

Supporting Information

Is $\text{Li}_4\text{Ti}_5\text{O}_{12}$ a solid-electrolyte-interphase-free electrode material in Li-ion batteries? Reactivity between $\text{Li}_4\text{Ti}_5\text{O}_{12}$ electrode and electrolyte

**Min-Sang Song,^a Ryoung-Hee Kim,^a Seung-Wook Baek,^a Kug-Seung Lee,^b
Kyusung Park^{*c} and Anass Benayad^{*d}**

^a Energy Storage Group, Samsung Advanced Institute of Technology (SAIT), San 14, Yongin 446-712, South Korea.

^b Pohang Accelerator Laboratory, Pohang 790-784, South Korea.

^c Texas Materials Institute, The University of Texas at Austin, Austin, Texas 78712, United States. E-mail: kspark37@gmail.com

^d Analytical Sciences Group, SAIT, San 14, Yongin 446-712, South Korea. E-mail: an_be2001@yahoo.fr

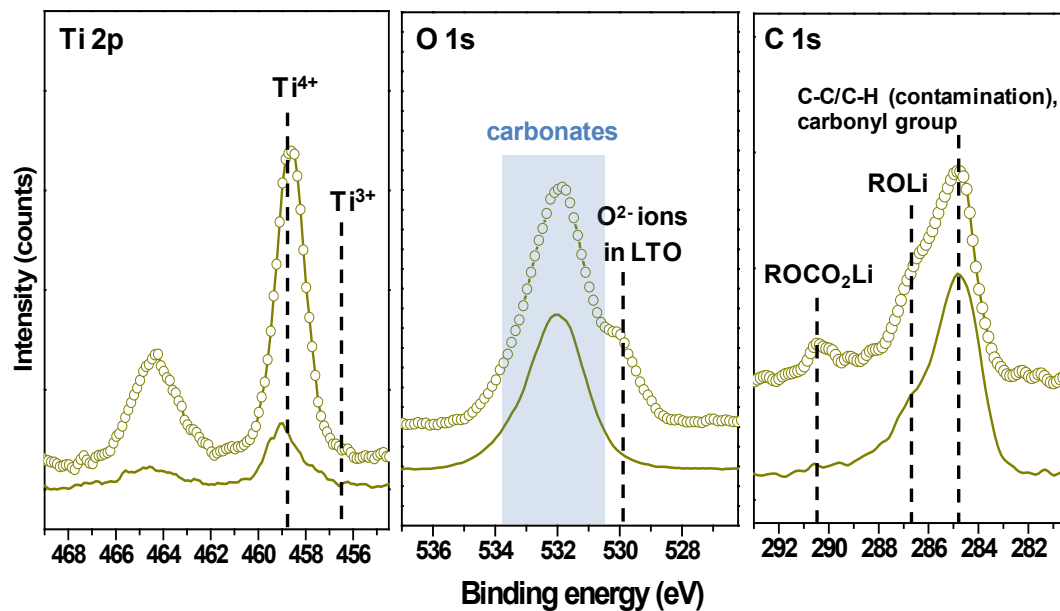


Fig. S1 Ti 2p, O 1s and C 1s XPS spectra of the carbon-free $\text{Li}_4\text{Ti}_5\text{O}_{12}$ electrode after 50 cycles: before and after mechanical scratching corresponding to solid and empty circles, respectively.

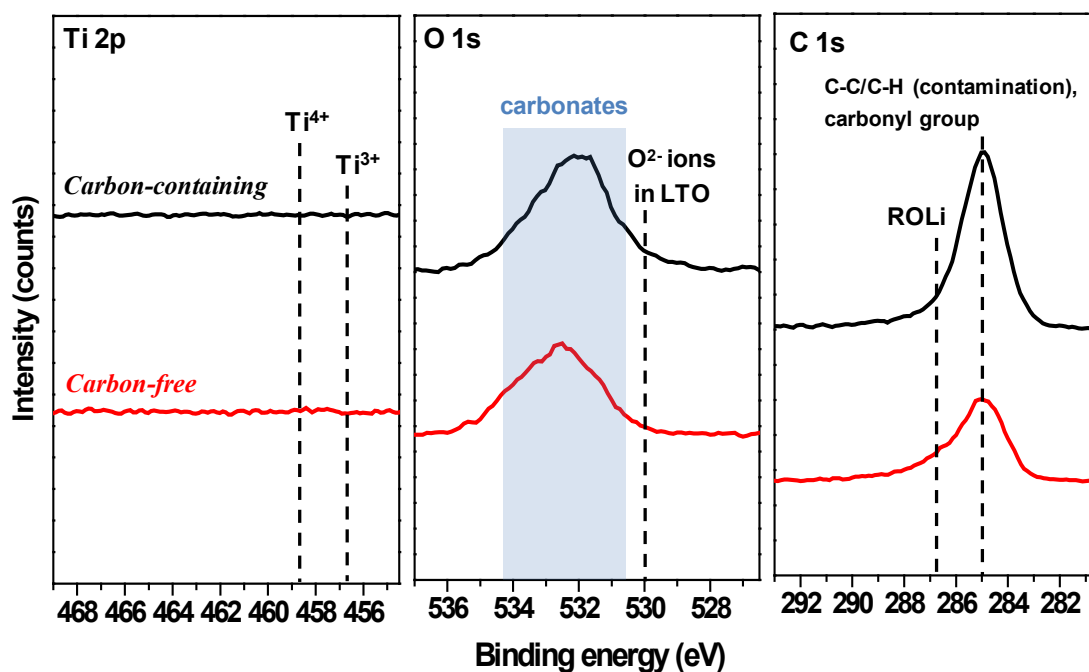


Fig. S2 Ti 2p, O 1s and C 1s XPS spectra of the carbon-free $Li_4Ti_5O_{12}$ electrode (red) and the carbon-containing electrode (black) after 280 cycles at 60°C.

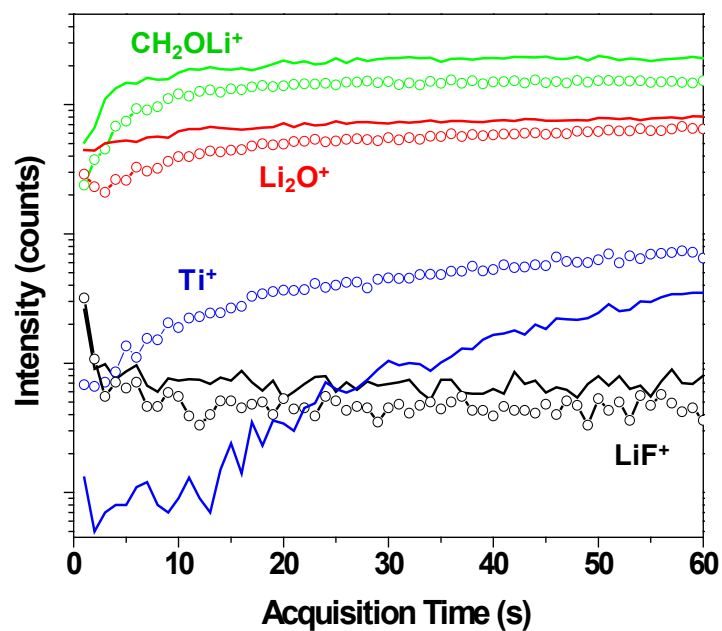


Fig. S3 SIMS depth profiles of $\text{Li}_4\text{Ti}_5\text{O}_{12}$ electrodes after 280 cycles at 60°C . Empty circles correspond to the carbon-free electrode and solid line for the carbon-containing electrode.

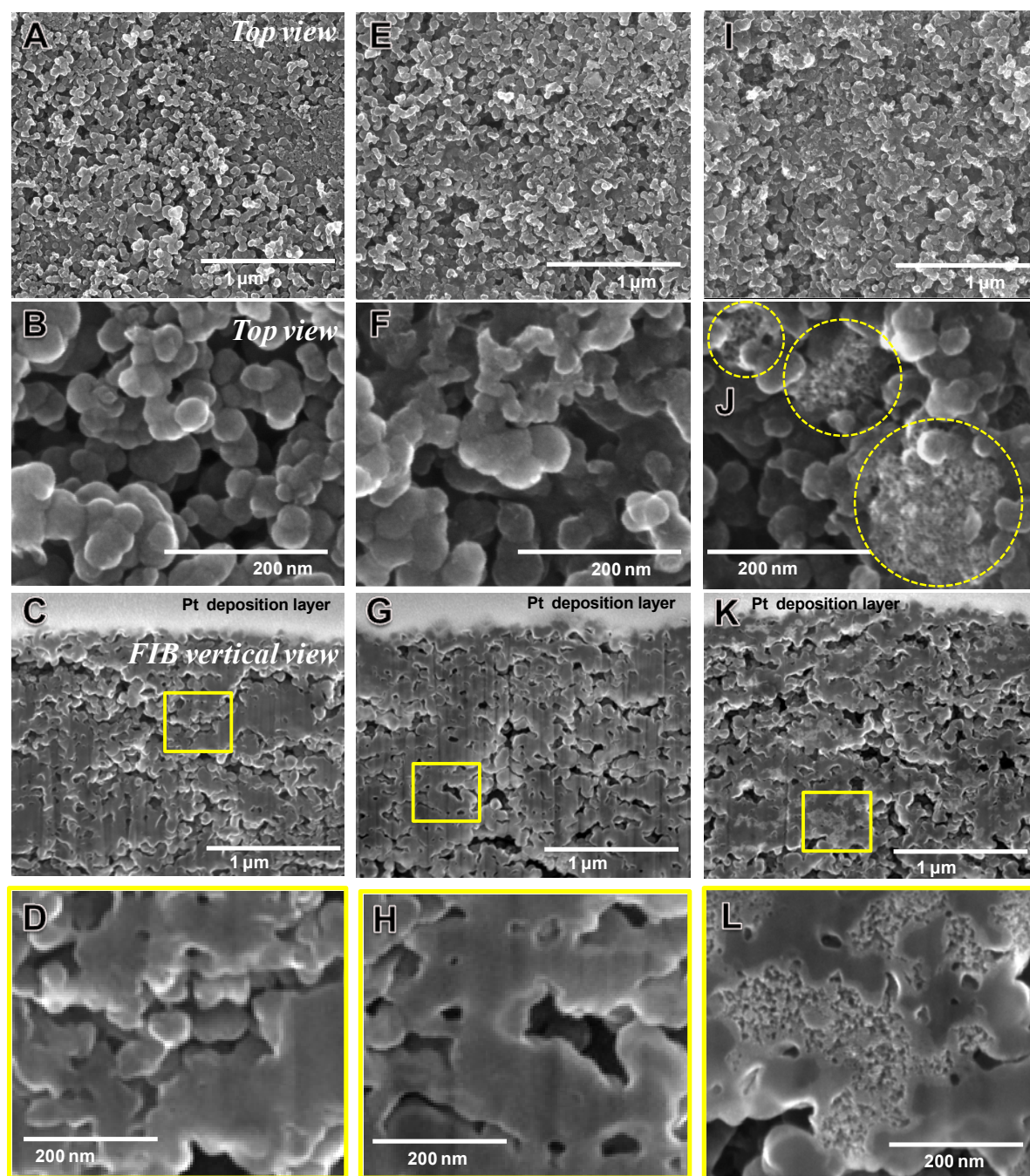


Fig. S4 SEM (top-view) and FIB cross-section images of the carbon electrodes without $\text{Li}_4\text{Ti}_5\text{O}_{12}$ (A-D) before and after the 1.55 V potentiostatic aging at (E-H) room and (I-L) high temperature.

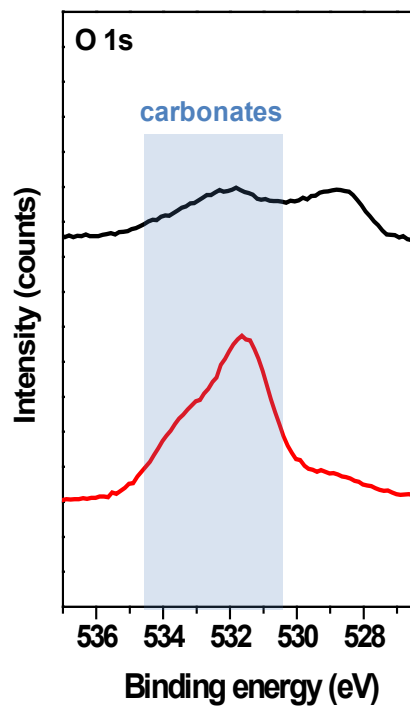


Fig. S5 O 1s XPS spectra of the carbon electrodes without $\text{Li}_4\text{Ti}_5\text{O}_{12}$ after 1.55 V potentiostatic aging at room (black line) and high (red line) temperature.