

Mixed-metal (Li, Al) amidoborane: synthesis and hydrogen storage properties

Guanglin Xia,^{b,a} Yingbin Tan,^a Xiaowei Chen,^a Zaiping Guo,^{*b,c} Huakun Liu,^b and Xuebin Yu,^{*a}

- a. Department of Materials Science, Fudan University, Shanghai 200433, China*
- b. Institute for Superconducting and Electronic Materials, University of Wollongong, North Wollongong, NSW 2522, Australia*
- c. School of Mechanical, Materials & Mechatronics Engineering, University of Wollongong, NSW 2522, Australia*

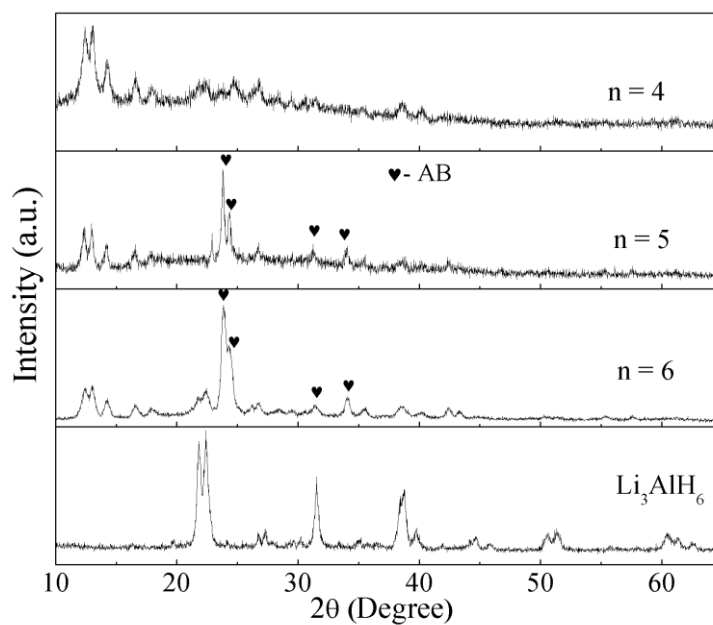


Figure S1. Powder XRD patterns of the as-prepared $\text{Li}_3\text{AlH}_6 - n \text{ AB}$ ($n = 4, 5,$ and 6) composites, with pristine Li_3AlH_6 included for comparison.

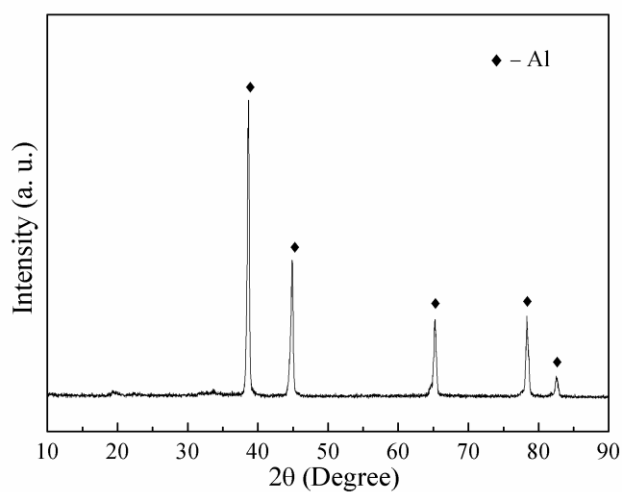


Figure S2. Powder XRD pattern for a hand-milled mixture of LiAlH_4 and AB in a molar ratio of 1:4.

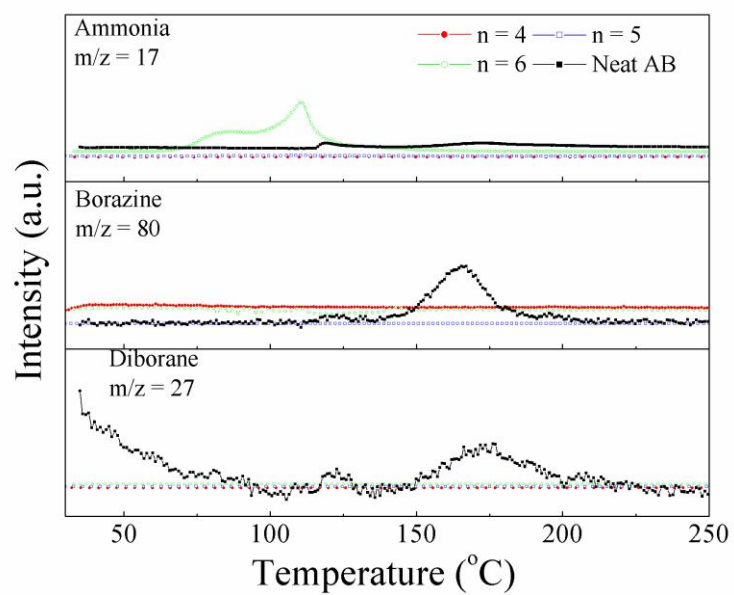


Figure S3. MS spectra for the by-products from the post-milled $\text{Li}_3\text{AlH}_6 - n \text{ AB}$ ($n = 4, 5,$ and 6) composites, including neat AB for comparison.

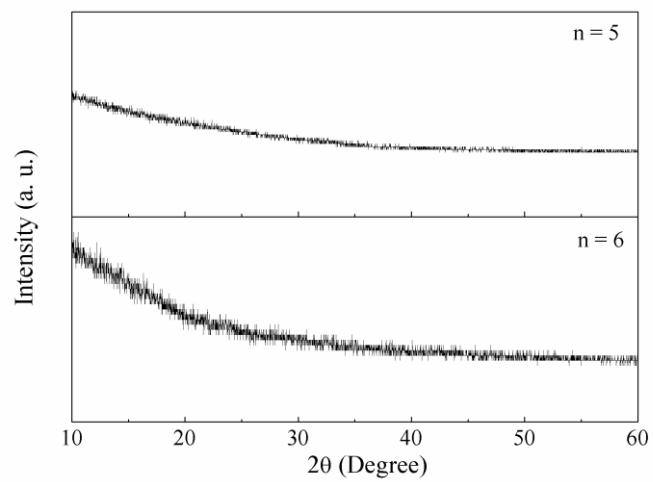


Figure S4. Powder XRD patterns of the as-prepared $\text{Li}_3\text{AlH}_6 - n \text{ AB}$ ($n = 5$ and 6) composites after heating to $250\text{ }^\circ\text{C}$

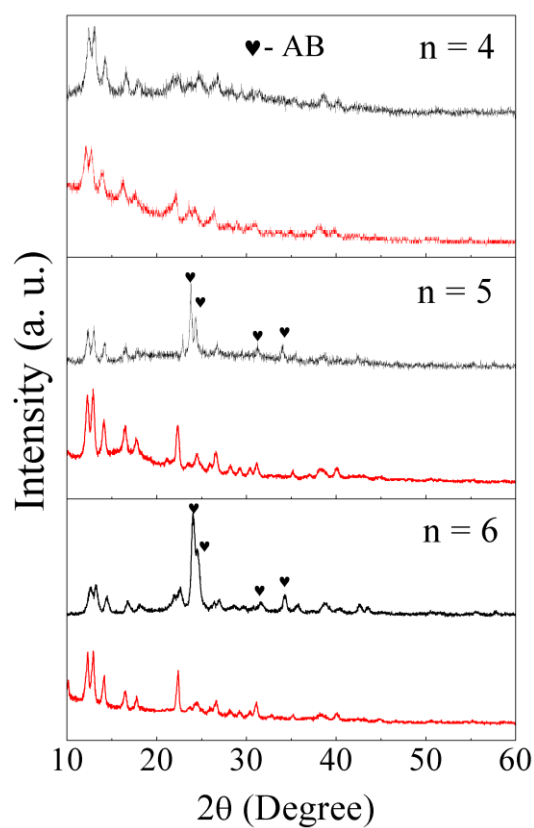


Figure S5. Powder XRD patterns for the as-prepared $\text{Li}_3\text{AlH}_6 - n \text{ AB}$ ($n = 4, 5,$ and 6) composites (black lines) and the related products cooled down to room temperature after heating to 90°C (red lines). For clarity, the intensity of the post-heated products was magnified 2, 2.5, and 3.5 times for $n = 4, 5,$ and $6,$ respectively.

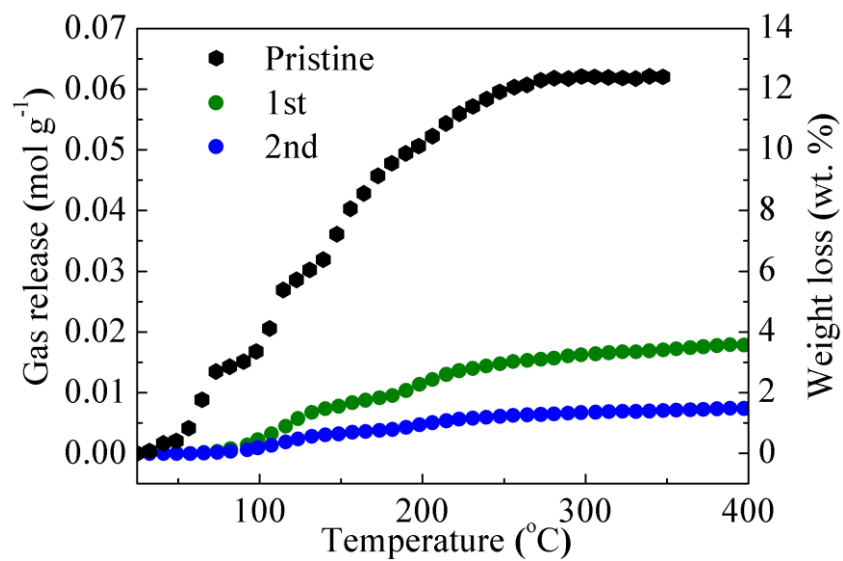


Figure S6. TPD results for the decomposition of the products after 1st and 2nd hydrogenation of the spent fuels from the post-milled $\text{Li}_3\text{AlH}_6 - 6\text{AB}$, including pristine $\text{Li}_3\text{AlH}_6 - 6\text{AB}$ for comparison.

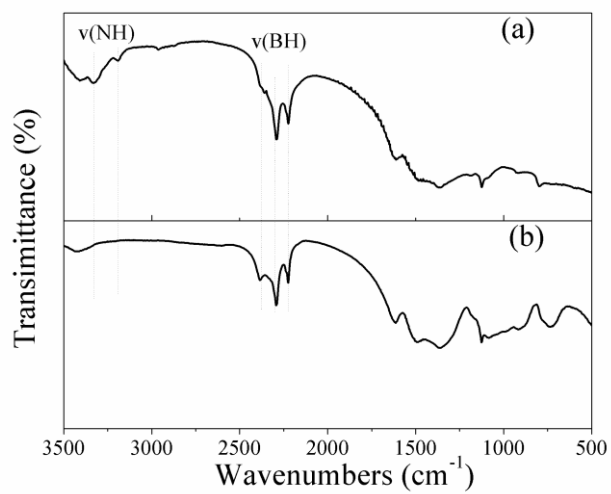


Figure S7. FTIR spectra of the regenerated $\text{Li}_3\text{AlH}_6 - 4\text{AB}$ (a) and the dehydrogenated products after heating to 250 °C (b).