

The average values of tensor components of transport properties vs. chemical potential (μ) for pristine graphene and its derivatives (H_2S adsorbed on bridge, hollow and top-site configuration) at 300 and 600 K are shown in Fig.1-4.

It is clear from Fig.1a,b that the p type carrier concentration shows greater average electrical conductivity (σ^{av}) as compared to n type carrier concentration for the two temperatures 300 K and 600 K. The σ^{av} shows a maximum value of $421 \times 10^6 (\Omega.m)^{-1}$, $190 \times 10^6 (\Omega.m)^{-1}$, $290 \times 10^6 (\Omega.m)^{-1}$ and $122 \times 10^6 (\Omega.m)^{-1}$ for pristine graphene, bridge, top and hollow-site configurations at 300 K. When the temperature is increased to 600 K the σ^{av} reduces to $400 \times 10^6 (\Omega.m)^{-1}$, $175 \times 10^6 (\Omega.m)^{-1}$, $276 \times 10^6 (\Omega.m)^{-1}$ and $104 \times 10^6 (\Omega.m)^{-1}$ for pristine graphene, bridge, top and hollow-site configuration. This could be related to the fact that at high temperature the scattering of phonons with electrons increases resulting in decreases the electrical conductivity [1].

The highly oscillatory spectra of the Seebeck coefficient (S^{av}) for four cases (pristine graphene, bridge, hollow and top-site configurations) shows same contribution of electron and hole concentrations. Pristine graphene and hollow-site configuration shows a maximum value of about $210 \mu V / K$ at 300 K and $126 \mu V / K$ at 600 K. The bridge and top-site configuration shows a maximum value of $150 \mu V / K$ and $540 \mu V / K$ for 300 K which reduced to $100 \mu V / K$ and $280 \mu V / K$ at 600 K as shown in Fig.2b.

The thermal conductivity (k^{av}) elucidates maximum value of n type concentration for pristine graphene and its derivative except the hollow-site configuration at 600 K which shows same electron-hole contribution. Here we also present k^{av} in p type region. The pristine graphene shows maximum value of $31 \times 10^2 W / mK$ at 300 K. The H_2S adsorbed on bridge, top and hollow-site results decrease in k^{av} to $11.5 \times 10^2 W / mK$, $19.5 \times 10^2 W / mK$ and $8.0 \times 10^2 W / mK$, respectively. For higher temperatures 600 K, the k^{av} shows significant increase in pristine graphene ($67.5 \times 10^2 W / mK$). The bridge, top and hollow-site configuration enhances the k^{av} to $24 \times 10^2 W / mK$, $35 \times 10^2 W / mK$ and $13 \times 10^2 W / mK$ at 600 K as shown Fig.3b. The greater value of thermal conductivity results lower values of figure of merit (ZT) provided that the lattice thermal conductivity is assumed to be similar.

The power factor ($P = \sigma S^2$) is the important quantity for measuring transport properties of thermoelectric materials. The calculated average power factor P^{av} for pristine graphene and its derivative at different temperatures are shown in Fig. 4a,b. As one moves from 300 K to 600 K the P^{av} of pristine graphene increases from $14 \times 10^{-3} W / mK^2$ to $27 \times 10^{-3} W / mK^2$ while for the bridge-site configuration it reduces from $31 \times 10^{-3} W / mK^2$ to $27 \times 10^{-3} W / mK^2$. For the top-site it shows a dramatic increase with the shift of $40.5 \times 10^{-3} W / mK^2$. The hollow-site increases from $14 \times 10^{-3} W / mK^2$ to $22 \times 10^{-3} W / mK^2$ as shown in Fig.4b.

[1] J. Hone, M. Whitney, C. Piskoti, and A. Zettl, Phys. Rev. B 59, R2514–R2516 (1999)

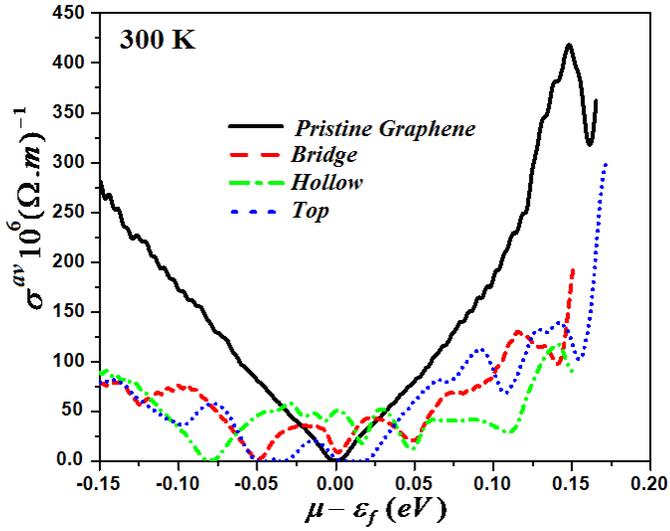
Figure Captions

Fig.1: Calculated average electrical conductivity (σ^{av}) of pristine graphene, H₂S adsorb on bridge site, H₂S adsorb on hollow site and H₂S adsorb on top site.

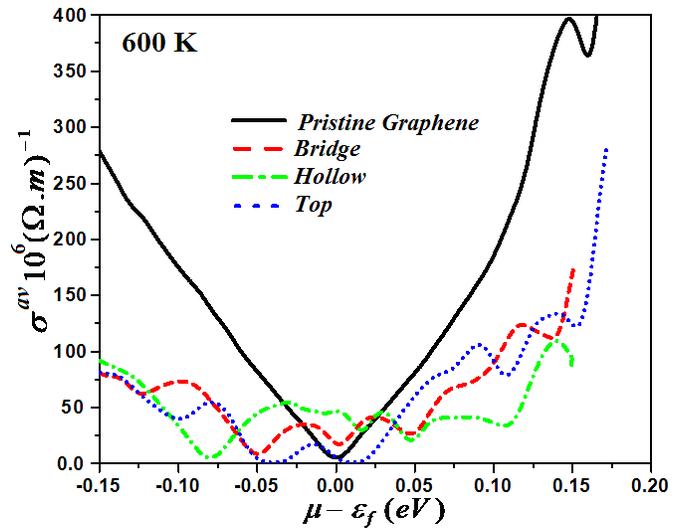
Fig.2: Calculated average Seebeck coefficient (S^{av}) of pristine graphene, H₂S adsorb on bridge site, H₂S adsorb on hollow site and H₂S adsorb on top site.

Fig.3: Calculated average thermal conductivity (k^{av}) of pristine graphene, H₂S adsorb on bridge site, H₂S adsorb on hollow site and H₂S adsorb on top site.

Fig.4: Calculated average value of power factor (P^{av}) of pristine graphene, H₂S adsorb on bridge site, H₂S adsorb on hollow site and H₂S adsorb on top site

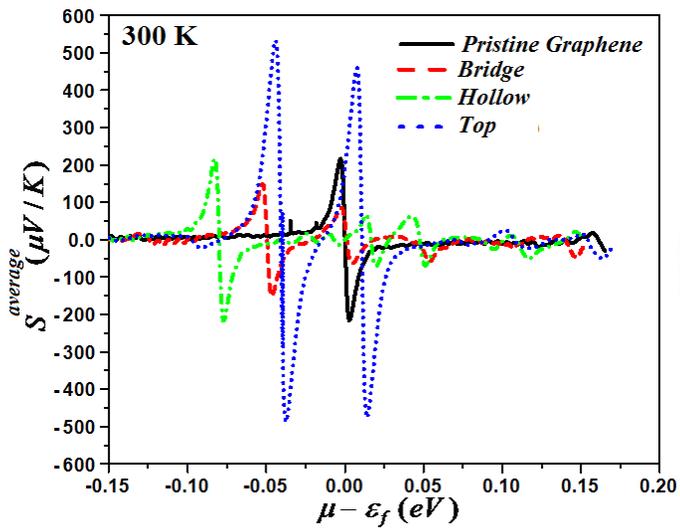


(a)

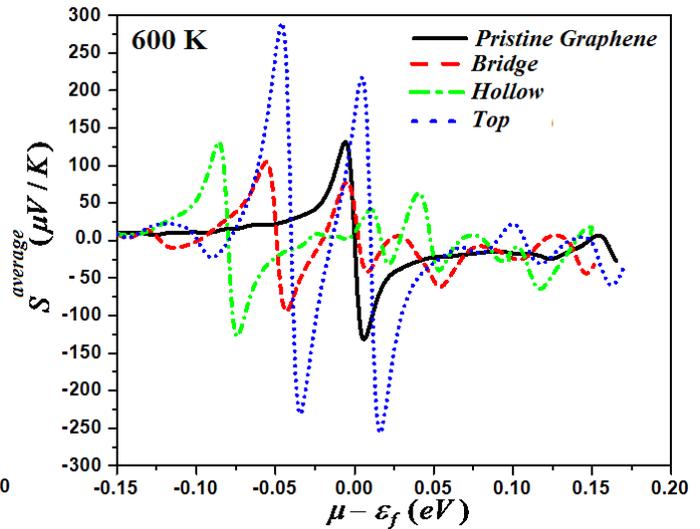


(b)

Fig.1



(a)



(b)

Fig.2

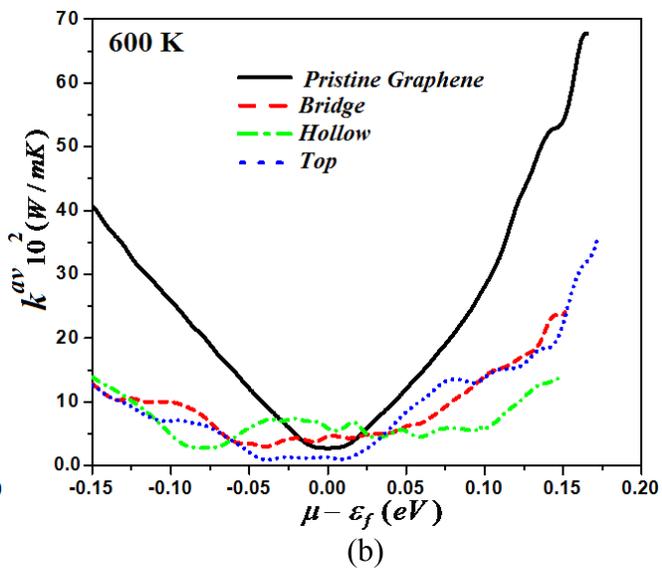
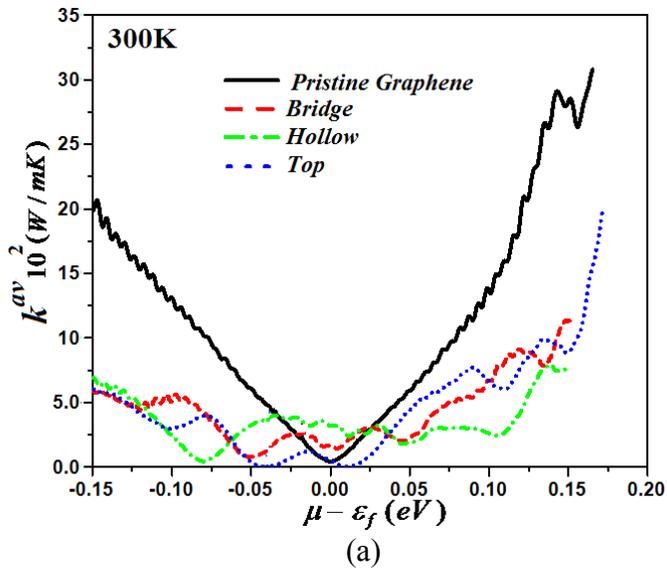


Fig.3

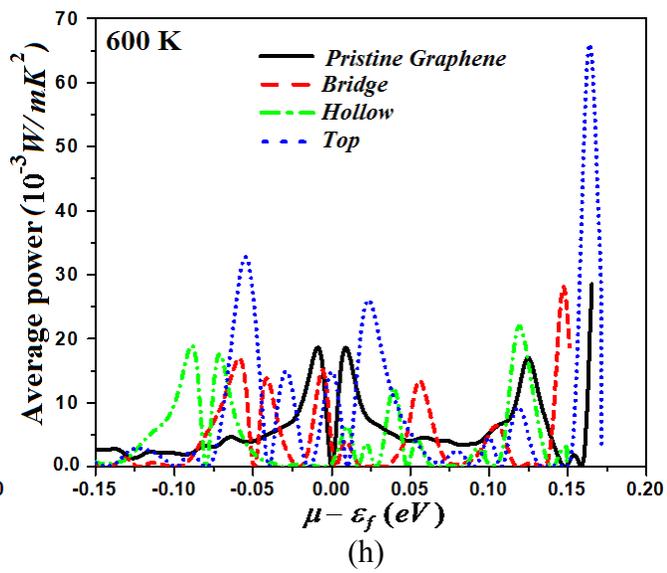
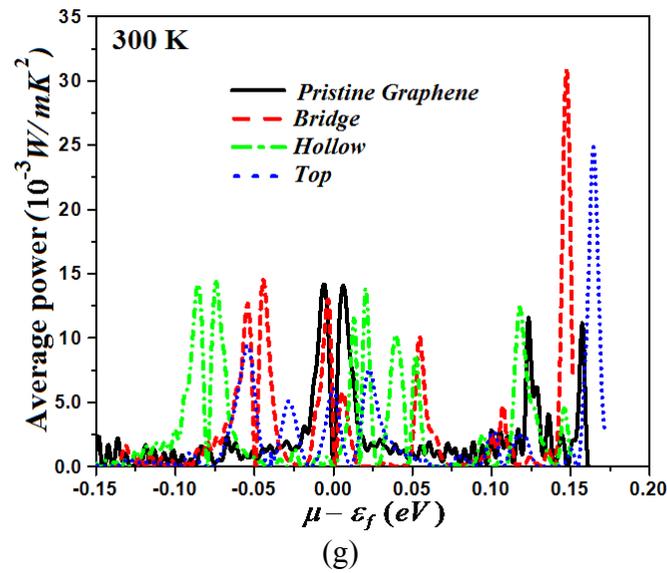


Fig.4