## **Supporting Information**

Click Synthesis of Podands Triazole-linked Gold Nanoparticles as Highly Selective and Sensitive Colorimetric Probes for Lead (II) ions

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#### Synthesis of 2-azidoethanol (AZE)

A suspension of 2-chloroethanol (1.61g, 0.02mol) and NaN<sub>3</sub> (2.60 g, 0.04 mmol) in DMF (20 mL) was stirred for 5 h at 90 °C. The mixture was cooled and then diluted with ethyl acetate (20 mL) and washed with water (3 × 10 mL). The organic phase was dried over magnesium sulphate, filtered and the solvent was removed under reduced pressure to afford the product 2-azidoethanol (AZE). IR:  $2110(-N_3)$ , 2935, 2874(-CH<sub>2</sub>-) cm<sup>-1</sup>.

#### Synthesis of 2-(2-(2-aminoethoxy)ethoxy)ethanol (AEE)

To a stirring solution of 2-(2-(2-azidoethoxy)ethoxy)ethanol (175 mg, 1 mmol) in fresh THF at room temperature was added PPh<sub>3</sub> (577 mg, 2.2 mmol). The mixture was stirred for 5 h and then added water (0.02 mL). 1 M HCl was used to regulate the solution until the value of pH is 2. The solution was extracted by Et<sub>2</sub>O ( $3 \times 10$ mL). Then the aqueous layer was regulated to pH=11 by 1 M NaOH solution. The solution was extracted by Et<sub>2</sub>O ( $3 \times 10$  mL) and the combined organic layers were dried over K<sub>2</sub>CO<sub>3</sub> and then concentrated under reduced pressure to afford the colorless liquid product 2-(2-(2-aminoethoxy)ethoxy)ethanol (AEE). IR: 3366,  $3216(-NH_2)$ , 2871, 1120(C–O) cm<sup>-1</sup>.

#### **Preparation of AZE-Au NPs**

2-azidoethanol (0.5 mL,  $10^{-3}$  M) was added to PA-Au NPs colloids (100 mL) in the condition of mixture of CuSO<sub>4</sub> and sodium ascorbic stirring for 3 h at 60°C, then the 2-azidoethanol modified Au NPs (AZE-Au NPs) were obtain. The synthesized

AZE-Au NPs were purified by centrifugation and redispersion in water twice. The particle concentration of the Au NPs (ca. 68.18 nM) was determined.<sup>1</sup>

### **Preparation of AEE-Au NPs**

HAuCl<sub>4</sub> 100 mL aqueous solution of (1mL, 8.9 mg/mL) and 2-(2-(2-aminoethoxy)ethanol(10<sup>-3</sup> M, 0.5 mL) which had been ultrasonic with CS<sub>2</sub> stoichiometry, were mixed stirring for 20 min, then the HAuCl<sub>4</sub> was reduced by fresh sodium borohydride solution (NaBH<sub>4</sub>, 1 mL, 4 mg/mL). After addition of NaBH<sub>4</sub>, stirred the gold colloidal solution for 2 h at room temperature to obtained 2-(2-(2-aminoethoxy)ethoxy)ethanol modified Au NPs (AEE-Au NPs). The synthesized AEE-Au NPs were purified by centrifugation and redispersion in water twice. The particle concentration of the Au NPs (ca. 62.73 nM) was determined.1

# Supplementary figures



Fig. S1 UV-vis absorption values of OEG-Au NPs at 512 nm by days.



Fig. S2 UV-vis absorption values of OEG-Au NPs affected by different pH at 512 nm.



Fig. S3 UV-vis absorption values of 10 batches of OEG-Au NPs at 512 nm.



Fig. S4 TEM images and the colours of dispersed OEG-Au NPs (a), and addition of  $Pb^{2+}$  (b), respectively. The final concentration of  $Pb^{2+}$  in the aggregated Au colloid is 14  $\mu$ M.



Fig. S5 The size distribution of OEG-Au NPs. 100 particles are measured to get the size distribution.



Fig. S6 The colors (a) and UV-vis spectra (b) of 1.5 mL OEG-Au NPs after adding 0.25 mL of  $10^{-4}$  M alkali metal and alkaline earth metal ions.



Fig. S7 TEM image (a) and the size distribution (b) of dispersed AZE-Au NPs. 150 particles are measured to get the size distribution.



Fig. S8 TEM image (a) and the size distribution (b) of dispersed AEE-Au NPs. 100 particles are measured to get the size distribution.

5

6

Size( nm)

9

10

11

12

8



Fig. S9 UV-vis spectra of 1.5 mL PA-Au NPs colloids after adding 0.25 mL of  $10^{-4}$  M transition metal ions.



Fig. S10 UV-vis spectra of 1.5 mL AZE-Au NPs colloids after adding 0.25 mL of  $10^{-4}$  M transition metal ions.



Fig. S11 UV-Vis spectra of 1.5 mL AEE-Au NPs colloids after adding 0.25 mL of  $10^{-4}$  M transition metal ions.

Reference

1 M. M. Maye, L. Han, N. N. Kariuki, N. K. Ly, W. B. Chan, J. Luo and C. J. Zhong, *Analytica Chimica Acta*, 2003, **496**, 17-27.