

Supporting Information

A universal route to fabricate hierarchically ordered macro- /mesoporous oxides with enhanced intrinsic activity

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Contents:

1. Tables
2. Figures
3. References

1. Tables

Table S1 Elemental compositions of hierarchically ordered macro-/mesoporous



Sample	Ce/Ti atomic ratio ^a	Ce/Ti atomic ratio ^b
Ce _{0.2} TiO _x	0.2	0.21
Ce _{0.3} TiO _x	0.3	0.33
Ce _{0.5} TiO _x	0.5	0.51
Ce _{0.7} TiO _x	0.7	0.73
Ce _{1.0} TiO _x	1.0	1.06

^a: The data are the nominal Ce/Ti atomic ratios in Ce–Ti mixed oxides.

^b: The data were obtained by quantitatively analyzing distribution mapping results.

Table S2 BET surface areas, pore volumes and average pore sizes of the hierarchically ordered macro-/mesoporous Ce_aTiO_x.

Sample	BET surface area			Pore volume			Average pore size	
	[m ² g ⁻¹]			[cm ³ g ⁻¹]			[nm]	
	Macropore	Mesopore	Total	Macropore	Mesopore	Total	Macropore ^a	Mesopore
	(> 50 nm)	(≤ 50 nm)		(> 50 nm)	(≤ 50 nm)		(> 50 nm)	(≤ 50 nm)
Ce _{0.2} TiO _x	1.18	55.15	56	0.0282	0.1112	0.140	195	8.2
Ce _{0.3} TiO _x	2.25	44.99	46	0.0484	0.0842	0.133	197	11.2
Ce _{0.5} TiO _x	1.64	40.88	43	0.0341	0.0640	0.098	197	9.5
Ce _{0.7} TiO _x	1.38	26.58	28	0.033	0.0622	0.095	194	15.1
Ce _{1.0} TiO _x	1.38	25.72	27	0.031	0.0543	0.085	197	17.0

^a: Estimated according to the SEM and TEM images.

Table S3. BET surface areas, pore volumes, and average pore sizes of the $\text{Ce}_x\text{TiO}_{3-x}$.

Sample	BET surface area	Pore volume	Average pore size
	$[\text{m}^2 \text{ g}^{-1}]$	$[\text{cm}^3 \text{ g}^{-1}]$	$[\text{nm}]$
$\text{Ce}_{0.2}\text{TiO}_{3-x}$	120 ^a	0.337	9.3
$\text{Ce}_{0.3}\text{TiO}_{3-x}$	122 ^a	0.424	14.3
$\text{Ce}_{0.5}\text{TiO}_{3-x}$	79 ^a	0.214	13.7
$\text{Ce}_{0.7}\text{TiO}_{3-x}$	61	0.146	7.9
$\text{Ce}_{1.0}\text{TiO}_{3-x}$	25	0.078	12.0

^a: From Ref. S1.

2. Figures

Figure S1

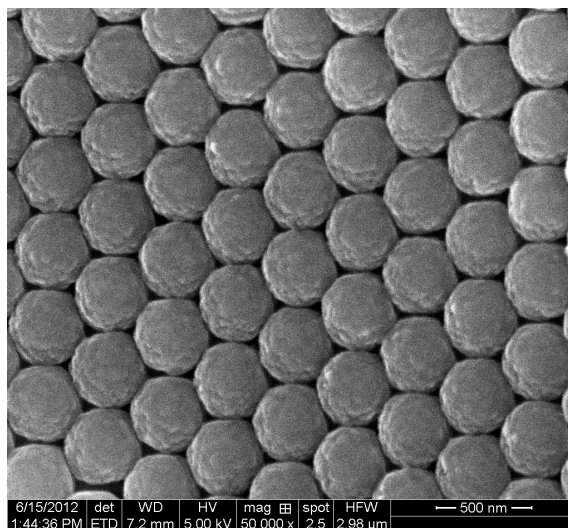


Fig. S1 SEM image of the colloidal crystal template assembled by 389 nm PMMA spheres.

Figure S2

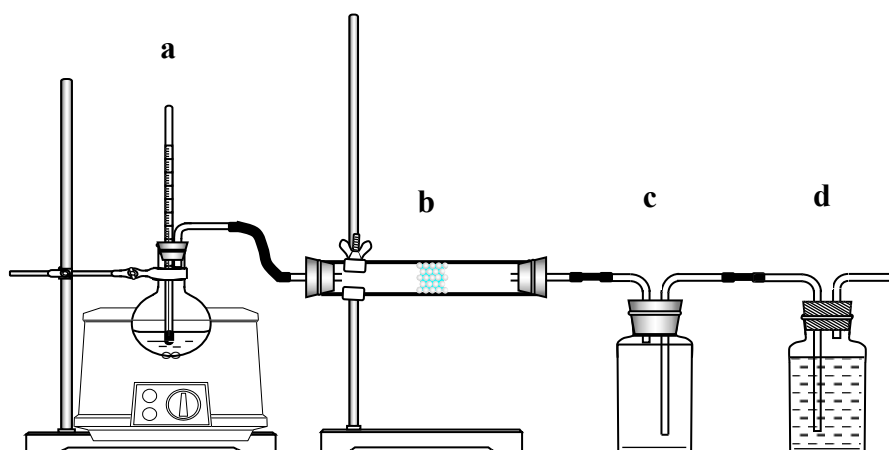


Fig. S2 Schematic illustration of a self-made wet ammonia gas infiltration-precipitation device: (a) wet ammonia gas generator; (b) infiltrated-precipitation device; (c) anti-reverse suction flask; (d) scrubbing bottle.

Figure S3

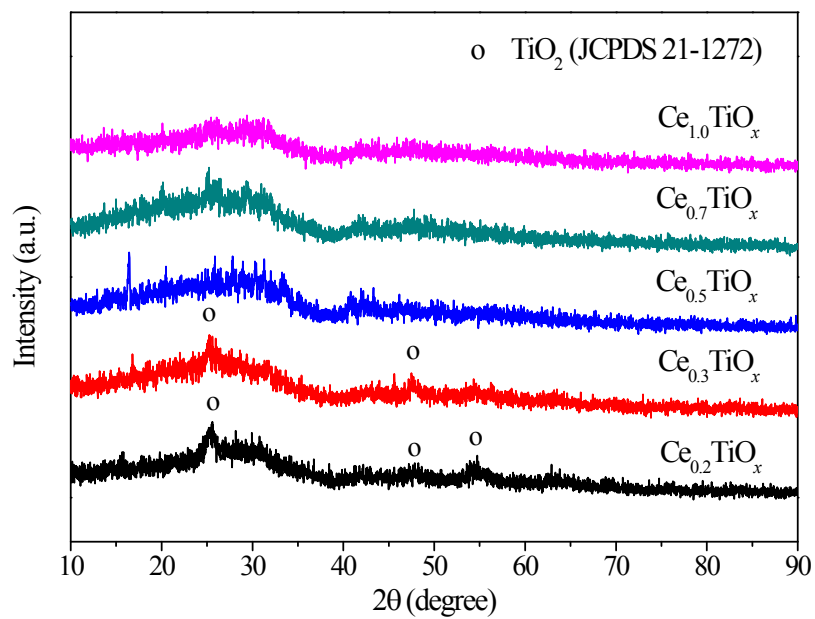


Fig. S3 XRD patterns of the as-fabricated hierarchical macro-/mesoporous Ce_aTiO_x .

Figure S4

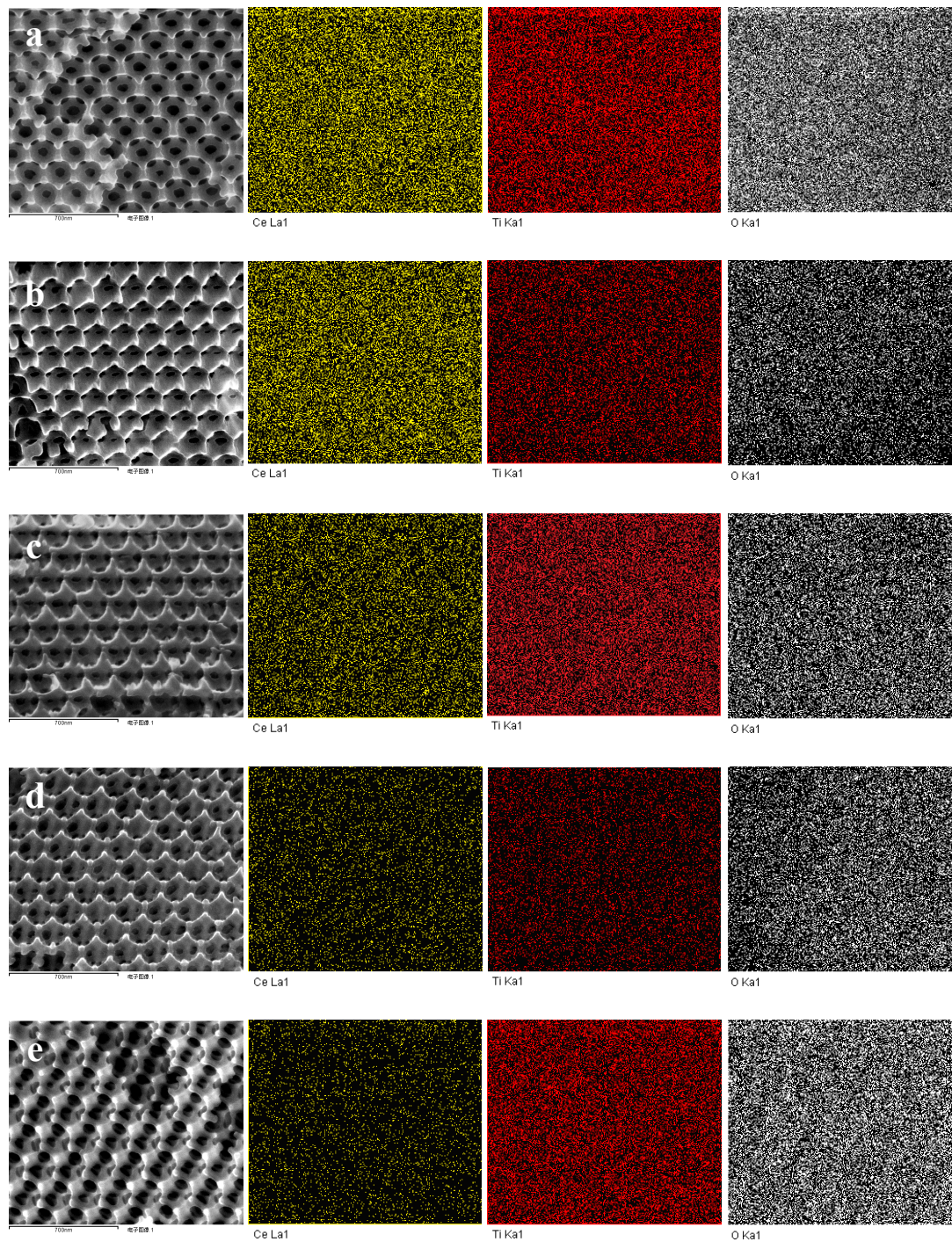
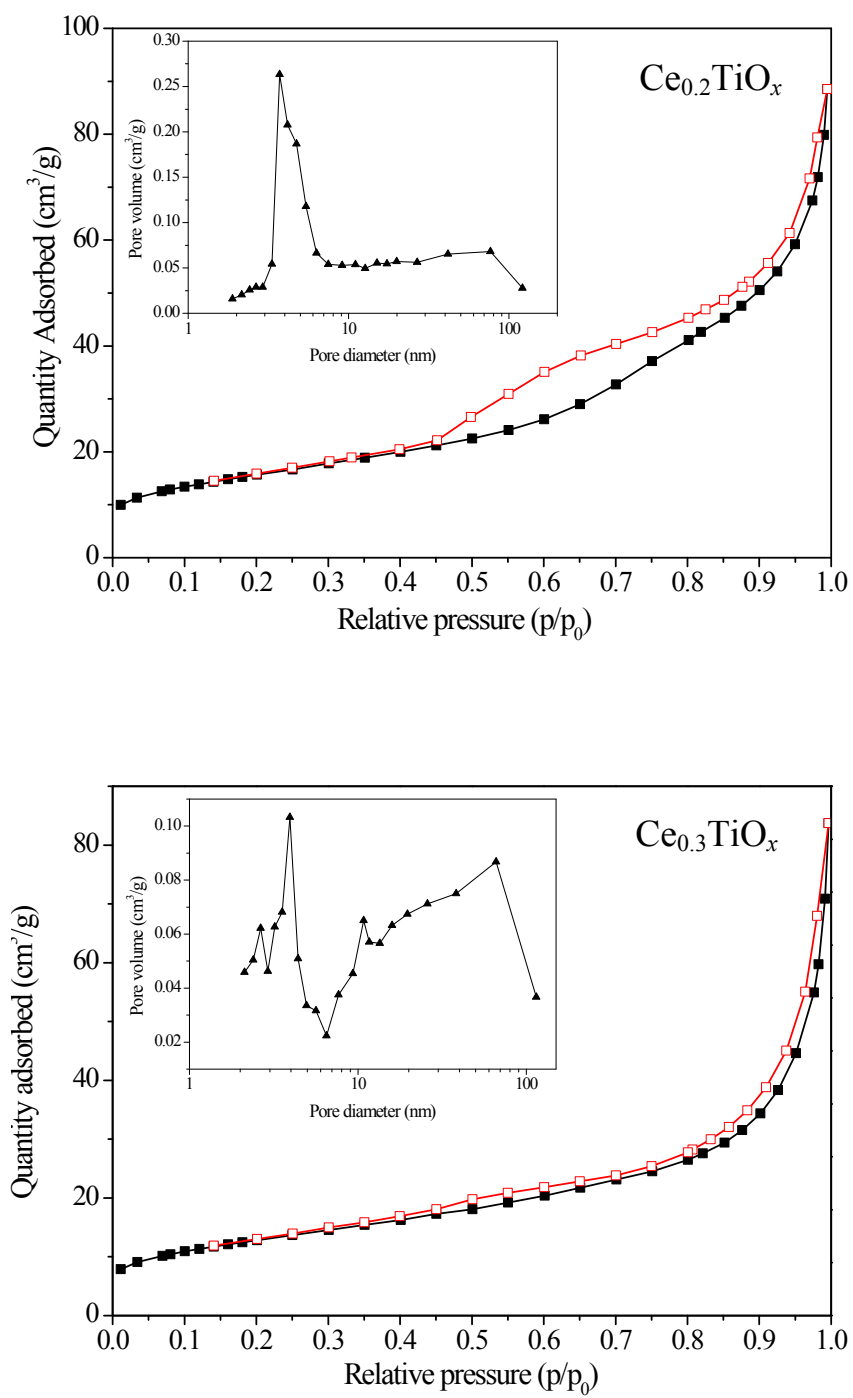
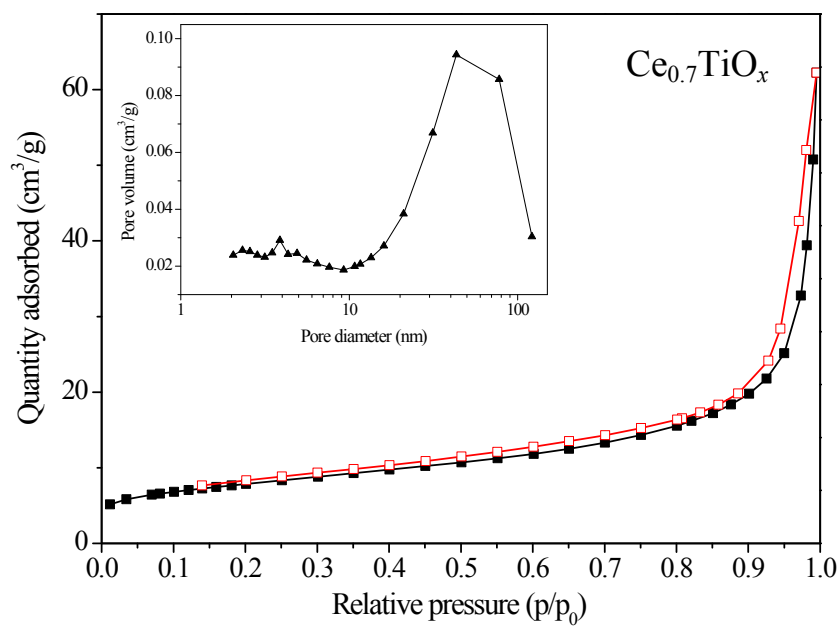
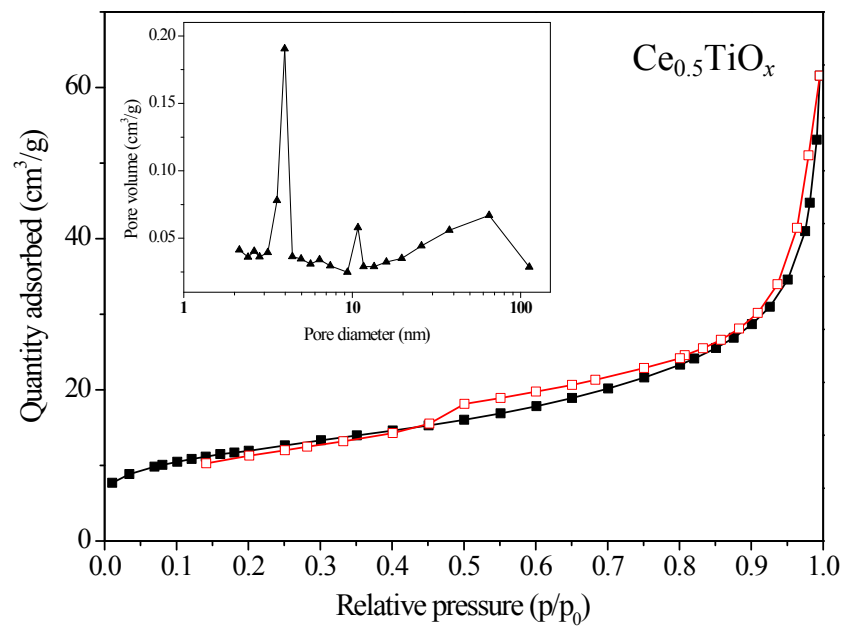


Fig. S4 SEM images and the corresponding distribution maps of Ce, Ti and O elements for the hierarchical macro-/mesoporous Ce_aTiO_x : (a) $\text{Ce}_{0.2}\text{TiO}_x$; (b) $\text{Ce}_{0.3}\text{TiO}_x$; (c) $\text{Ce}_{0.5}\text{TiO}_x$; (d) $\text{Ce}_{0.7}\text{TiO}_x$; (e) $\text{Ce}_{1.0}\text{TiO}_x$.

Figure S5





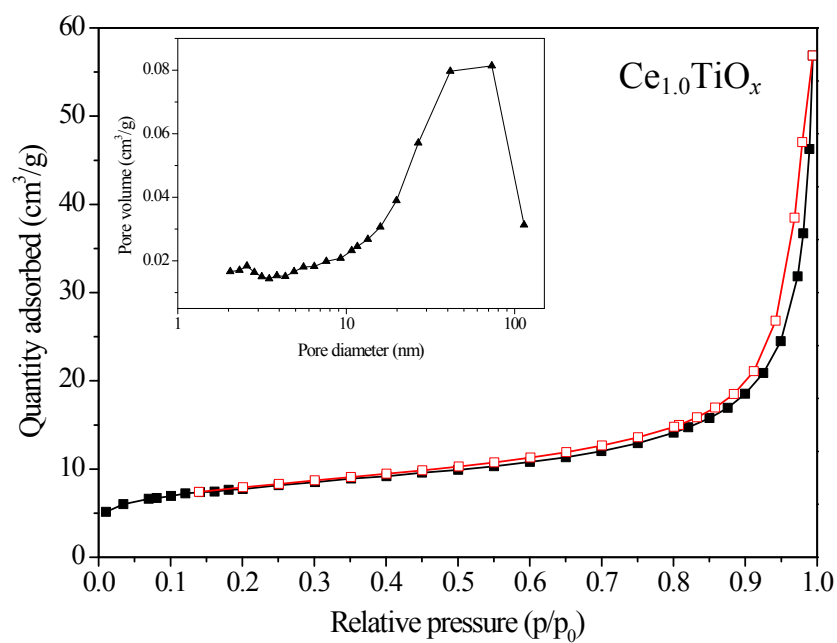


Fig. S5 N_2 adsorption/desorption isotherms and pore size distribution curves (insets)
for Ce_aTiO_x .

Figure S6

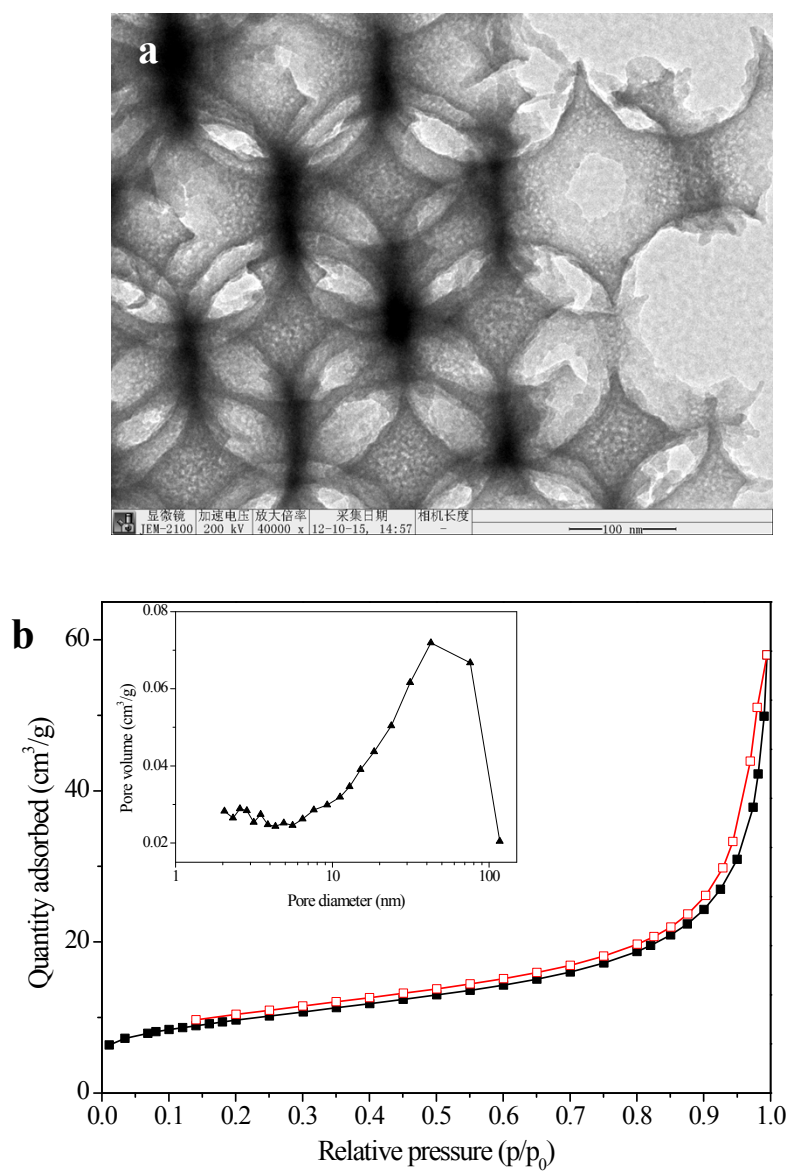


Fig. S6 (a) TEM image and (b) N₂ adsorption/desorption isotherms and pore size distribution curve (inset) of Ce_{0.3}TiO_x-air.

Figure S7

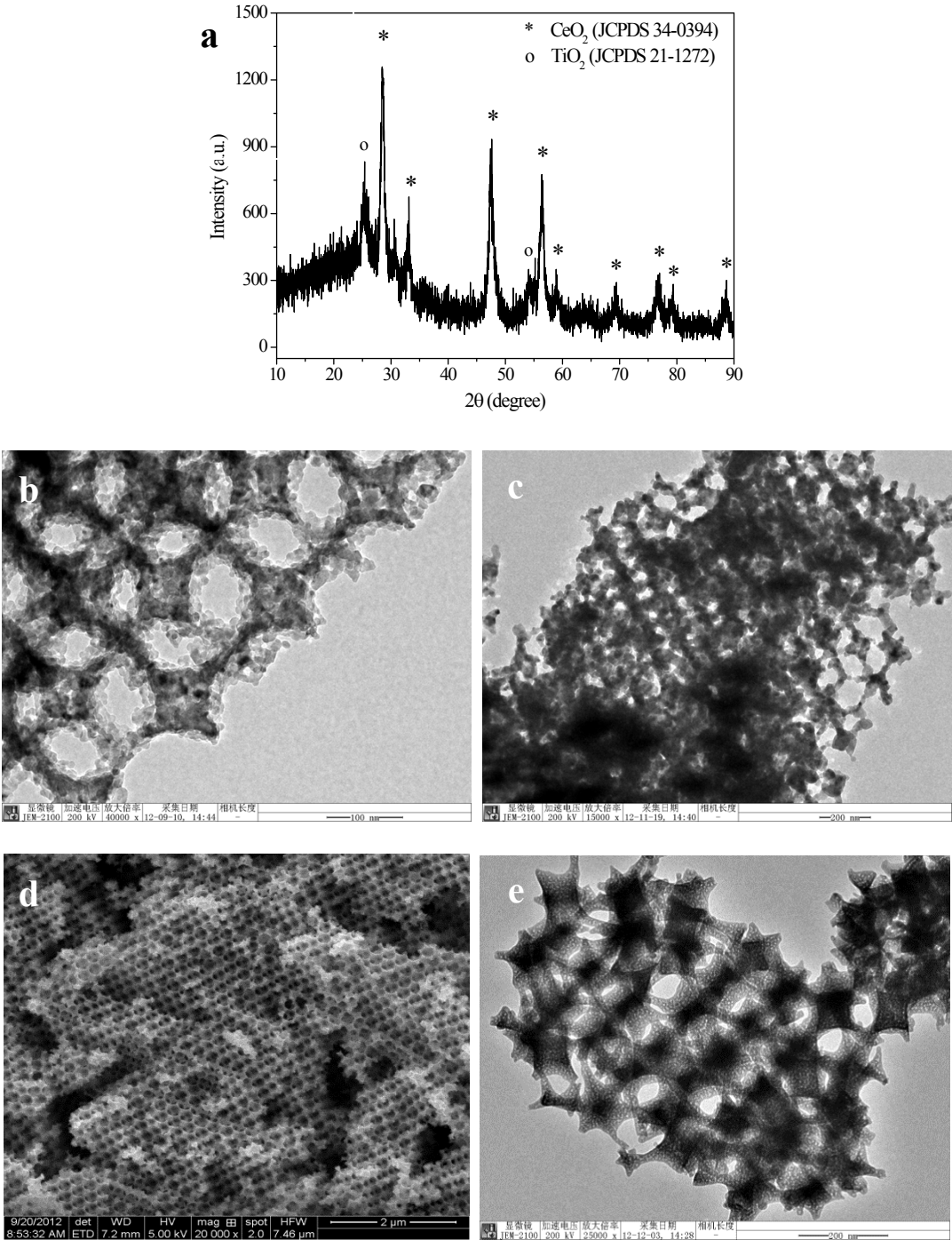


Fig. S7 (a) XRD pattern and (b) TEM image of $\text{Ce}_{0.3}\text{TiO}_x$ -550, (c) TEM image of $\text{Ce}_{0.3}\text{TiO}_x$ -650, (d) SEM and (e) TEM images of $\text{Ce}_{1.0}\text{TiO}_x$ after a mild pressing and grinding.

Figure S8

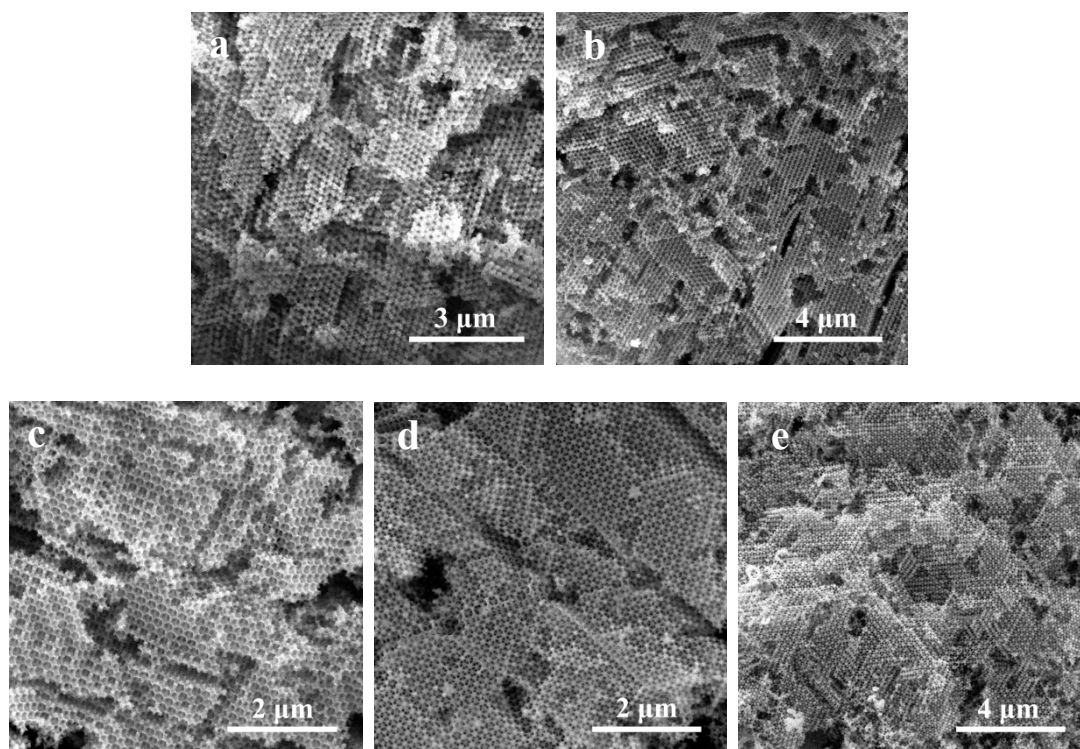


Fig. S8 SEM images of the hierarchically macro-/mesoporous Ce_aTiO_x after SCR reactions: (a) $\text{Ce}_{0.2}\text{TiO}_x$; (b) $\text{Ce}_{0.3}\text{TiO}_x$; (c) $\text{Ce}_{0.5}\text{TiO}_x$; (d) $\text{Ce}_{0.7}\text{TiO}_x$; (e) $\text{Ce}_{1.0}\text{TiO}_x$.

Figure S9

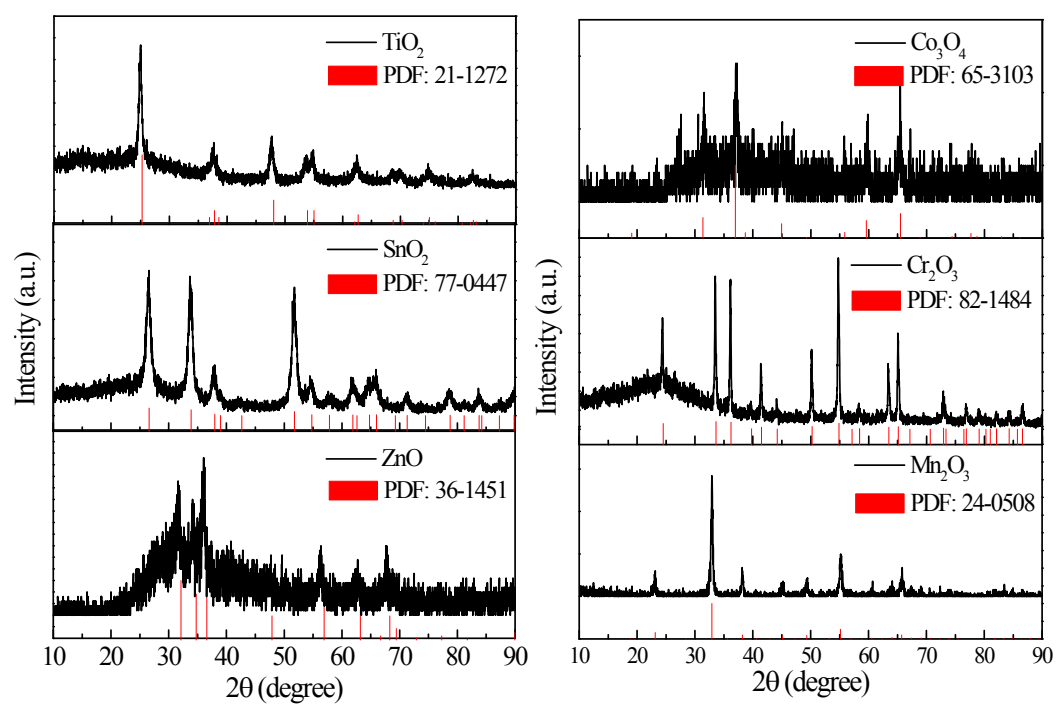


Fig. S9 XRD patterns of the hierarchical macro-/mesoporous unitary oxides.

Figure S10

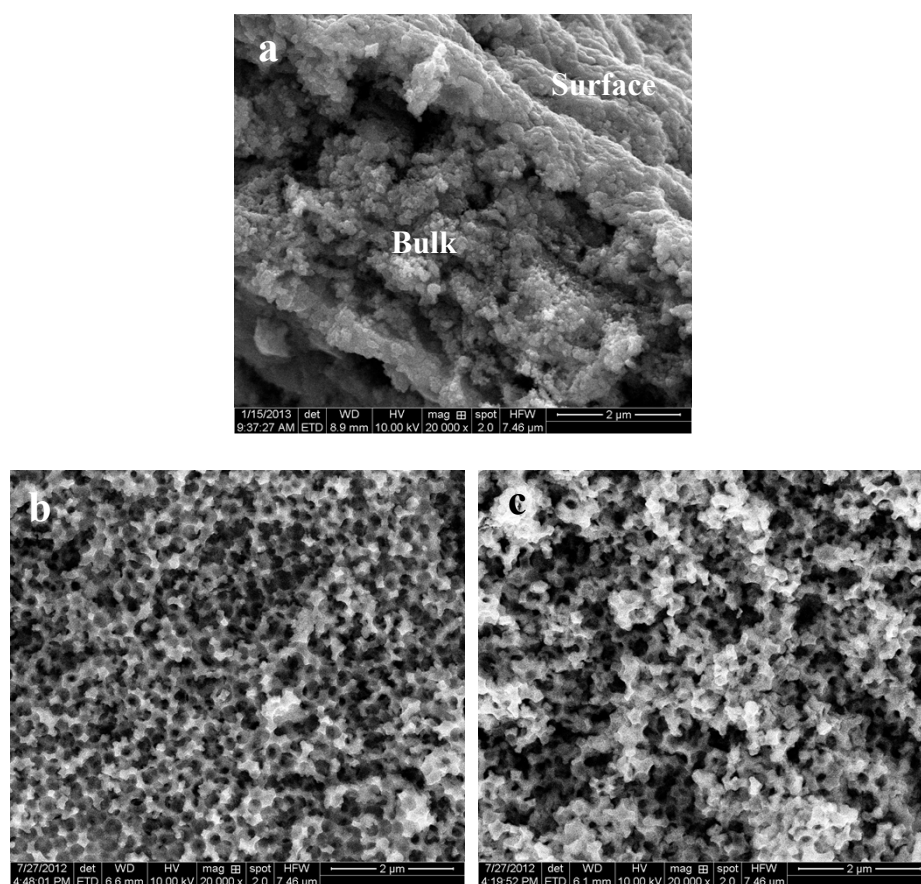


Fig. S10 SEM images of Mn₂O₃ synthesized using pure ammonia gas (a), Ce_{0.3}TiO_x synthesized using ammonia water (b) and Ce_{0.3}TiO_x synthesized without using wet ammonia gas infiltration-precipitation route (c).

3. References

- [S1] P. Li, Y. Xin, Q. Li, Z. P. Wang, Z. L. Zhang and L. R. Zheng, *Environ. Sci. Technol.*, 2012, **46**, 9600–9605.