvent as internal standard. DOSY experiments were referenced using the value 1.92×10⁻⁹ m² s⁻¹ for the DHO signal in D₂O at 298 K¹² and the value 1.97×10⁻⁹ m² s⁻¹ for the CHD₂NO₂ signal in CD₃NO₂ at 298 K.¹³ Mass spectrometry experiments were carried out in a LCQ-q-TOF Applied Biosystems QSTAR Elite spectrometer for low- and high-resolution ESI. UV-Vis spectra were recorded on a Perkin Elmer Lambda 900 spectrometer. Fluorescence spectra were recorded at room temperature on a Perkin Elmer LS 50B fluorescence spectrometer using a 1% T filter and a slit width of 8 nm. Melting points were measured using Stuart Scientific SMP3 apparatus. Microanalyses for C, H and N were performed by the elemental analyses general service of the University of A Coruña.

Crystal structure analysis

The structure was solved by direct methods and refined with the full-matrix least-squares procedure (SHELX-97)¹⁴ against *F*². The X-ray diffraction data were collected on a Bruker X8 ApexII. Non-Solvent hydrogen atoms were placed in idealized positions with *U*_{eg}(H) = 1.2*U*_{eg}(C) and were allowed to ride on their parent atoms. Solvent hydrogen atoms were placed in idealized positions with *U*_{eg}(H) = 1.5*U*_{eg}(C) and were allowed to ride on their parent atoms.

Ligand 4·PF₆



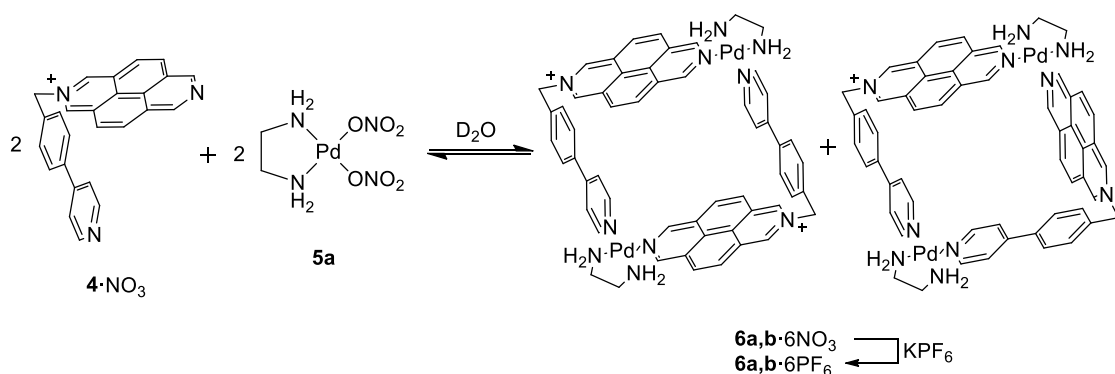
To a solution of 2,7-diazapyrene (1.44 g, 7.00 mmol) and a catalytic amount of KI in refluxing CH₃CN (90 mL) was slowly added a solution of 4-(4'-cloromethylphenyl)pyridine

(0.95 g, 4.66 mmol) cooled to 0 °C in CH₃CN (70 mL). The reaction was refluxed for 72 h; after cooling, the solvent was evaporated in vacuo to give a crude product which was purified by column chromatography (SiO₂, acetone/NH₄Cl 1.5M/MeOH 5:4:1). The product-containing fractions were combined and the solvents were removed in vacuo. The residue was dissolved in H₂O/CH₃OH (50/30, 800 mL) and an excess of KPF₆ was added until no further precipitation was observed. The solid was filtered and washed with water to give **4**·PF₆ (0.76 g, 31%) as a yellow solid. Mp: 189-191 °C (dec.). ¹H NMR (500 MHz, CD₃NO₂) δ: 6.45 (2H, s); 7.84 (2H, d, *J* = 8.5 Hz); 8.00 (2H, d, *J* = 6.4 Hz); 8.03 (2H, d, *J* = 6.4 Hz); 8.61 (2H, d, *J* = 9.1 Hz); 8.77 (4H, m); 9.83 (2H, s); 9.88 (2H, s); ¹³C NMR (125 MHz, CD₃NO₂) δ: 67.1 (CH₂); 124.6 (CH); 125.5 (C); 127.2 (CH); 127.5 (C); 130.0 (CH); 130.4 (C); 130.9 (C); 131.5 (CH); 132.8 (CH); 137.2 (C); 139.2 (C); 139.5 (CH); 147.4 (CH); 149.8 (CH); 153.7 (C). MS-ESI (*m/z*): 372.2 [M – PF₆[–]]⁺. Anal. Calcd. C, 53.90; H, 3.13; N, 9.67. Found. C, 53.67; H, 3.22; N, 9.60.

Ligand **4**·NO₃

To a solution of **4**·PF₆ (197.0 mg, 0.38 mmol) in CH₃CN (90 mL) Bu₄NNO₃ was added until no further precipitation is observed. The mixture was stirred at rt for 6h and the precipitate was filtered and washed with CH₃CN to yield **4**·NO₃ (130.1 mg, 79 %) as a yellow solid. Mp: 161-163 °C. ¹H NMR (500 MHz, D₂O) δ: 6.20 (2H, s); 7.69 (4H, m); 7.80 (2H, d, *J* = 8.3 Hz); 8.31 (2H, d, *J* = 9.1 Hz); 8.44 (4H, m); 9.51 (2H, s); 9.75 (2H, s); ¹³C NMR (125 MHz, D₂O) δ: 65.9 (CH₂); 123.2 (CH); 124.6 (C); 126.5 (CH); 126.7 (C); 128.6 (C); 129.0 (CH); 129.7 (C); 130.6 (CH); 131.7 (CH); 135.9 (C); 138.5 (C); 138.8 (CH); 147.3 (CH); 147.6 (CH); 150.9 (C). MS-ESI (*m/z*): 372.2 [M – NO₃[–]]⁺. Anal. Calcd. C, 71.88; H, 4.18; N, 12.90. Found. C, 71.99; H, 4.10; N, 12.72.

Metalloccycles **6a,b**·6NO₃



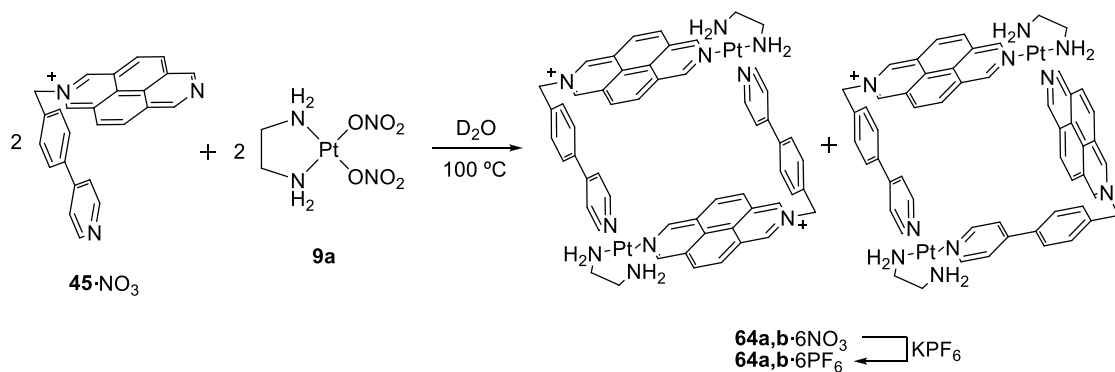
A solution of **4**·NO₃ (8.7 mg; 0.020 mmol) and (en)Pd(NO₃)₂ (**5a**) (5.8 mg, 0.020 mmol) in D₂O (4.0 mL) was stirred at 60 °C for 1 h. ¹H NMR (500 MHz, D₂O) δ: 3.00 (8H, m); 6.28 (4H, s); 7.59-7.69 (12H, m); 8.50 (4H, m); 8.55-8.61 (6H, m); 8.77 (2H, d, *J* = 6.8 Hz); 9.95 (4H, s); 9.98 (2H, s); 10.15 (2H, s); ¹³C NMR (125 MHz, D₂O) δ: 46.8 (CH₂); 46.9 (CH₂);

47.1 (CH₂); 47.2 (CH₂); 65.9 (CH₂); 124.1 (CH); 124.2 (CH); 125.0 (C); 125.1 (C); 127.4 (C); 127.4 (C); 127.9 (CH); 128.0 (CH); 128.1 (CH); 128.4 (C); 128.5 (C); 129.5 (C); 129.5 (C); 129.7 (CH); 129.7 (CH); 129.9 (CH); 129.9 (CH); 136.1 (C); 136.1 (C); 136.4 (C); 136.6 (C); 139.4 (CH); 139.5 (CH); 148.1 (CH); 148.1 (CH); 150.6 (C); 150.7 (C); 151.0 (CH); 151.0 (CH).

Metallocycles **6a,b**·6PF₆

A solution of **4**·NO₃ (21.7 mg; 0.050 mmol) and **5a** (14.5 mg, 0.050 mmol) in H₂O (10.0 mL) was stirred at 60 °C for 1h. An excess of KPF₆ was added until no further precipitation was observed. The solid was filtered and washed with water to obtain **6a,b**·6PF₆ (32.0 mg, 66%) as a pale brown solid. Mp: 195-197 °C. ¹H NMR (500 MHz, CD₃NO₂) δ: 3.22 (8H, m); 4.59 (2H, m); 4.76 (2H, m); 6.34 (4H, s); 7.66 (12H, m); 8.58 (4H, m); 8.63 (4H, m); 8.69 (2H, d, *J* = 6.4 Hz); 8.83 (2H, d, *J* = 6.5 Hz); 9.84 (4H, m); 10.00 (2H, s); 10.18 (2H, s); ¹³C NMR (125 MHz, CD₃NO₂) δ: 48.5 (CH₂); 48.6 (CH₂); 48.7 (CH₂); 48.9 (CH₂); 67.6 (CH₂); 125.4 (CH); 125.5 (CH); 126.5 (C); 129.0 (C); 129.1 (C); 129.5 (CH); 129.5 (CH); 129.6 (CH); 129.9 (C); 130.0 (C); 131.0 (CH); 131.0 (C); 131.1 (C); 131.1 (CH); 131.6 (CH); 131.6 (CH); 137.7 (C); 137.8 (C); 137.9 (C); 141.0 (CH); 141.0 (CH); 150.1 (CH); 150.2 (CH); 152.0 (C); 152.0 (C); 152.9 (CH); 153.0 (CH).

Metallocycles **7a,b**·6PF₆



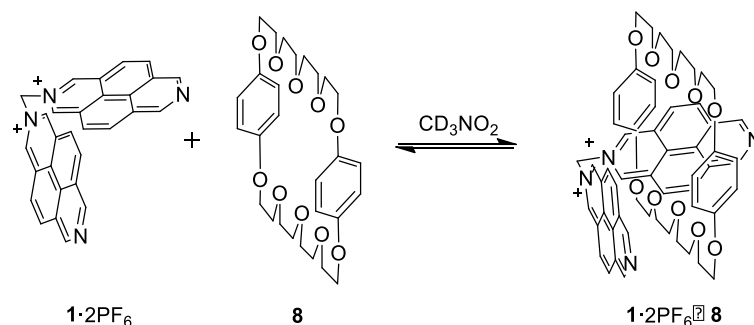
A solution of ligand **4**·NO₃ (39.0 mg, 0.090 mmol) and (en)Pt(NO₃)₂ (**5b**) (34.0 mg, 0.090 mmol) in H₂O (24.0 mL) was stirred at 100 °C for 8d. Upon cooling to room temperature, an excess of KPF₆ was added until no further precipitation was observed. The solid was filtered to yield **7a,b**·6PF₆ (88.1 mg, 92 %) as a pale brown solid. Mp: 243-245 °C (dec.). ¹H NMR (500 MHz, CD₃NO₂) δ: 3.17 (8H, m); 4.93 (2H, m); 5.13 (4H, m); 5.33 (2H, m); 6.34 (4H, s); 7.67 (12H, m); 8.62 (8H, m); 8.71 (2H, d, *J* = 5,6 Hz); 8.86 (2H, d, *J* = 5,7 Hz); 9.85 (4H, m); 10.01 (2H, s); 10.19 (2H, s). ¹³C NMR (125 MHz, CD₃NO₂) δ: 49.6 (CH₂); 49.7 (CH₂); 49.8 (CH₂); 49.9 (CH₂); 67.6 (CH₂); 67.6 (CH₂); 125.7 (CH); 125.8 (CH); 126.4 (C); 126.4 (C); 128.9 (C); 129.0 (C); 129.2 (CH); 129.3 (CH); 129.5 (CH); 130.4 (C); 130.4 (C); 131.0 (CH); 131.1 (C); 131.1 (C); 131.1 (CH); 131.5 (CH); 131.6

(CH); 137.5 (C); 137.7 (C); 137.8 (C); 137.9 (C); 141.0 (CH); 151.1 (CH); 151.1 (CH); 151.9 (C); 153.6 (CH); 153.7 (CH). HRMS-ESI (m/z): Calcd para $[M - 2PF_6^-]^{2+}$ 917-1114, found 917.1085; calcd. for $[M - 3PF_6^-]^{3+}$ 563.0860, found 563.0843; calcd. for $[M - 4PF_6^-]^{4+}$ 386.0733, found. 386.0726; calcd. for $[M - 5PF_6^-]^{5+}$ 289.8657, found 279.8645.

Metallocycles **7a,b**·6NO₃

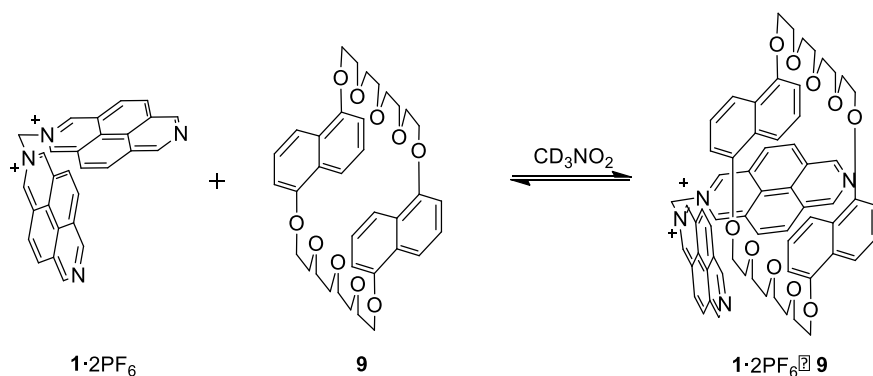
To a solution of **7a,b**·6PF₆ (55.0 mg, 0.026 mmol) in CH₃CN (4 mL) Bu₄NNO₃ was added until no further precipitation is observed. The precipitate was filtered and washed with CH₃CN to yield **7a,b**·6NO₃ (31.1 mg, 74 %) as a pale brown solid. ¹H NMR (500 MHz, D₂O) δ: 2.92 (8H, m); 6.28 (4H, s); 7.61-7.69 (12H, m); 8.50-8.58 (8H, m); 8.63 (2H, d, J = 6.9 Hz); 8.80 (2H, d, J = 6.9 Hz); 9.96 (4H, s); 9.99 (2H, s); 10.17 (2H, s); ¹³C NMR (125 MHz, D₂O) δ: 47.4 (CH₂); 47.6 (CH₂); 47.6 (CH₂); 47.8 (CH₂); 65.8 (CH₂); 65.9 (CH₂); 124.4 (CH); 124.4 (CH); 124.8 (C); 124.9 (C); 127.2 (C); 127.3 (C); 128.0 (CH); 128.0 (CH); 128.1 (CH); 128.9 (C); 128.9 (C); 129.5 (C); 129.5 (C); 129.7 (CH); 129.7 (CH); 136.1 (C); 136.1 (C); 136.2 (C); 136.3 (C); 139.4 (CH); 139.4 (CH); 149.0 (CH); 149.0 (CH); 150.3 (C); 150.5 (C); 151.6 (CH); 151.7 (CH).

Pseudorotaxane **1**·2PF₆⊂**8**



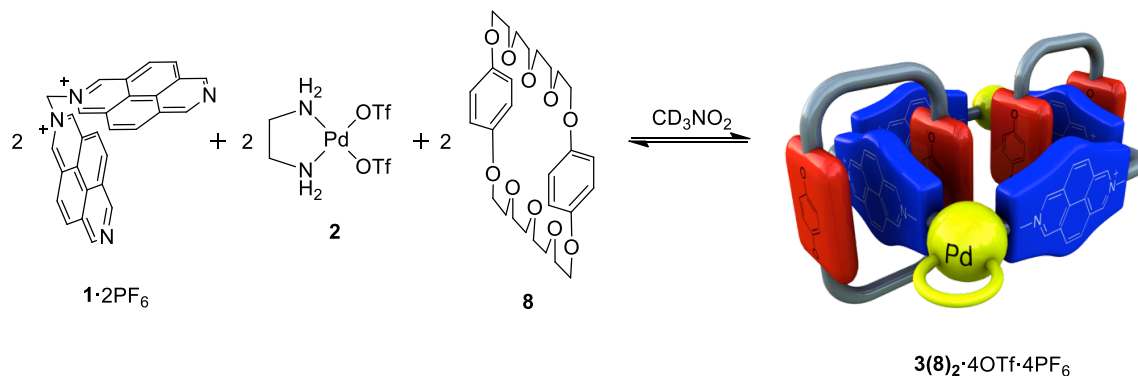
To a solution of **1**·2PF₆ (2.1 mg; 3.0×10^{-3} mmol) in CD₃NO₂ (0.6 mL) **8** (1.6 mg, 3.0×10^{-3} mmol) was added. ¹H NMR (500 MHz, CD₃NO₂) δ: 10.14 (4H, s), 9.98 (4H, s), 8.80 (4H, d, J = 9.1 Hz), 8.58 (4H, d, J = 9.1 Hz), 8.41 (2H, s), 5.90 (8H, s), 3.93–3.72 (24H, m), 3.64 (8H, m). ¹³C NMR (125 MHz, CD₃NO₂) δ: 153.0 (C), 150.8 (CH), 139.0 (CH), 134.3 (CH), 131.2 (C), 131.1 (C), 127.3 (C), 126.9 (CH), 125.1 (C), 115.1 (CH), 82.0 (CH₂), 71.8 (CH₂), 71.6 (CH₂), 71.0 (CH₂), 68.9 (CH₂).

Pseudorotaxane **1**·2PF₆⊂**9**



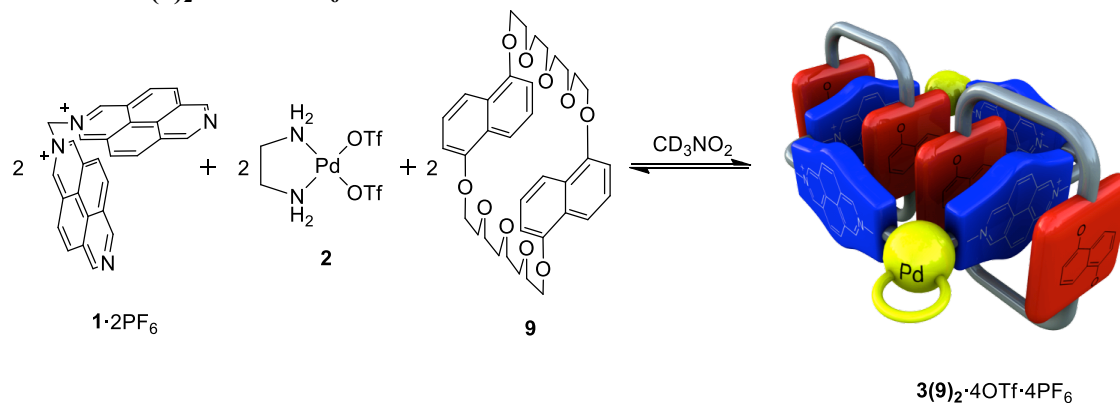
To a solution of **1**·2PF₆ (2.1 mg; 3.0×10^{-3} mmol) in CD₃NO₂ (0.6 mL) **9** (1.9 mg, 3.0×10^{-3} mmol) was added. ¹H NMR (500 MHz, CD₃NO₂) δ: 10.10 (4H, s), 9.88 (4H, s), 8.70 (4H, d, J = 9.1 Hz), 8.42 (4H, d, J = 9.1 Hz), 8.26 (2H, s), 6.43 (8H, m), 6.15 (4H, d, J = 7.1 Hz), 4.06 (8H, m), 4.02 (8H, m), 3.95 (16H, m). ¹³C NMR (125 MHz, CD₃NO₂) δ: 154.3 (C) 150.5 (CH), 138.4 (CH), 134.1 (CH), 130.7 (C), 130.3 (C), 127.3 (C), 126.7 (C), 125.9 (C), 125.9 (CH), 124.3 (C), 113.9 (CH), 106.0 (CH), 82.2 (CH₂), 72.3 (CH₂), 72.2 (CH₂), 71.2 (CH₂), 69.2 (CH₂).

Catenane **3**(**8**)₂·4OTf·4PF₆



To a solution of **1**·2PF₆ (7.1 mg; 0.010 mmol) and **2** (5.6 mg, 0.024 mmol) in CD₃NO₂ (4.0 mL) **8** (5.4 mg, 0.010 mmol) was added. ¹H NMR (500 MHz, CD₃NO₂) δ: 3.11 (4H, m); 3.32 (8H, br s); 3.83 (58H, m); 4.09 (4H, m); 4.64 (2H, m); 5.00 (4H, br s); 5.30 (4H, br s); 5.50 (8H, s); 8.26 (4H, s); 8.33 (4H, d, J = 9.2 Hz); 8.48 (4H, d, J = 9.3 Hz); 8.82 (4H, d, J = 9.4 Hz); 8.89 (4H, d, J = 9.4 Hz); 10.11 (4H, s); 10.36 (4H, s); 10.53 (4H, s); 10.61 (4H, s); ¹³C NMR (125 MHz, CD₃NO₂) δ: 49.1 (CH₂); 49.4 (CH₂); 66.9 (CH₂); 69.2 (CH₂); 70.5 (CH₂); 71.0 (CH₂); 71.2 (CH₂); 71.7 (CH₂); 72.1 (CH₂); 82.6 (CH₂); 115.0 (CH); 119.1 (C); 121.0 (C); 123.5 (C); 124.7 (C); 125.6 (C); 128.6 (C); 128.8 (CH); 129.2 (C); 129.7 (C); 129.9 (CH); 130.1 (C); 130.4 (C); 131.4 (C); 132.6 (CH); 133.0 (CH); 140.4 (C); 141.6 (C); 151.4 (CH); 152.3 (C).

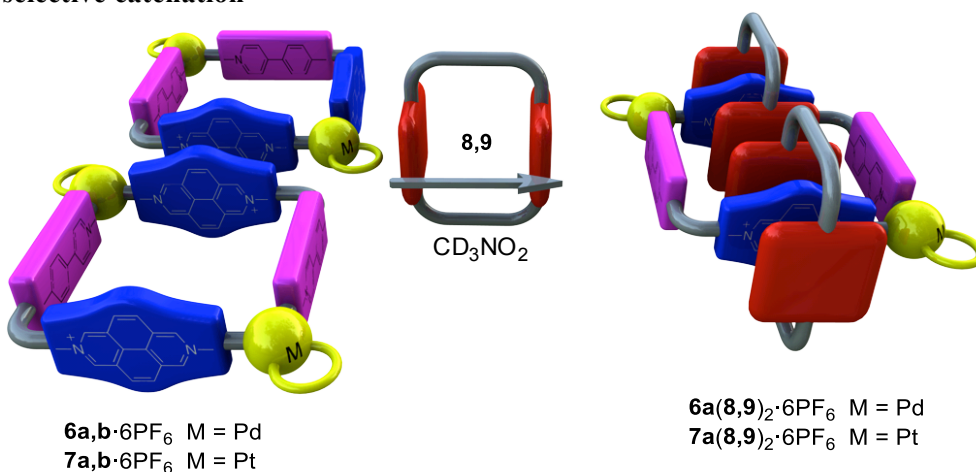
Catenane **3(9)**₂·4OTf·4PF₆



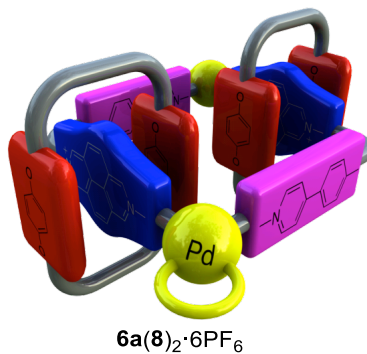
To a solution of **1**·2PF₆ (7.1 mg; .010 mmol) and (en)Pd(OTf)₂ (**2**) (5.6 mg, 0.024 mmol) in CD₃NO₂ (4.0 mL) DNP34C10 (**9**) (6.4 mg, 0.010 mmol) was added. ¹H NMR (500 MHz, CD₃NO₂): The complexity of the spectrum precluded its analysis (Figure S46).

X-ray diffraction quality single crystals of the catenane were grown by slow diffusion of diethyl ether into a solution of **1**·2PF₆, (en)Pd(OTf)₂ (**2**) and **9** in acetonitrile.

Regioselective catenation



Catenane **6a(8)**₂·6PF₆

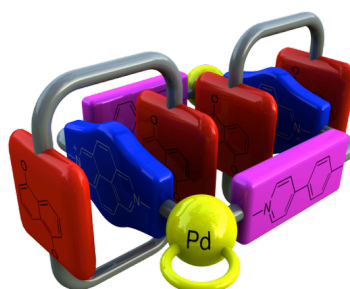


To a solution of **6a,b**·6PF₆ (5.8 mg, 3.0 × 10⁻³ mmol) in CD₃NO₂ (0.6 mL) **8** (3.2 mg, 6.0 × 10⁻³ mmol) was added. ¹H NMR (500 MHz, CD₃NO₂) δ: 3.23 (16H, m); 3.44 (12H, m); 3.59-4.19

(52H, m); 4.60 (4H, m); 4.85 (4H, m); 5.11 (4H, s); 5.39 (8H, s); 6.24 (4H, s); 7.73 (4H, d, $J = 8.4$ Hz); 7.83 (4H, d, $J = 6.7$ Hz); 7.92 (4H, d, $J = 8.4$ Hz); 8.27 (8H, m); 9.19 (4H, d, $J = 6.6$ Hz); 9.77 (4H, s); 9.87 (4H, s); ^{13}C NMR (500 MHz, CD_3NO_2) δ : 48.6 (CH_2); 49.2 (CH_2); 66.7 (CH_2); 67.2 (CH_2); 68.8 (CH_2); 69.2 (CH_2); 70.7 (CH_2); 70.8 (CH_2); 71.0 (CH_2); 71.4 (CH_2); 71.6 (CH_2); 71.7 (CH_2); 71.8 (CH_2); 112.1 (CH); 114.4 (CH); 125.1 (C); 126.2 (CH); 127.9 (C); 129.2 (CH); 129.3 (CH); 129.5 (C); 129.9 (C); 131.0 (CH); 132.0 (CH); 137.8 (C); 138.1 (C); 141.0 (CH); 149.6 (CH); 152.1 (C); 153.6 (CH).

Single crystals suitable for X-ray diffraction analysis were grown by slow diffusion of diethyl ether into a solution of **1**· PF_6 , (en) $\text{Pd}(\text{OTf})_2$ (**2**) and **8** in nitromethane.

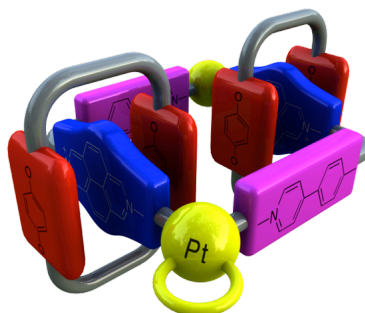
Catenane **6a(9)**₂·6 PF_6



6a(9)₂·6 PF_6

To a solution of **6a,b**·6 PF_6 (5.8 mg, 3.0×10^{-3} mmol) in CD_3NO_2 (0.6 mL) **9** (3.8 mg, 6.0×10^{-3} mmol) was added. ^1H NMR (500 MHz, CD_3NO_2). The complexity of the spectrum precluded its analysis (Figure S55).

Catenane **7a(8)**₂·6 PF_6

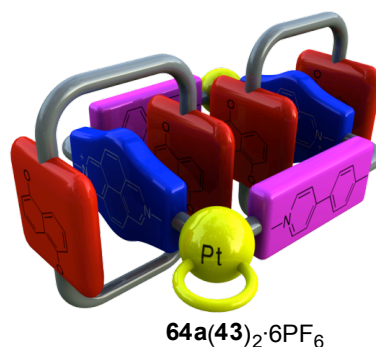


7a(8)₂·6 PF_6

A solution of **7a,b**·6 PF_6 (25.0 mg, 0.012 mmol) and BPP34C10 (**8**) (25.2 mg, 0.047 mmol) in CH_3NO_2 (2.4 mL) was stirred at 100 °C for 7d. After cooling to rt, the solvent was removed under reduced pressure without heating. The resulting residue was suspended in water (12 mL) and Amberlite™ IRA-402 (0.50 g) was added. The mixture was stirred at rt for 24h. The resin was removed by filtration and the filtrate was concentrated under reduced pressure. The resulting crude was purified by flash chromatography (SiO_2 , acetone/ $\text{NH}_4\text{Cl}_{\text{aq}}$ 1.5M/MeOH 5:4:1). The

product containing fractions were combined and the solvent removed under reduced pressure to afford a residue that was dissolved in H₂O (15 mL). An excess of KPF₆ was added until no further precipitation is observed. The solid was filtered and washed with water to yield catenane **7a(8)**₂·6PF₆ (24.0 mg, 64 %) as a yellow solid. ¹H NMR (500 MHz, CD₃NO₂) δ: 3.16 (16H, m); 3.45 (12H, m); 3.59-4.19 (52H, m); 4.88 (4H, m); 5.16 (4H, s); 5.41 (8H, s); 6.25 (4H, s); 7.75 (4H, d, *J* = 8.6 Hz); 7.81 (4H, d, *J* = 7.0 Hz); 7.93 (4H, d, *J* = 8.6 Hz); 8.27 (8H, m); 9.21 (4H, d, *J* = 6.9 Hz); 9.78 (4H, s); 9.90 (4H, s); ¹³C NMR (125 MHz, CD₃NO₂) δ: 49.4 (CH₂); 50.2 (CH₂); 66.7 (CH₂); 67.2 (CH₂); 68.8 (CH₂); 70.7 (CH₂); 70.9 (CH₂); 71.0 (CH₂); 71.5 (CH₂); 71.7 (CH₂); 71.8 (CH₂); 112.1 (CH); 114.5 (CH); 125.0 (C); 126.4 (CH); 127.7 (C); 129.4 (CH); 129.5 (CH); 129.6 (C); 130.0 (C); 131.0 (CH); 132.0 (CH); 137.9 (C); 137.9 (C); 141.1 (CH); 149.8 (CH); 152.1 (C); 153.8 (CH). HRMS-ES (*m/z*): calc. for [M – 3PF₆]³⁺ 920.5941, found 920.5905; calc. for [M – 4PF₆]⁴⁺ 654.2044, found 654.2035; calc. for [M – 5PF₆]⁵⁺ 494.3705, found 494,3700.

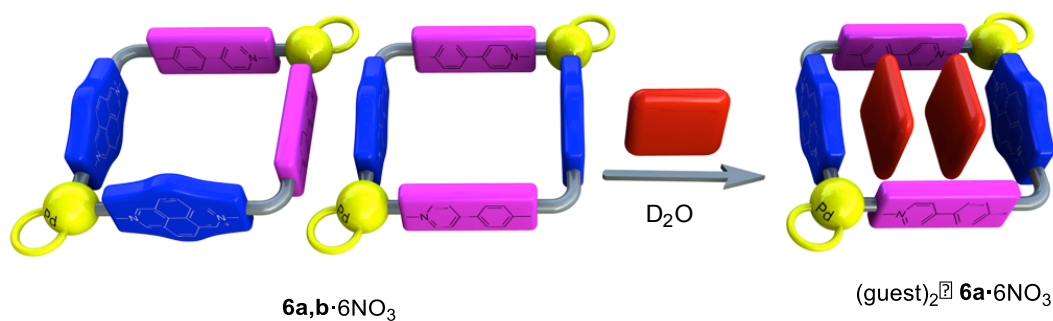
Catenane **7a(9)**₂·6PF₆



A solution of **7a,b**·6PF₆ (25.0 mg, 0.012 mmol) and **9** (30.0 mg, 0.047 mmol) in CH₃NO₂ (2.4 mL) was stirred at 100 °C for 7d. After cooling to rt, the solvent was removed under reduced pressure without heating. The resulting residue was suspended in water (12 mL) and Amberlite™ IRA-402 (0,50 g) was added. The mixture was stirred at rt for 24h. The resin was removed by filtration and the filtrate was concentrated under reduce pressure. The resulting crude was purified by flash chromatography (SiO₂, acetone/NH₄Cl_{aq} 1.5M/MeOH 5:4:1). The product containing fractions were combined and the solvent removed under reduced pressure to afford a residue that was dissolved in H₂O (15 mL). An excess of KPF₆ was added until no further precipitation is observed. The solid was filtered and washed with water to yield catenane **7a(9)**₂·6PF₆ (25.3 mg, 63 %) as a yellow solid.¹H NMR (500 MHz, CD₃NO₂) The complexity of the spectrum precluded its analysis (Figure S58).

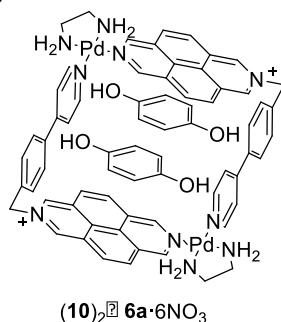
HRMS-ES (*m/z*): calc. for [M – 3PF₆]³⁺ 987.2816, found 987.2800; calc. for [M – 4PF₆]⁴⁺ 704.2200, found 704.2212; calc. for [M – 5PF₆]⁵⁺ 534.3831, found 534.3828; calc. for [M – 6PF₆]⁶⁺ 421.1585, found 421.1593.

General procedure for the regioselective formation of inclusion complexes



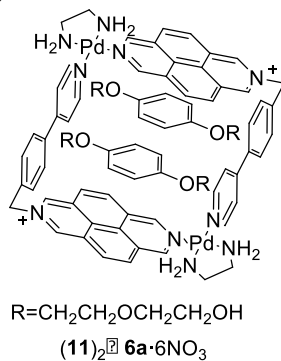
To a solution of **4**·NO₃ (4.3 mg; 0.010 mmol) and (en)Pd(NO₃)₂ (**5a**) (2.9 mg; 0.010 mmol) in D₂O (2.0 mL) guests **15**, **17**, **19**, **20**, **21** (0.010 mmol), **11**, **13**, **16**, **18** (0.020 mmol) or **10** (0.070 mmol) were added and the mixture was stirred at 60°C for 1h.

Inclusion complex (**10**)₂·**6a**·6NO₃



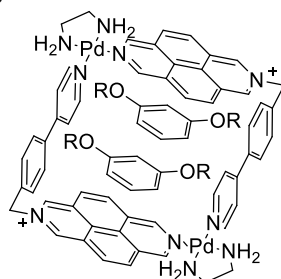
¹H NMR (500 MHz, D₂O) δ: 3.01 (8H, s); 6.13 (4H, s); 6.24 (56H, s); 7.74 (4H, d, *J* = 8.5 Hz); 7.84 (8H, m); 8.31 (4H, d, *J* = 9.2 Hz); 8.44 (4H, d, *J* = 9.2 Hz); 9.01 (4H, d, *J* = 6.8 Hz); 9.80 (4H, s); 10.01 (4H, s); ¹³C NMR (125 MHz, D₂O) δ: 46.4 (CH₂); 46.5 (CH₂); 65.4 (CH₂); 115.3 (CH); 123.8 (CH); 124.2 (C); 125.7 (C); 127.5 (CH); 127.7 (C); 127.9 (CH); 128.8 (C); 129.1 (CH); 129.6 (CH); 135.8 (C); 135.9 (C); 138.0 (CH); 147.3 (CH); 148.0 (C); 150.2 (C); 150.8 (CH).

Inclusion complex (**11**)₂·**6a**·6NO₃



^1H NMR (500 MHz, D_2O) δ : 3.00 (8H, s); 3.72-3.83 (64H, m); 6.18 (4H, s); 7.81 (4H, d, J = 8.5 Hz); 7.90 (8H, m); 8.36 (4H, d, J = 9.2 Hz); 8.48 (4H, d, J = 9.2 Hz); 9.11 (4H, d, J = 6.8 Hz); 9.90 (4H, s); 10.08 (4H, s); ^{13}C NMR (125 MHz, D_2O) δ : 46.8 (CH_2); 46.9 (CH_2); 60.5 (CH_2); 65.8 (CH_2); 67.2 (CH_2); 68.9 (CH_2); 71.9 (CH_2); 114.4 (CH); 124.3 (CH); 126.1 (C); 127.8 (CH); 128.1 (C); 128.3 (CH); 129.1 (C); 129.8 (CH); 130.2 (CH); 136.4 (C); 136.6 (C); 138.7 (CH); 147.9 (CH); 150.4 (C); 151.3 (C); 151.5 (CH).

Inclusion complex $(13)_2 \subset 6a \cdot 6\text{NO}_3$

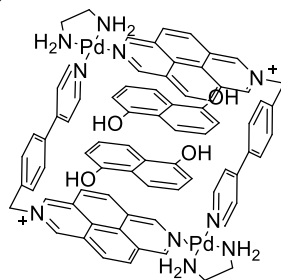


R = $\text{CH}_2\text{CH}_2\text{OCH}_2\text{CH}_2\text{OH}$

$(13)_2 \subset 6a \cdot 6\text{NO}_3$

^1H NMR (500 MHz, D_2O) δ : 3.00 (8H, s); 3.72-3.81 (64H, m); 5.93 (6H, br); 6.18 (4H, s); 7.80 (4H, d, J = 8.5 Hz); 7.89 (8H, m); 8.35 (4H, d, J = 9.2 Hz); 8.46 (4H, d, J = 9.2 Hz); 9.07 (4H, d, J = 6.8 Hz); 9.90 (4H, s); 10.05 (4H, s); ^{13}C NMR (125 MHz, D_2O) δ : 46.8 (CH_2); 46.9 (CH_2); 60.5 (CH_2); 65.8 (CH_2); 66.8 (CH_2); 68.9 (CH_2); 71.9 (CH_2); 100.5 (CH); 106.5 (CH); 124.3 (CH); 126.3 (C); 127.9 (CH); 128.1 (C); 128.3 (CH); 129.0 (C); 129.3 (CH); 129.7 (CH); 130.4 (CH); 136.3 (C); 136.5 (C); 138.8 (CH); 147.9 (CH); 150.5 (C); 151.5 (CH); 158.3 (C).

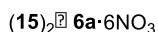
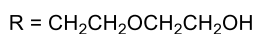
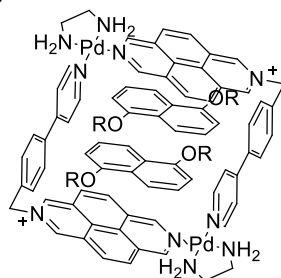
Inclusion complex $(14)_2 \subset 6a \cdot 6\text{NO}_3$



$(14)_2 \subset 6a \cdot 6\text{NO}_3$

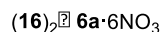
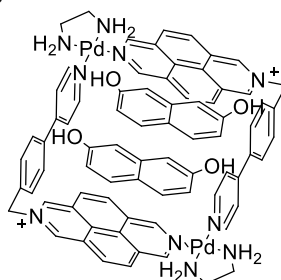
^1H NMR (500 MHz, D_2O) δ : 2.99 (8H, m); 5.60 (8H, m); 6.01 (4H, s); 7.77 (4H, d, J = 9.1 Hz); 7.92 (8H, m); 7.97 (4H, d, J = 8.6 Hz); 8.08 (4H, d, J = 6.8 Hz); 9.13 (4H, d, J = 6.9 Hz); 9.47 (4H, s); 9.72 (4H, s); ^{13}C NMR (125 MHz, D_2O) δ : 46.7 (CH_2); 46.8 (CH_2); 65.8 (CH_2); 107.1 (CH); 111.4 (CH); 123.1 (C); 124.5 (CH); 124.9 (C); 126.7 (CH); 127.5 (C); 128.2 (C); 128.4 (CH); 128.8 (CH); 130.5 (CH); 135.4 (C); 136.6 (C); 137.7 (CH); 147.3 (CH); 149.5 (C); 150.7 (C); 151.4 (CH).

Inclusion complex (15)₂⊂6a·6NO₃



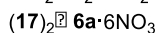
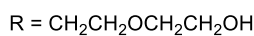
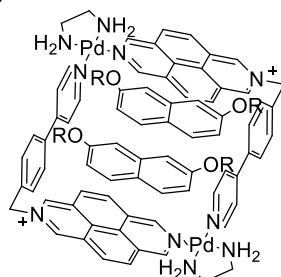
¹H NMR (500 MHz, D₂O, 353 K) δ: 3.40-3.49 (8H, m); 4.06 (8H, m); 4.33 (8H, m); 4.41 (8H, m); 4.50 (8H, m); 5.90-6.07 (6H, m); 6.55 (4H, s); 8.38 (4H, d, *J* = 9.1 Hz); 8.48-8.59 (12H, m); 8.68 (4H, d, *J* = 5.7 Hz); 9.71 (4H, d, *J* = 6.7 Hz); 10.02 (4H, s); 10.21 (4H, s).

Inclusion complex (16)₂⊂6a·6NO₃



¹H NMR (500 MHz, D₂O) δ: 3.01 (8H, s); 6.03 (4H, s); 7.81 (4H, d, *J* = 9.2 Hz); 7.94 (4H, d, *J* = 8.7 Hz); 8.00 (4H, d, *J* = 8.7 Hz); 8.05 (4H, d, *J* = 9.2 Hz); 8.12 (4H, d, *J* = 7.0 Hz); 9.19 (4H, d, *J* = 7.0 Hz); 9.50 (4H, s); 9.87 (4H, s); ¹³C NMR (125 MHz, D₂O) δ: 46.8 (CH₂); 46.9 (CH₂); 65.7 (CH₂); 105.9 (CH); 113.3 (CH); 121.1 (C); 123.5 (C); 124.5 (CH); 124.9 (C); 126.8 (CH); 127.2 (C); 127.8 (C); 128.5 (CH); 128.6 (C); 128.9 (CH); 130.4 (CH); 133.1 (C); 136.4 (C); 136.6 (C); 137.8 (CH); 147.4 (CH); 150.5 (C); 151.6 (CH); 152.0 (C).

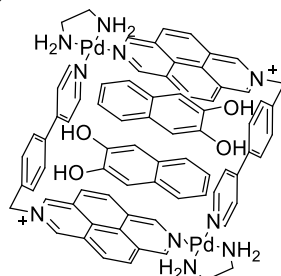
Inclusion complex (17)₂⊂6a·6NO₃



¹H NMR (500 MHz, D₂O) δ: 3.00 (8H, s); 3.29 (4H, m); 3.88-4.00 (32H, m); 4.51 (4H, m); 5.46 (m); 6.11 (4H, s); 7.89 (4H, d, *J* = 9.2 Hz); 8.00 (8H, m); 8.11 (4H, d, *J* = 9.1 Hz); 8.16 (4H, d, *J*

= 5.4 Hz); 9.33 (4H, m); 9.69 (4H, s); 9.96 (4H, s); ^{13}C NMR (125 MHz, D_2O) δ : 46.8 (CH_2); 46.9 (CH_2); 60.7 (CH_2); 65.7 (CH_2); 68.8 (CH_2); 72.3 (CH_2); 102.8 (CH); 114.2 (CH); 123.3 (C); 124.3 (CH); 125.2 (C); 126.9 (CH); 127.9 (C); 128.3 (CH); 128.7 (C); 129.4 (CH); 130.4 (CH); 136.3 (C); 137.1 (C); 138.1 (CH); 147.6 (CH); 150.4 (C); 152.0 (CH); 154.1 (C).

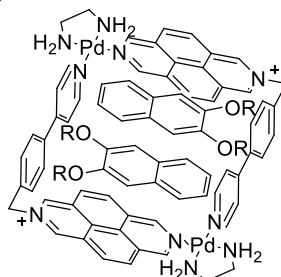
Inclusion complex $(18)_2 \subset 6a \cdot 6\text{NO}_3$



(56)₂ · 63a · 6NO₃

^1H NMR (500 MHz, D_2O) δ : 3.00 (8H, m); 4.26 (8H, m); 5.51 (4H, m); 6.05 (4H, s); 7.86 (4H, d, $J = 9.1$ Hz); 7.95 (4H, d, $J = 8.5$ Hz); 7.99 (4H, d, $J = 8.2$ Hz); 8.10 (8H, m); 9.18 (4H, d, $J = 6.3$ Hz); 9.53 (4H, s); 9.87 (4H, s); ^{13}C NMR (125 MHz, D_2O) δ : 46.8 (CH_2); 46.9 (CH_2); 65.7 (CH_2); 108.2 (CH); 121.5 (CH); 122.9 (CH); 123.7 (C); 124.5 (CH); 125.1 (C); 125.9 (C); 126.9 (CH); 127.9 (C); 128.5 (CH); 128.7 (C); 129.0 (CH); 130.5 (CH); 136.5 (C); 136.5 (C); 138.0 (CH); 142.8 (C); 147.4 (CH); 150.6 (C); 151.6 (CH).

Inclusion complex $(19)_2 \subset 6a \cdot 6\text{NO}_3$

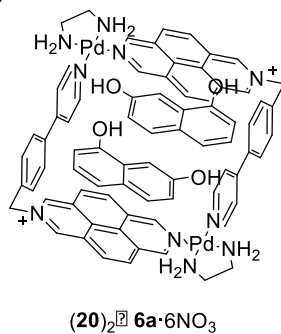


R = $\text{CH}_2\text{CH}_2\text{OCH}_2\text{CH}_2\text{OH}$

(57)₂ · 63a · 6NO₃

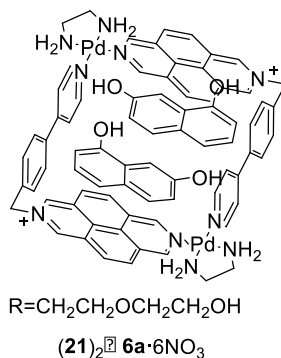
^1H NMR (500 MHz, D_2O) δ : 2.98 (8H, m); 3.90-3.98 (32H, m); 5.99 (4H, m); 6.14 (4H, s); 7.89 (4H, m); 8.05 (4H, d, $J = 8.6$ Hz); 8.10 (4H, d, $J = 8.5$ Hz); 8.22 (4H, d, $J = 6.5$ Hz); 9.32 (4H, d, $J = 6.7$ Hz); 9.68 (4H, s); 9.94 (4H, s); ^{13}C NMR (125 MHz, D_2O) δ : 46.9 (CH_2); 47.0 (CH_2); 60.6 (CH_2); 65.7 (CH_2); 68.4 (CH_2); 72.3 (CH_2); 104.4 (CH); 123.0 (CH); 123.4 (C); 124.2 (CH); 125.1 (C); 127.0 (CH); 128.0 (C); 128.4 (CH); 128.8 (C); 129.2 (CH); 130.7 (CH); 136.1 (C); 137.3 (C); 138.2 (CH); 145.2 (CH); 147.6 (CH); 150.4 (C); 151.1 (CH).

Inclusion complex (20)₂⊂6a·6NO₃



¹H NMR (500 MHz, D₂O) δ: 2.98 (8H, m); 5.28 (4H, br m); 5.52 (4H, br m); 6.01 (4H, s); 7.75 (4H, d, *J* = 9.1 Hz); 7.99 (16H, m); 9.06 (4H, d, *J* = 7.0 Hz); 9.45 (4H, s); 9.82 (4H, s); ¹³C NMR (125 MHz, D₂O) δ: 46.7 (CH₂); 46.8 (CH₂); 65.8 (CH₂); 101.7 (CH); 106.3 (CH); 115.8 (CH); 117.0 (C); 121.6 (CH); 122.5 (C); 123.6 (C); 124.6 (CH); 125.0 (C); 126.5 (CH); 127.0 (C); 127.6 (C); 128.5 (CH); 128.8 (CH); 130.4 (CH); 135.9 (C); 137.0 (C); 137.8 (CH); 147.5 (CH); 148.5 (C); 150.5 (C); 151.2 (CH).

Inclusion complex (21)₂⊂6a·6NO₃



¹H NMR (500 MHz, D₂O, 338 K) δ: 3.35-3.42 (8H, m); 4.11-4.29 (24H, m); 4.46 (8H, m); 6.50 (4H, s); 6.78 (8H, m); 7.10 (8H, m); 8.23 (4H, d, *J* = 9.5 Hz); 8.33 (4H, d, *J* = 8.3 Hz); 8.41 (8H, m); 8.49 (4H, d, *J* = 6.6 Hz); 9.60 (4H, d, *J* = 6.2 Hz); 9.95 (4H, s); 10.25 (4H, s).

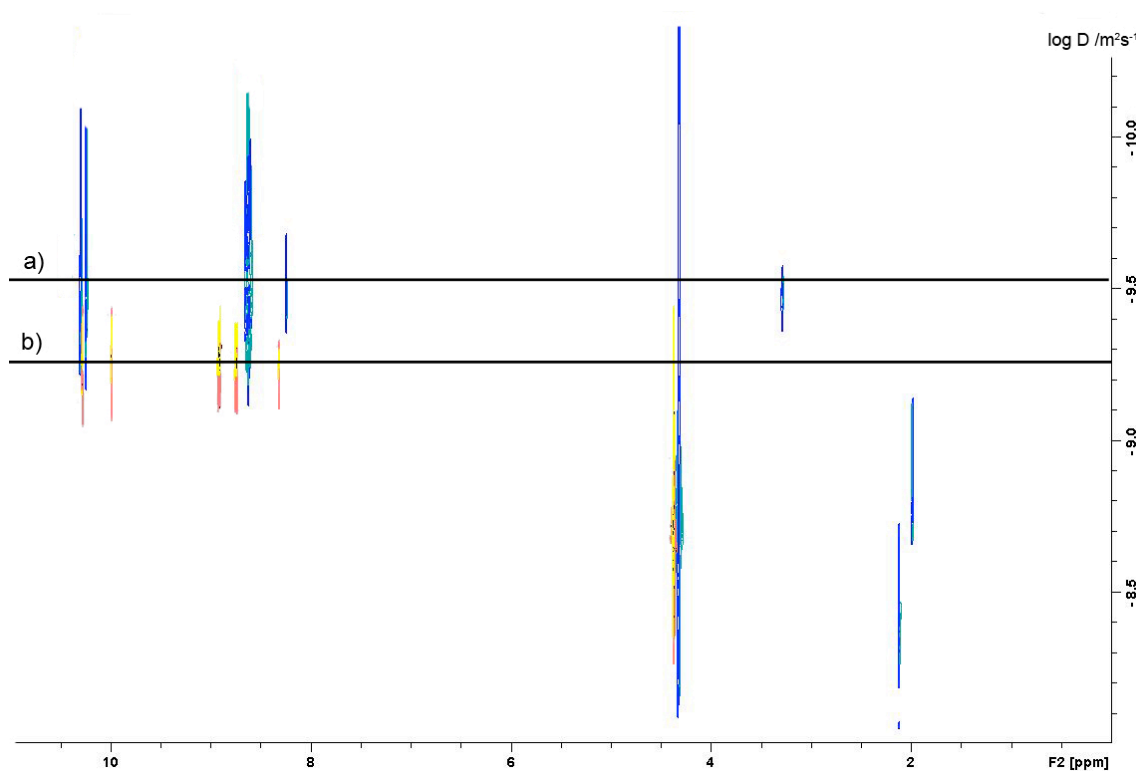


Figure S1: Superposed DOSY (CD_3NO_2 , 500 MHz, 298 K) experiment of: a) 1.25 mM solution of $\mathbf{3} \cdot 4\text{OTf} \cdot 4\text{PF}_6$ (blue). b) 2.5 mM solution of ligand $\mathbf{1} \cdot 2\text{PF}_6$ (yellow).

1-(4-(pyridin-4-yl)benzyl)-2,7-diazapyren-1-ium hexafluorophosphate ($4 \cdot \text{PF}_6$)

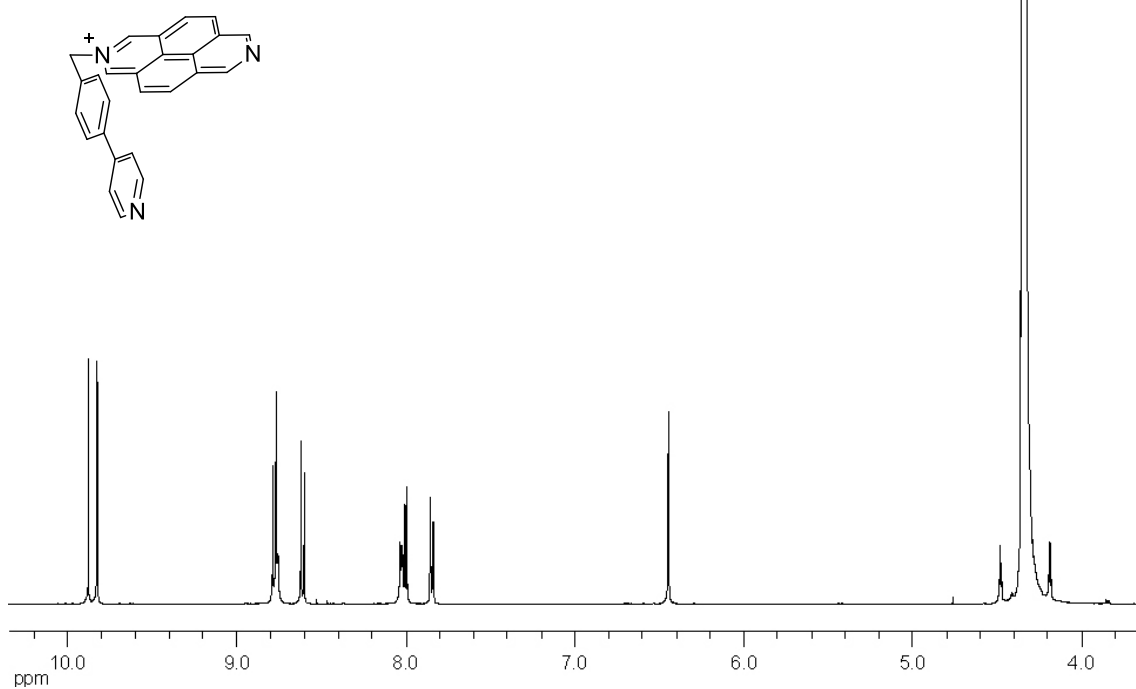


Figure S2: ^1H NMR (CD_3NO_2 , 500 MHz) spectrum of $4 \cdot \text{PF}_6$

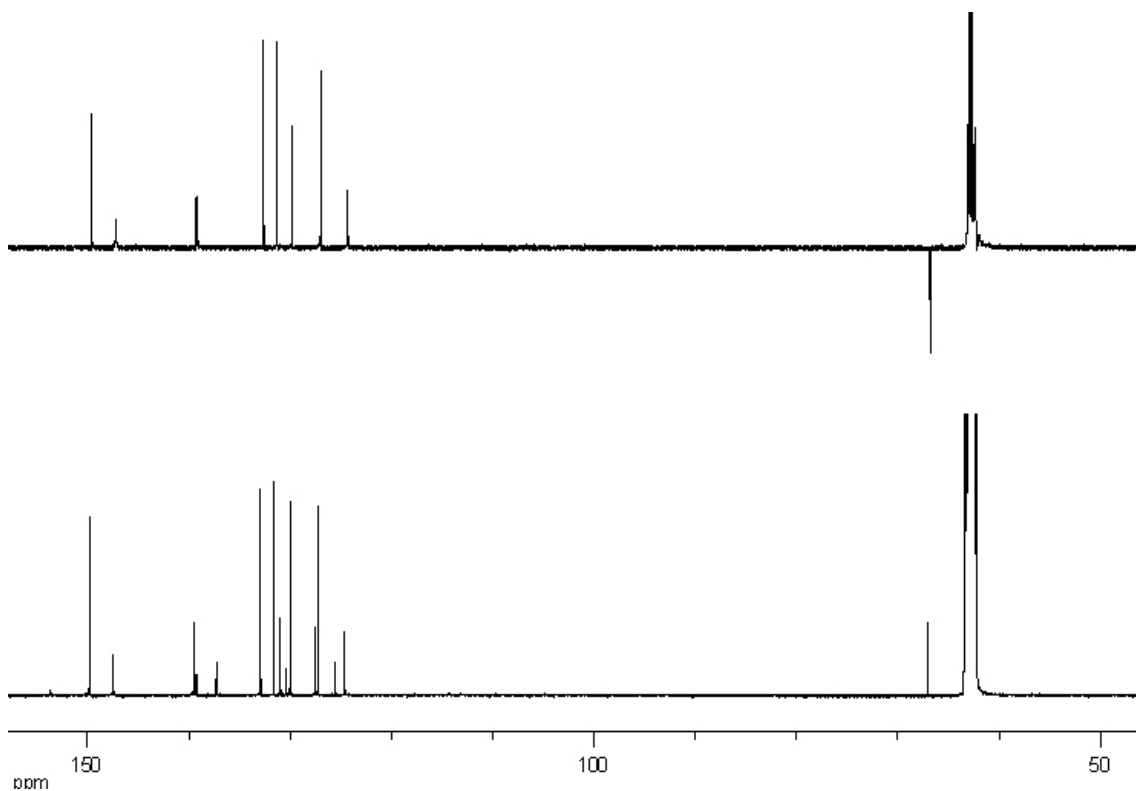


Figure S3: ^{13}C NMR (CD_3NO_2 , 125 MHz) spectrum of $4 \cdot \text{PF}_6$.

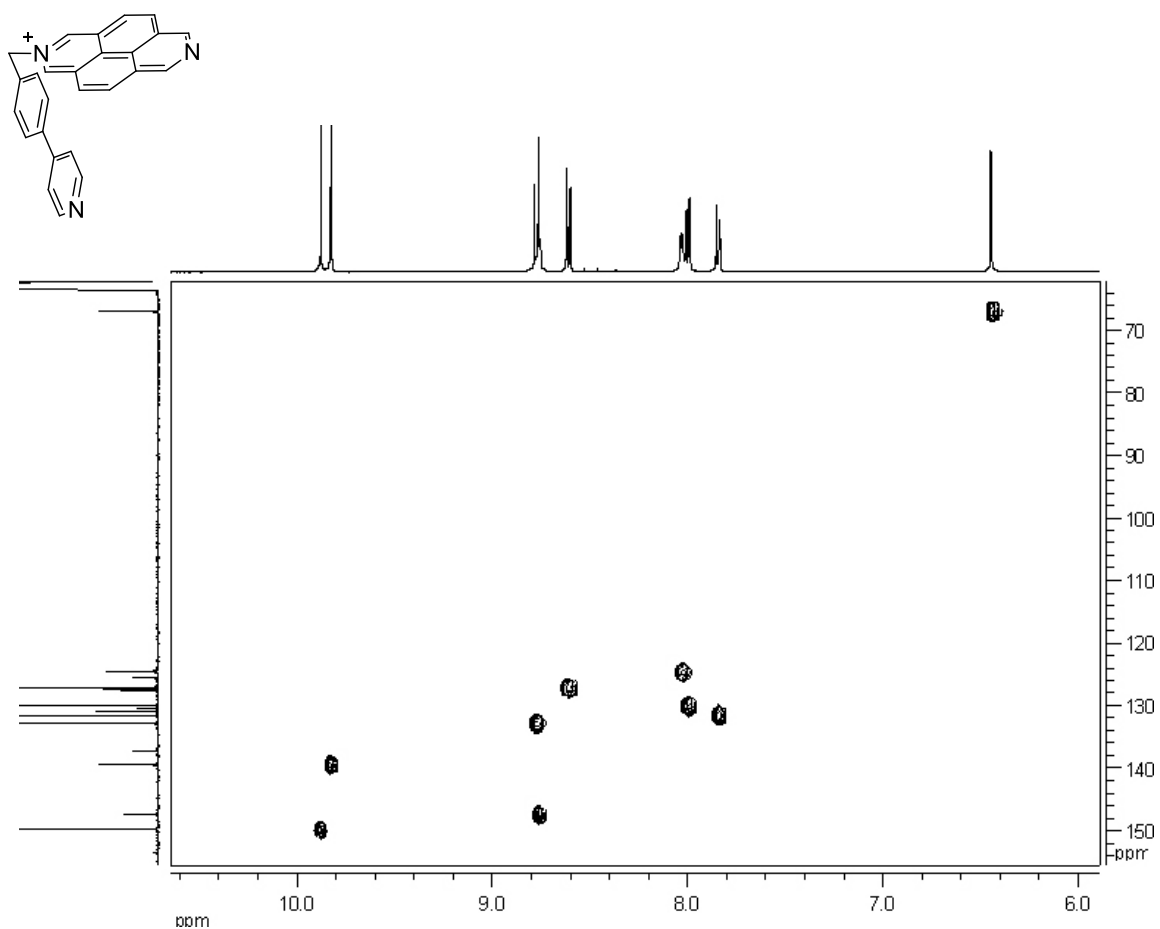


Figure S4: HSQC (CD₃NO₂, 500 and 125 MHz) spectrum of 4·PF₆.

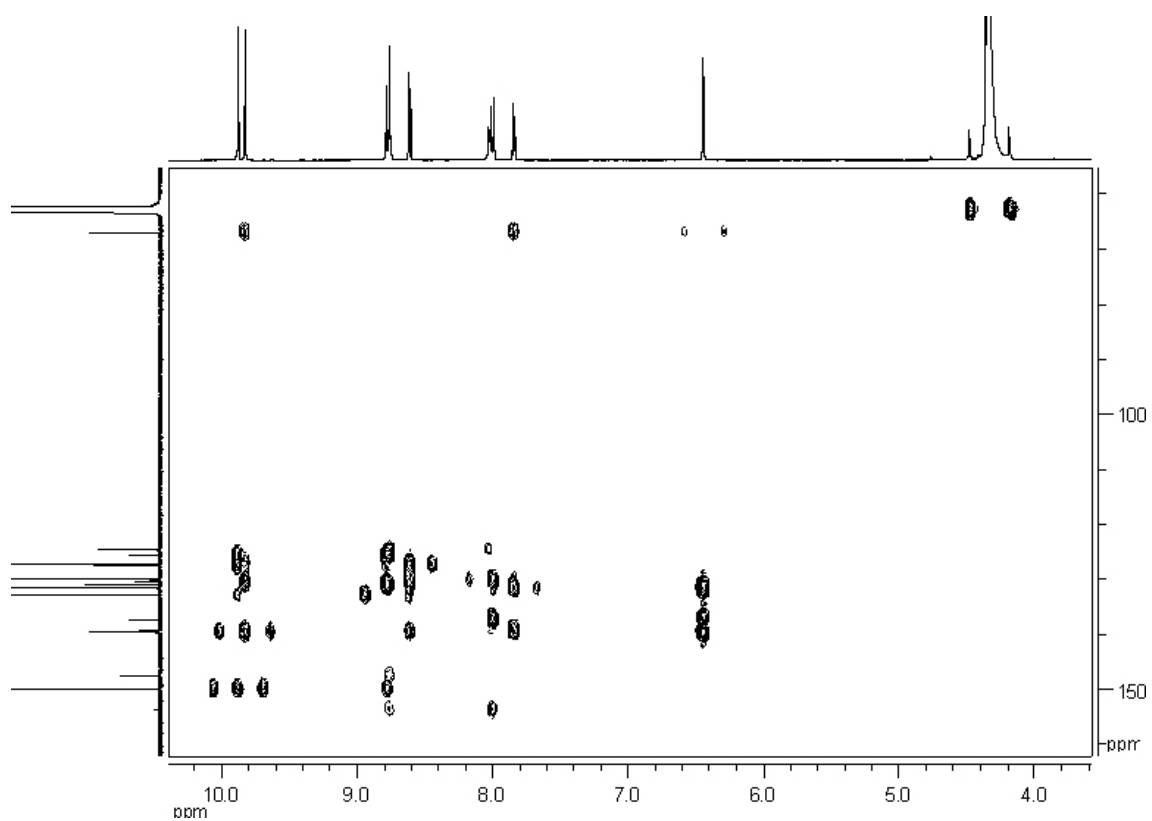


Figure S5: HMBC (CD₃NO₂, 500 and 125 MHz) spectrum of 4·PF₆

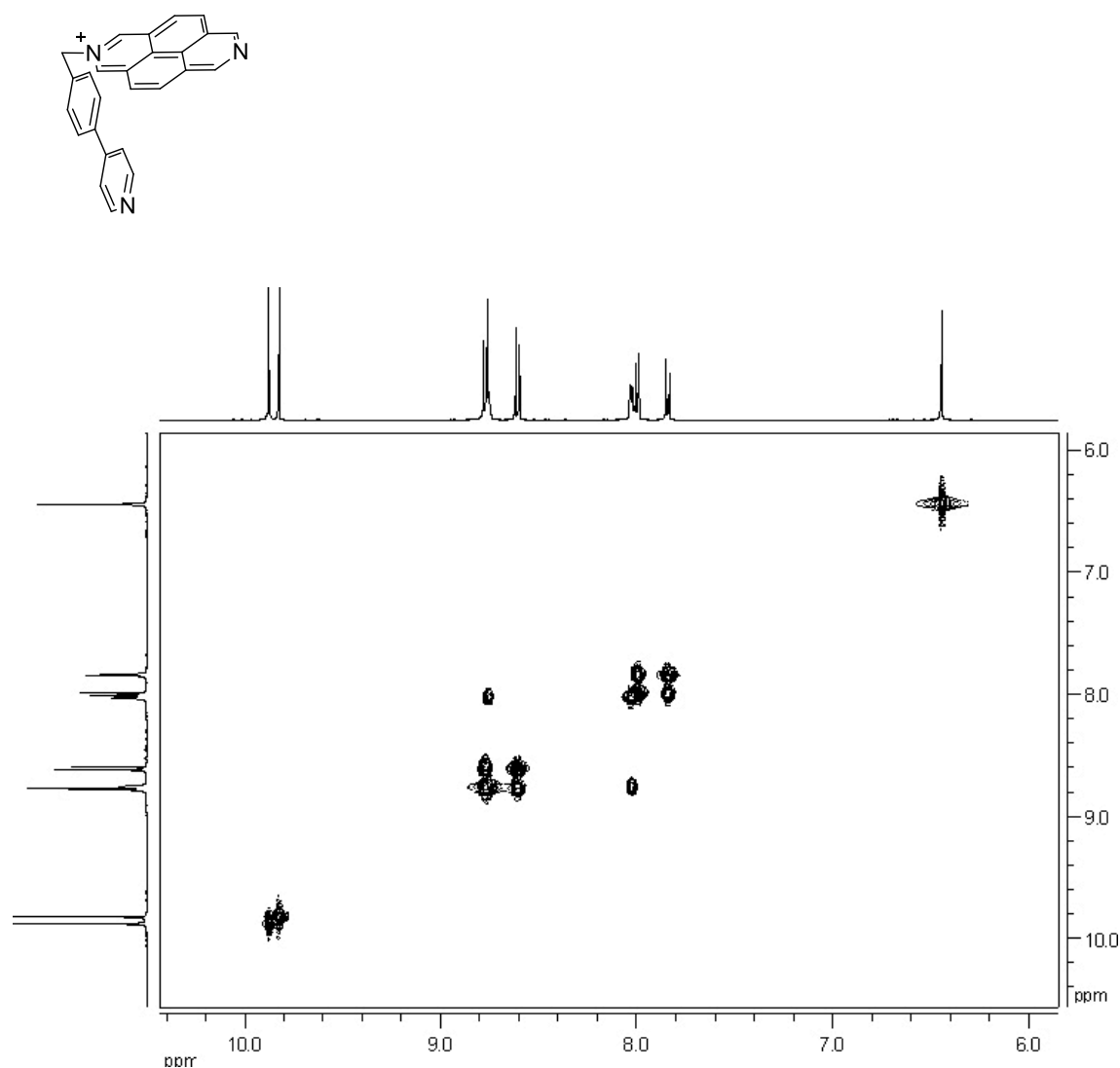


Figure S6: COSY (CD₃NO₂, 500 MHz) spectrum of **4**·PF₆.

1-(4-(pyridin-4-yl)benzyl)-2,7-diazapyren-1-ium nitrate ($4 \cdot \text{NO}_3$)

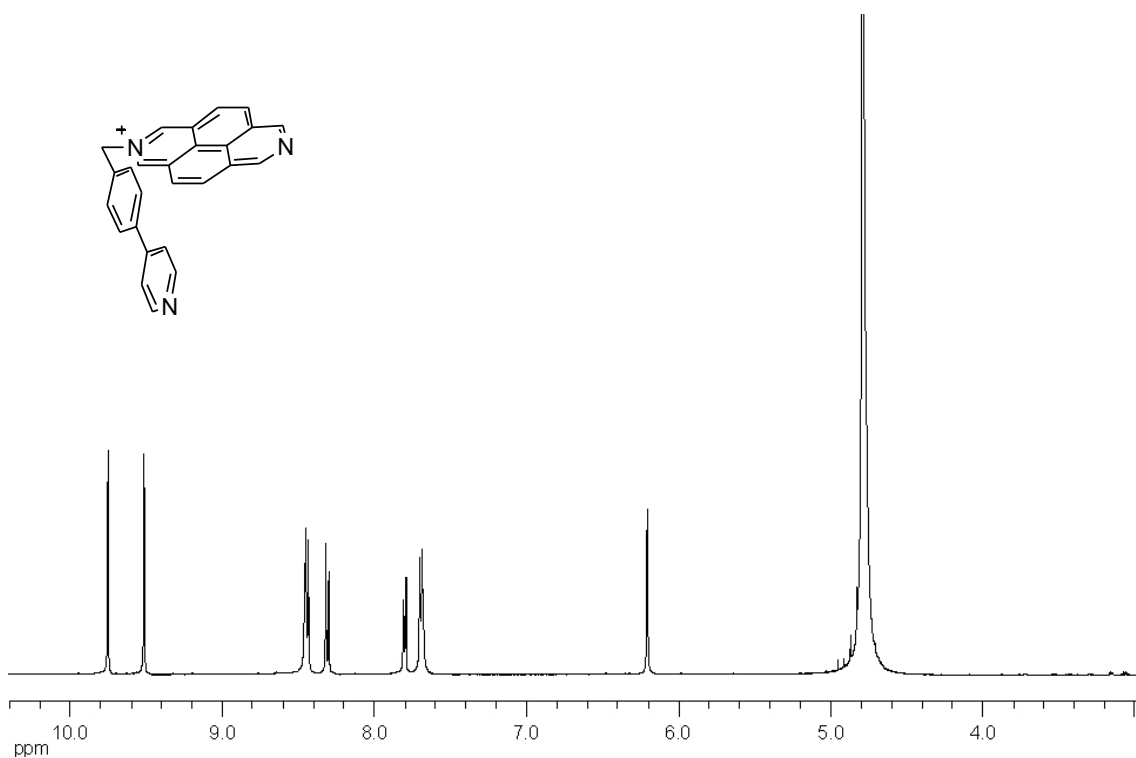


Figure S7: ^1H NMR (D_2O , 500 MHz) spectrum of $4 \cdot \text{NO}_3$

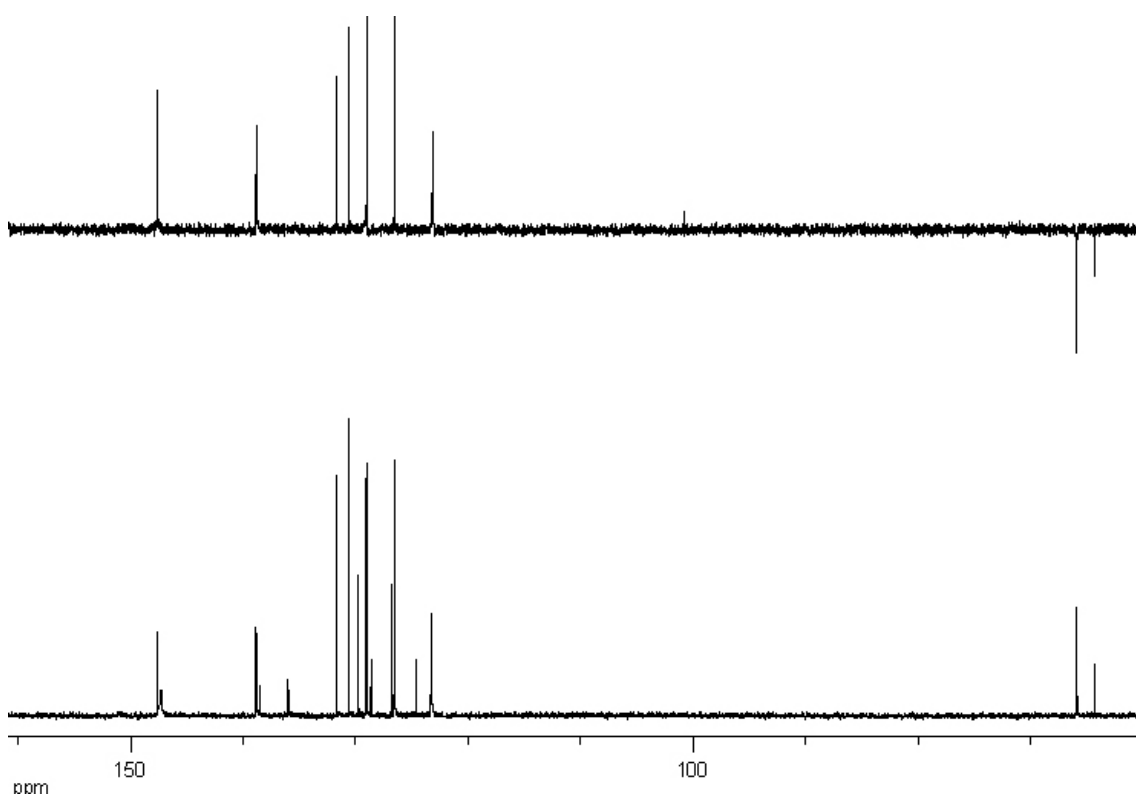


Figure S8: ^{13}C NMR (D_2O , 125 MHz) spectrum of $4 \cdot \text{NO}_3$.

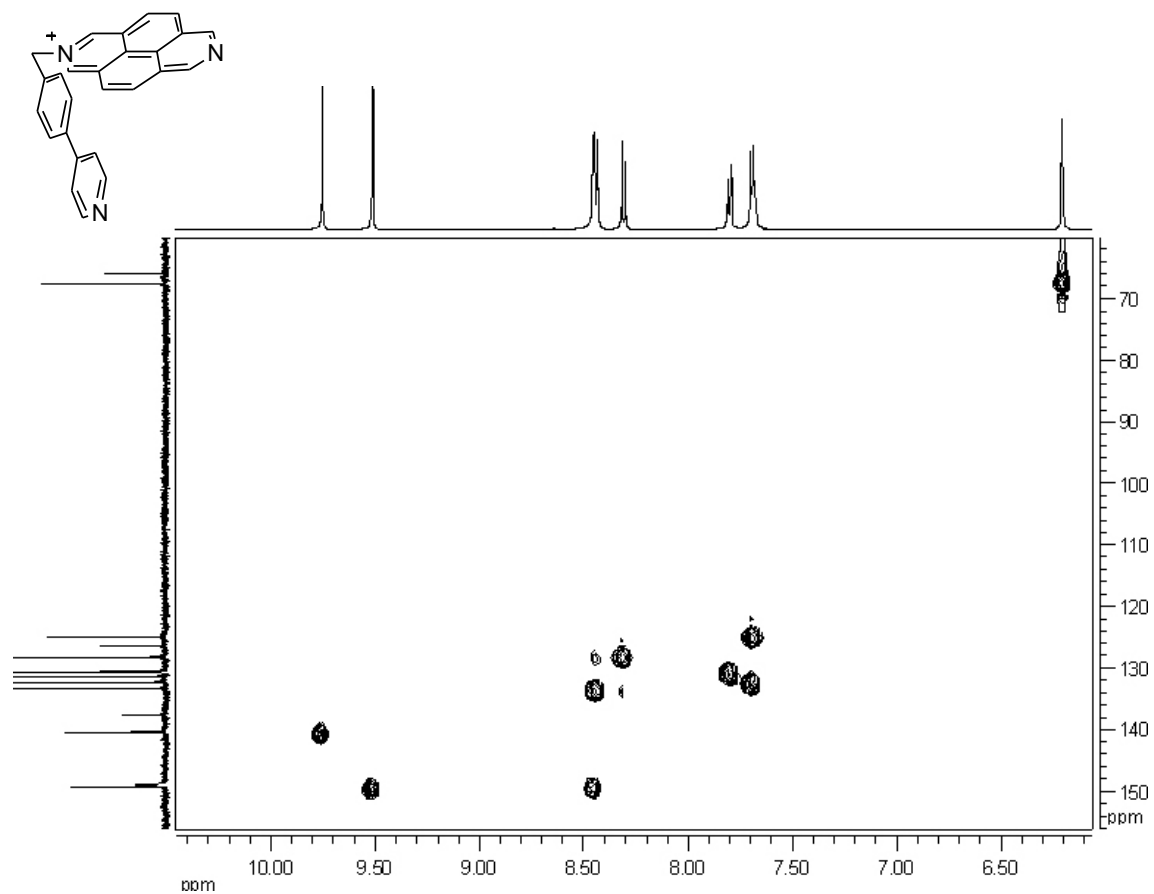


Figure S9: HSQC (D₂O, 500 and 125 MHz) spectrum of 4·NO₃

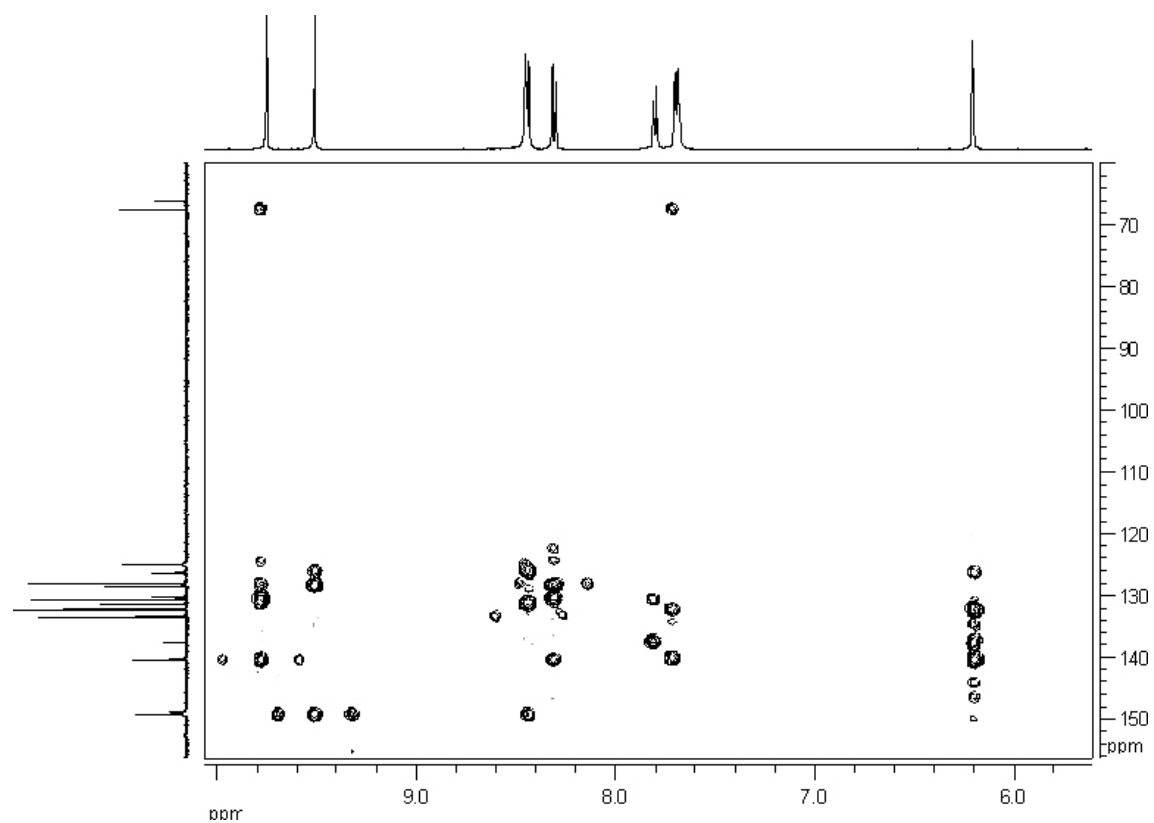


Figure S10: HMBC (D₂O, 500 and 125 MHz) spectrum of 4·NO₃.

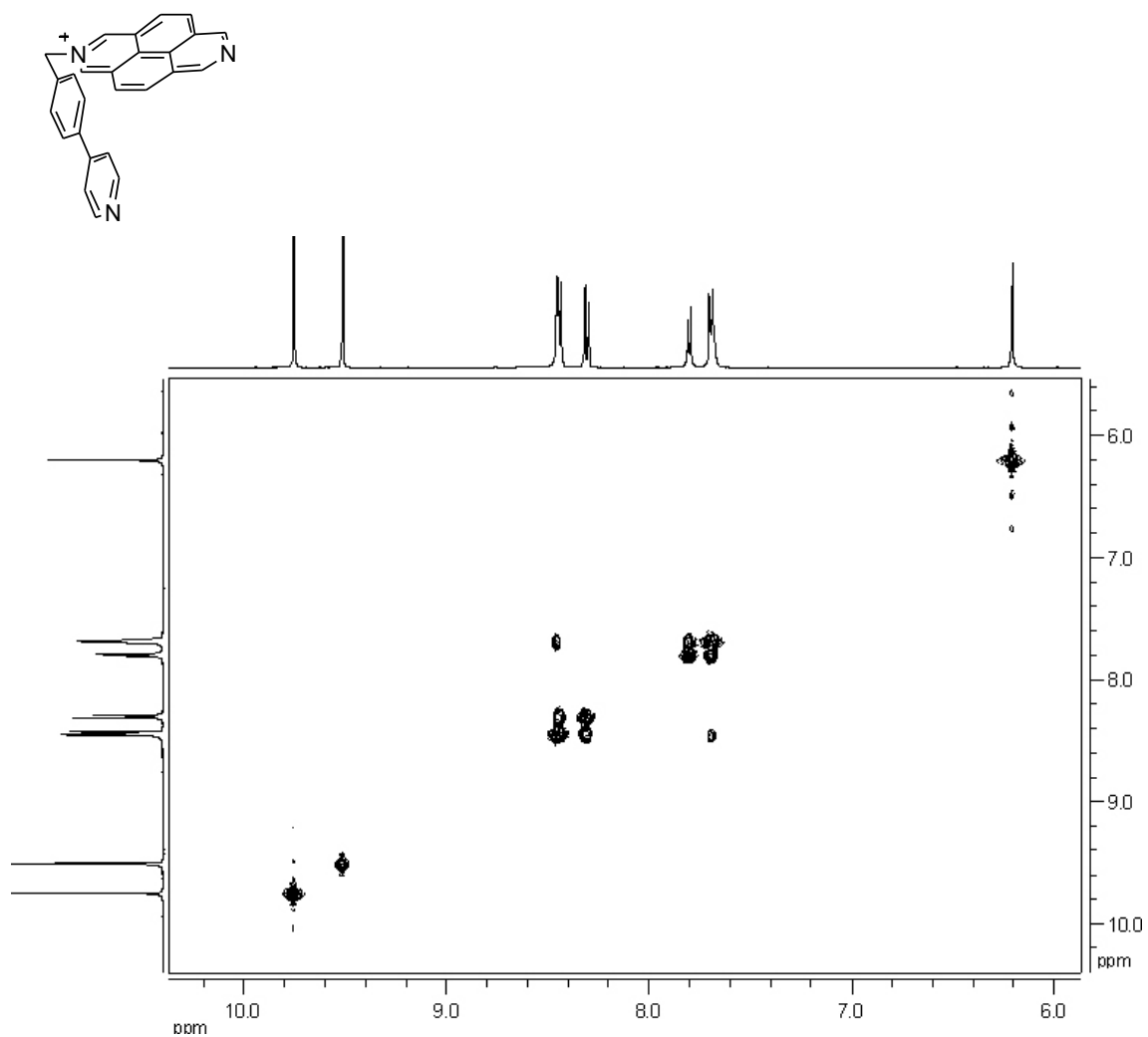


Figure S11: COSY (D₂O, 500 MHz) spectrum of 4·NO₃.

Metallocycles **6a,b·6NO₃**

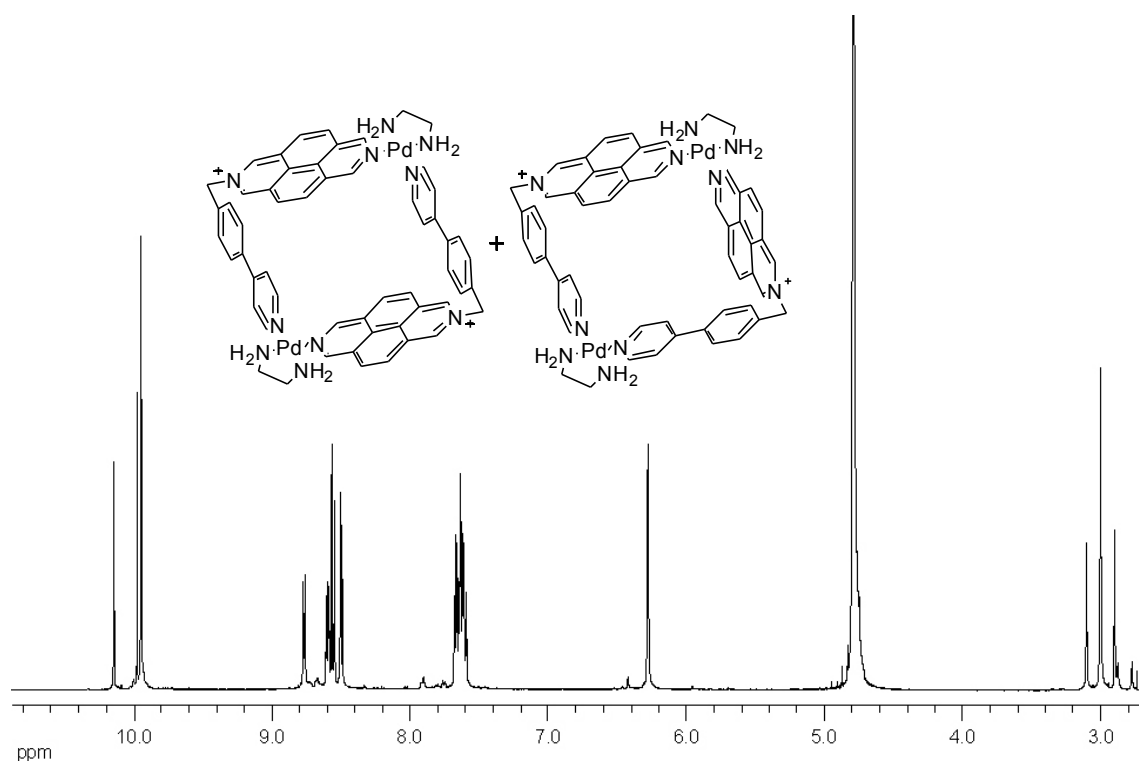


Figure S12: ¹H NMR (D₂O, 500 MHz) spectrum of **6a,b**·6NO₃

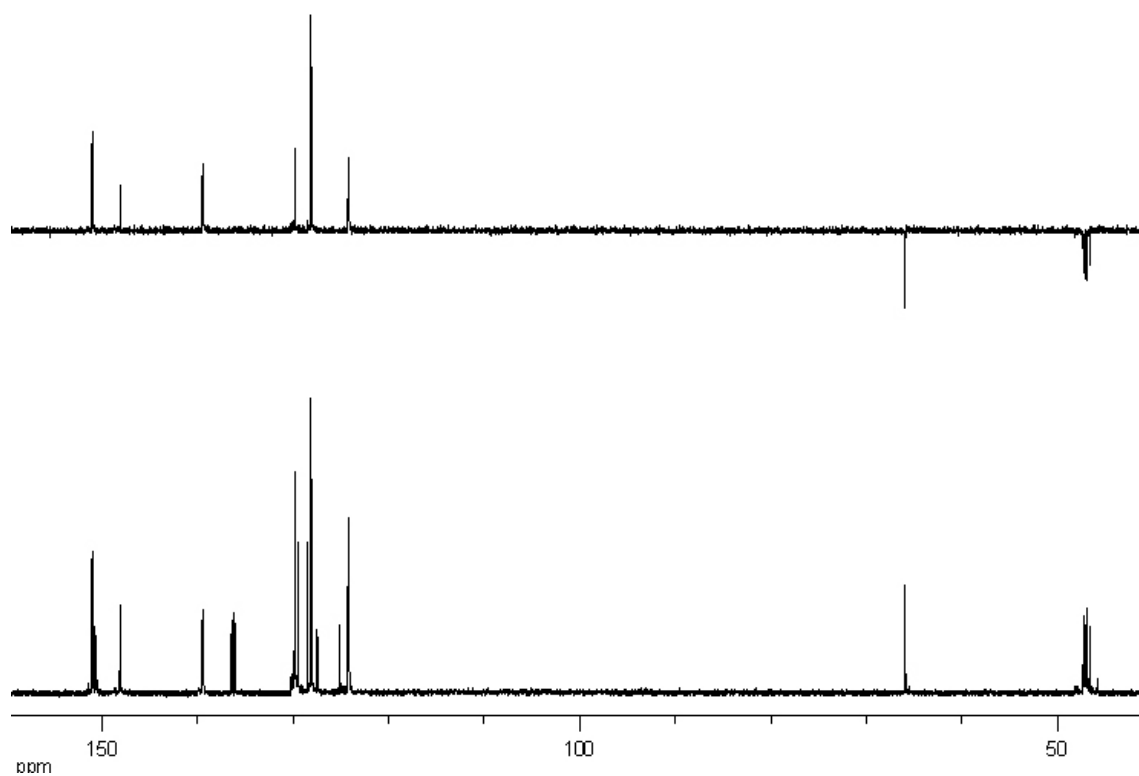


Figure S13: ¹³C NMR (D₂O, 125 MHz) spectrum of **6a,b**·6NO₃.

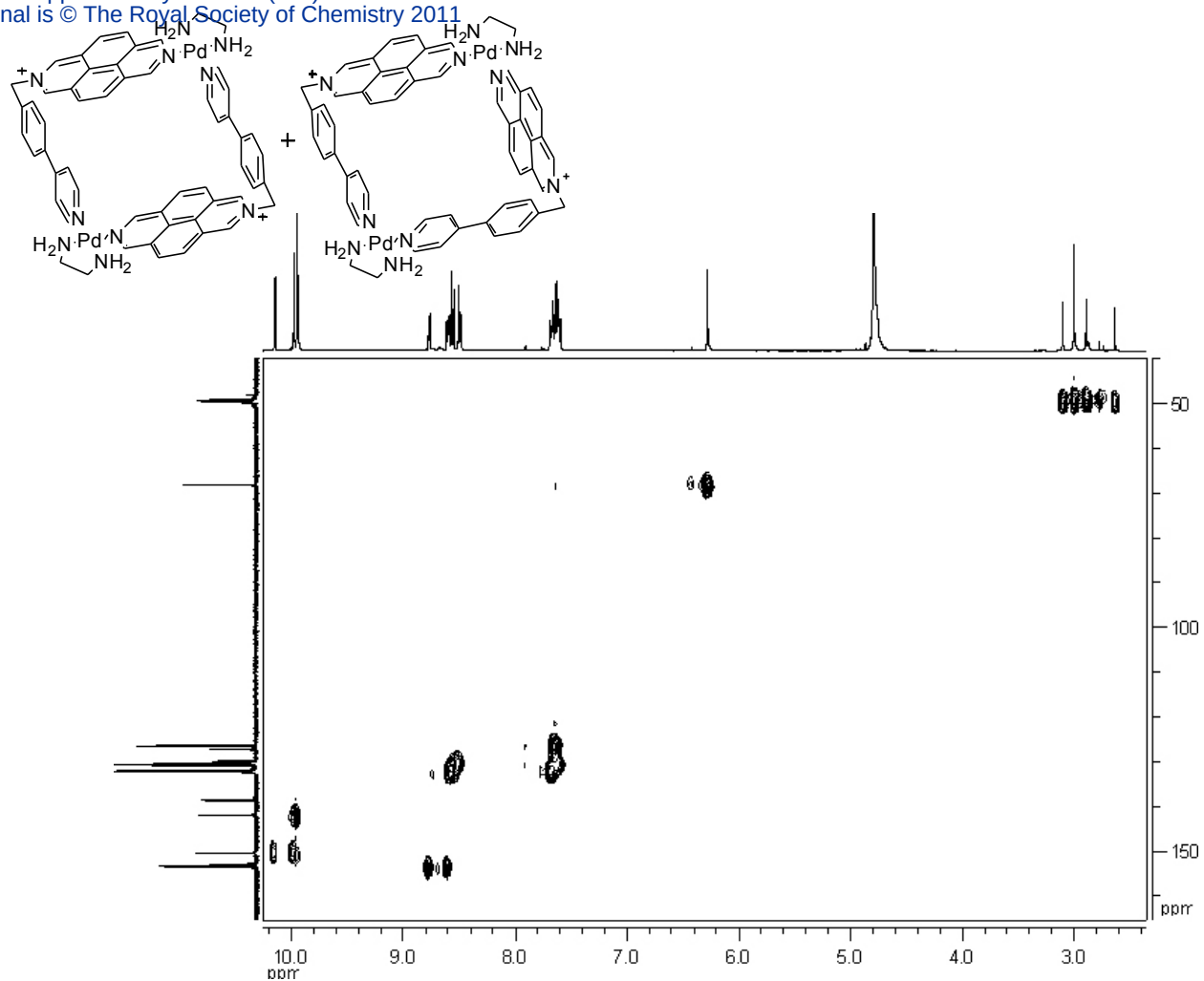


Figure S14: HSQC (D_2O , 500 and 125 MHz) spectrum of **6a,b**· 6NO_3 .

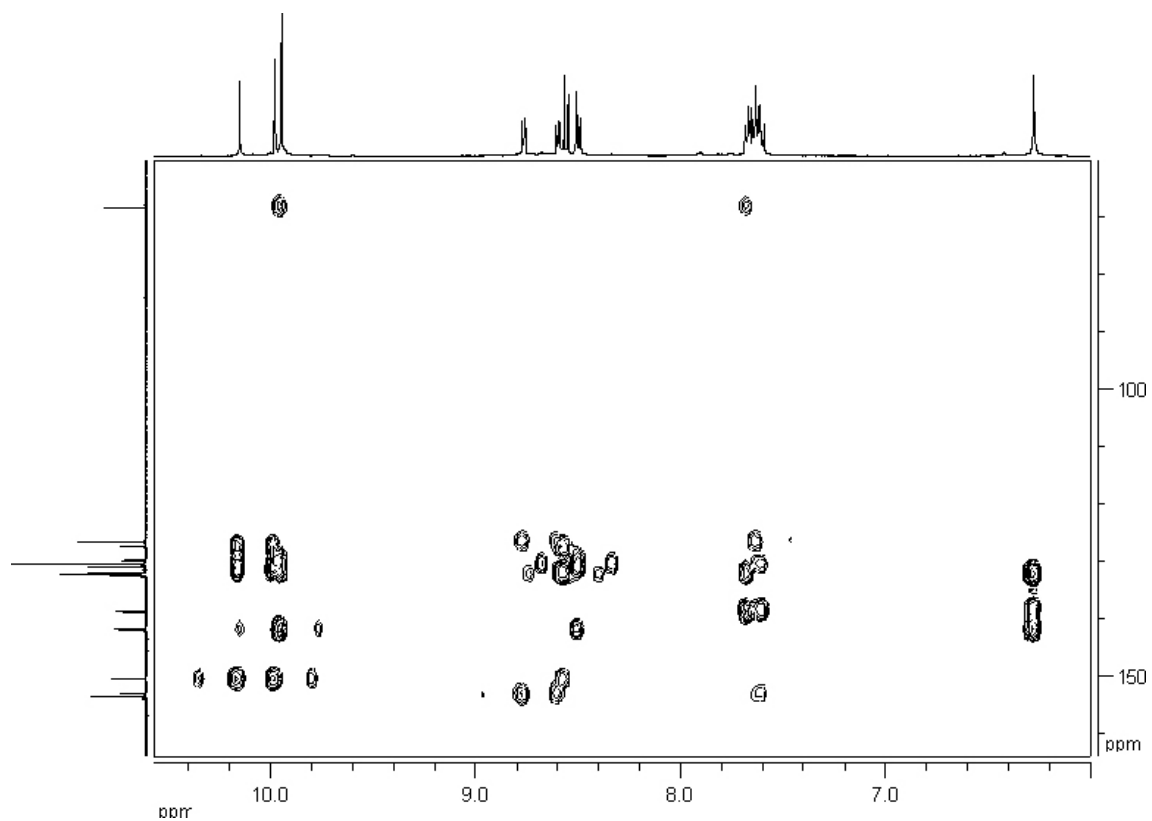


Figure S15: HMBC (D_2O , 500 and 125 MHz) spectrum of **6a,b**· 6NO_3 .

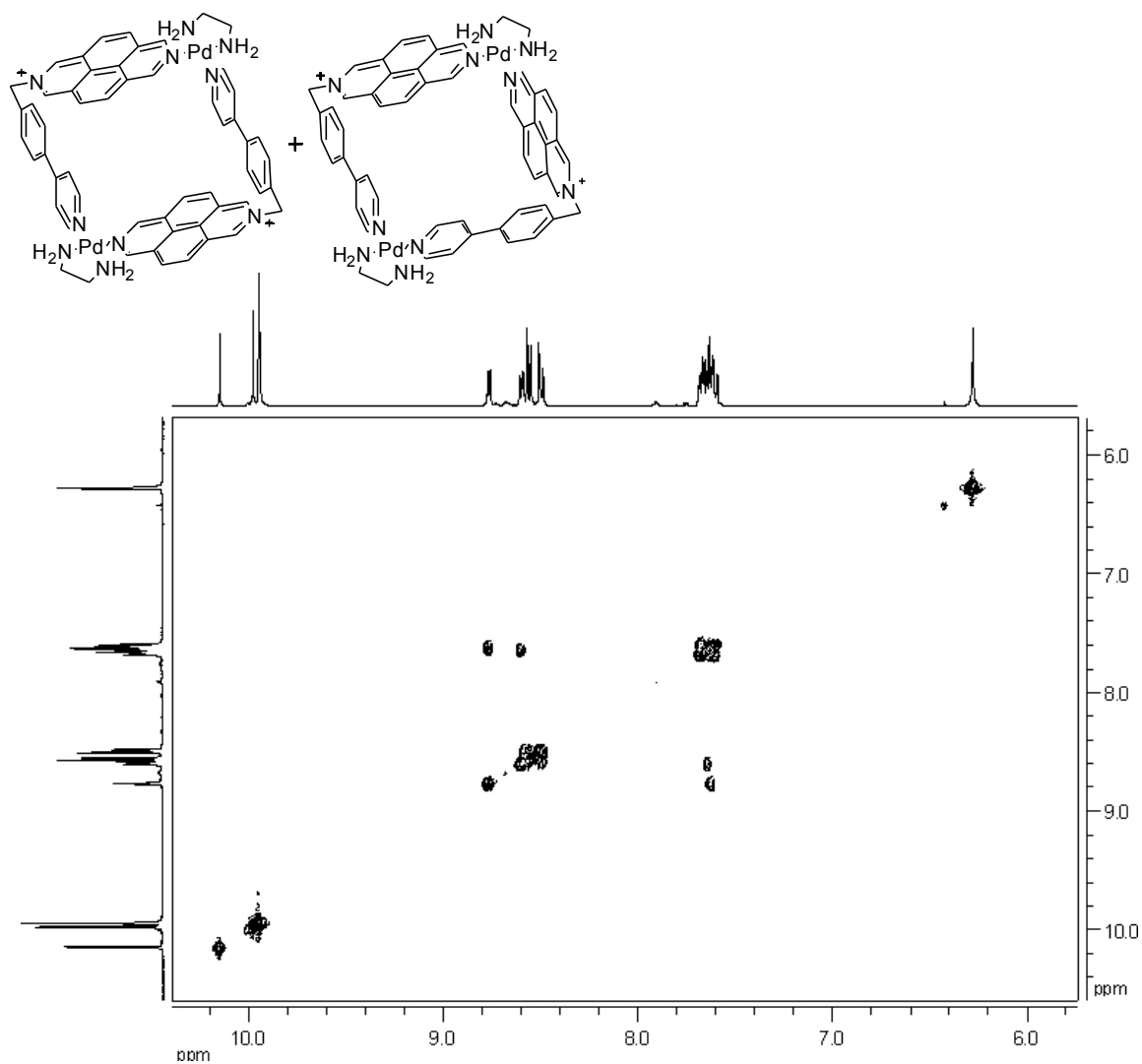


Figure S16: COSY (D₂O, 500 MHz) spectrum of **6a,b**·6NO₃.

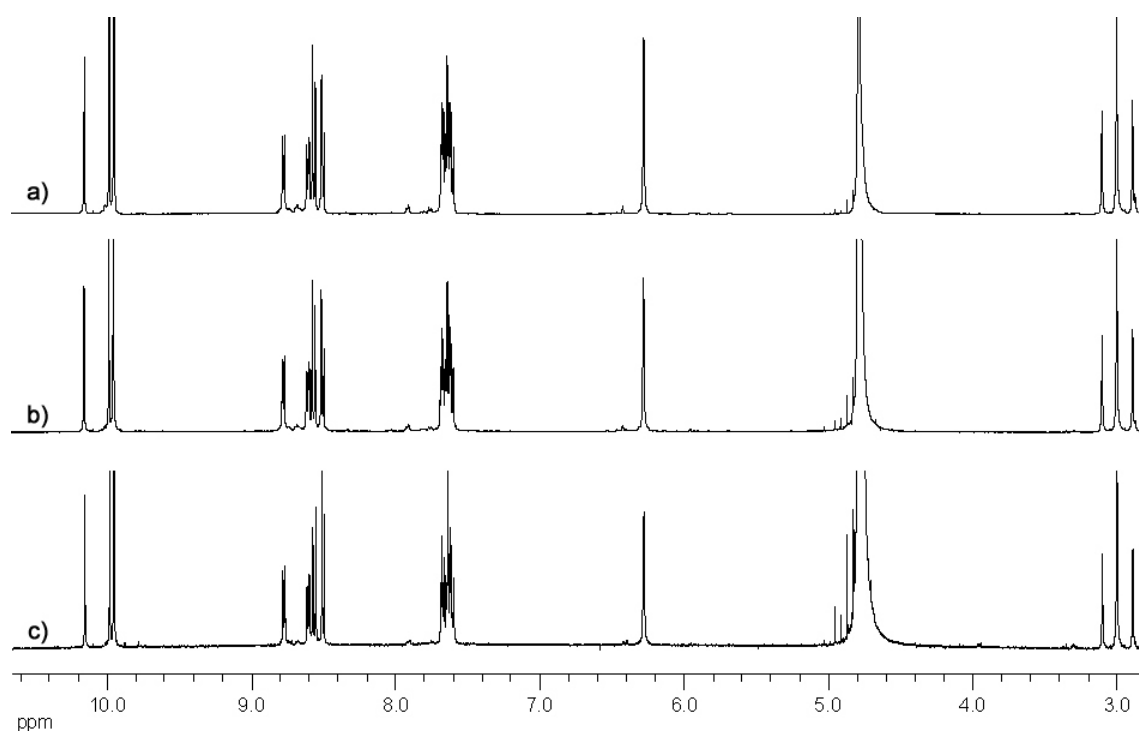


Figure S17: ^1H NMR (D_2O , 300 MHz) spectra of **6a,b**· 6NO_3 at different concentrations: a) 2.5 mM. b) 1 mM. c) 0.25 mM.

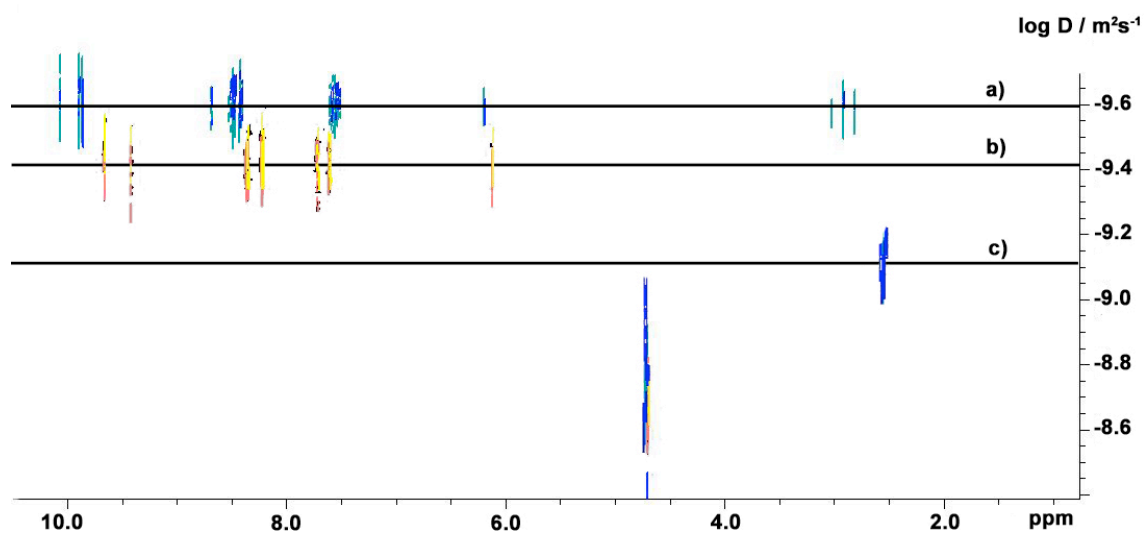


Figure S18: Superposed DOSY (D_2O , 500 MHz, 298 K) experiments of: a) 10 mM solution of $4\cdot\text{NO}_3$ and **5a**. b) 10 mM solution of $4\cdot\text{NO}_3$. c) **5a**.

Metalocycles **6a,b·6PF₆**

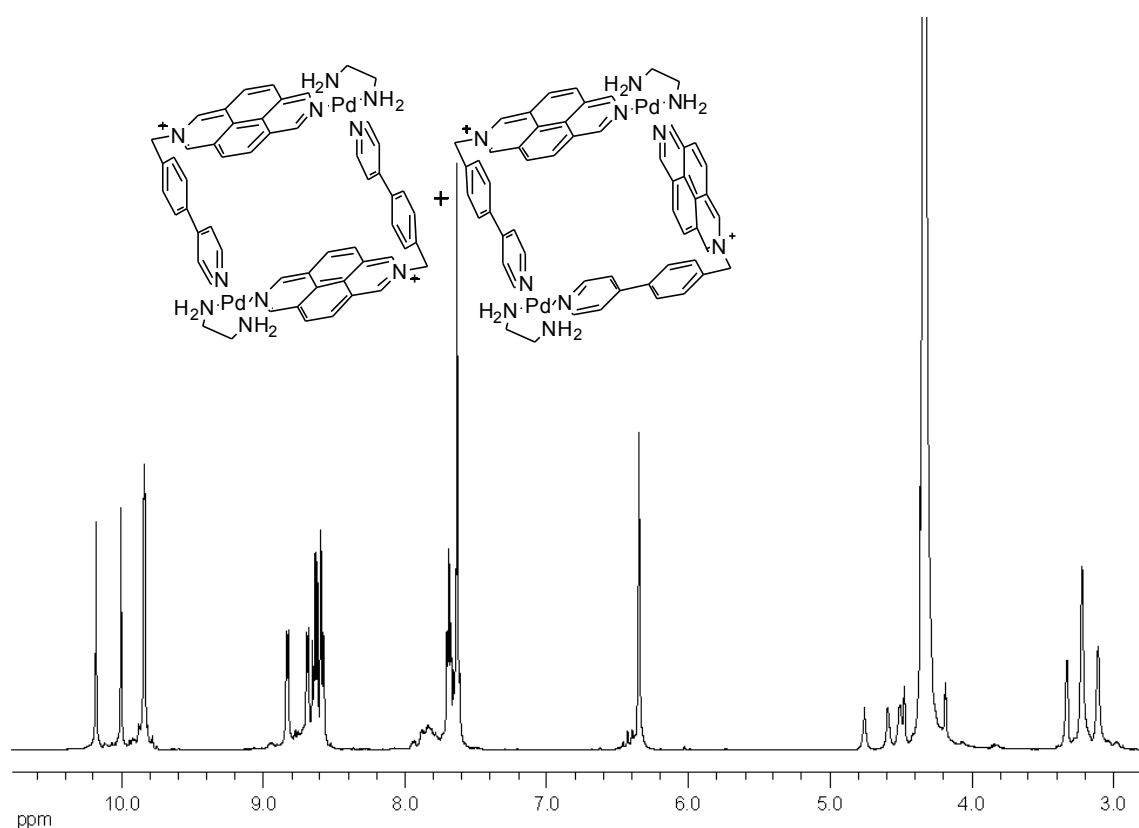


Figure S19: ¹H NMR (CD₃NO₂, 500 MHz) spectrum of **6a,b**·6PF₆

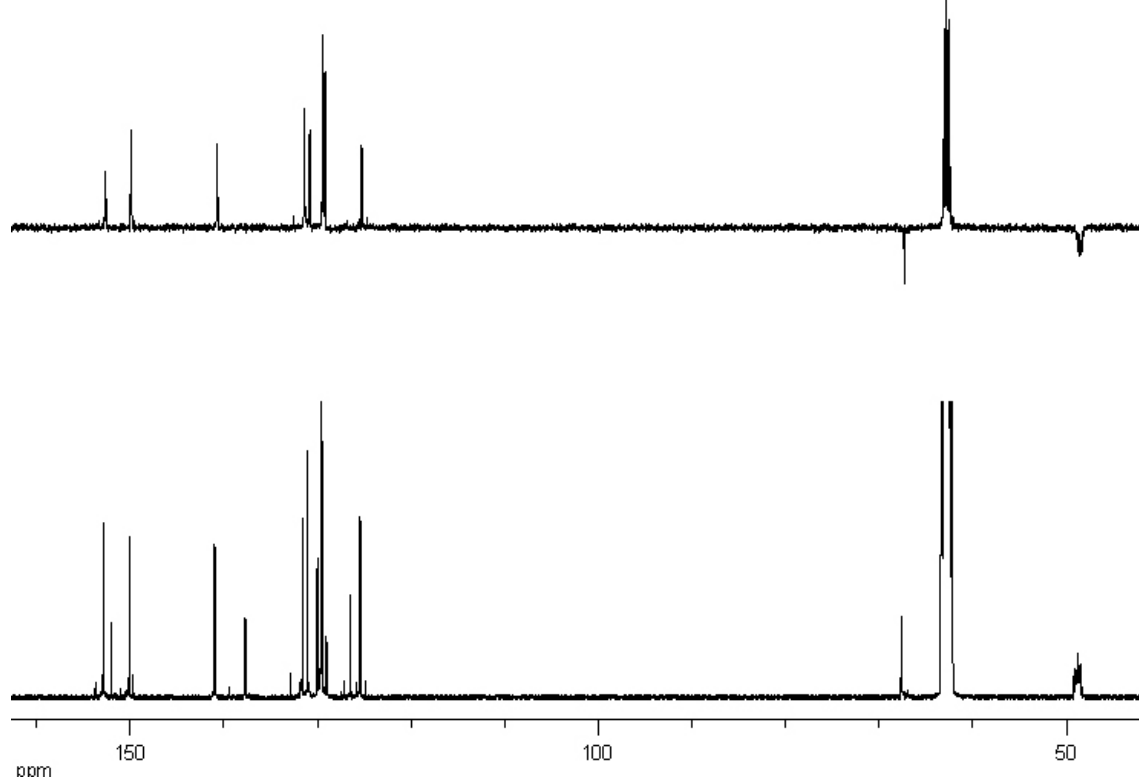


Figure S20: ¹³C NMR (CD₃NO₂, 125 MHz) spectrum of **6a,b**·6PF₆.

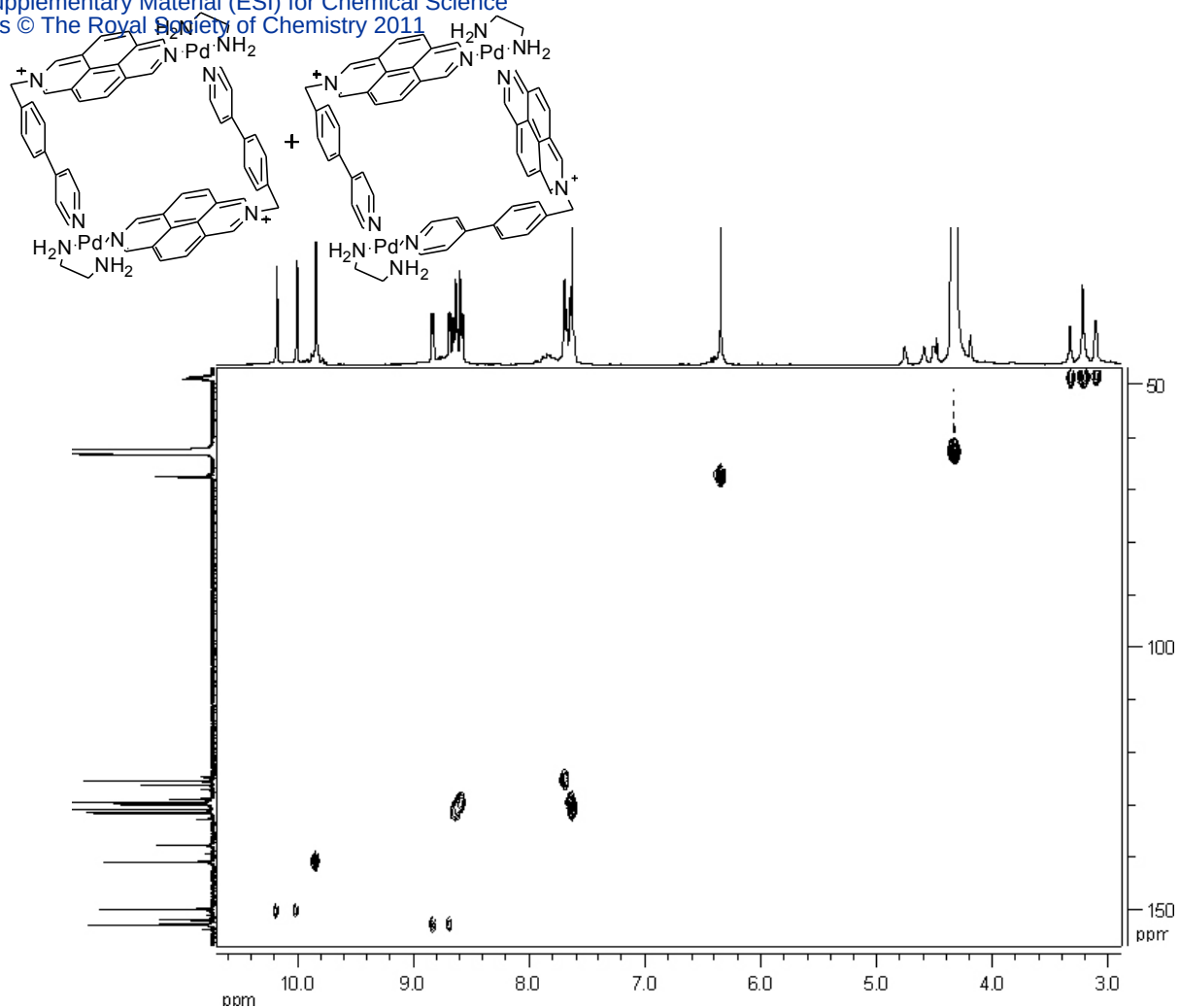


Figure S21: HSQC (CD_3NO_2 , 500 and 125 MHz) spectrum of **6a,b**· 6PF_6

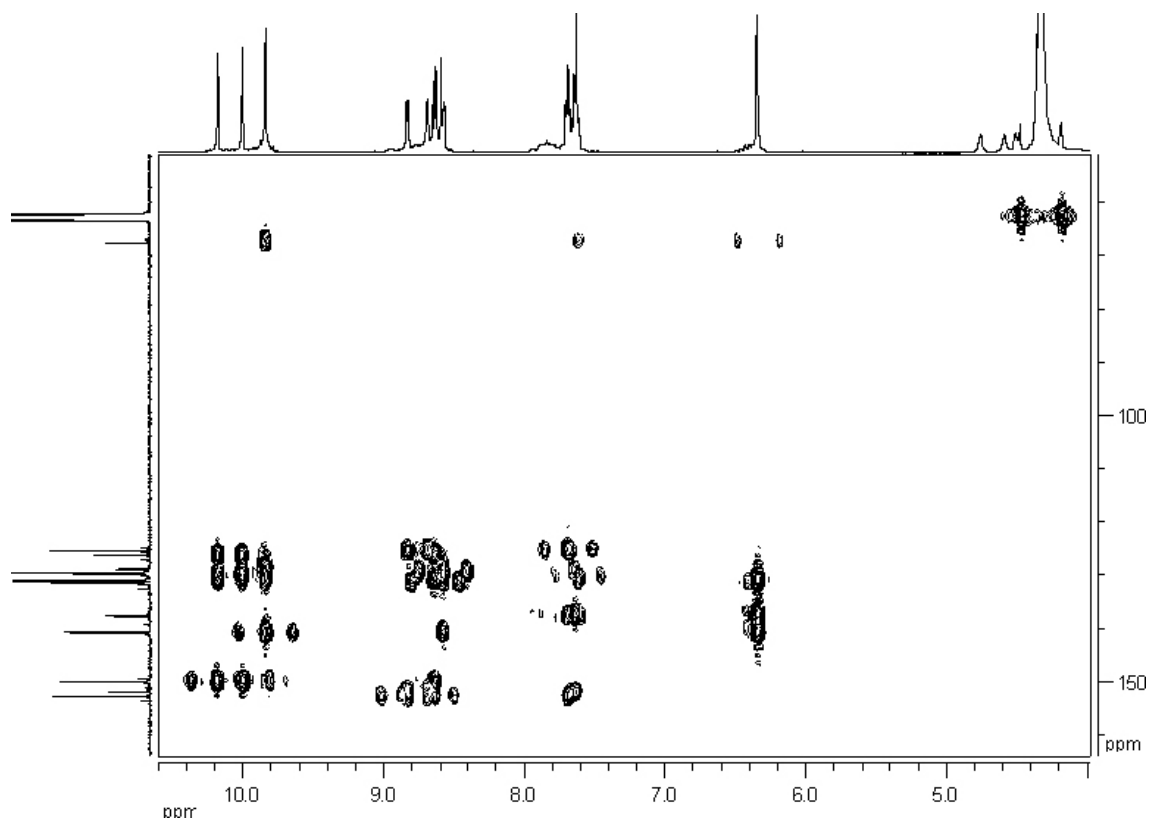


Figure 22: HMBC (CD_3NO_2 , 500 and 125 MHz) spectrum of **6a,b**· 6PF_6

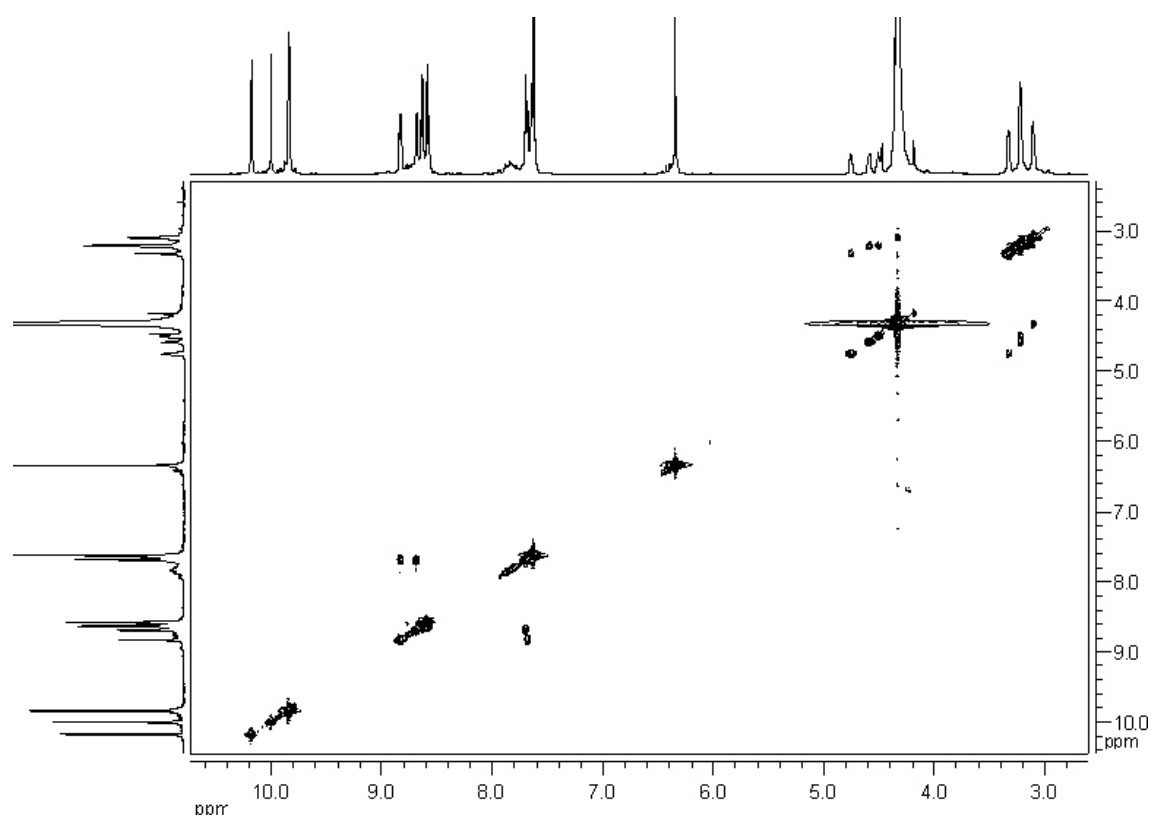


Figure S23: COSY (CD_3NO_2 , 500 MHz) spectrum of **6a,b**· 6PF_6 .

Metalocycles 7a,b·6PF₆

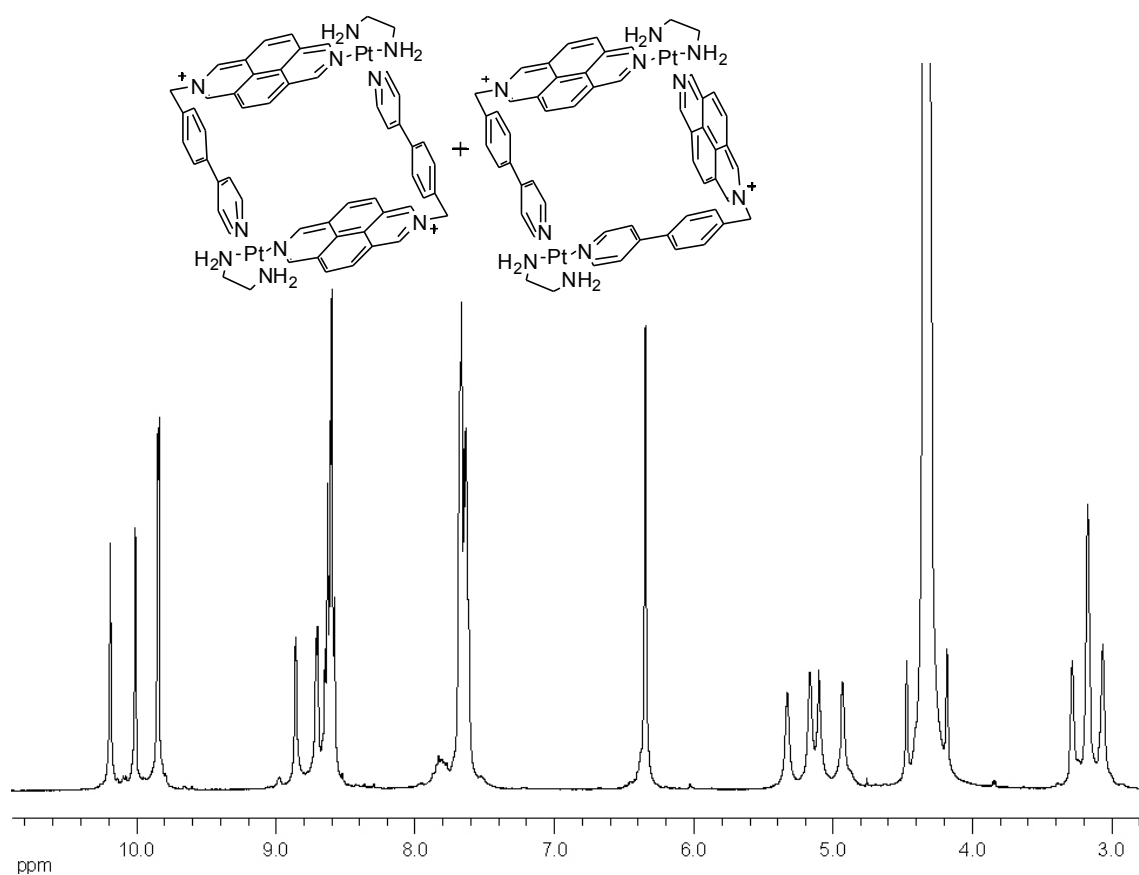


Figure S24: ¹H NMR (CD₃NO₂, 500 MHz) spectrum of 7a,b·6PF₆.

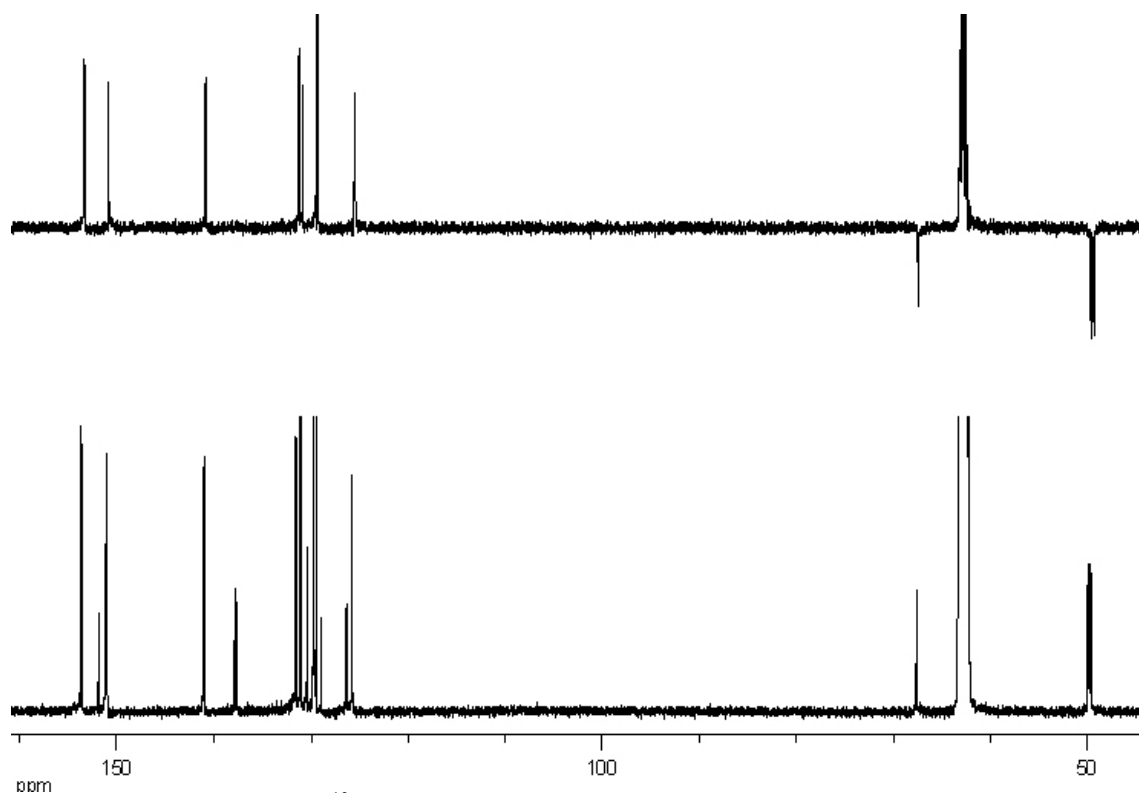


Figure S25: ¹³C NMR (CD₃NO₂, 125 MHz) spectrum of 7a,b·6PF₆.

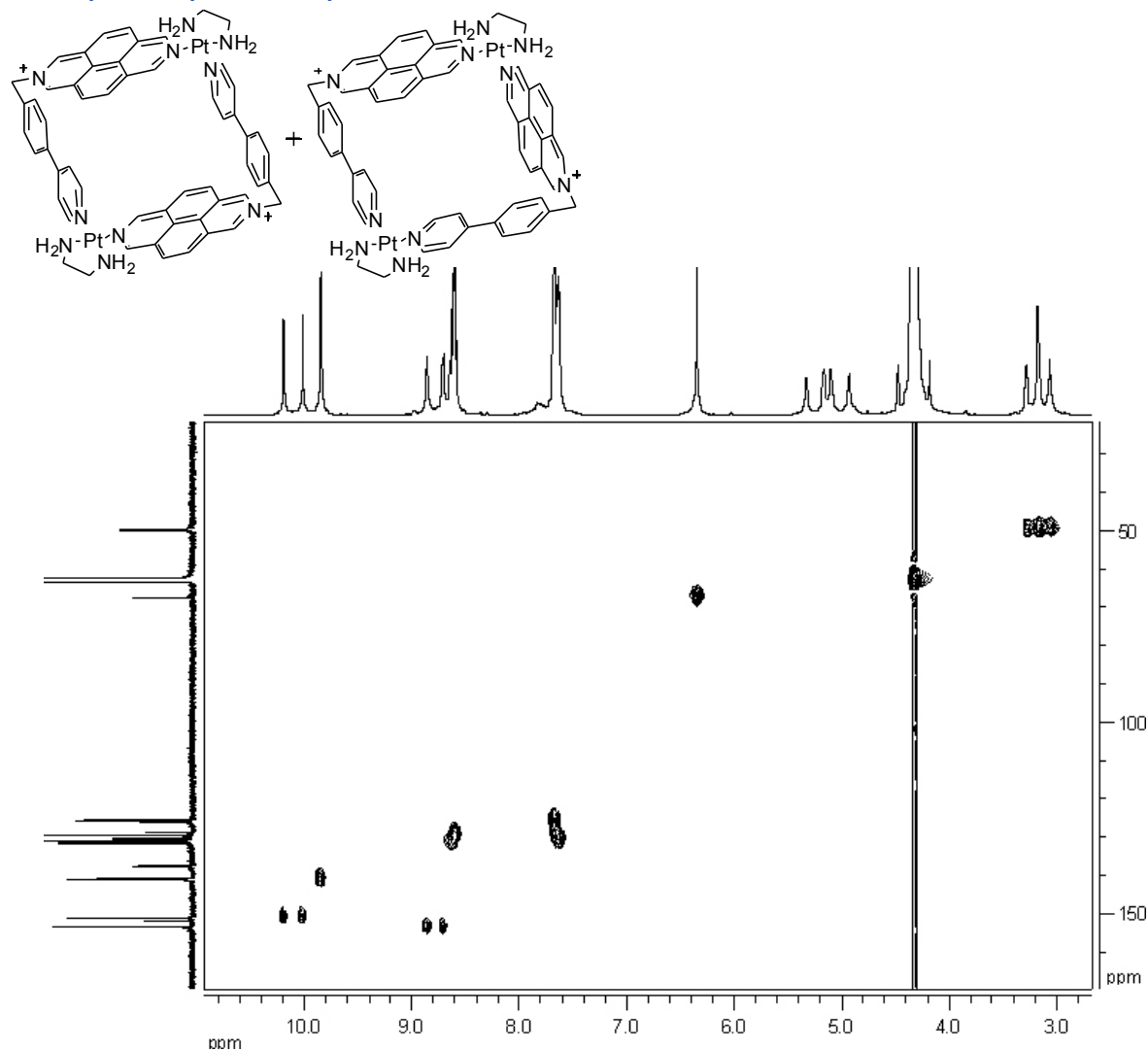


Figure S26: HSQC (CD₃NO₂, 500 and 125 MHz) spectrum of **7a,b**·6PF₆.

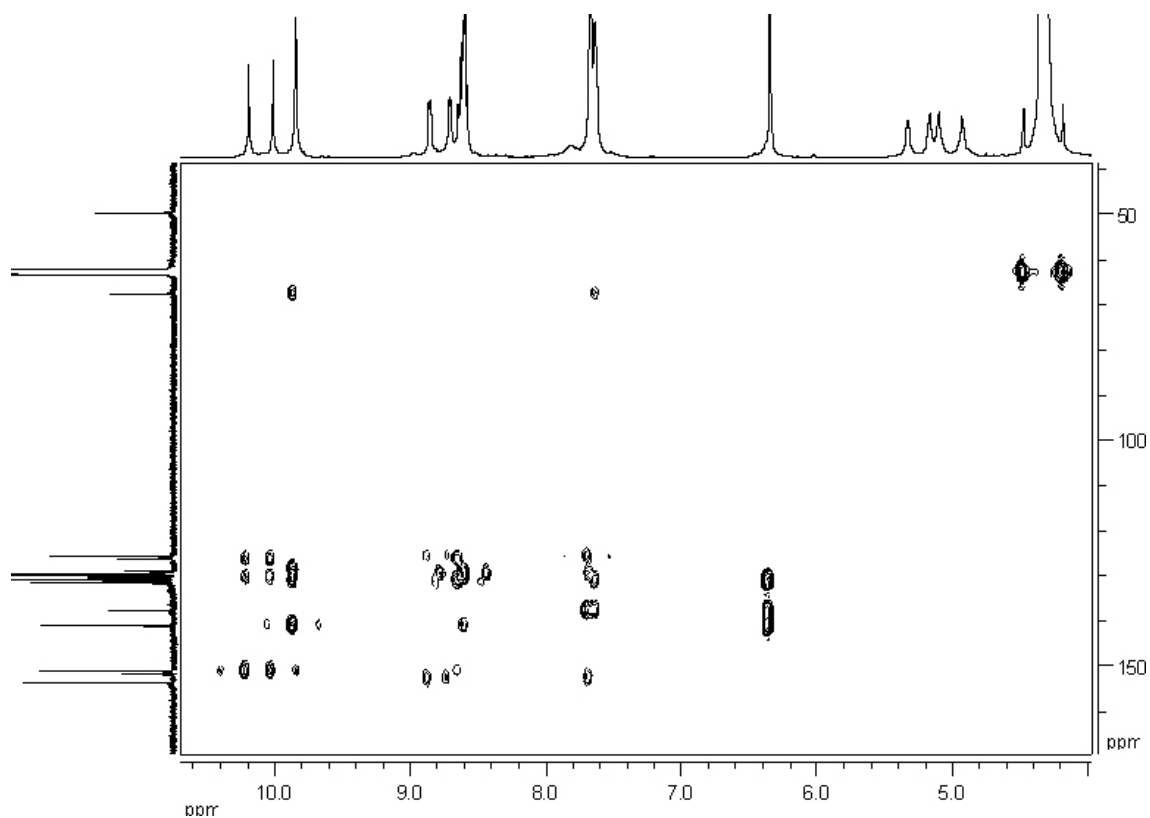


Figure S27: HMBC (CD₃NO₂, 500 and 125 MHz) spectrum of **7a,b**·6PF₆.

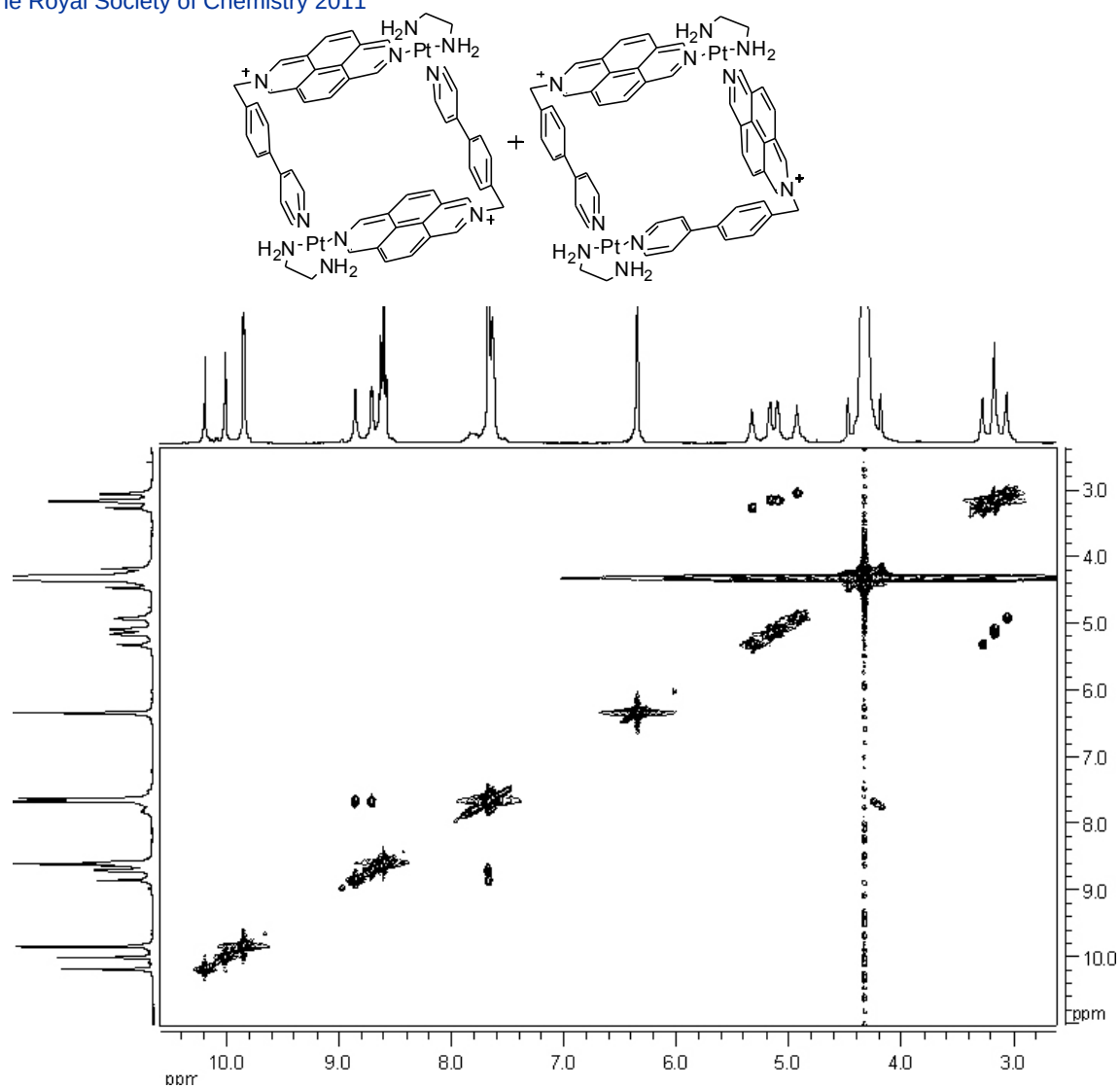


Figure S28: COSY (CD₃NO₂, 500 MHz) spectrum of **7a,b**·6PF₆

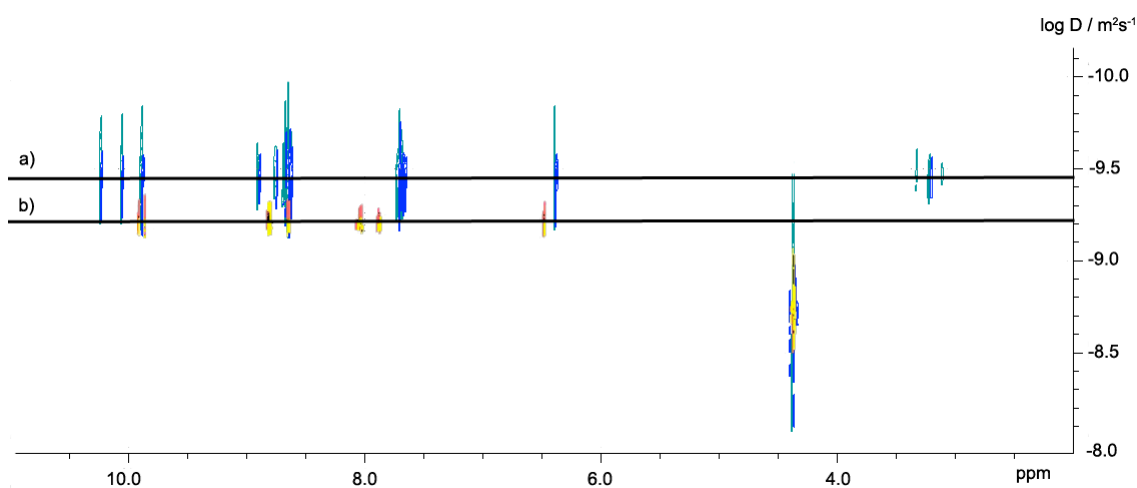


Figure S29: Superposed DOSY (CD₂NO₂, 500 MHz, 298 K) experiments of: a) metallobis(quinoline) complexes **7a,b**·6PF₆ (blue). b) ligand **4**·PF₆ (yellow).

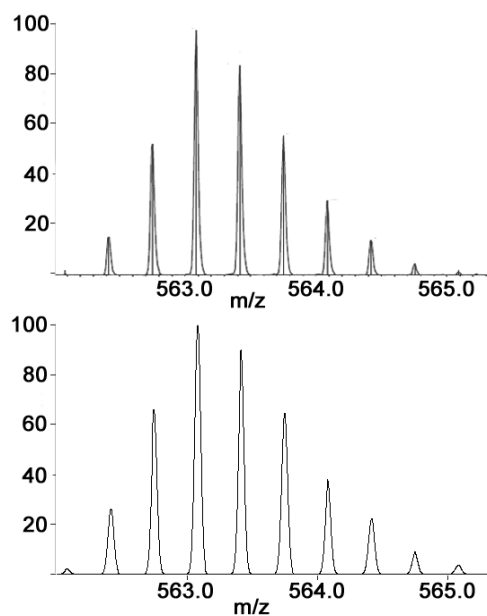


Figure S30: Observed (top) and theoretical (bottom) isotopic distribution for fragment $[7\mathbf{a},\mathbf{b}-3\text{PF}_6]^{3+}$ (exp. $m/z = 563.0843$, theoretical $m/z = 563.0860$).

Metalocycles 7a,b·6NO₃

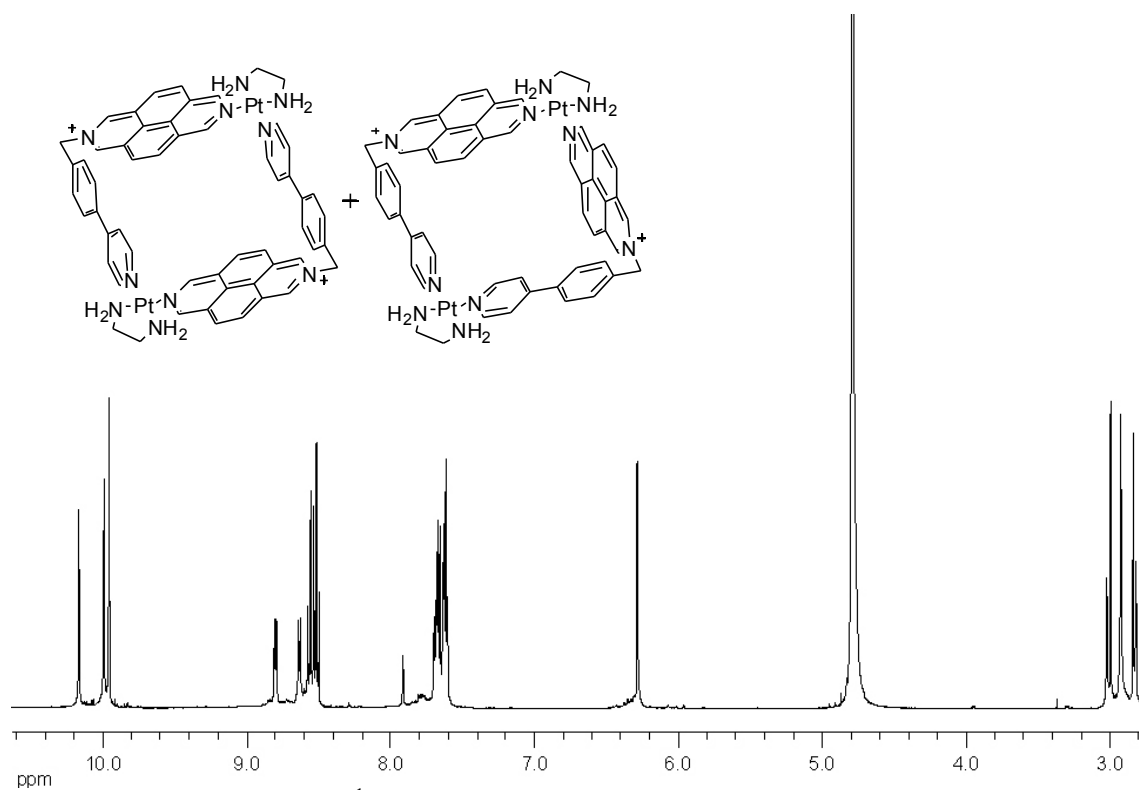


Figure S31: ¹H NMR (D₂O, 500 MHz) spectrum of **7a,b**·6NO₃.

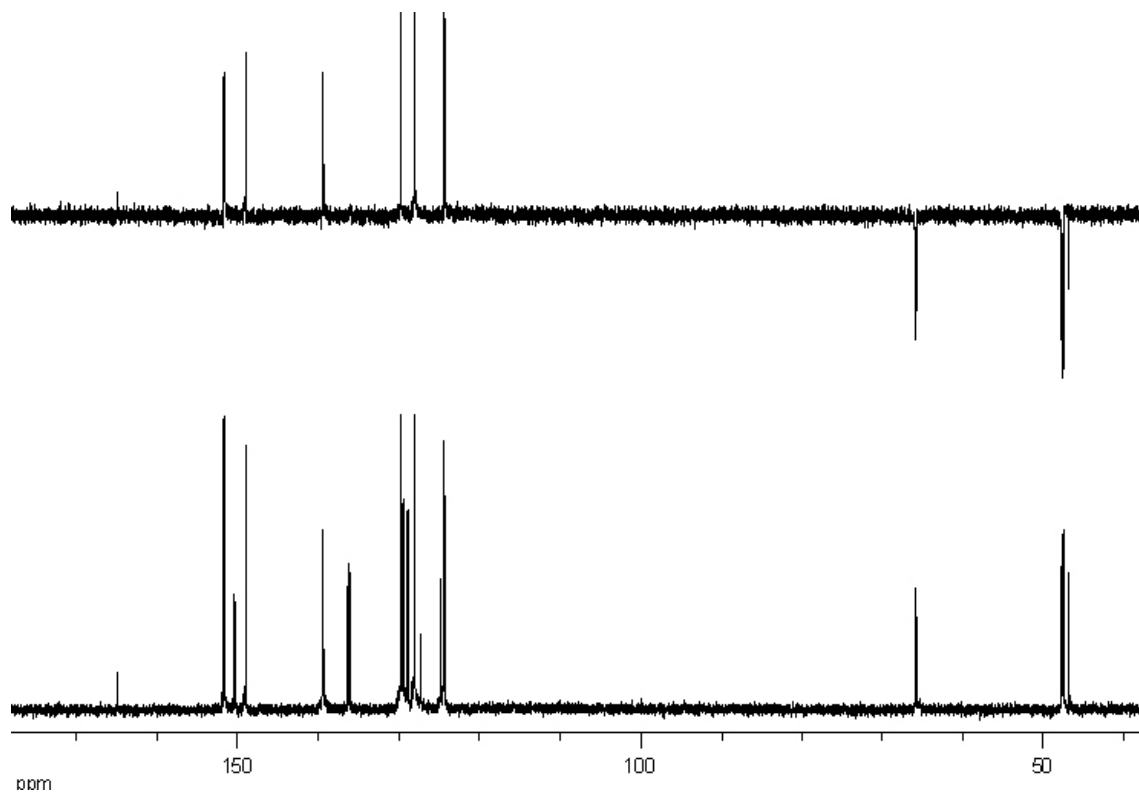


Figure S32: ¹³C NMR (D₂O, 125 MHz) spectrum of **7a,b**·6NO₃.

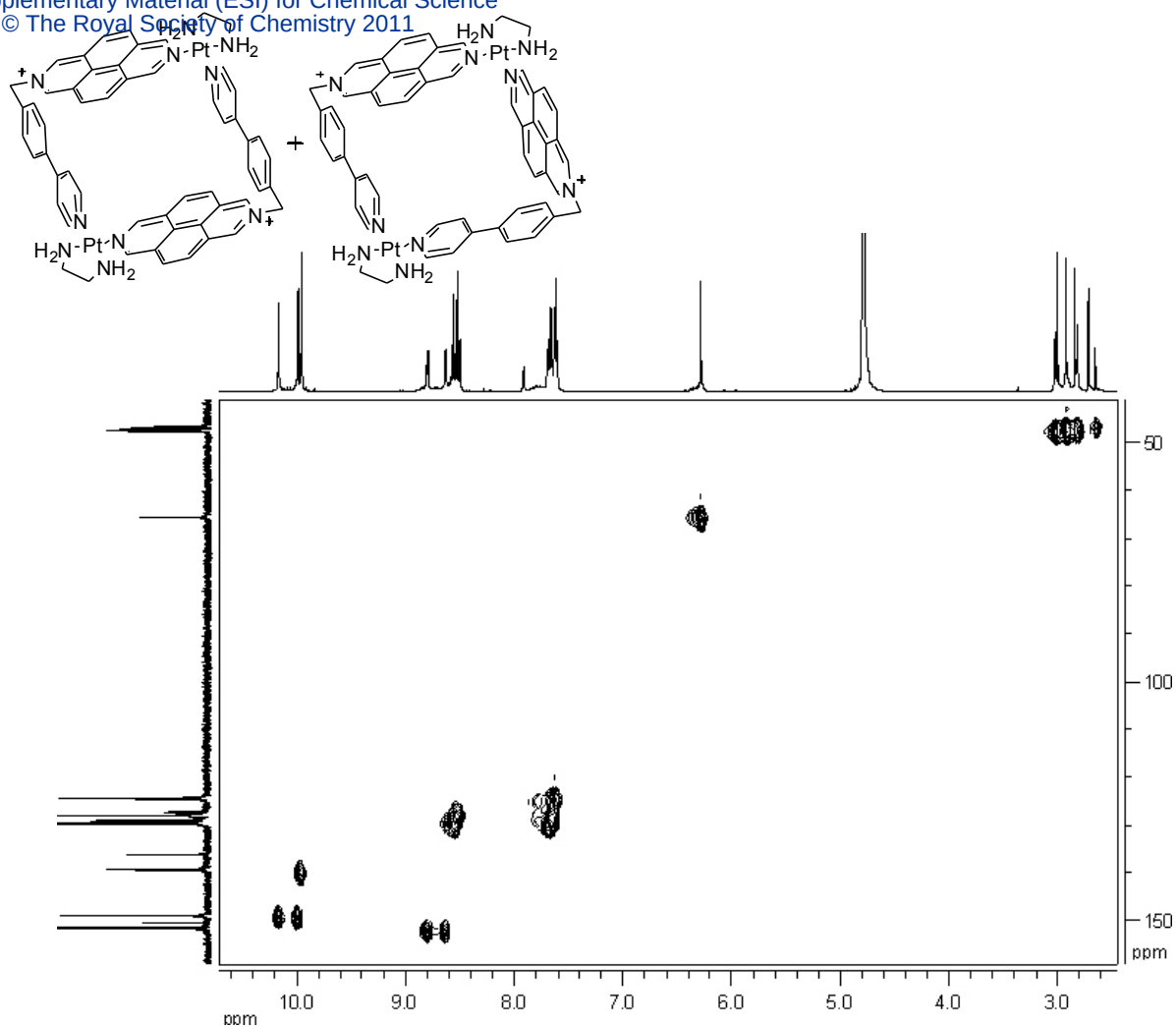


Figure S33: HSQC (D₂O, 500 and 125 MHz) spectrum of **7a,b**·6NO₃.

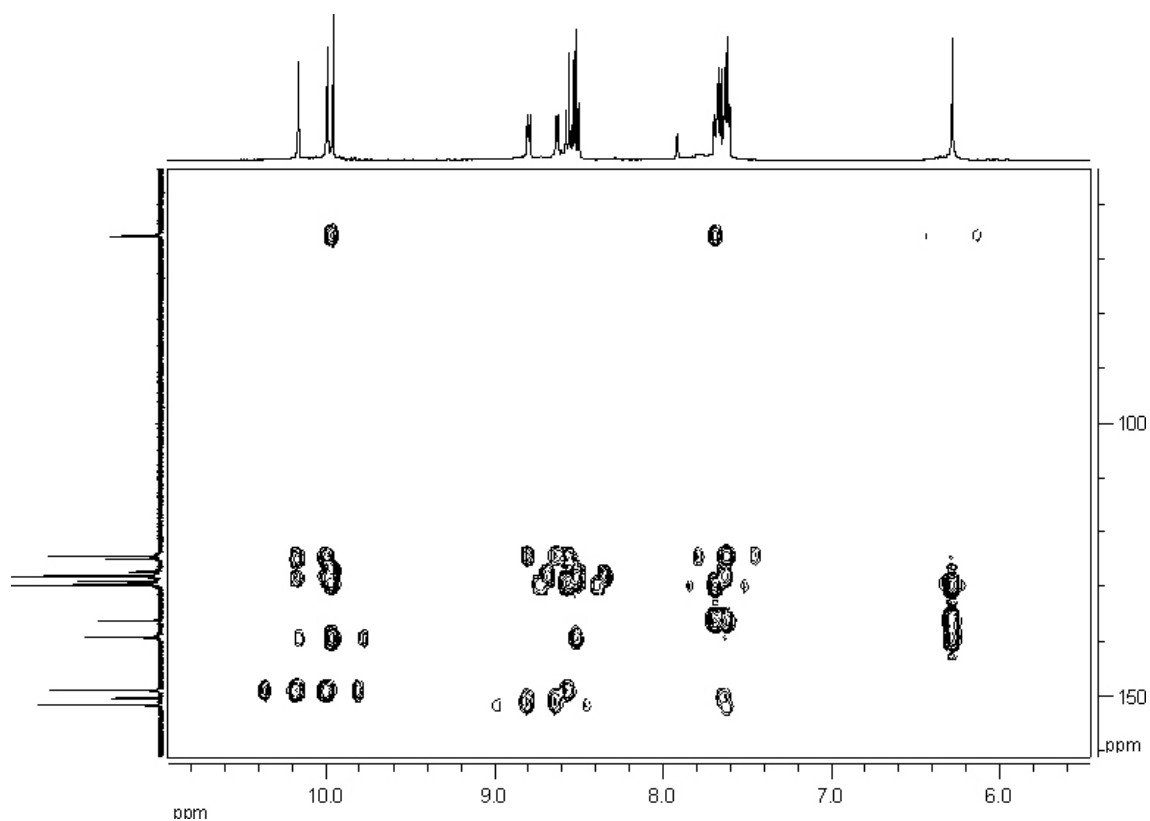


Figure S34: HMBC (D₂O, 500 and 125 MHz) spectrum of **7a,b**·6NO₃.

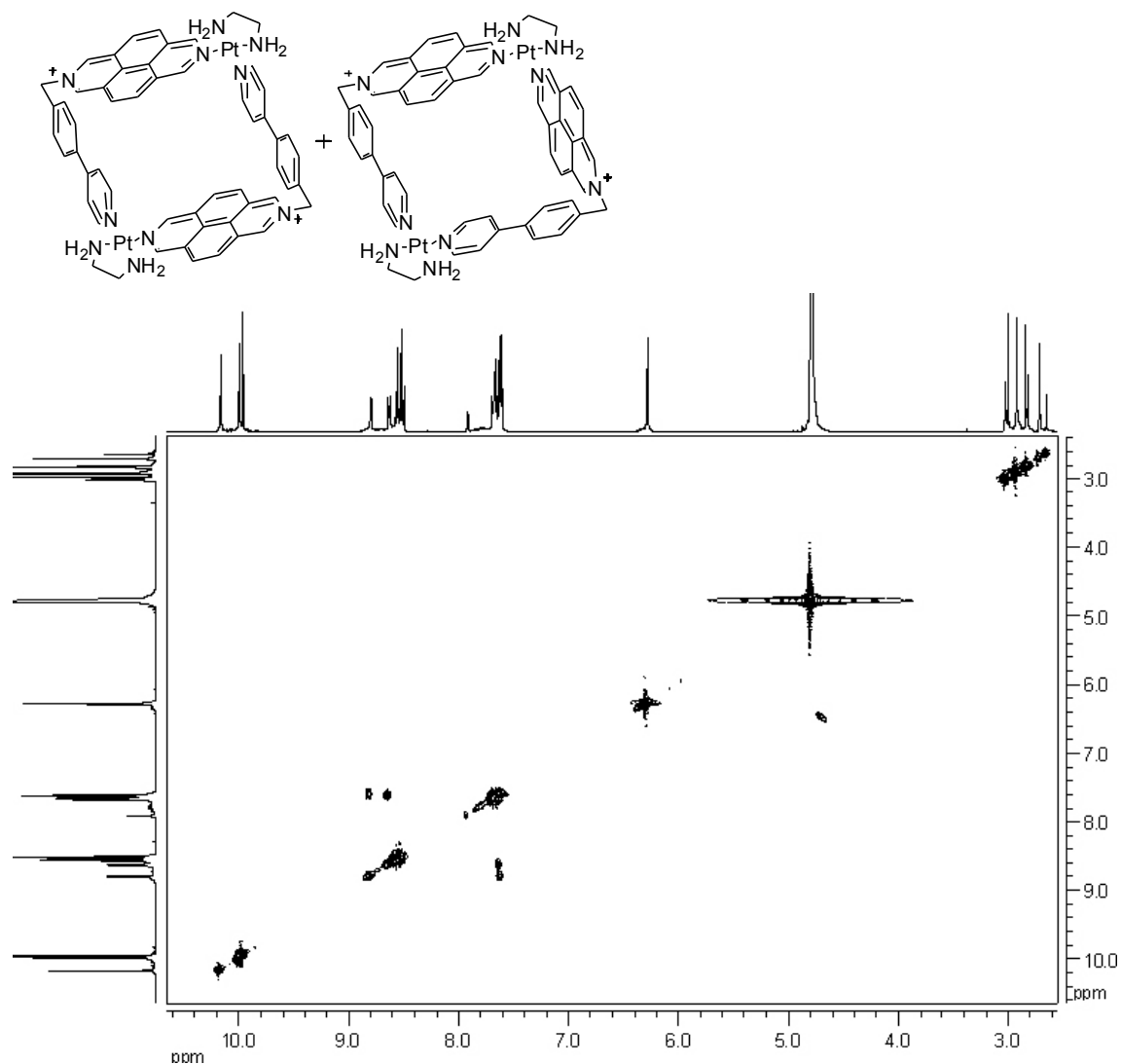


Figure S35: COSY (D₂O, 500 MHz) spectrum of **7a,b**·6NO₃.

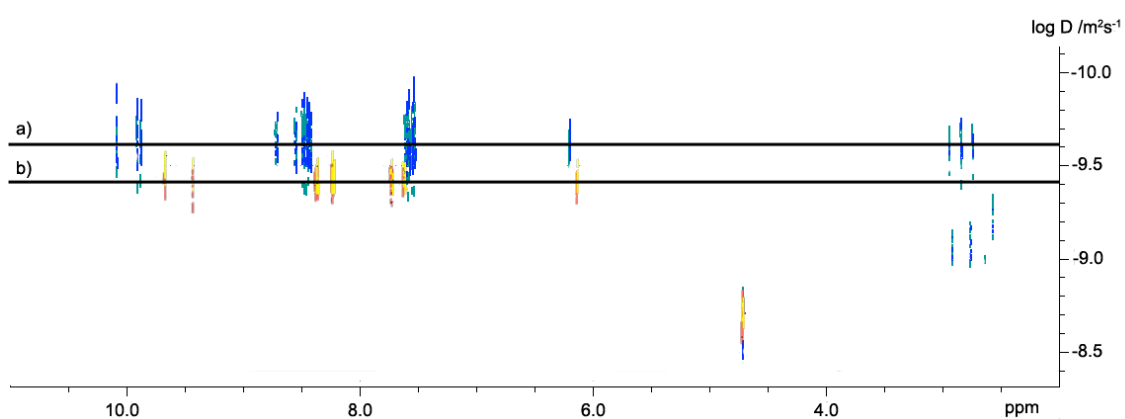


Figure S36: Superposed DOSY (D₂O, 500 MHz, 298 K) experiments of: a) metallocycles **7a,b**·6NO₃ (blue). b) ligand **4**·NO₃ (yellow).

Pseudorotaxane 1·2PF₆·8

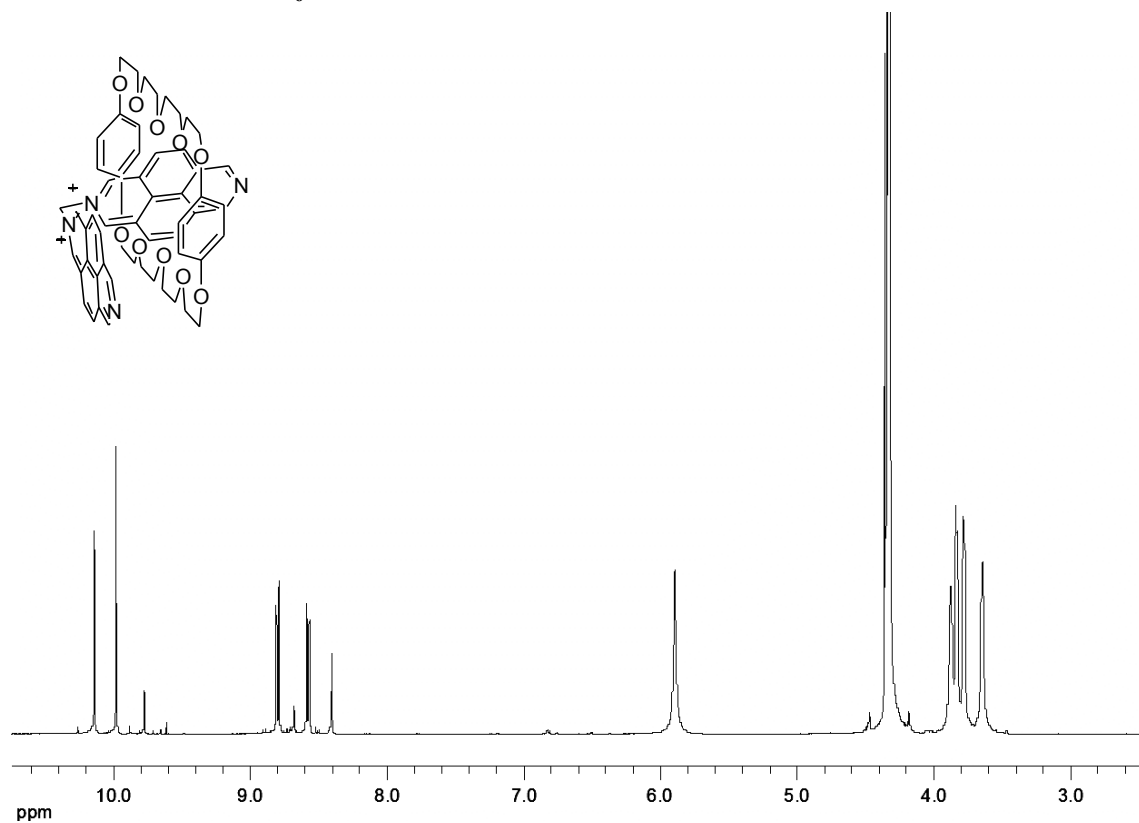


Figure S37: ¹H NMR (CD₃NO₂, 500 MHz) spectrum of 1·2PF₆·8.

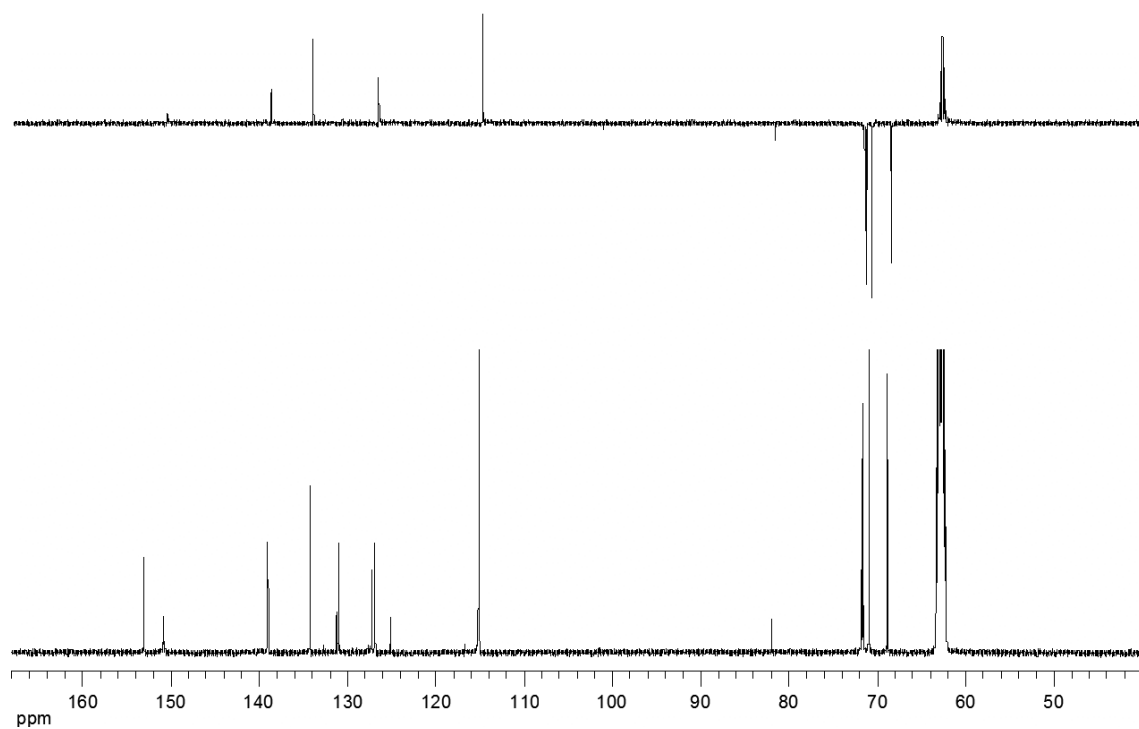


Figure S38: ¹³C NMR (CD₃NO₂, 500 MHz) spectrum of 1·2PF₆·8.

Pseudorotaxane 1·2PF₆·9

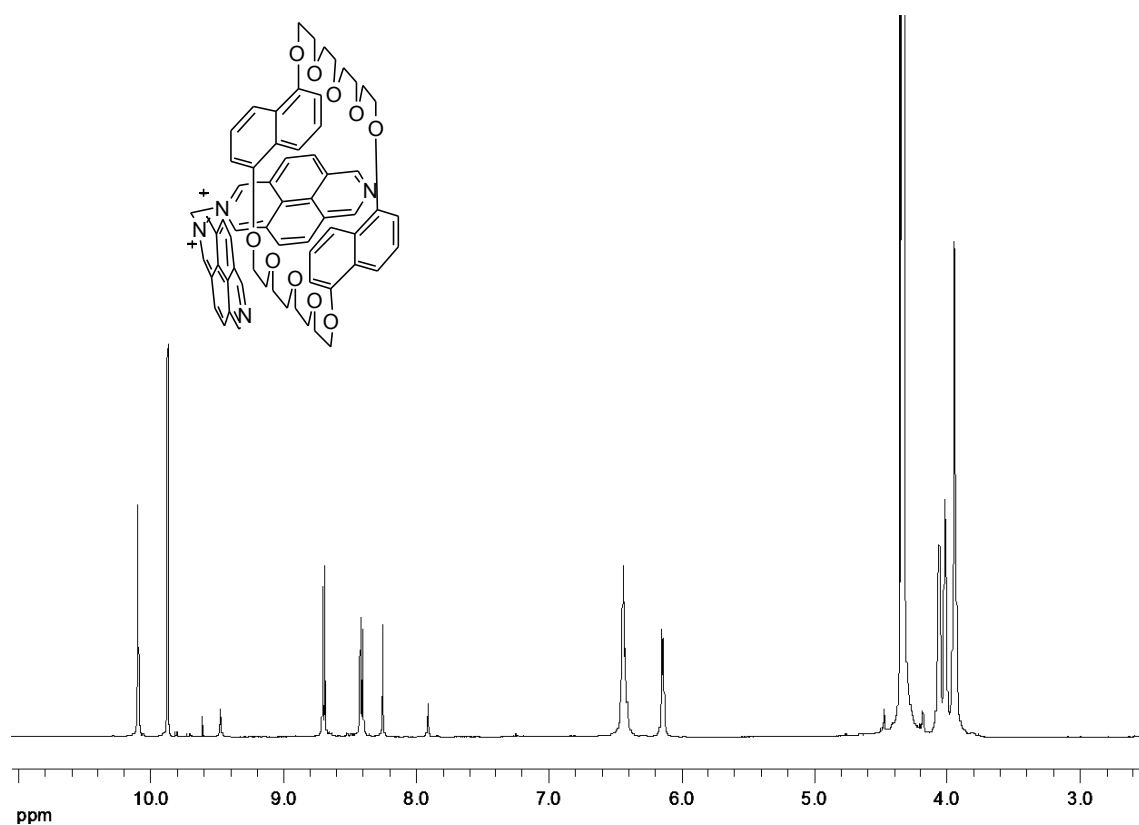


Figure S39: ¹H NMR (CD₃NO₂, 500 MHz) spectrum of 1·2PF₆·9.

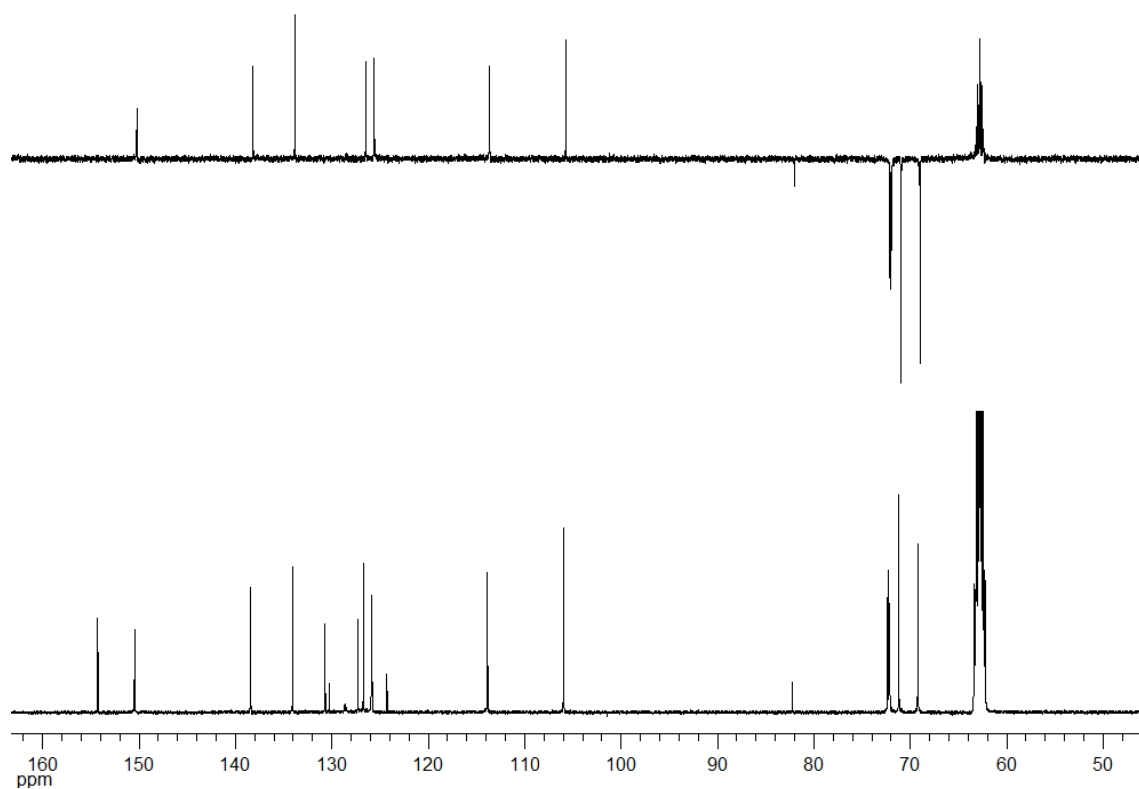


Figure S40: ¹³C NMR (CD₃NO₂, 500 MHz) spectrum of 1·2PF₆·9.

Catenane $3(\mathbf{8})_2 \cdot 4\text{OTf} \cdot 4\text{PF}_6$

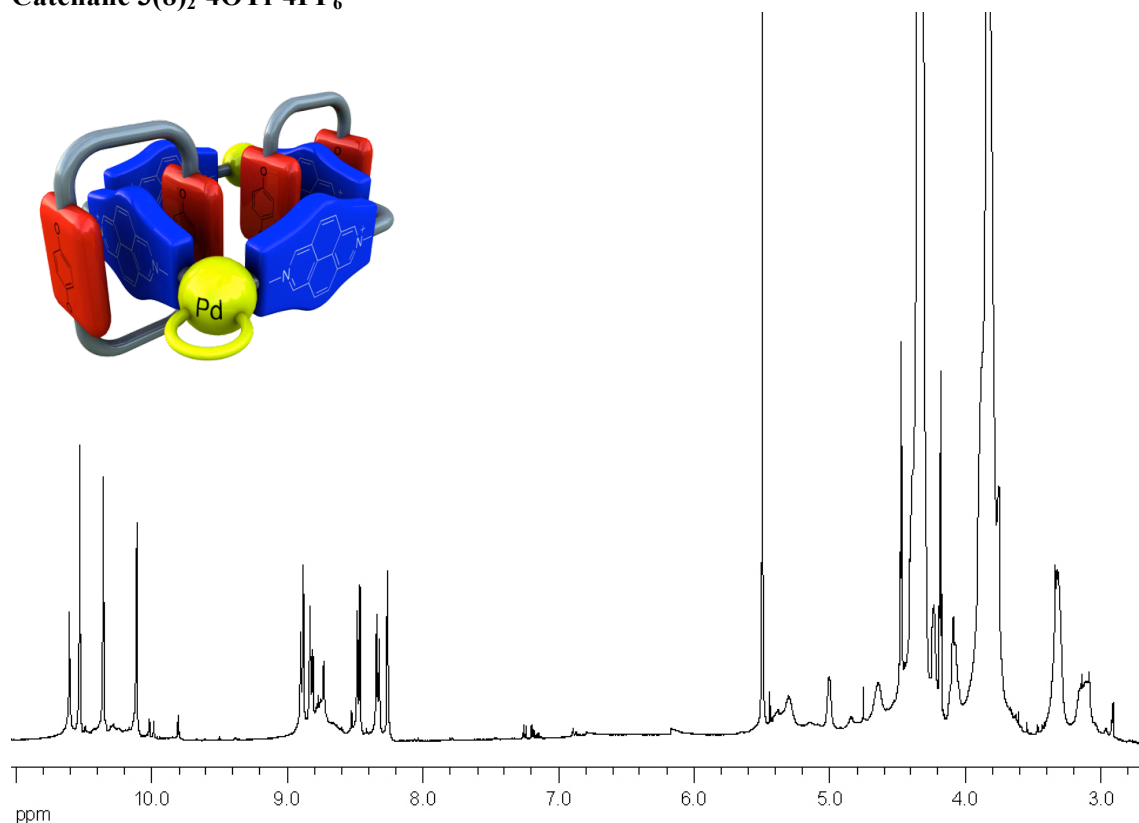


Figure S41: ^1H NMR (CD_3NO_2 , 500 MHz) spectrum of $3(\mathbf{8})_2 \cdot 4\text{OTf} \cdot 4\text{PF}_6$.

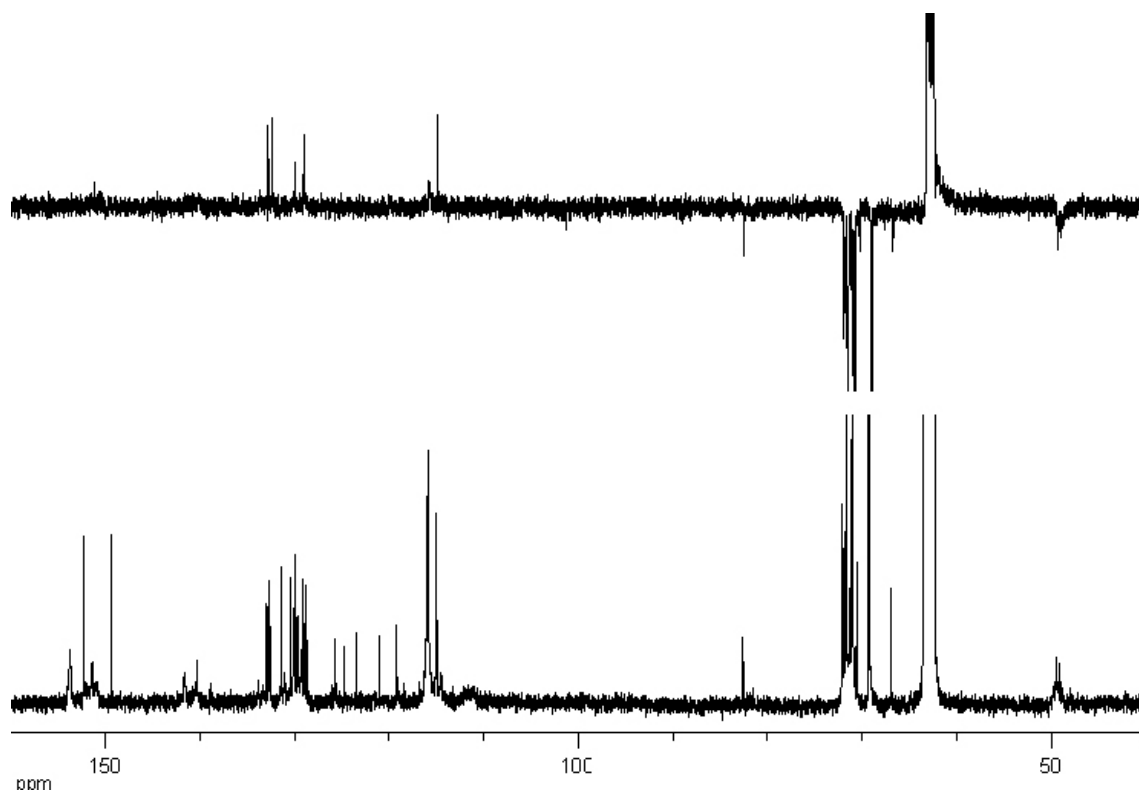


Figure S42: ^{13}C NMR (CD_3NO_2 , 125 MHz) spectrum of $3(\mathbf{8})_2 \cdot 4\text{OTf} \cdot 4\text{PF}_6$.

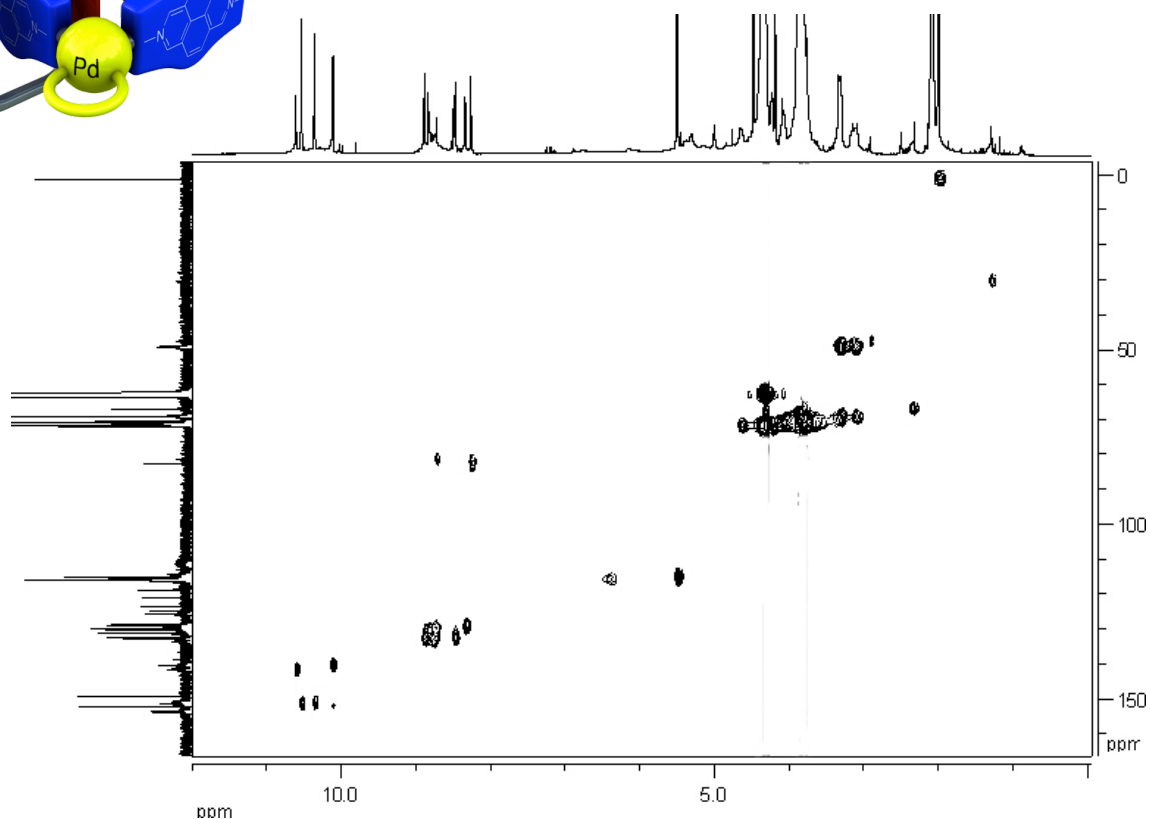
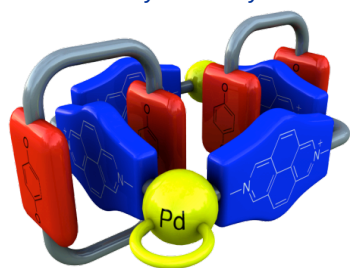


Figure S43: HSQC (CD_3NO_2 , 500 and 125 MHz) spectrum of $3(8)_2 \cdot 4\text{OTf} \cdot 4\text{PF}_6$.

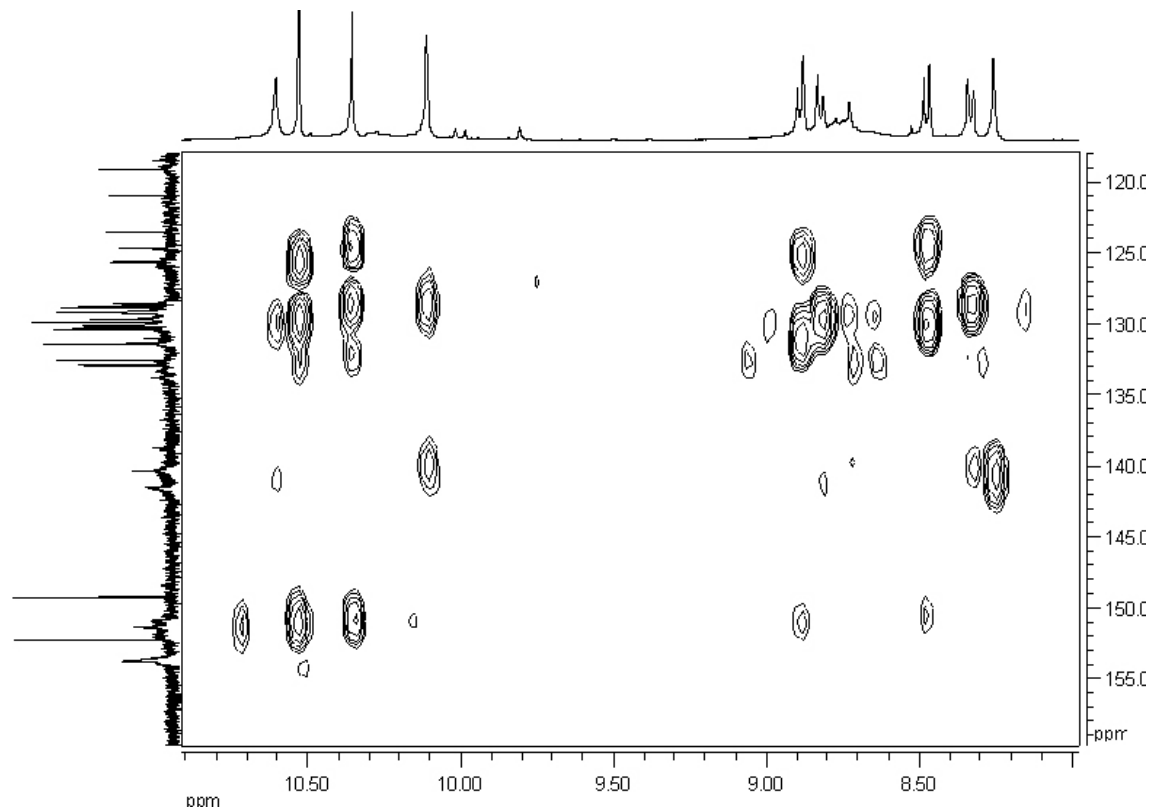


Figure S44: HMBC (CD_3NO_2 , 500 and 125 MHz) spectrum of $3(8)_2 \cdot 4\text{OTf} \cdot 4\text{PF}_6$.

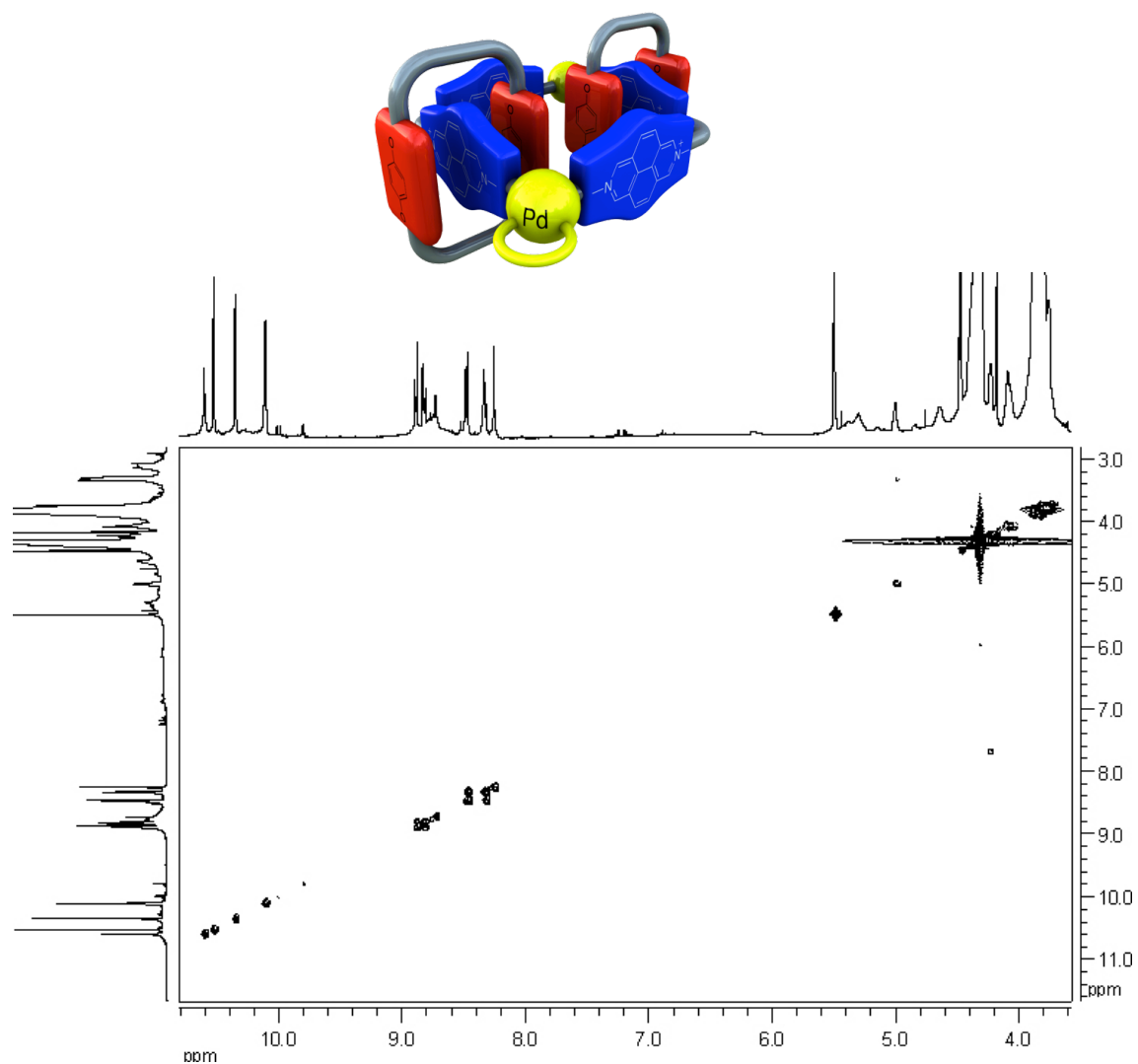


Figure S45: COSY (CD_3NO_2 , 500 MHz) spectrum of $3(8)_2 \cdot 4OTf \cdot 4PF_6$.

Catenane $3(9)_2 \cdot 4OTf \cdot 4PF_6$

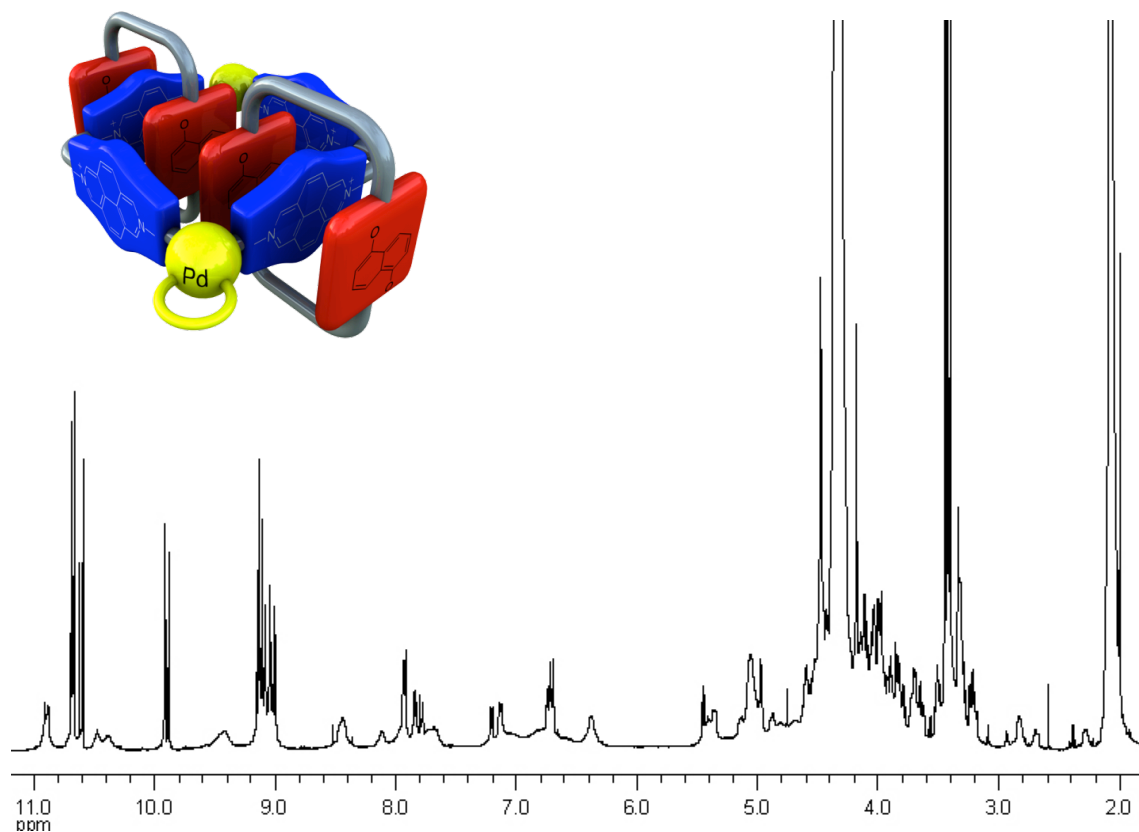


Figure S46: ^1H NMR (CD₃NO₂, 500 MHz, 298 K) spectrum of $3(9)_2 \cdot 4OTf \cdot 4PF_6$.

Catenane $6a(8)_2 \cdot 6PF_6$

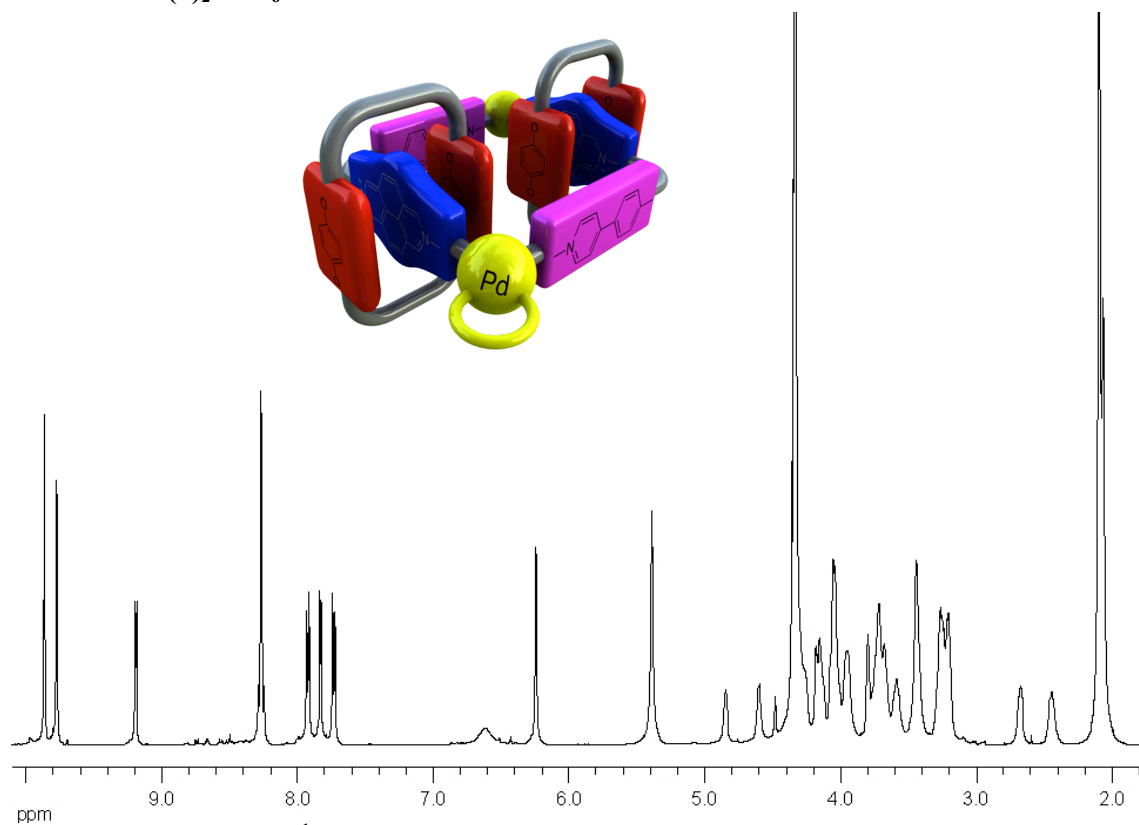


Figure S47: 1H NMR (CD₃NO₂, 500 MHz) spectrum of $6a(8)_2 \cdot 6PF_6$.

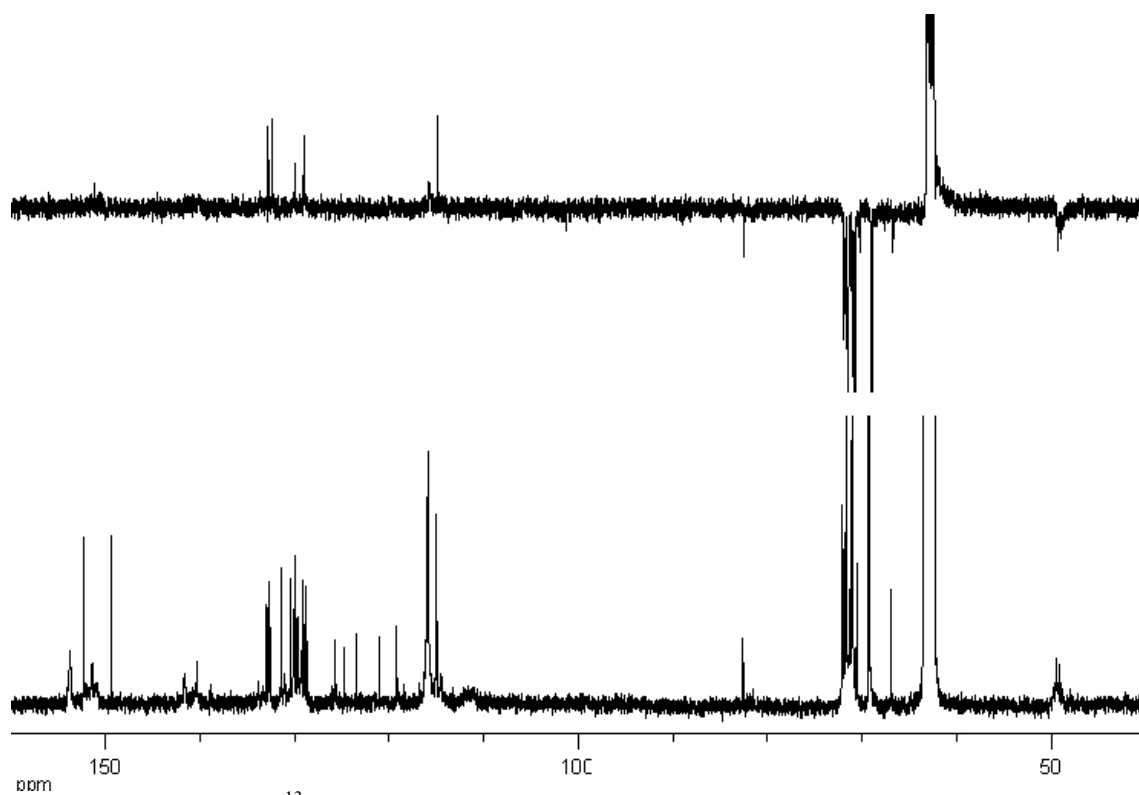


Figure S48: ^{13}C NMR (CD₃NO₂, 500 y 125 MHz) spectrum of $6a(8)_2 \cdot 6PF_6$.

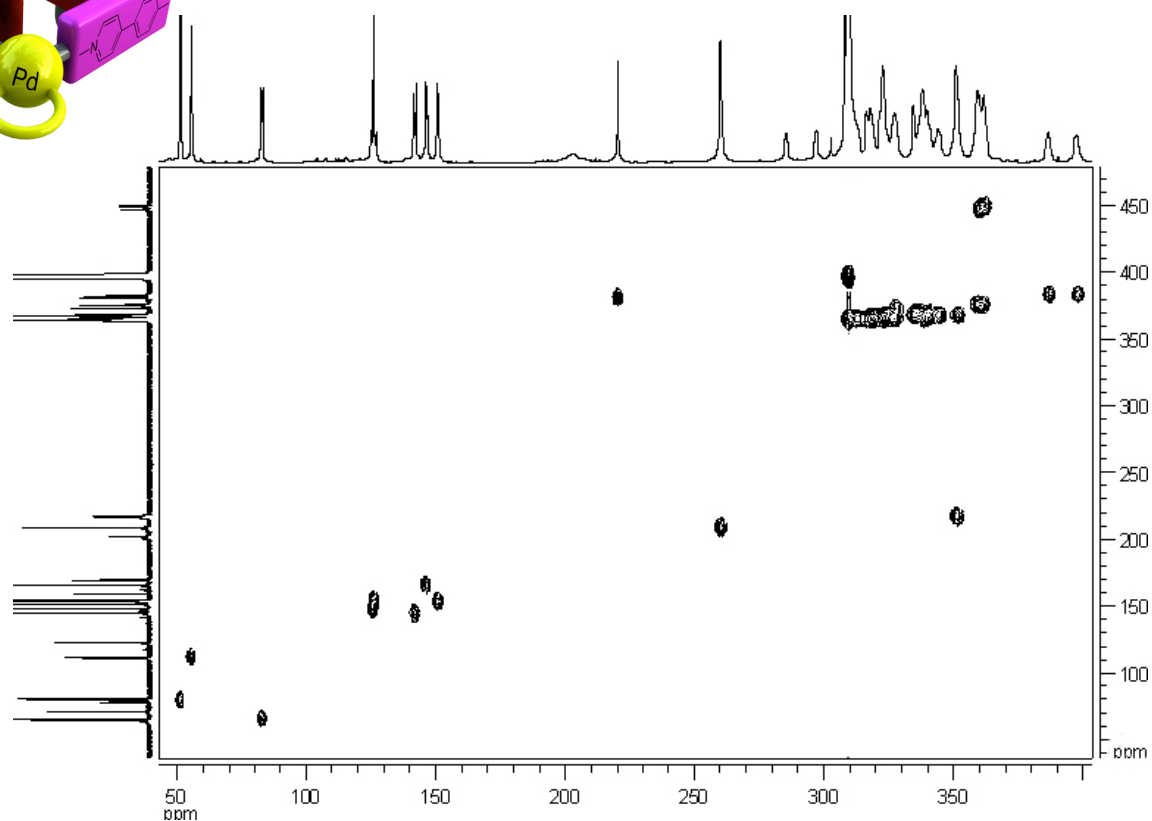
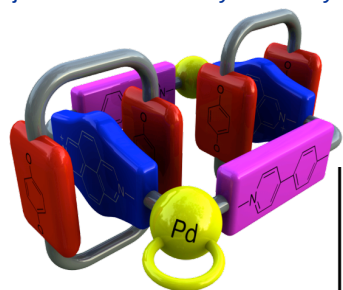


Figure S49: HSQC (CD_3NO_2 , 500 and 125 MHz) spectrum of **6a(8)₂·6PF₆**.

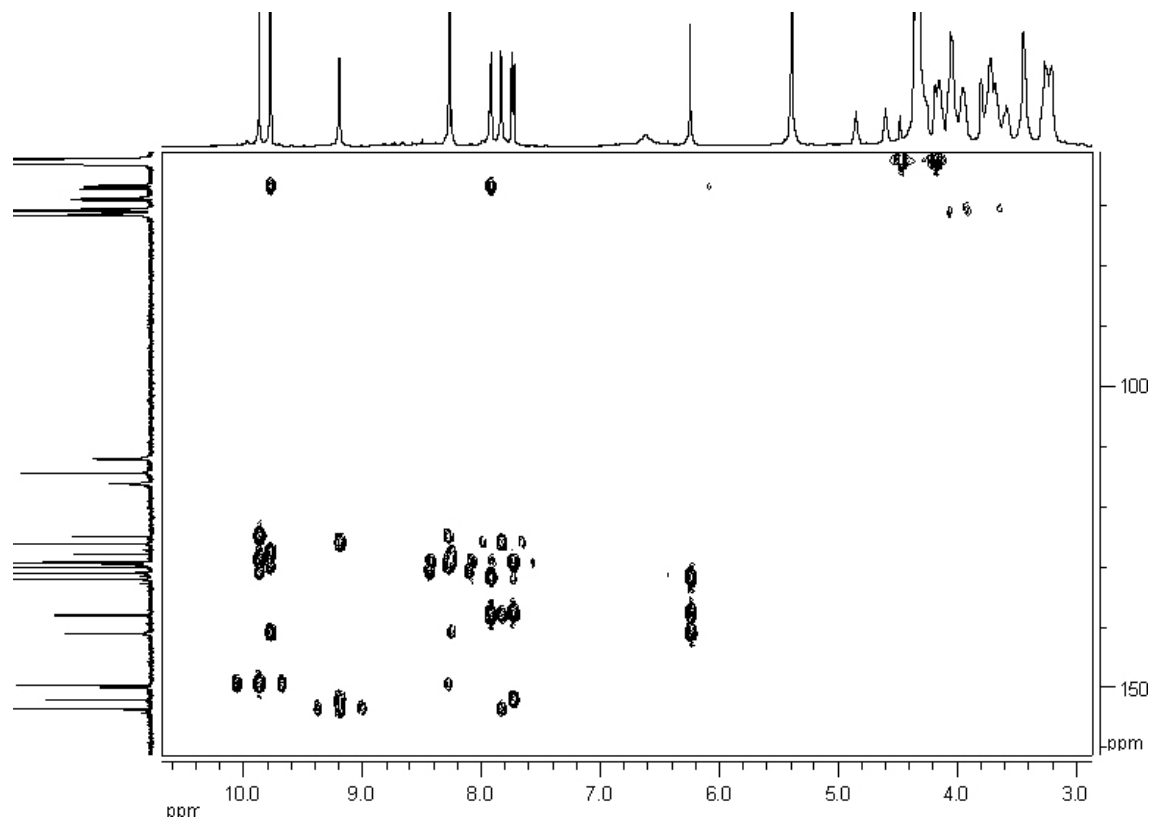


Figure S50: HMBC (CD_3NO_2 , 500 and 125 MHz) spectrum of **6a(8)₂·6PF₆**.

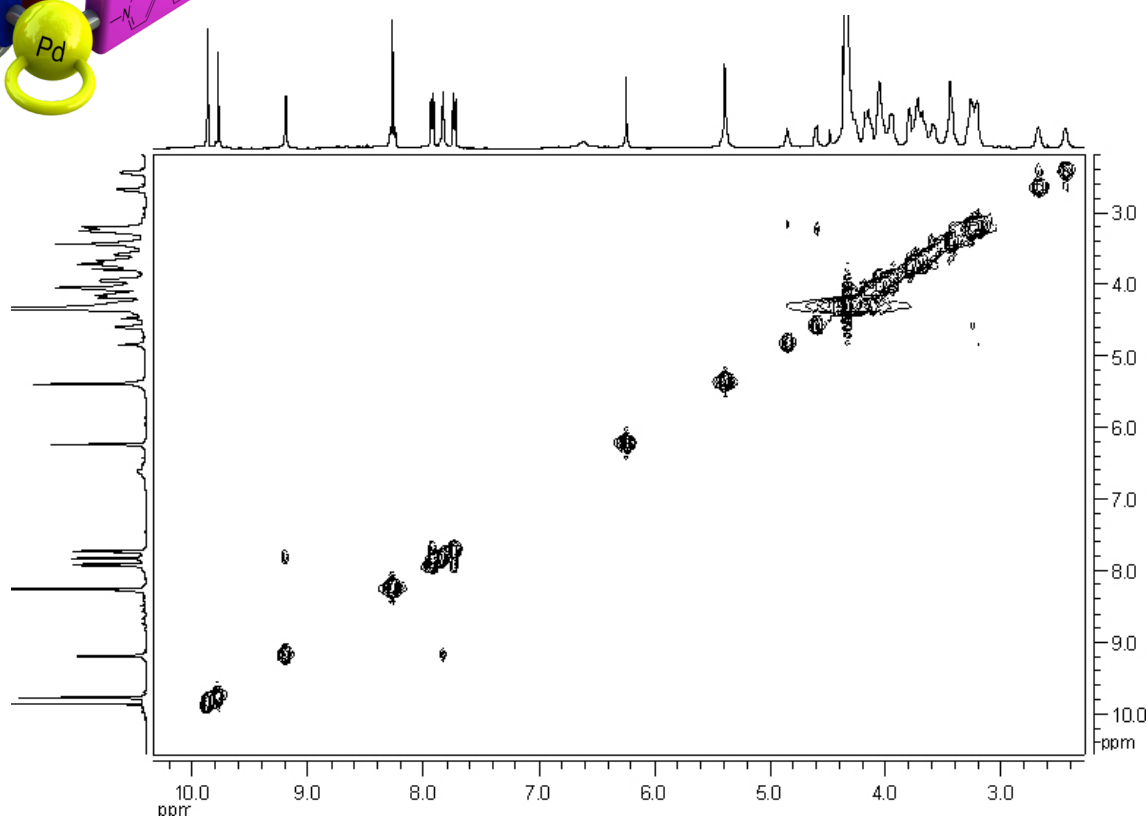
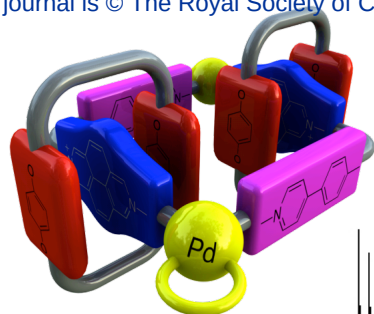


Figure S51: COSY (CD_3NO_2 , 500 MHz) spectrum of **6a(8)**₂·6PF₆.

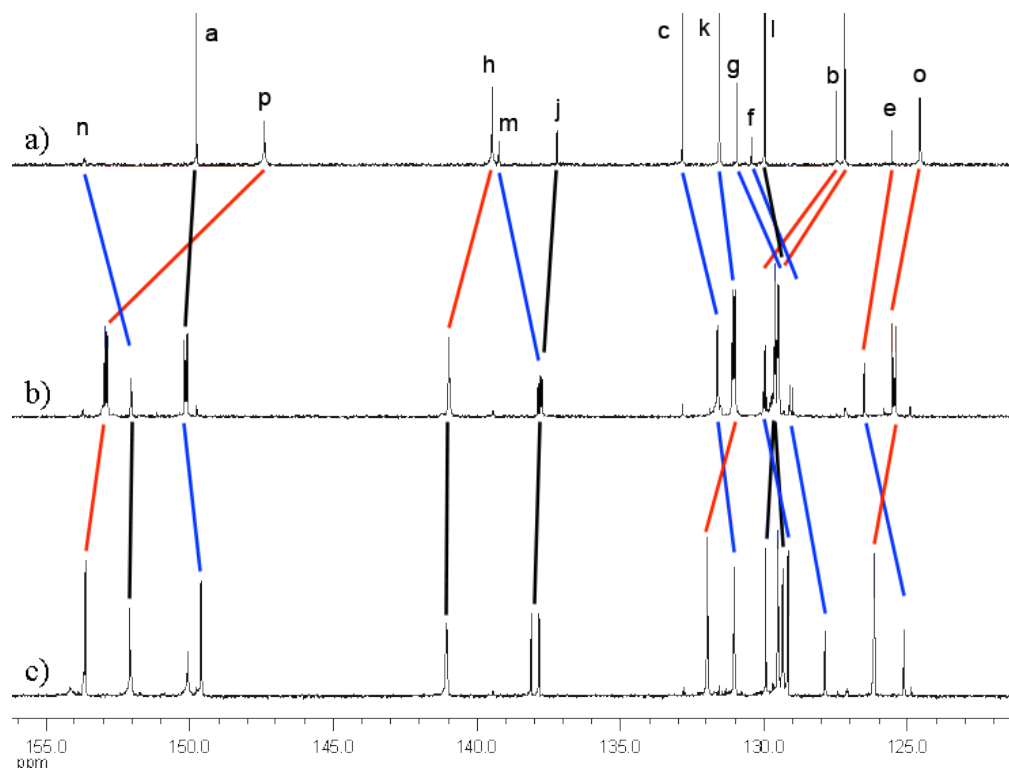


Figure S52: Partial ^{13}C NMR (CD_3NO_2 , 125 MHz) spectra of: (a) ligand **4**·PF₆, (b) metallocycles **6a,b**·6PF₆, and (c) catenane **6a(8)**₂·6PF₆. Peak labels are defined in Scheme 1.

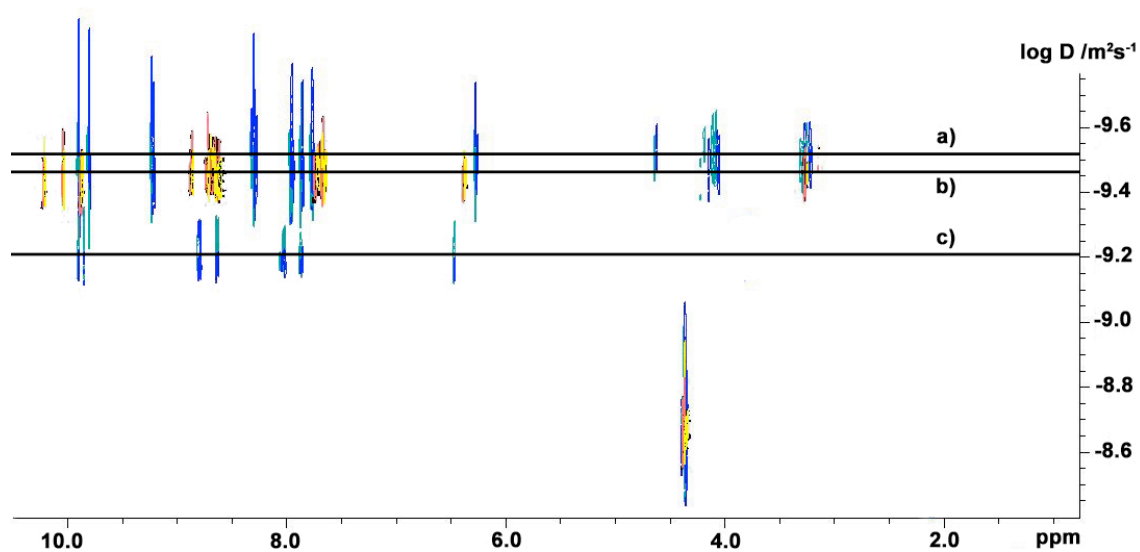


Figure S53: Superposed DOSY (CD_3NO_2 , 500 MHz, 298 K) experiment of: a) 5 mM solution of **6a(8)**₂·6PF₆ (blue). b) 5 mM solution of **6a,b**·6PF₆ (yellow). c) 10 mM solution of ligand **4**·PF₆ (blue).

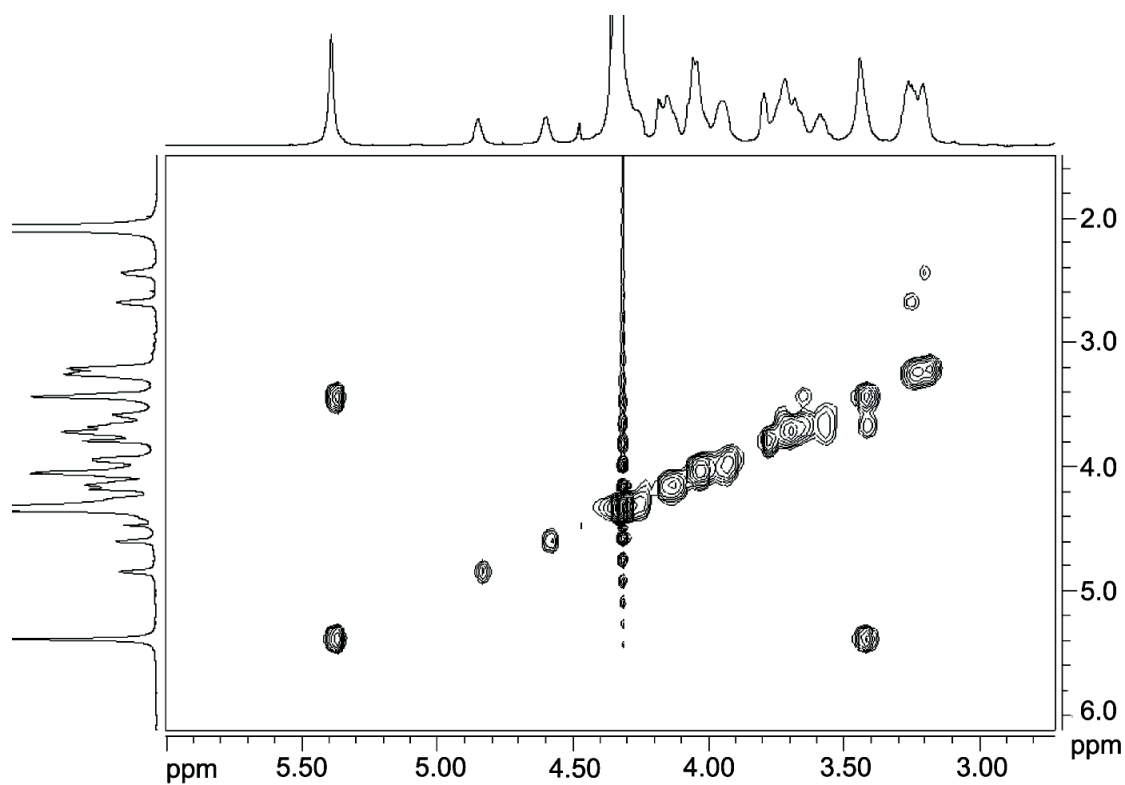


Figure S54: Partial EXSY (CD_3NO_2 , 500 MHz, 298 K) spectrum showing the correlation between HQ_{out} and HQ_{in}

Catenane $6a(9)_2 \cdot 6PF_6$

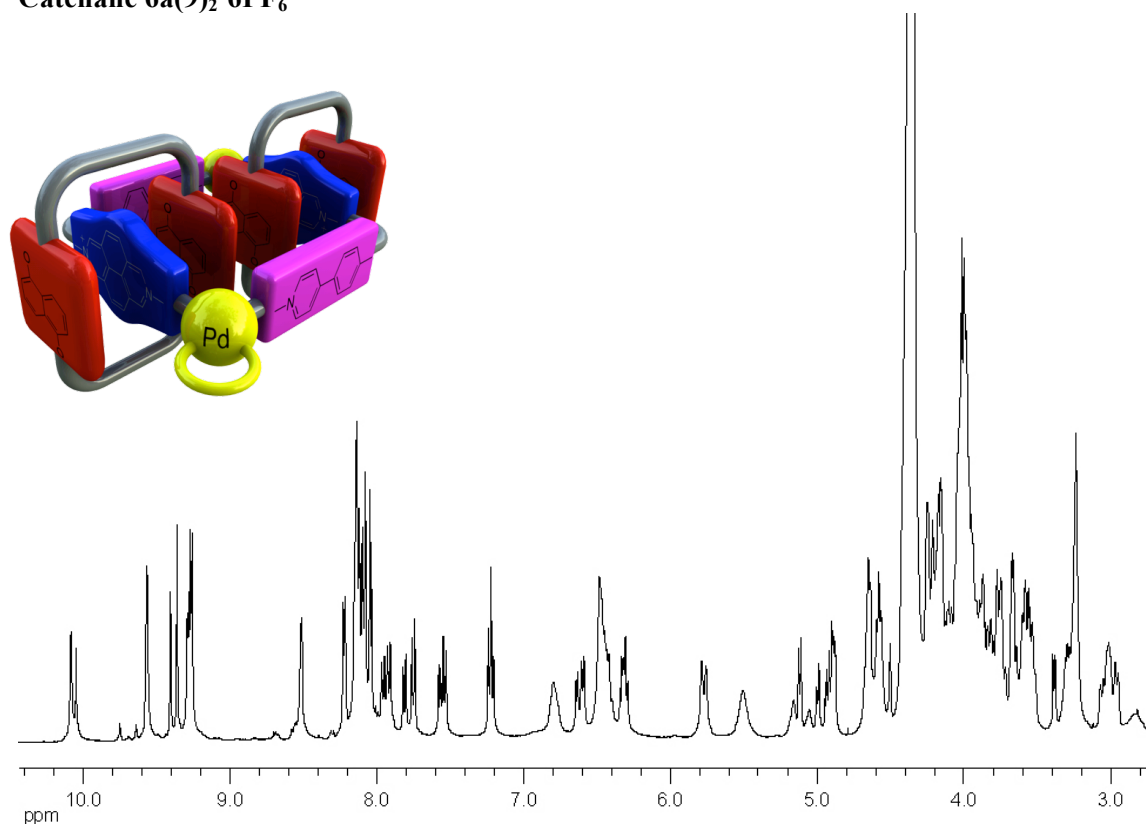


Figure S55: 1H NMR (CD_3NO_2 , 500 MHz, 298 K) spectrum of $6a(9)_2 \cdot 6PF_6$.

Catenane $7a(8)_2 \cdot 6PF_6$

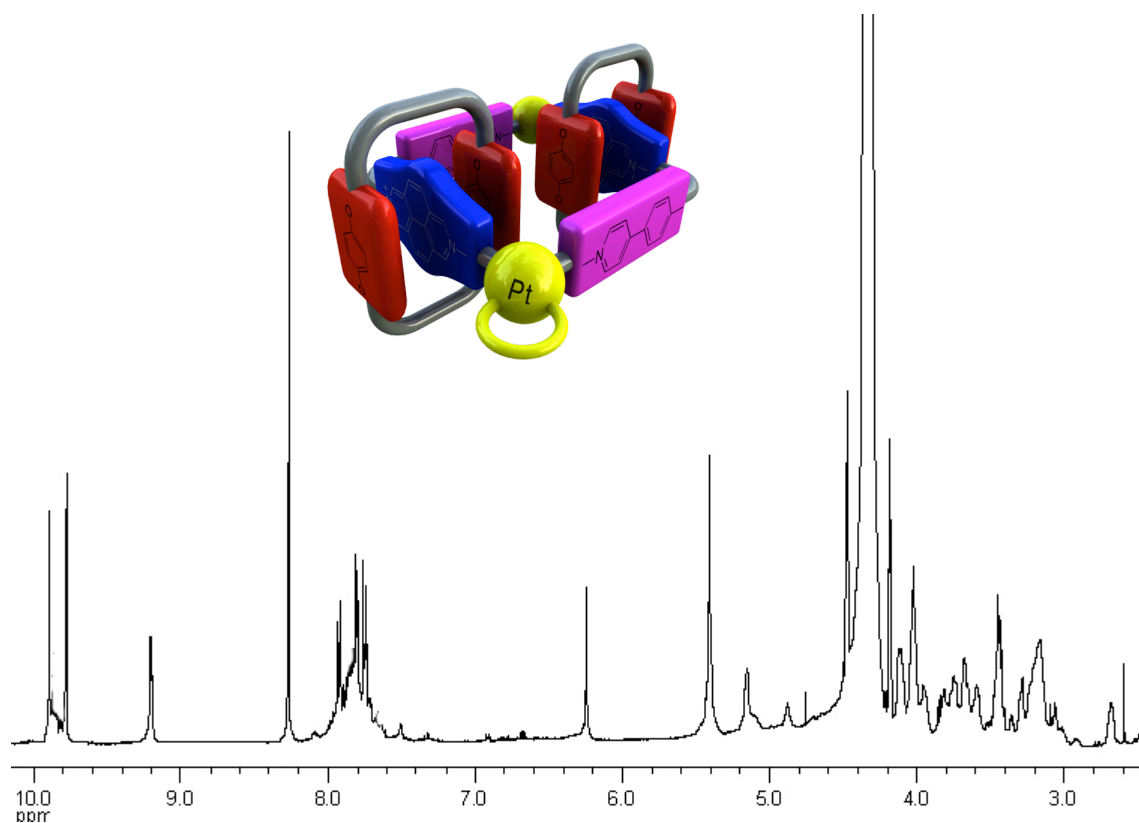


Figure S56: 1H NMR (CD_3NO_2 , 500 MHz) spectrum of $7a(8)_2 \cdot 6PF_6$.

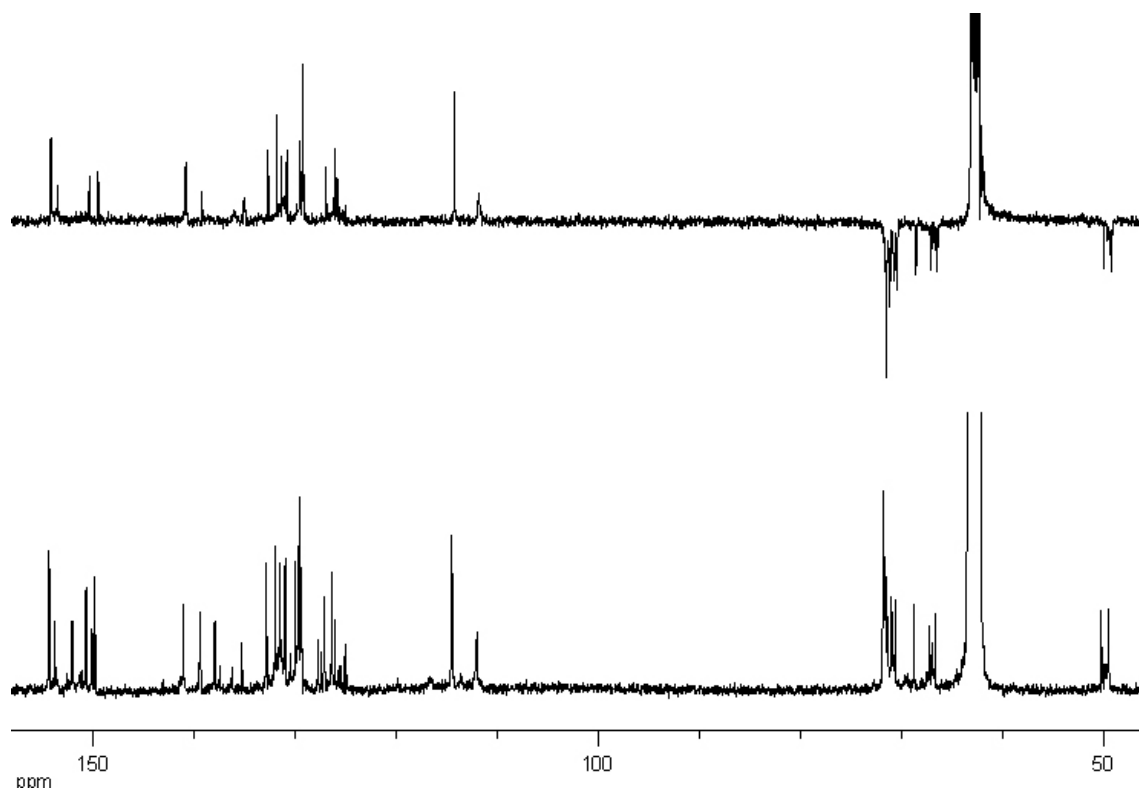


Figure S57: ^{13}C NMR (CD_3NO_2 , 125 MHz) spectrum of $7a(8)_2 \cdot 6PF_6$.

Catenane $7a(9)_2 \cdot 6PF_6$

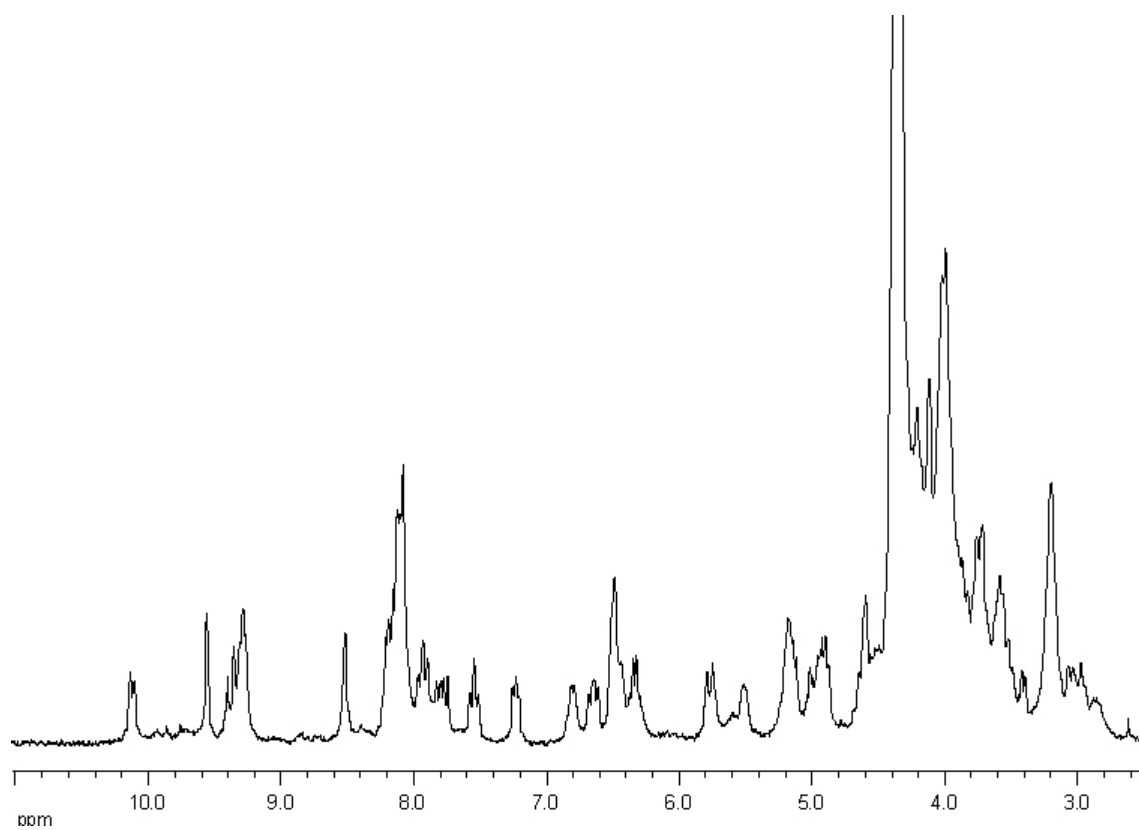


Figure S58: ^1H NMR (CD_3NO_2 , 500 MHz) spectrum de $7a(9)_2 \cdot 6PF_6$.

Inclusion complex $(10)_2 \subset 6a \cdot 6NO_3$

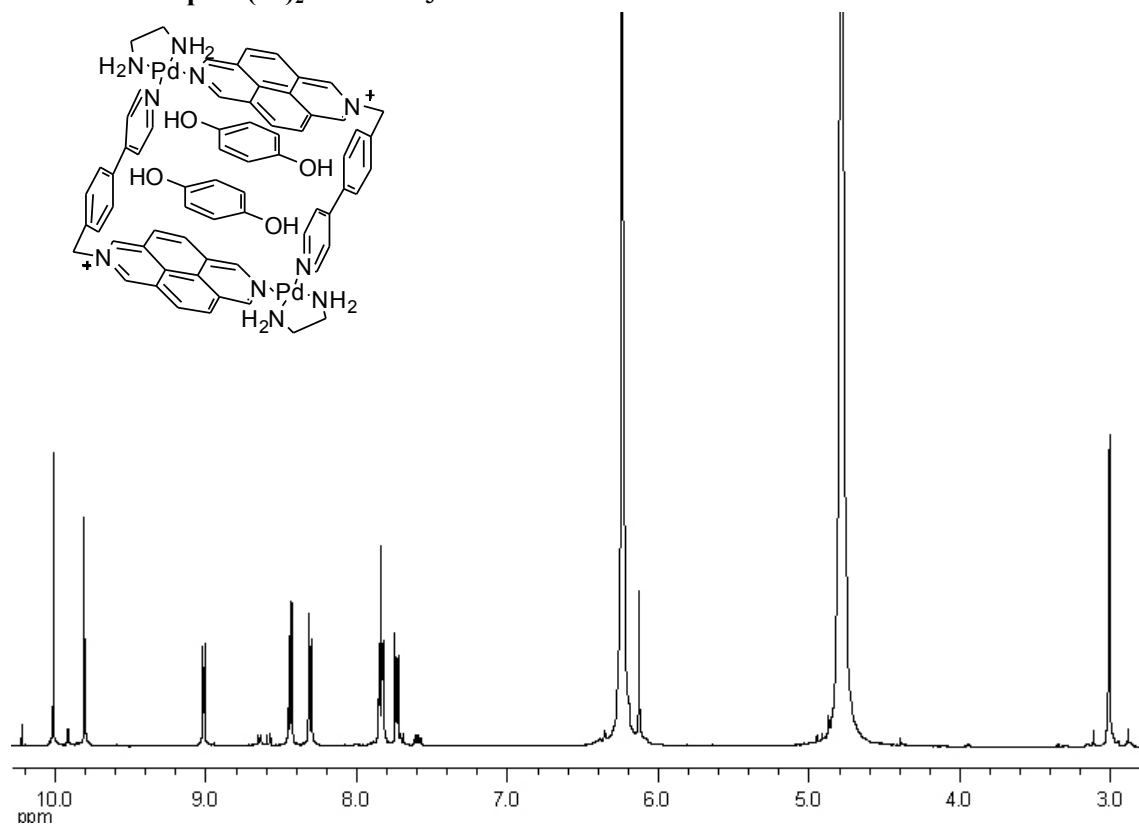


Figure S59: 1H NMR (D_2O , 500 MHz) spectrum of $(10)_2 \subset 6a \cdot 6NO_3$

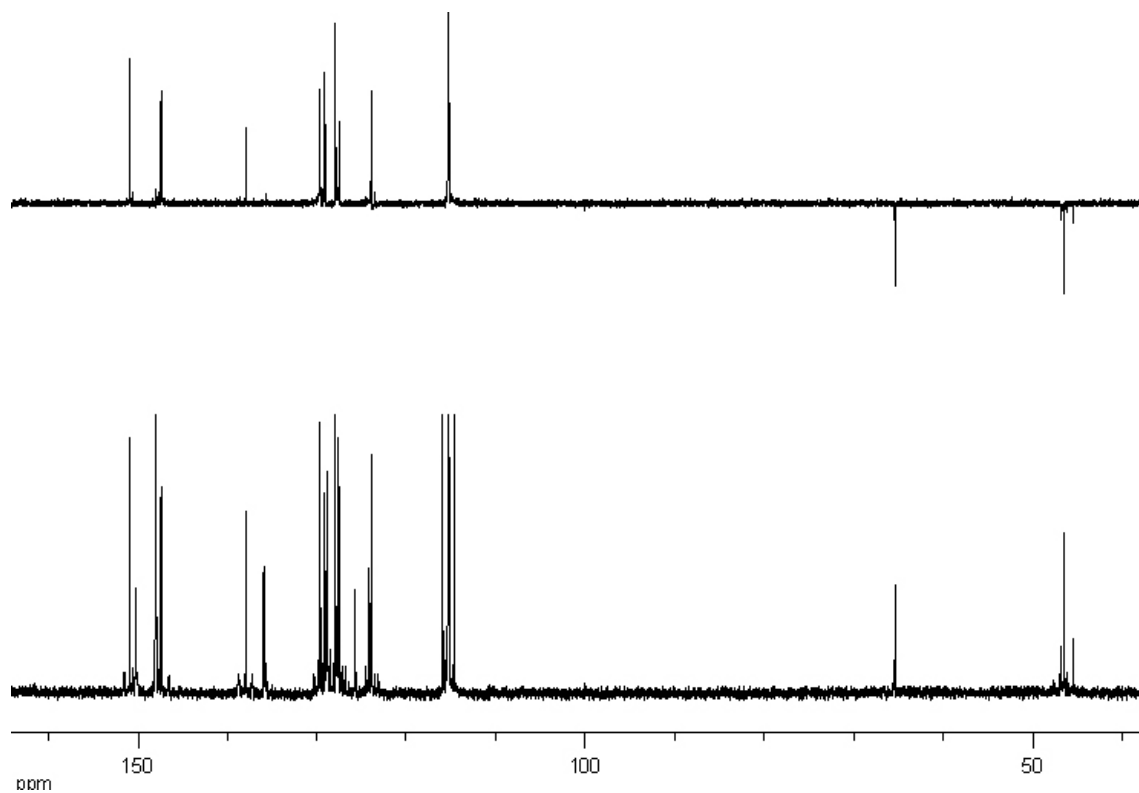


Figure S60: ^{13}C NMR (D_2O , 125 MHz) of $(10)_2 \subset 6a \cdot 6NO_3$.

Inclusion complex $(11)_2 \subset 6a \cdot 6NO_3$

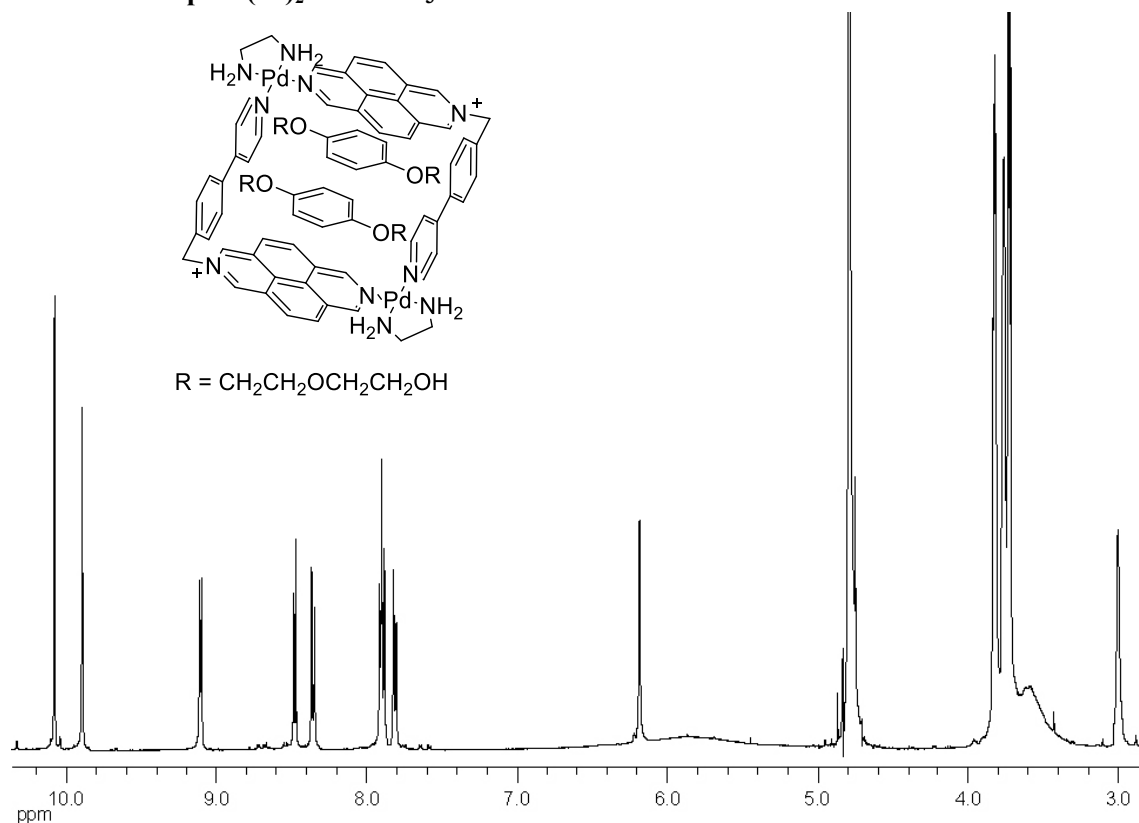


Figure S61: 1H NMR (D₂O, 500 MHz) spectrum of $(11)_2 \subset 6a \cdot 6NO_3$.

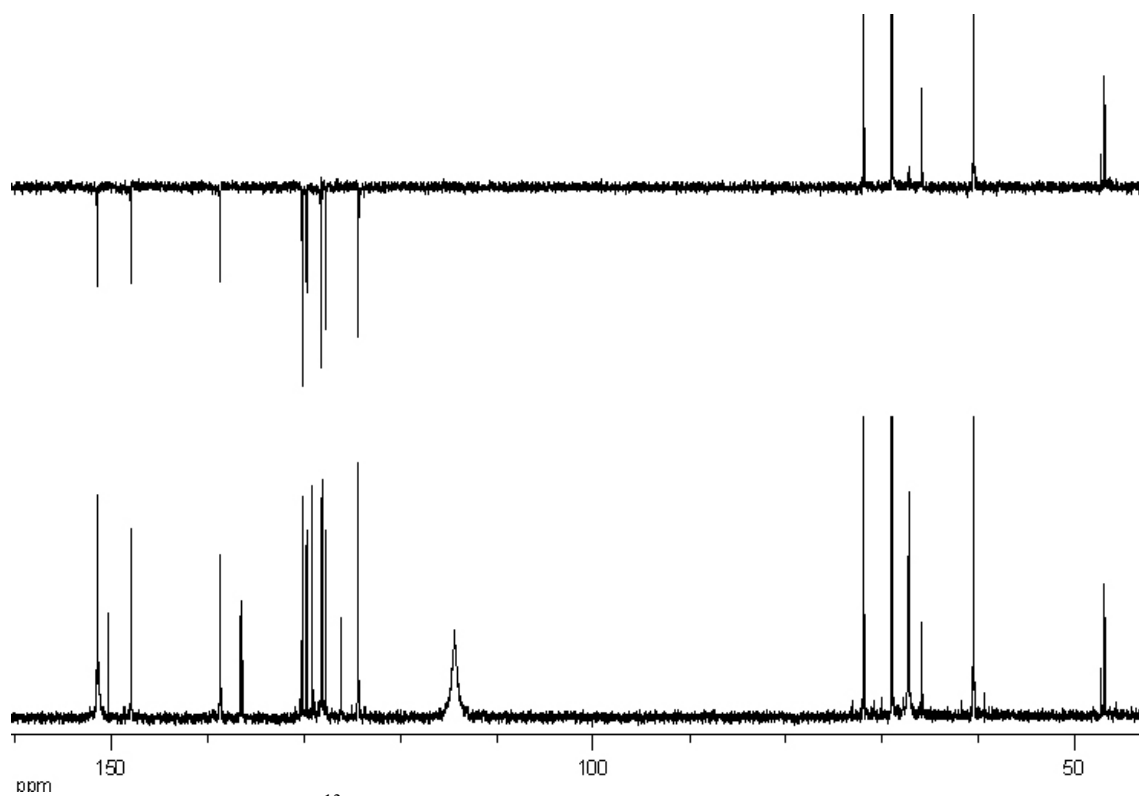


Figure S62: ^{13}C NMR (D₂O, 125 MHz) spectrum of $(11)_2 \subset 6a \cdot 6NO_3$.

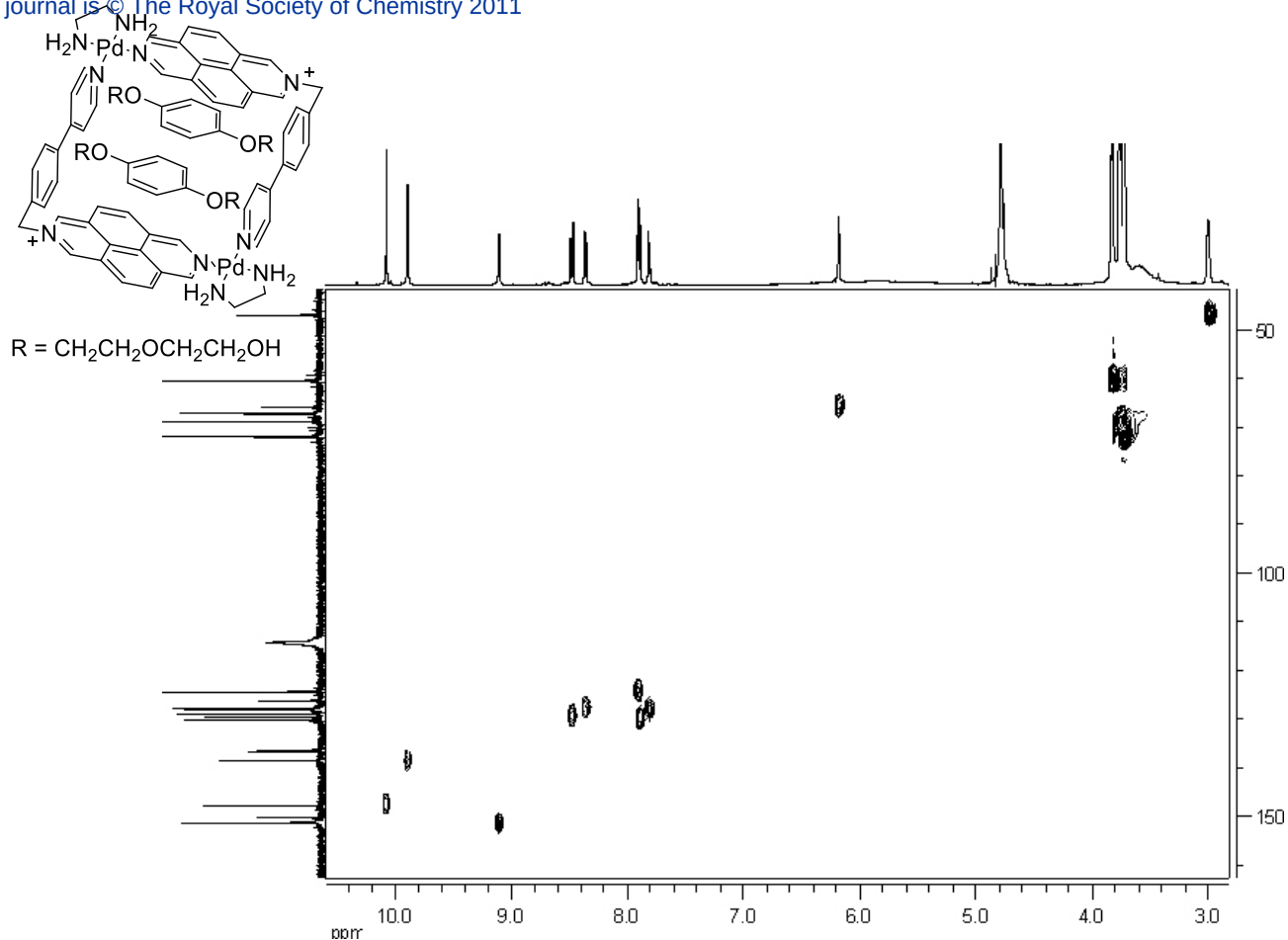


Figure S63: HSQC (D₂O, 500 and 125 MHz) spectrum of $(\mathbf{11})_2\cdot\mathbf{6a}\cdot 6\text{NO}_3$.

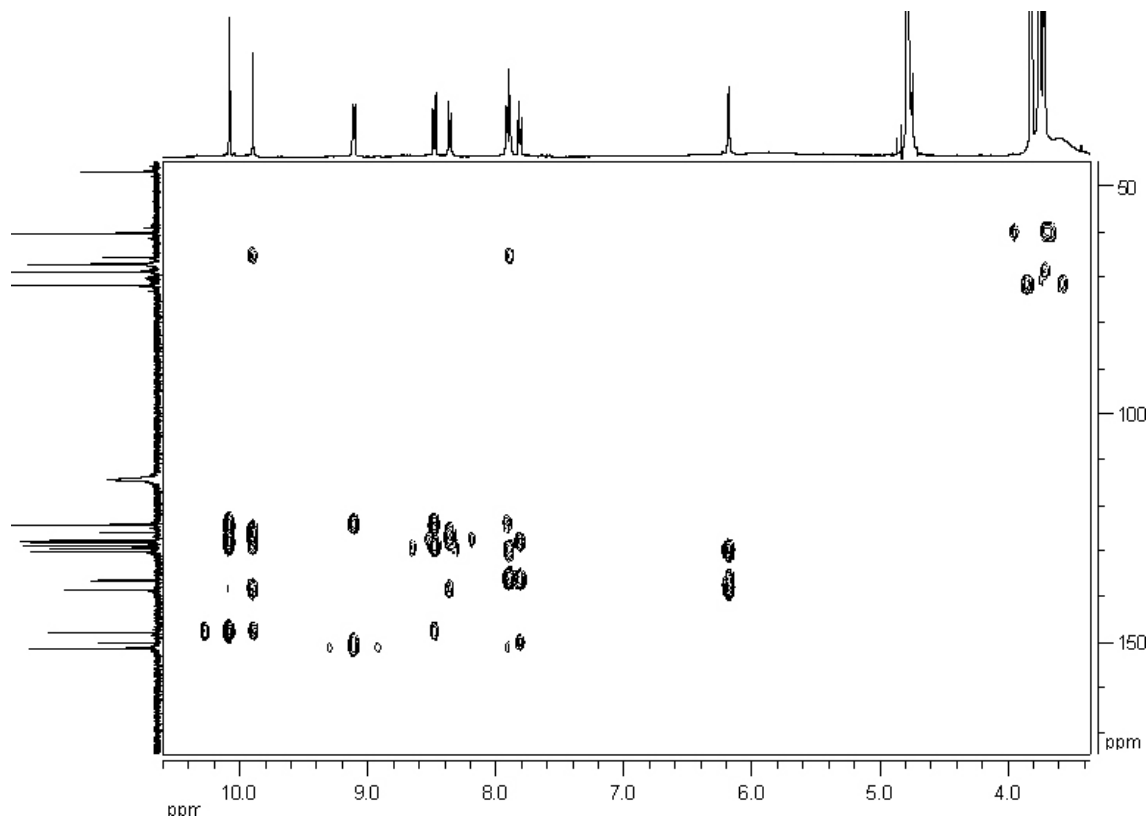


Figure S64: HMBC (D₂O, 500 and 125 MHz) spectrum of $(\mathbf{11})_2\cdot\mathbf{6a}\cdot 6\text{NO}_3$.

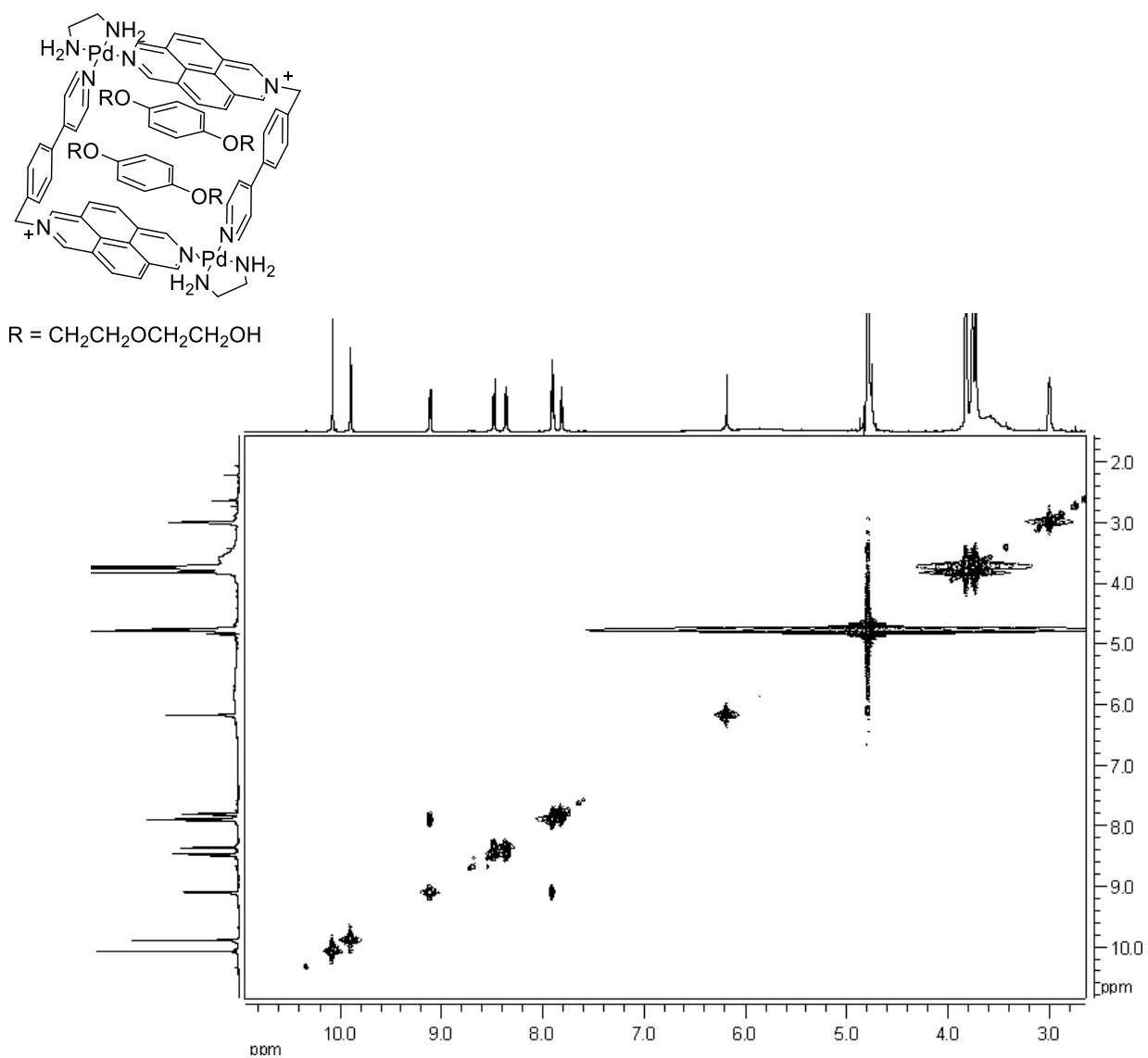


Figure S65: COSY (D_2O , 500 MHz) spectrum of $(11)_2 \cdot 6a \cdot 6NO_3$.

Inclusion complex $(\mathbf{13})_2\subset\mathbf{6a}\cdot 6\text{NO}_3$

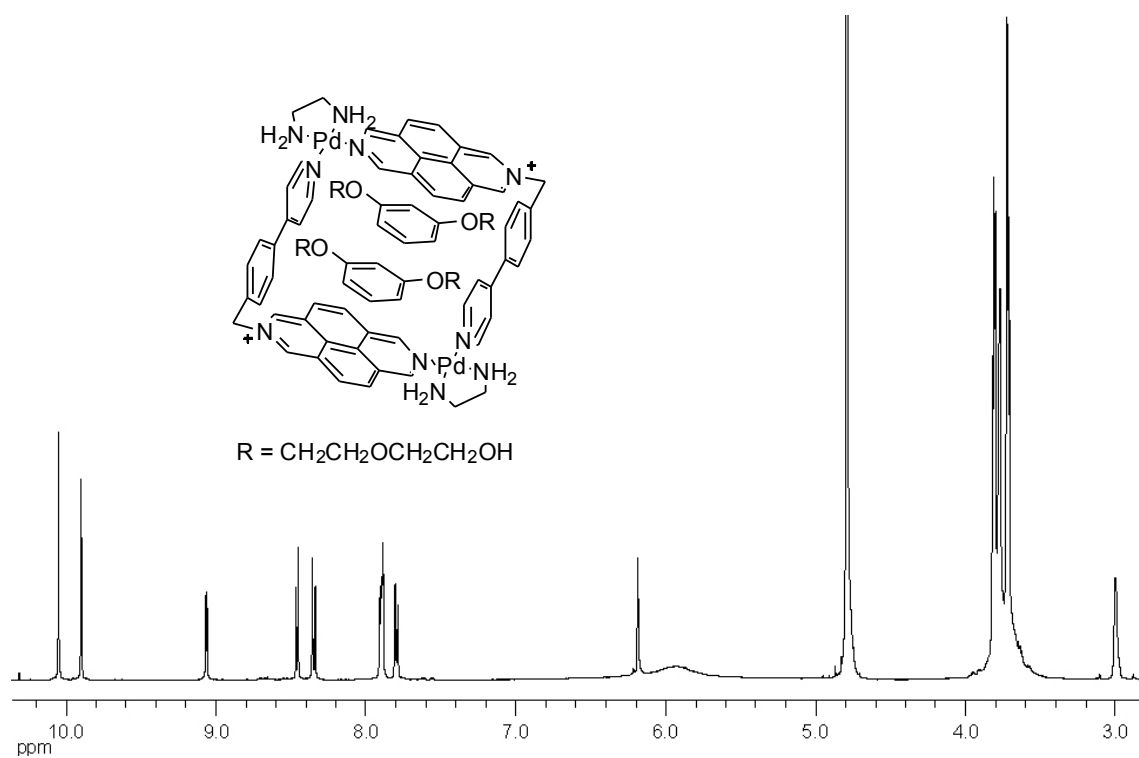


Figure S66: ^1H NMR (D₂O, 500 MHz) spectrum of $(\mathbf{13})_2\subset\mathbf{6a}\cdot 6\text{NO}_3$

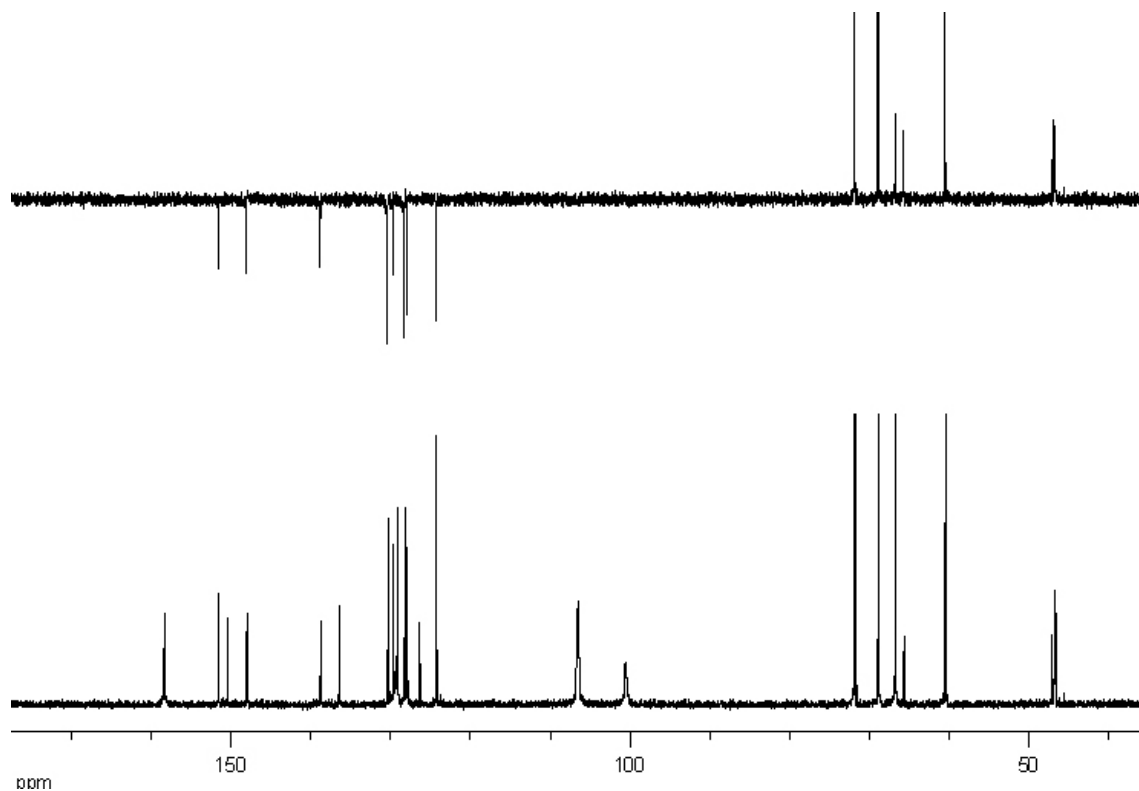


Figure S67: ^{13}C NMR (D₂O, 125 MHz) spectrum of $(\mathbf{13})_2\subset\mathbf{6a}\cdot 6\text{NO}_3$

Inclusion complex $(14)_2 \subset 6a \cdot 6NO_3$

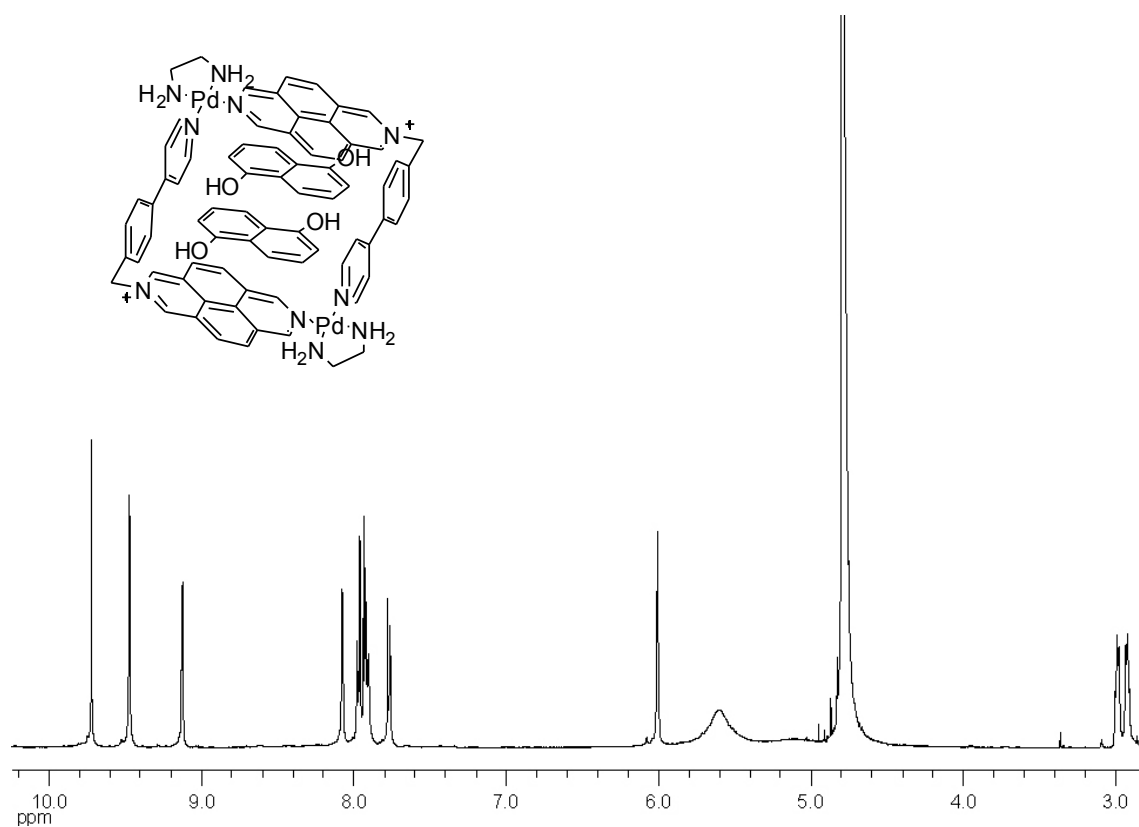


Figure S68: 1H NMR (D₂O, 500 MHz) spectrum of $(14)_2 \subset 6a \cdot 6NO_3$.

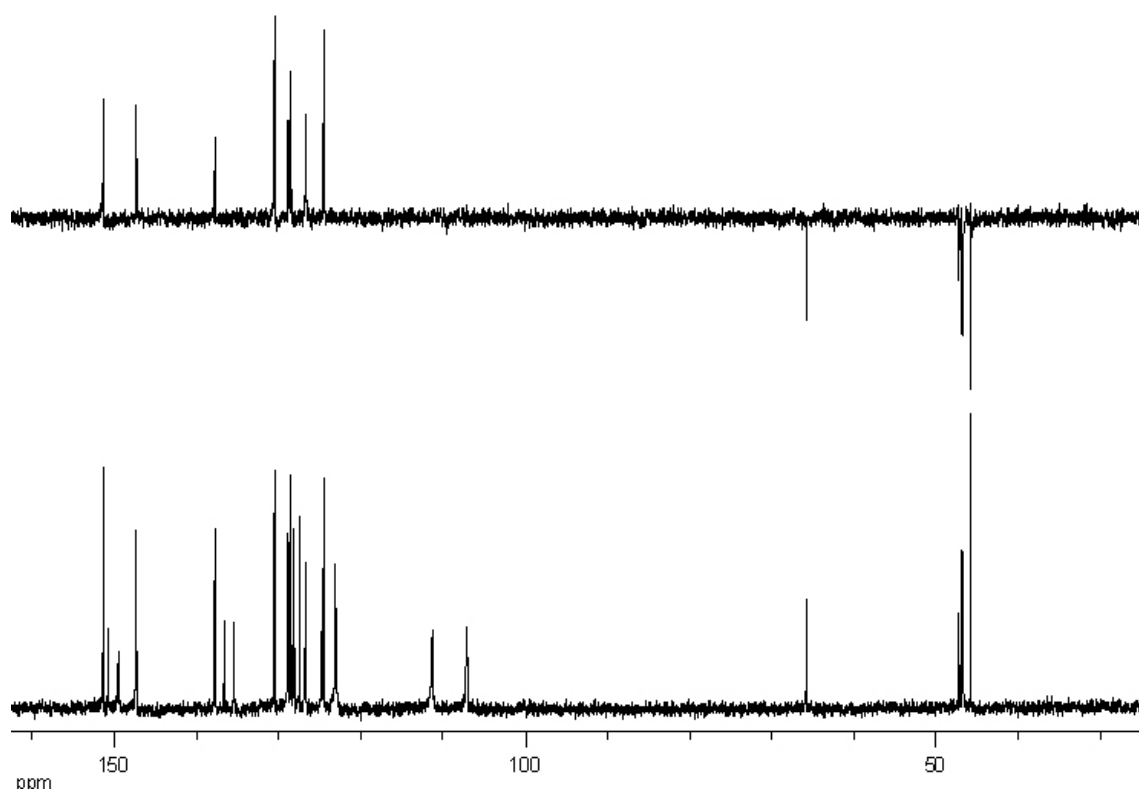


Figure S69: ^{13}C NMR (D₂O, 125 MHz) spectrum of $(14)_2 \subset 6a \cdot 6NO_3$.

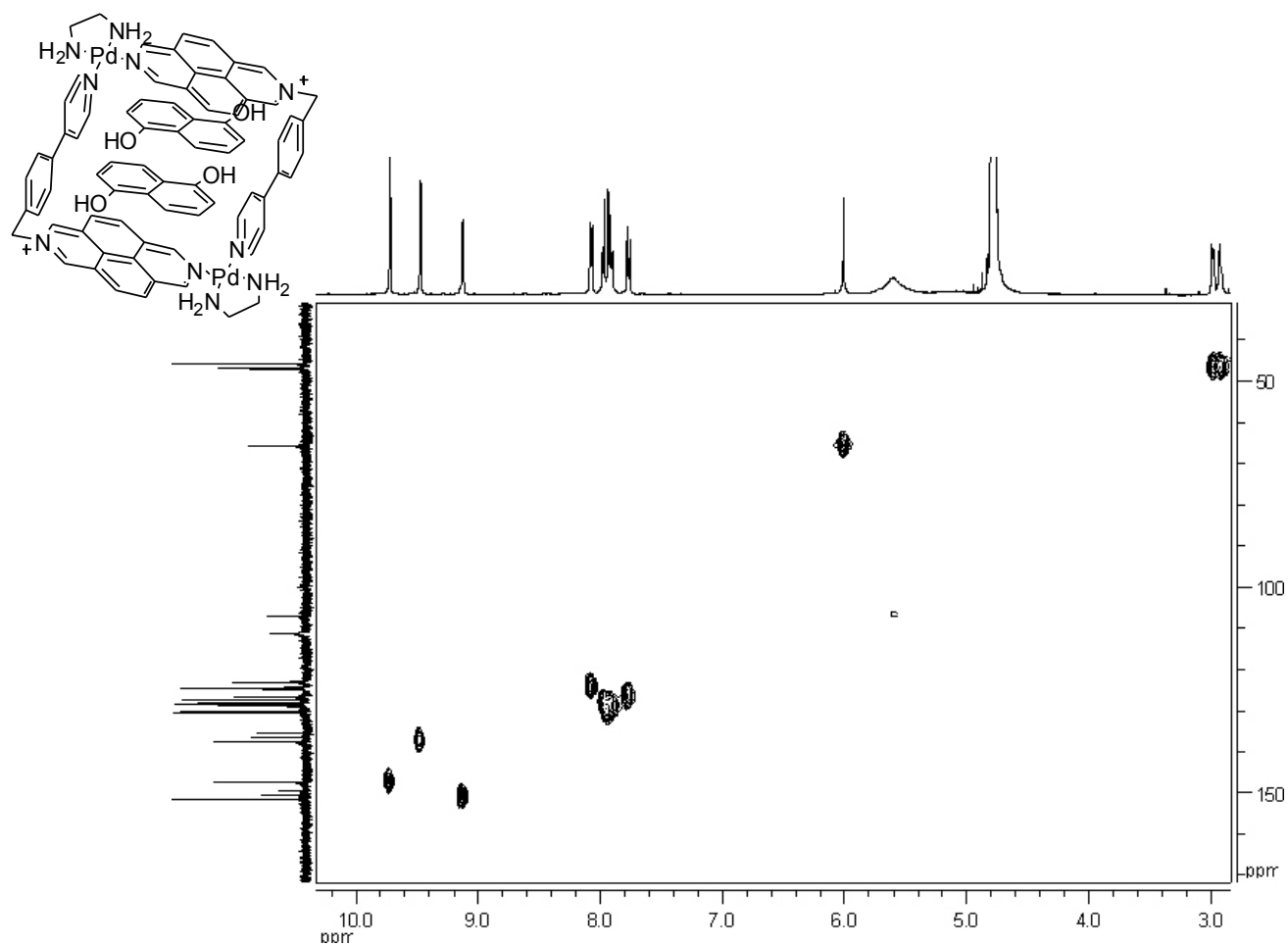


Figure S70: HSQC (D₂O, 500 and 125 MHz) spectrum of (14)₂·6a·6NO₃.

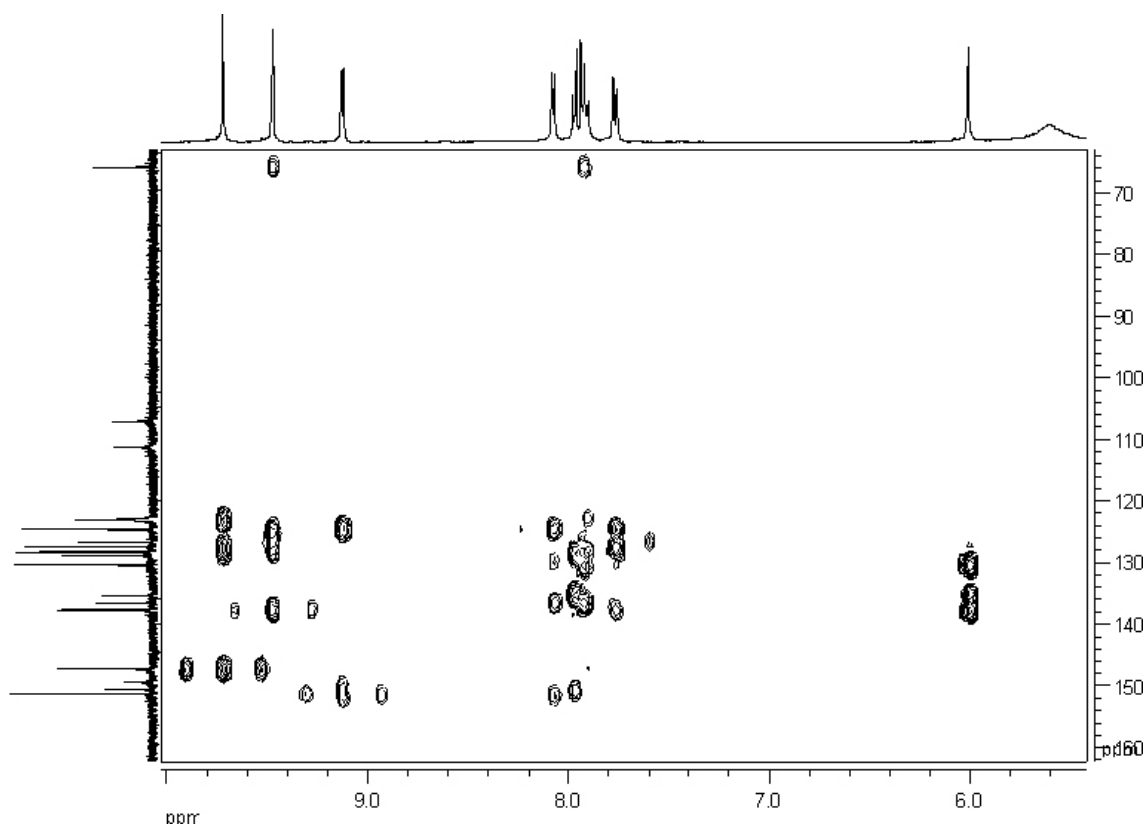


Figure S71: HMBC (D₂O, 500 and 125 MHz) spectrum of (14)₂·6a·6NO₃.

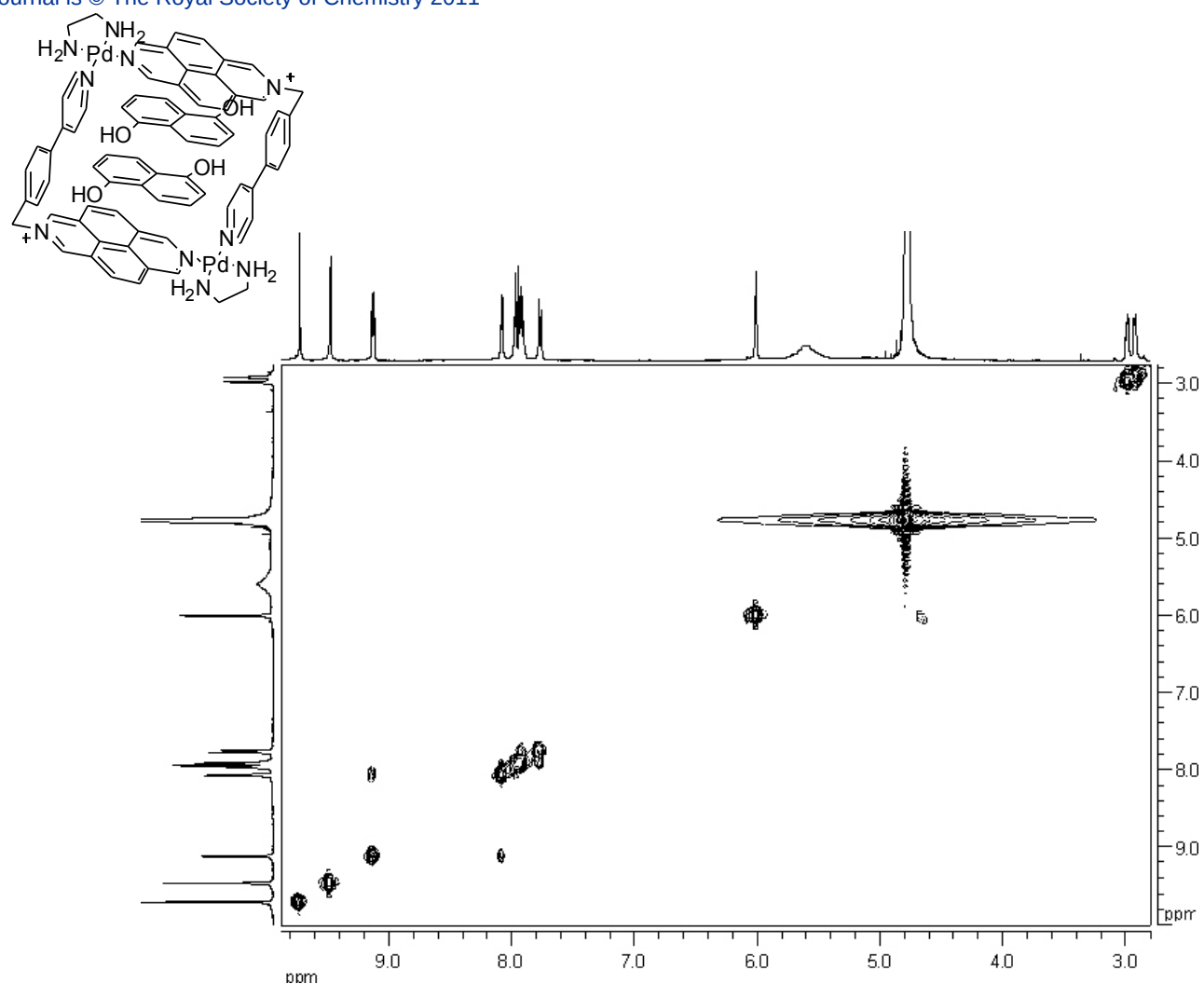


Figure S72: COSY (D₂O, 500 MHz) spectrum of (14)₂·6a·6NO₃.

Inclusion complex $(15)_2 \subset 6a \cdot 6NO_3$

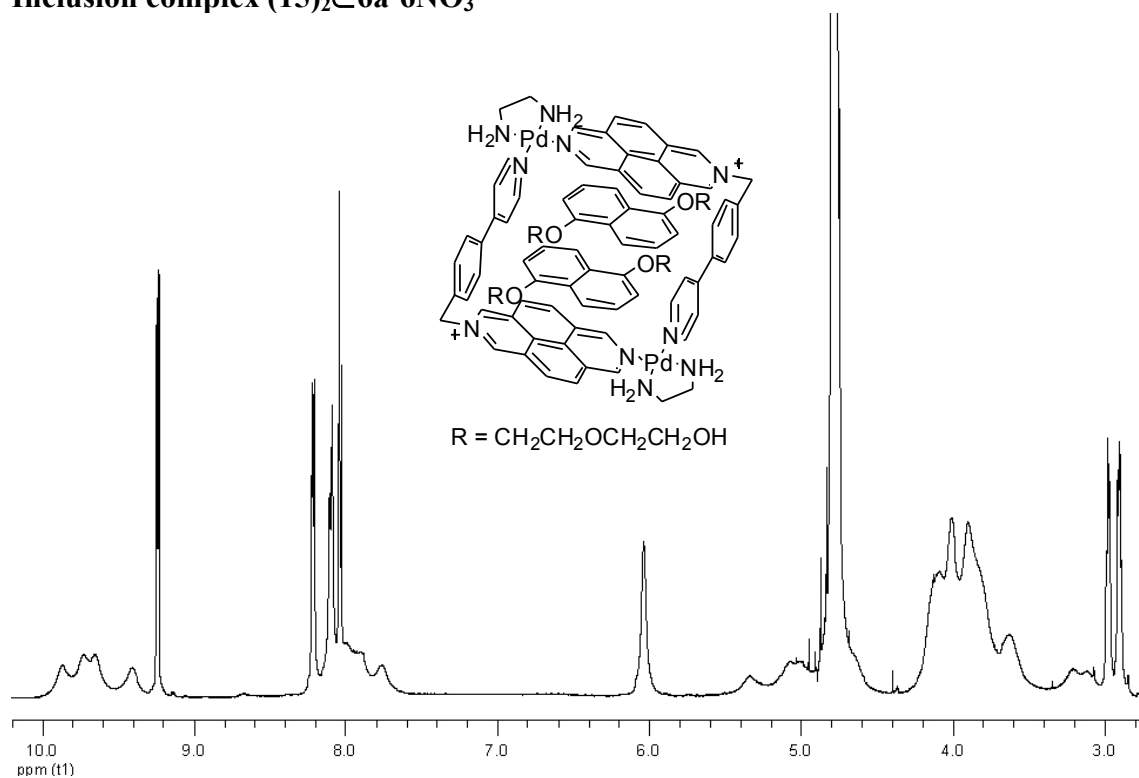


Figure S73: 1H NMR (D_2O , 500 MHz, 298 K) spectrum of $(15)_2 \subset 6a \cdot 6NO_3$.

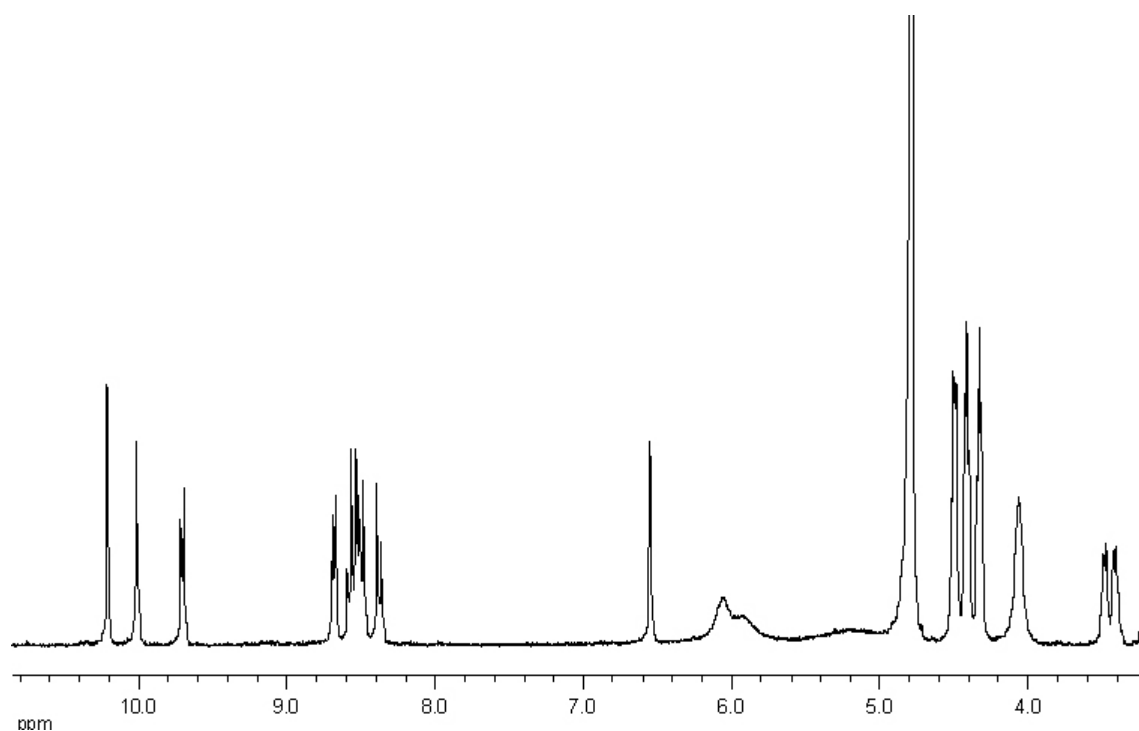


Figure S74: 1H NMR (D_2O , 300 MHz, 353 K) spectrum of $(15)_2 \subset 6a \cdot 6NO_3$.

Inclusion complex $(16)_2 \subset 6a \cdot 6NO_3$

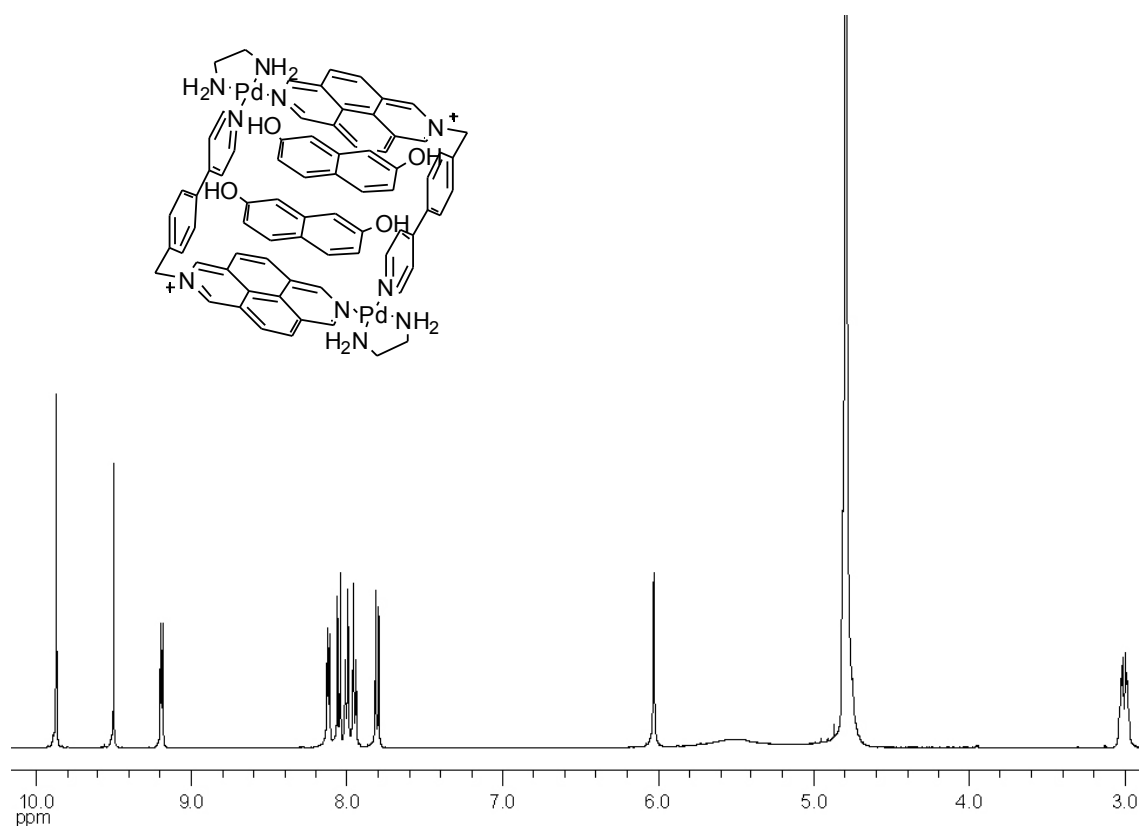


Figure S75: 1H NMR (D₂O, 500 MHz) spectrum of $(16)_2 \subset 6a \cdot 6NO_3$.

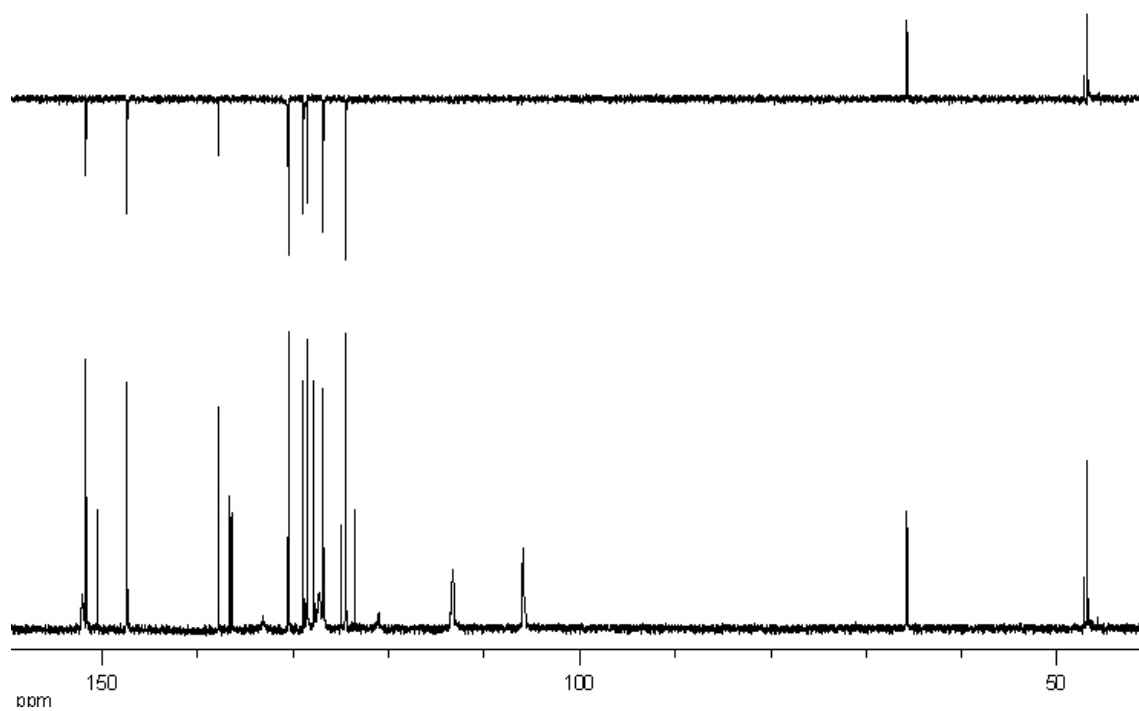


Figure S76: ^{13}C NMR (D₂O, 125 MHz) spectrum of $(16)_2 \subset 6a \cdot 6NO_3$.

Inclusion complex $(17)_2 \subset 6a \cdot 6NO_3$

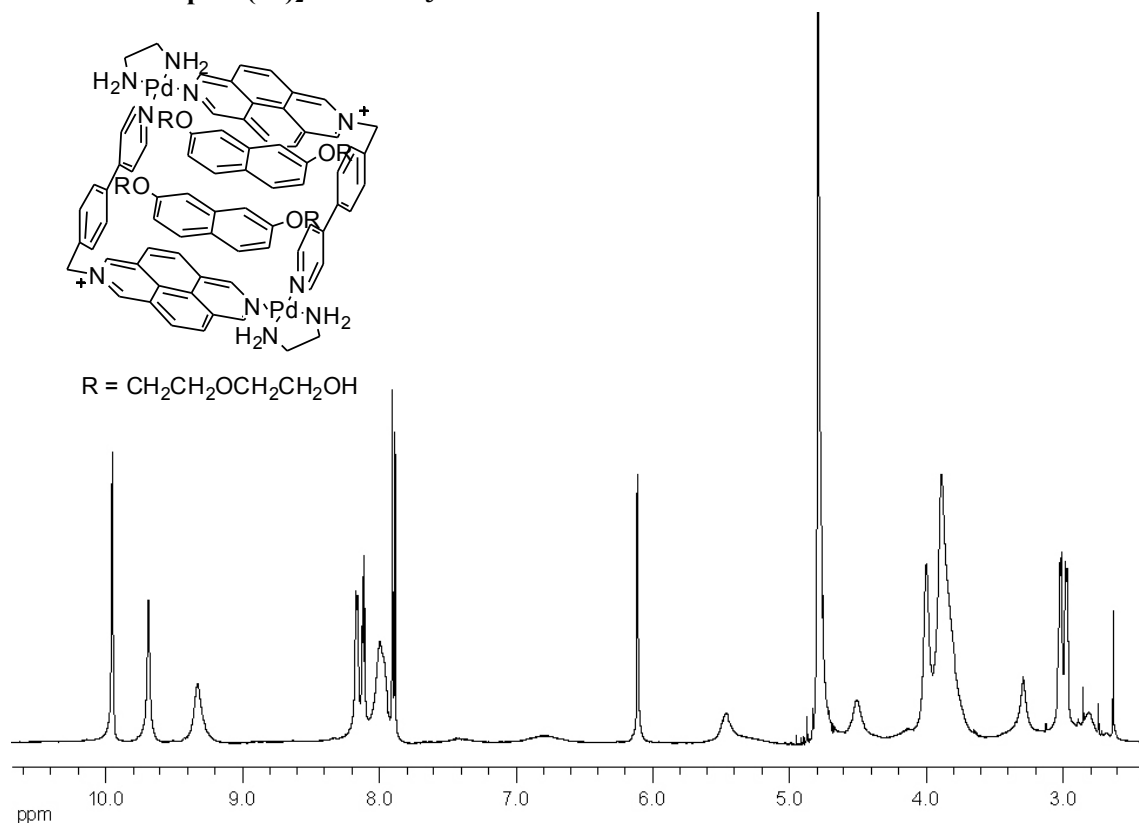


Figure S77: 1H NMR (D₂O, 500 MHz) spectrum of de $(17)_2 \subset 6a \cdot 6NO_3$.

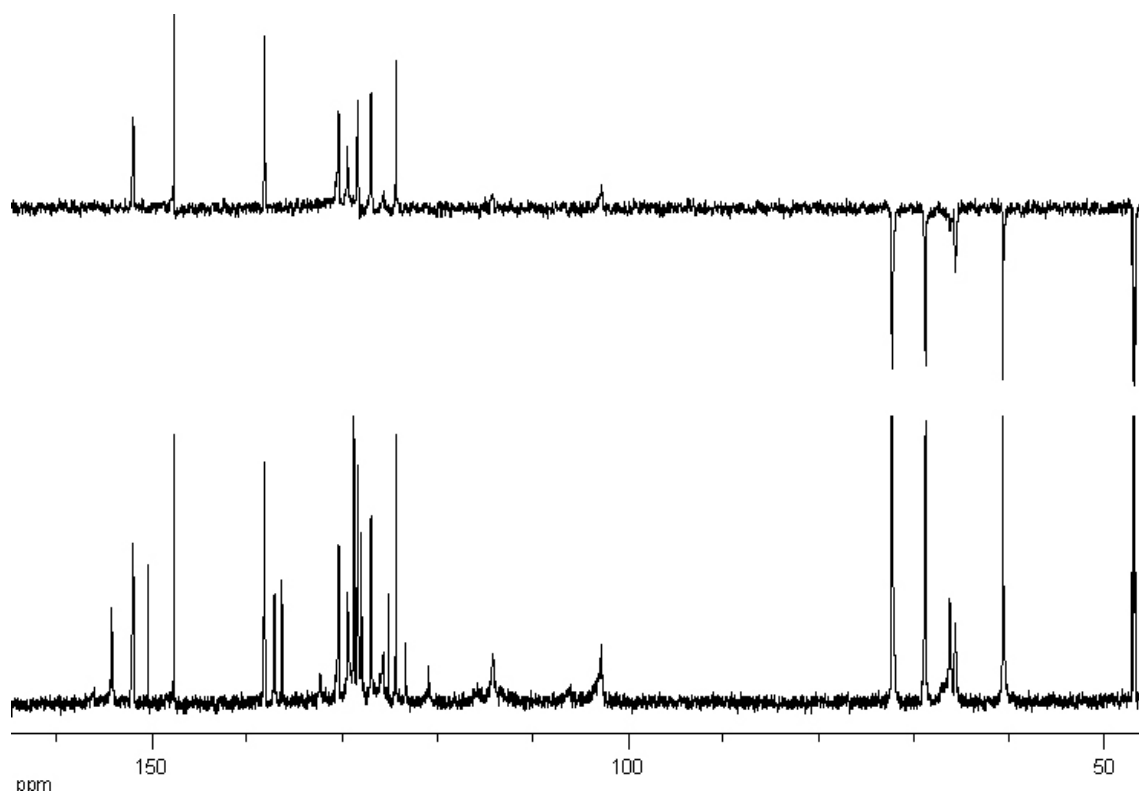


Figure S78: ^{13}C NMR (D₂O, 125 MHz) spectrum of $(17)_2 \subset 6a \cdot 6NO_3$.

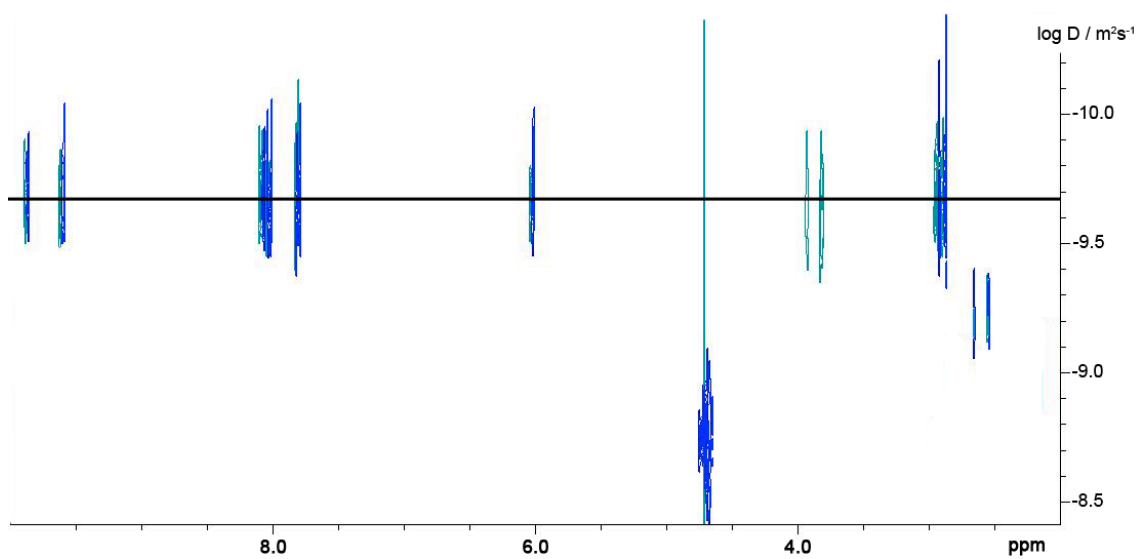


Figure S79: DOSY (D_2O , 500 MHz, 298 K) experiment of inclusion complex $(17)_2 \cdot 6a \cdot 6\text{NO}_3$.

Inclusion complex $(18)_2 \subset 6a \cdot 6NO_3$

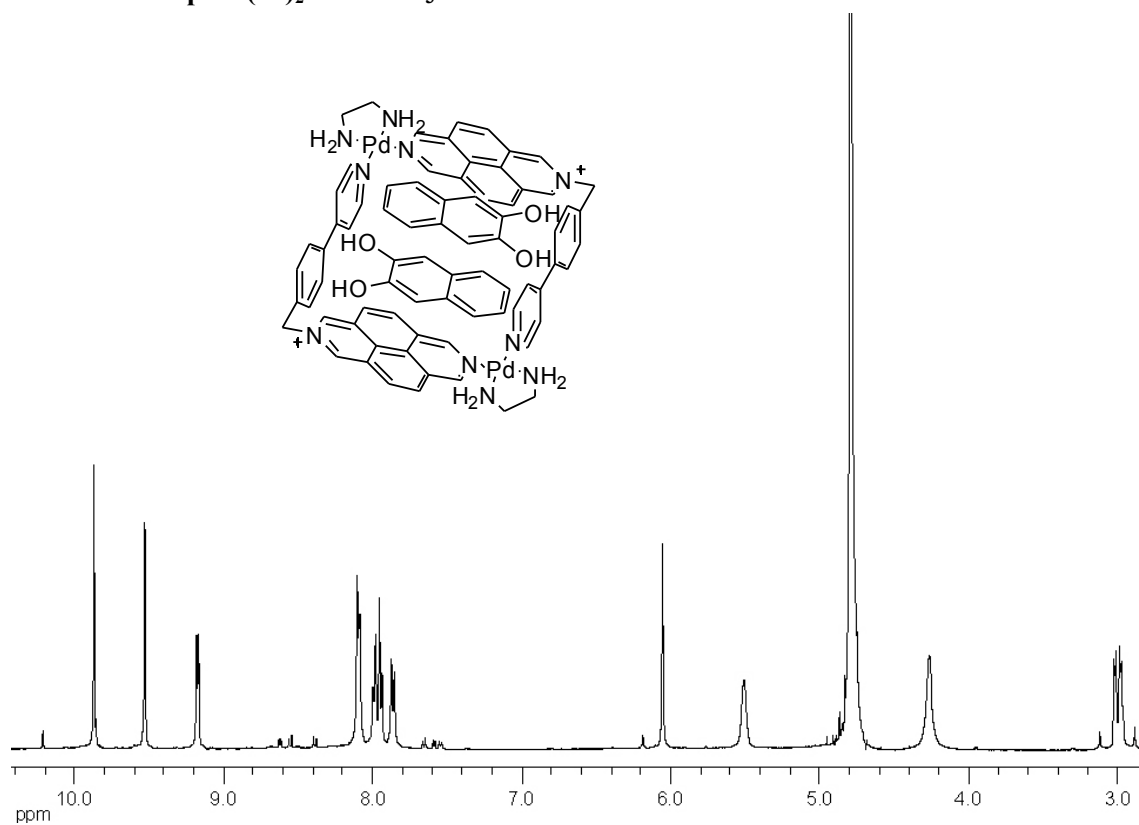


Figure S80: 1H NMR (D₂O, 500 MHz) spectrum of $(18)_2 \subset 6a \cdot 6NO_3$.

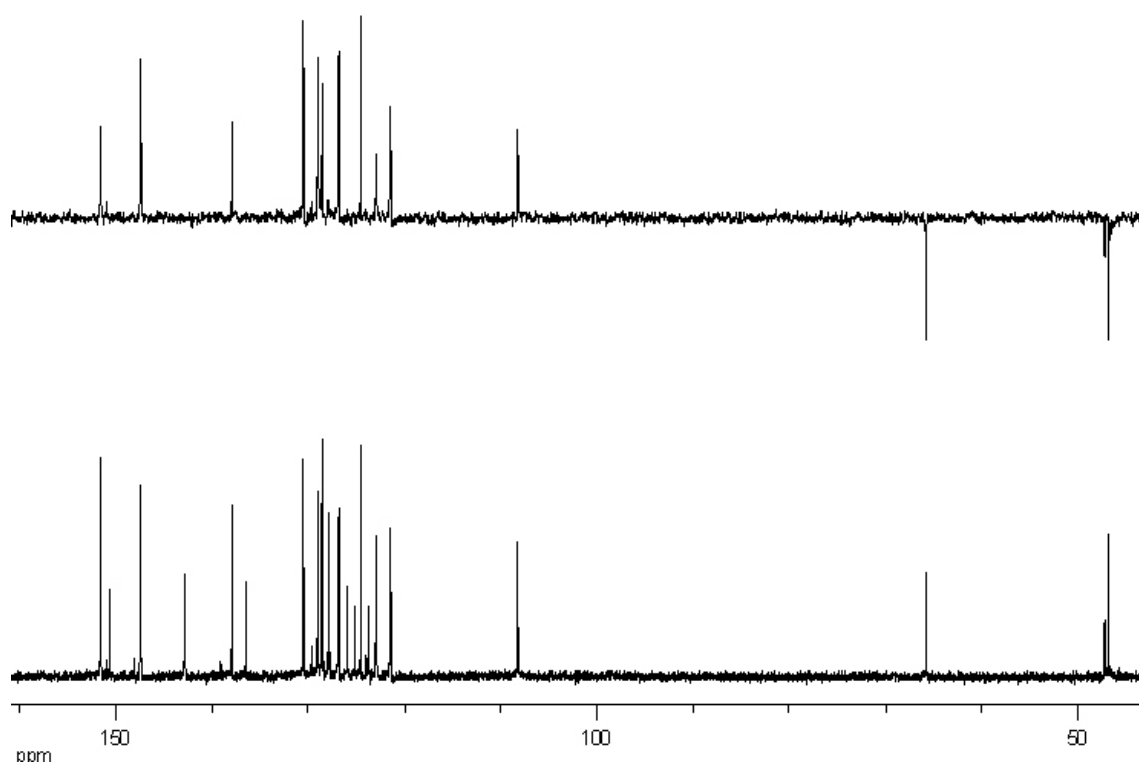


Figure S81: ^{13}C NMR (D₂O, 125 MHz) spectrum of $(18)_2 \subset 6a \cdot 6NO_3$.

Inclusion complex $(19)_2 \subset 6a \cdot 6NO_3$

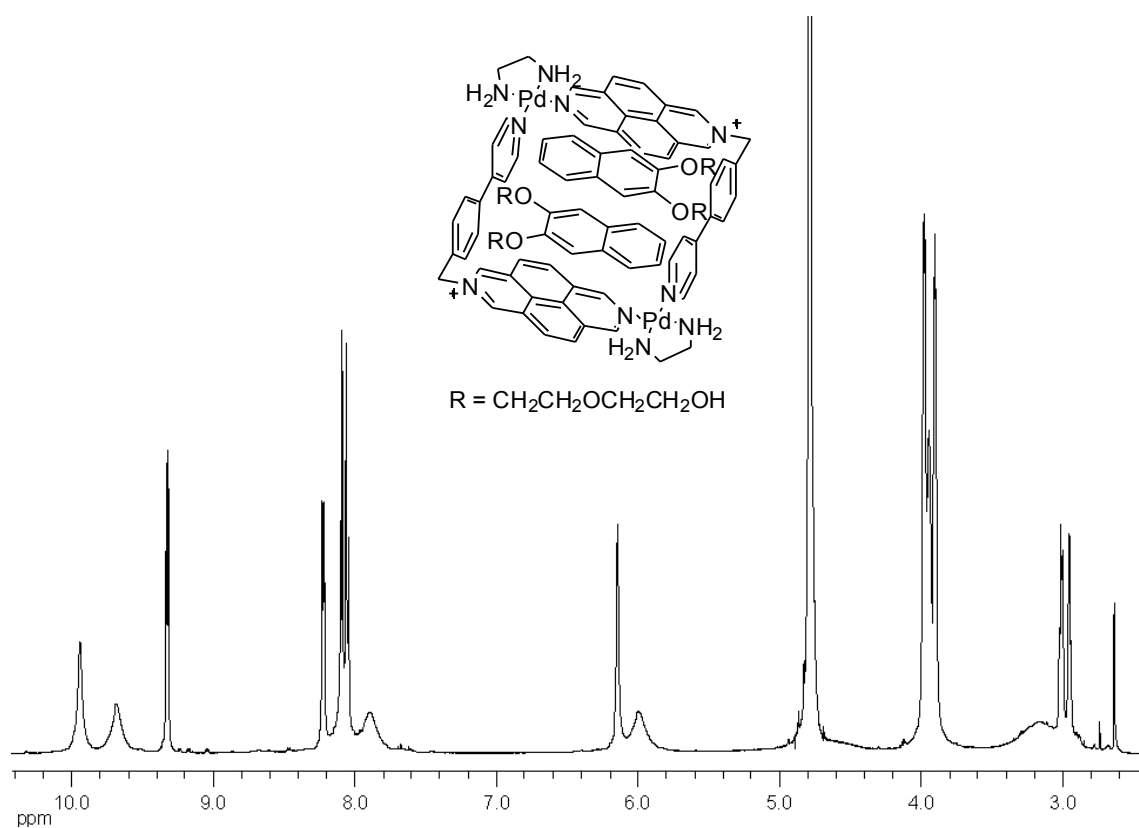


Figure S82: 1H NMR (D_2O , 500 MHz) spectrum of $(19)_2 \subset 6a \cdot 6NO_3$.

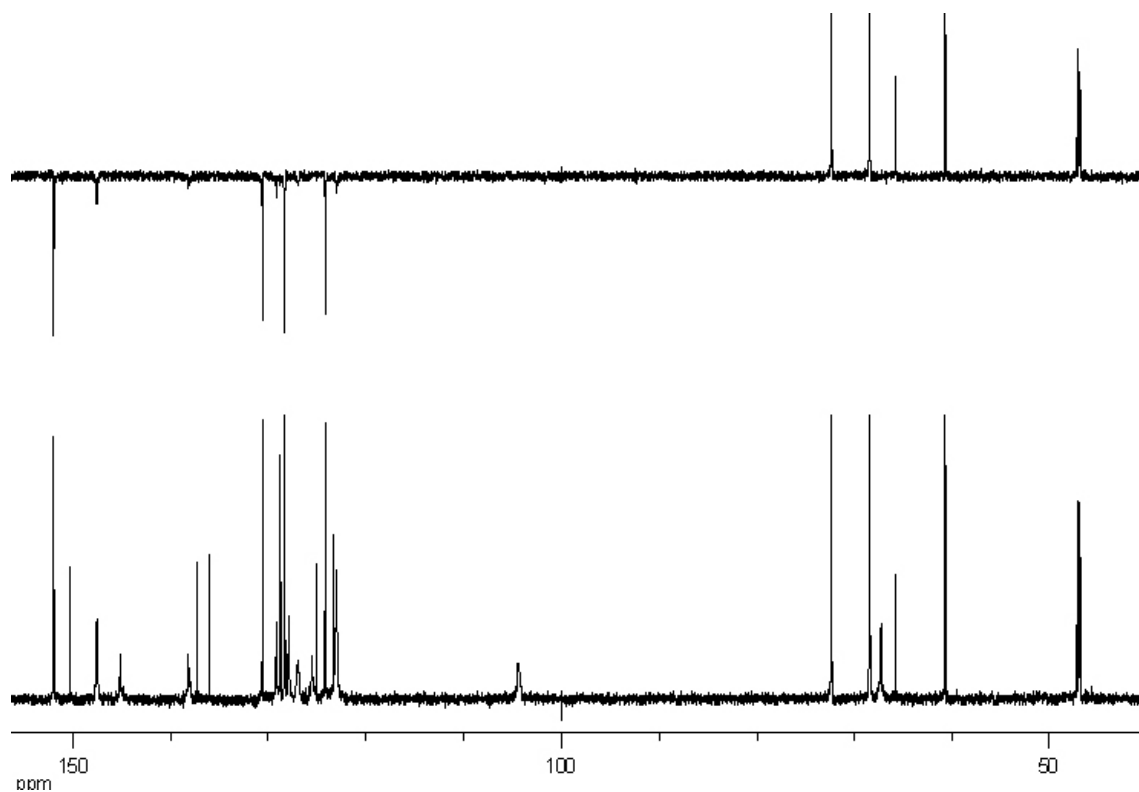


Figure S83: ^{13}C NMR (D_2O , 125 MHz) spectrum of $(19)_2 \subset 6a \cdot 6NO_3$.

Inclusion complex $(20)_2 \subset 6a \cdot 6NO_3$

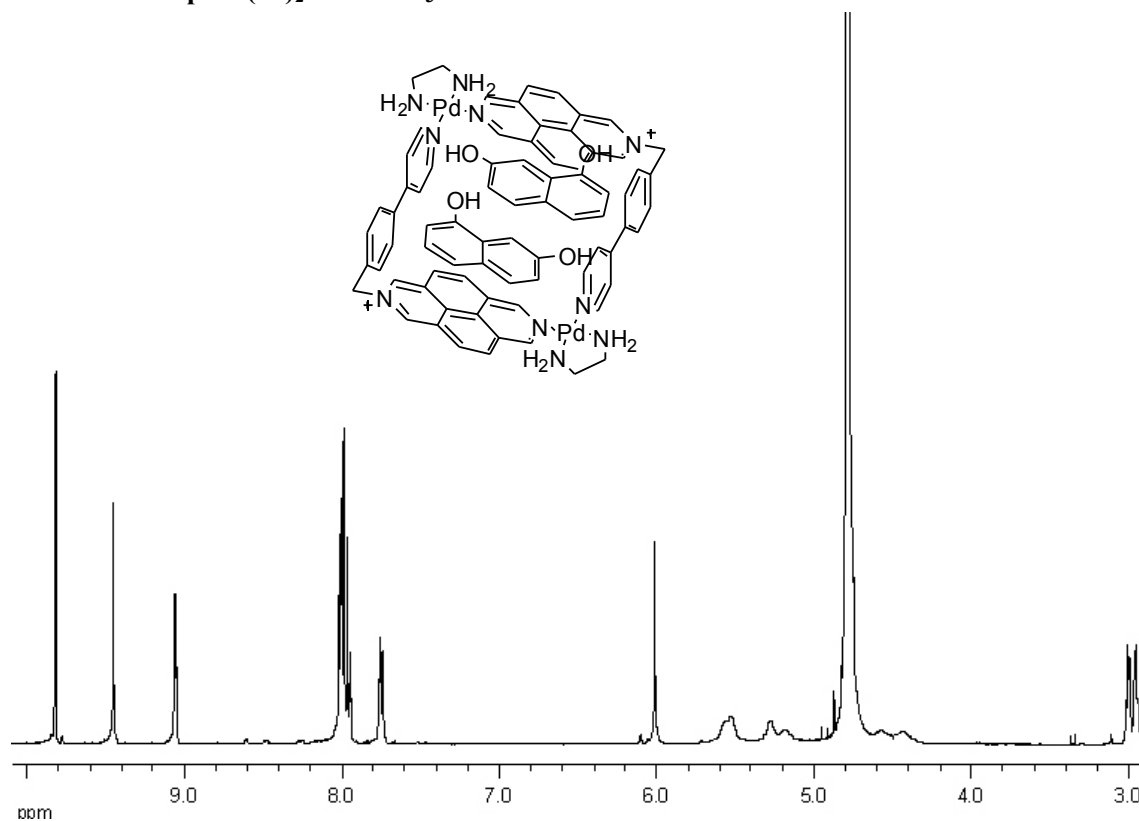


Figure S84: 1H NMR (D_2O , 500 MHz) spectrum of $(20)_2 \subset 6a \cdot 6NO_3$.

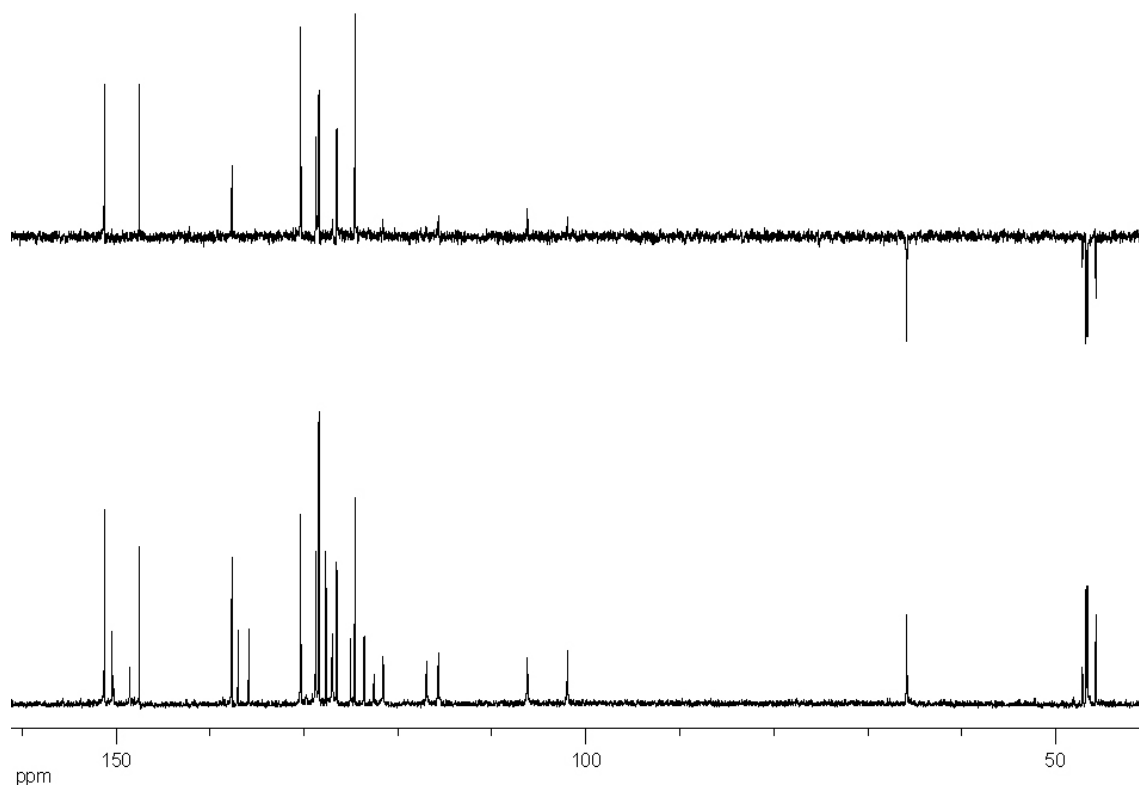


Figure S85: ^{13}C NMR (D_2O , 125 MHz) spectrum of $(20)_2 \subset 6a \cdot 6NO_3$.

Inclusion complex $(\mathbf{21})_2\subset\mathbf{6a}\cdot 6\text{NO}_3$

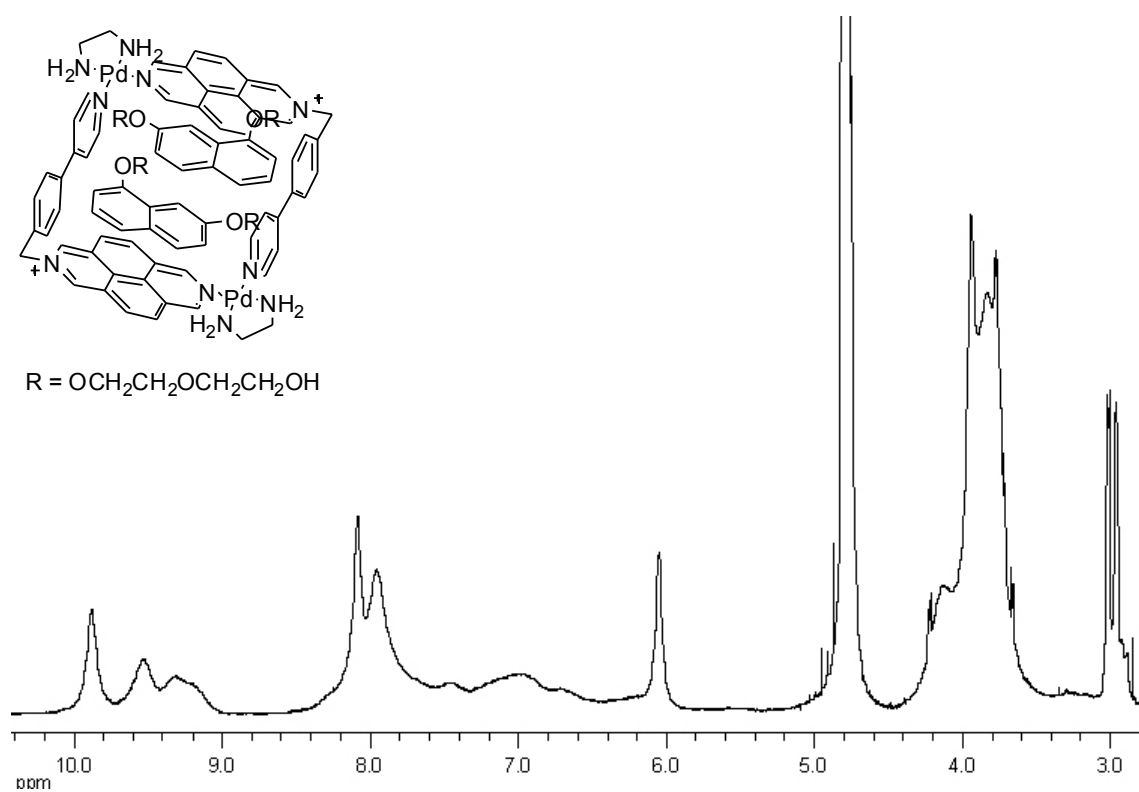


Figure S86: ^1H NMR (D₂O, 300 MHz, 298 K) spectrum of $(\mathbf{21})_2\subset\mathbf{6a}\cdot 6\text{NO}_3$.

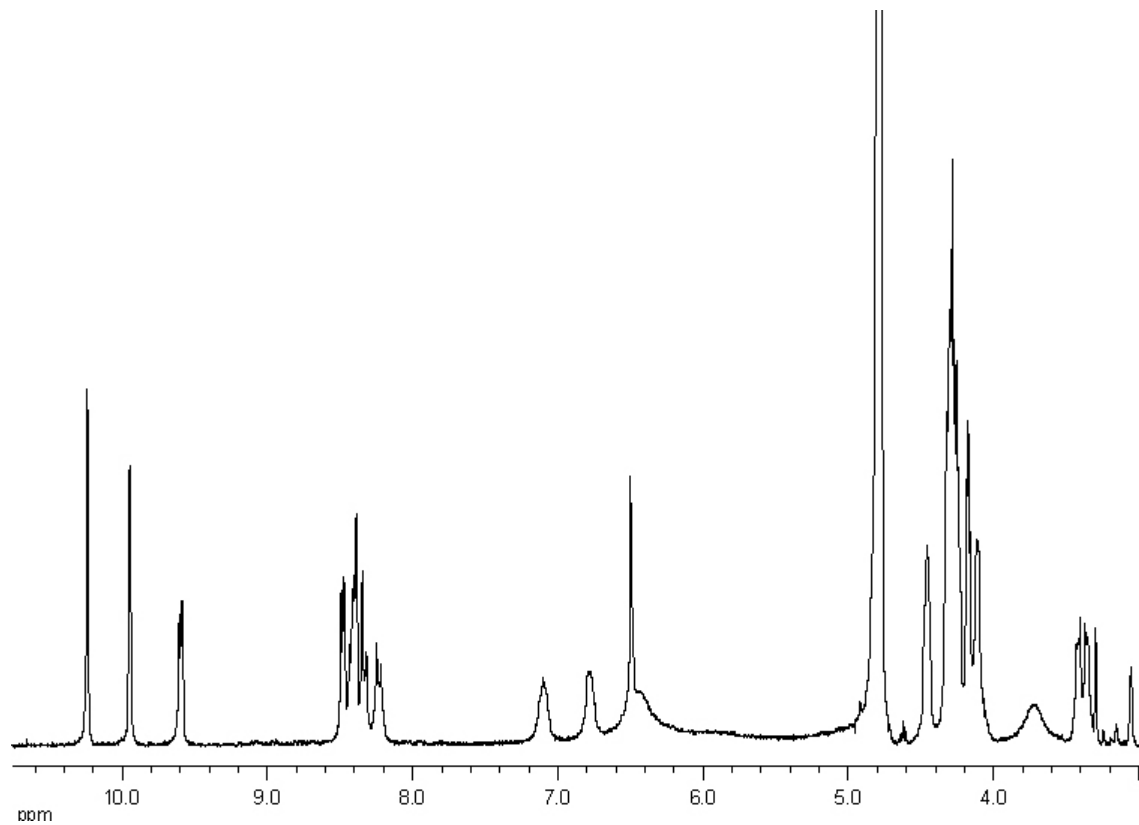


Figure S87: ^1H NMR (D₂O, 300 MHz, 338 K) spectrum of $(\mathbf{21})_2\subset\mathbf{6a}\cdot 6\text{NO}_3$.

ORTEP X-ray structures

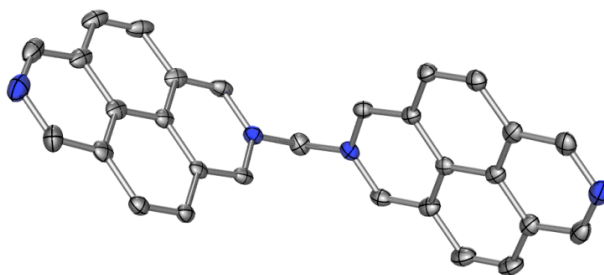


Figure S88: ORTEP drawing of the X-ray structure of ligand **1**·2PF₆. The displacement ellipsoids are drawn for 50% probability. H atoms, counterions and solvent molecules are omitted for clarity.

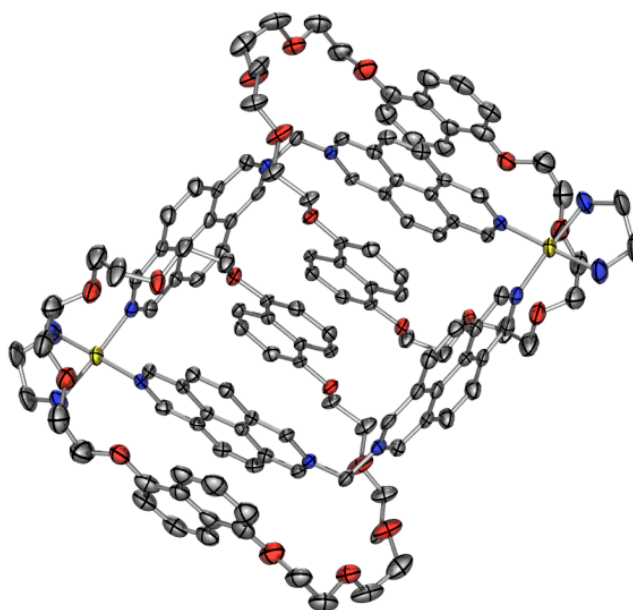


Figure S89: ORTEP representation of the X-ray structure of **3(9)**₂·4OTf·4PF₆ showing the 50% probability displacement ellipsoids. H atoms, counterions and solvent molecules are omitted for clarity.

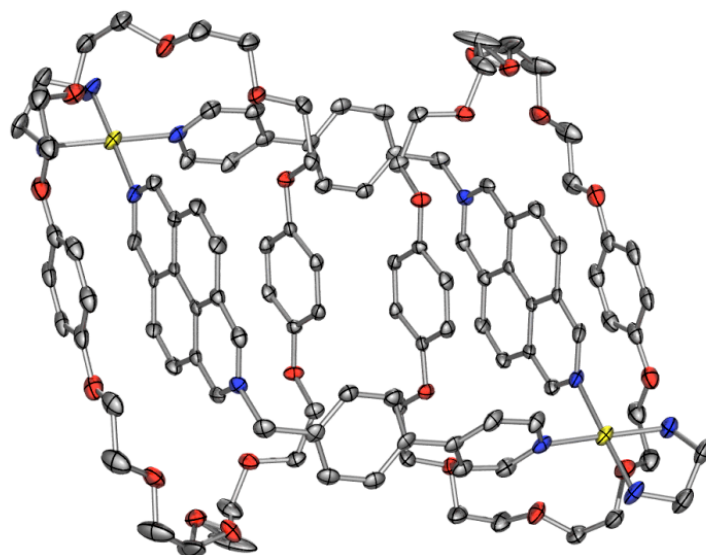


Figure S90: ORTEP drawing of the solid state structure of catenane **6a(8)**₂·6PF₆ showing the 50% probability displacement ellipsoids. H atoms, counterions and solvent molecules are omitted for clarity.

UV-Vis and fluorescence spectra

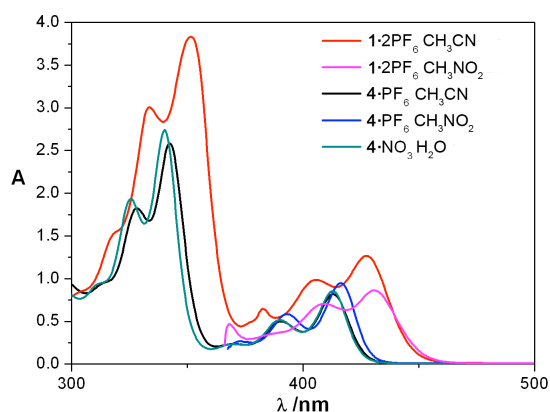


Figure S91: UV-Vis absorption spectra of ligands $1 \cdot 2PF_6$ (1.0×10^{-4} mM), $4 \cdot PF_6$ (1.0×10^{-4} M) and $4 \cdot NO_3$ (1.0×10^{-4} mM).

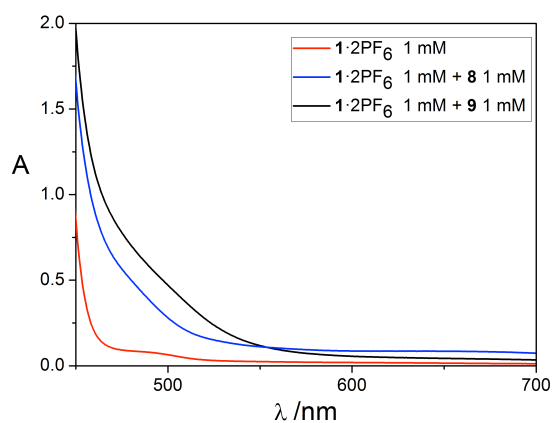


Figure S92: Charge transfer band region of the UV-Vis spectra (CH_3CN) of ligand $1 \cdot 2PF_6$ and pseudorotaxanes $1 \cdot 2PF_6 \cdot 8$ and $1 \cdot 2PF_6 \cdot 9$.

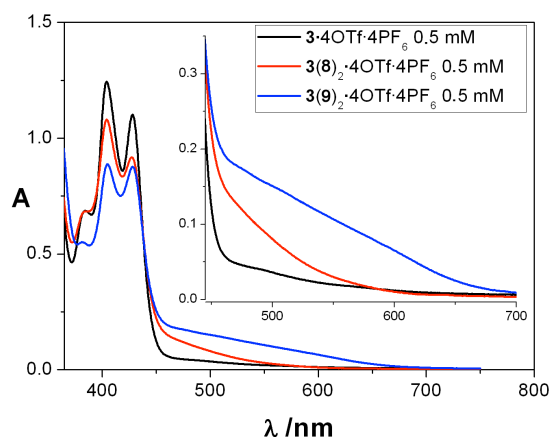


Figure S93: UV-Vis spectra (CH_3NO_2 , optical path length 1.0 mm) of metallocycle $3 \cdot 4OTf \cdot 4PF_6$ and catenanes $3(8)_2 \cdot 4OTf \cdot 4PF_6$ and $3(9)_2 \cdot 4OTf \cdot 4PF_6$. Inset: detail of the charge transfer band region (450-650 nm).

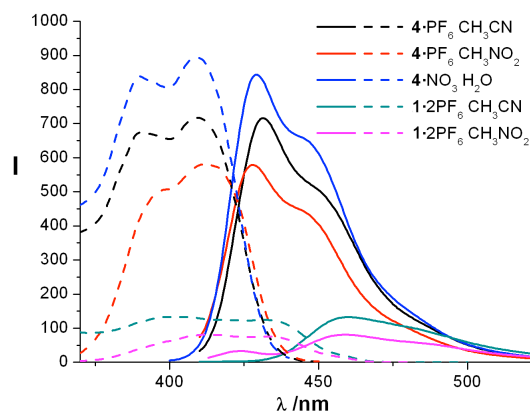


Figure S94: Fluorescence emission spectra (solid lines) and excitation spectra (dashed lines) (293 K) of ligands $1 \cdot 2PF_6$ (1.0×10^{-4} mM), $4 \cdot PF_6$ (1.0×10^{-4} mM) and $4 \cdot NO_3$ (1.0×10^{-4} mM).

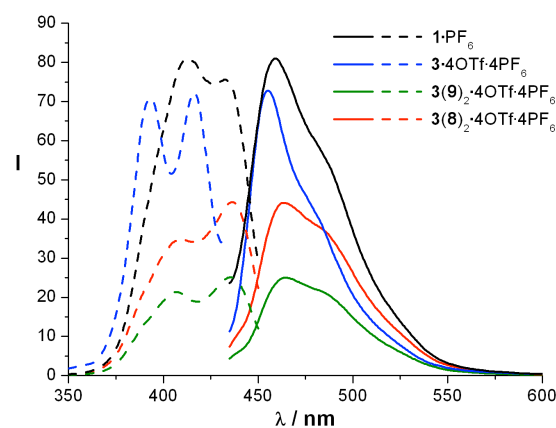


Figure S95: Fluorescence emission spectra (solid lines) and excitation spectra (dashed lines) (CH_3NO_2 , 293 K) of ligand $1 \cdot 2PF_6$ (1.0×10^{-4} M), metallocycle $3 \cdot 4OTf \cdot 4PF_6$ (5.0×10^{-5} M) and catenanes $3(8)_2 \cdot 4OTf \cdot 4PF_6$ (5.0×10^{-5} M) and $3(9)_2 \cdot 4OTf \cdot 4PF_6$ (5.0×10^{-5} M).

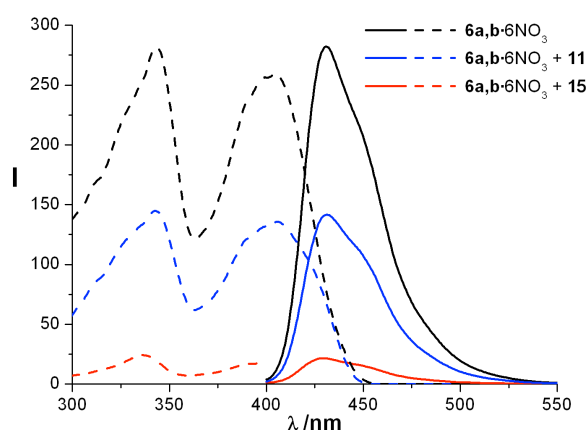


Figure S96: Fluorescence emission spectra (solid lines) and excitation spectra (dashed lines) (H_2O , 293 K) of metallocycles $6a,b \cdot NO_3$ (5.0×10^{-5} M) and inclusion complexes $(11)_2 \cdot 6a \cdot 6NO_3$ (5.0×10^{-5} mM) and $(15)_2 \cdot 6a \cdot 6NO_3$ (5.0×10^{-5} M).

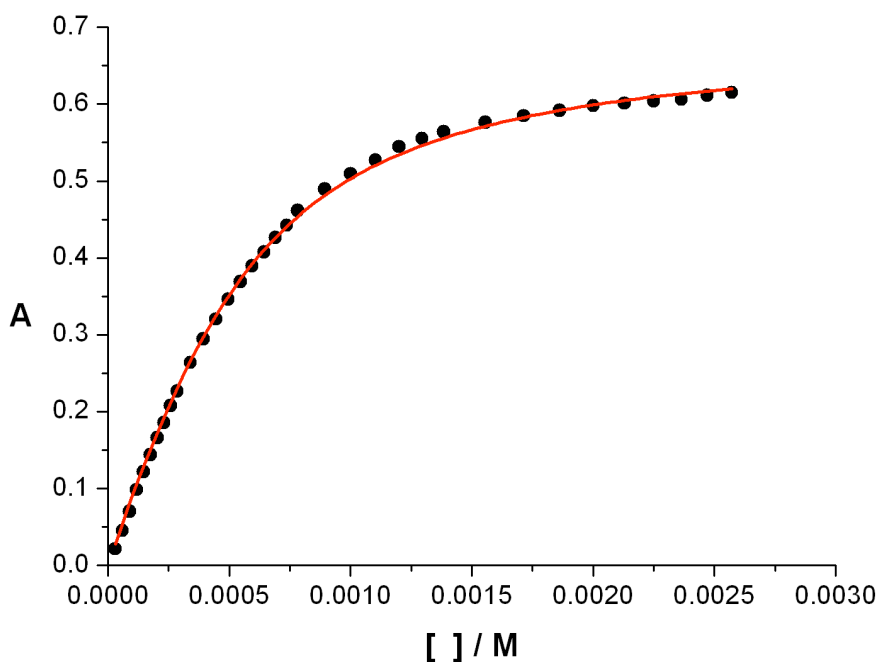
Determination of binding constant (K_a) between $1 \cdot 2PF_6$ and DN38C10 (**9**) by UV/Vis titration method.

A 0.5 mM solution of host **9** in acetonitrile and a solution of guest $1 \cdot 2PF_6$ (6 mM) and host **9** (0.5 mM) in acetonitrile were separately prepared. Then, aliquots of the guest/host solution (initially 10 μ L, then 20 μ L; 50 μ L and, finally 100 μ L) were added to the host solution (2 mL), recording the spectrum of the mixture after each addition (overall 35 points). The association constants were determined by using the nonlinear least squares fitting of the titration curves plotting the corrected A ($\epsilon_{1 \cdot 2PF_6} = 101.7 \text{ L mol}^{-1}$ at 470 nm) of the host-guest complex charge-transfer band against the concentration of the corresponding guest. The titration curve fits perfectly to the 1:1 binding isotherm ($R^2 = 0.999$).

Temperature: 298 K

$\lambda_{\text{max}} = 470 \text{ nm}$

$K_a = 4268 \pm 124 \text{ M}^{-1}$



Absorbance	[1] /M	Total added volume (μL)	Absorbance	[1] /M	Total added volume (μL)
0.0213	0.000030	10	0.4421	0.000737	280
0.0448	0.000059	20	0.4616	0.000783	300
0.0700	0.000089	30	0.4892	0.000894	350
0.0983	0.000118	40	0.5095	0.001000	400
0.1216	0.000146	50	0.5268	0.001102	450
0.1437	0.000175	60	0.5445	0.001200	500
0.1657	0.000203	70	0.5554	0.001385	550
0.1854	0.000231	80	0.5638	0.001556	600
0.2077	0.000258	90	0.5760	0.001714	700
0.2265	0.000286	100	0.5848	0.001862	800
0.2635	0.000340	120	0.5919	0.002000	900
0.2945	0.000393	140	0.5978	0.002129	1000
0.3202	0.000444	160	0.6010	0.002250	1100
0.3460	0.000495	180	0.6041	0.002364	1200
0.3691	0.000545	200	0.6059	0.002471	1300
0.3892	0.000595	220	0.6115	0.002571	1400
0.4073	0.000643	240	0.6150	0.002667	1500
0.4261	0.000690	260			

- ¹ (a) P. R. Ashton, S. E. Boyd, A. Brindle, S. J. Langford, L. Pérez-García, J. A. Preece, F. M. Raymo, N. Spencer, J. F. Stoddart, A. J. P. White, D. J. Williams, *New J. Chem.* 1999, **23**, 587; (b) J. A. Blake, G. Baum, N. R. Champness, S. S. M. Chung, P. A. Cooke, D. Fenske, A. N. Khlobystov, D. A. Lemenovskii, W.-S. Li, M. Schröder, *J. Chem. Soc., Dalton Trans.* **2000**, 4285.
- ² C. Peinador, V. Blanco, J. M. Quintela, *J. Am. Chem. Soc.* 2009, **131**, 920.
- ³ 2: (a) C. Diver, G. A. Lawrance *J. Chem. Soc., Dalton Trans.* **1988**, 931. (b) P. de Wolf, S. L. Heath; J. M. Thomas, *Inorg. Chim. Acta* **2003**, 280. **5a**: (c) H. D. K. Drew, F. W. Pinkard, G. H. Preston, W. J. Wardlaw. *Chem. Soc.* **1932**, 1895. **5b**: (d) L. V. Popov; N. N. Zheligovskaya, A. M. Grevtsev, E. A. Kharina, V. I. Spitsyn. *Seriya Khimicheskaya* 1977, **7**, 1677. (e) M. Fujita, J. Yazaki, K. Ogura, *Chem. Lett.* **1991**, 1031.
- ⁴ V. Blanco, D. Abella, E. Pía, C. Platas-Iglesias, C. Peinador, J. M. Quintela, *Inorg. Chem.* 2009, **48**, 4098.
- ⁵ D. B. Amabilino, P. L. Anelli, P. R. Ashton, G. R. Brown, E. Córdova, L. A. Godínez, W. Hayes, A. E. Kaifer, D. Philp, A. M. Z. Slawin, N. Spencer, J. F. Stoddart, M. S. Tolley, D. J. Williams, *J. Am. Chem. Soc.* 1995, **117**, 11142.
- ⁶ D. B. Amabilino, P. R. Ashton, V. Balzani, S. E. Boyd, A. Credi, J. Y. Lee, S. Menzer, J. F. Stoddart, M. Venturi, D. J. Williams, *J. Am. Chem. Soc.* 1998, **120**, 4295.
- ⁷ P. R. Anelli, P. R. Ashton, R. Ballardini, V. Balzani, M. Delgado, M. T. Gandolfi, T. T. Goodnow, A. E. Kaifer, D. Philp, M. Pietraszkiewicz, L. Prodi, M. V. Reddington, A. M. Z. Slawin, N. Spencer, J. F. Stoddart, C. Vicent, D. J. Williams, *J. Am. Chem. Soc.* 1992, **114**, 193.
- ⁸
- ⁹ M. Asakawa, P. R. Ashton, S. E. Boyd, C. L. Brown, R. E. Gillard, O. Kocian, F. M. Raymo, J. F. Stoddart, M. S. Tolley, A. J. P. White, D. J. Williams, *J. Org. Chem.* 1997, **62**, 26.
- ¹⁰ B. Czech, A. Czech, R. A. Bartsch, *J. Heterocycl. Chem.* 1984, **21**, 341.
- ¹¹ V. Blanco, A. Gutiérrez, C. Platas-Iglesias, C. Peinador, J. M. Quintela, *J. Org. Chem.* 2009, **74**, 6577.
- ¹² L. G. Longworth, *J. Phys. Chem.* 1960, **64**, 1914.
- ¹³ T. Megyes, H. Jude, T. Grósz, I. Bakó, R. Radnai, G. Tárkányi, G. Pálkás, P. J. Stang, *J. Am. Chem. Soc.* 2005, **127**, 10731.
- ¹⁴ *SHELX-97*, Release 97-2; G. M. Sheldrick, University of Göttingen, Germany, 1997.