

**Non-covalent functionalization of reduced graphene oxide using sulfanilic acid
azocromotrop and its application as supercapacitor electrode material**

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Tables

Table S1. Comparative data's of Raman study, electrical conductivity and electrochemical properties of similarly modified graphene.

Sample	I_D/I_G	σ (S m ⁻¹)	Electrochemical performances				Refer-ence
			Cell configuration	SC (F g ⁻¹)	ED (W h Kg ⁻¹)	PD (W kg ⁻¹)	
ADS-G	1.03	-----	3-electrode	210 at 4.5 A g ⁻¹	-----	-----	1
Functionalized graphene	1.1	-----	3-electrode	276 at 0.1 A g ⁻¹	20.0	34.5×10 ³	2
Poly(p-phenylene diamine)/graphene	>1	294.9	2-electrode	248 at 2 A g ⁻¹	5.8	0.5 ×10 ³	3
Graphene oxide hydrogel	1.3	1351	2- electrode	232 at 1 A g ⁻¹	33.5	500	4
SPEEK modified graphene	1.01	-----	3-electrode	476 at 6.6 A g ⁻¹	-----	-----	5
Microwave generated graphene	>1	-----	2- electrode	219 at 1 A g ⁻¹	83.56	1.66 ×10 ³	6
Chemically derived graphene	1.15	431	3- electrode	253 at 0.2 A g ⁻¹	-----	-----	7
3,4-propylenedioxythiophene	>1	22.5	3- electrode	201 at 10 mV s ⁻¹	-----	-----	8
ACA-G (chemical)	-----	-----	3- electrode	148	-----	-----	9
9-anthracene carboxylic acid modified graphene	1.41	-----	3- electrode	577	-----	-----	10
NC-TEG	1.29	-----	3- electrode	635	-----	-----	11
n-doped graphene	1.28	800	Three electrode	295 at 5 A g ⁻¹	40.3	2500	12
O-TEG	1.44	271	3- electrode	175 at 1 A g ⁻¹	-----	-----	13
PCA-graphene	-----	-----	2- electrode	202 at 10 mV s ⁻¹	-----	-----	14
Graphene-polypyrrole nanocomposite	-----	6.56	3- electrode	267 at 100 mV s ⁻¹	94.93	3797.2	15
SPEEK modified graphene	0.49	-----	3- electrode	244 at 2.2 A g ⁻¹	-----	-----	16
TCNQ-graphene	1.04	-----	3- electrode	324 at of 1 A g ⁻¹	86.9	300	17
ANEG	0.32	-----	3- electrode	115 at 4 A g ⁻¹	-----	-----	18
ACA-rGO	1.2	-----	2- electrode*	610 at 0.8 A g ⁻¹	41.3	200	19
Graphene materials (GMs)	-----	100	2- electrode	205 at 0.1 A g ⁻¹	28.5	10000	20
Thermally Reduced graphene	-----	-----	2- electrode	132 at 10 m V s ⁻¹	6.74	190	21
KOH modified graphene	-----	-----	3- electrode	136 at 10mVs ⁻¹	18.9	-----	22
SAC-RGO	0.97	551	3- electrode	366 at 1.2 A g ⁻¹	-----	-----	Prese- nt work

Where, σ , SC, ED and PD refer to electrical conductivity, specific capacitance, energy density and power density, respectively. The 2-electrode asymmetric configuration has been represented by the symbol “*”.

Table S2. Comparative study of electrochemical properties of various electro active materials with asymmetric configurations.

Materials	SC (F g ⁻¹)	ED (W h Kg ⁻¹)	PD (W Kg ⁻¹)	References
MnO ₂ / graphene	37 at 5 mA cm ⁻¹	25.2	100	23
Carbon Nanotube /Graphene and Mn ₃ O ₄ Nanoparticle	72 at 0.5 A g ⁻¹	32.7	-	24
Manganosite-microwave exfoliated graphene oxide	51.5 at 0.1 A g ⁻¹	2.6	9024	25
Polyaniline/Graphene/Carbon Nanotube	107 at 0.2 A g ⁻¹	20.5	2500	26
Ni(OH) ₂ /graphene and RuO ₂ /graphene	87 at 0.5 A g ⁻¹	31	420	27
Graphene-Nickel Cobaltite Nanocomposite	288 at 0.5 A g ⁻¹	7.57	5600	28
3D MnO ₂ /graphene hydroge	59.6 at 1 A g ⁻¹	21.2	8200	29
Reduced graphene oxide/carbon nanotube and carbon fiber paper/polypyrrole	82.4 at 0.5 A g ⁻¹	28.6	15100	30
Nickel oxide/graphene foamand hierarchical porous nitrogen-doped carbon nanotubes	116 at 1 A g ⁻¹	32	700	31
HydrousRuO ₂ -ionic liquid functionalized-chemically modified graphene	-----	19.7	6800	32
Graphene hydrogel (GH) with 3D interconnected pores	-----	23.2	1000	33
Functionalized graphene aerogel composites	175.8	13.9	13300	34
SAC-RGO//RGO	95 at 1.4 A g ⁻¹	25.86	980	Present work

Figures

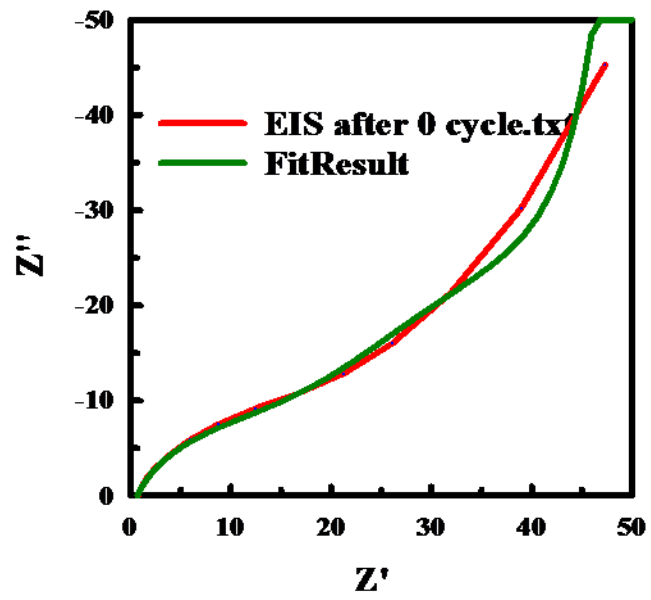


Fig. S1. Z-view fitted curve of SAC-RGO //RGO at 1st cycle

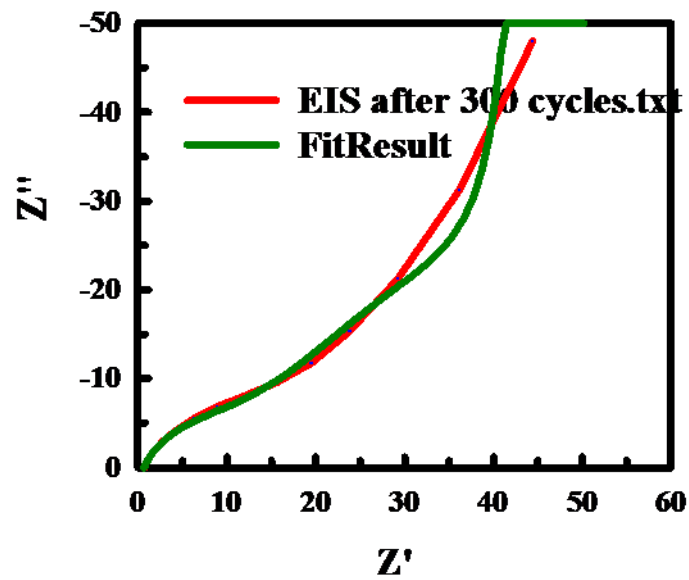


Fig. S2. Z-view fitted curve of SAC-RGO //RGO after 300 cycles

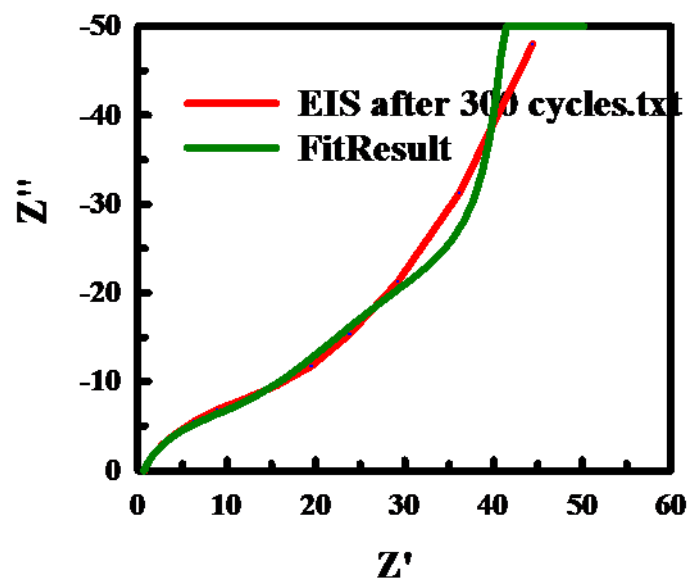


Fig. S3. Z-view fitted curve of SAC-RGO //RGO after 1000 cycles

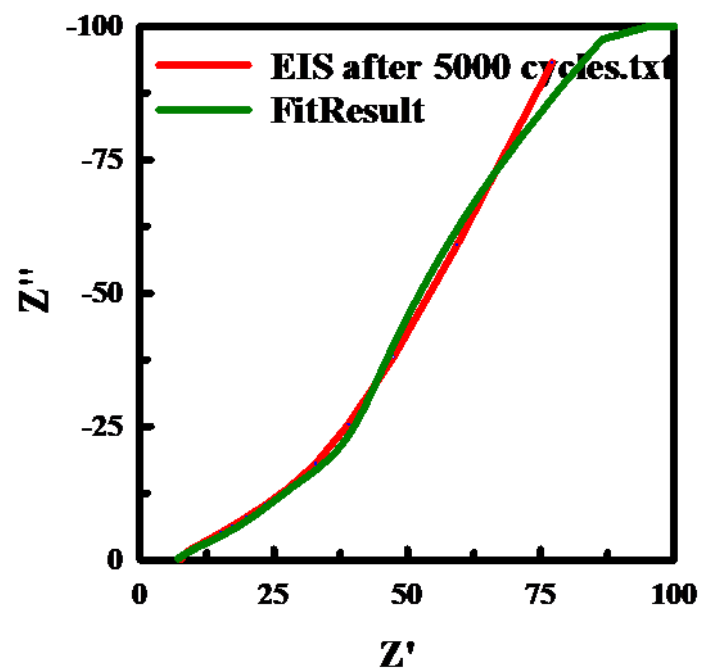


Fig. S4. Z-view fitted curve of SAC-RGO //RGO after 1000 cycles

Reference supporting information

1. D. S. Yu, T. Kuila, N. H. Kim, P. Khanra, J. H. Lee, *Carbon*, 2013, **54**, 310.
2. Z. Lin, Y. Liu, Y. Yao, O. J. Hildreth, Z. Li, K. Moon, C. Wong, *J. Phys. Chem. C*, 2011, **115**, 7120.
3. Jaidev, S. Ramaprabhu, *J. Mater. Chem.*, 2012, **22**, 18775.
4. V. H. Luan, H. N. Tien, L. T. Hoa, N. Thi M. Hien, E.-S. Oh, J. S. Chung, E. J. Kim, W. M. Choi, B.-S. Kong, S. H. Hur, *J. Mater. Chem. A*, 2013, **1**, 208.
5. T. Kuila, A. K. Mishra, P. Khanra, N. H. Kim, Md. E. Uddin, J. H. Lee, *Langmuir*, 2012, **28**, 9825.
6. B. Subramanya, D. K. Bhat, *New J. Chem.* DOI: 10.1039/c4nj01359j
7. B. Rajagopalan, J. S. Chung, *Nanoscale Res. Lett.*, 2014, **9**, 535.
8. N. A. Kumar, H. J. Choi, A. Bund, J.-B. Baek, Y. T. Jeong, *J. Mater. Chem.*, 2012, **22**, 12268.
9. S. Bose, T. Kuila, A. K. Mishra, N. H. Kim, J. H. Lee, *Nanotechnology* 2011, **22**, 405603.
10. P. Khanra, T. Kuila, S. H. Bae, N. H. Kim, J. H. Lee, *J. Mater. Chem.*, 2012, **22**, 24403.
11. Y. Yan, T. Kuila, N. H. Kim, S. H. Lee, J. H. Lee, *Carbon*, 2015, **85**, 60.
12. H. Zhang, T. Kuila, N. H. Kim, D. S. Yu, J. H. Lee, *Carbon*, 2014, **69**, 66.
13. Y. Yan, T. Kuila, N. H. Kim, J. H. Lee, *Carbon*, 2014, **74**, 195.
14. S. Ghosh, X. An, R. Shah, D. Rawat, B. Dave, S. Kar, S. Talapatra, *J. Phys. Chem. C*, 2012, **116**, 20688.
15. S. Bose, N. H. Kim, T. Kuila, K. Lau, J. H. Lee, *Nanotechnology*, 2011, **22**, 295202
16. T. Kuila, P. Khanra, N. H. Kim, J. K. Lim, J. H. Lee, *J. Mater. Chem. A*, 2013, **1**, 9294.

17. P. Khanra, C. N. Lee, T. Kuila, N. H. Kim, M. J. Parka, J. H. Lee, *Nanoscale*, 2014, **6**, 4864.
18. T. Kuila, P. Khanra, N. H. Kim, S. K. Choi, H. J. Yun, J. H. Lee, *Nanotechnology*, 2013, **24**, 365706.
19. P. Khanra, Md. E. Uddin, N. H. Kim, T. Kuila, S. H. Lee, J. H. Lee, *RSC Adv.*, 2015, **5**, 6443.
20. Y. Wang, Z. Shi, Y. Huang, Y. Ma, C. Wang, M. Chen, Y. Chen, *J. Phys. Chem. C*, 2009, **113**, 13103.
21. L. T. Le, M. H. Ervin, H. Qiu, B. E. Fuchs, W. Y. Lee, *Electrochem. Commun.* 2011, **13**, 355.
22. Y. Li, M. V. Zijll, S. Chiang, N. Pan, J. P. Sources, 2011, **196**, 6003.
23. J. Cao, Y. Wang, Y. Zhou, J. H. Ouyang, D. Jia, L. Guo, *J. Electroanal. Chem.* 2013, **689**, 201.
24. H. Gao, F. Xiao, C. B. Ching, H. Duan, *ACS Appl. Mater. Interfaces*, 2012, **4**, 7020.
25. D. Antiohosa, K. Pingmuangb, M. S. Romanoa, S. Beirnea, T. Romeoa, P. Aitchisonc, A. Minettd, G. Wallacea, S. Phanichphantb, J. Chena, *Electrochim. Acta* 2013, **101**, 99.
26. J. Shen, C. Yang, X. Li, G. Wang, *ACS Appl. Mater. Interfaces*, 2013, **5**, 8467.
27. H. Wang, Y. Liang, T. Mirfakhrai, Z. Chen, H. S. Casalongue, H. Dai, *Nano Res.* 2011, **4**, 729.
28. H. Wang, C. M. B. Holt, Z. Li, X. Tan, B. S. Amirkhiz, Z. Xu, B. C. Olsen, T. Stephenson, D. Mitlin, *Nano Res.* 2012, **5**, 605.
29. S. Wu, W. Chen, L. Yan, *J. Mater. Chem. A*, 2014, **2**, 2765.
30. C. Yang, J. Shen, C. Wang, H. Fei, H. Bao, G. Wang, *J. Mater. Chem. A*, 2014, **2**, 1458.
31. H. Wang, H. Yi, X. Chen, X. Wang, *J. Mater. Chem. A*, 2014, **2**, 3223.

32. B. G. Choi, S. J. Chang, H. W. Kang, C. P. Park, H. J. Kim, W. H. Hong, S. G. Lee, Y. S. Huh, *Nanoscale*, 2012, **4**, 4983-4988.
33. H. Gao, F. Xiao, C. Bun Ching, H. Duan, *ACS Appl. Mater. Interfaces*, 2012, **4**, 2801.
34. Z. Yu, M. McInnis, J. Calderon, S. Seal, L. Zhai, J. Thomas, *Nano energy*, 2015, **11**, 611.