

Electronic Supplementary Information

Surface modification of nano-silica with amides and imides for use in polyester nanocomposites

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Calculation of Number of reacted silanol groups per square nanometer on silica surface

Percentage organic content on silica surface after APS modification is 11.8 wt-%.

Silica : Degussa, Aerosil 200[®], diameter = 12 nm

Specific surface area of silica particle = 200 m²/g (from the manufacturer)

 Surface Area of a sphere is given by: $4\pi r^2$

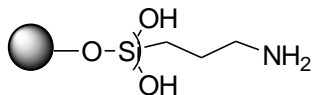
∴ surface area of silica with 12 nm diameter is:

$$4 \times 3.14 \times (6 \times 10^{-9})^2 = 4.52 \times 10^{-16} \text{ m}^2$$

Since Mass = surface Area / specific surface area

$$\begin{aligned} \therefore \text{mass of the given silica} &= 4.52 \times 10^{-16} \text{ m}^2 / 200 \text{ m}^2/\text{g} \\ &= 2.26 \times 10^{-18} \text{ g} \end{aligned}$$

Considering that all the organic content on the silica degrades, including the amino-propyl group and the two-hydroxy groups of the attached APS, the molecular weight of the organic content coming off is **92.11**



Molar mass of organic content on modified silica = molecular weight of the organic content divided by Avogadro's constant

$$\text{Molar mass of the organic content on modified silica} = 92.11 / 6.023 \times 10^{23} = 1.53 \times 10^{-22} \text{ g}$$

$$\text{Total weight of organic content on silica} = \text{Mass of Silica} \times \left\{ \frac{\% \text{ organic content on silica} / 100}{1 - \% \text{ organic content on silica} / 100} \right\}$$

$$\text{Total weight of organic content on silica} = 2.26 \times 10^{-18} \times \left\{ \frac{11.8/100}{1-11.8/100} \right\}$$

$$\text{Total weight of organic content on silica} = 2.26 \times 10^{-18} \times 0.13$$

$$\text{Total weight of organic on silica} = 2.94 \times 10^{-19} \text{ g}$$

Dividing this by the molar mass of the organic content on modified silica (1.53×10^{-22} g calculated from TGA) gives the total number of reacted silanol groups present on silica surface.

$$\begin{aligned} \therefore \text{Total number of reacted silanol groups on silica surface} &= 2.94 \times 10^{-19} / 1.53 \times 10^{-22} \\ &= 1921 \end{aligned}$$

Number of reacted silanol groups per square nanometer on silica surface = Total no. of
reacted silanol groups present on silica surface / surface area of silica particle

Number of reacted silanol groups per square nanometer on silica surface =

$$1921 / 452 = \mathbf{4.2 \text{ reacted OH (silanol) groups / nm}^2}$$

Fig. S1 ^{13}C NMR of SiAP

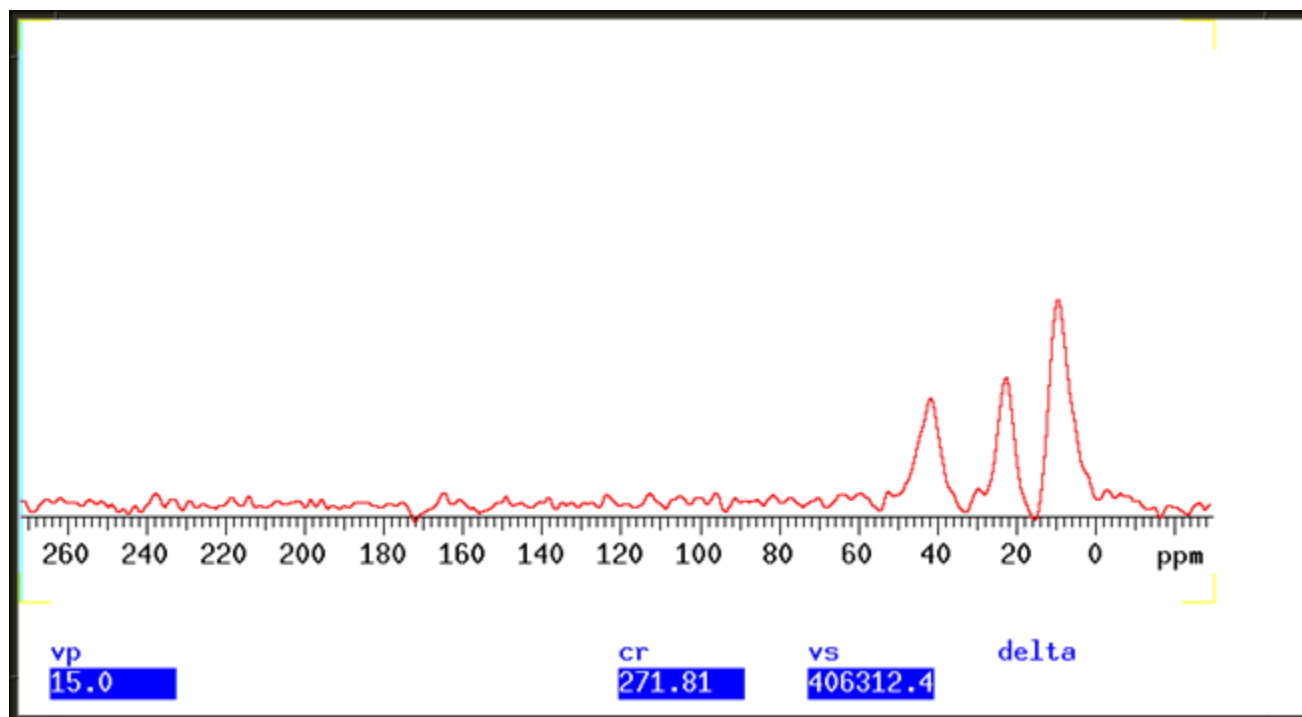
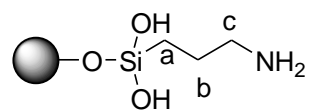


Fig. S2 ^{13}C NMR of SiAc

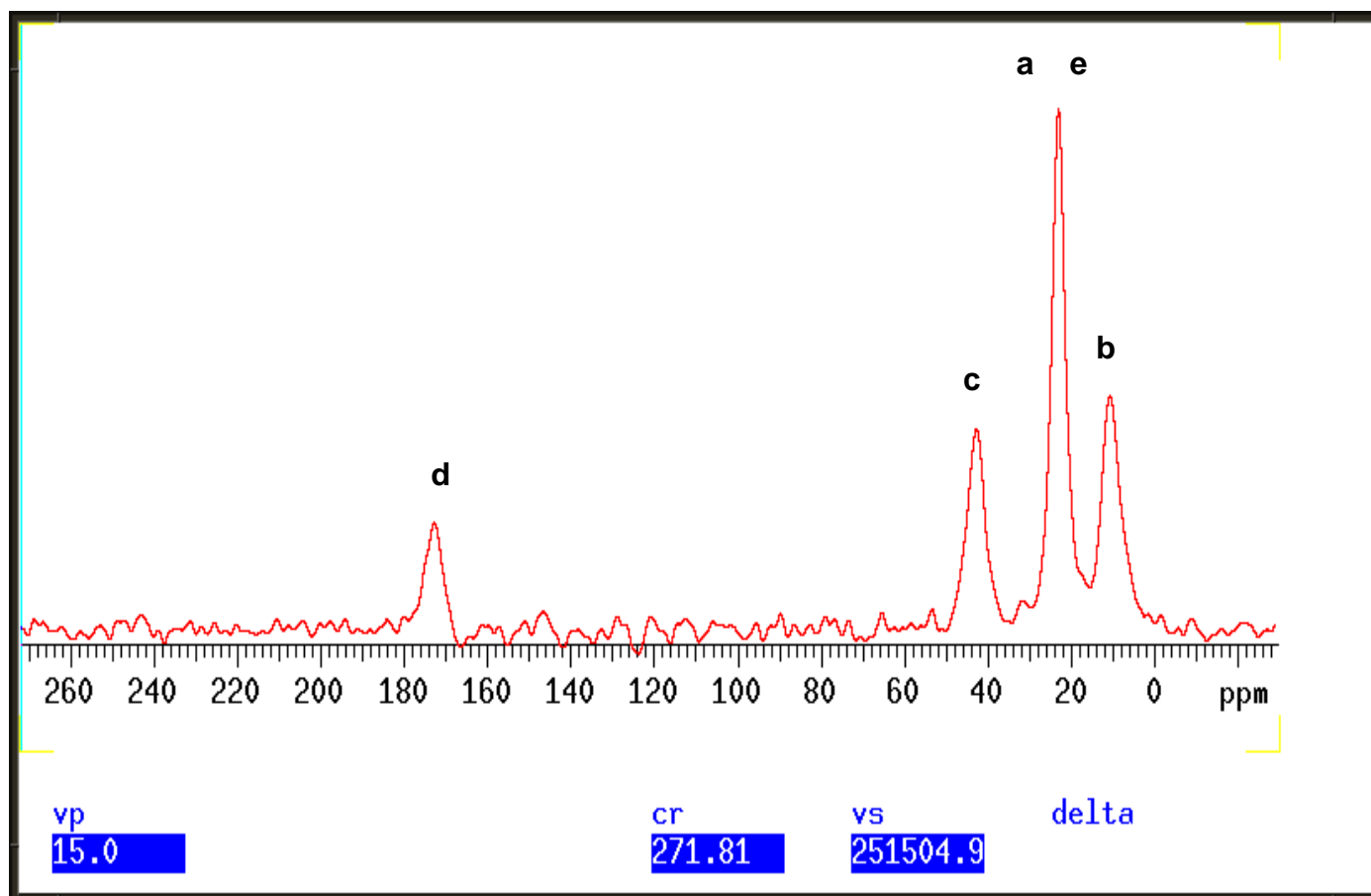
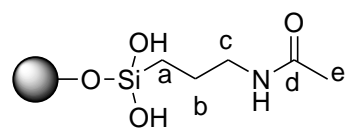


Fig. S3 ^{13}C NMR of SiSA

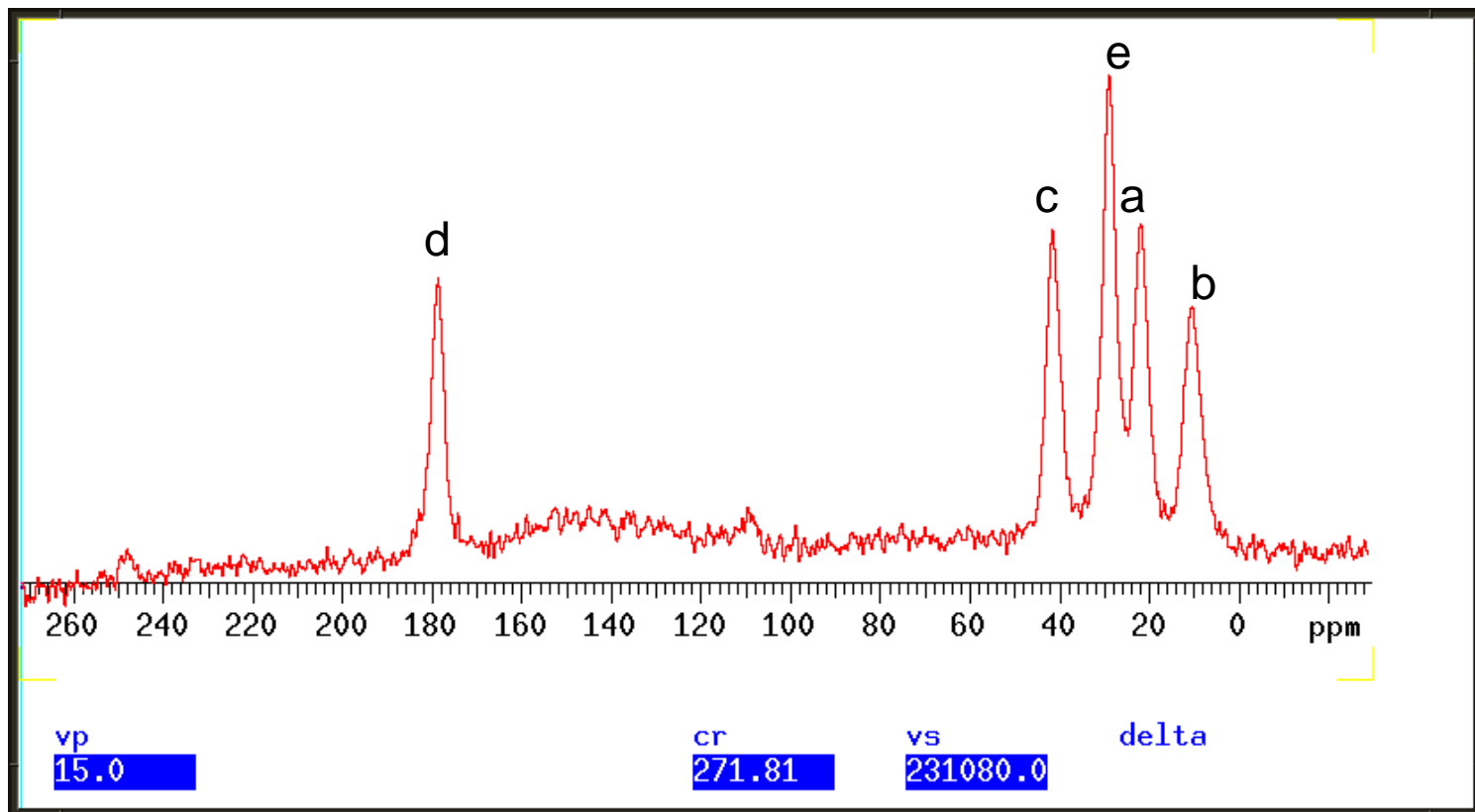
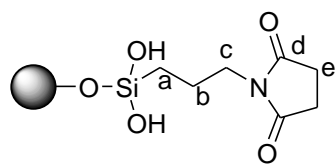


Fig. S4 ^{13}C NMR of SiMA

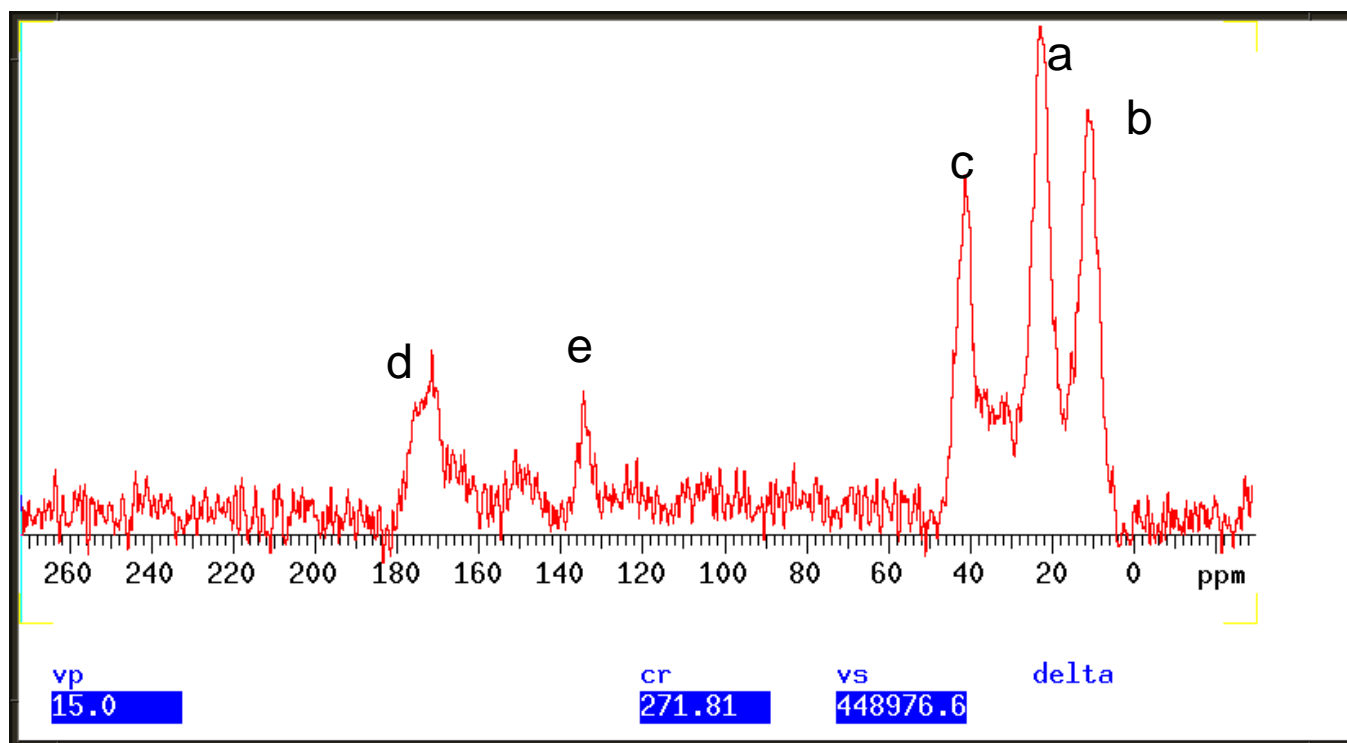
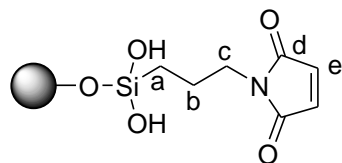


Fig. S5 ^{13}C NMR of SiPA

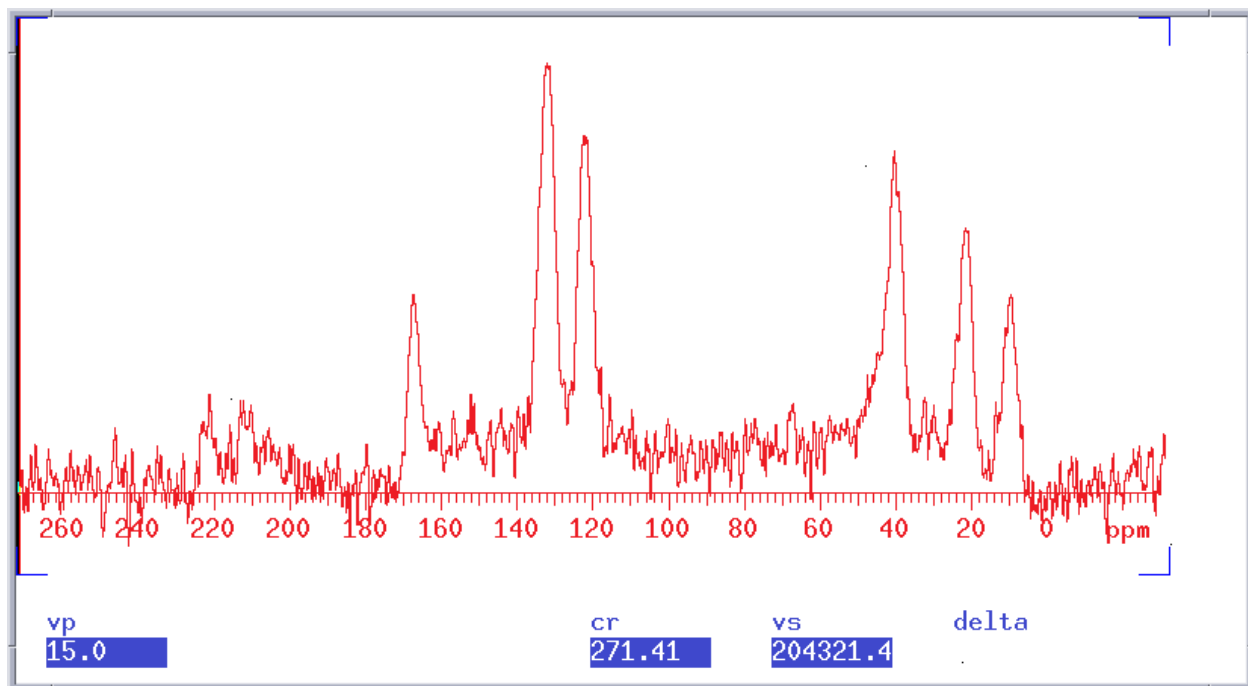
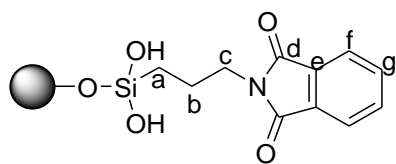


Fig. S6 ^{13}C NMR of SiTPA

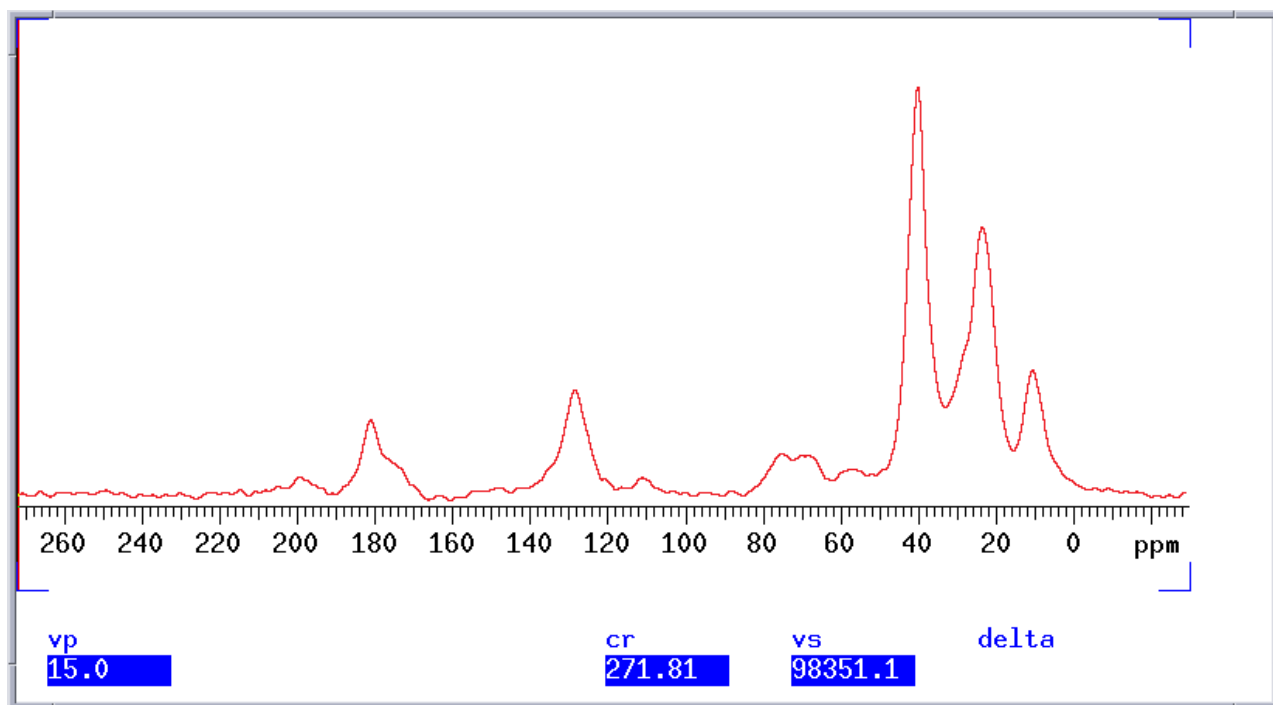
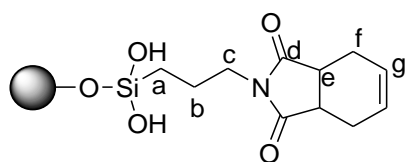


Fig. S7 FT-IR spectra of (1) untreated nano-SiO₂ and after modification with (2) APS, (3) APS/acetic anhydride, (4) APS/succinnic anhydride, (5) APS/maleic anhydride, (6) APS/phthalic anhydride, (7) APS/tetrahydrophthalic anhydride.

