

High Photoelectrocatalytic Performance of MoS₂/SiC Hybrid Structure for Hydrogen Evolution Reaction

Xiaoning Guoa, Chengmeng Chena, Yunwei Wanga,b, Xili Tonga, Guoqiang Jina,
and Xiang-Yun Guoa,*

^a State Key Laboratory of Coal Conversion, Institute of Coal Chemistry, Taiyuan
030001, PR China

^b University of the Chinese Academy of Sciences, Beijing 100039, PR China

*Corresponding author:

Fax/Tel: +86-351 4065282

Email: xyguo@sxicc.ac.cn (X. Y. Guo)

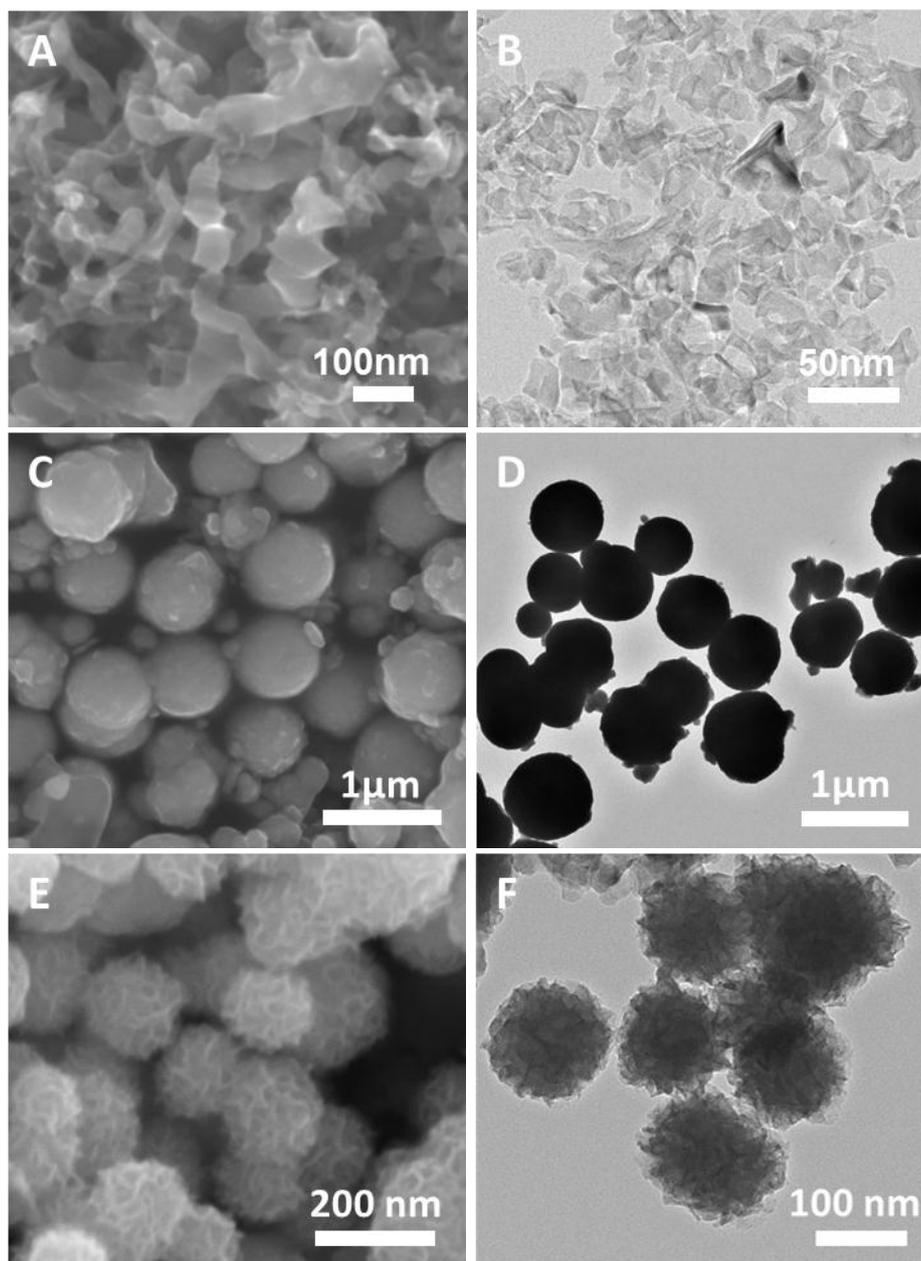


Fig. S1. SEM and TEM images of SiC nanosheets before (A) and after (B) etching, free MoS₂ (C, D) and the MoS₂/SiC hybrid (E, F). The free MoS₂ are 3D-like spherical particles and their surfaces are almost smooth, whereas the MoS₂/SiC hybrid is composed of uniformly flower-like spheres assembled by folded flakes with thickness of about several nanometers.

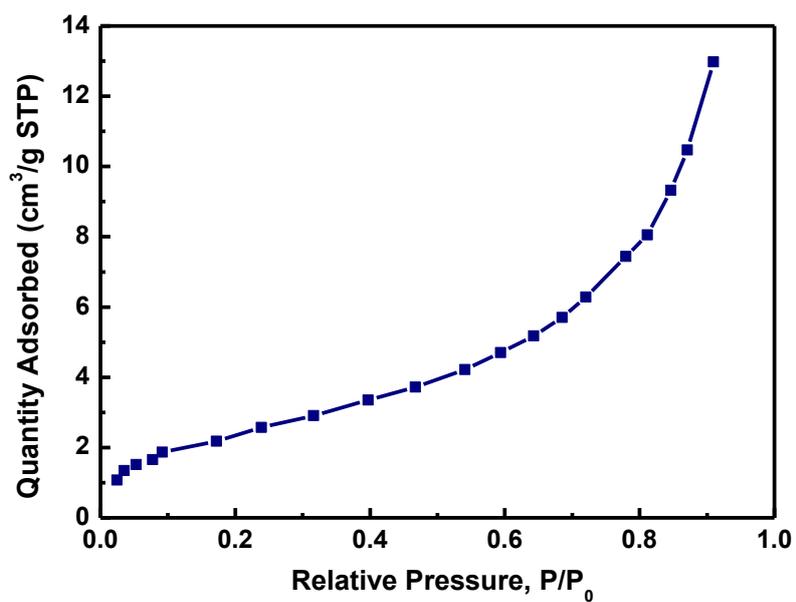


Fig. S2. Adsorption isotherm of p-xylene vapor on the MoS₂/SiC hybrid. The equilibrium adsorption value is only 13.0 cm³/g, which is far less than the value for water vapor (104.6 cm³/g), indicating that the nanopores in the hybrid catalyst are strongly favorable to the adsorption of H₂O molecules.

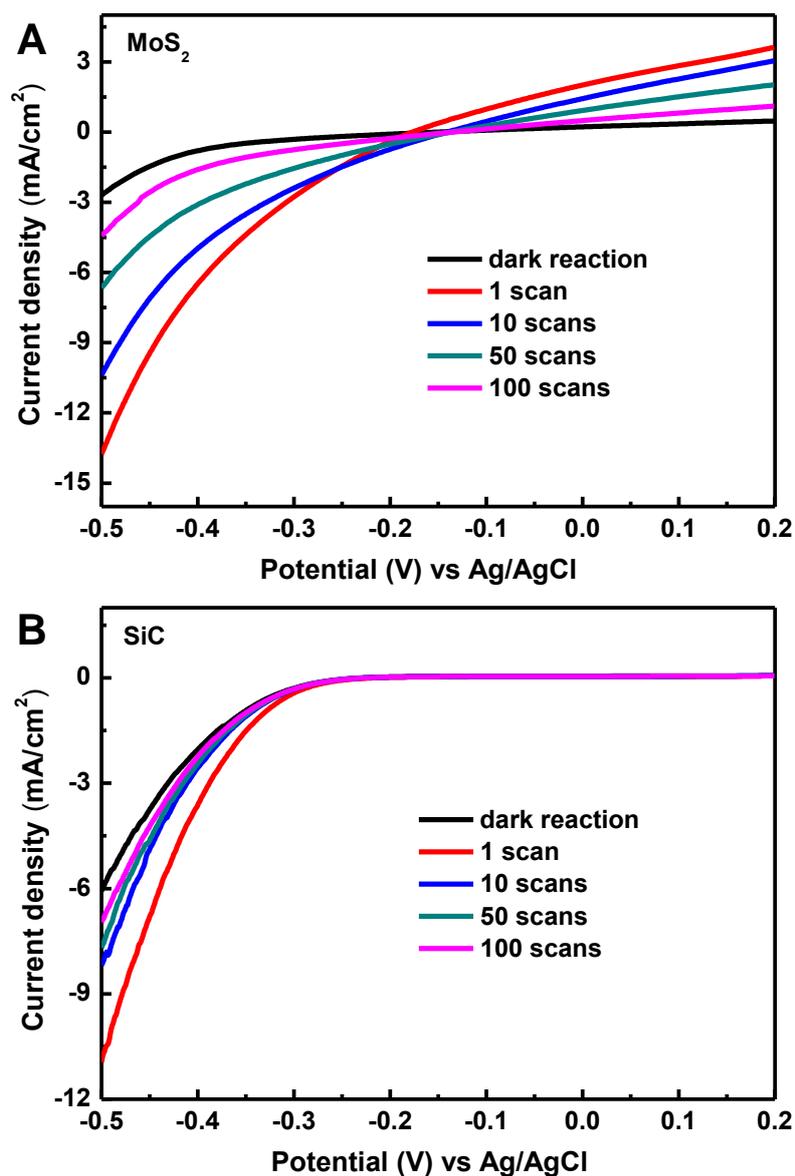


Fig. S3. The photoelectric conversion efficiency and catalytic stability of free MoS₂ and SiC. Both MoS₂ and SiC can effectively response visible light and generate photocurrent. However, MoS₂ and SiC showed a high overpotential (η) of about 0.2 and 0.32 V at the first scan for HER, suggesting that their catalytic activities are less than the MoS₂/SiC (0.04V). Meanwhile, either MoS₂ or SiC shows poor catalytic stability, and the current density rapidly decreases with the scan times.

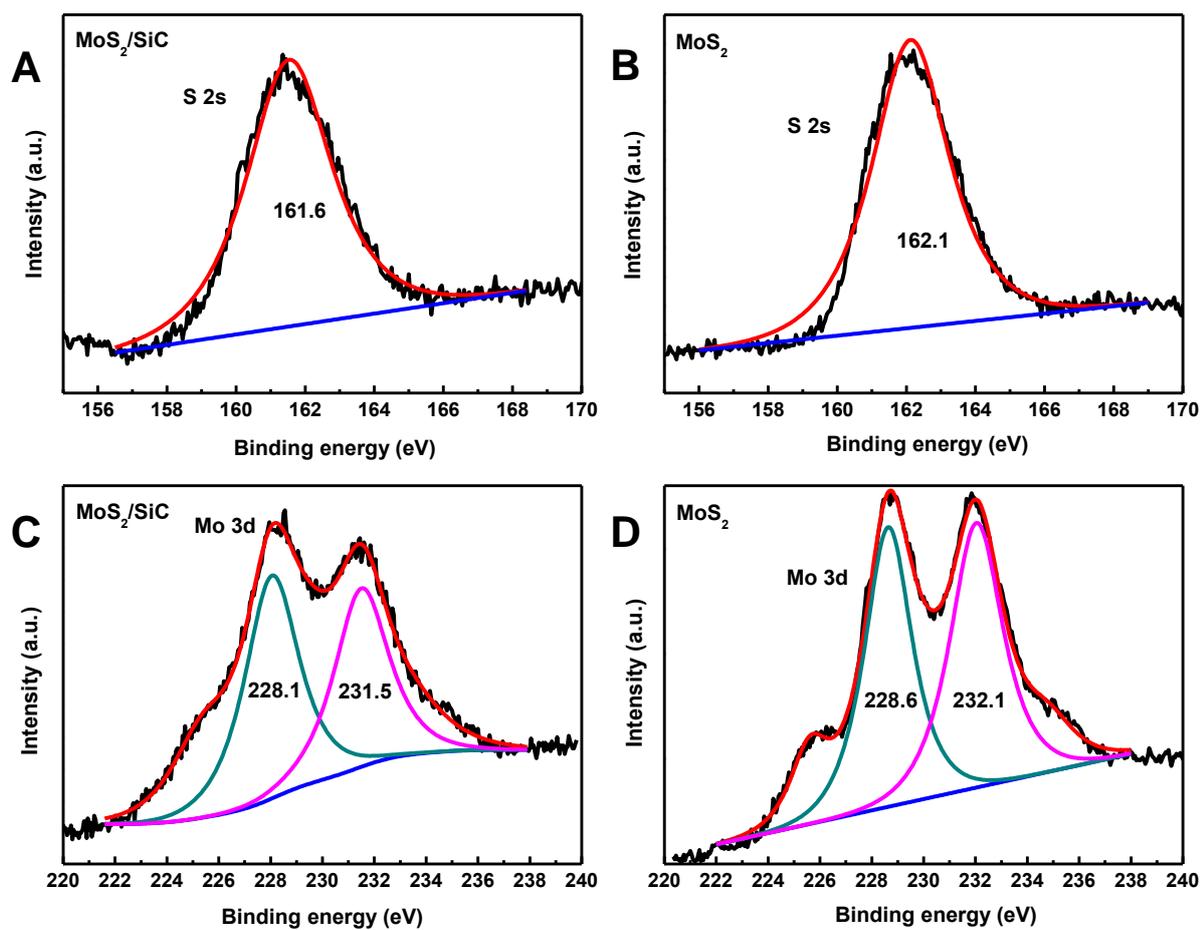


Fig. S4. XPS spectra of S_{2s} (A and B) and Mo_{3d} (C and D) levels of the MoS_2/SiC hybrid and free MoS_2 particles. Both the S_{2s} and Mo_{3d} BE values of MoS_2/SiC are lower than those of free MoS_2 , suggesting that electron cloud densities around those atoms increase.

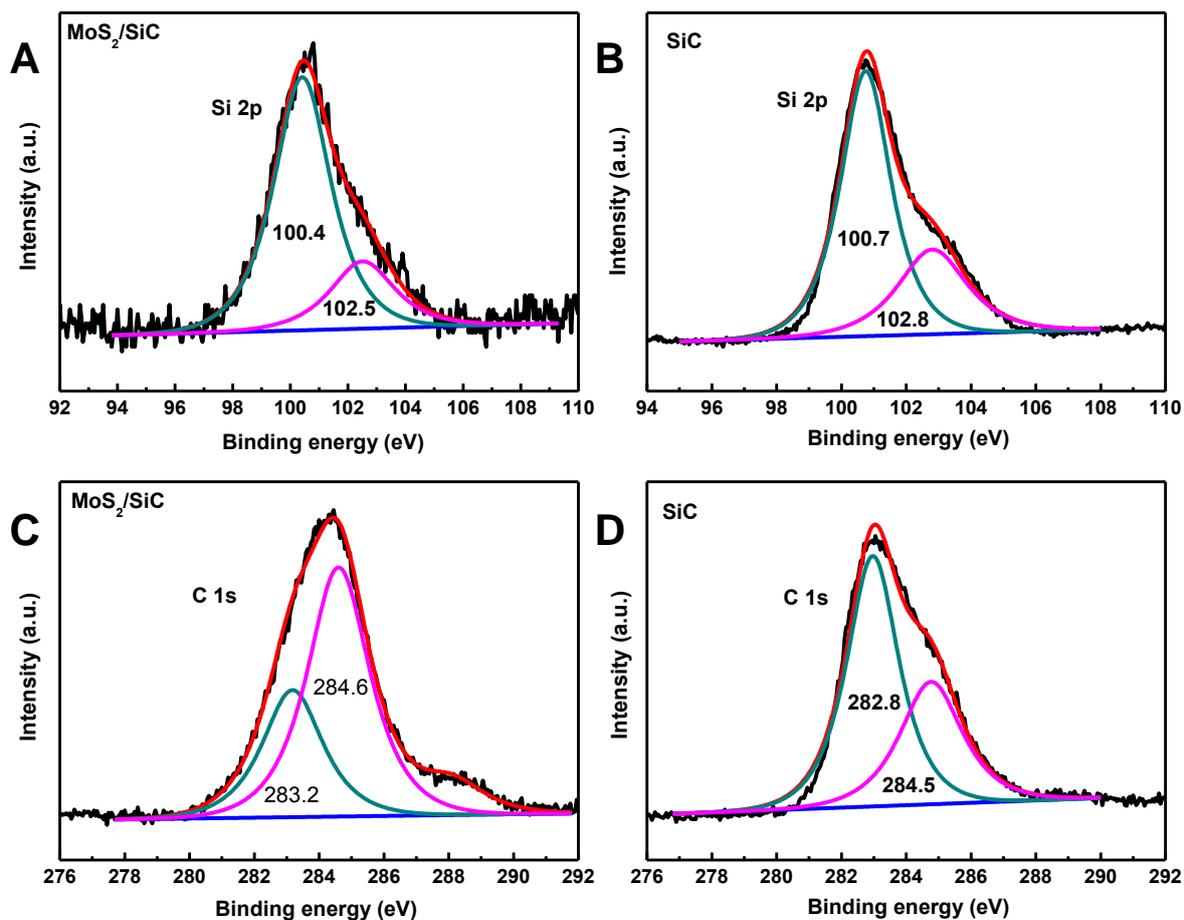


Fig. S5. XPS spectra of Si_{2p} (A and B) and C_{1s} (C and D) levels of the MoS₂/SiC hybrid and free SiC. The Si_{2p} BE value of MoS₂/SiC is lower than that of free SiC, suggesting that the electron cloud density around Si increases. While the C_{1s} peaks in the MoS₂/SiC shift to higher binding energies comparing with free SiC, indicating that the surrounding electrons of C have transferred to other atoms.